

Complete On/Off Responsive ParaCEST MRI Contrast Agents for Copper and Zinc

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SUPPORTING INFORMATION

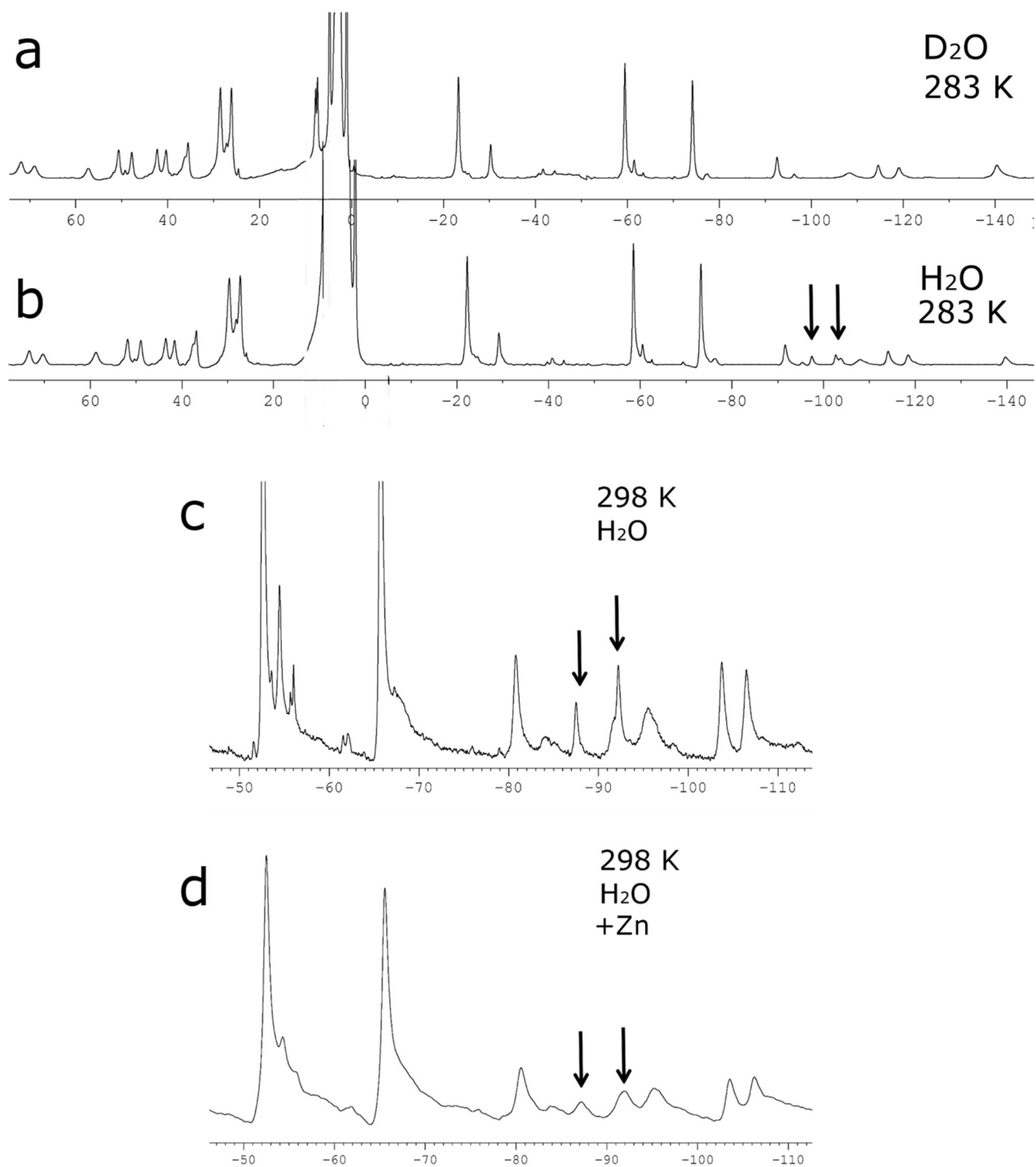


Fig. S1 a) ^1H -NMR spectra of Tm-DOTAm-py in D_2O and b) in H_2O ($B_0 = 14\text{ T}$, $10\text{ }^\circ\text{C}$, $\text{pH } 7.2$); c) ^1H -NMR spectra of Tm-DOTAm-py in H_2O at 298K ($B_0 = 14\text{ T}$, $\text{pH } 7.2$); d) ^1H -NMR spectra of Tm-DOTAm-py in H_2O at 298K in presence of one equivalent of Zn^{2+} ($B_0 = 14\text{ T}$, $\text{pH } 7.2$).

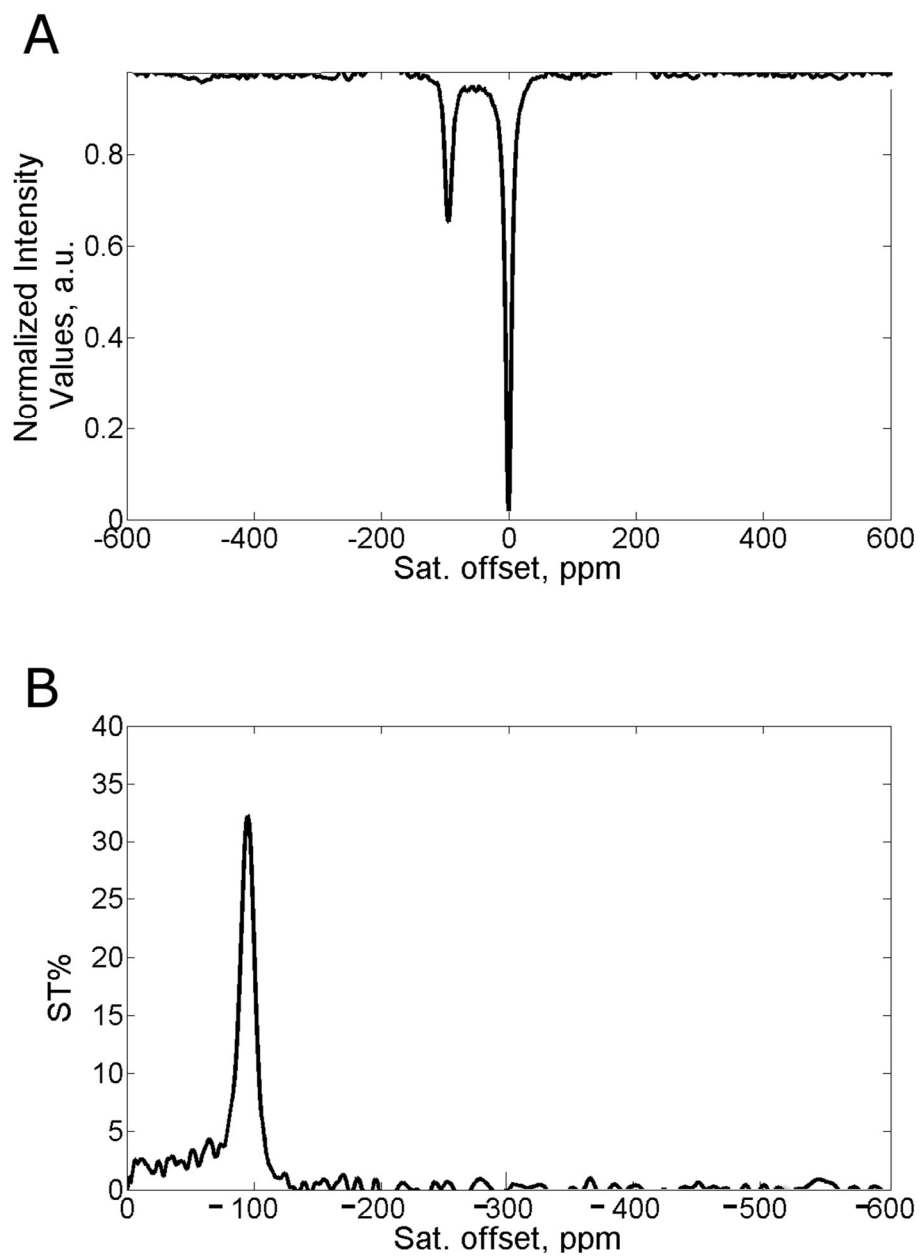


Fig. S2 a) Z- and b) ST- spectra of Tm-DOTAm-py. Experimental condition: [Tm-DOTAm-py] = 10 mM, [HEPES] = 3.8 mM, pH = 7.2, B_1 = 24 μ T, T = 21 $^{\circ}$ C.

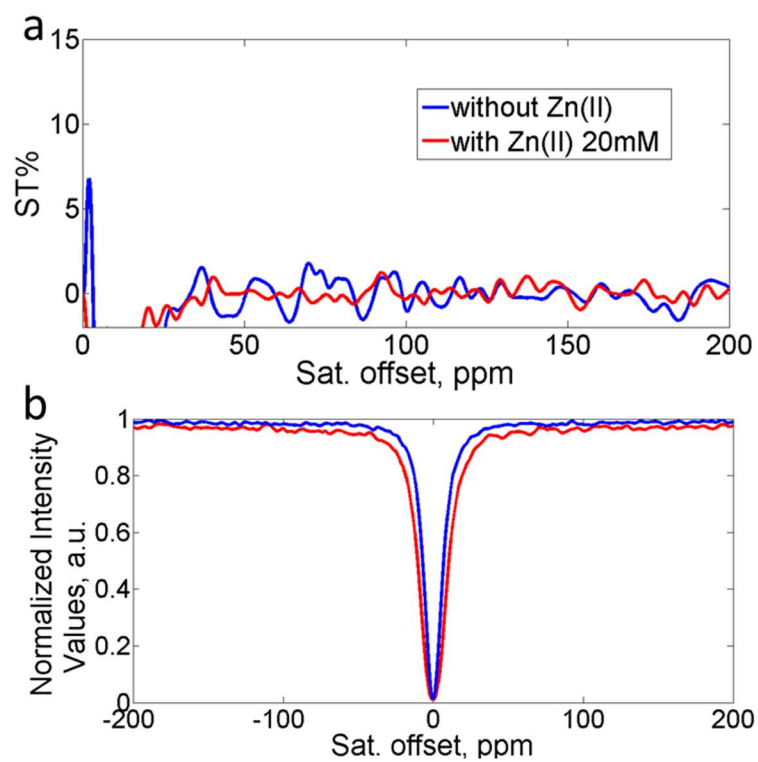


Fig. S3 a) Saturation transfer % (ST%) spectra and b) Z-spectra of Eu-DOTAm-py in the absence and presence of 20 mM ZnCl_2 . Experimental conditions: $[\text{Eu-DOTAm-py}] = 10 \text{ mM}$, $[\text{HEPES}] = 3.8 \text{ mM}$, $\text{pH} = 7.2$, $B_1 = 24 \text{ } \mu\text{T}$, $T = 21 \text{ } ^\circ\text{C}$.

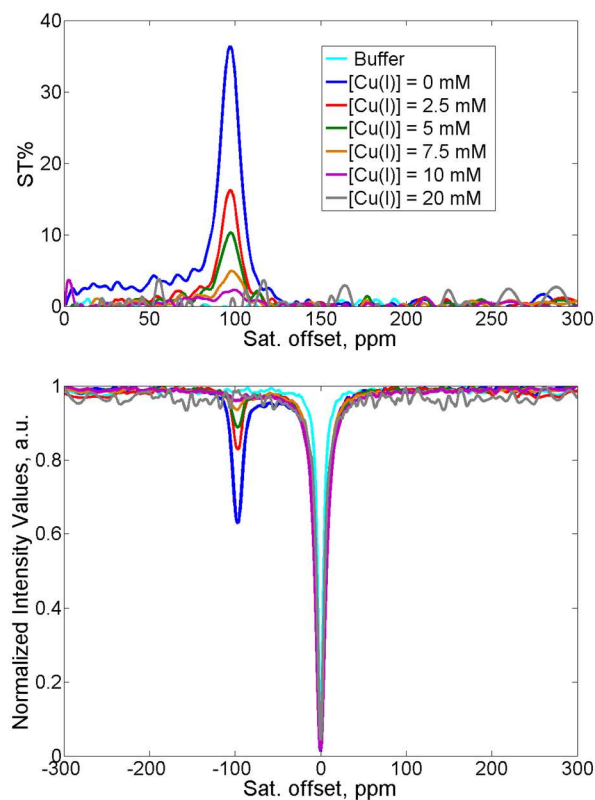


Fig. S4 a) Saturation transfer % (ST%) spectra and b) Z-spectra of Tm-DOTAm-py upon addition of different concentration of $[(CH_3CN)_4Cu]PF_6$. Experimental conditions: $[Tm-DOTAm-py] = 10$ mM, $[HEPES] = 3.8$ mM, $pH = 7.2$, $B_1 = 24$ μT , $T = 21$ $^{\circ}C$.

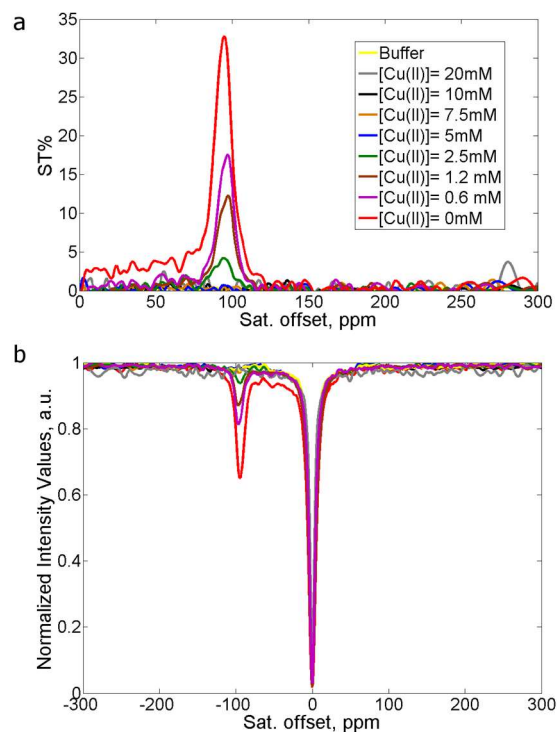


Fig. S5 a) Saturation transfer % (ST%) spectra and b) Z-spectra of Tm-DOTAm-py upon addition of different concentration of $CuCl_2$. Experimental conditions: $[Tm-DOTAm-py] = 10$ mM, $[HEPES] = 3.8$ mM, $pH = 7.2$, $B_1 = 24$ μT , $T = 21$ $^{\circ}C$.

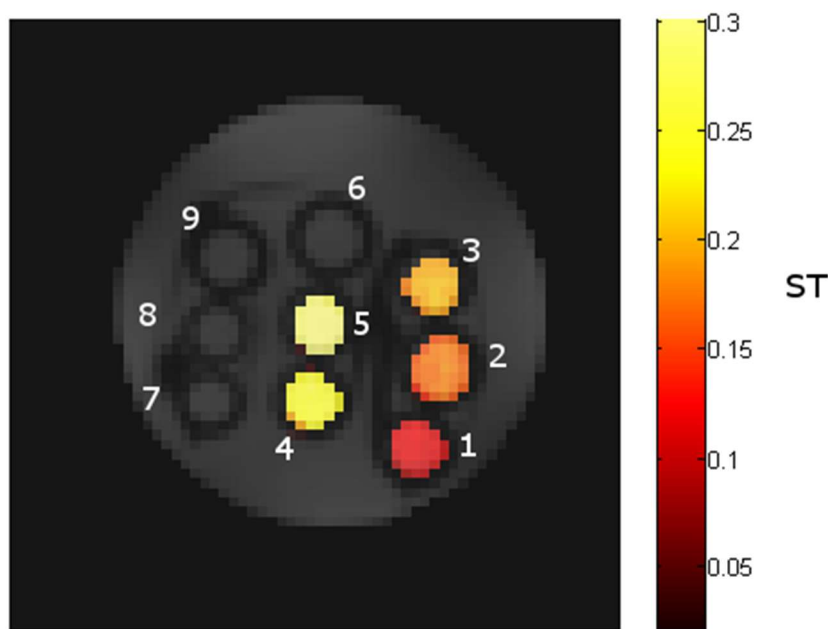


Fig. S6 CEST-MR phantom image of glass capillaries filled with aqueous solutions of 1) 10 mM Tm-DOTAm-py + 10 mM $[(\text{CH}_3\text{CN})_4\text{Cu}]\text{PF}_6$, 2) 10 mM Tm-DOTAm-py + 7.5 mM $[(\text{CH}_3\text{CN})_4\text{Cu}]\text{PF}_6$, 3) 10 mM Tm-DOTAm-py + 5 mM $[(\text{CH}_3\text{CN})_4\text{Cu}]\text{PF}_6$, 4) 10 mM Tm-DOTAm-py + 2.5 mM $[(\text{CH}_3\text{CN})_4\text{Cu}]\text{PF}_6$, 5) 10 mM Tm-DOTAm-py, 6) 10mM $[(\text{CH}_3\text{CN})_4\text{Cu}]\text{PF}_6$, 7) 7.5 mM $[(\text{CH}_3\text{CN})_4\text{Cu}]\text{PF}_6$, 8) 5 mM $[(\text{CH}_3\text{CN})_4\text{Cu}]\text{PF}_6$, 9) 5 mM CuCl_2 , and 10) 2.5 mM $[(\text{CH}_3\text{CN})_4\text{Cu}]\text{PF}_6$. Experimental conditions: $[\text{HEPES}] = 3.8 \text{ mM}$, $\text{pH} = 7.2$, $T = 21 \text{ }^\circ\text{C}$.

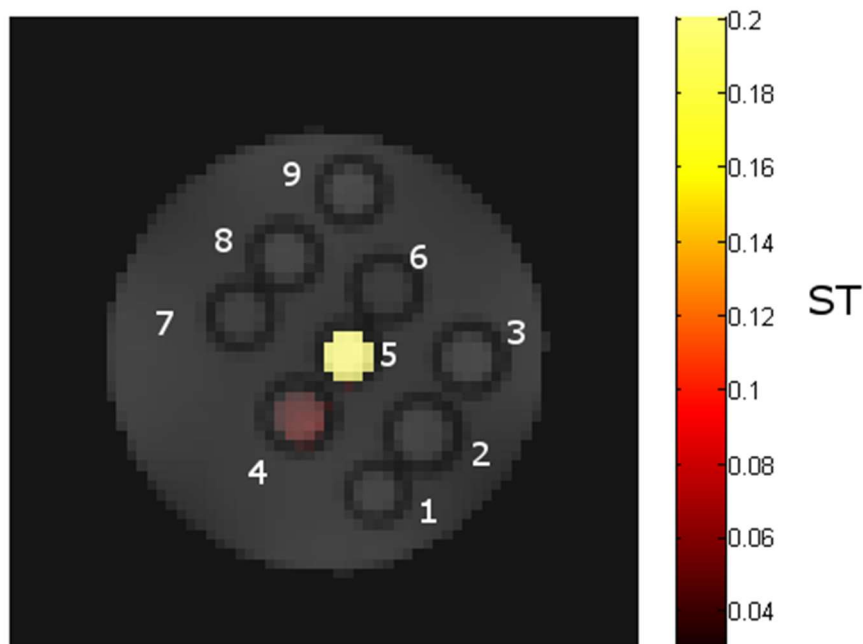


Fig. S7 CEST-MR phantom image of glass capillaries filled with aqueous solutions of 1) 10 mM Tm-DOTAm-py + 10 mM CuCl_2 , 2) 10 mM Tm-DOTAm-py + 7.5 mM CuCl_2 , 3) 10 mM Tm-DOTAm-py + 5 mM CuCl_2 , 4) 10 mM Tm-DOTAm-py + 2.5 mM CuCl_2 , 5) 10 mM Tm-DOTAm-py, 6) 10 mM CuCl_2 , 7) 7.5 mM CuCl_2 , 8) 5 mM CuCl_2 , 9) 2.5 mM CuCl_2 , Experimental conditions: $[\text{HEPES}] 3.8 \text{ mM}$, $\text{pH} = 7.2$, $T = 21 \text{ }^\circ\text{C}$.

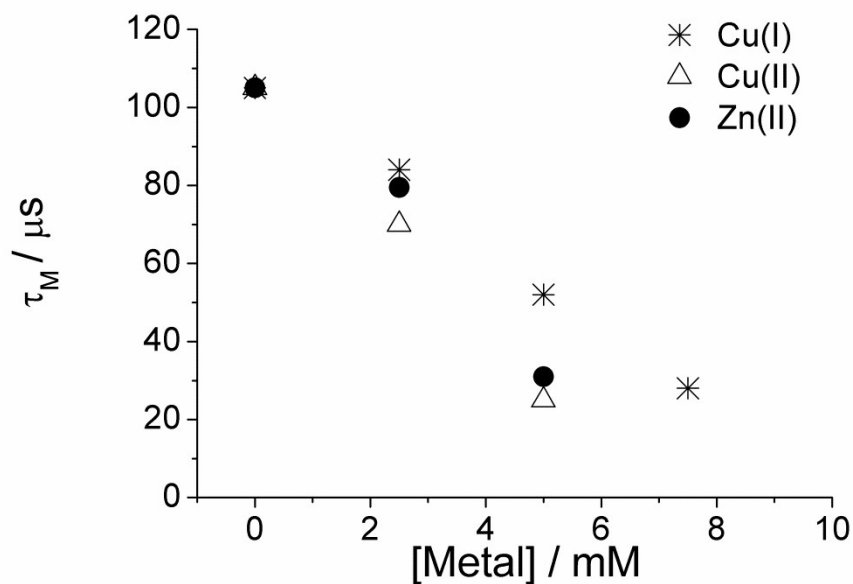


Fig. S8 τ_M of Tm-DOTAm-py upon addition of different concentration of CuCl_2 , $[(\text{CH}_3\text{CN})_4\text{Cu}]\text{PF}_6$ and ZnCl_2 . Experimental conditions: $[\text{Tm-DOTAm-py}] = 10 \text{ mM}$, $[\text{HEPES}] = 3.8 \text{ mM}$, $\text{pH} = 7.2$, $T = 21^\circ\text{C}$.

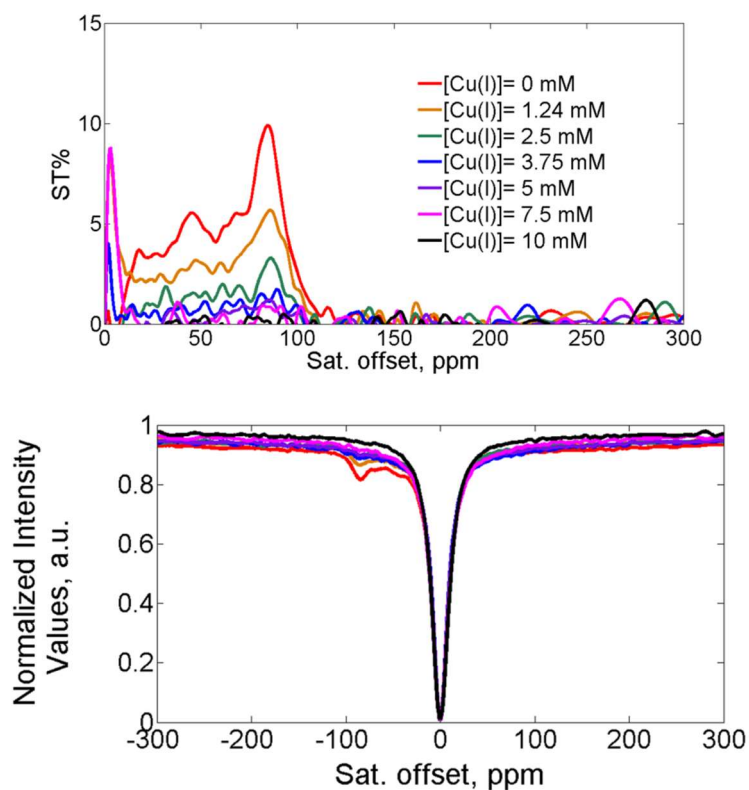


Fig. S9 a) Saturation transfer % (ST%) spectra and b) Z-spectra of Tm-DOTAm- β Ala-py upon addition of different concentration of $[(\text{CH}_3\text{CN})_4\text{Cu}]\text{PF}_6$. Experimental conditions: $[\text{Tm-DOTAm-}\beta\text{Ala-py}] = 5 \text{ mM}$, $[\text{HEPES}] = 3.8 \text{ mM}$, $\text{pH} = 7.2$, $B_1 = 48 \mu\text{T}$, $T = 21^\circ\text{C}$.

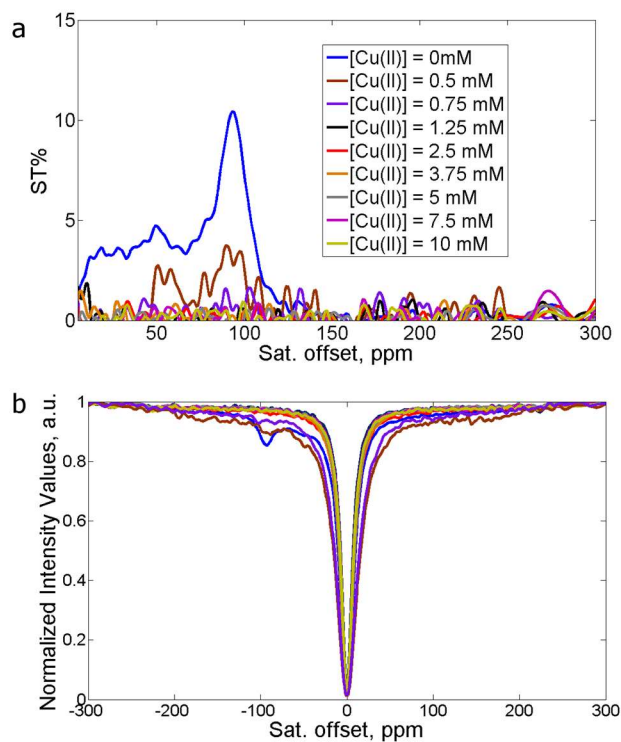


Fig. S10 a) Saturation transfer % (ST%) spectra and b) Z-spectra of Tm-DOTAm- β Ala-py upon addition of different concentration of CuCl_2 . Experimental conditions: [Tm-DOTAm- β Ala-py] = 5 mM, [HEPES] = 3.8 mM, pH = 7.2, B_1 = 48 μT , T = 21 $^\circ\text{C}$.

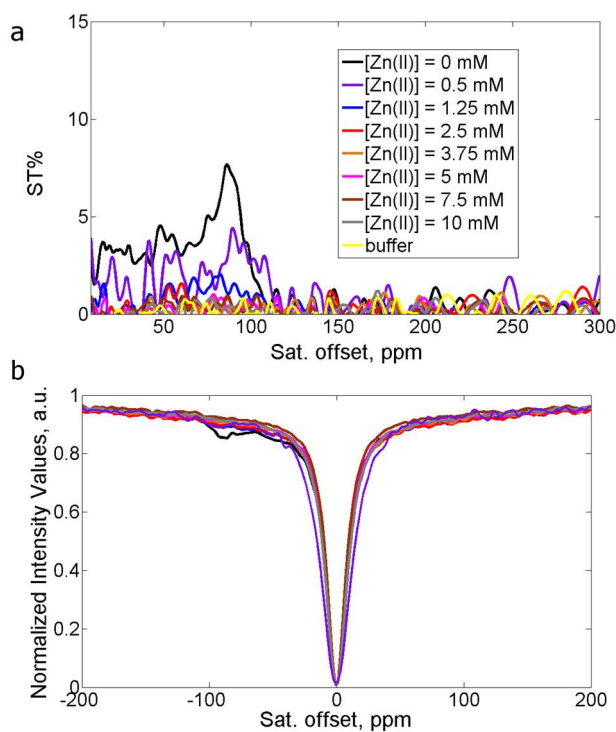


Fig. S11 a) Saturation transfer % (ST%) spectra and b) Z-spectra of Tm-DOTAm- β Ala-py upon addition of different concentration of ZnCl_2 . Experimental conditions: [Tm-DOTAm- β Ala-py] = 5 mM, [HEPES] = 3.8 mM, pH = 7.2, B_1 = 48 μT , T = 21 $^\circ\text{C}$.

DOTAm-py

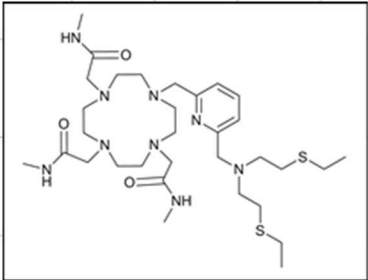


Fig. S12 ^1H NMR spectrum of DOTAm-py (CDCl_3 , 500 MHz).

DOTAm-py

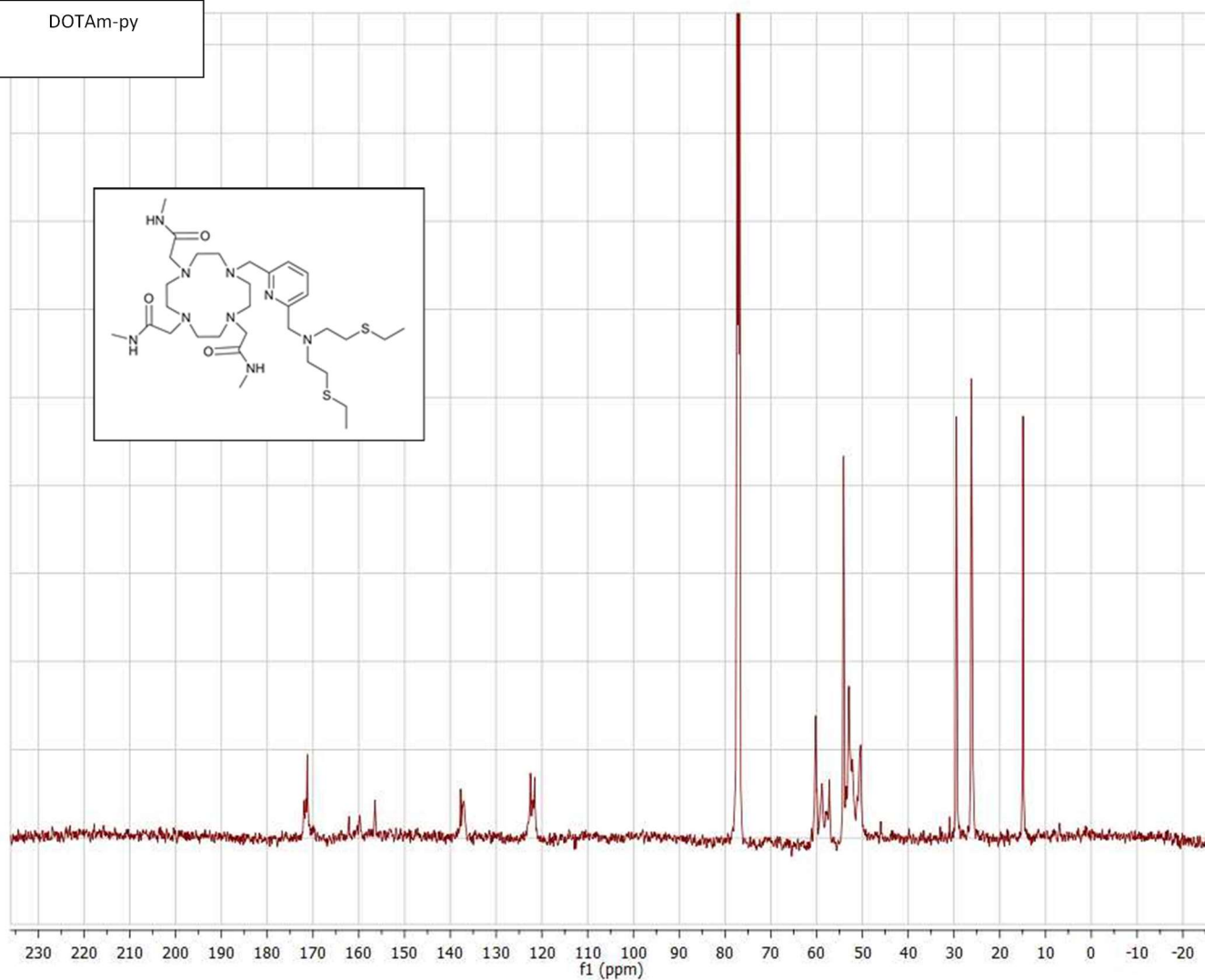
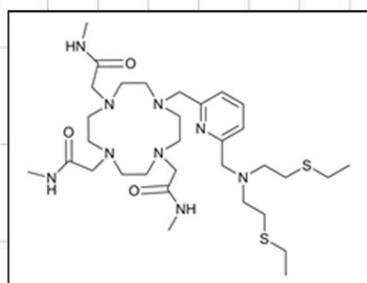


Fig. S13 ^{13}C NMR spectrum of DOTAm-py (CDCl_3 , 126 MHz).

Tm-DOTAm-py

Chemical structure of Tm-DOTAm-py is shown in the inset. The structure features a central Tm atom coordinated by four N atoms and two O atoms, forming a macrocyclic complex. The structure includes a pyridine ring and a thioether group.

Tm-DOTAm-py

Chemical structure of Tm-DOTAm-py is shown in the inset. The structure features a central Tm atom coordinated by four N atoms and two O atoms, forming a macrocyclic complex. The structure includes a pyridine ring and a thioether group.

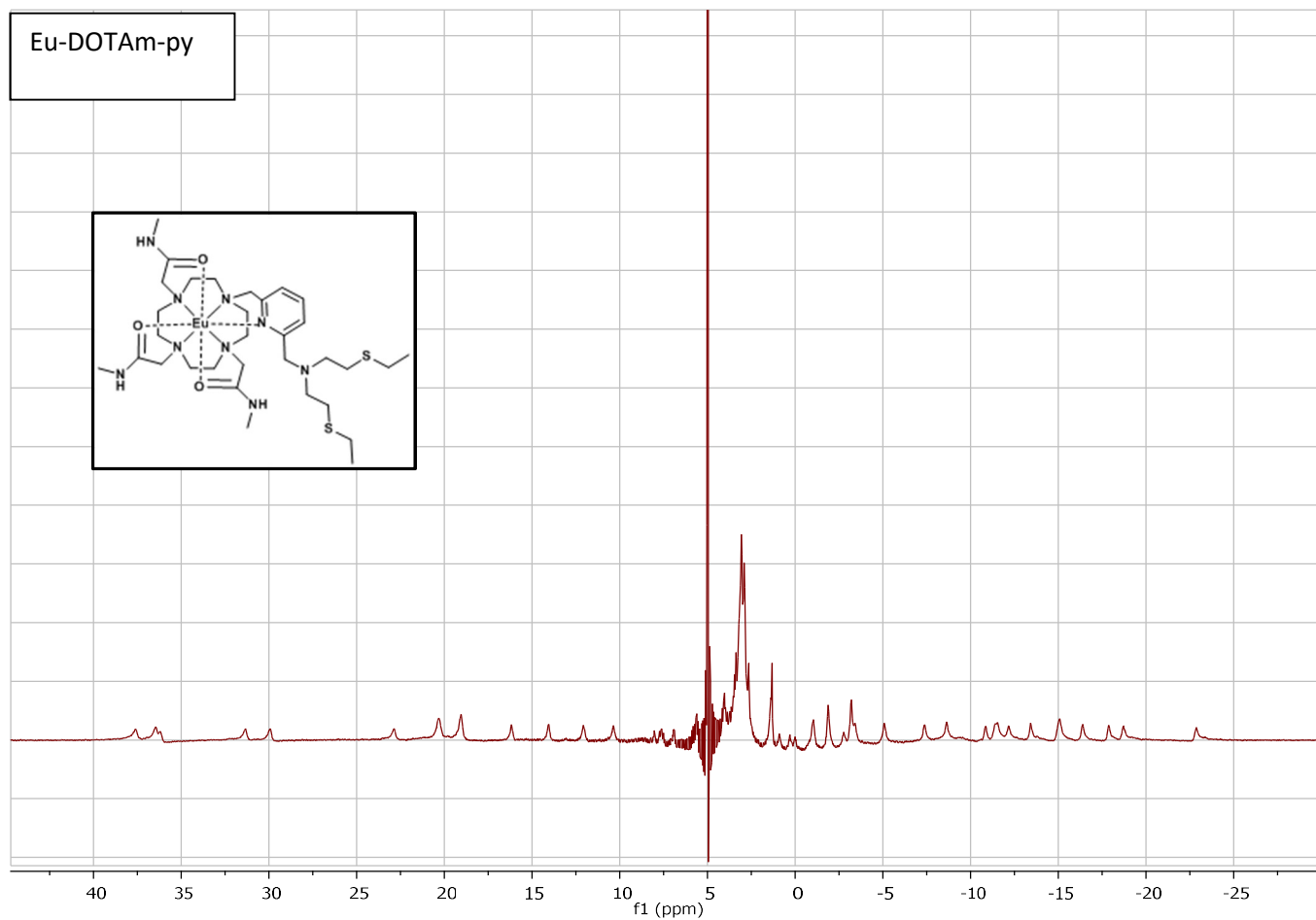


Fig. S15 ^1H NMR spectrum of Eu-DOTAm-py (D_2O , 500 MHz).

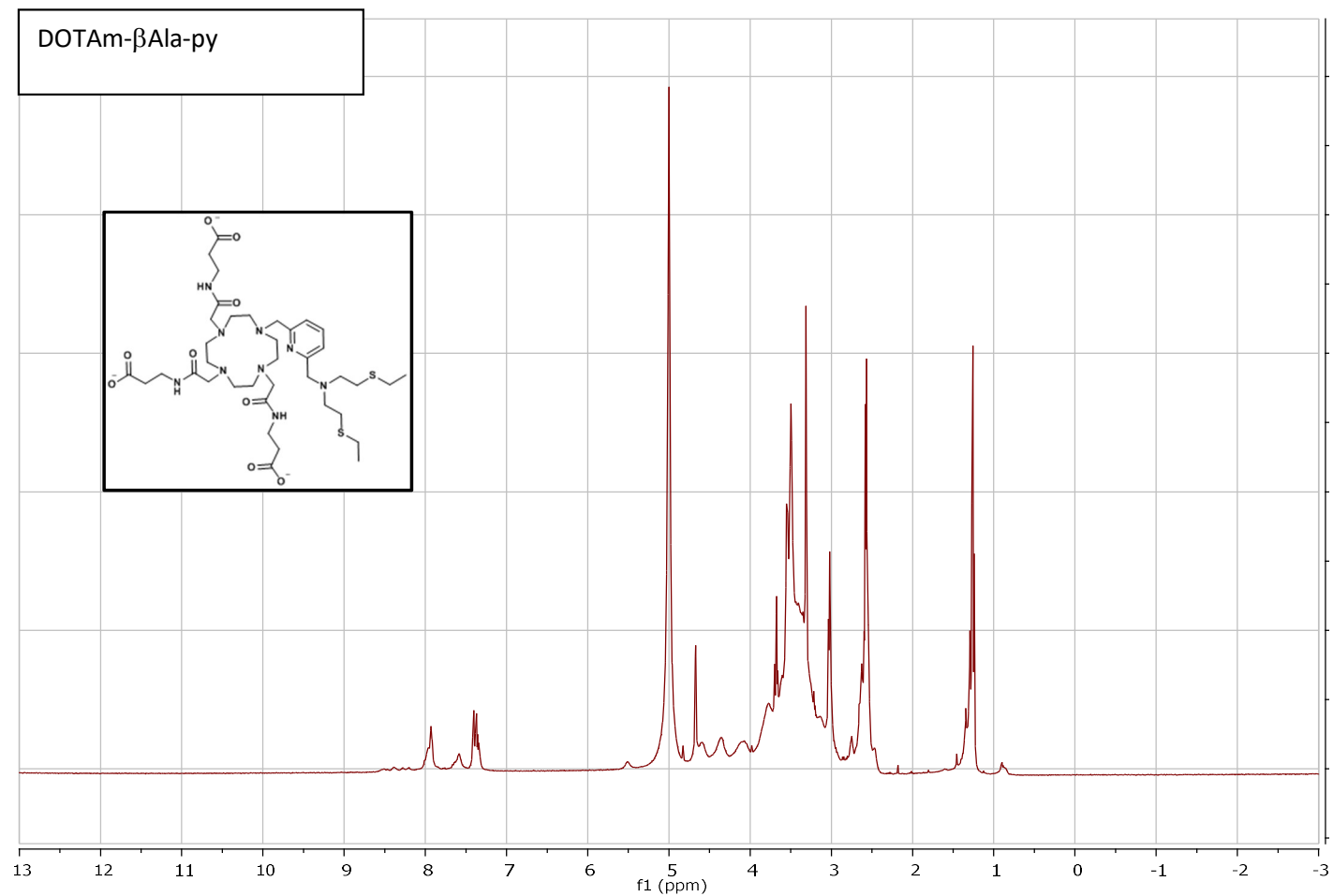


Fig. S16 ¹H NMR spectrum of DOTAm-βAla-py (CD₃OD, 500 MHz).

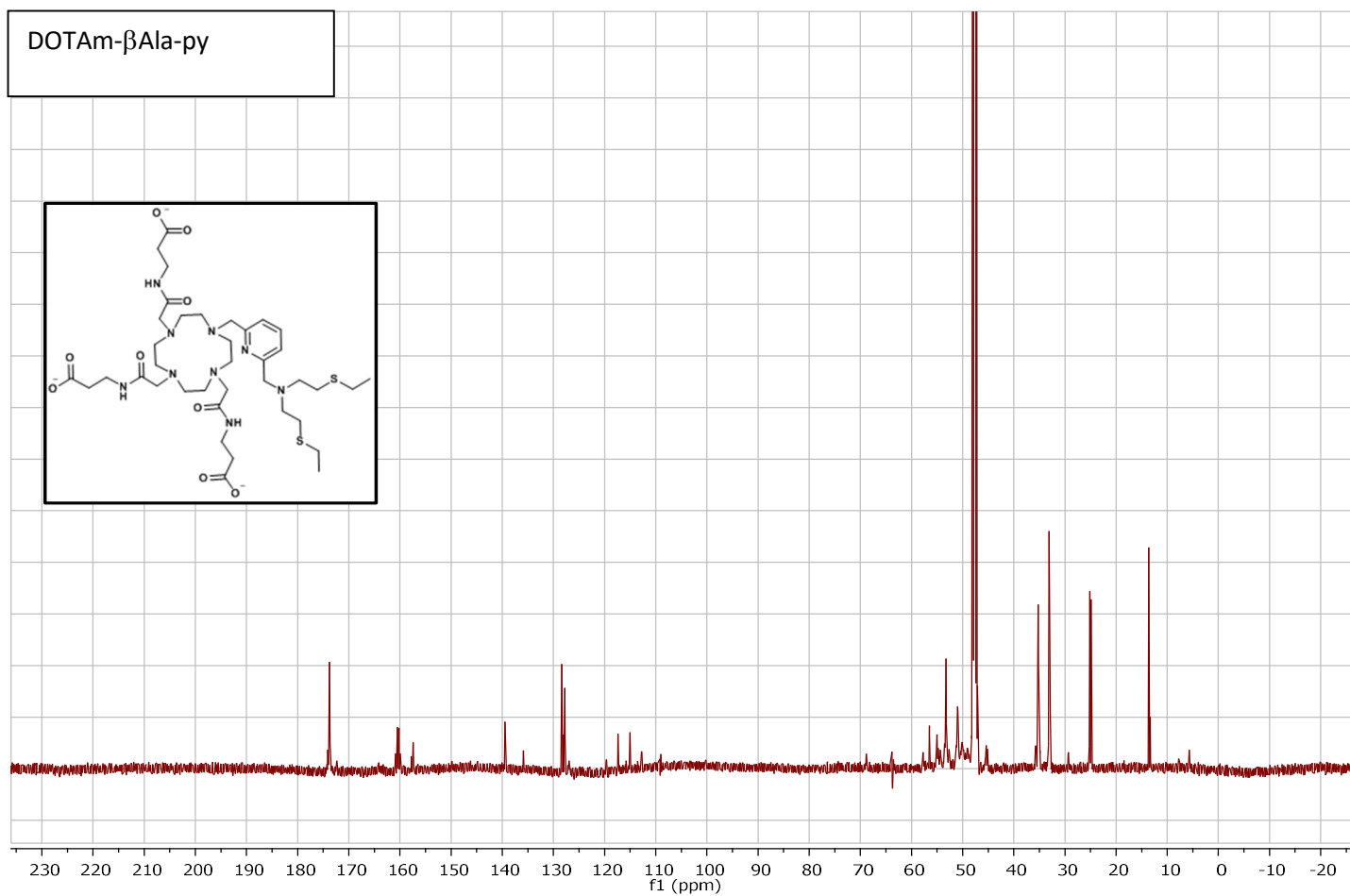


Fig. S17 ^{13}C NMR spectrum of DOTAm- β Ala-py (CD_3OD , 126 MHz).

