

## Supporting Information for

### Microporous 2D indium metal-organic frameworks for selective CO<sub>2</sub> capture and their application in the catalytic CO<sub>2</sub>-cycloaddition of epoxides

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**Table S1.** Crystallographic Data.

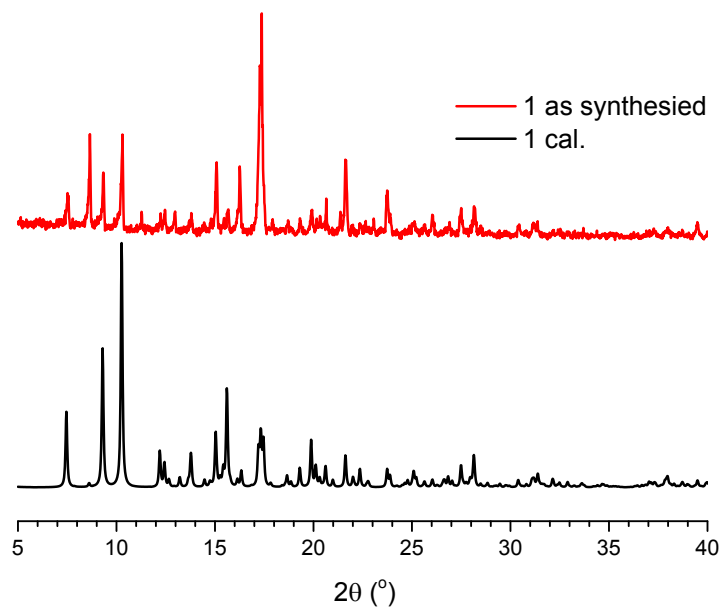
	<b>1</b>	<b>2</b>	<b>3</b>	<b>4</b>
formula	C <sub>28</sub> H <sub>16</sub> InO <sub>12</sub> S	C <sub>105</sub> H <sub>108</sub> In <sub>4</sub> O <sub>44</sub> N <sub>4</sub> S	C <sub>11</sub> H <sub>6.50</sub> In <sub>0.50</sub> N <sub>0.50</sub> O <sub>5</sub>	C <sub>70</sub> H <sub>40</sub> Cl <sub>2</sub> In <sub>3</sub> O <sub>2</sub>
	2	4	S <sub>0.50</sub>	0
fw	723.37	362.36	299.1	1614.38
space group	<i>P</i> -1	<i>Ccca</i>	<i>Ccca</i>	<i>P2<sub>1</sub>2<sub>1</sub>2</i>
<i>a</i> , Å	7.2978(4)	14.5653(3)	15.0622(4)	29.6876(5)
<i>b</i> , Å	12.1206(7)	21.0484(5)	20.9202(5)	33.7481(5)
<i>c</i> , Å	20.9806(11)	23.9140(9)	23.4817(7)	9.4487(5)
<i>α</i> , °	78.626(3)	90	90	90
<i>β</i> , °	85.181(3)	90	90	90
<i>γ</i> , °	84.564(3)	90	90	90
volume, Å <sup>3</sup>	1807.12(17)	7331.5(3)	7399.2(4)	9466.7(2)
<i>Z</i>	2	2	16	4
<i>D</i> <sub>cacl</sub> , g cm <sup>-3</sup>	1.329	1.060	1.074	1.134
<i>μ</i> , mm <sup>-1</sup>	0.821	0.736	0.732	6.755
<i>T</i> , °C	296(2)	295(2)	296(2)	296(2)
<i>λ</i> , Å	0.71073	0.71073	0.71073	1.54178
reflections collected	27446	31859	32917	63217
independent reflections	8698	4517	4598	16415
<i>R</i> (int)	0.0896	0.1087	0.0913	0.0414
Goodness-of-fit on <i>F</i> <sup>2</sup>	0.902	0.807	0.886	1.086
<i>R</i> <sub>1</sub> [ <i>I</i> >2σ( <i>I</i> )]	0.0280	0.0876	0.0501	0.0656
<i>wR</i> <sub>2</sub> [ <i>I</i> >2σ( <i>I</i> )]	0.0654	0.2662	0.1407	0.2044
<i>R</i> <sub>1</sub> [all data]	0.0353	0.1408	0.1121	0.0737
<i>wR</i> <sub>2</sub> [all data]	0.0679	0.2852	0.1449	0.2162
CCDC No.	1823435	1823434	1823437	1823436

**Table S2.** Selected bond lengths (Å).

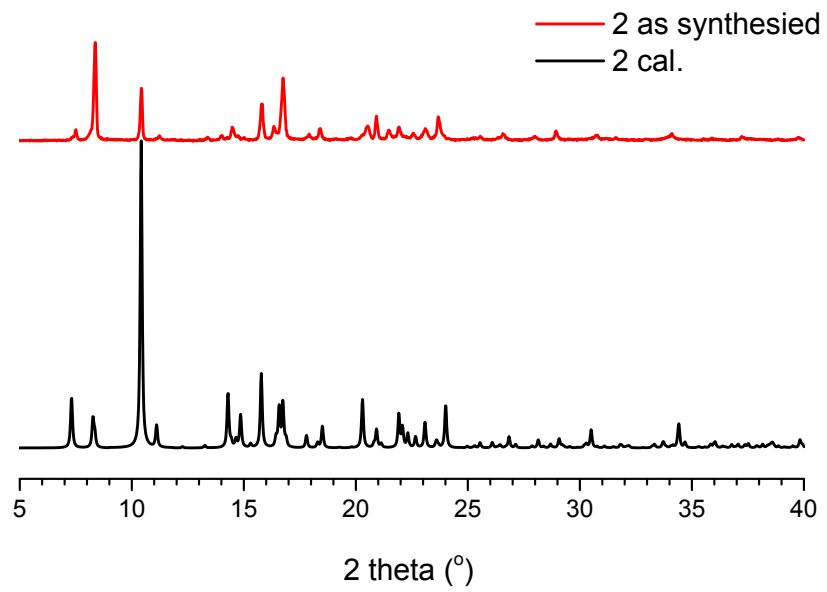
<b>1</b>			
In(1)-O(1)	2.351(3)	In(1)-O(2)	2.169(3)
In(1)-O(3)	2.207(3)	In(1)-O(4)	2.422(3)
In(1)-O(5)	2.286(3)	In(1)-O(6)	2.257(3)
In(1)-O(7)#1	2.415(4)	In(1)-O(8)#1	2.222(4)
<b>2</b>			
In(1)-O(1)	2.305(3)	In(1)-O(1)#1	2.305(3)
In(1)-O(2)	2.257(3)	In(1)-O(2)#1	2.257(3)
In(1)-O(4)	2.300(3)	In(1)-O(4)#1	2.300(3)
In(1)-O(5)	2.237(3)	In(1)-O(5)#1	2.236(3)
<b>3</b>			
In(1)-O(1)	2.306(6)	In(1)-O(4)	2.312(5)
In(1)-O(1)#1	2.306(6)	In(1)-O(4)#1	2.312(5)
In(1)-O(2)	2.270(6)	In(1)-O(5)	2.235(5)
In(1)-O(2)#1	2.270(5)	In(1)-O(5)#1	2.235(5)
<b>4</b>			
In(1)-O(15)	2.154(9)	In(1)-O(17)	2.230(7)
In(1)-O(19)	2.239(9)	In(1)-O(4)	2.246(8)
In(1)-O(3)	2.247(8)	In(1)-O(18)	2.318(7)
In(1)-O(20)	2.358(11)	In(1)-O(16)	2.539(12)
In(2)-O(8)	2.147(8)	In(2)-O(12)	2.155(7)
In(2)-O(9)	2.168(6)	In(2)-O(10)	2.322(8)
In(2)-O(11)	2.430(9)	In(2)-O(7)	2.564(9)
In(2)-Cl(1)	2.385(3)	In(3)-O(5)	2.162(8)
In(3)-O(14)	2.169(7)	In(3)-O(1)	2.234(7)
In(3)-O(2)	2.244(6)	In(3)-O(13)	2.457(9)
In(3)-O(6)	2.491(9)	In(3)-Cl(2)	2.386(3)

Symmetry transformations used to generate equivalent atoms: for **2**, #1  $-x+1/2, -y, z$ , #2  $-x, y, -z+1/2$ ; for **3**, #1  $-x+1/2, -y, z$  #2  $-x+0, -y-1/2, z$  #3  $-x+1, -y, -z+1$ ; for **4**, #1  $x+1/2, y-1/2, -z$ , #2  $x-1/2, y+1/2, -z$ , #3  $-x, -y, -z$ .

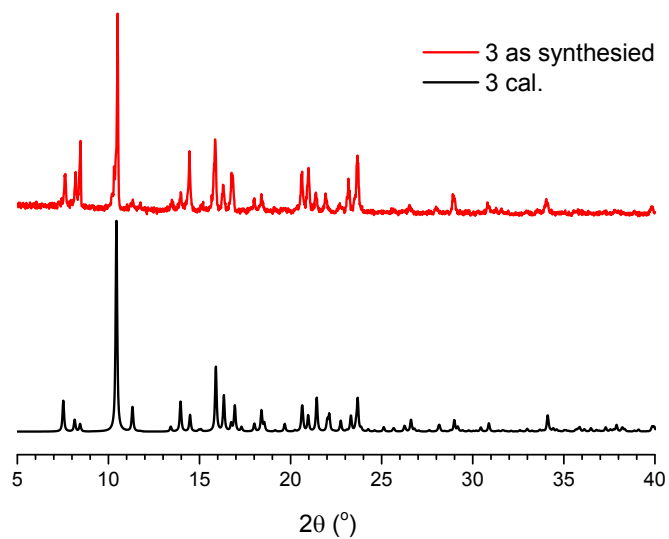
(a)



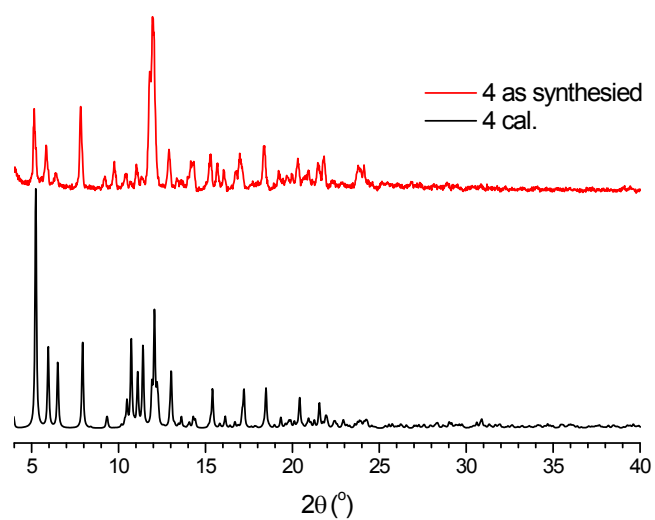
(b)



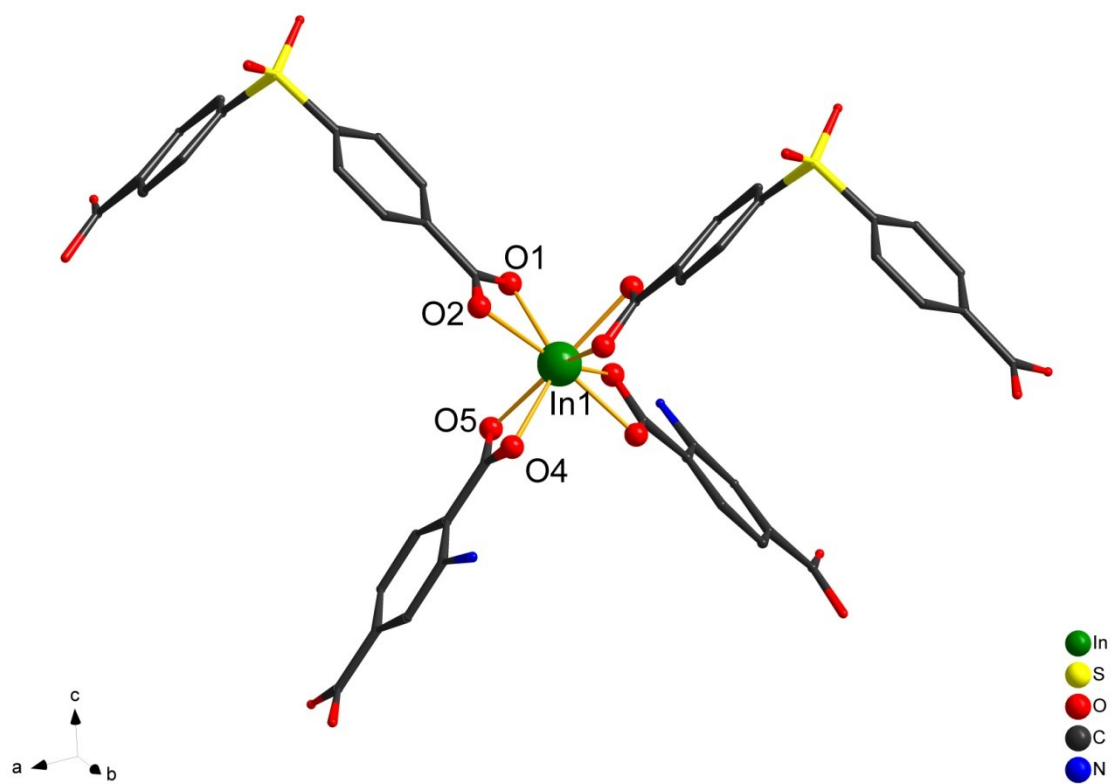
(c)



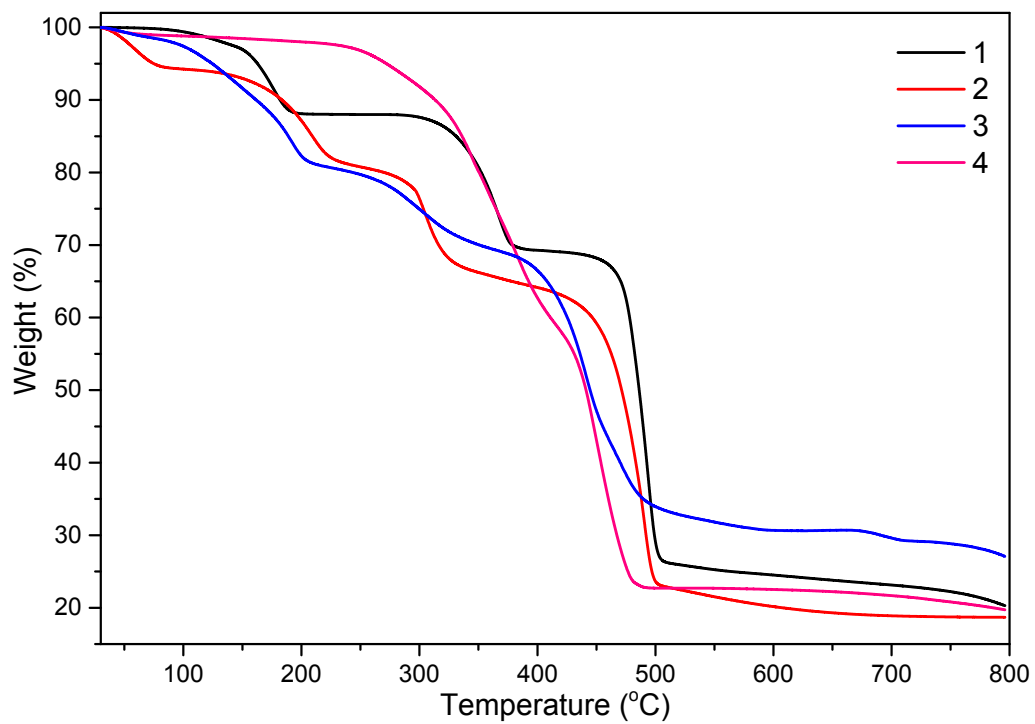
(d)



**Figure S1.** The PXRD patterns of compounds **1** (a), **2** (b), **3** (c), **4** (d).

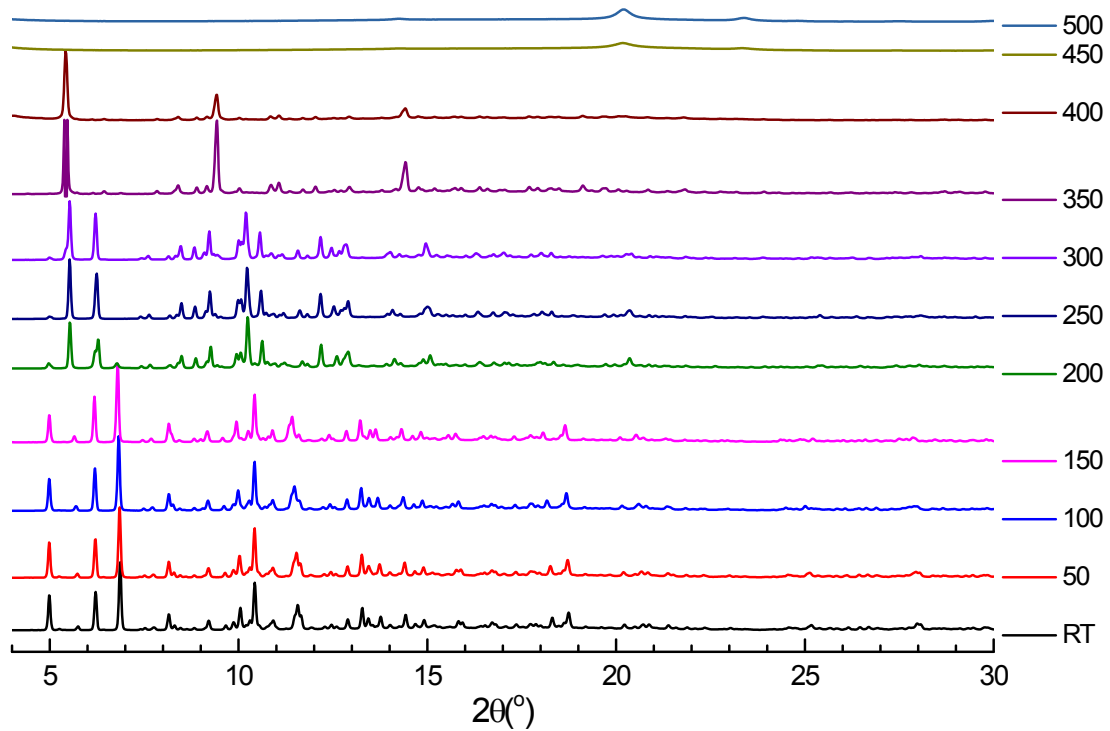


**Figure S2.** (a) The coordination environment around Incenters in **3**

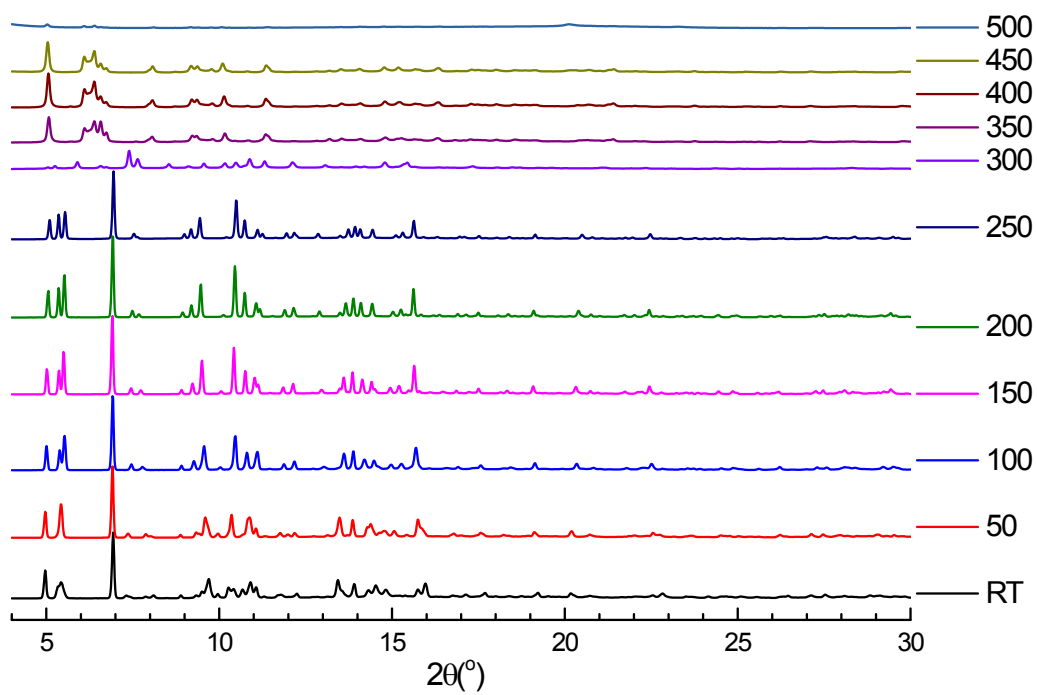


**Figure S3.** TGA curves of compound **1** (black), **2** (red), **3** (blue), and **4** (pink).

(a)

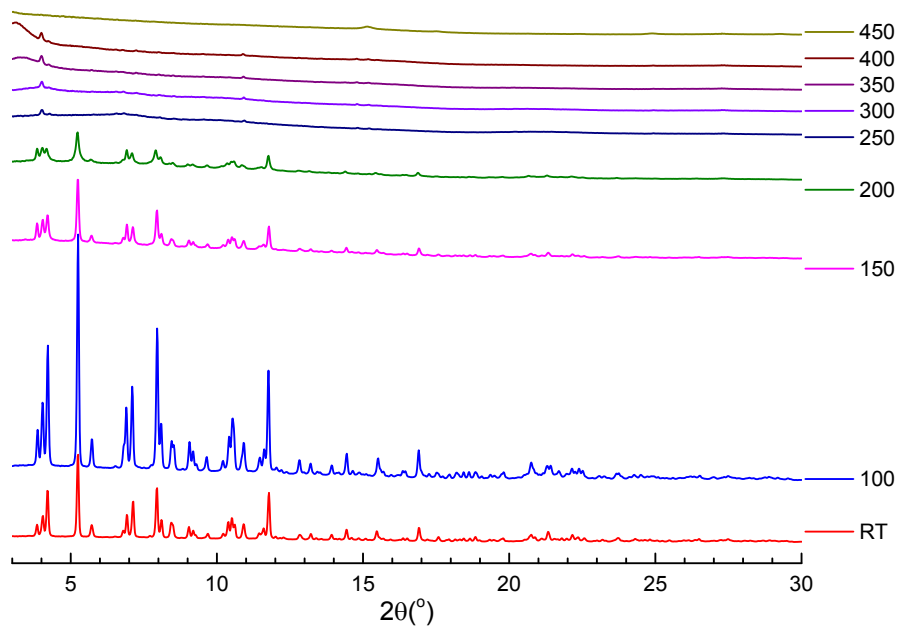


(b)

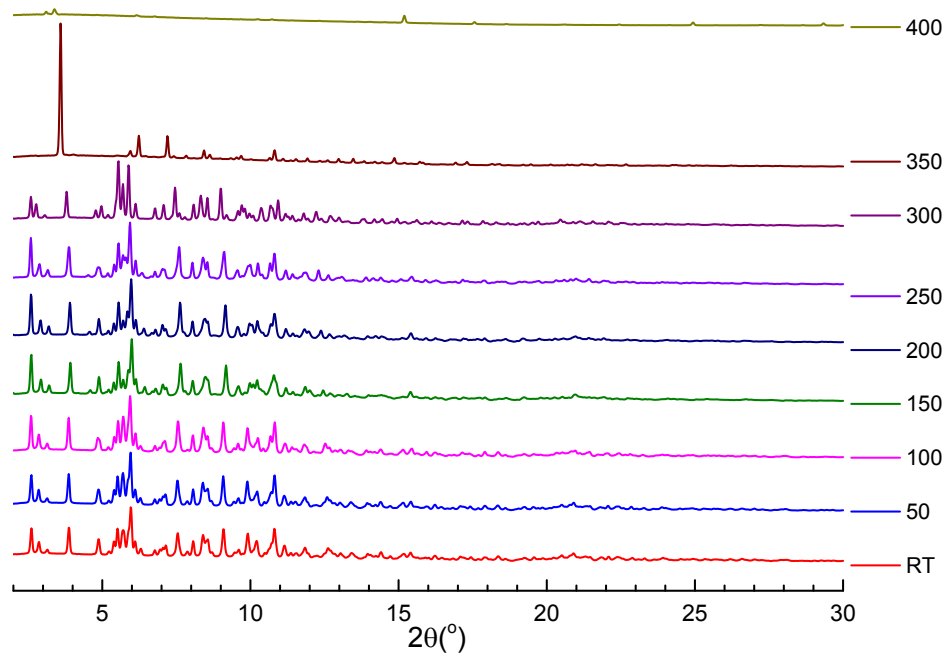




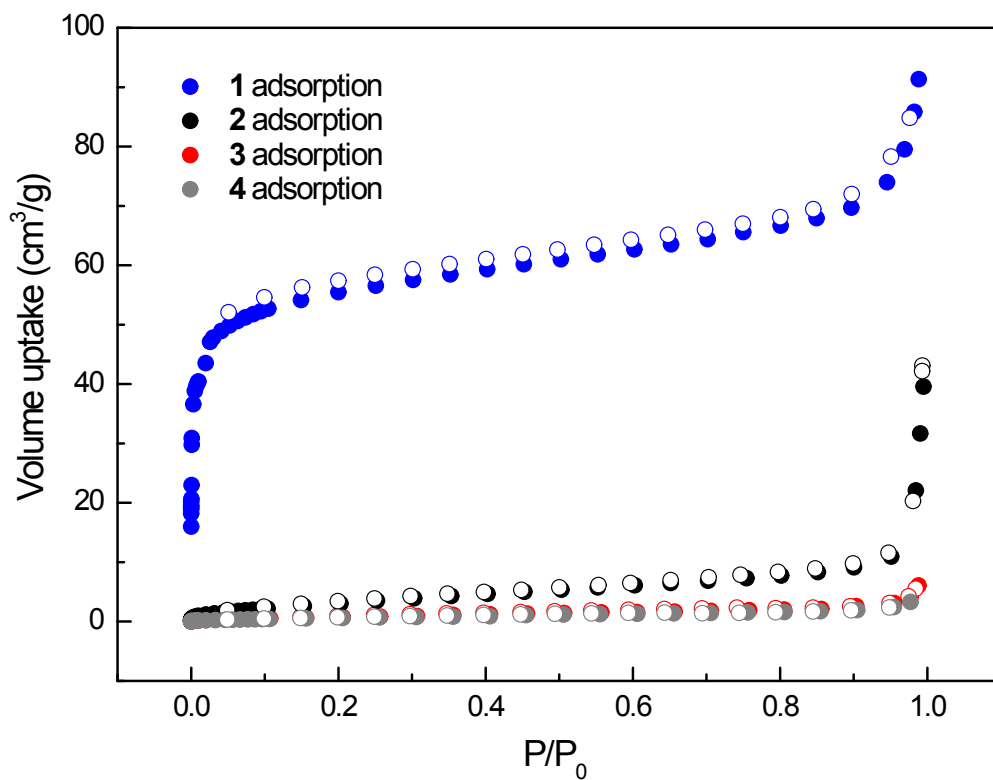
(c)



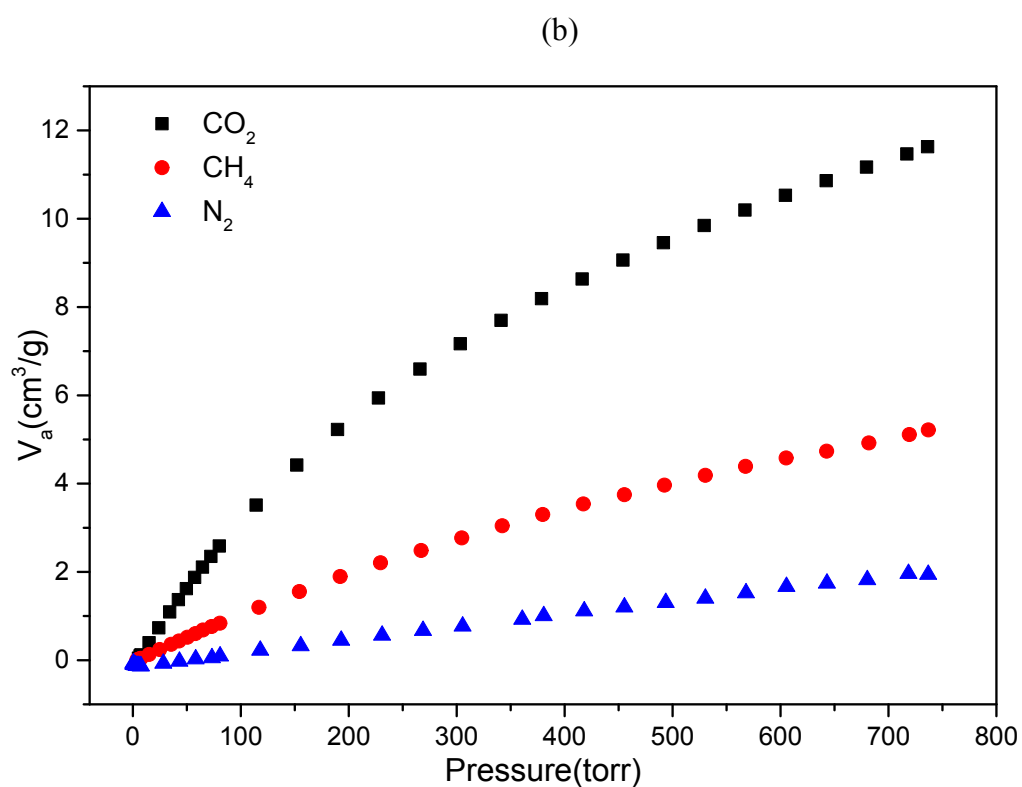
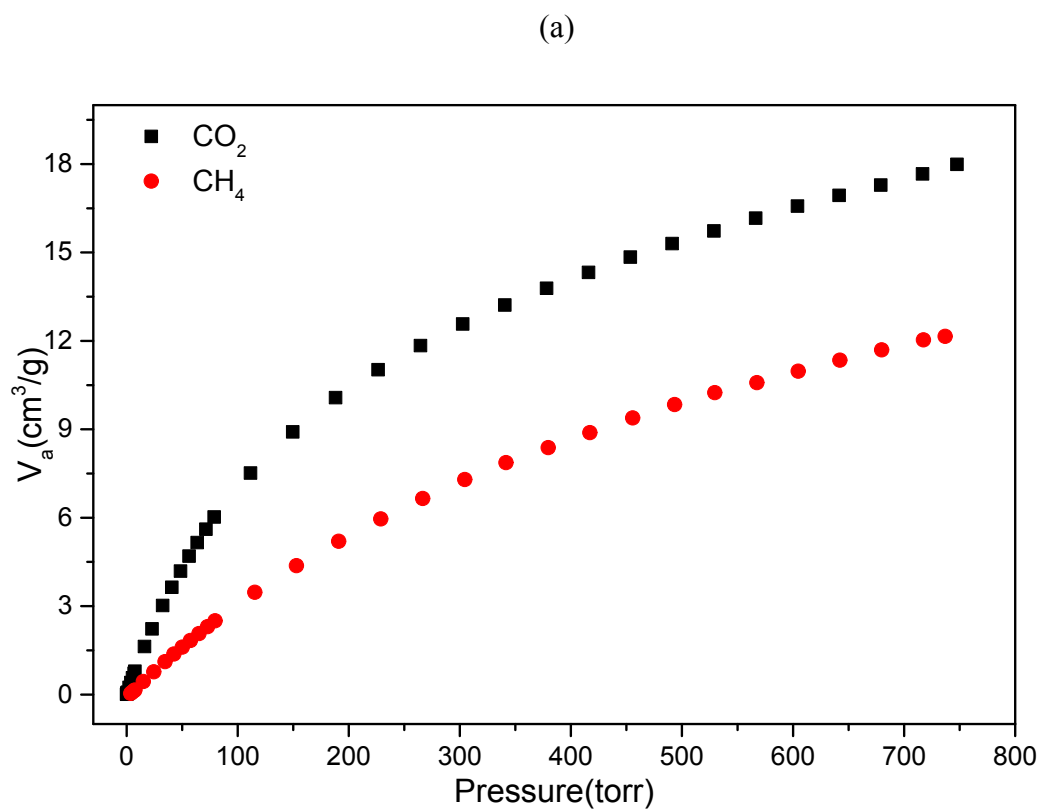
(d)



**Figure S4.** The varied temperature PXRD patterns of **1** (a), **2**(b), **3** (c), and **4** (d).

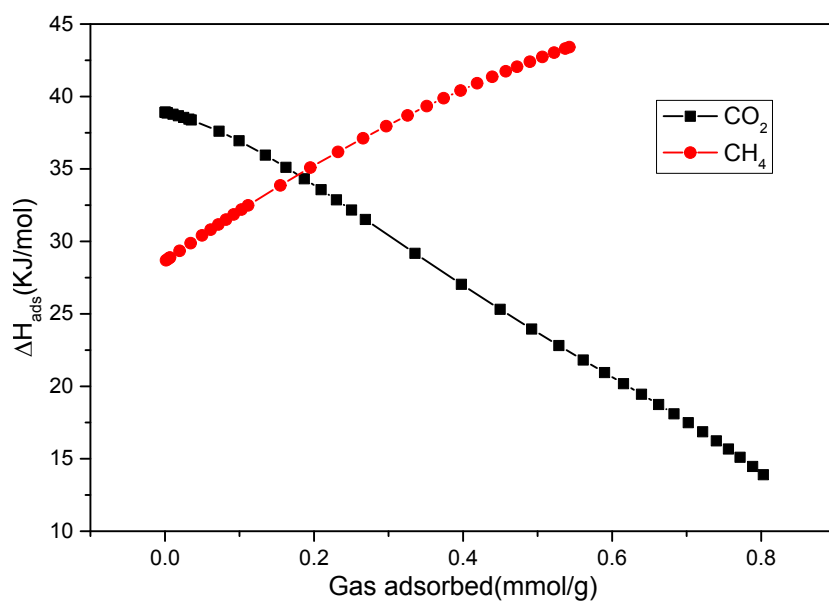


**Figure S5.** The N<sub>2</sub> adsorption-desorption isotherms of compound **1** (blue), **2** (black), **3** (red), and **4** (Gray) at 77 K (the closed and open symbol represents the adsorption and desorption respectively).

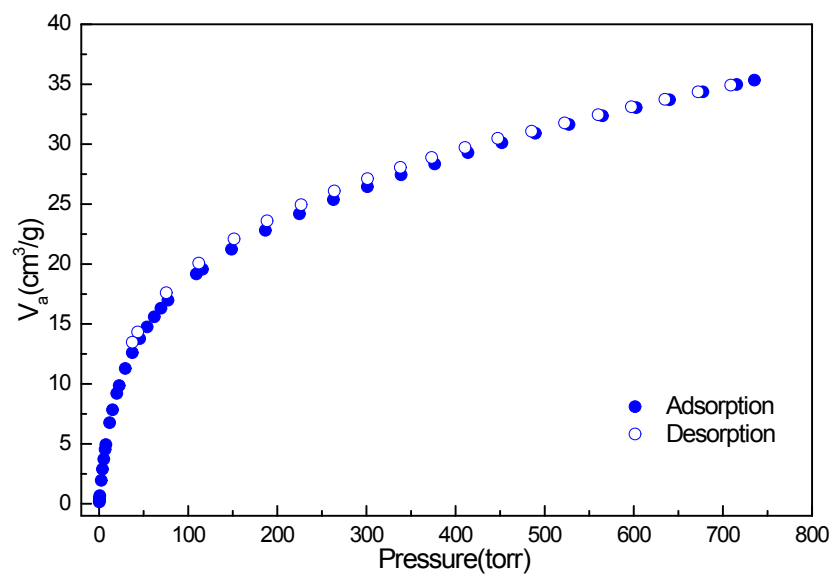


**Figure S6.** (a) The CO<sub>2</sub> and CH<sub>4</sub> gas adsorption isotherms of **1** recorded at 273 K. (b) The CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub> gas adsorption isotherms of **1** recorded at 298 K.

(a)

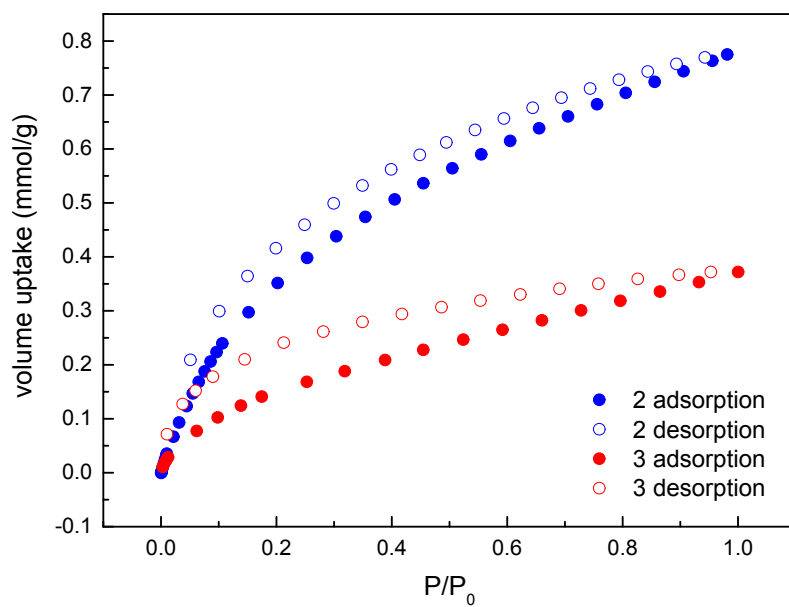


(b)

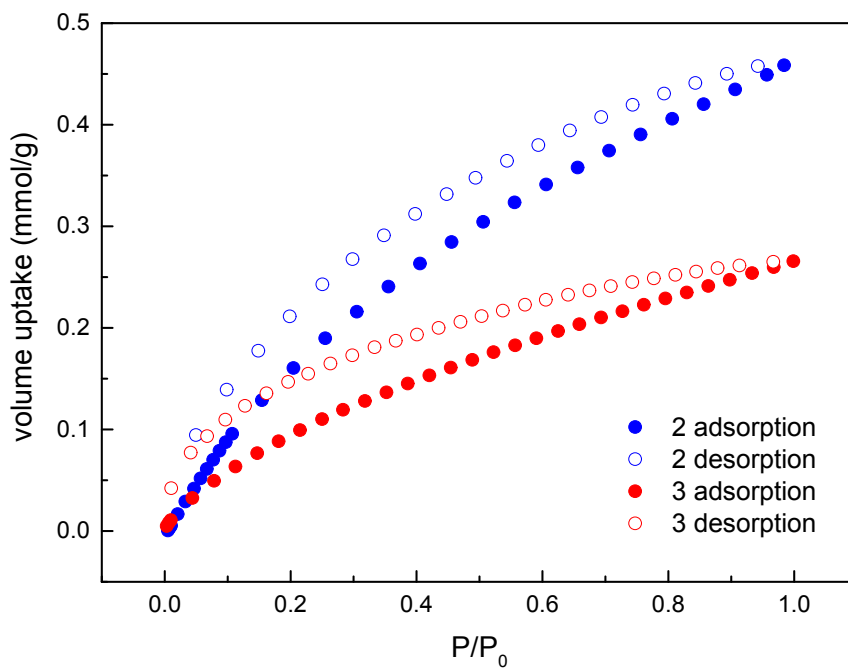


**Figure S7.** (a) The  $Q_{\text{st}}$  of compound **1** for  $\text{CO}_2$  and  $\text{CH}_4$ . (b)  $\text{H}_2$  adsorption-desorption isotherms of compound **1** at 77 K.

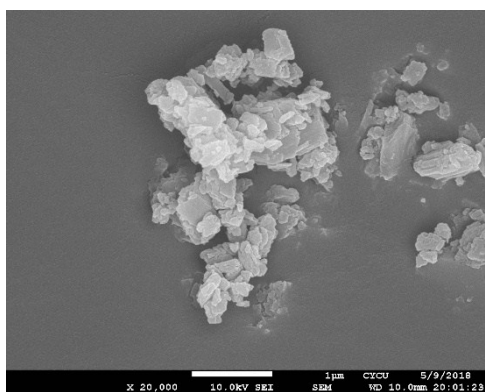
(a)



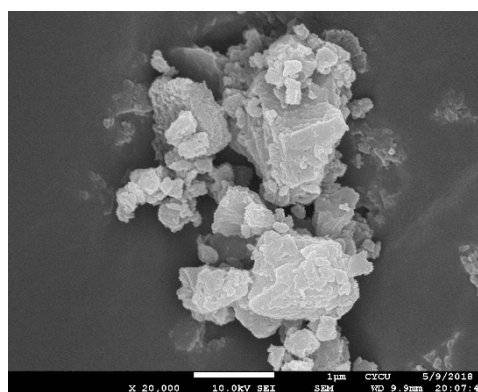
(b)



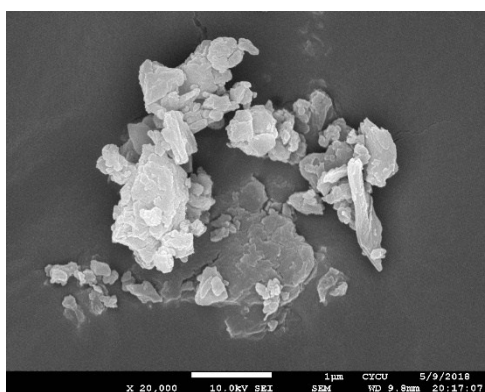
**Figure S8.** The CO<sub>2</sub> isotherm at **2** and **3** at (a) 273 K (b) 298 K.



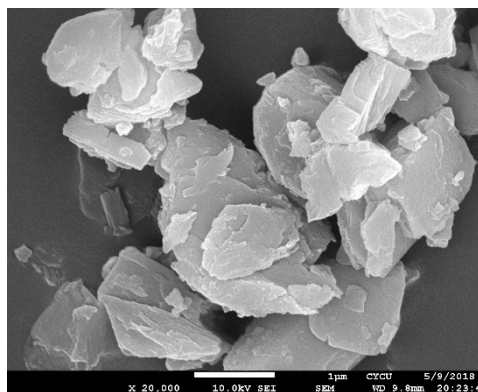
(a)



(b)

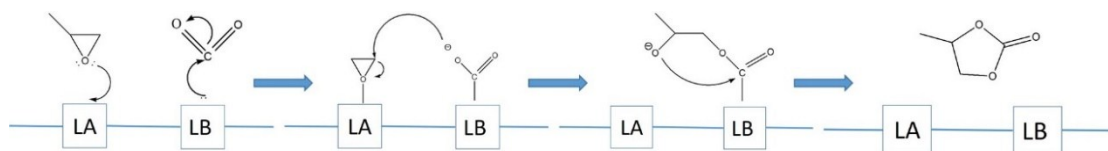


(b)



(d)

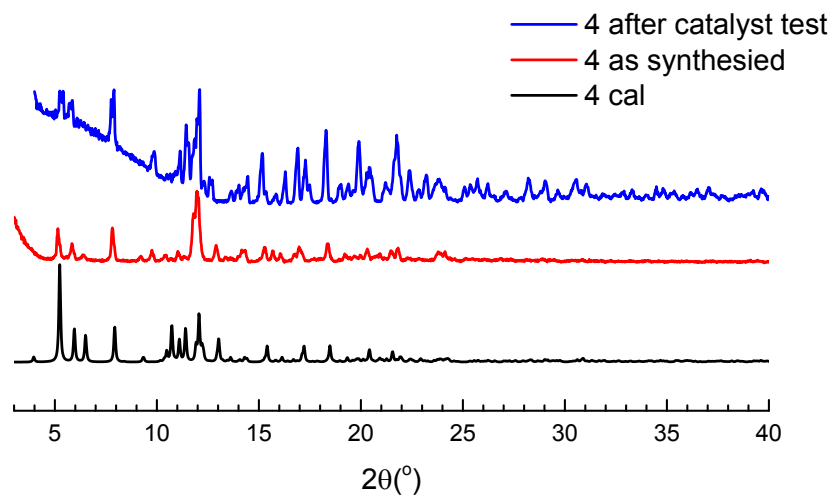
**Figure S9.** The SEM images for compound (a) **1**, (b) **2**, (c) **3**, and (d) **4**.



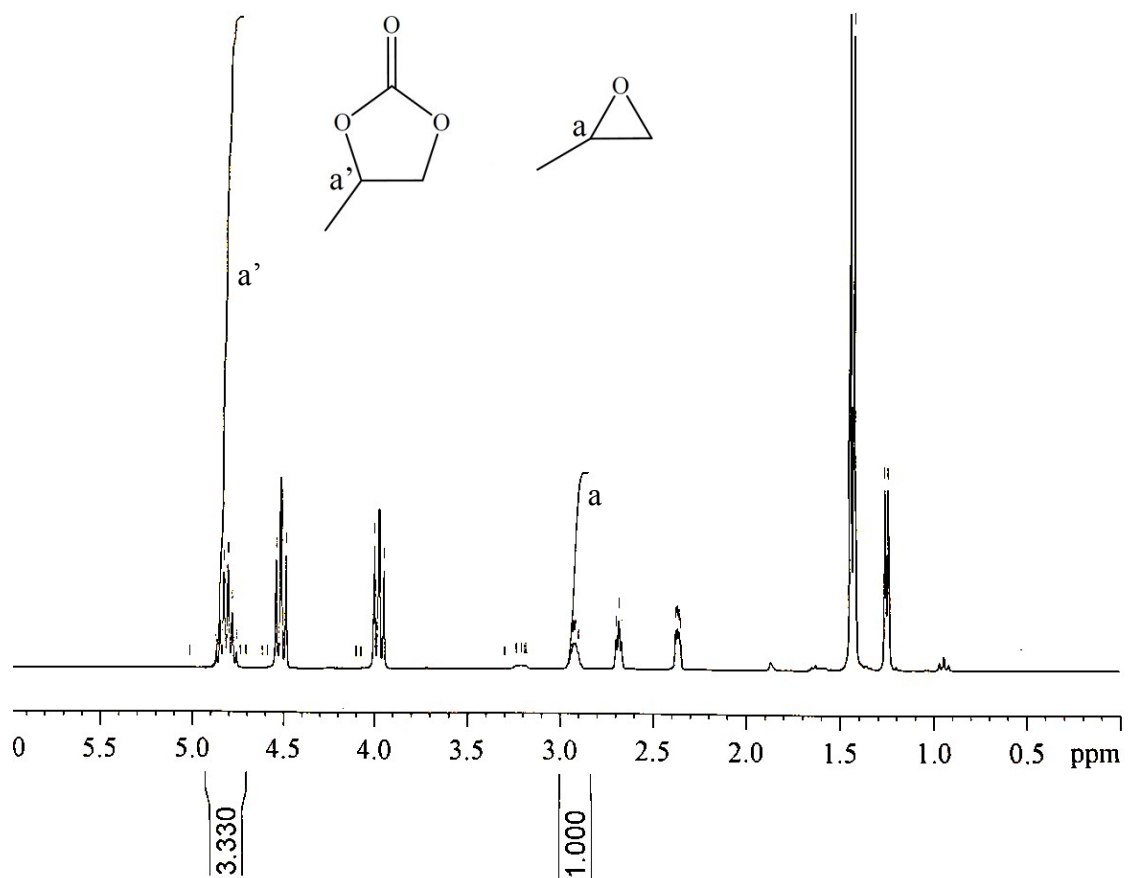
**Figure S10.** The proposed reaction mechanism for compound **3**.

**Table S3.** Coupling of propylene oxide (PO) and CO<sub>2</sub> to propylene carbonate (PC) catalyzed by various MOF/n-Bu<sub>4</sub>NBr catalytic systems.

Catalyst	Catalyst(g)	Co-catalyst (mol%)	Temperature(°C)	Pressure(MPa)	Reaction Times <i>T</i> (h)	Conversion (%)	ref.
MOF-5	2.5	2.5	60	6	4	98	37
In+Dpa+1,10 phen	0.35	0.35	60	1.2	6	92	38
In+Dpa+1,10 phen	0.35	0.35	50	1.2	6	66	38
HKUST-1	0.2	10	25	0.1	48	49	39
MOF1	0.2	10	25	0.1	48	96	39
Cr-MIL-101	1.2	0.62	25	0.8	34	82	40
CHB	1.6	1.6	120	1.2	6	62	41
PCN-224	-	-	100	2	4	42	42
Hf-Nu-1000	4	1/10	25	0.1	56	100	43
MMCF-2	0.13	7.2	25	0.1	48	95.4	44
MMPF-9	0.13	7.2	25	0.1	48	87.4	45
Zn-Glu	0.47	0.94	80	1.2	6	99	46
Ni-saldpen-MOF	0.7	2	80	2	4	86	47
MIL-47	0.1	2.5	50	2	24	95	48
MOF-205	0.6	0.6	25	1.2	24	89	49
<b>1</b>	0.1	1	80	2	24	85	+
<b>2</b>	0.1	1	80	2	24	89	+
<b>3</b>	0.1	1	80	2	24	92	+
<b>4</b>	0.1	1	80	2	24	98	+



**Figure S11.** The PXRD after CO<sub>2</sub> cycloaddition of compound 4.

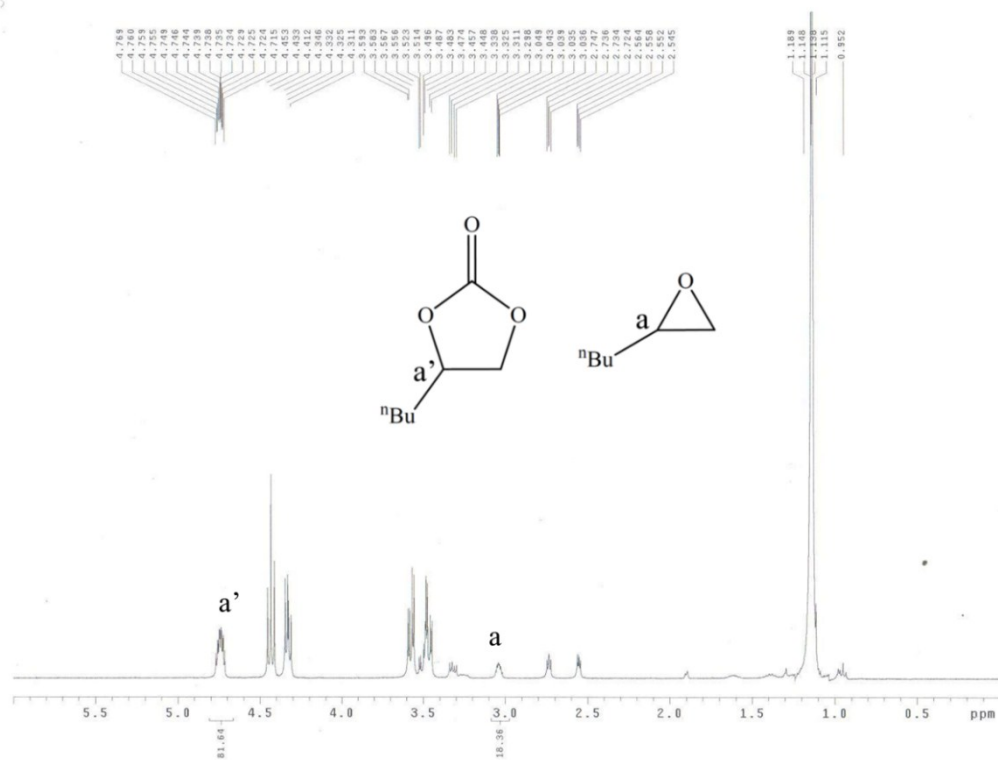


**Figure S12.** The <sup>1</sup>H NMR spectrum for PO to PC.

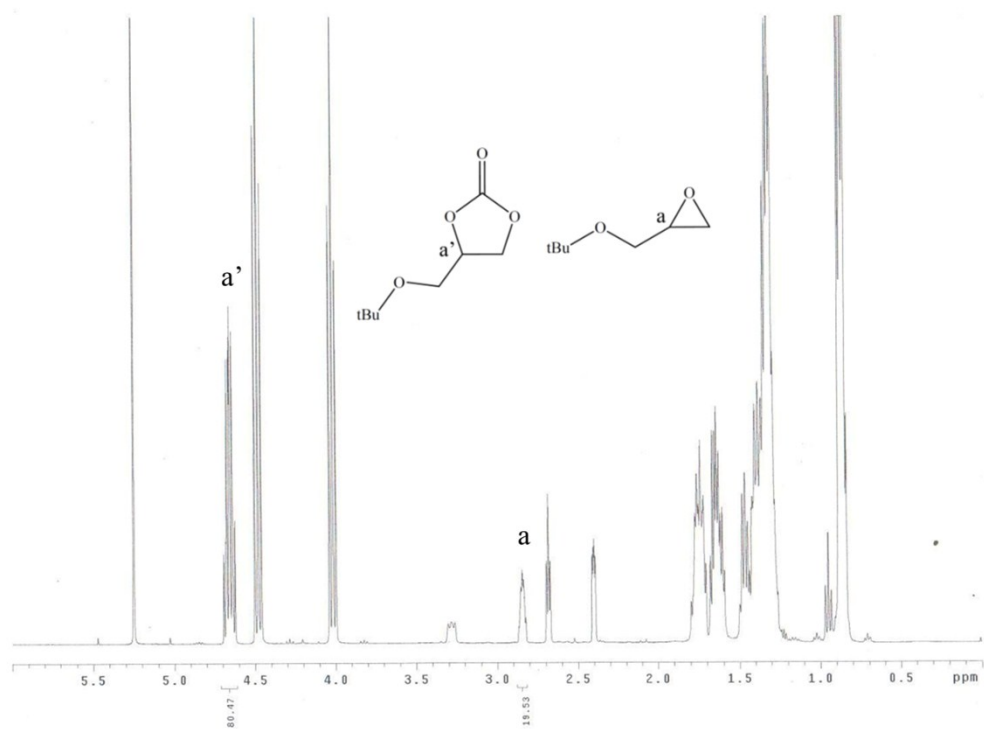




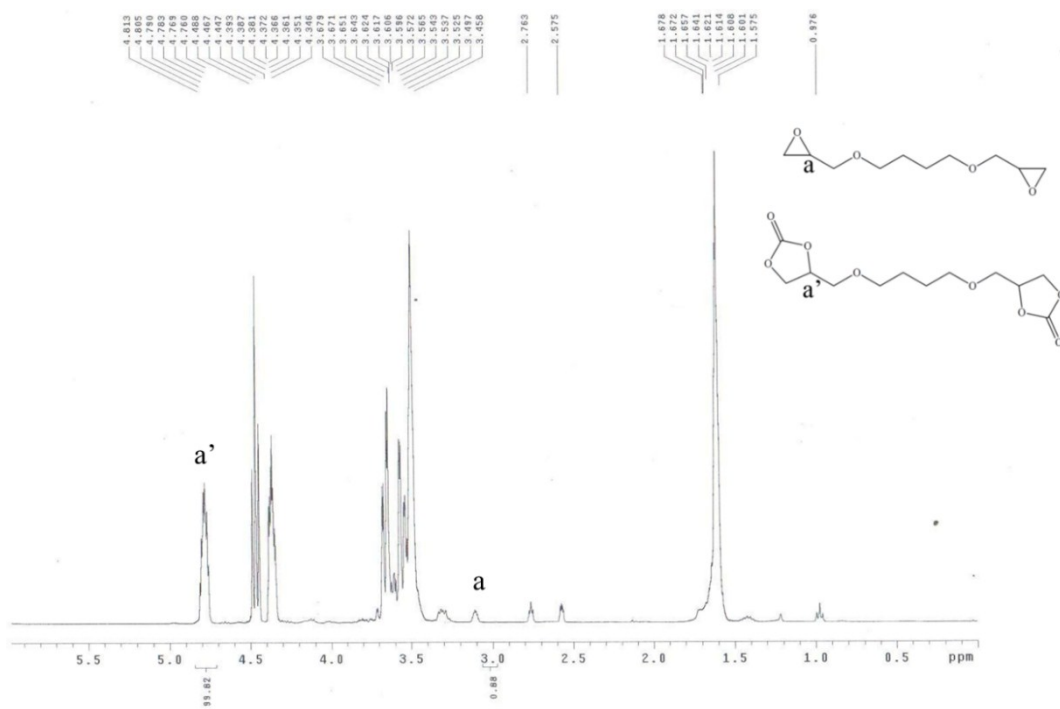
**Figure S13.** The  $^1\text{H}$  NMR spectrum for entry 1.



**Figure S14.** The  $^1\text{H}$  NMR spectrum for entry 2.



**Figure S15.** The  $^1\text{H}$  NMR spectrum for entry 3.



**Figure S16.** The <sup>1</sup>H NMR spectrum for entry 4.