Supplementary Information for: 'Viologen-modified Electrodes for Protection of Hydrogenases from High Potential Inactivation while Performing H₂ Oxidation at Low Overpotential' by Alaa A. Oughli, Marisela Vélez, James Birrell, Wolfgang Schuhmann, Wolfgang Lubitz, Nicolas Plumeré and Olaf Rüdiger



Figure S1. Electrode characterization. A) CVs of a gold electrode modified covalently with viologen with scan rates of 5, 25, 50, 75, 100, 150, 175, 200 and 250 mV s⁻¹. B) Anodic and cathodic peak currents versus scan rate showing a linear response. Conditions: aqueous buffer pH 7, 25°C, N₂.



Figure S2. Viologen stability of the film at negative potentials. 100 consecutive CVs of a Ag electrode modified with a viologen film. Conditions: 25° C, aqueous buffer pH 7, N₂, 0.1 V s⁻¹.



Figure S3. AFM topographic images, in 3D representation of a bare (A), viologen modified (B) and protein modified (C) $1x1 \ \mu m^2$ area of a Au-coated substrate. A line profile is shown on the left.



Figure S4. Determination of the E_{sw} using the derivative of the reversed scan of the CV of DdHydAB activity. The second more positive peak is attributed to another inactivation process, which is still currently under investigation.



Figure S5. Chronoamperometry experiment of *Dd*HydAB in DET mode on PGE (red) and on a viologen-modified GCE (blue). Potentials of 450, 500, 550, 600 mV vs. SHE were applied for 2 min each and a potential of -170 mV vs. SHE was applied for 5 min after each high potential to reactivate the enzyme. Conditions: 25° C, aqueous buffer pH 7, 1000 rpm, 100% H₂.



Figure S6. ¹H NMR spectra of the viologen derivative (2) in D_2O (blue), in HCl 0.5 N solution in D_2O (red) and with addition of NaNO₂ (green). The new peaks after NaNO₂ addition appear as a result of forming the diazonium salt of the viologen.