

***Supplementary Information***

Anion-induced 3d-4f luminescent coordination clusters: structural  
characters and chemical fixation of CO<sub>2</sub> under mild conditions

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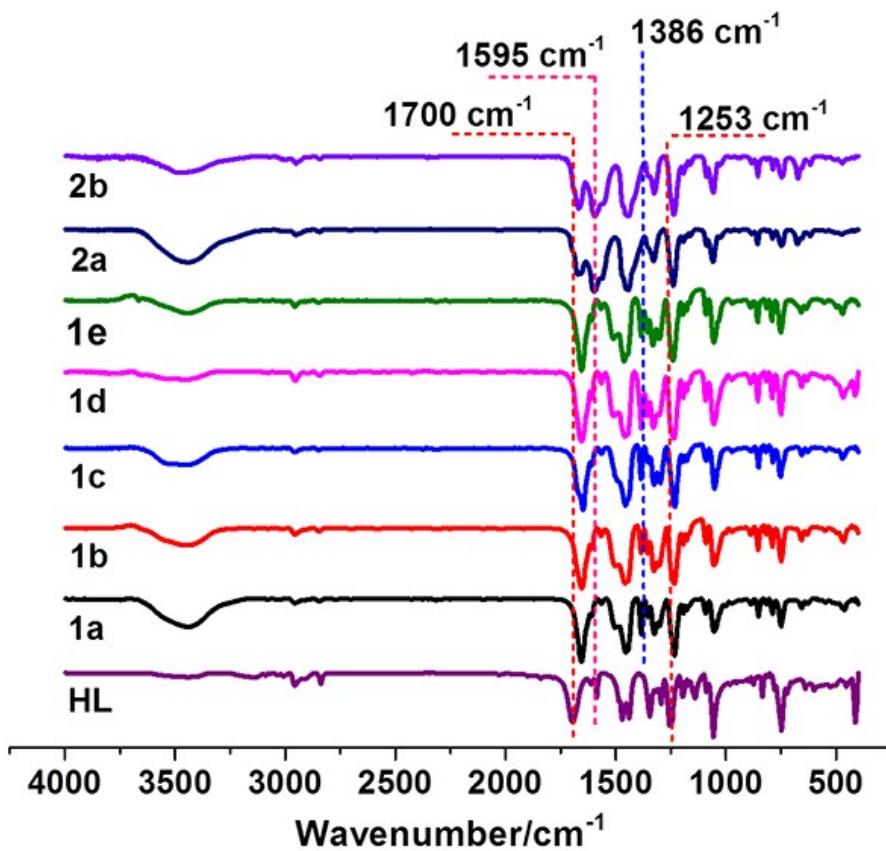
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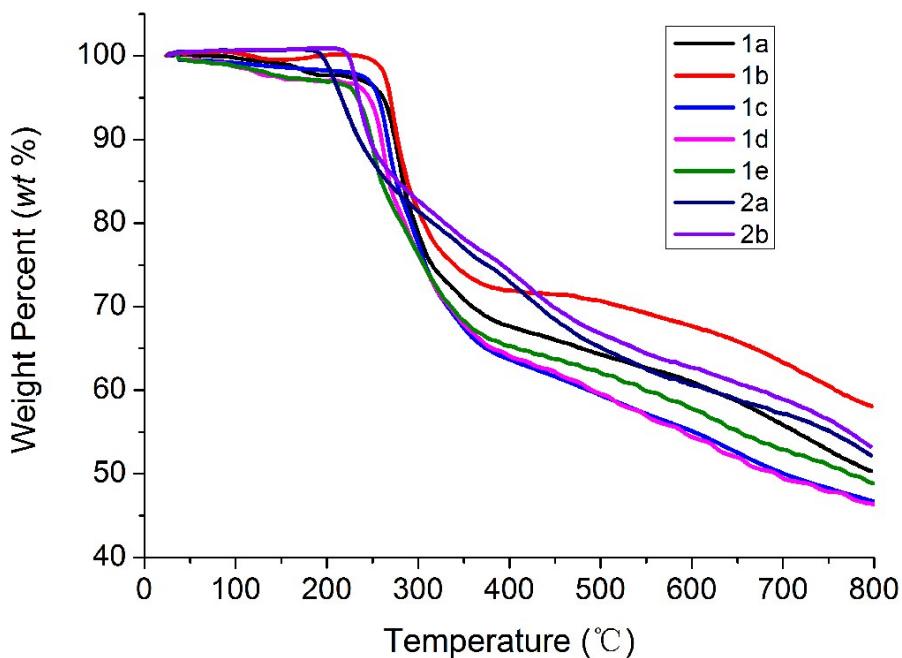
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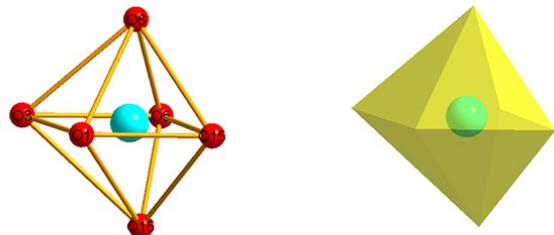
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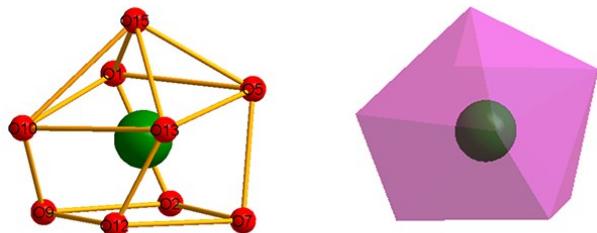
**Figure S1** IR spectra of the ligand and complexes in the solid state (KBr disk) at room temperature.



**Figure S2** TGA spectra of the complexes.

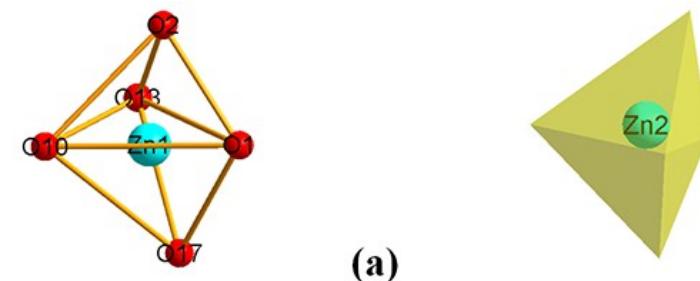


(a)

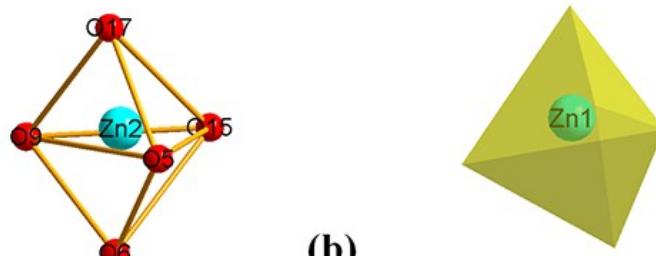


(b)

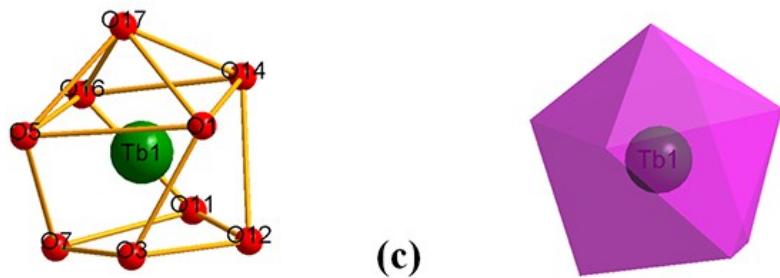
**Figure S3.** Coordination polyhedra of zinc ion (a) and terbium ion (b) of complex **1b**. Solvent molecule, extra ligands and coordination anions are omitted for clarity.



(a)

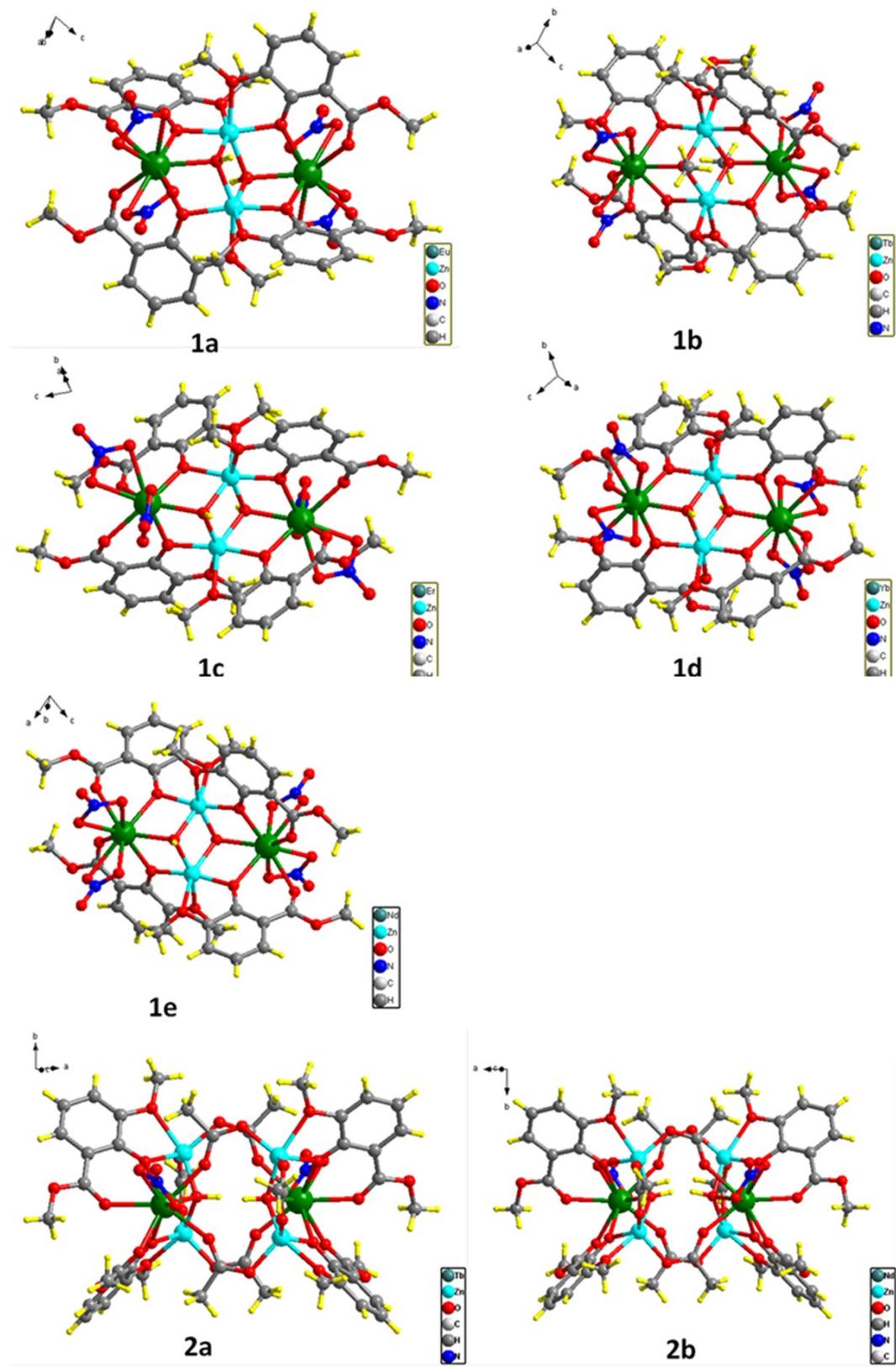


(b)



(c)

**Figure S4.** Coordination polyhedra of zinc ion (a)(b) and terbium ion (c) of complex **2a**. Solvent molecule, extra ligands and coordination anions are omitted for clarity.



**Figure S5.** Crystal structures of complexes **1a-1e** and **2a-2b**. Solvent molecules are omitted for clarity.

**Table S1** Crystal Data and Structure Refinement Parameters for Complexes **1a-2b**.

	<b>1a</b> (CCDC-1575742)	<b>1b</b> (CCDC-1575745)	<b>1c</b> (CCDC-1575741)	<b>1d</b> (CCDC-1575743)
Formula	C <sub>36</sub> H <sub>38</sub> Eu <sub>2</sub> N <sub>4</sub> O <sub>30</sub> Zn <sub>2</sub> ·2(C H <sub>3</sub> CN)	C <sub>38</sub> H <sub>42</sub> N <sub>4</sub> O <sub>30</sub> Tb <sub>2</sub> Zn <sub>2</sub> ·2(C H <sub>3</sub> CN)	C <sub>36</sub> H <sub>38</sub> N <sub>4</sub> O <sub>30</sub> Er <sub>2</sub> Zn <sub>2</sub> ·4(C H <sub>3</sub> CN)	C <sub>36</sub> H <sub>38</sub> N <sub>4</sub> O <sub>30</sub> Yb <sub>2</sub> Zn <sub>2</sub> ·2(C H <sub>3</sub> CN)
Formula Weight	1523.47	1565.44	1636.18	1565.63
Crystal System	Triclinic	Triclinic	Triclinic	Triclinic
Space Group	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1
<i>a</i> (Å)	11.1206 (8)	10.8386 (9)	11.0490 (10)	10.5088 (12)
<i>b</i> (Å)	11.2718 (8)	11.0943 (5)	11.0806 (10)	10.9921 (6)
<i>c</i> (Å)	13.3306 (11)	13.3764 (9)	14.7671 (12)	13.2571 (8)
$\alpha$ (°)	74.493 (7)	111.335 (5)	83.834 (7)	111.997 (5)
$\beta$ (°)	72.689 (7)	101.544 (7)	68.566 (8)	100.721 (7)
$\gamma$ (°)	62.178 (7)	102.840 (6)	64.269 (9)	101.715 (7)
<i>Z</i>	1	1	1	1
Volume (Å <sup>3</sup> )	1394.5 (2)	1389.11 (17)	1513.2 (3)	1331.15 (19)
<i>D</i> <sub>calcd</sub> (g/cm <sup>3</sup> )	1.814	1.871	1.796	1.953
$\mu$ (mm <sup>-1</sup> )	3.163	3.464	3.621	4.470
<i>F</i> (000)	752.0	772.0	806.0	766.0
Temperature	293.00(10)	290.83 (10)	293 (2)	293 (2)
Crystal Size (mm)	0.25 × 0.24 × 0.22	0.35 × 0.3 × 0.2	0.23 × 0.22 × 0.21	0.13 × 0.12 × 0.1
Reflections collected	9529	9549	10897	9027
Independent ref.	5469	5447	5961	5228
Data/restraints/ parameters	5469/30/389	5447/30/376	5961/42/417	5228/19/370
<i>R</i> <sub>int</sub>	0.0302	0.0353	0.0392	0.0474
Final <i>R</i> indices [ <i>I</i> >2.0σ]	<i>R</i> <sub><i>I</i></sub> = 0.0374	<i>R</i> <sub><i>I</i></sub> = 0.0373	<i>R</i> <sub><i>I</i></sub> = 0.0374	<i>R</i> <sub><i>I</i></sub> = 0.0408
( <i>I</i> )	<i>wR</i> <sub><i>2</i></sub> = 0.0914	<i>wR</i> <sub><i>2</i></sub> = 0.0732	<i>wR</i> <sub><i>2</i></sub> = 0.0708	<i>wR</i> <sub><i>2</i></sub> = 0.0674
<i>R</i> indices	<i>R</i> <sub><i>I</i></sub> = 0.0464	<i>R</i> <sub><i>I</i></sub> = 0.0495	<i>R</i> <sub><i>I</i></sub> = 0.0478	<i>R</i> <sub><i>I</i></sub> = 0.0565
(all data)	<i>wR</i> <sub><i>2</i></sub> = 0.0998	<i>wR</i> <sub><i>2</i></sub> = 0.0817	<i>wR</i> <sub><i>2</i></sub> = 0.0782	<i>wR</i> <sub><i>2</i></sub> = 0.0764
GOF	1.063	1.042	1.051	0.987

Table S1 (*Contd.*)

	<b>1e</b> (CCDC-1577169)	<b>2a</b> (CCDC-1575744)	<b>2b</b> (CCDC-1575746)
Formula	C <sub>36</sub> H <sub>38</sub> N <sub>4</sub> Nd <sub>2</sub> O <sub>30</sub> Zn <sub>2</sub> ·2(CH <sub>3</sub> CN)	C <sub>48</sub> H <sub>56</sub> N <sub>2</sub> O <sub>36</sub> Tb <sub>2</sub> Zn <sub>4</sub> ·5(CH <sub>3</sub> CN)	C <sub>48</sub> H <sub>56</sub> Nd <sub>2</sub> O <sub>36</sub> Zn <sub>4</sub> ·4(CH <sub>3</sub> CN)
Formula	1508.03	2018.51	1951.12
Weight			
Crystal System	Triclinic	Monoclinic	Monoclinic
Space Group	<i>P</i> -1	<i>C</i> 2/ <i>c</i>	<i>C</i> 2/ <i>c</i>
<i>a</i> (Å)	11.1988(12)	22.5467 (5)	22.2896 (5)
<i>b</i> (Å)	11.2992(14)	17.8961 (5)	18.4936 (5)
<i>c</i> (Å)	13.3744(14)	19.5235 (5)	19.2391 (4)
$\alpha$ (°)	74.797(10)	90	90
$\beta$ (°)	72.700(9)	100.736 (2)	99.957 (2)
$\gamma$ (°)	62.097(12)	90	90
<i>Z</i>	1	4	4

Volume (Å <sup>3</sup> )	1412.7(3)	7739.8 (3)	7811.2 (3)
D <sub>calcd</sub> (g/cm <sup>3</sup> )	1.773	1.732	1.659
μ (mm <sup>-1</sup> )	2.739	3.114	2.602
F(000)	746.0	4012.0	3896.0
Temperature	173.00(10)	293.23 (10)	173.00 (10)
Crystal Size (mm)	0.35 × 0.3 × 0.2	0.35 × 0.3 × 0.2	0.35 × 0.3 × 0.2
Reflections collected	9130	15624	15970
Independent ref.	5536	7597	7676
Data/restraints/ parameters	5536/32/389	7597/48/515	7676/62/501
R <sub>int</sub>	0.0559	0.0342	0.0445
Final R indices	R <sub>I</sub> = 0.0586	R <sub>I</sub> = 0.0355	R <sub>I</sub> = 0.0452
[I>2.0σ (I)]	wR <sub>2</sub> = 0.1247	wR <sub>2</sub> = 0.0735	wR <sub>2</sub> = 0.1212
R indices (all date)	R <sub>I</sub> = 0.0895	R <sub>I</sub> = 0.0504	R <sub>I</sub> = 0.0591
	wR <sub>2</sub> = 0.1489	wR <sub>2</sub> = 0.0828	wR <sub>2</sub> = 0.1348
GOF	1.036	1.059	1.106

**Table S2** Selected Bond Lengths (Å) and Angles (°) for All Complexes.

Complex <b>1a</b>					
Eu(1)—Zn(1)	3.4974(7)	Eu(1)—O(6)	2.413(4)	Zn(1)—O(8)	2.221(4)
Eu(1)—(13)	2.542(4)	Eu(1)—O(2)	2.395(4)	Zn(1)—O(1)l	2.034(3)
Eu(1)—O(1)	2.384(3)	Eu(1)—O(10)	2.524(4)	Zn(1)—O(15)l	2.050(3)
Eu(1)—O(15)	2.361(3)	Eu(1)—N(2)	2.931(5)	Zn(1)—O(15)	2.059(4)
Eu(1)—O(5)	2.393(3)	Eu(1)—N(1)	2.910(5)	Zn(1)—O(5)	2.045(3)
Eu(1)—O(12)	2.485(4)	Zn(1)—Zn(1)l	3.0115(11)	Zn(1)—O(4)l	2.185(4)
Eu(1)—O(9)	2.478(4)				
O(13)—Eu(1)—Zn(1)	132.70(8)	O(12)—Eu(1)—O(13)	50.88(12)	Zn(1) <sup>l</sup> —Zn(1)—Eu(1)	64.65(2)
O(13)—Eu(1)—N(2)	25.58(12)	O(12)—Eu(1)—O(10)	112.14(14)	O(8)—Zn(1)—Eu(1) <sup>l</sup>	99.88(9)
O(13)—Eu(1)—N(1)	63.50(14)	O(12)—Eu(1)—N(2)	25.30(12)	O(8)—Zn(1)—Eu(1)	116.83(9)
O(1)—Eu(1)—Zn(1)	75.33(8)	O(12)—Eu(1)—N(1)	93.67(15)	O(8)—Zn(1)—Zn(1) <sup>l</sup>	136.37(10)
O(1)—Eu(1)—O(13)	145.83(12)	O(9)—Eu(1)—Zn(1)	109.26(9)	O(1) <sup>l</sup> —Zn(1)—Eu(1) <sup>l</sup>	41.15(10)
O(1)—Eu(1)—O(5)	89.38(12)	O(9)—Eu(1)—O(13)	68.40(13)	O(1) <sup>l</sup> —Zn(1)—Eu(1)	142.08(11)
O(1)—Eu(1)—O(12)	145.28(12)	O(9)—Eu(1)—O(12)	78.06(14)	O(1) <sup>l</sup> —Zn(1)—Zn(1) <sup>l</sup>	92.28(10)
O(1)—Eu(1)—O(9)	85.02(13)	O(9)—Eu(1)—O(10)	51.22(13)	O(1) <sup>l</sup> —Zn(1)—O(8)	100.72(14)
O(1)—Eu(1)—O(6)	132.29(13)	O(9)—Eu(1)—N(2)	71.99(14)	O(1) <sup>l</sup> —Zn(1)—O(15)	101.72(14)
O(1)—Eu(1)—O(2)	70.17(12)	O(9)—Eu(1)—N(1)	25.83(13)	O(1) <sup>l</sup> —Zn(1)—O(15) <sup>l</sup>	81.64(13)
O(1)—Eu(1)—O(10)	78.22(13)	O(6)—Eu(1)—Zn(1)	100.79(9)	O(1) <sup>l</sup> —Zn(1)—O(5)	169.22(14)
O(1)—Eu(1)—N(2)	156.48(13)	O(6)—Eu(1)—O(13)	69.21(13)	O(1) <sup>l</sup> —Zn(1)—O(4) <sup>l</sup>	76.07(13)
O(1)—Eu(1)—N(1)	83.15(14)	O(6)—Eu(1)—O(12)	77.26(14)	O(15) <sup>l</sup> —Zn(1)—Eu(1)	97.52(9)

O(15)—Eu(1)—Zn(1)	34.76(9)	O(6)—Eu(1)—O(9)	137.56(14)	O(15)—Zn(1)—Eu(1) <sup>1</sup>	97.12(9)
O(15)—Eu(1)—O(13)	120.89(12)	O(6)—Eu(1)—O(10)	109.58(13)	O(15) <sup>1</sup> —Zn(1)—Eu(1) <sup>1</sup>	40.60(9)
O(15)—Eu(1)—O(1)	68.45(11)	O(6)—Eu(1)—N(2)	70.92(14)	O(15)—Zn(1)—Eu(1)	40.84(9)
O(15)—Eu(1)—O(5)	69.32(12)	O(6)—Eu(1)—N(1)	124.43(13)	O(15)—Zn(1)—Zn(1) <sup>1</sup>	42.75(9)
O(15)—Eu(1)—O(12)	77.86(12)	O(2)—Eu(1)—Zn(1)	115.92(10)	O(15) <sup>1</sup> —Zn(1)—Zn(1) <sup>1</sup>	42.98(10)
O(15)—Eu(1)—O(9)	74.68(12)	O(2)—Eu(1)—O(13)	103.75(13)	O(15) <sup>1</sup> —Zn(1)—O(8)	97.76(14)
O(15)—Eu(1)—O(6)	131.60(12)	O(2)—Eu(1)—O(12)	144.41(13)	O(15)—Zn(1)—O(8)	157.56(13)
O(15)—Eu(1)—O(2)	134.65(12)	O(2)—Eu(1)—O(9)	119.20(14)	O(15) <sup>1</sup> —Zn(1)—O(15)	85.72(14)
O(15)—Eu(1)—O(10)	118.11(13)	O(2)—Eu(1)—O(6)	69.48(14)	O(15) <sup>1</sup> —Zn(1)—O(4) <sup>1</sup>	157.70(13)
O(15)—Eu(1)—N(2)	99.82(12)	O(2)—Eu(1)—O(10)	69.39(14)	O(15)—Zn(1)—O(4) <sup>1</sup>	98.55(14)
O(15)—Eu(1)—N(1)	97.95(13)	O(2)—Eu(1)—N(2)	125.40(13)	O(5)—Zn(1)—Eu(1)	41.65(10)
O(5)—Eu(1)—Zn(1)	34.61(8)	O(2)—Eu(1)—N(1)	94.77(15)	O(5)—Zn(1)—Eu(1) <sup>1</sup>	148.87(10)
O(5)—Eu(1)—O(13)	124.78(12)	O(10)—Eu(1)—Zn(1)	148.54(10)	O(5)—Zn(1)—Zn(1) <sup>1</sup>	97.37(10)
O(5)—Eu(1)—O(12)	86.41(13)	O(10)—Eu(1)—O(13)	68.51(13)	O(5)—Zn(1)—O(8)	75.46(13)
O(5)—Eu(1)—O(9)	143.04(12)	O(10)—Eu(1)—N(2)	90.72(14)	O(5)—Zn(1)—O(15)	82.42(13)
O(5)—Eu(1)—O(6)	68.29(12)	O(10)—Eu(1)—N(1)	25.61(13)	O(5)—Zn(1)—O(15) <sup>1</sup>	108.73(13)
O(5)—Eu(1)—O(2)	92.67(13)	N(2)—Eu(1)—Zn(1)	107.15(10)	O(5)—Zn(1)—O(4) <sup>1</sup>	93.54(14)
O(5)—Eu(1)—O(10)	160.76(13)	N(1)—Eu(1)—Zn(1)	132.52(10)	O(4) <sup>1</sup> —Zn(1)—Eu(1)	99.94(10)
O(5)—Eu(1)—N(2)	105.88(13)	N(1)—Eu(1)—N(2)	78.29(15)	O(4) <sup>1</sup> —Zn(1)—Eu(1) <sup>1</sup>	117.14(9)
O(5)—Eu(1)—N(1)	166.98(12)	Eu(1)—Zn(1)—Eu(1) <sup>1</sup>	129.054(17)	O(4) <sup>1</sup> —Zn(1)—Zn(1) <sup>1</sup>	137.00(11)
O(12)—Eu(1)—Zn(1)	81.93(9)	Zn(1) <sup>1</sup> —Zn(1)—Eu(1) <sup>1</sup>	64.40(2)	O(4) <sup>1</sup> —Zn(1)—O(8)	86.63(15)
<b>Complex 1b</b>					
Tb(1)—Zn(1)	3.5033(7)	Tb(1)—O(11)	2.479(4)	Zn(1)—O(7)	2.193(3)
Tb(1)—O(3)	2.518(4)	Tb(1)—N(1)	2.856(5)	Zn(1)—O(5)	2.070(4)
Tb(1)—O(6) <sup>1</sup>	2.372(4)	Tb(1)—O(14)	2.533(4)	Zn(1)—O(1)	2.055(3)
Tb(1)—O(5) <sup>1</sup>	2.365(3)	Tb(1)—O(13)	2.472(4)	Zn(1)—O(2)	2.076(4)
Tb(1)—O(1)	2.338(3)	Tb(1)—N(2)	2.929(6)	Zn(1)—O(9)	2.054(3)
Tb(1)—O(9) <sup>1</sup>	2.356(4)	Zn(1)—Zn(1) <sup>1</sup>	3.0842(10)	Zn(1)—O(9) <sup>1</sup>	2.106(3)
Tb(1)—O(10)	2.446(4)				
O(3)—Tb(1)—Zn(1)	96.82(8)	O(9) <sup>1</sup> —Tb(1)—O(6) <sup>1</sup>	134.99(12)	Zn(1) <sup>1</sup> —Zn(1)—Tb(1) <sup>1</sup>	63.80(2)
O(3)—Tb(1)—N(1)	123.17(15)	O(9) <sup>1</sup> —Tb(1)—O(5) <sup>1</sup>	68.83(12)	O(7)—Zn(1)—Tb(1) <sup>1</sup>	115.43(10)
O(3)—Tb(1)—O(14)	68.78(14)	O(9) <sup>1</sup> —Tb(1)—O(10)	77.82(13)	O(7)—Zn(1)—Tb(1)	100.76(10)
O(3)—Tb(1)—N(2)	70.52(16)	O(9) <sup>1</sup> —Tb(1)—O(11)	124.36(14)	O(7)—Zn(1)—Zn(1) <sup>1</sup>	134.48(11)
O(6) <sup>1</sup> —Tb(1)—Zn(1)	113.50(9)	O(9) <sup>1</sup> —Tb(1)—N(1)	102.13(16)	O(5)—Zn(1)—Tb(1) <sup>1</sup>	40.74(9)
O(6) <sup>1</sup> —Tb(1)—O(3)	68.63(13)	O(9) <sup>1</sup> —Tb(1)—O(14)	117.58(15)	O(5)—Zn(1)—Tb(1)	137.52(9)
O(6) <sup>1</sup> —Tb(1)—O(10)	119.18(14)	O(9) <sup>1</sup> —Tb(1)—O(13)	73.83(13)	O(5)—Zn(1)—Zn(1) <sup>1</sup>	88.60(8)
O(6) <sup>1</sup> —Tb(1)—O(11)	69.92(15)	O(9) <sup>1</sup> —Tb(1)—N(2)	95.78(16)	O(5)—Zn(1)—O(7)	74.95(13)
O(6) <sup>1</sup> —Tb(1)—N(1)	95.04(17)	O(10)—Tb(1)—Zn(1)	113.51(11)	O(5)—Zn(1)—O(2)	98.74(15)
O(6) <sup>1</sup> —Tb(1)—O(14)	107.41(15)	O(10)—Tb(1)—O(3)	137.91(13)	O(5)—Zn(1)—O(9) <sup>1</sup>	97.06(14)
O(6) <sup>1</sup> —Tb(1)—O(13)	144.30(14)	O(10)—Tb(1)—O(11)	51.75(16)	O(1)—Zn(1)—Tb(1)	40.01(8)
O(6) <sup>1</sup> —Tb(1)—N(2)	127.99(17)	O(10)—Tb(1)—N(1)	26.01(16)	O(1)—Zn(1)—Tb(1) <sup>1</sup>	149.01(9)
O(5) <sup>1</sup> —Tb(1)—Zn(1)	74.66(8)	O(10)—Tb(1)—O(14)	69.65(14)	O(1)—Zn(1)—Zn(1) <sup>1</sup>	96.05(9)
O(5) <sup>1</sup> —Tb(1)—O(3)	130.63(13)	O(10)—Tb(1)—O(13)	82.65(14)	O(1)—Zn(1)—O(7)	95.53(14)
O(5) <sup>1</sup> —Tb(1)—O(6) <sup>1</sup>	70.91(13)	O(10)—Tb(1)—N(2)	75.00(15)	O(1)—Zn(1)—O(5)	169.94(14)

O(5) <sup>l</sup> —Tb(1)—O(10)	86.85(13)	O(11)—Tb(1)—Zn(1)	155.90(12)	O(1)—Zn(1)—O(2)	84.15(15)
O(5) <sup>l</sup> —Tb(1)—O(11)	84.79(13)	O(11)—Tb(1)—O(3)	106.19(14)	O(1)—Zn(1)—O(9) <sup>l</sup>	80.80(13)
O(5) <sup>l</sup> —Tb(1)—N(1)	87.11(13)	O(11)—Tb(1)—N(1)	25.86(16)	O(2)—Zn(1)—Tb(1) <sup>l</sup>	94.38(11)
O(5) <sup>l</sup> —Tb(1)—O(14)	152.36(12)	O(11)—Tb(1)—O(14)	69.41(15)	O(2)—Zn(1)—Tb(1)	123.65(13)
O(5) <sup>l</sup> —Tb(1)—O(13)	142.55(14)	O(11)—Tb(1)—N(2)	92.56(16)	O(2)—Zn(1)—Zn(1) <sup>l</sup>	135.75(12)
O(5) <sup>l</sup> —Tb(1)—N(2)	158.56(16)	N(1)—Tb(1)—Zn(1)	137.74(14)	O(2)—Zn(1)—O(7)	89.18(16)
O(1)—Tb(1)—Zn(1)	34.40(9)	N(1)—Tb(1)—N(2)	81.54(15)	O(2)—Zn(1)—O(9) <sup>l</sup>	163.89(16)
O(1)—Tb(1)—O(3)	64.75(11)	O(14)—Tb(1)—Zn(1)	127.43(11)	O(9) <sup>l</sup> —Zn(1)—Tb(1) <sup>l</sup>	95.56(9)
O(1)—Tb(1)—O(6) <sup>l</sup>	89.33(12)	O(14)—Tb(1)—N(1)	65.37(14)	O(9) <sup>l</sup> —Zn(1)—Tb(1)	40.87(10)
O(1)—Tb(1)—O(5) <sup>l</sup>	87.96(11)	O(14)—Tb(1)—N(2)	25.64(14)	O(9)—Zn(1)—Tb(1)	96.74(9)
O(1)—Tb(1)—O(9) <sup>l</sup>	70.13(11)	O(13)—Tb(1)—Zn(1)	76.89(10)	O(9)—Zn(1)—Tb(1) <sup>l</sup>	40.36(10)
O(1)—Tb(1)—O(10)	147.21(14)	O(13)—Tb(1)—O(3)	76.40(15)	O(9)—Zn(1)—Zn(1) <sup>l</sup>	42.80(9)
O(1)—Tb(1)—O(11)	159.24(15)	O(13)—Tb(1)—O(11)	115.04(14)	O(9) <sup>l</sup> —Zn(1)—Zn(1) <sup>l</sup>	41.51(8)
O(1)—Tb(1)—N(1)	171.96(16)	O(13)—Tb(1)—N(1)	98.14(15)	O(9) <sup>l</sup> —Zn(1)—O(7)	97.93(14)
O(1)—Tb(1)—O(14)	119.68(12)	O(13)—Tb(1)—O(14)	50.84(15)	O(9)—Zn(1)—O(7)	155.58(15)
O(1)—Tb(1)—O(13)	82.01(12)	O(13)—Tb(1)—N(2)	25.21(15)	O(9)—Zn(1)—O(5)	80.64(14)
O(1)—Tb(1)—N(2)	101.07(13)	N(2)—Tb(1)—Zn(1)	101.94(13)	O(9)—Zn(1)—O(1)	108.80(14)
O(9) <sup>l</sup> —Tb(1)—Zn(1)	35.80(8)	Tb(1)—Zn(1)—Tb(1) <sup>l</sup>	127.820(16)	O(9)—Zn(1)—O(2)	95.20(14)
O(9) <sup>l</sup> —Tb(1)—O(3)	128.42(11)	Zn(1) <sup>l</sup> —Zn(1)—Tb(1)	64.01(2)	O(9)—Zn(1)—O(9) <sup>l</sup>	84.31(13)
<b>Complex <b>1e</b></b>					
Er(1)—Zn(1)	3.4754(6)	Er(1)—O(6)	2.354(4)	Zn(1)—O(15) <sup>l</sup>	2.041(3)
Er(1)—O(15)	2.339(3)	Er(1)—O(2)	2.378(4)	Zn(1)—O(15)	2.084(3)
Er(1)—O(1)	2.305(3)	Er(1)—O(9)	2.444(3)	Zn(1)—O(1)	2.031(3)
Er(1)—O(5)	2.326(3)	Er(1)—N(1)	2.865(5)	Zn(1)—O(5) <sup>l</sup>	2.053(3)
Er(1)—O(13)	2.503(3)	Er(1)—N(2)	2.893(4)	Zn(1)—O(3)	2.204(3)
Er(1)—O(10)	2.441(4)	Zn(1)—Zn(1) <sup>l</sup>	2.9354(11)	Zn(1)—O(7) <sup>l</sup>	2.194(3)
Er(1)—O(12)	2.442(3)				
O(15)—Er(1)—Zn(1)	35.69(8)	O(10)—Er(1)—O(13)	72.54(13)	Zn(1) <sup>l</sup> —Zn(1)—Er(1) <sup>l</sup>	64.25(2)
O(15)—Er(1)—O(13)	149.25(13)	O(10)—Er(1)—O(12)	70.52(13)	O(15) <sup>l</sup> —Zn(1)—Er(1)	101.09(9)
O(15)—Er(1)—O(10)	76.89(12)	O(10)—Er(1)—O(9)	51.94(14)	O(15)—Zn(1)—Er(1)	40.88(10)
O(15)—Er(1)—O(12)	114.47(13)	O(10)—Er(1)—N(1)	25.97(13)	O(15) <sup>l</sup> —Zn(1)—Er(1) <sup>l</sup>	39.23(9)
O(15)—Er(1)—O(6)	130.13(11)	O(10)—Er(1)—N(2)	71.07(13)	O(15)—Zn(1)—Er(1) <sup>l</sup>	98.43(9)
O(15)—Er(1)—O(2)	128.83(12)	O(12)—Er(1)—Zn(1)	139.98(8)	O(15) <sup>l</sup> —Zn(1)—Zn(1) <sup>l</sup>	45.24(9)
O(15)—Er(1)—O(9)	78.36(12)	O(12)—Er(1)—O(13)	51.51(12)	O(15)—Zn(1)—Zn(1) <sup>l</sup>	44.05(9)
O(15)—Er(1)—N(1)	76.62(12)	O(12)—Er(1)—O(9)	116.88(13)	O(15) <sup>l</sup> —Zn(1)—O(15)	89.29(13)
O(15)—Er(1)—N(2)	136.27(13)	O(12)—Er(1)—N(1)	93.60(14)	O(15) <sup>l</sup> —Zn(1)—O(5) <sup>l</sup>	77.82(13)
O(1)—Er(1)—Zn(1)	34.11(8)	O(12)—Er(1)—N(2)	26.01(12)	O(15) <sup>l</sup> —Zn(1)—O(3)	93.15(14)
O(1)—Er(1)—O(15)	69.41(12)	O(6)—Er(1)—Zn(1)	104.46(8)	O(15)—Zn(1)—O(3)	153.15(13)
O(1)—Er(1)—O(5)	86.90(11)	O(6)—Er(1)—O(13)	76.91(12)	O(15)—Zn(1)—O(7) <sup>l</sup>	102.34(13)
O(1)—Er(1)—O(13)	137.15(13)	O(6)—Er(1)—O(10)	144.07(12)	O(15) <sup>l</sup> —Zn(1)—O(7) <sup>l</sup>	151.89(13)
O(1)—Er(1)—O(10)	133.97(12)	O(6)—Er(1)—O(12)	75.84(13)	O(1)—Zn(1)—Er(1)	39.53(9)
O(1)—Er(1)—O(12)	152.73(12)	O(6)—Er(1)—O(2)	72.01(12)	O(1)—Zn(1)—Er(1) <sup>l</sup>	153.82(9)
O(1)—Er(1)—O(6)	81.77(12)	O(6)—Er(1)—O(9)	143.29(13)	O(1)—Zn(1)—Zn(1) <sup>l</sup>	99.57(10)
O(1)—Er(1)—O(2)	70.39(11)	O(6)—Er(1)—N(1)	153.25(12)	O(1)—Zn(1)—O(15) <sup>l</sup>	114.69(13)

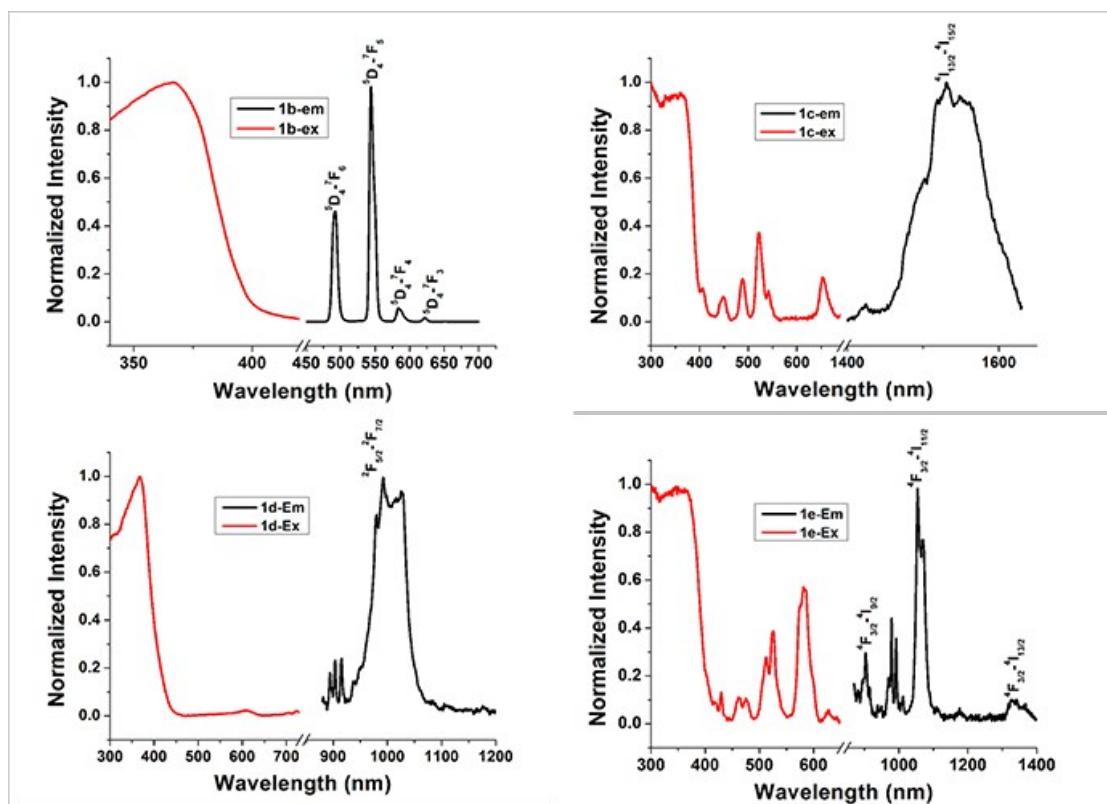
O(1)—Er(1)—O(9)	90.38(13)	O(6)—Er(1)—N(2)	73.19(13)	O(1)—Zn(1)—O(15)	79.94(13)
O(1)—Er(1)—N(1)	113.16(13)	O(2)—Er(1)—Zn(1)	101.52(8)	O(1)—Zn(1)—O(5) <sup>l</sup>	167.00(12)
O(1)—Er(1)—N(2)	152.62(13)	O(2)—Er(1)—O(13)	67.81(12)	O(1)—Zn(1)—O(3)	74.76(12)
O(5)—Er(1)—Zn(1)	70.76(7)	O(2)—Er(1)—O(10)	111.97(13)	O(1)—Zn(1)—O(7) <sup>l</sup>	92.68(13)
O(5)—Er(1)—O(15)	66.90(11)	O(2)—Er(1)—O(12)	115.86(12)	O(5) <sup>l</sup> —Zn(1)—Er(1) <sup>l</sup>	38.97(8)
O(5)—Er(1)—O(13)	119.97(11)	O(2)—Er(1)—O(9)	71.57(14)	O(5) <sup>l</sup> —Zn(1)—Er(1)	137.99(9)
O(5)—Er(1)—O(10)	107.94(13)	O(2)—Er(1)—N(1)	91.60(14)	O(5) <sup>l</sup> —Zn(1)—Zn(1) <sup>l</sup>	86.82(9)
O(5)—Er(1)—O(12)	71.58(11)	O(2)—Er(1)—N(2)	91.09(13)	O(5) <sup>l</sup> —Zn(1)—O(15)	97.38(13)
O(5)—Er(1)—O(6)	71.88(11)	O(9)—Er(1)—Zn(1)	87.19(9)	O(5) <sup>l</sup> —Zn(1)—O(3)	109.28(12)
O(5)—Er(1)—O(2)	139.52(12)	O(9)—Er(1)—O(13)	85.31(12)	O(5) <sup>l</sup> —Zn(1)—O(7) <sup>l</sup>	75.38(12)
O(5)—Er(1)—O(9)	143.78(12)	O(9)—Er(1)—N(1)	25.98(13)	O(3)—Zn(1)—Er(1) <sup>l</sup>	100.29(8)
O(5)—Er(1)—N(1)	128.60(13)	O(9)—Er(1)—N(2)	103.02(13)	O(3)—Zn(1)—Er(1)	112.70(8)
O(5)—Er(1)—N(2)	95.49(12)	N(1)—Er(1)—Zn(1)	99.36(9)	O(3)—Zn(1)—Zn(1) <sup>l</sup>	132.21(10)
O(13)—Er(1)—Zn(1)	168.51(9)	N(1)—Er(1)—N(2)	86.52(13)	O(7) <sup>l</sup> —Zn(1)—Er(1) <sup>l</sup>	113.02(9)
O(13)—Er(1)—N(1)	77.34(12)	N(2)—Er(1)—Zn(1)	165.86(10)	O(7) <sup>l</sup> —Zn(1)—Er(1)	104.41(8)
O(13)—Er(1)—N(2)	25.60(12)	Er(1)—Zn(1)—Er(1) <sup>l</sup>	130.471(17)	O(7) <sup>l</sup> —Zn(1)—Zn(1) <sup>l</sup>	139.91(10)
O(10)—Er(1)—Zn(1)	109.27(9)	Zn(1) <sup>l</sup> —Zn(1)—Er(1)	66.223(18)	O(7) <sup>l</sup> —Zn(1)—O(3)	87.81(14)
<b>Complex 1d</b>					
Yb(1)—Zn(1)	3.4470(7)	Yb(1)—O(7)	2.488(4)	Zn(1)—O(15)	2.090(3)
Yb(1)—O(13)	2.406(4)	Yb(1)—O(2)	2.311(4)	Zn(1)—O(15) <sup>l</sup>	2.047(4)
Yb(1)—O(15)	2.297(4)	Yb(1)—O(9)	2.418(4)	Zn(1)—O(5)	2.043(4)
Yb(1)—O(5)	2.261(3)	Yb(1)—N(2)	2.866(6)	Zn(1)—O(1) <sup>l</sup>	2.075(4)
Yb(1)—O(1)	2.277(4)	Yb(1)—N(1)	2.803(6)	Zn(1)—O(6)	2.052(4)
Yb(1)—O(12)	2.495(4)	Zn(1)—Zn(1) <sup>l</sup>	3.0682(13)	Zn(1)—O(3) <sup>l</sup>	2.187(4)
Yb(1)—O(10)	2.382(4)				
O(13)—Yb(1)—Zn(1)	76.66(11)	O(1)—Yb(1)—N(2)	157.38(15)	Zn(1) <sup>l</sup> —Zn(1)—Yb(1) <sup>l</sup>	63.42(2)
O(13)—Yb(1)—O(12)	51.83(16)	O(1)—Yb(1)—N(1)	86.03(15)	O(15) <sup>l</sup> —Zn(1)—Yb(1)	96.70(11)
O(13)—Yb(1)—O(7)	76.70(15)	O(12)—Yb(1)—Zn(1)	128.15(12)	O(15)—Zn(1)—Yb(1) <sup>l</sup>	95.49(10)
O(13)—Yb(1)—O(9)	115.79(15)	O(12)—Yb(1)—N(2)	25.96(15)	O(15) <sup>l</sup> —Zn(1)—Yb(1) <sup>l</sup>	39.78(11)
O(13)—Yb(1)—N(2)	25.88(15)	O(12)—Yb(1)—N(1)	65.20(16)	O(15)—Zn(1)—Yb(1)	40.38(10)
O(13)—Yb(1)—N(1)	97.57(16)	O(10)—Yb(1)—Zn(1)	112.27(11)	O(15) <sup>l</sup> —Zn(1)—Zn(1) <sup>l</sup>	42.66(10)
O(15)—Yb(1)—Zn(1)	36.12(9)	O(10)—Yb(1)—O(13)	81.18(16)	O(15)—Zn(1)—Zn(1) <sup>l</sup>	41.58(9)
O(15)—Yb(1)—O(13)	72.03(14)	O(10)—Yb(1)—O(12)	69.53(15)	O(15) <sup>l</sup> —Zn(1)—O(15)	84.25(14)
O(15)—Yb(1)—O(12)	116.81(15)	O(10)—Yb(1)—O(7)	137.18(14)	O(15) <sup>l</sup> —Zn(1)—O(1) <sup>l</sup>	78.62(15)
O(15)—Yb(1)—O(10)	76.20(14)	O(10)—Yb(1)—O(9)	52.70(15)	O(15) <sup>l</sup> —Zn(1)—O(6)	96.46(16)
O(15)—Yb(1)—O(7)	128.61(13)	O(10)—Yb(1)—N(2)	74.01(16)	O(15) <sup>l</sup> —Zn(1)—O(3) <sup>l</sup>	153.71(16)
O(15)—Yb(1)—O(2)	136.71(13)	O(10)—Yb(1)—N(1)	26.44(16)	O(15)—Zn(1)—O(3) <sup>l</sup>	96.36(15)
O(15)—Yb(1)—O(9)	123.45(15)	O(7)—Yb(1)—Zn(1)	97.70(9)	O(5)—Zn(1)—Yb(1) <sup>l</sup>	150.02(11)
O(15)—Yb(1)—N(2)	94.63(16)	O(7)—Yb(1)—O(12)	67.94(15)	O(5)—Zn(1)—Yb(1)	39.02(10)
O(15)—Yb(1)—N(1)	100.92(17)	O(7)—Yb(1)—N(2)	70.09(15)	O(5)—Zn(1)—Zn(1) <sup>l</sup>	95.95(11)
O(5)—Yb(1)—Zn(1)	34.67(9)	O(7)—Yb(1)—N(1)	123.03(16)	O(5)—Zn(1)—O(15) <sup>l</sup>	110.27(15)
O(5)—Yb(1)—O(13)	81.99(14)	O(2)—Yb(1)—Zn(1)	113.43(10)	O(5)—Zn(1)—O(15)	79.21(14)
O(5)—Yb(1)—O(15)	70.64(13)	O(2)—Yb(1)—O(13)	144.08(15)	O(5)—Zn(1)—O(1) <sup>l</sup>	169.23(14)
O(5)—Yb(1)—O(1)	88.96(13)	O(2)—Yb(1)—O(12)	106.47(15)	O(5)—Zn(1)—O(6)	85.14(16)

O(5)—Yb(1)—O(12)	119.94(14)	O(2)—Yb(1)—O(10)	121.20(16)	O(5)—Zn(1)—O(3) <sup>l</sup>	95.58(15)
O(5)—Yb(1)—O(10)	146.06(15)	O(2)—Yb(1)—O(7)	67.95(14)	O(1) <sup>l</sup> —Zn(1)—Yb(1) <sup>l</sup>	39.44(10)
O(5)—Yb(1)—O(7)	65.40(13)	O(2)—Yb(1)—O(9)	70.45(15)	O(1) <sup>l</sup> —Zn(1)—Yb(1)	136.25(10)
O(5)—Yb(1)—O(2)	88.86(14)	O(2)—Yb(1)—N(2)	127.30(17)	O(1) <sup>l</sup> —Zn(1)—Zn(1) <sup>l</sup>	86.60(11)
O(5)—Yb(1)—O(9)	159.22(14)	O(2)—Yb(1)—N(1)	96.37(17)	O(1) <sup>l</sup> —Zn(1)—O(15)	96.10(15)
O(5)—Yb(1)—N(2)	101.35(15)	O(9)—Yb(1)—Zn(1)	154.59(12)	O(1) <sup>l</sup> —Zn(1)—O(3) <sup>l</sup>	75.17(15)
O(5)—Yb(1)—N(1)	171.31(16)	O(9)—Yb(1)—O(12)	69.85(16)	O(6)—Zn(1)—Yb(1)	123.52(13)
O(1)—Yb(1)—Zn(1)	74.86(10)	O(9)—Yb(1)—O(7)	106.58(15)	O(6)—Zn(1)—Yb(1) <sup>l</sup>	95.49(12)
O(1)—Yb(1)—O(13)	141.48(15)	O(9)—Yb(1)—N(2)	93.03(17)	O(6)—Zn(1)—Zn(1) <sup>l</sup>	136.74(13)
O(1)—Yb(1)—O(15)	69.64(13)	O(9)—Yb(1)—N(1)	26.39(15)	O(6)—Zn(1)—O(15)	163.52(17)
O(1)—Yb(1)—O(12)	151.09(15)	N(2)—Yb(1)—Zn(1)	102.36(13)	O(6)—Zn(1)—O(1) <sup>l</sup>	100.19(16)
O(1)—Yb(1)—O(10)	86.16(14)	N(1)—Yb(1)—Zn(1)	136.75(14)	O(6)—Zn(1)—O(3) <sup>l</sup>	90.25(17)
O(1)—Yb(1)—O(7)	132.38(13)	N(1)—Yb(1)—N(2)	81.01(17)	O(3) <sup>l</sup> —Zn(1)—Yb(1) <sup>l</sup>	114.38(11)
O(1)—Yb(1)—O(2)	72.37(14)	Yb(1)—Zn(1)—Yb(1) <sup>l</sup>	127.25(2)	O(3) <sup>l</sup> —Zn(1)—Yb(1)	100.67(11)
O(1)—Yb(1)—O(9)	83.13(15)	Zn(1) <sup>l</sup> —Zn(1)—Yb(1)	63.83(2)	O(3) <sup>l</sup> —Zn(1)—Zn(1) <sup>l</sup>	132.34(11)
<b>Complex 1e</b>					
Nd(1)—Zn(1) <sup>l</sup>	3.5274(11)	Nd(1)—O(9)	2.542(6)	Zn(1)—O(5)	2.039(5)
Nd(1)—O(5)	2.432(5)	Nd(1)—O(13)	2.559(6)	Zn(1)—O(15) <sup>l</sup>	2.066(6)
Nd(1)—O(15)	2.407(5)	Nd(1)—N(1)	2.974(7)	Zn(1)—O(15)	2.056(5)
Nd(1)—O(1)	2.450(5)	Nd(1)—N(2)	2.941(8)	Zn(1)—O(1) <sup>l</sup>	2.047(5)
Nd(1)—O(2)	2.449(5)	Nd(1)—O(12)	2.517(6)	Zn(1)—O(7)	2.188(5)
Nd(1)—O(6)	2.439(6)	Zn(1)—Zn(1) <sup>l</sup>	3.0283(17)	Zn(1)—O(3) <sup>l</sup>	2.230(6)
Nd(1)—O(10)	2.563(6)				
O(5)—Nd(1)—Zn(1) <sup>l</sup>	75.29(13)	O(2)—Nd(1)—N(2)	124.9(2)	Zn(1) <sup>l</sup> —Zn(1)—Nd(1)	64.43(3)
O(5)—Nd(1)—O(1)	89.48(18)	O(2)—Nd(1)—O(12)	138.2(2)	O(5)—Zn(1)—Nd(1)	41.75(14)
O(5)—Nd(1)—O(2)	131.1(2)	O(6)—Nd(1)—Zn(1) <sup>l</sup>	116.25(16)	O(5)—Zn(1)—Nd(1) <sup>l</sup>	142.68(16)
O(5)—Nd(1)—O(6)	69.38(18)	O(6)—Nd(1)—O(1)	93.8(2)	O(5)—Zn(1)—Zn(1) <sup>l</sup>	93.13(14)
O(5)—Nd(1)—O(10)	146.77(19)	O(6)—Nd(1)—O(2)	70.0(2)	O(5)—Zn(1)—O(15) <sup>l</sup>	101.8(2)
O(5)—Nd(1)—O(9)	145.51(19)	O(6)—Nd(1)—O(10)	104.4(2)	O(5)—Zn(1)—O(15)	82.8(2)
O(5)—Nd(1)—O(13)	78.7(2)	O(6)—Nd(1)—O(9)	145.0(2)	O(5)—Zn(1)—O(1) <sup>l</sup>	167.9(2)
O(5)—Nd(1)—N(1)	157.2(2)	O(6)—Nd(1)—O(13)	69.4(2)	O(5)—Zn(1)—O(7)	75.9(2)
O(5)—Nd(1)—N(2)	84.2(2)	O(6)—Nd(1)—N(1)	126.0(2)	O(5)—Zn(1)—O(3) <sup>l</sup>	99.6(2)
O(5)—Nd(1)—O(12)	85.8(2)	O(6)—Nd(1)—N(2)	94.3(2)	O(15)—Zn(1)—Nd(1) <sup>l</sup>	97.23(16)
O(15)—Nd(1)—Zn(1) <sup>l</sup>	34.65(13)	O(6)—Nd(1)—O(12)	118.4(2)	O(15) <sup>l</sup> —Zn(1)—Nd(1) <sup>l</sup>	41.49(14)
O(15)—Nd(1)—O(5)	68.07(17)	O(10)—Nd(1)—Zn(1) <sup>l</sup>	131.90(13)	O(15) <sup>l</sup> —Zn(1)—Nd(1)	96.70(14)
O(15)—Nd(1)—O(1)	69.05(17)	O(10)—Nd(1)—N(1)	25.51(19)	O(15)—Zn(1)—Nd(1)	41.15(14)
O(15)—Nd(1)—O(2)	131.16(19)	O(10)—Nd(1)—N(2)	63.4(2)	O(15)—Zn(1)—Zn(1) <sup>l</sup>	42.84(16)
O(15)—Nd(1)—O(6)	133.88(19)	O(9)—Nd(1)—Zn(1) <sup>l</sup>	81.89(15)	O(15) <sup>l</sup> —Zn(1)—Zn(1) <sup>l</sup>	42.59(14)
O(15)—Nd(1)—O(10)	120.98(18)	O(9)—Nd(1)—O(10)	50.1(2)	O(15)—Zn(1)—O(15) <sup>l</sup>	85.4(2)
O(15)—Nd(1)—O(9)	78.42(19)	O(9)—Nd(1)—O(13)	111.5(2)	O(15) <sup>l</sup> —Zn(1)—O(7)	99.0(2)
O(15)—Nd(1)—O(13)	117.63(19)	O(9)—Nd(1)—N(1)	24.65(19)	O(15)—Zn(1)—O(7)	158.7(2)
O(15)—Nd(1)—N(1)	100.01(19)	O(9)—Nd(1)—N(2)	93.1(2)	O(15)—Zn(1)—O(3) <sup>l</sup>	97.0(2)
O(15)—Nd(1)—N(2)	98.3(2)	O(13)—Nd(1)—Zn(1) <sup>l</sup>	148.45(14)	O(15) <sup>l</sup> —Zn(1)—O(3) <sup>l</sup>	158.6(2)
O(15)—Nd(1)—O(12)	75.3(2)	O(13)—Nd(1)—O(10)	68.9(2)	O(1) <sup>l</sup> —Zn(1)—Nd(1) <sup>l</sup>	42.60(15)

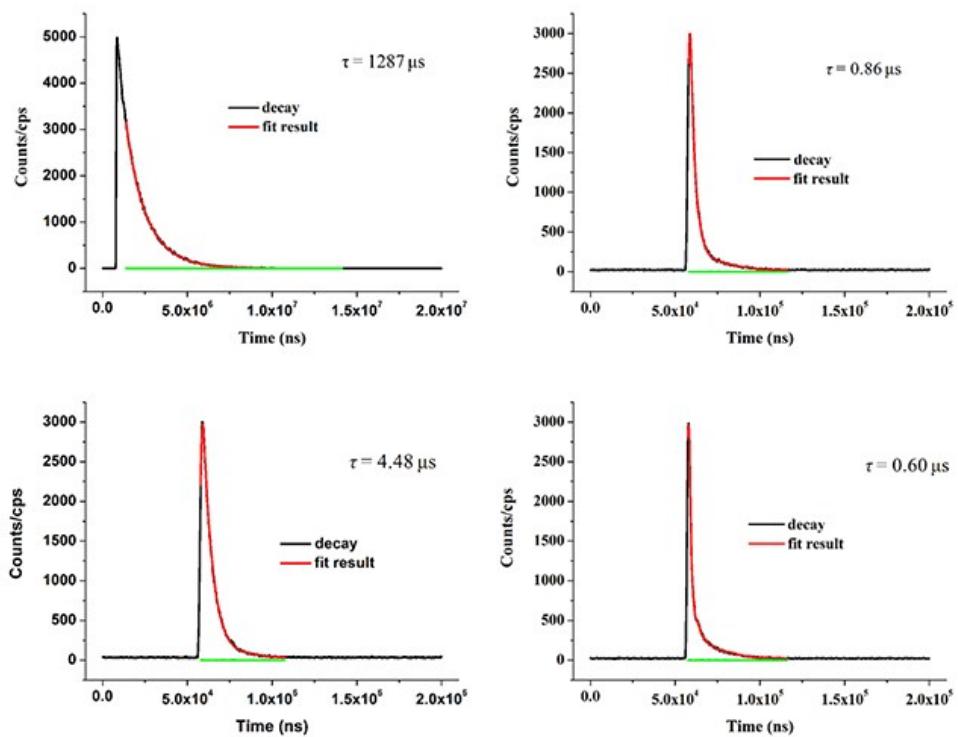
O(1)—Nd(1)—Zn(1) <sup>l</sup>	34.45(11)	O(13)—Nd(1)—N(1)	91.0(2)	O(1) <sup>l</sup> —Zn(1)—Nd(1)	148.92(15)
O(1)—Nd(1)—O(10)	123.74(18)	O(13)—Nd(1)—N(2)	25.1(2)	O(1) <sup>l</sup> —Zn(1)—Zn(1) <sup>l</sup>	98.25(15)
O(1)—Nd(1)—O(9)	85.9(2)	N(1)—Nd(1)—Zn(1) <sup>l</sup>	106.42(15)	O(1) <sup>l</sup> —Zn(1)—O(15) <sup>l</sup>	84.0(2)
O(1)—Nd(1)—O(13)	161.97(19)	N(2)—Nd(1)—Zn(1) <sup>l</sup>	132.88(16)	O(1) <sup>l</sup> —Zn(1)—O(15)	108.4(2)
O(1)—Nd(1)—N(1)	104.7(2)	N(2)—Nd(1)—N(1)	78.3(2)	O(1) <sup>l</sup> —Zn(1)—O(7)	92.8(2)
O(1)—Nd(1)—N(2)	167.30(19)	O(12)—Nd(1)—Zn(1) <sup>l</sup>	109.66(15)	O(1) <sup>l</sup> —Zn(1)—O(3) <sup>l</sup>	75.0(2)
O(1)—Nd(1)—O(12)	143.08(18)	O(12)—Nd(1)—O(10)	68.2(2)	O(7)—Zn(1)—Nd(1)	117.56(15)
O(2)—Nd(1)—Zn(1) <sup>l</sup>	99.89(14)	O(12)—Nd(1)—O(9)	77.8(2)	O(7)—Zn(1)—Nd(1) <sup>l</sup>	99.79(17)
O(2)—Nd(1)—O(1)	67.26(18)	O(12)—Nd(1)—O(13)	50.4(2)	O(7)—Zn(1)—Zn(1) <sup>l</sup>	137.64(18)
O(2)—Nd(1)—O(10)	70.1(2)	O(12)—Nd(1)—N(1)	72.1(2)	O(7)—Zn(1)—O(3) <sup>l</sup>	86.4(2)
O(2)—Nd(1)—O(9)	77.8(2)	O(12)—Nd(1)—N(2)	25.6(2)	O(3) <sup>l</sup> —Zn(1)—Nd(1) <sup>l</sup>	117.32(15)
O(2)—Nd(1)—O(13)	110.7(2)	Nd(1) <sup>l</sup> —Zn(1)—Nd(1)	129.25(3)	O(3) <sup>l</sup> —Zn(1)—Nd(1)	99.05(15)
O(2)—Nd(1)—N(1)	71.6(2)	Zn(1) <sup>l</sup> —Zn(1)—Nd(1) <sup>l</sup>	64.82(3)	O(3) <sup>l</sup> —Zn(1)—Zn(1) <sup>l</sup>	135.92(15)
<b>Complex 2a</b>					
Tb(1)—Zn(1)	3.4729(6)	Tb(1)—O(7)	2.416(3)	Zn(1)—O(15)	1.956(3)
Tb(1)—Zn(2)	3.4910(5)	Tb(1)—O(12)	2.481(3)	Zn(1)—O(11)	1.971(3)
Tb(1)—O(9)	2.401(3)	Tb(1)—O(13)	2.501(3)	Zn(2)—O(9)	2.000(3)
Tb(1)—O(5)	2.364(3)	Tb(1)—N(1)	2.877(4)	Zn(2)—O(5)	2.036(3)
Tb(1)—O(16) <sup>l</sup>	2.394(3)	Zn(1)—O(9)	1.991(3)	Zn(2)—O(17)	1.956(3)
Tb(1)—O(1)	2.380(3)	Zn(1)—O(2)	2.257(3)	Zn(2)—O(10)	1.966(3)
Tb(1)—O(18) <sup>l</sup>	2.350(3)	Zn(1)—O(1)	2.038(3)	Zn(2)—O(6)	2.211(3)
Tb(1)—O(3)	2.437(3)				
Zn(1)—Tb(1)—Zn(2)	57.268(13)	O(1)—Tb(1)—O(12)	129.76(11)	N(1)—Tb(1)—Zn(1)	122.22(10)
O(9)—Tb(1)—Zn(1)	33.78(7)	O(1)—Tb(1)—O(13)	82.05(11)	N(1)—Tb(1)—Zn(2)	172.66(9)
O(9)—Tb(1)—Zn(2)	33.66(7)	O(1)—Tb(1)—N(1)	105.91(12)	O(9)—Zn(1)—Tb(1)	42.10(8)
O(9)—Tb(1)—O(3)	120.66(10)	O(18) <sup>l</sup> —Tb(1)—Zn(1)	107.49(7)	O(9)—Zn(1)—O(2)	154.26(13)
O(9)—Tb(1)—O(7)	132.30(11)	O(18) <sup>l</sup> —Tb(1)—Zn(2)	75.96(7)	O(9)—Zn(1)—O(1)	81.20(12)
O(9)—Tb(1)—O(12)	134.84(11)	O(18) <sup>l</sup> —Tb(1)—O(9)	76.57(10)	O(2)—Zn(1)—Tb(1)	112.55(9)
O(9)—Tb(1)—O(13)	137.23(11)	O(18) <sup>l</sup> —Tb(1)—O(5)	83.05(10)	O(1)—Zn(1)—Tb(1)	41.81(8)
O(9)—Tb(1)—N(1)	141.54(12)	O(18) <sup>l</sup> —Tb(1)—O(16) <sup>l</sup>	80.74(10)	O(1)—Zn(1)—O(2)	73.18(12)
O(5)—Tb(1)—Zn(1)	85.16(7)	O(18) <sup>l</sup> —Tb(1)—O(1)	142.27(10)	O(15)—Zn(1)—Tb(1)	119.69(9)
O(5)—Tb(1)—Zn(2)	34.34(7)	O(18) <sup>l</sup> —Tb(1)—O(3)	142.78(11)	O(15)—Zn(1)—O(9)	110.35(12)
O(5)—Tb(1)—O(9)	67.92(10)	O(18) <sup>l</sup> —Tb(1)—O(7)	77.41(11)	O(15)—Zn(1)—O(2)	85.12(13)
O(5)—Tb(1)—O(16) <sup>l</sup>	138.19(10)	O(18) <sup>l</sup> —Tb(1)—O(12)	72.95(11)	O(15)—Zn(1)—O(1)	130.15(13)
O(5)—Tb(1)—O(1)	90.01(10)	O(18) <sup>l</sup> —Tb(1)—O(13)	123.79(10)	O(15)—Zn(1)—O(11)	103.10(14)
O(5)—Tb(1)—O(3)	75.46(11)	O(18) <sup>l</sup> —Tb(1)—N(1)	97.90(12)	O(11)—Zn(1)—Tb(1)	132.28(11)
O(5)—Tb(1)—O(7)	69.81(10)	O(3)—Tb(1)—Zn(1)	100.64(8)	O(11)—Zn(1)—O(9)	105.68(14)
O(5)—Tb(1)—O(12)	137.79(11)	O(3)—Tb(1)—Zn(2)	100.45(8)	O(11)—Zn(1)—O(2)	89.83(14)
O(5)—Tb(1)—O(13)	143.15(10)	O(3)—Tb(1)—O(12)	103.63(12)	O(11)—Zn(1)—O(1)	120.67(14)
O(5)—Tb(1)—N(1)	150.13(12)	O(3)—Tb(1)—O(13)	68.12(11)	O(9)—Zn(2)—Tb(1)	41.72(8)
O(16) <sup>l</sup> —Tb(1)—Zn(1)	63.81(7)	O(3)—Tb(1)—N(1)	86.87(12)	O(9)—Zn(2)—O(5)	82.52(12)
O(16) <sup>l</sup> —Tb(1)—Zn(2)	104.05(7)	O(7)—Tb(1)—Zn(1)	153.93(7)	O(9)—Zn(2)—O(6)	151.18(12)
O(16) <sup>l</sup> —Tb(1)—O(9)	70.91(10)	O(7)—Tb(1)—Zn(2)	101.08(8)	O(5)—Zn(2)—Tb(1)	40.90(8)
O(16) <sup>l</sup> —Tb(1)—O(3)	134.59(11)	O(7)—Tb(1)—O(3)	66.83(11)	O(5)—Zn(2)—O(6)	74.31(11)

O(16) <sup>1</sup> —Tb(1)—O(7)	141.33(10)	O(7)—Tb(1)—O(12)	71.34(11)	O(17)—Zn(2)—Tb(1)	111.28(9)
O(16) <sup>1</sup> —Tb(1)—O(12)	71.93(11)	O(7)—Tb(1)—O(13)	90.39(12)	O(17)—Zn(2)—O(9)	109.54(12)
O(16) <sup>1</sup> —Tb(1)—O(13)	75.87(11)	O(7)—Tb(1)—N(1)	81.21(12)	O(17)—Zn(2)—O(5)	105.42(13)
O(16) <sup>1</sup> —Tb(1)—N(1)	70.64(12)	O(12)—Tb(1)—Zn(1)	134.73(8)	O(17)—Zn(2)—O(10)	113.10(15)
O(1)—Tb(1)—Zn(1)	34.81(7)	O(12)—Tb(1)—Zn(2)	148.90(8)	O(17)—Zn(2)—O(6)	93.21(12)
O(1)—Tb(1)—Zn(2)	77.59(7)	O(12)—Tb(1)—O(13)	51.38(11)	O(10)—Zn(2)—Tb(1)	130.52(11)
O(1)—Tb(1)—O(9)	66.53(10)	O(12)—Tb(1)—N(1)	25.43(11)	O(10)—Zn(2)—O(9)	101.90(13)
O(1)—Tb(1)—O(16) <sup>1</sup>	80.13(10)	O(13)—Tb(1)—Zn(1)	106.47(8)	O(10)—Zn(2)—O(5)	136.66(14)
O(1)—Tb(1)—O(3)	68.57(11)	O(13)—Tb(1)—Zn(2)	159.28(8)	O(10)—Zn(2)—O(6)	84.22(13)
O(1)—Tb(1)—O(7)	134.30(10)	O(13)—Tb(1)—N(1)	26.02(11)	O(6)—Zn(2)—Tb(1)	114.05(8)
<b>Complex 2b</b>					
Nd(1)—Zn(1)	3.5362(7)	Nd(1)—O(13)	2.563(4)	Zn(1)—O(11)	1.977(4)
Nd(1)—Zn(2)	3.5287(7)	Nd(1)—O(16) <sup>1</sup>	2.417(4)	Zn(1)—O(15)	1.962(4)
Nd(1)—O(1)	2.422(4)	Nd(1)—O(18) <sup>1</sup>	2.455(4)	Zn(2)—O(5)	2.021(4)
Nd(1)—O(3)	2.470(4)	Nd(1)—N(1)	2.947(5)	Zn(2)—O(6)	2.266(4)
Nd(1)—O(5)	2.452(4)	Zn(1)—O(1)	2.032(3)	Zn(2)—O(9)	2.004(4)
Nd(1)—O(7)	2.493(4)	Zn(1)—O(2)	2.217(4)	Zn(2)—O(10)	1.978(4)
Nd(1)—O(9)	2.473(4)	Zn(1)—O(9)	1.998(4)	Zn(2)—O(17)	1.966(4)
Nd(1)—O(12)	2.528(4)				
Zn(2)—Nd(1)—Zn(1)	56.687(15)	O(9)—Nd(1)—Zn(1)	33.23(9)	N(1)—Nd(1)—Zn(1)	172.45(13)
O(1)—Nd(1)—Zn(1)	33.76(8)	O(9)—Nd(1)—Zn(2)	33.52(8)	N(1)—Nd(1)—Zn(2)	122.01(12)
O(1)—Nd(1)—Zn(2)	83.67(8)	O(9)—Nd(1)—O(7)	119.41(13)	O(1)—Zn(1)—Nd(1)	41.48(10)
O(1)—Nd(1)—O(3)	68.26(12)	O(9)—Nd(1)—O(12)	136.76(14)	O(1)—Zn(1)—O(2)	74.73(14)
O(1)—Nd(1)—O(5)	88.86(13)	O(9)—Nd(1)—O(13)	136.03(14)	O(2)—Zn(1)—Nd(1)	114.92(9)
O(1)—Nd(1)—O(7)	76.20(14)	O(9)—Nd(1)—N(1)	141.30(16)	O(9)—Zn(1)—Nd(1)	42.72(10)
O(1)—Nd(1)—O(9)	66.87(12)	O(12)—Nd(1)—Zn(1)	149.95(10)	O(9)—Zn(1)—O(1)	84.03(15)
O(1)—Nd(1)—O(12)	138.33(13)	O(12)—Nd(1)—Zn(2)	136.23(10)	O(9)—Zn(1)—O(2)	151.64(14)
O(1)—Nd(1)—O(13)	144.82(14)	O(12)—Nd(1)—O(13)	50.32(14)	O(11)—Zn(1)—Nd(1)	130.97(13)
O(1)—Nd(1)—O(18) <sup>1</sup>	136.59(12)	O(12)—Nd(1)—N(1)	25.09(14)	O(11)—Zn(1)—O(1)	137.46(18)
O(1)—Nd(1)—N(1)	151.58(15)	O(13)—Nd(1)—Zn(1)	159.12(11)	O(11)—Zn(1)—O(2)	83.65(16)
O(3)—Nd(1)—Zn(1)	99.04(9)	O(13)—Nd(1)—Zn(2)	105.86(11)	O(11)—Zn(1)—O(9)	100.40(16)
O(3)—Nd(1)—Zn(2)	150.75(9)	O(13)—Nd(1)—N(1)	25.36(15)	O(15)—Zn(1)—Nd(1)	111.11(12)
O(3)—Nd(1)—O(7)	67.43(13)	O(16) <sup>1</sup> —Nd(1)—Zn(1)	75.17(9)	O(15)—Zn(1)—O(1)	104.62(15)
O(3)—Nd(1)—O(9)	130.36(12)	O(16) <sup>1</sup> —Nd(1)—Zn(2)	107.65(9)	O(15)—Zn(1)—O(2)	92.87(15)
O(3)—Nd(1)—O(12)	73.00(13)	O(16) <sup>1</sup> —Nd(1)—O(1)	82.31(12)	O(15)—Zn(1)—O(9)	110.79(15)
O(3)—Nd(1)—O(13)	93.43(14)	O(16) <sup>1</sup> —Nd(1)—O(3)	77.47(12)	O(15)—Zn(1)—O(11)	112.76(18)
O(3)—Nd(1)—N(1)	84.23(16)	O(16) <sup>1</sup> —Nd(1)—O(5)	141.41(12)	O(5)—Zn(2)—Nd(1)	42.46(11)
O(5)—Nd(1)—Zn(1)	76.78(9)	O(16) <sup>1</sup> —Nd(1)—O(7)	143.60(13)	O(5)—Zn(2)—O(6)	73.29(15)
O(5)—Nd(1)—Zn(2)	33.81(8)	O(16) <sup>1</sup> —Nd(1)—O(9)	76.63(12)	O(6)—Zn(2)—Nd(1)	113.37(11)
O(5)—Nd(1)—O(3)	133.19(12)	O(16) <sup>1</sup> —Nd(1)—O(12)	74.81(13)	O(9)—Zn(2)—Nd(1)	42.95(10)
O(5)—Nd(1)—O(7)	67.67(13)	O(16) <sup>1</sup> —Nd(1)—O(13)	124.22(14)	O(9)—Zn(2)—O(5)	82.77(15)
O(5)—Nd(1)—O(9)	65.41(12)	O(16) <sup>1</sup> —Nd(1)—O(18) <sup>1</sup>	80.17(13)	O(9)—Zn(2)—O(6)	155.95(15)
O(5)—Nd(1)—O(12)	130.04(13)	O(16) <sup>1</sup> —Nd(1)—N(1)	99.07(15)	O(10)—Zn(2)—Nd(1)	133.01(13)
O(5)—Nd(1)—O(13)	82.46(14)	O(18) <sup>1</sup> —Nd(1)—Zn(1)	103.05(9)	O(10)—Zn(2)—O(5)	122.45(17)

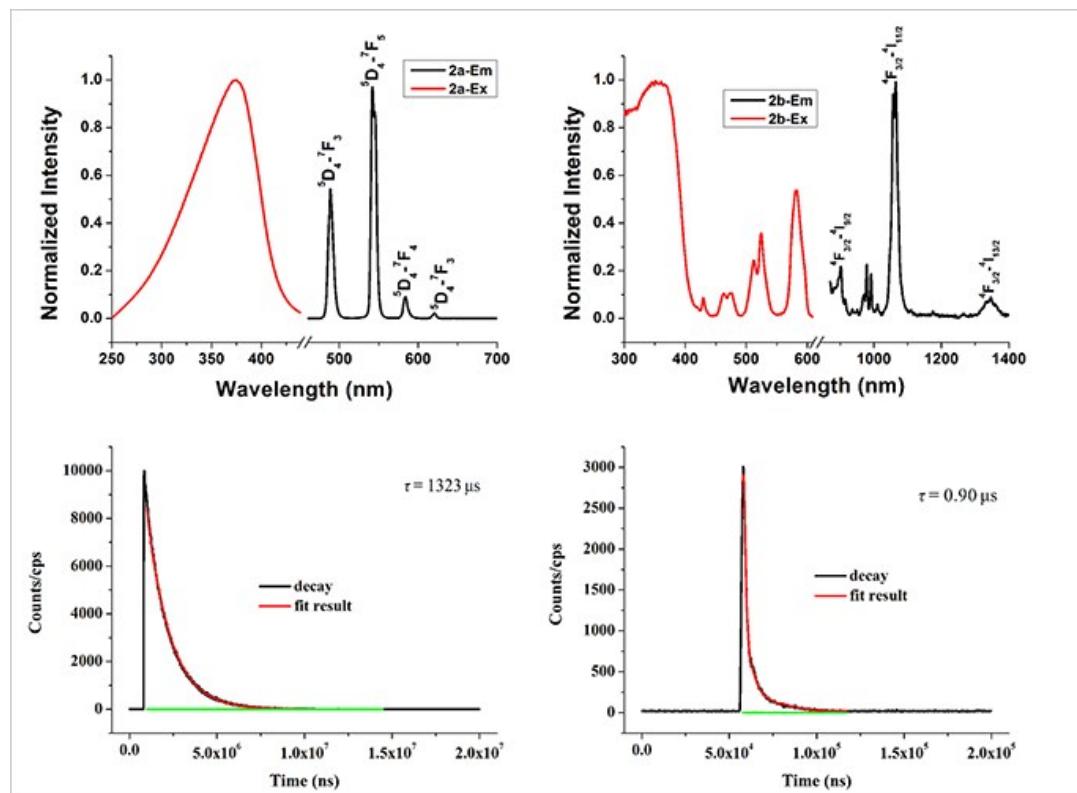
O(5)—Nd(1)—O(18) <sup>l</sup>	80.93(13)	O(18) <sup>l</sup> —Nd(1)—Zn(2)	64.70(9)	O(10)—Zn(2)—O(6)	90.42(18)
O(5)—Nd(1)—N(1)	105.98(15)	O(18) <sup>l</sup> —Nd(1)—O(3)	143.18(13)	O(10)—Zn(2)—O(9)	104.64(17)
O(7)—Nd(1)—Zn(1)	100.49(11)	O(18) <sup>l</sup> —Nd(1)—O(7)	134.80(14)	O(17)—Zn(2)—Nd(1)	118.43(12)
O(7)—Nd(1)—Zn(2)	98.85(10)	O(18) <sup>l</sup> —Nd(1)—O(9)	70.55(12)	O(17)—Zn(2)—O(5)	128.35(17)
O(7)—Nd(1)—O(12)	102.72(15)	O(18) <sup>l</sup> —Nd(1)—O(12)	73.11(14)	O(17)—Zn(2)—O(6)	84.33(16)
O(7)—Nd(1)—O(13)	68.95(15)	O(18) <sup>l</sup> —Nd(1)—O(13)	75.60(14)	O(17)—Zn(2)—O(9)	109.64(15)
O(7)—Nd(1)—N(1)	87.04(17)	O(18) <sup>l</sup> —Nd(1)—N(1)	70.82(16)	O(17)—Zn(2)—O(10)	103.14(17)



**Figure S6.** Excitation spectrum and emission of the complexes **1b-1e**.

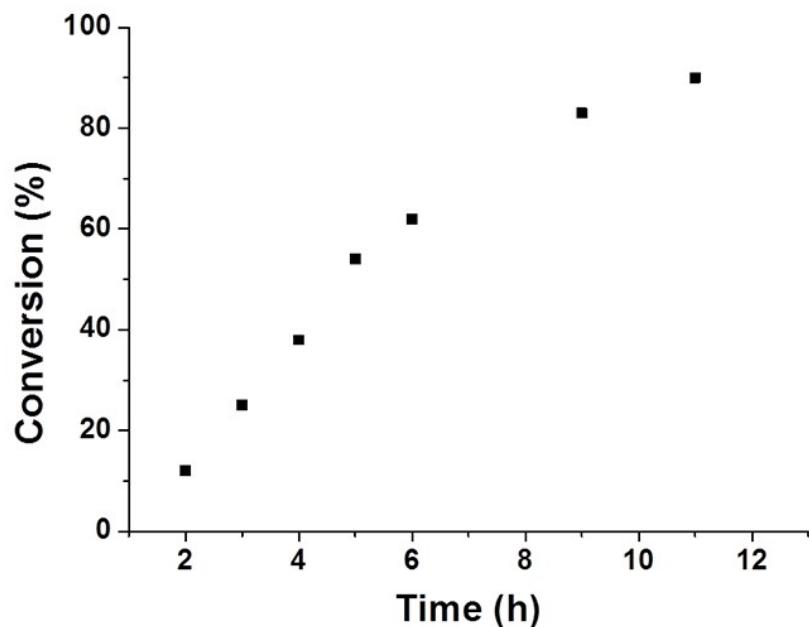


**Figure S7.** Decay curves of the complex **1b** ( $\text{Tb}^{\text{III}}(^5\text{D}_4-^7\text{F}_5)$ ), complex **1c** ( $\text{Er}^{\text{III}}(^4\text{I}_{13/2}-^4\text{I}_{15/2})$ ), complex **1d** ( $\text{Yb}^{\text{III}}(^2\text{F}_{5/2}-^2\text{F}_{7/2})$ ) and complex **1e** ( $\text{Nd}^{\text{III}}(^4\text{F}_{3/2}-^4\text{I}_{11/2})$ ), respectively.



**Figure S8.** Excitation spectrum and emission of the complexes **2a**, **2b** and their decay

curves.



**Figure S9.** Kinetic curve of catalytic reaction at atmospheric pressure and 120 °C (styrene oxide (10 mmol), 0.01 mol % catalyst **1b**, 0.8 mol % co-catalyst, solvent-free).

**Table S3** Cycloaddition of CO<sub>2</sub> and styrene oxide under various pressure <sup>a</sup>

Entry	Cat.	Co-cat.	T (°C)	P (MPa)	t (h)	Conversion <sup>b</sup> (%)
1	<b>1b</b>	TBAB	120	1	2	97
2	<b>1b</b>	TBAB	120	0.8	2	96
3	<b>1b</b>	TBAB	120	0.6	2	92
4	<b>1b</b>	TBAB	120	0.5	2	91

a. Reaction conditions: styrene oxide (10 mmol), 0.01 mol % catalyst **1b**, 0.8 mol % co-catalyst, solvent-free, 120 °C, 2 h. b. Determined by <sup>1</sup>H NMR spectroscopy.

**Table S4.** Overview of reported Ln-complexes catalysts under mild conditions

Cat. (%)	Co-cat.	Catalyst/Epoxyde	P CO <sub>2</sub>	T (°C)	Time (h)	Conversion (%)	TON	TOF (h <sup>-1</sup> )	Ref
Nd(C <sub>5</sub> H <sub>5</sub> ) <sub>3</sub> (THF)	TBAI	(styrene oxide)	1	85	24	93	465	19	1
	(0.8)	1:500							
Ionic Rare Earth	-	(styrene oxide)	1	90	12	93	465	38	2
Metal Complexes		1:500							
{[Eu(BTB)(phen)]·4.5DMF·2H <sub>2</sub> O} <sub>n</sub>	TBAB	(styrene oxide)	1	80	12	68	24	2	3
	(2.5)	1:286							
Tb <sub>4</sub> MOF	TBAB	(styrene oxide)	1	60	12	95	48	4	2
	(2.5)	1:50							
<b>1·Gd</b>	TBAB	(styrene oxide)	1	60	12	70	12	1	4
	(2.5)	1:133							
3d-4f MOF	TBAB	(styrene oxide)	1	70	12	77	78	6.5	5
	(2.5)	1:100							
Yb-mesocate	TBAB	(styrene oxide)	10	120	2.5	95	950	380	6
	(0.75)	1:1000							
3d–4f helicates	TBAB	(styrene oxide)	10	80	6	94	3762	627	7
	(0.75)	1:4000							
[La{N(SiHMe <sub>2</sub> ) <sub>2</sub> } <sub>2</sub> ] <sup>-</sup>	TBAB	(propylene oxide)	10	70	4	95	236	59	8
{k <sup>3</sup> -bpzcp}]	(0.05)	)							
Yb-DDPY	TBAB	(styrene oxide)	10	60	12	52	204	17	9
	(0.5)	1:400							
<b>Complex 1b</b>	TBAB	(styrene oxide)	1	80	14	<b>92.6</b>	<b>9260</b>	<b>661</b>	This work
	(0.8)	<b>1:10000</b>							
<b>Complex 2a</b>	TBAB	(styrene oxide)	1	80	14	<b>93.5</b>	<b>9350</b>	<b>668</b>	This work
	(0.8)	<b>1:10000</b>							

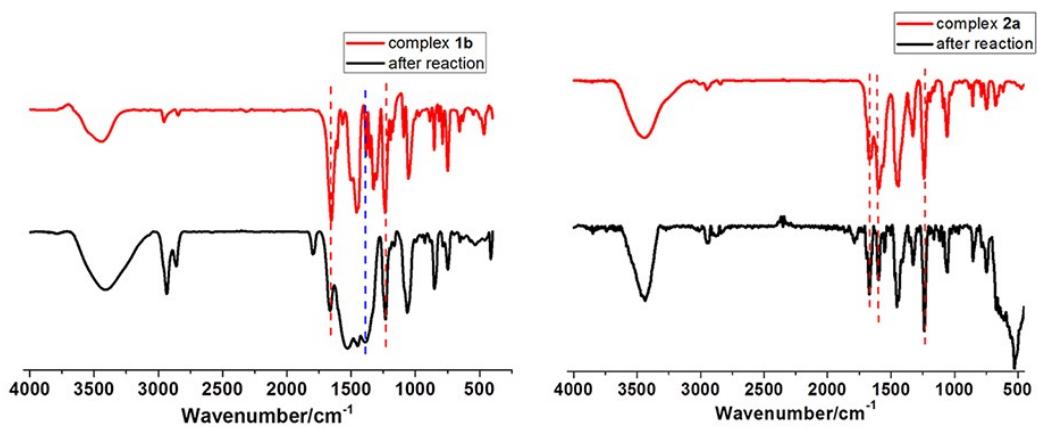
**Table S5.** Overview of reported Mg, Al, Zn, Fe-catalysts under mild conditions

Cat. (%)	Co-cat.	Catalyst/Epoxyd	p <sub>CO<sub>2</sub></sub>	T (°C)	Tim (h)	Conversion(% )	TON	TOF (h <sup>-1</sup> )	Ref
bimetallic Fe-catalyst	[PPN]C	(styrene oxide)	1	80	24	83	830	8	10
	1	1:1000							
<b>Complex 1b</b>	TBAB	(styrene oxide)	1	80	14	<b>92.6</b>	<b>9260</b>	<b>661</b>	This work
	(0.8)	<b>1:10000</b>							

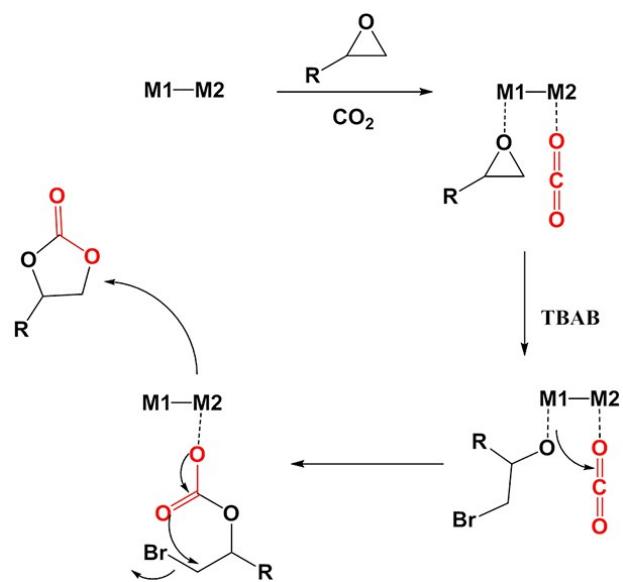
<b>Complex 2a</b>	<b>TBAB</b>	<b>(styrene oxide)</b>	<b>1</b>	<b>80</b>	<b>14</b>	<b>93.5</b>	<b>9350</b>	<b>668</b>	<b>This</b>
		<b>(0.8)</b>	<b>1:10000</b>						<b>wor</b>
Zn <sub>4</sub> (OCOCF <sub>3</sub> ) <sub>6</sub>	TBAI	(propylene oxide)	1	25	6	94	94	15.6	11
O	(3.0)	oxide)							
dimetallic Al-	TBAB	(styrene oxide)	1	25	24	98	39.2	-	12
complex	(2.5)	1:40							
Al-iPOP-2	-	(propylene oxide)	1	25	8	>99	992	124	16
Al-HCP	TBAB	(propylene oxide)	1	25	5	99	396	79.2	17
	(2)	oxide)							
<b>Complex 1b</b>	<b>TBAB</b>	<b>(styrene oxide)</b>	<b>1</b>	<b>25</b>	<b>48</b>	<b>93.5</b>	<b>748</b>	<b>15.6</b>	<b>This</b>
		<b>(3.6)</b>	<b>1:800</b>						<b>wor</b>
<b>Complex 2a</b>	<b>TBAB</b>	<b>(styrene oxide)</b>	<b>1</b>	<b>25</b>	<b>48</b>	<b>95</b>	<b>760</b>	<b>15.8</b>	<b>This</b>
		<b>(3.6)</b>	<b>1:800</b>						<b>wor</b>
salen Al-	-	(AGE)	10	100	2.5	96	182.5	76	13
complexes		1:200							
organocatalysts	TBAI	(styrene oxide)	10	50	18	>99	>66	>3.7	14
	(5.0)	1:66							
binary Al-	TBAB	(styrene oxide)	10	90	2	13	1300	650	15
catalysts	(0.05)	1:1000							
Al-iPOP	-	(propylene oxide)	10	40	3	99	990	330	16
		oxide)							
Al-HCP	TBAB	(propylene oxide)	10	40	1	99	396	396	17
	(2.0)	oxide)							
helical	TBAB	(propylene oxide)	10	r.t.	24	96	19.2	0.8	18
bimetallic	(5.0)	oxide)							
Mg-complex	-	(propylene oxide)	15	120	6	99	1980	3300	19
		oxide)					0		
Salen-Zn	TBAI	(styrene oxide)	10	80	6	89	594	99	20
	(0.5)	1:670							
Zn <sub>3</sub>	-	(epichlorohydrin)	10	85	18	80	32.4	1.8	21
		)							
<b>Complex 1b</b>	<b>TBAB</b>	<b>(propylene oxide)</b>	<b>10</b>	<b>80</b>	<b>8</b>	<b>98</b>	<b>9800</b>	<b>1225</b>	<b>This</b>
		<b>(0.8)</b>	<b>oxide)</b>						<b>wor</b>
<b>Complex 2a</b>	<b>TBAB</b>	<b>(propylene oxide)</b>	<b>10</b>	<b>80</b>	<b>8</b>	<b>&gt;99</b>	<b>&gt;990</b>	<b>&gt;123</b>	<b>This</b>
		<b>(0.8)</b>	<b>oxide)</b>				<b>0</b>	<b>7</b>	<b>wor</b>

**Table S6.** Overview of reported catalysts for the conversion of CO<sub>2</sub> to cyclic carbonates under ambient conditions

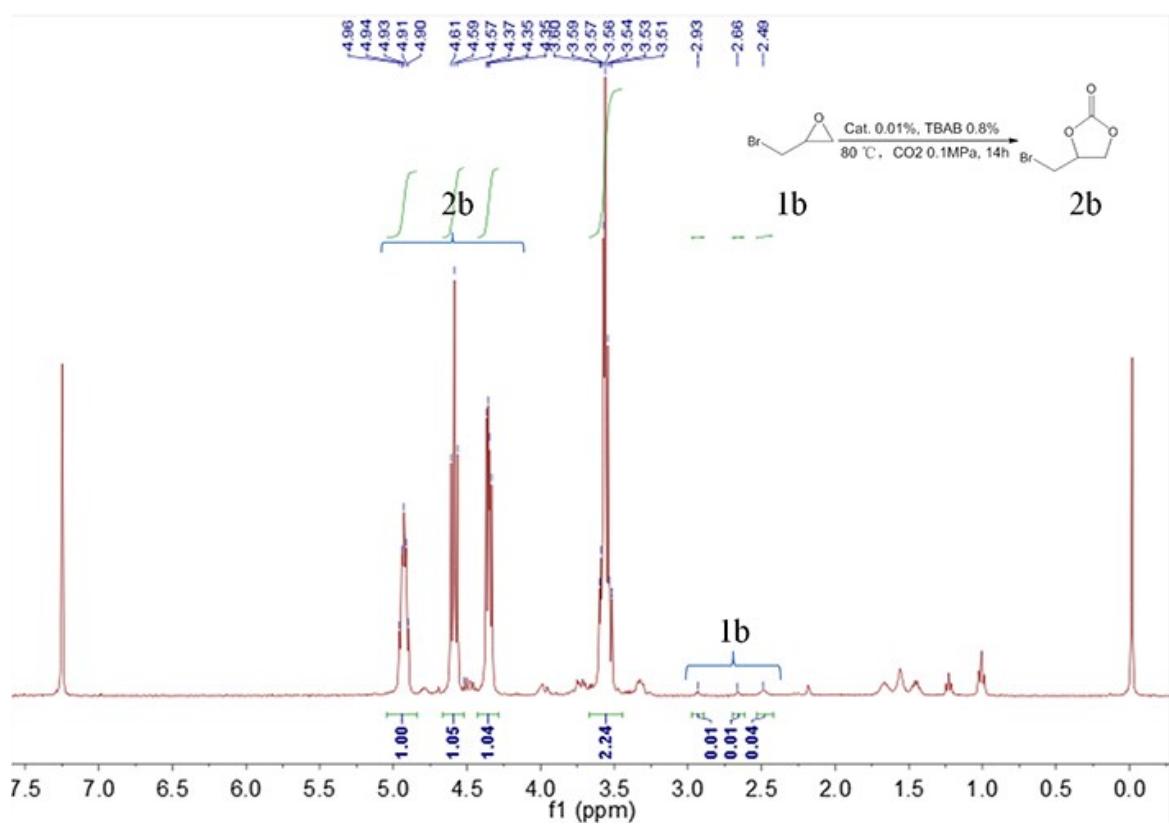
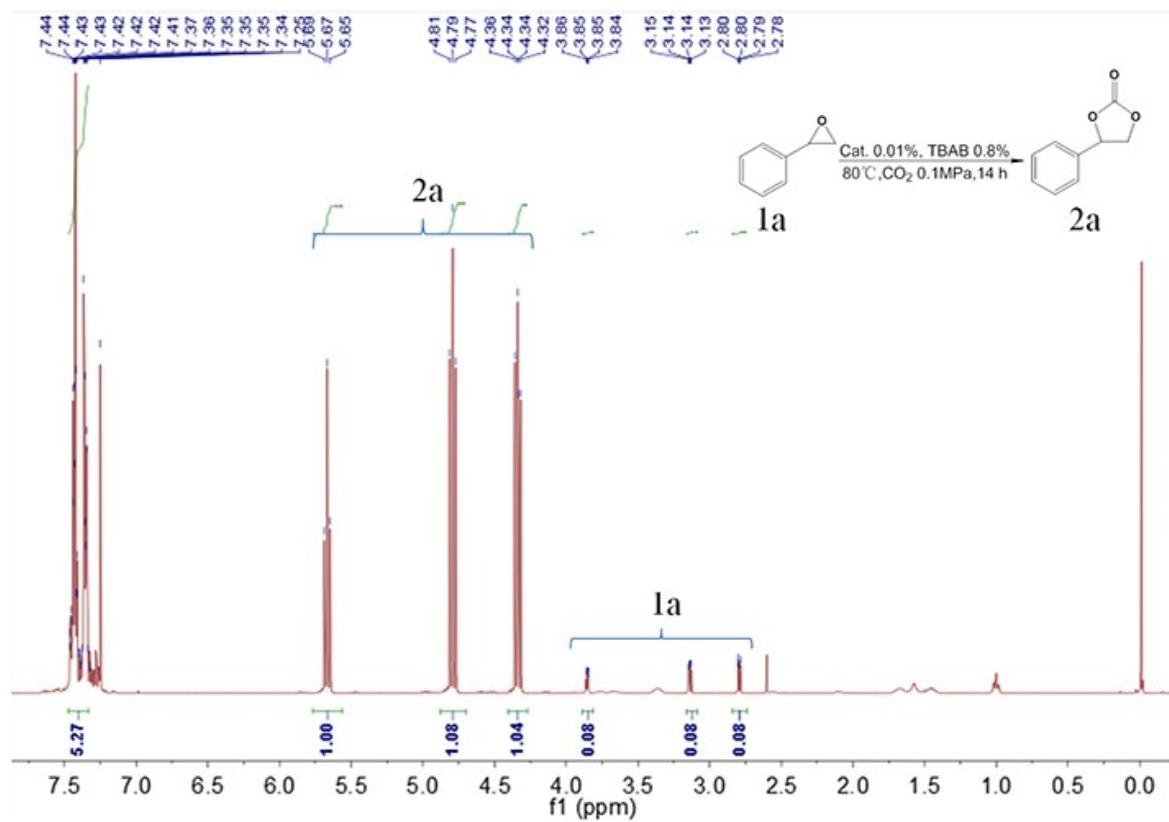
Cat. (%)	Co-cat. (%)	Catalyst/Epoxyde (Mole ratio)	Time (h)	Yields (%)	TON	TOF (h <sup>-1</sup> )	Ref
MMPF-9 (0.125mol % MOF)	TBAB (7.2)	(propylene oxide)	48	87	696	14.5	22
MMCF-2 (0.125mol % MOF)	TBAB (7.2)	(propylene oxide)	48	95	760	15.8	23
HKUST-1 (0.125mol % MOF)	TBAB (7.2)	(propylene oxide)	48	49	392	8.2	23
MOF-505 (0.125mol % MOF)	TBAB (7.2)	(propylene oxide)	48	48	384	8	23
heterometallic helicate	TBAB (7.5)	(styrene oxide) 1:100	48	81	648	13.5	8
MIL-53 (1mol % MOF)	TBAB (6.5)	(propylene oxide)	72	54	-	-	24
MOF-253 (1mol % MOF)	TBAB (6.5)	(propylene oxide)	72	82	-	-	24
MIL-101 (1mol % MOF)	TBAB (6.5)	(propylene oxide)	72	31	-	-	24
UiO-66 (1mol % MOF)	TBAB (6.5)	(propylene oxide)	72	55	-	-	24
<b>Complex 1b</b>	<b>TBAB</b> (7.2)	<b>(styrene oxide)</b> <b>1:800</b>	<b>48</b>	<b>&gt;99</b>	<b>&gt;792</b>	<b>&gt;16.5</b>	<b>This work</b>
<b>Complex 1b</b>	<b>TBAB</b> (3.6)	<b>(styrene oxide)</b> <b>1:800</b>	<b>48</b>	<b>93.5</b>	<b>748</b>	<b>15.6</b>	<b>This work</b>
<b>Complex 2a</b>	<b>TBAB</b> (7.2)	<b>(styrene oxide)</b> <b>1:800</b>	<b>48</b>	<b>&gt;99</b>	<b>&gt;792</b>	<b>&gt;16.5</b>	<b>This work</b>
<b>Complex 2a</b>	<b>TBAB</b> (3.6)	<b>(styrene oxide)</b> <b>1:800</b>	<b>48</b>	<b>95</b>	<b>760</b>	<b>15.8</b>	<b>This work</b>

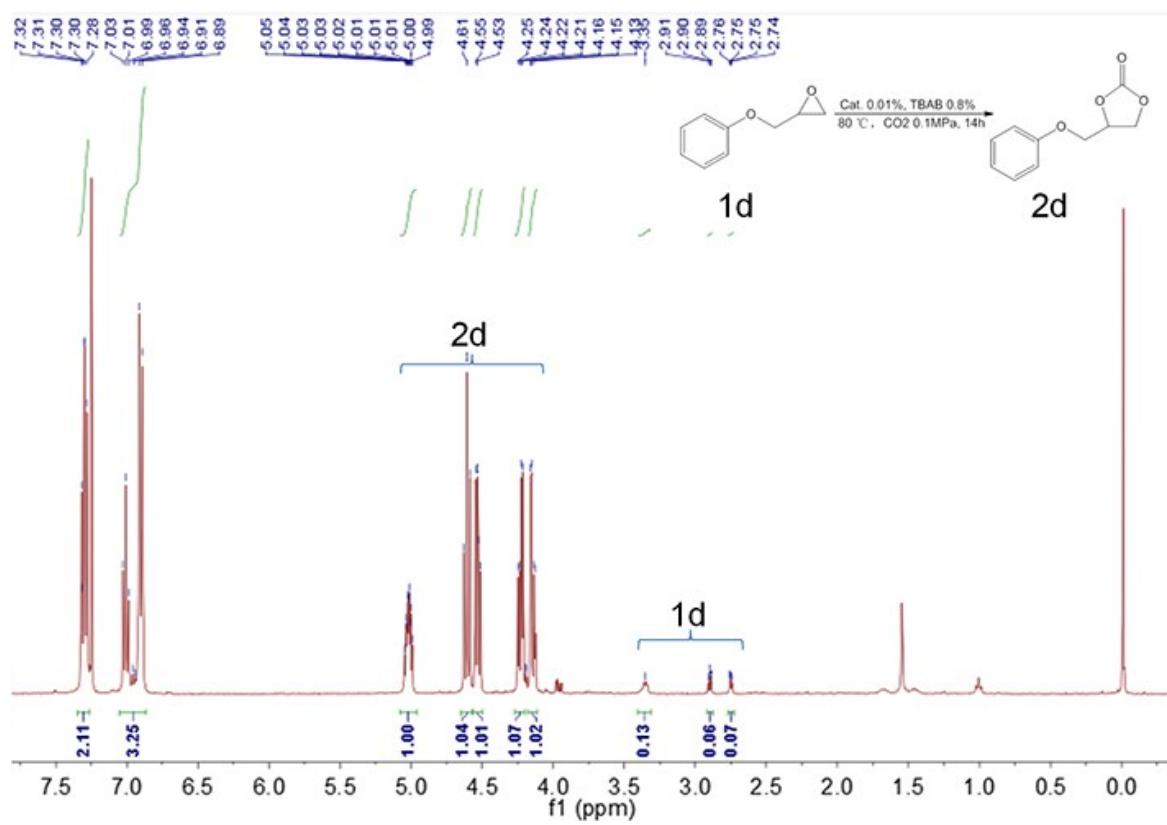
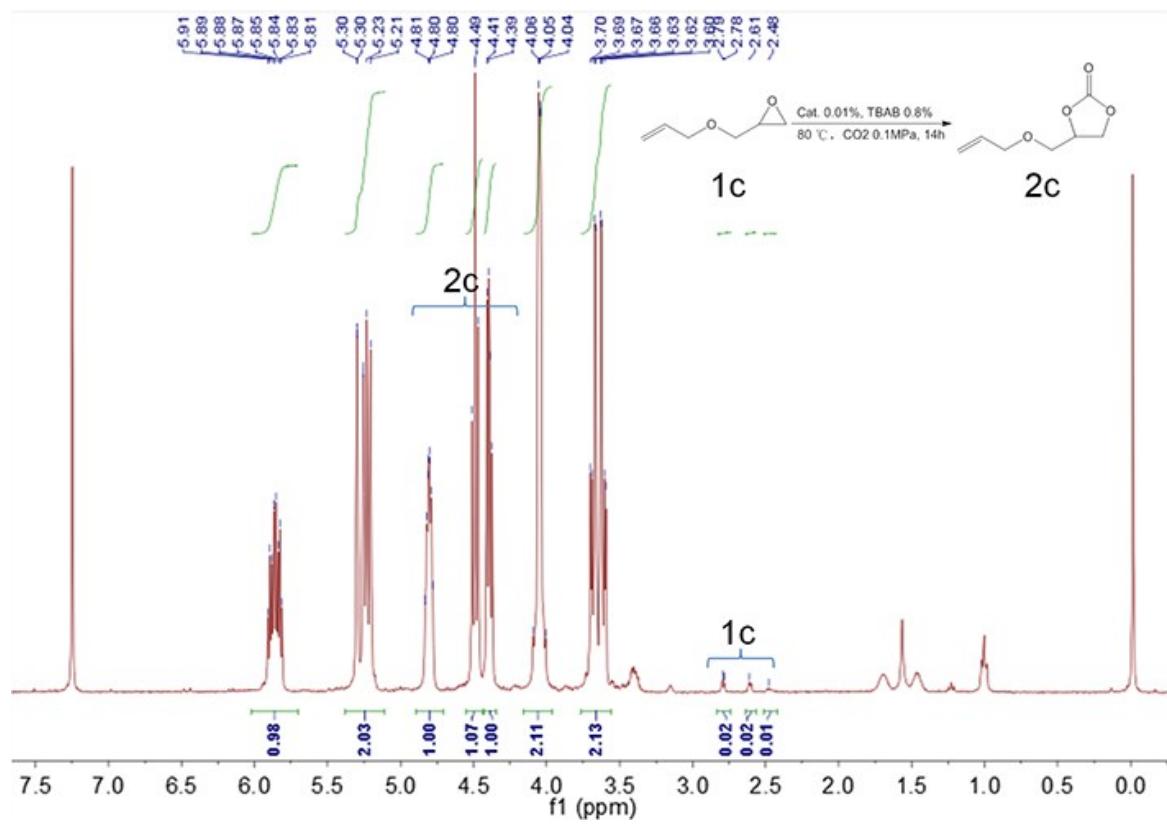


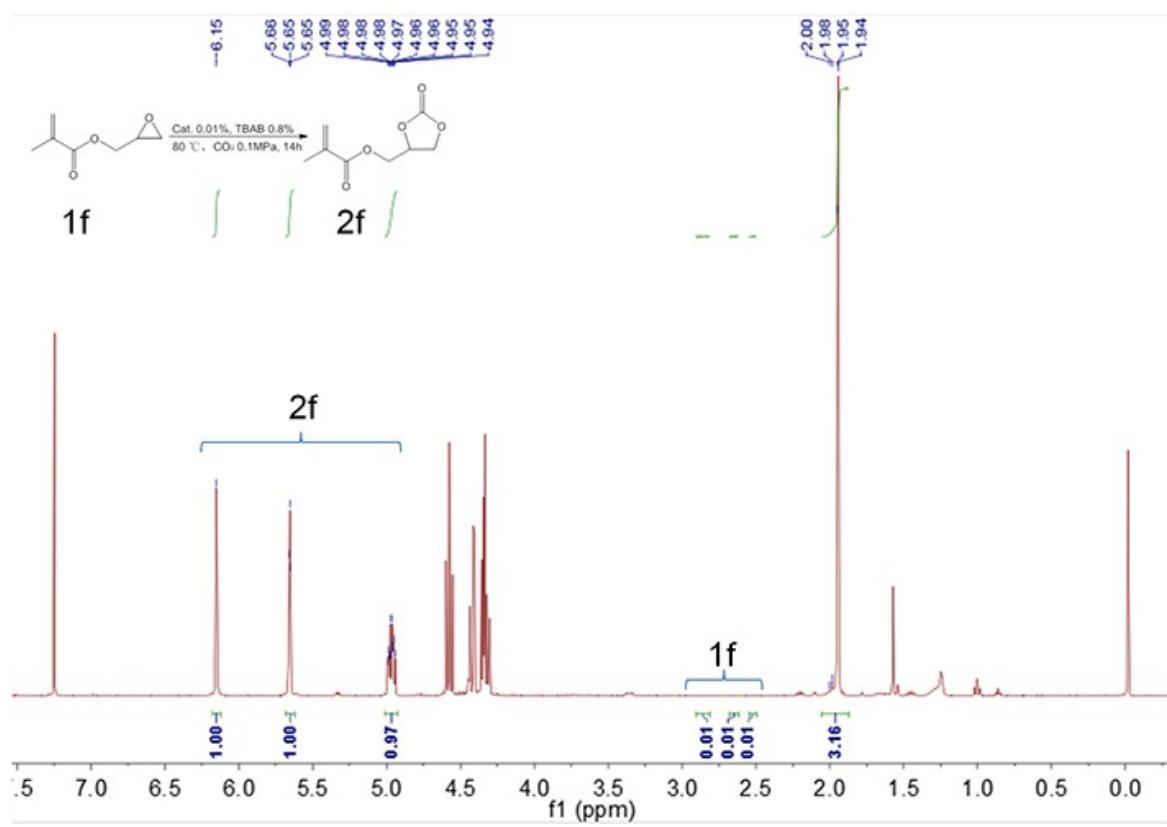
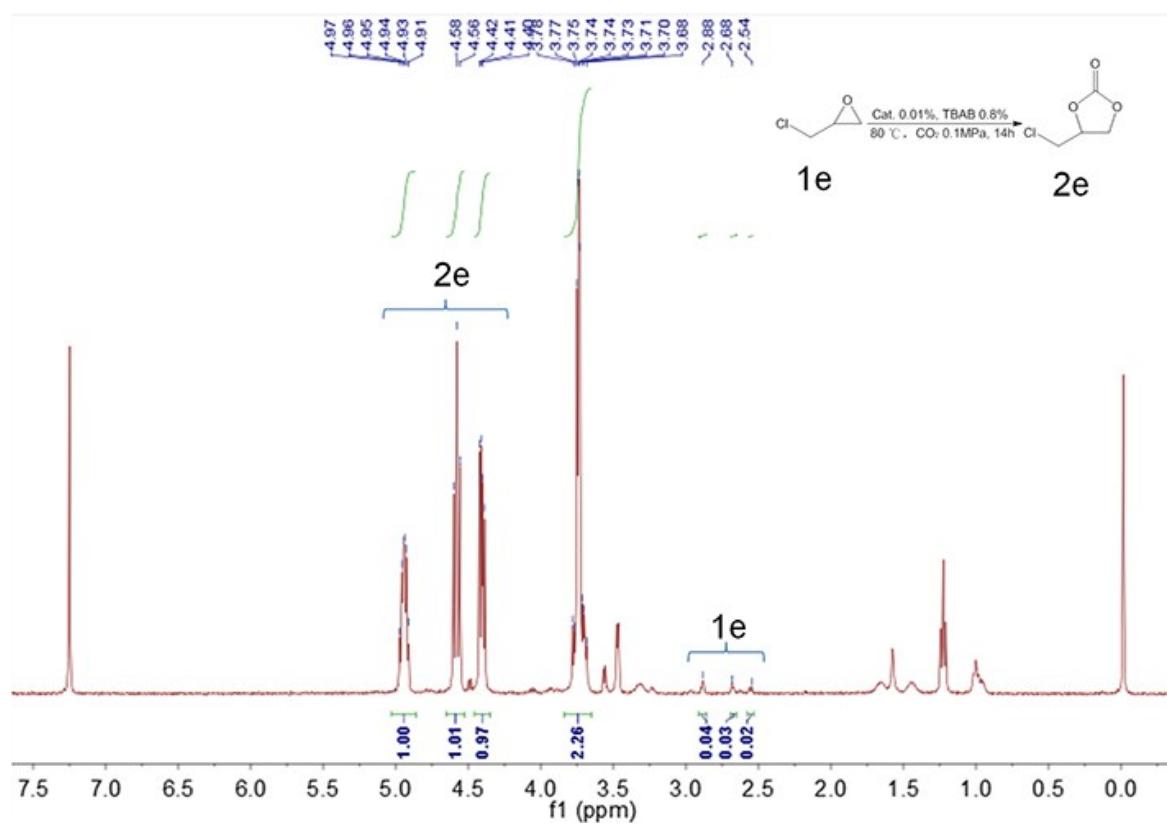
**Figure 10.** FT-IR spectra of complex **1b** and **2a** before and after the cycloaddition of epoxides with  $\text{CO}_2$ . The FT-IR peaks are almost no changed, indicates that complexes didn't decomposed during the catalyzed reaction.

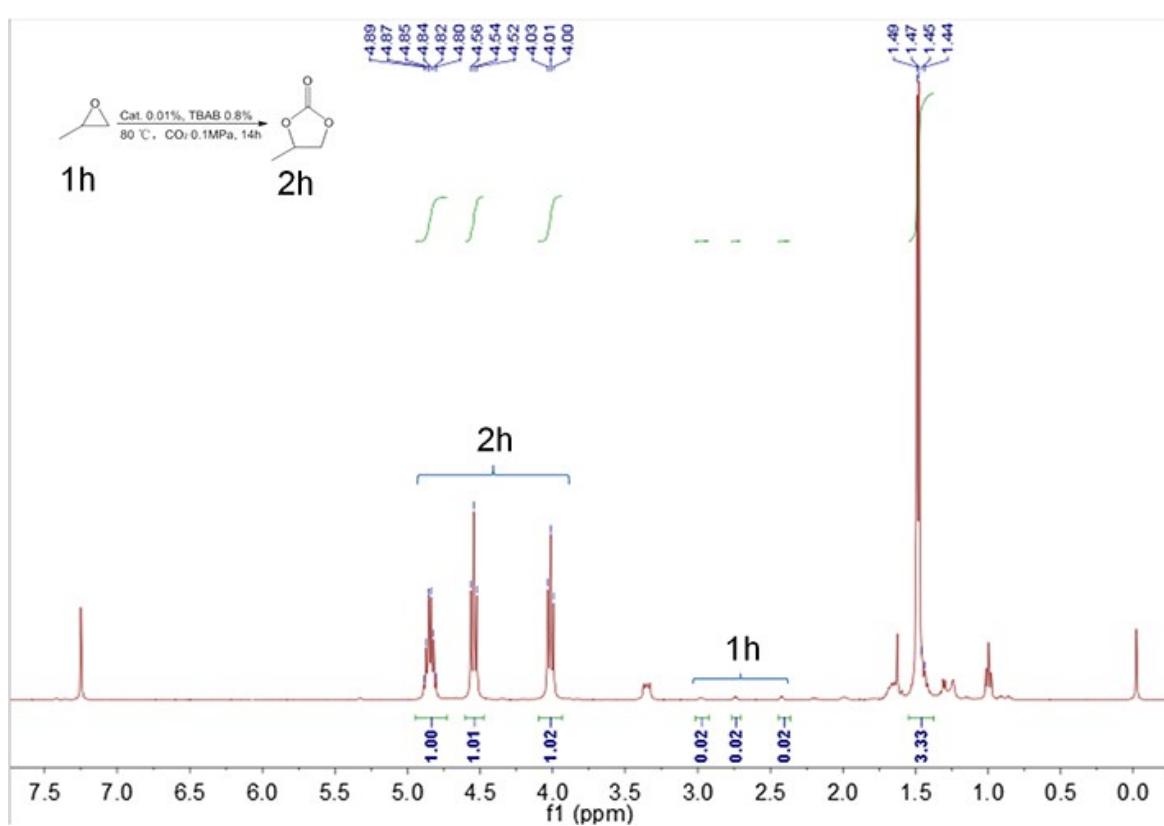
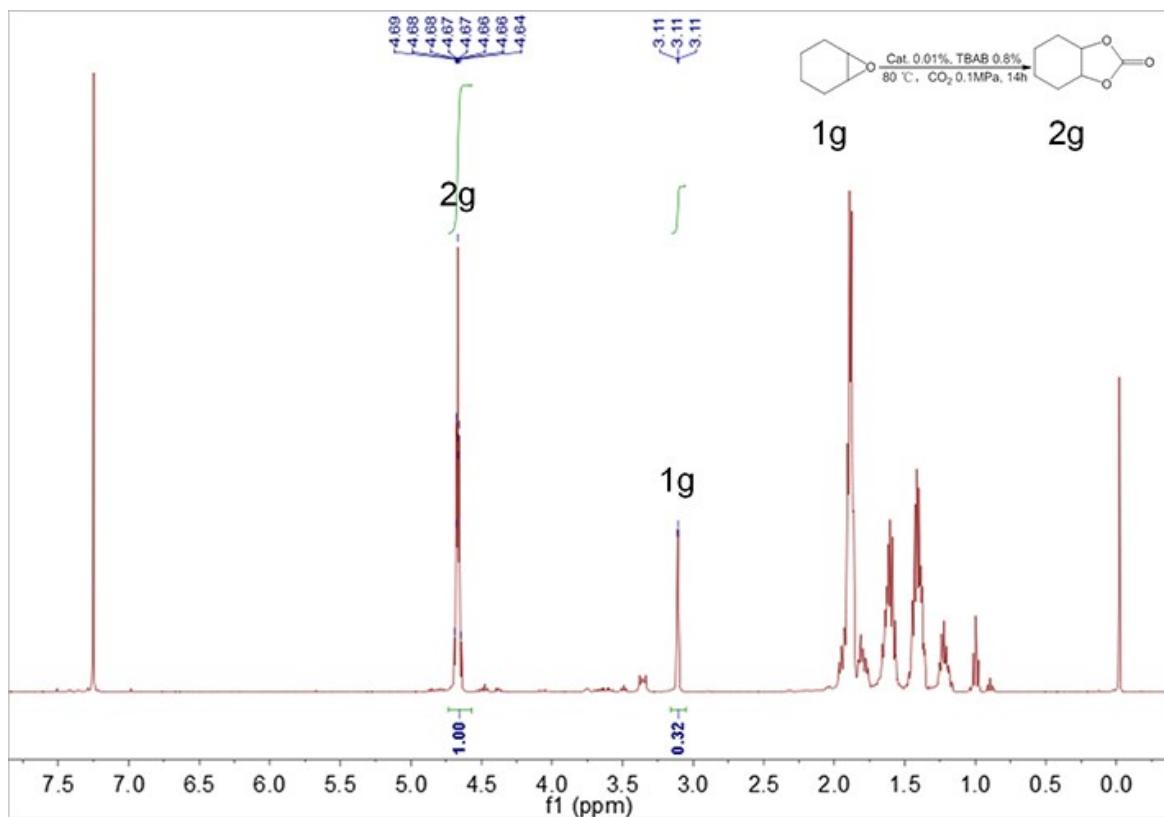


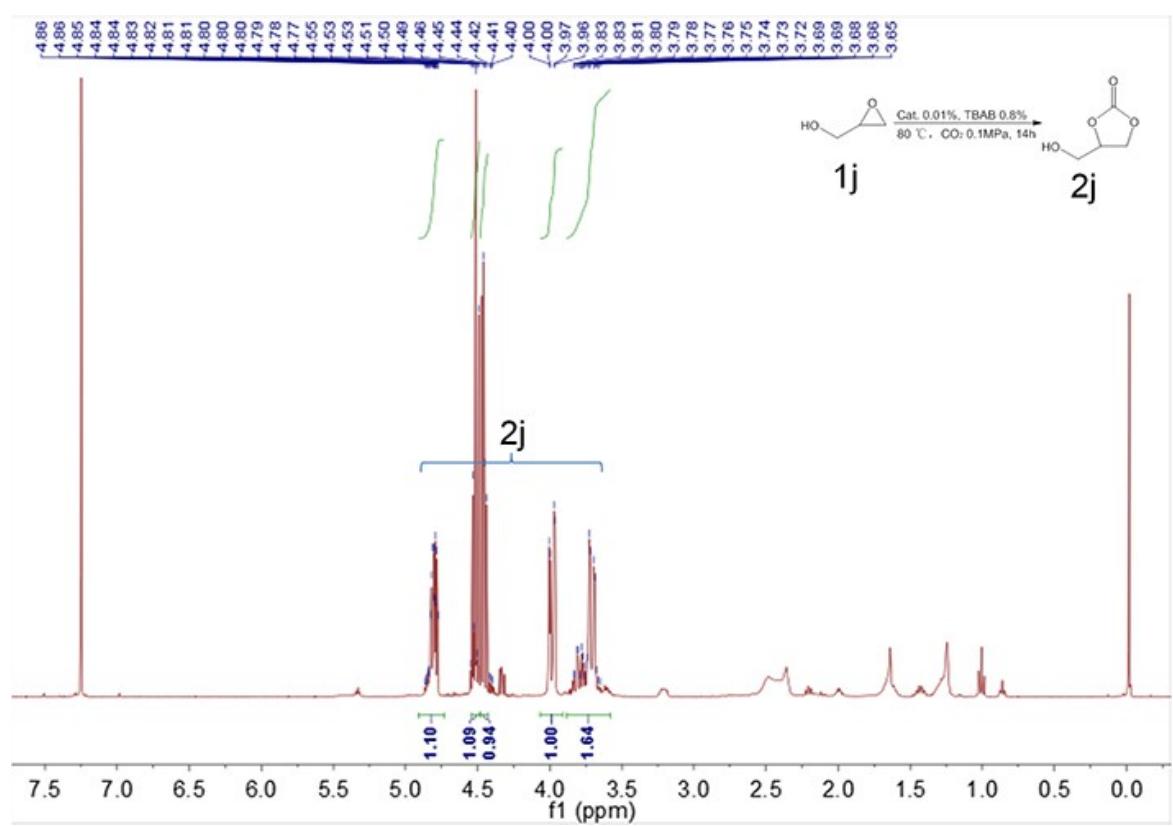
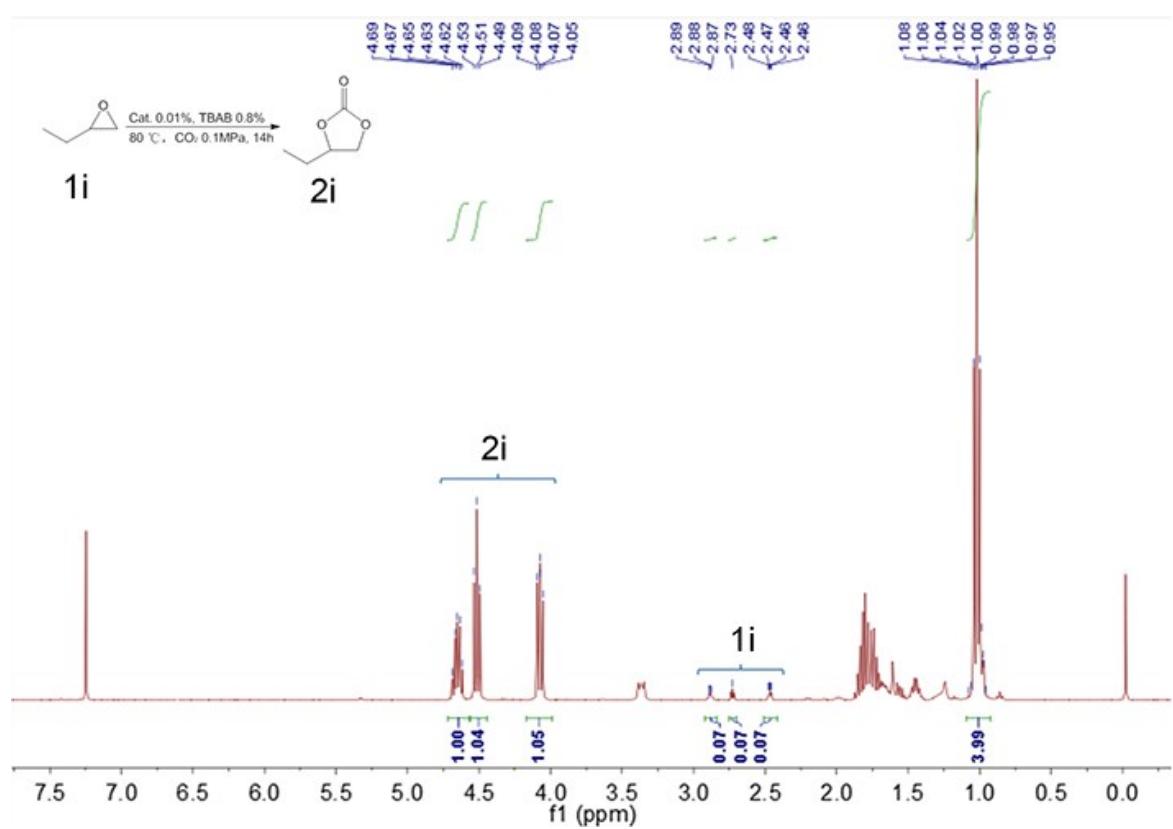
**Figure S11.** A possible mechanism for the cycloaddition reaction.

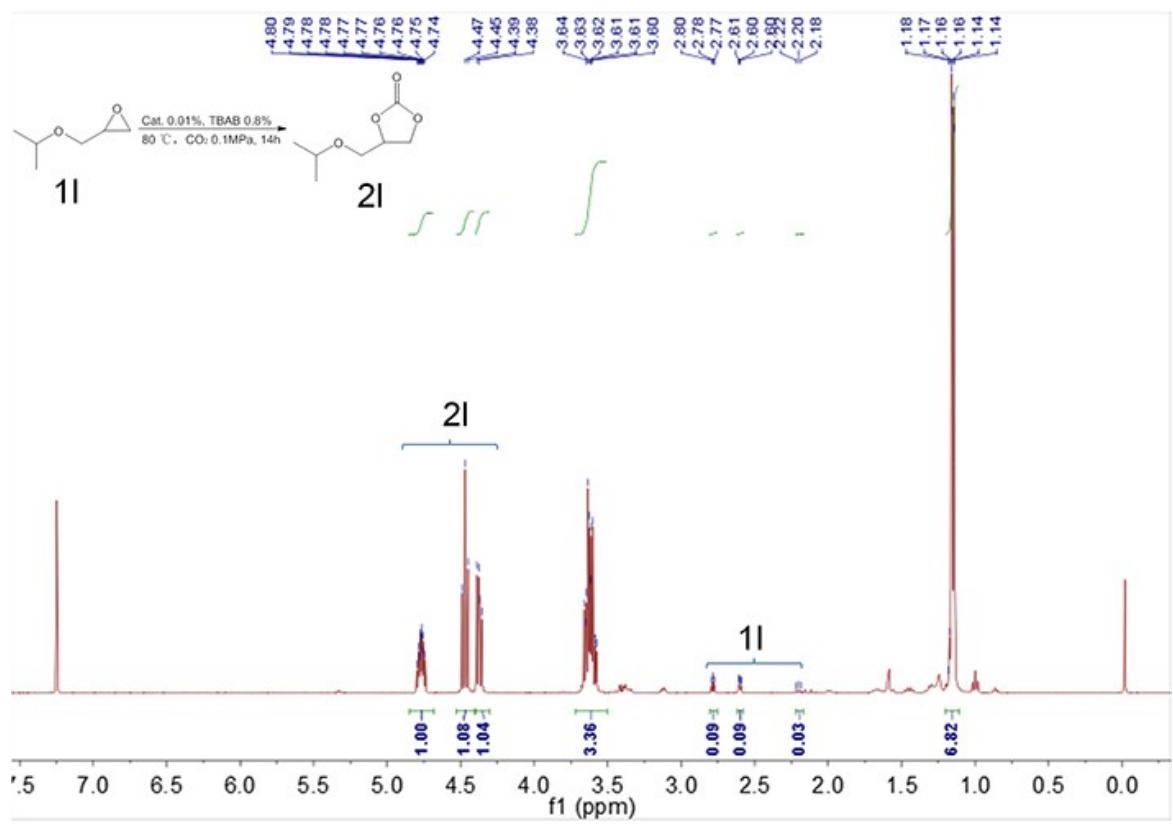
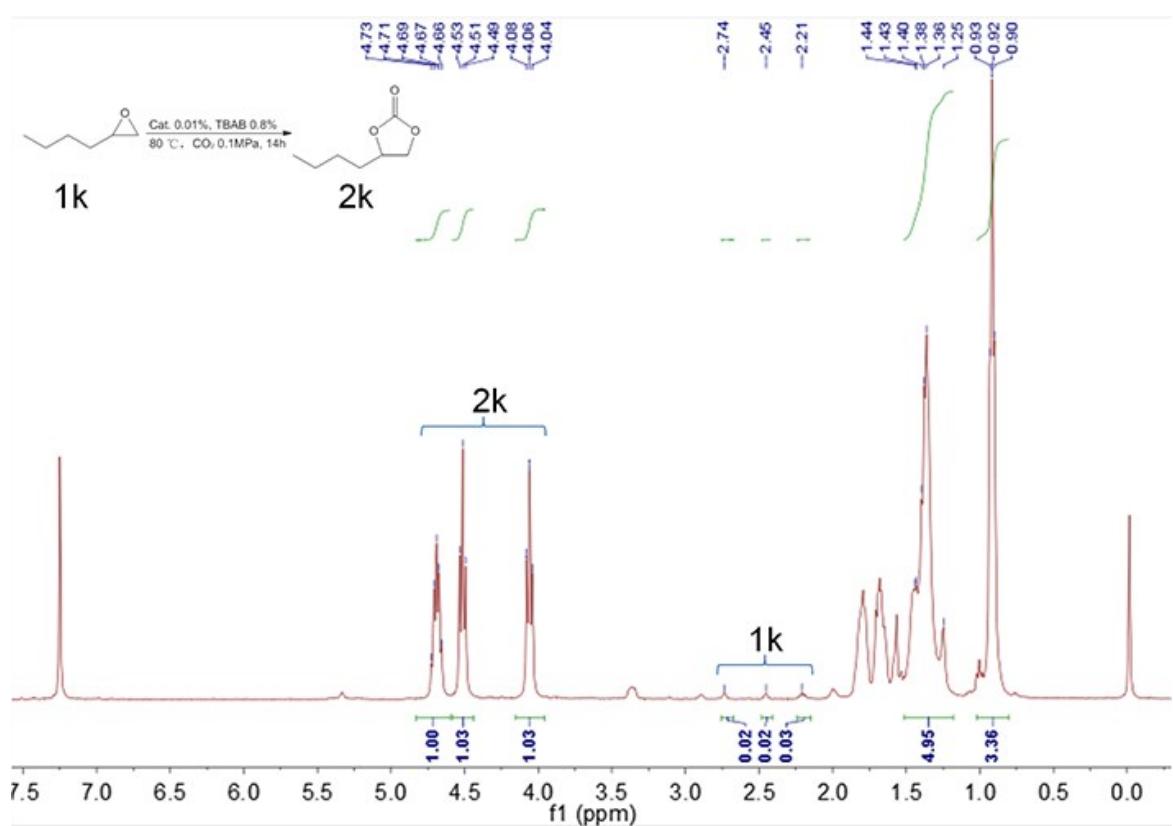












## References:

- (1) Qin, J.; Wang, P.; Li, Q.; Zhang, Y.; Yuan, D.; Yao, Y. *Chem. Commun.*, 2014, 50, 10952.
- (2) J. Dong, H. Xu, S. L. Hou, Z. L. Wu and B. Zhao, *Inorg. Chem.*, 2017, 56, 6244.
- (3) H. Xu, B. Zhai, C. S. Cao and B. Zhao, *Inorg. Chem.*, 2016, 55, 9671.
- (4) J. Dong, P. Cui, P.-F. Shi, P. Cheng and B. Zhao, *J. Am. Chem. Soc.*, 2015, 137, 15988.
- (5) W. Z. Qiao, H. Xu, P. Cheng and B. Zhao, *Cryst. Growth Des.*, 2017, 17, 3128.
- (6) Q. Han, L. Wang, Z. Shi, C. Xu, Z. Dong, Z. Mou and W. Liu, *Chem. Asian. J.*, 2017, **12**, 1364.
- (7) L. Wang, C. Xu, Q. Han, X. Tang, P. Zhou, R. Zhang, G. Gao, B. Xu, W. Qin and W. Liu, *Chem. Commun.*, 2018, **54**, 2212.
- (8) A. Otero, J. Martinez, J. Fernandez-Baeza, L. F. Sanchez-Barba, J. A. Castro-Osma and A. Lara-Sanchez, *ChemSusChem.*, 2017, **10**, 2886.
- (9) N. Wei, Y. Zhang, L. Liu, Z. B. Han and D. Q. Yuan, *Appl. Catal., B.*, 2017, **219**, 603.
- (10) Buchard, A.; Kember, M. R.; Sandeman, K. G.; Williams, C. K. *Chem. Commun.*, 2011, 47, 212.
- (11) Yang, Y.; Hayashi, Y.; Fujii, Y.; Nagano, T.; Kita, Y.; Ohshima, T.; Mashima, K. *Catal. Sci. Technol.*, 2012, 2, 509.
- (12) Meléndez, J.; North, M.; Pasquale, R. *Eur. J. inorg. chem.*, 2007, 2007, 3323.
- (13) Luo, R.; Zhou, X.; Chen, S.; Li, Y.; Zhou, L.; Ji, H. *Green Chem.*, 2014, 16, 1496.
- (14) Martínez - Rodríguez, L., Otalora Garmilla, J., and Kleij, A. W. *ChemSusChem*, 2016, 9, 749.
- (15) Rintjema, J. and Kleij, A. W. *ChemSusChem*, 2017, 10, 1274.
- (16) Chen, Y., Luo, R., Xu, Q., Jiang, J., Zhou, X. and Ji, H. *ChemSusChem*, 2017, 10, 2534.
- (17) Chen, Y., Luo, R., Xu, Q., Zhang, W., Zhou, X. and Ji, H. *ChemCatChem*, 2017, 9, 767.

- (18) Castro-Osma, J. A., Lara-Sánchez, A., North, M., Otero, A. and Villuendas, P. *Catal. Sci. Technol.*, 2012, 2, 1021.
- (19) Ema, T., Miyazaki, Y., Koyama, S., Yano, Y. and Sakai, T. *Chem. Commun.*, 2012, 48, 4489.
- (20) T. T Wang, Y. Xie and W. Q Deng, *J. Phys. Chem. A*, 2014, 118, 9239.
- (21) M. V. Escárcega-Bobadilla, M. Martínez Belmonte, E. Martín, E. C. Escudero-Adán and A. W. Kleij, *Chem. Eur. J.*, 2013, 19, 2641.
- (22) W. Y Gao, L. Wojtas, S. Ma, *Chem. Commun.*, 2014, 50, 5316.
- (23) W. Y Gao, Y. Chen, Y. Niu, K. Williams, L. Cash, P. J. Perez, and S. Ma, *Angew. Chem. Int. Ed. Engl.*, 2014, 53, 2615.
- (24) Z. R. Jiang, H. Wang, Y. Hu, J. Lu and H. L. Jiang, *ChemSusChem.*, 2015, 8, 878.