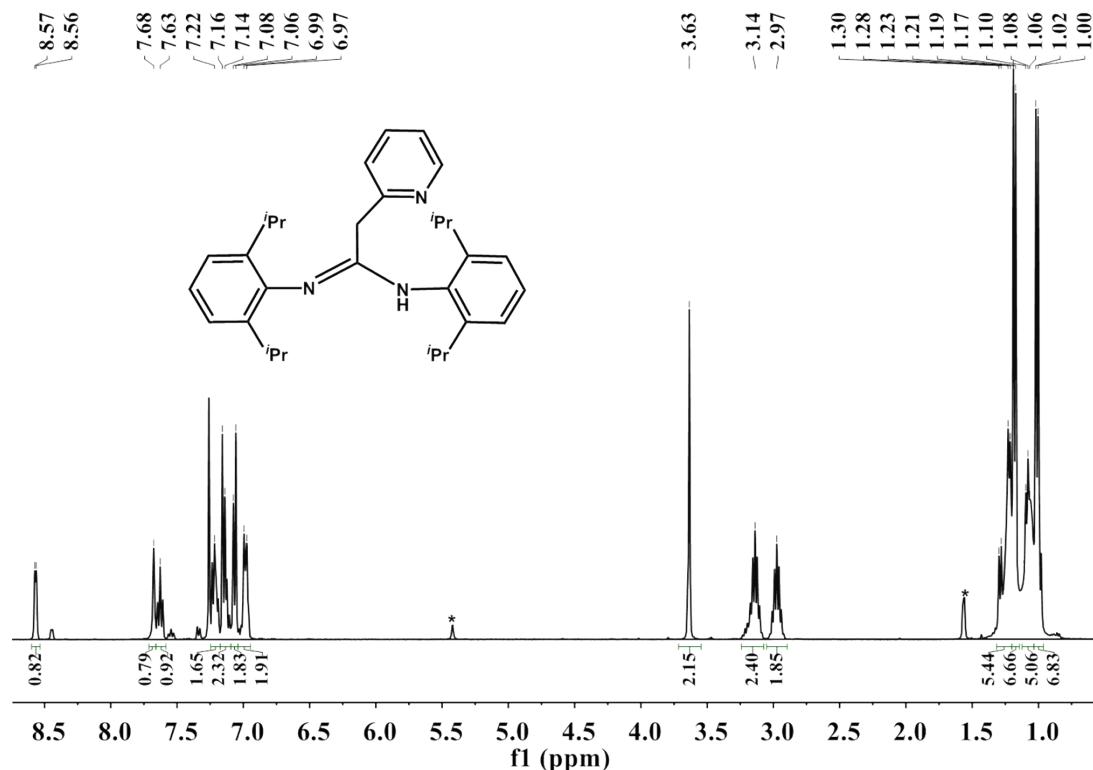
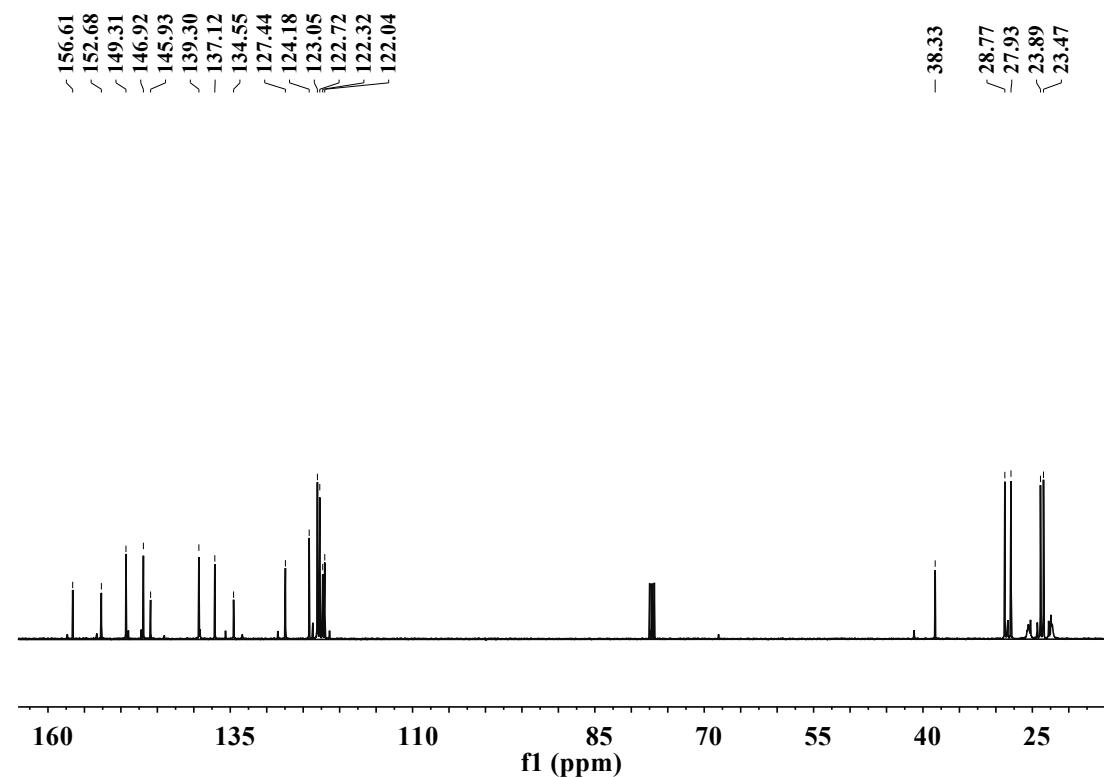


*Supporting information*

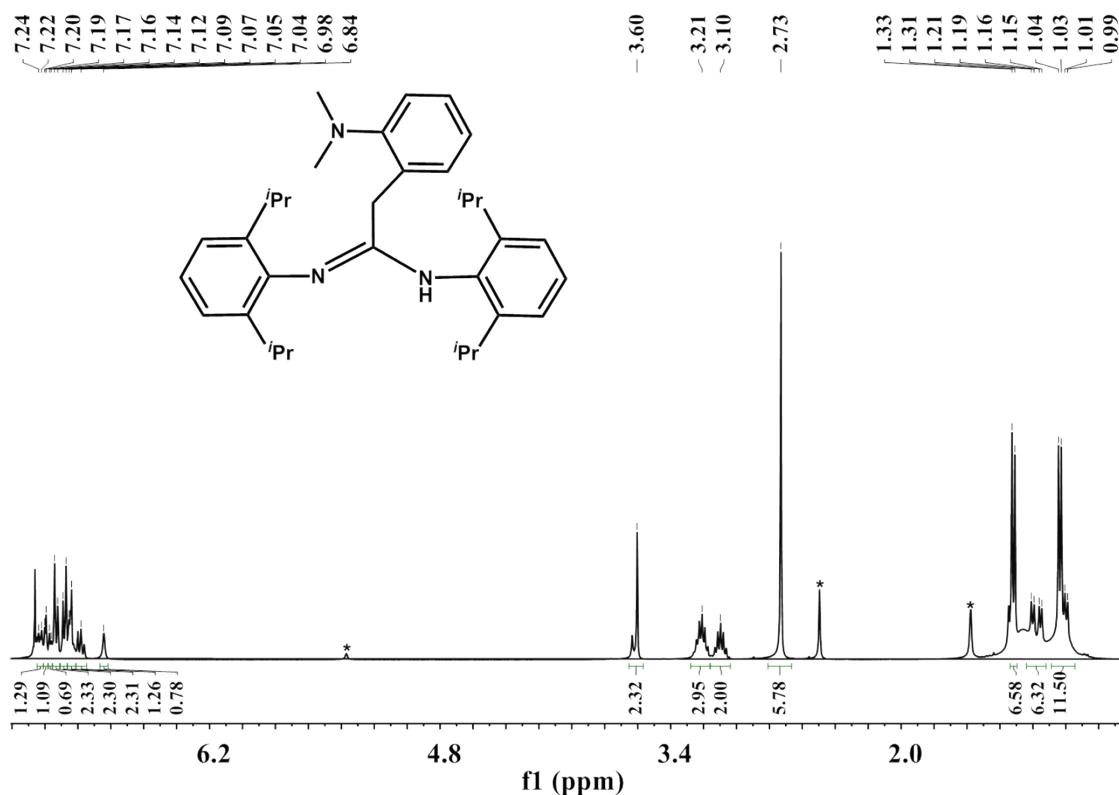
**Alkaline earth metal complexes stabilized by amidine and guanidine ligands:  
synthesis, structure and their catalytic activity towards polymerization of rac-lactide**



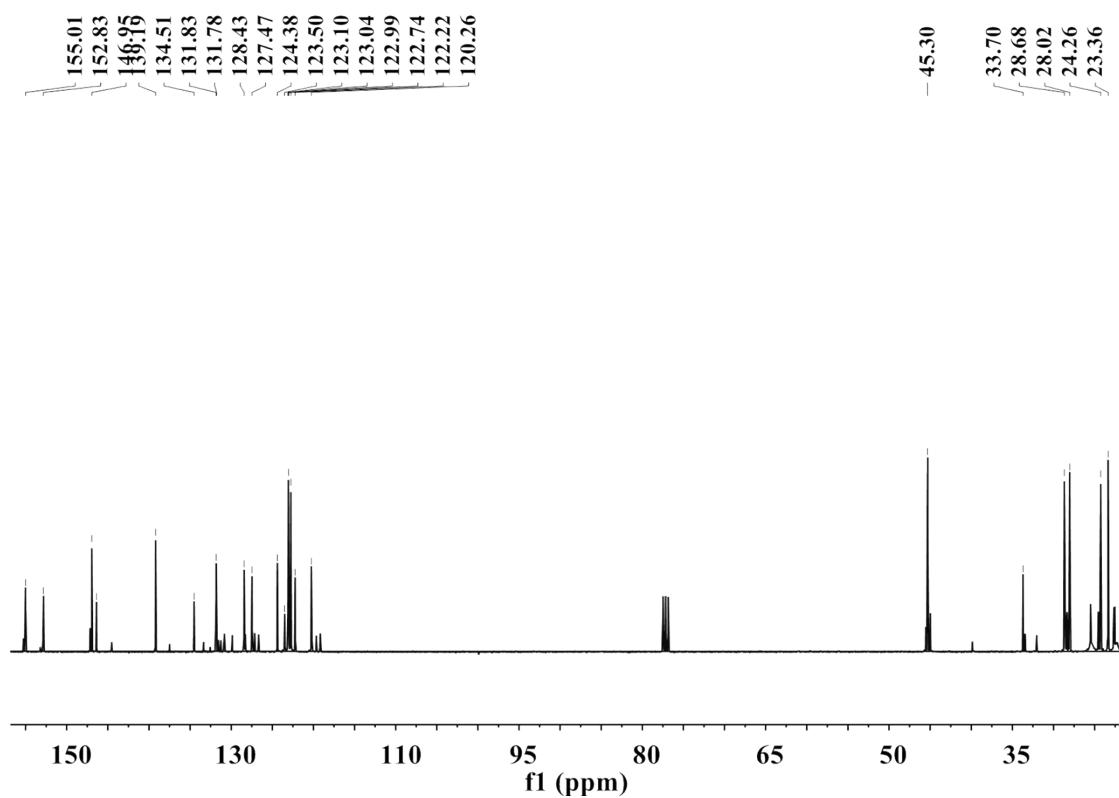
**Fig. S1.** <sup>1</sup>H NMR spectrum of HL<sup>1</sup> (CDCl<sub>3</sub>, 400 MHz, 25 °C)



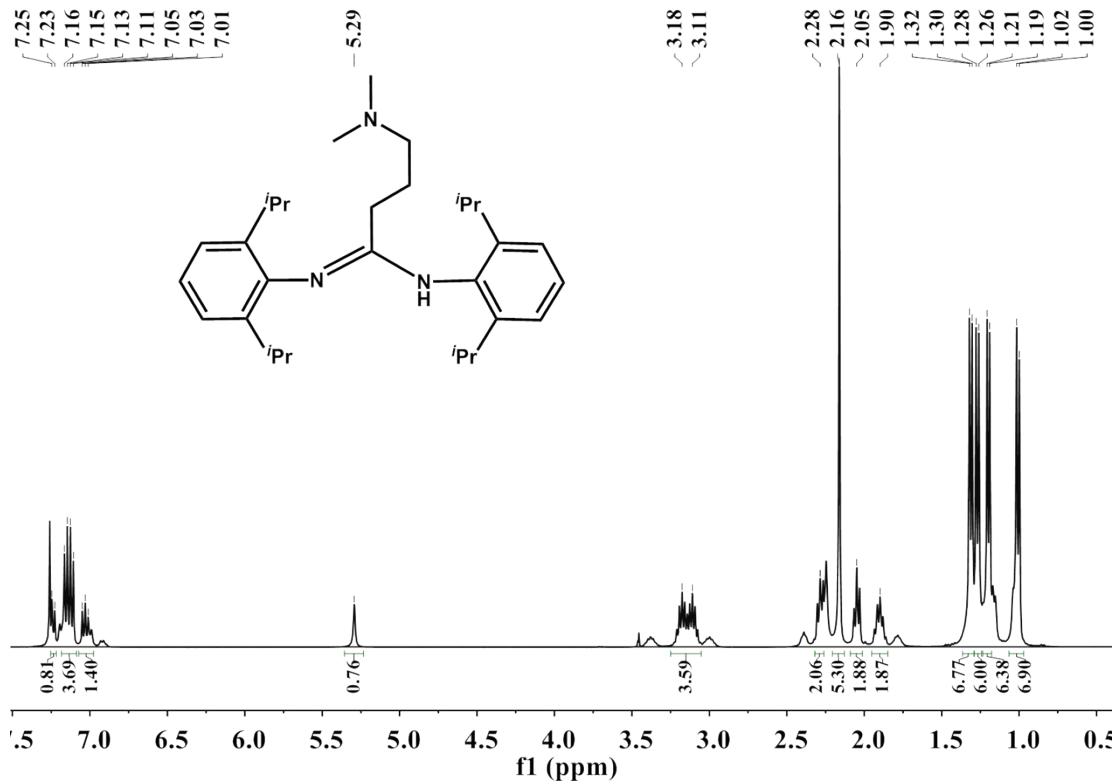
**Fig. S2.**  $^{13}\text{C}$  NMR spectrum of  $\text{HL}^1$  ( $\text{CDCl}_3$ , 100 MHz, 25 °C)



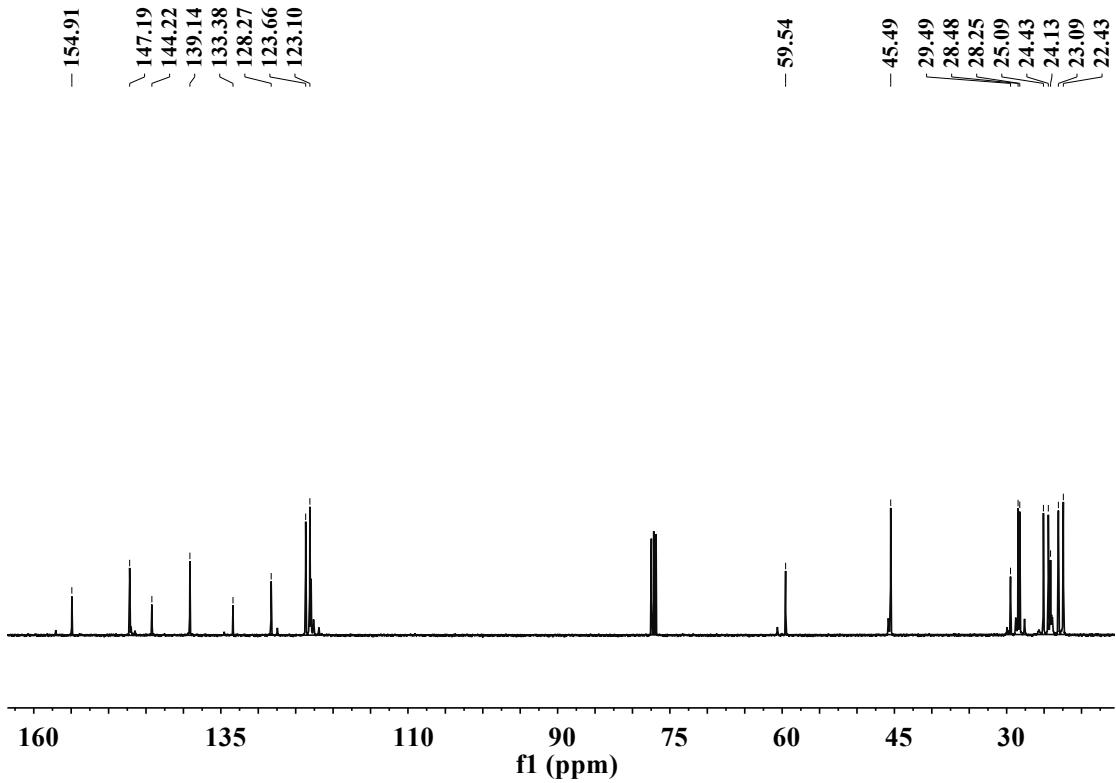
**Fig. S3.**  $^1\text{H}$  NMR spectrum of  $\text{HL}^2$  ( $\text{CDCl}_3$ , 400 MHz, 25 °C)



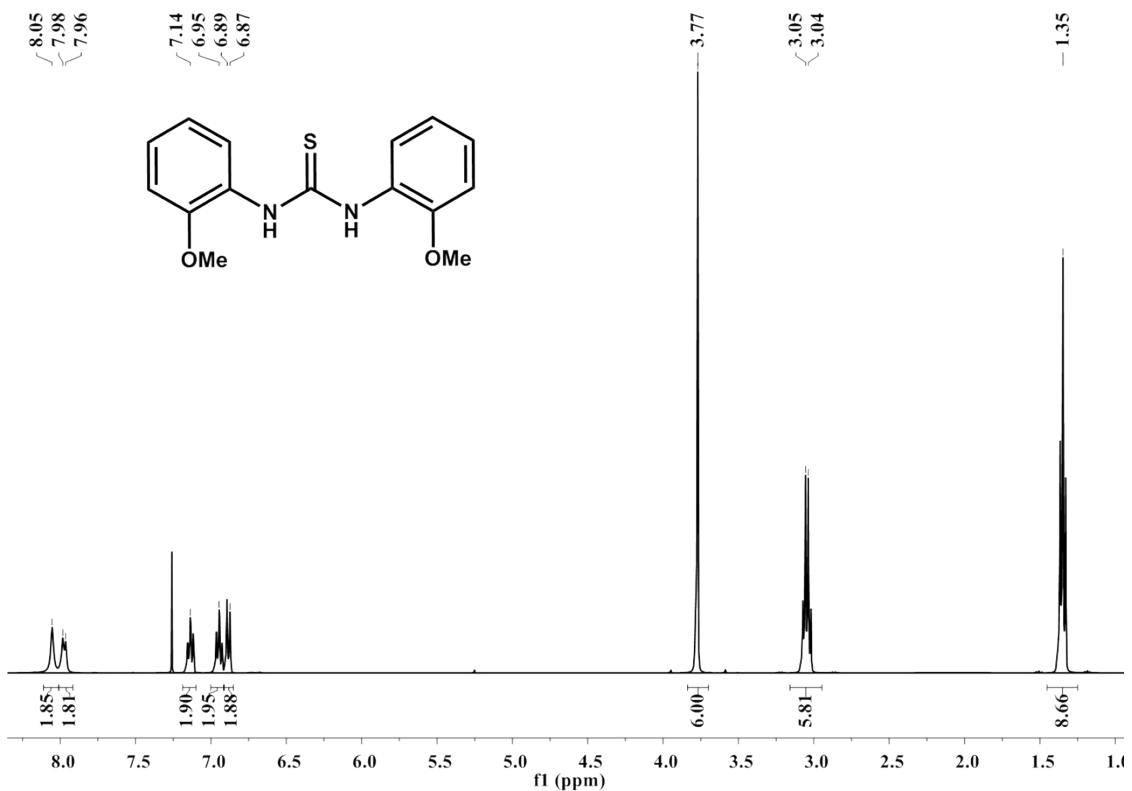
**Fig. S4.**  $^{13}\text{C}$  NMR spectrum of  $\text{HL}^2$  ( $\text{CDCl}_3$ , 100 MHz, 25 °C)



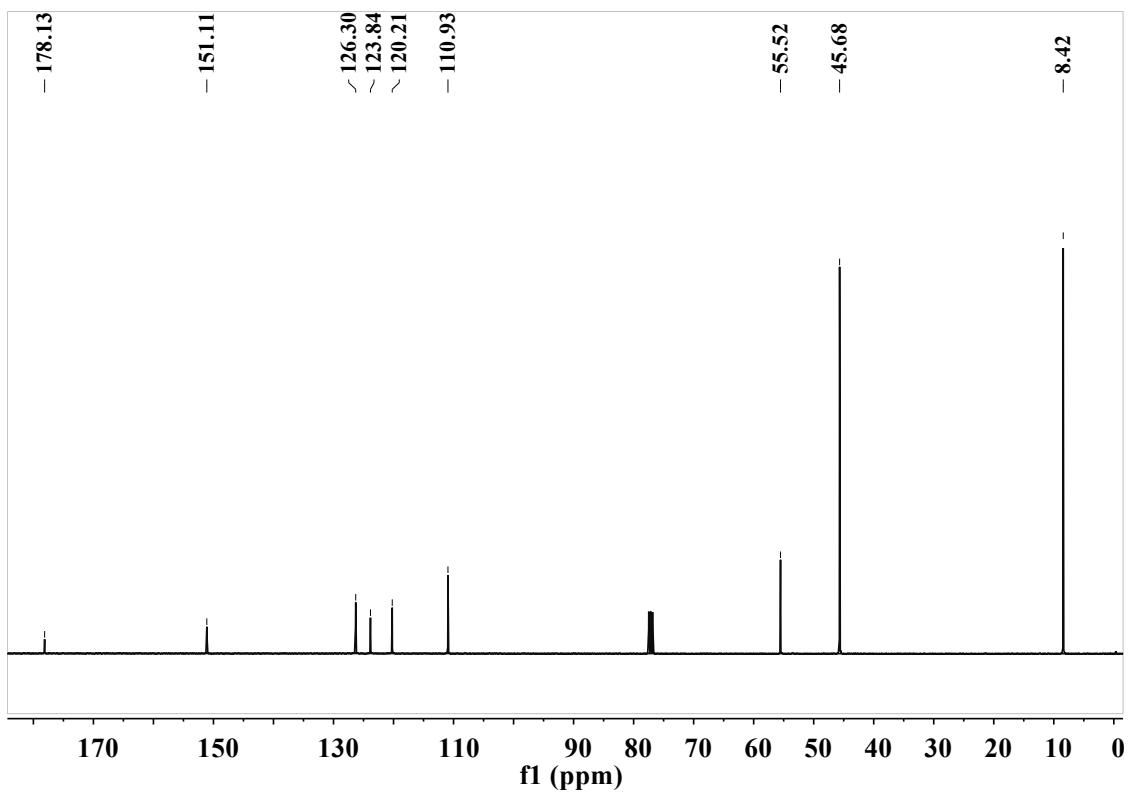
**Fig. S5.** <sup>1</sup>H NMR spectrum of HL<sup>3</sup> ( $\text{CDCl}_3$ , 400 MHz, 25 °C)



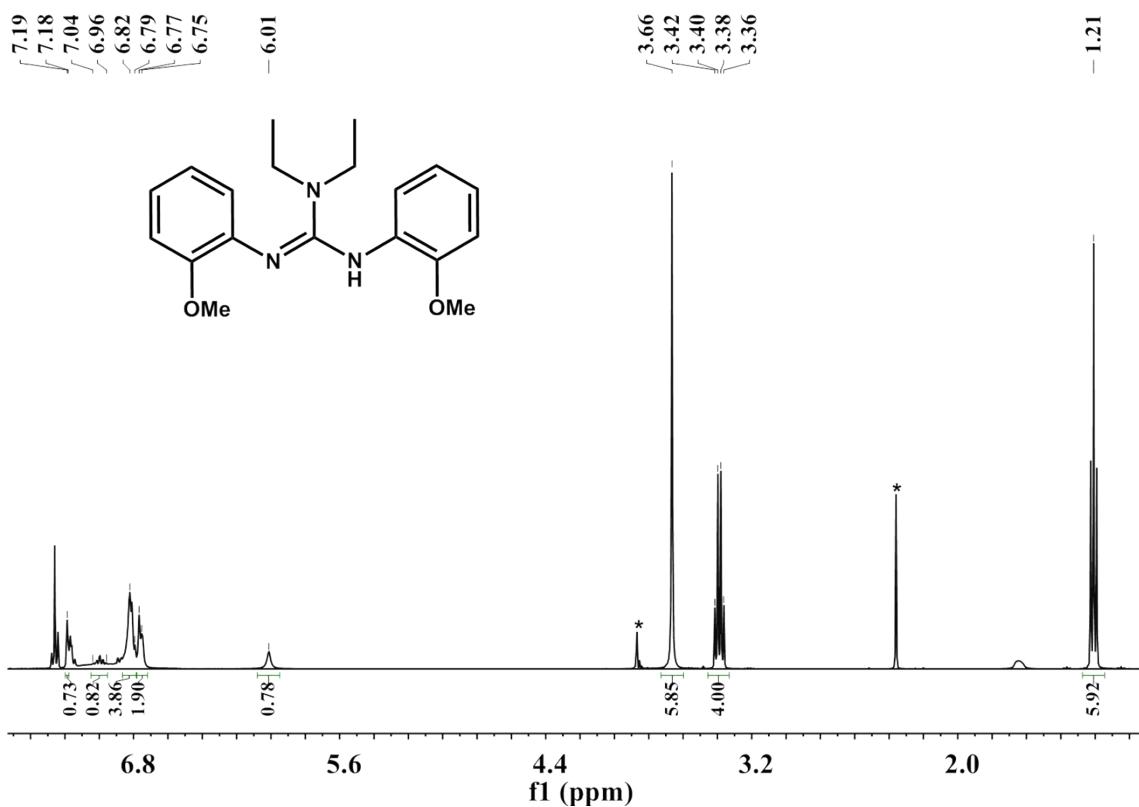
**Fig. S6.** <sup>13</sup>C NMR spectrum of HL<sup>3</sup> ( $\text{CDCl}_3$ , 100 MHz, 25 °C)



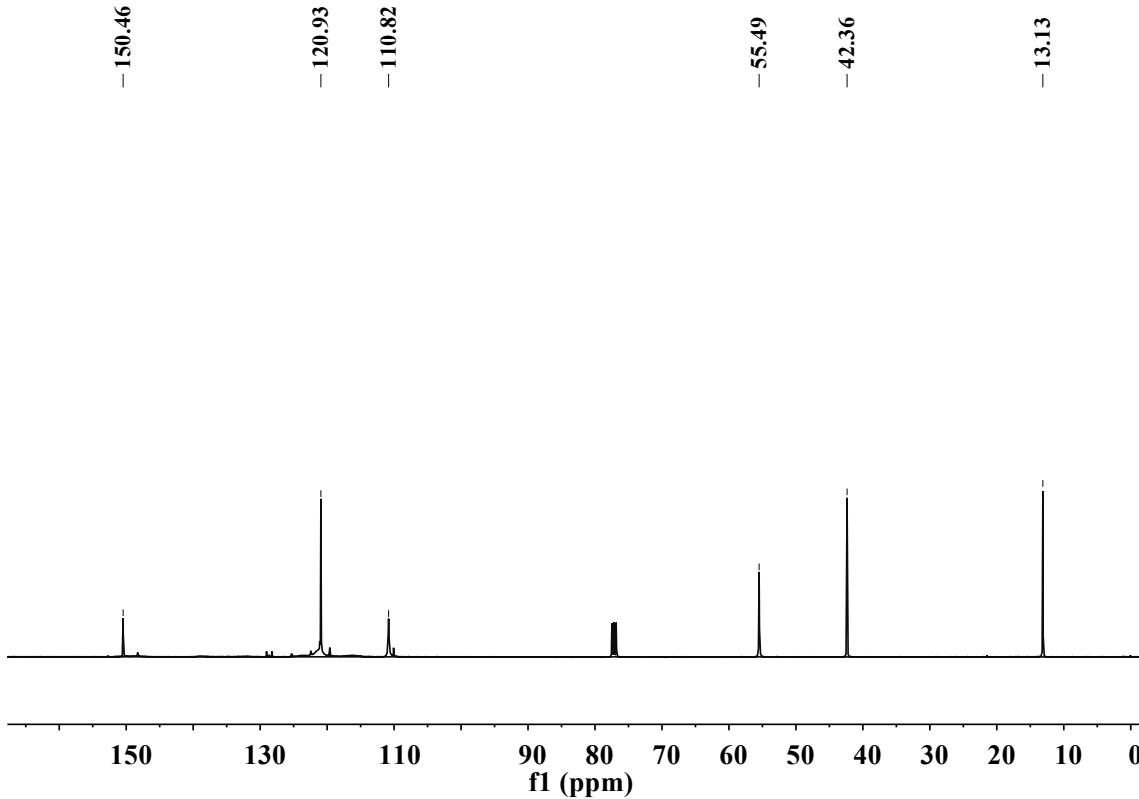
**Fig. S7.** <sup>1</sup>H NMR spectrum of thiourea (**I**) (CDCl<sub>3</sub>, 400 MHz, 25 °C)



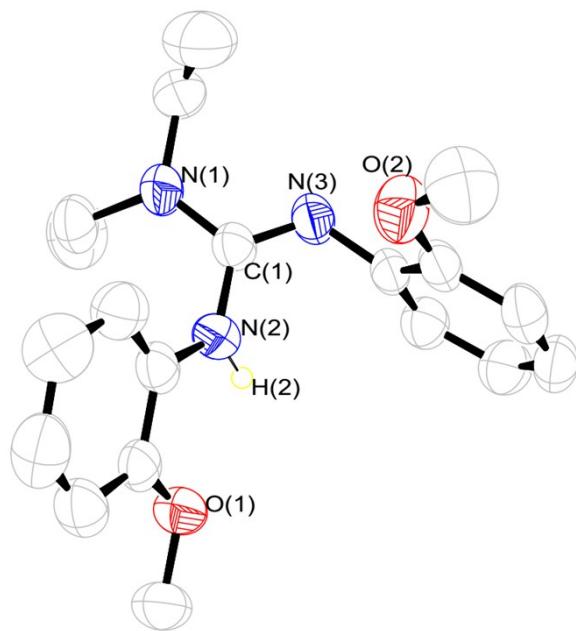
**Fig. S8.** <sup>13</sup>C NMR spectrum of thiourea (**I**) (CDCl<sub>3</sub>, 100 MHz, 25 °C)



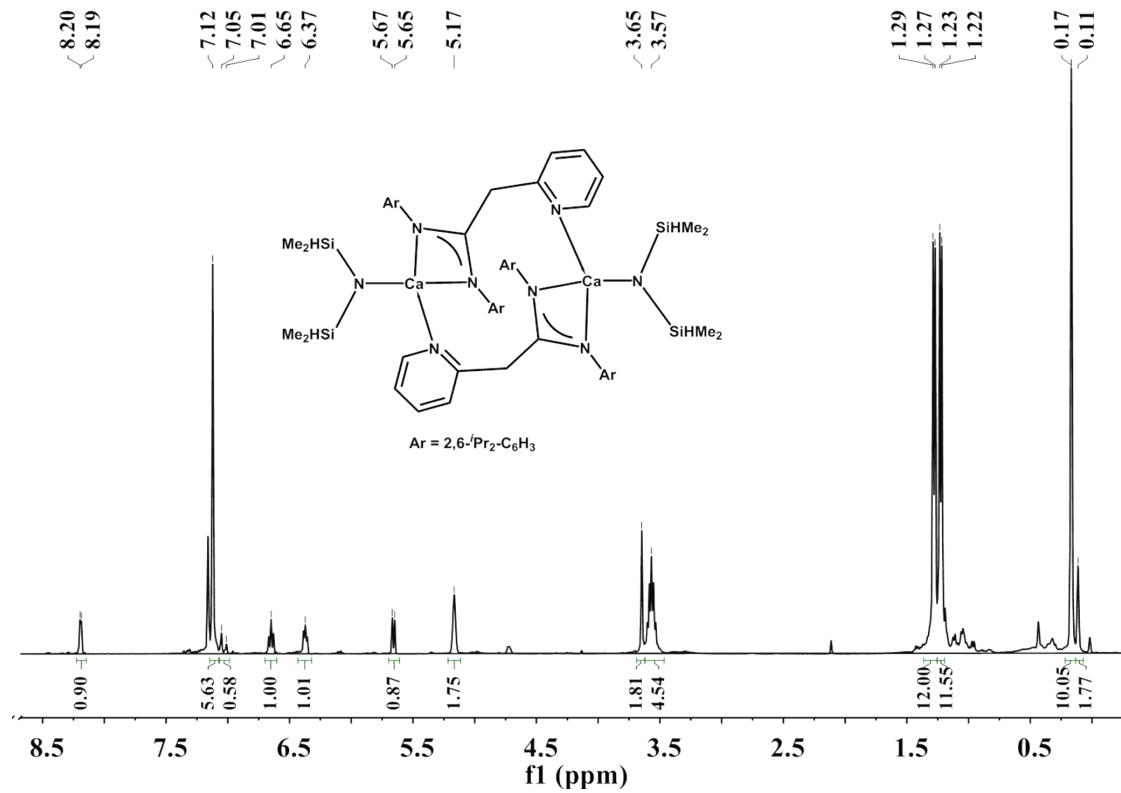
**Fig. S9.** <sup>1</sup>H NMR spectrum of HL<sup>4</sup> (CDCl<sub>3</sub>, 400 MHz, 25 °C)



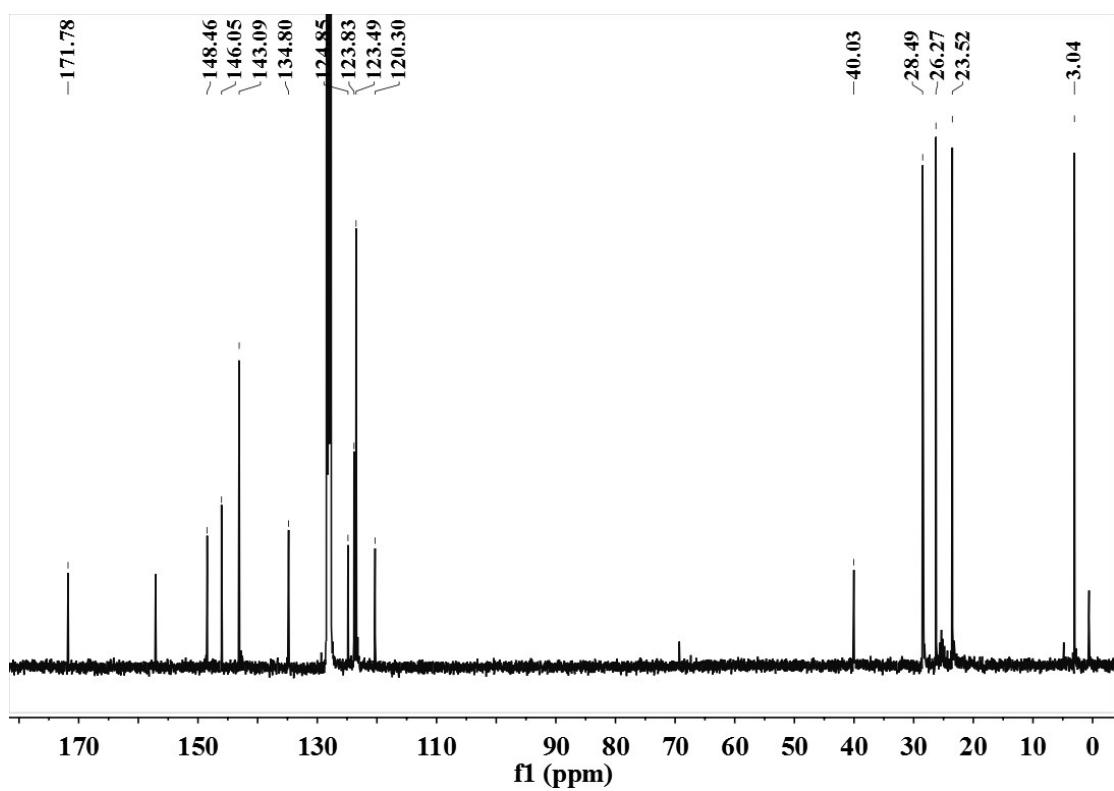
**Fig. S10.** <sup>13</sup>C NMR spectrum of HL<sup>4</sup> (CDCl<sub>3</sub>, 100 MHz, 25 °C)



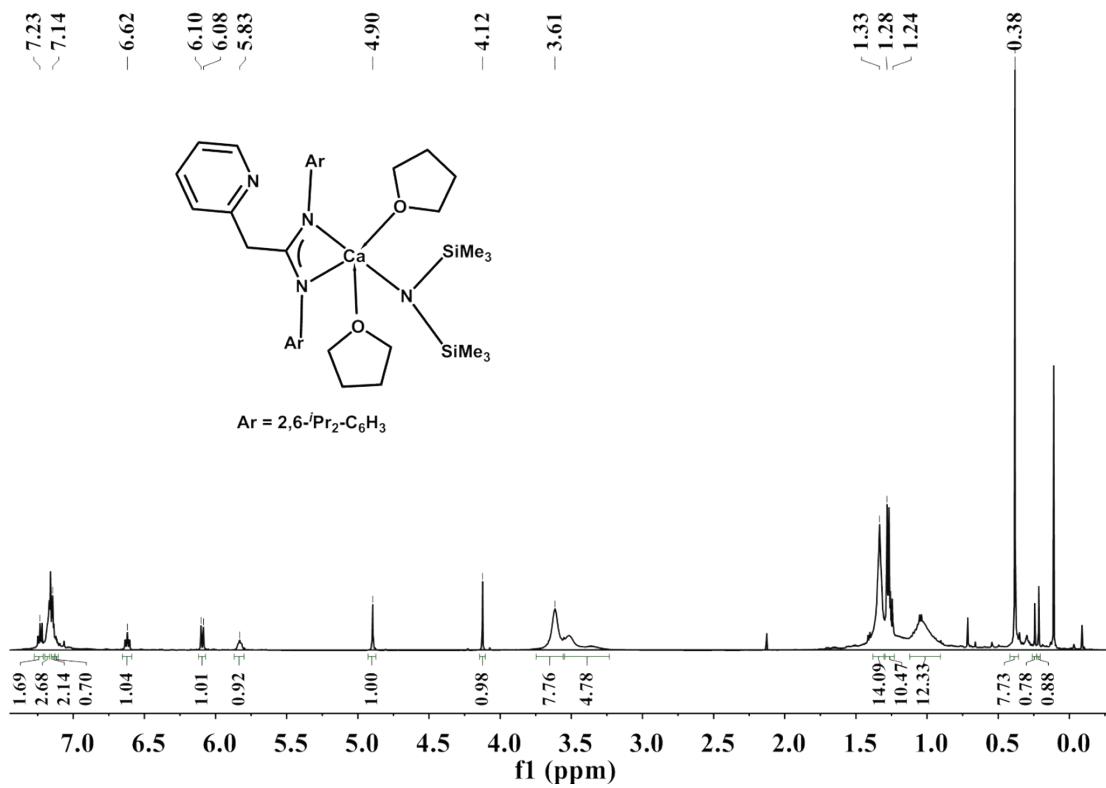
**Fig. S11.** Crystal structures of guanidine ligand **HL<sup>4</sup>**. Thermal ellipsoids are drawn at the 35% probability level. Hydrogen atoms except N-H are omitted for clarity. Selected bond lengths ( $\text{\AA}$ ) and angles (deg): N(3)-C(1) 1.273, N(2)-C(1) 1.391, N(1)-C(1) 1.359, N(2)-H(2) 0.860, N(3)-C(1)-N(1) 120.38, N(2)-C(1)-N(1) 116.84, H(2)-N(2)-C(1) 116.78.



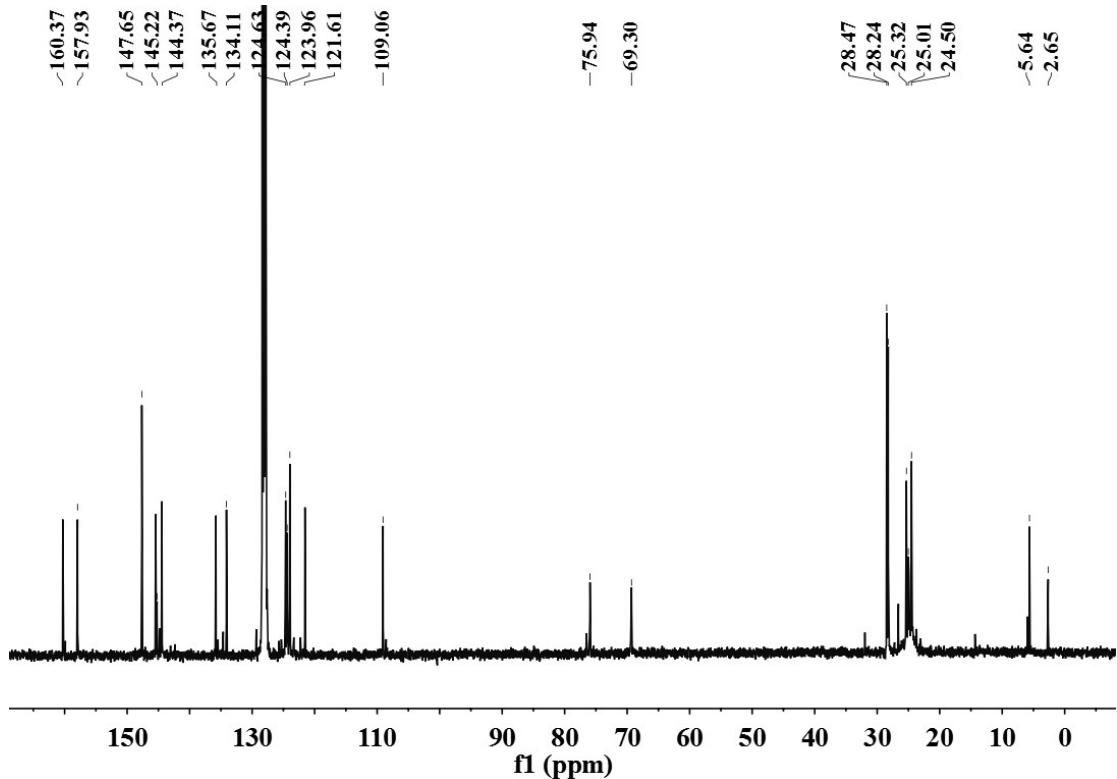
**Fig. S12.**  $^1\text{H}$  NMR spectrum of complex **1** ( $\text{C}_6\text{D}_6$ , 400 MHz, 25 °C)



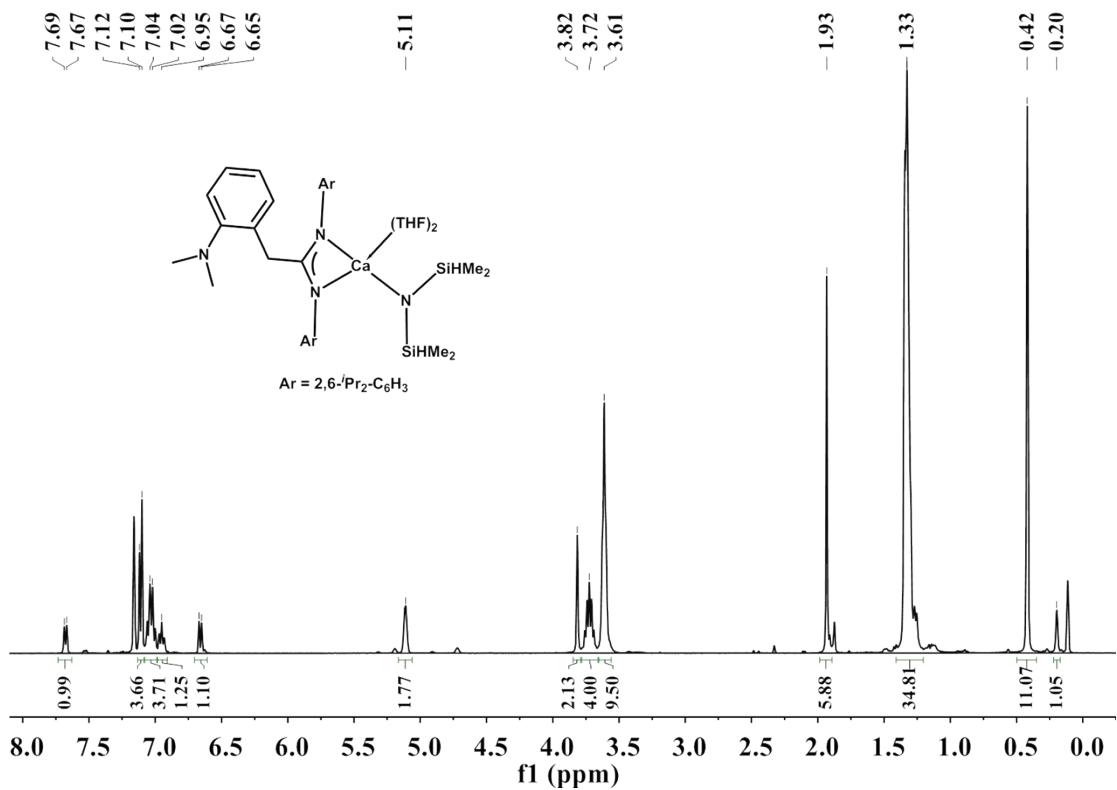
**Fig. S13.**  $^{13}\text{C}$  NMR spectrum of complex **1** ( $\text{C}_6\text{D}_6$ , 100 MHz, 25 °C)



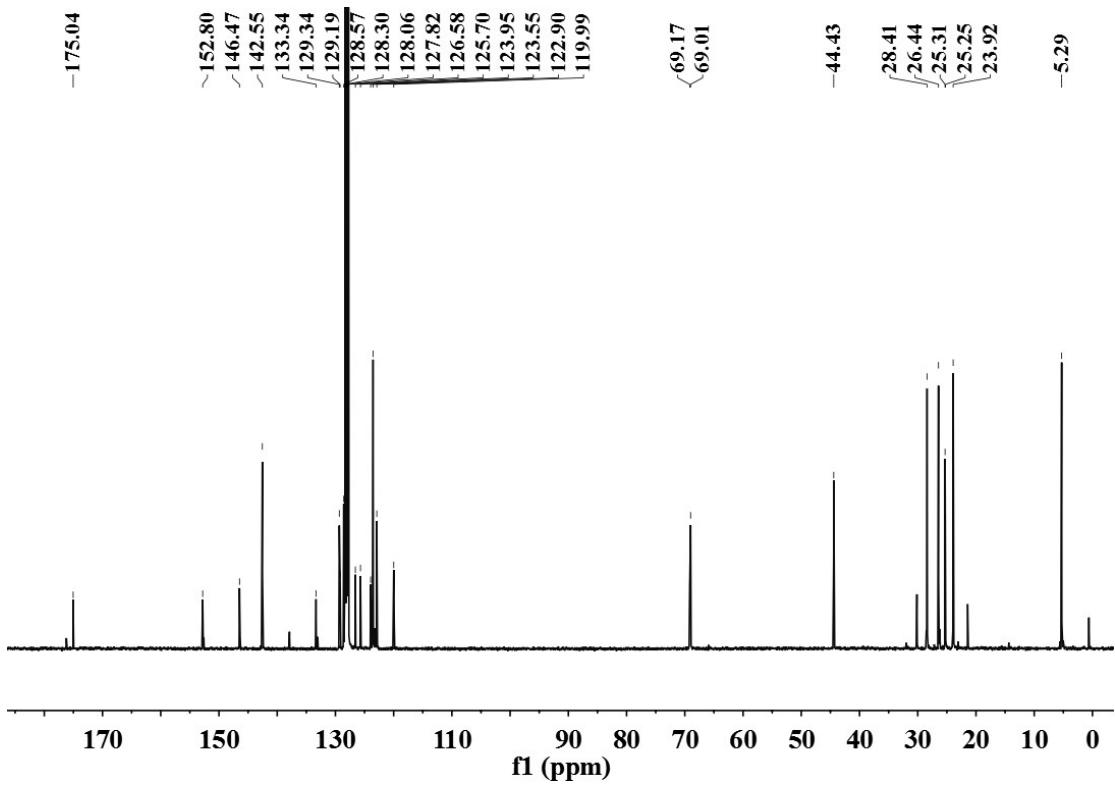
**Fig. S14.**  $^1\text{H}$  NMR spectrum of complex **2** ( $\text{C}_6\text{D}_6$ , 500 MHz, 25 °C)



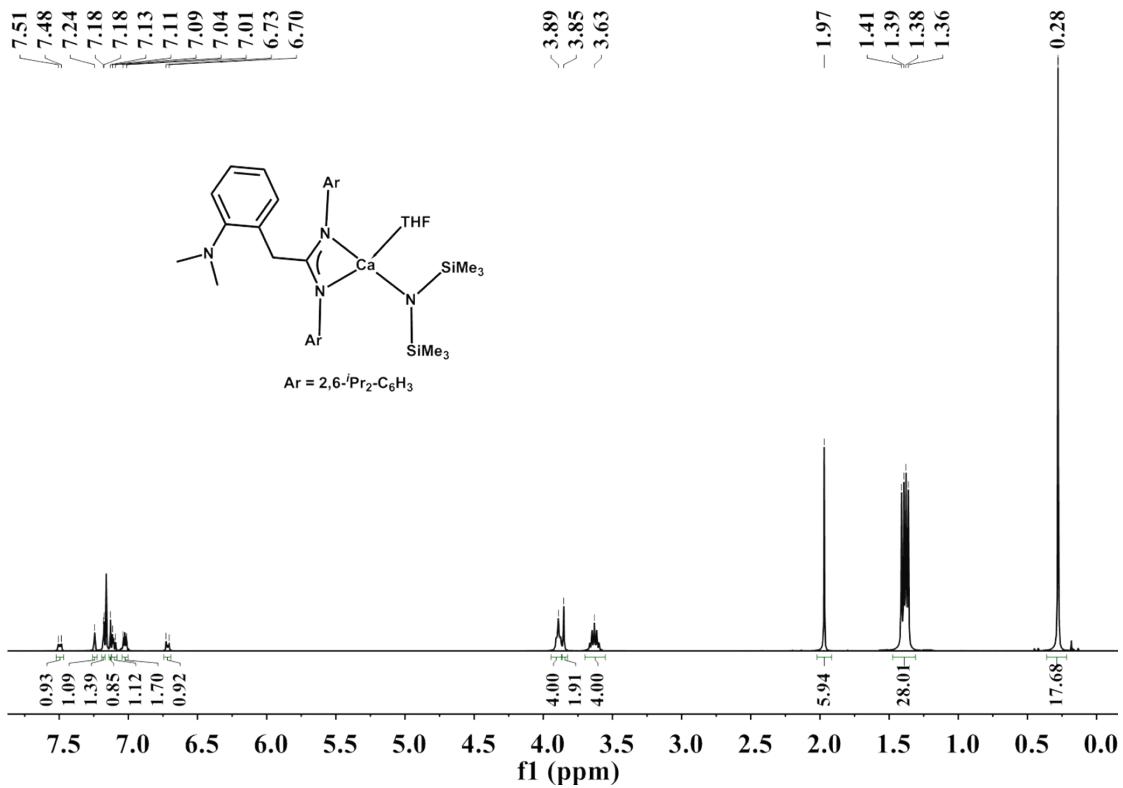
**Fig. S15.**  $^{13}\text{C}$  NMR spectrum of complex 2 ( $\text{C}_6\text{D}_6$ , 126 MHz, 25 °C)



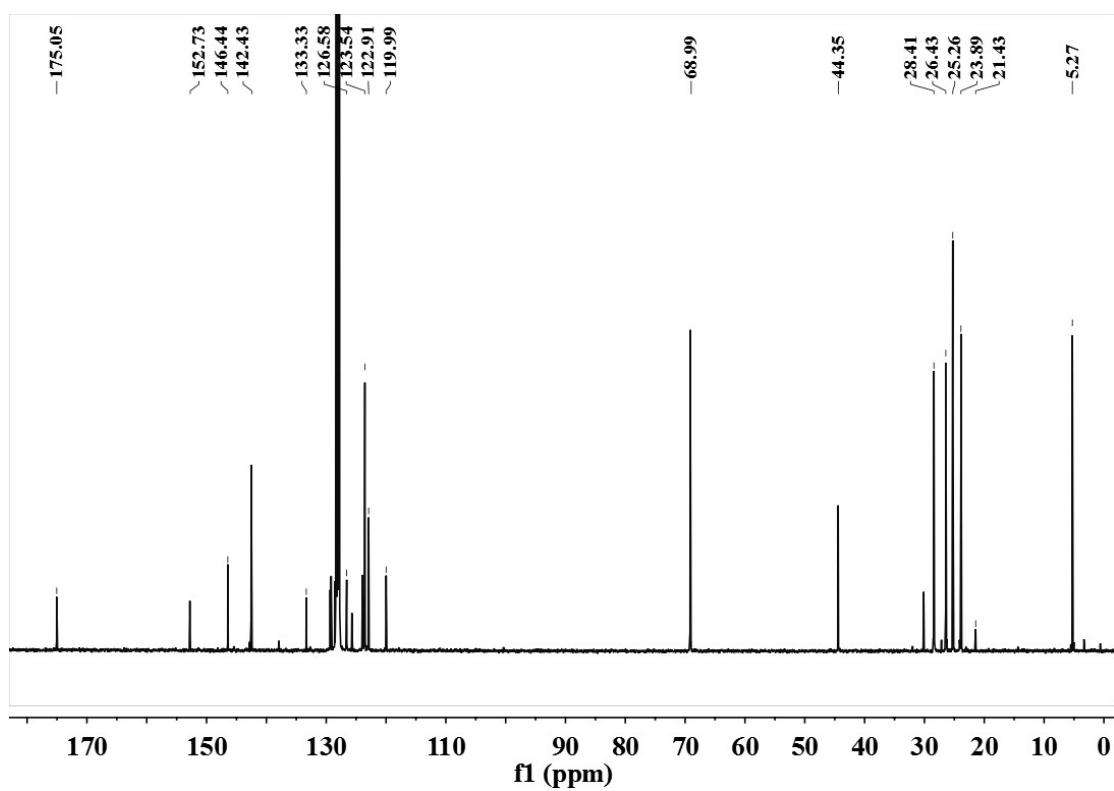
**Fig. S16.**  $^1\text{H}$  NMR spectrum of complex 3 ( $\text{C}_6\text{D}_6$ , 400 MHz, 25 °C)



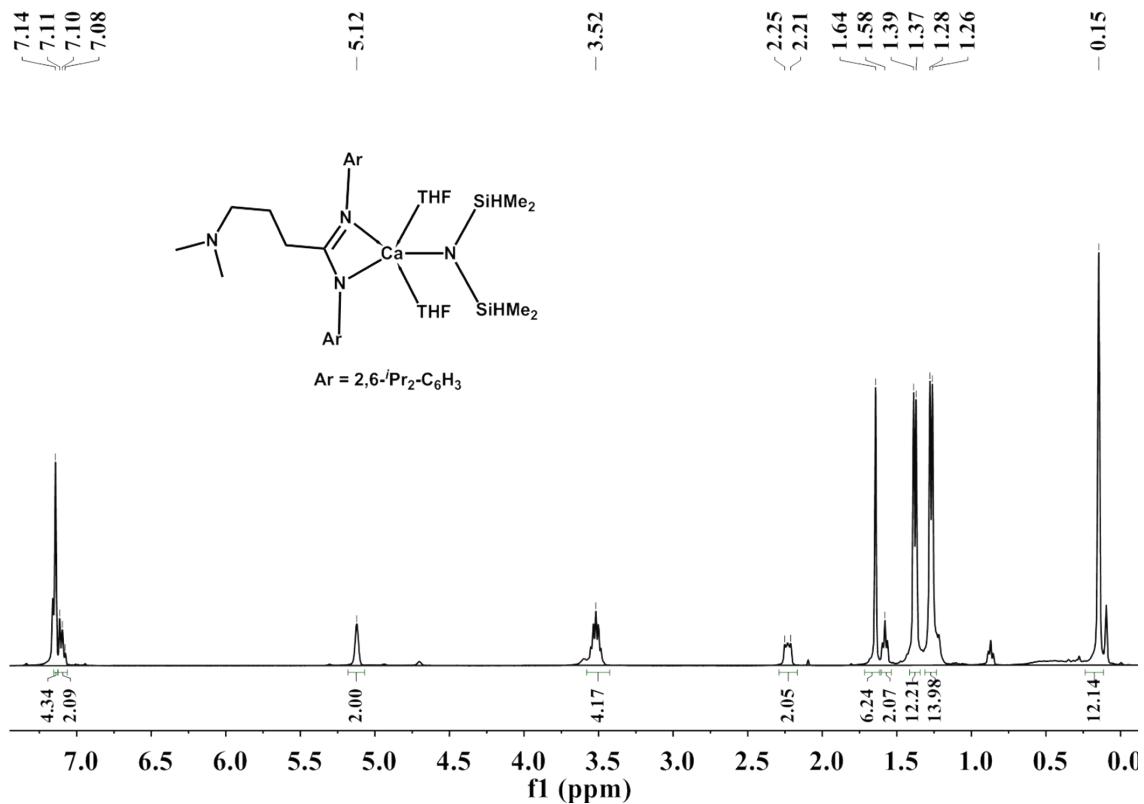
**Fig. S17.**  $^{13}\text{C}$  NMR spectrum of complex 3 ( $\text{C}_6\text{D}_6$ , 100 MHz, 25 °C)



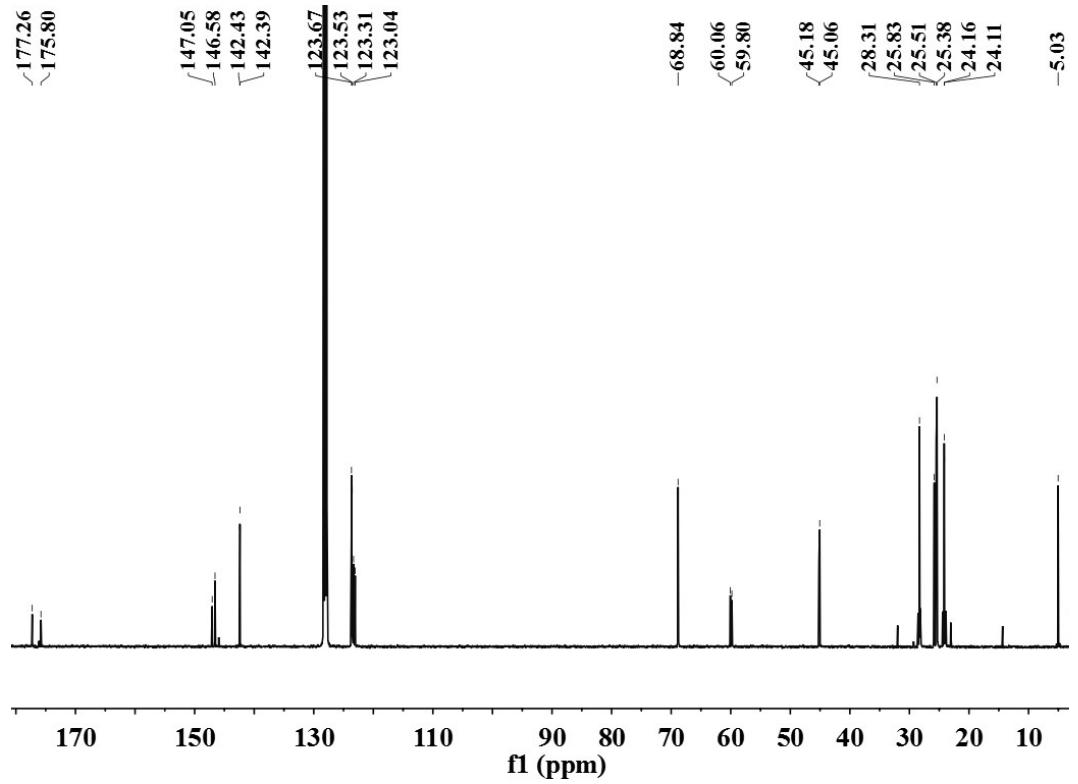
**Fig. S18.**  $^1\text{H}$  NMR spectrum of complex 4 ( $\text{C}_6\text{D}_6$ , 400 MHz, 25 °C)



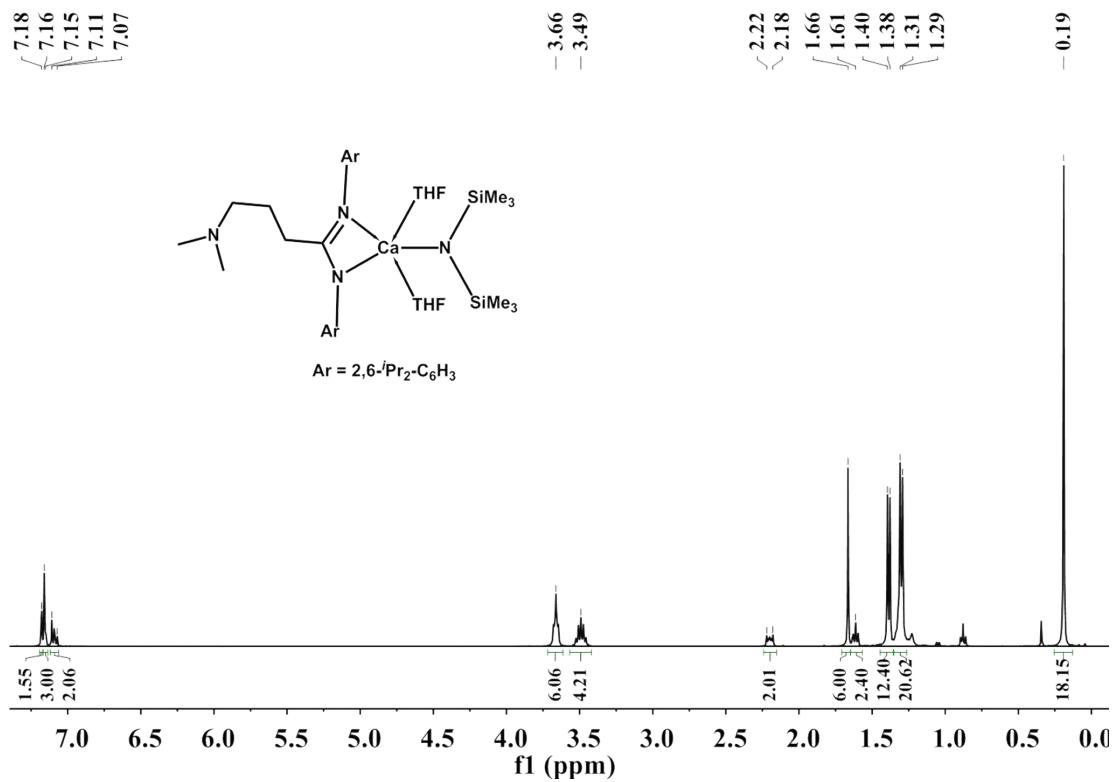
**Fig. S19.** <sup>13</sup>C NMR spectrum of complex 4 (C<sub>6</sub>D<sub>6</sub>, 100 MHz, 25 °C)



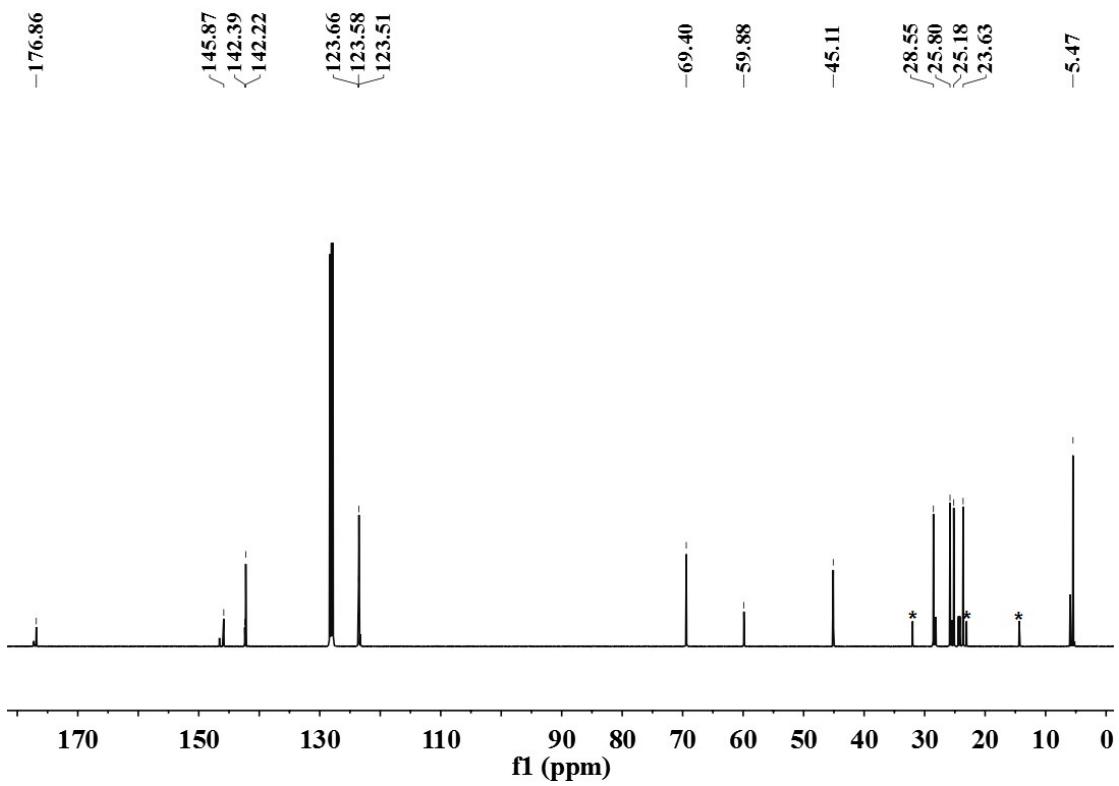
**Fig. S20.** <sup>1</sup>H NMR spectrum of complex 5 (C<sub>6</sub>D<sub>6</sub>, 400 MHz, 25 °C)



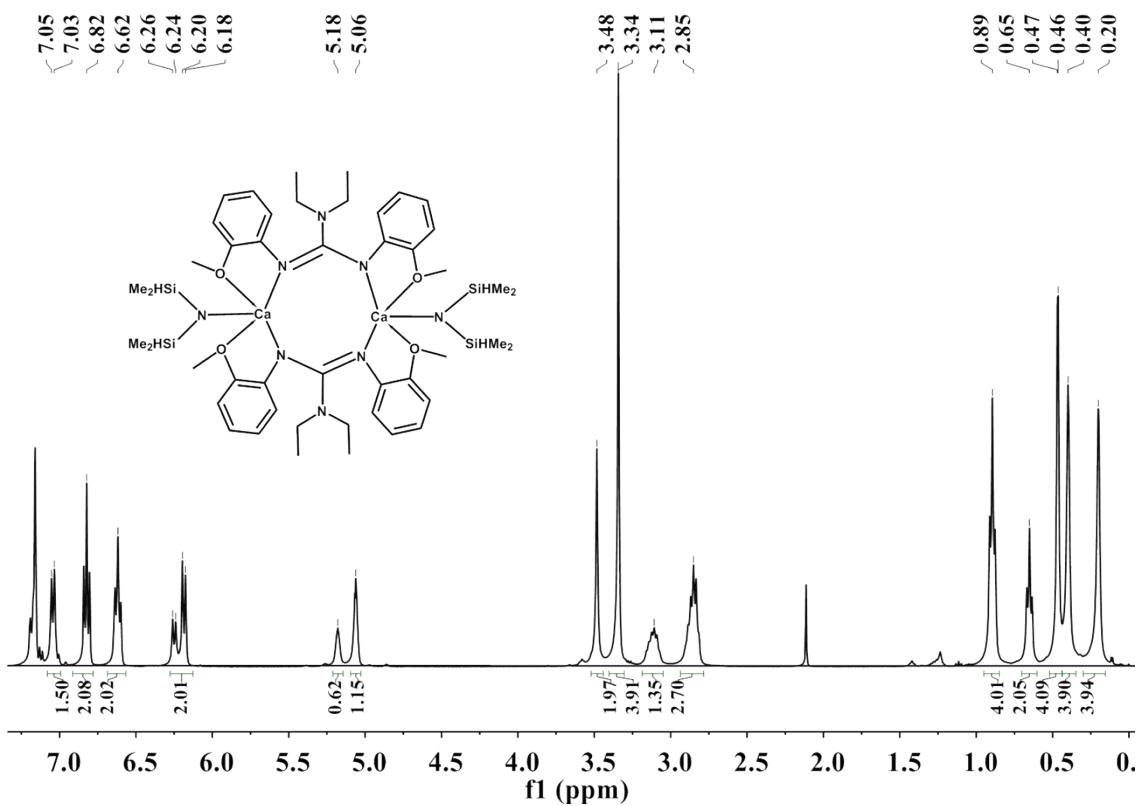
**Fig. S21.** <sup>13</sup>C NMR spectrum of complex 5 (C<sub>6</sub>D<sub>6</sub>, 100 MHz, 25 °C)



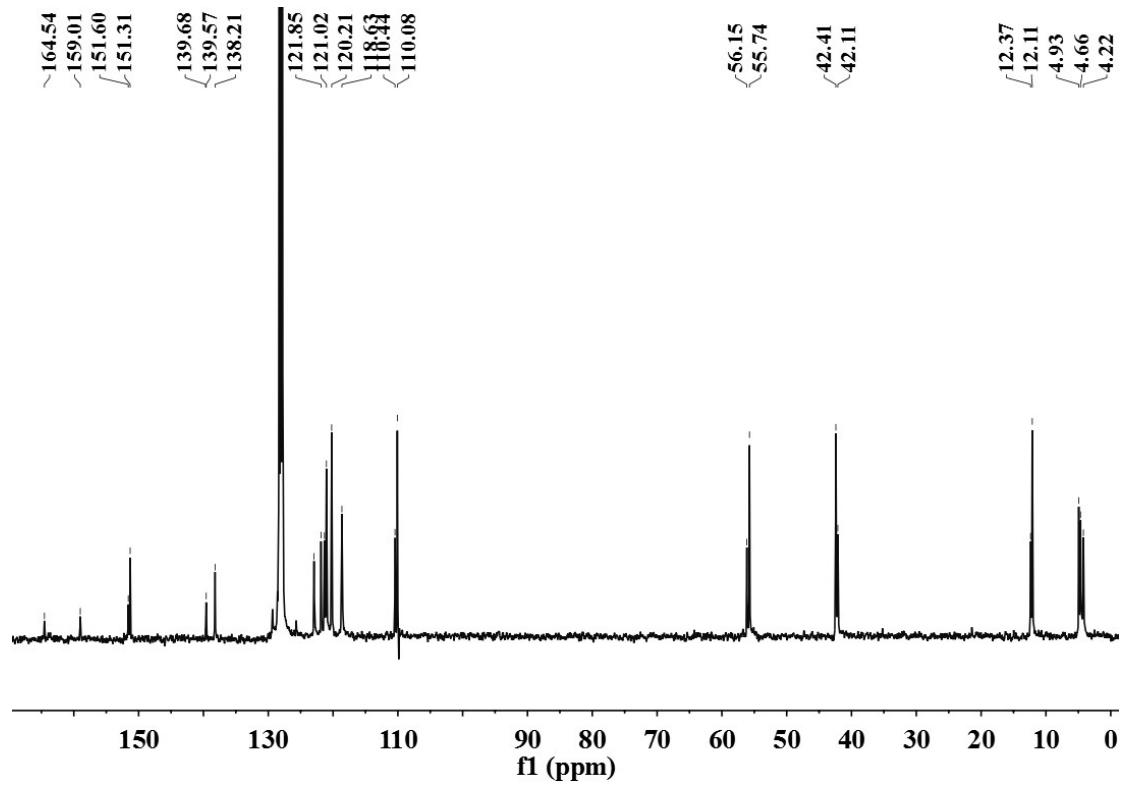
**Fig. S22.** <sup>1</sup>H NMR spectrum of complex 6 (C<sub>6</sub>D<sub>6</sub>, 400 MHz, 25 °C)



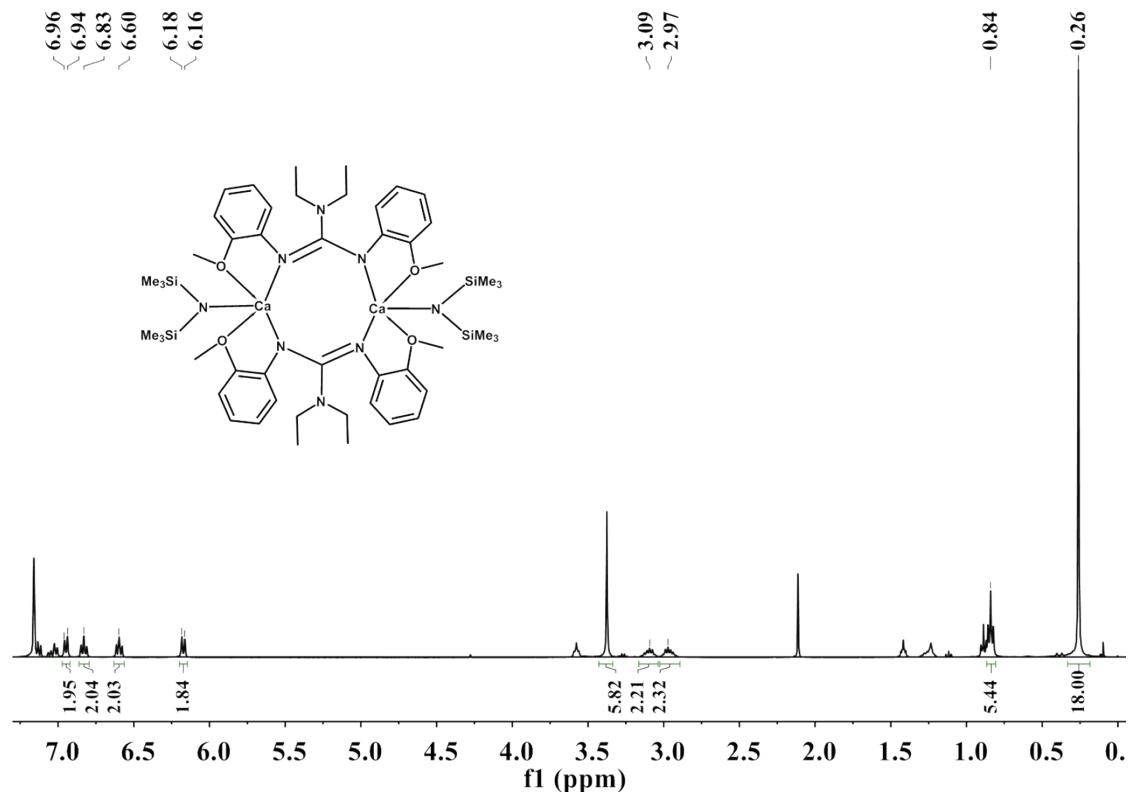
**Fig. S23.**  $^{13}\text{C}$  NMR spectrum of complex 6 ( $\text{C}_6\text{D}_6$ , 100 MHz, 25 °C)



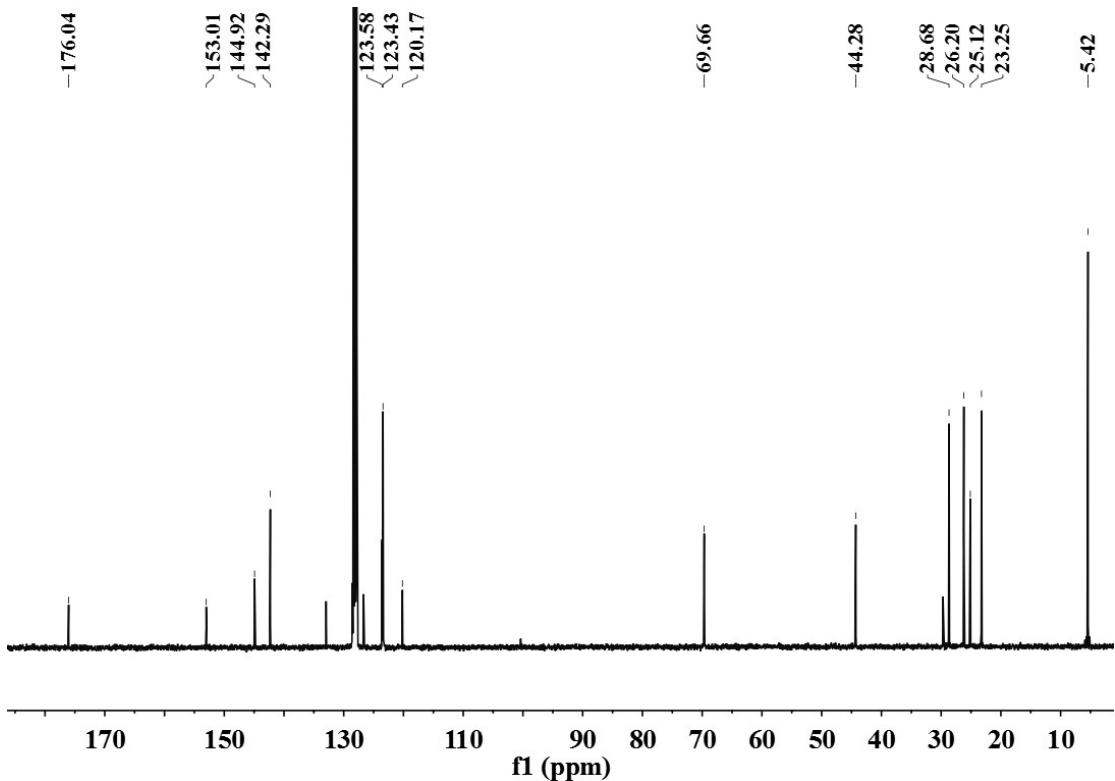
**Fig. S24.**  $^1\text{H}$  NMR spectrum of complex 7 ( $\text{C}_6\text{D}_6$ , 400 MHz, 25 °C)



**Fig. S25.**  $^{13}\text{C}$  NMR spectrum of complex 7 ( $\text{C}_6\text{D}_6$ , 100 MHz, 25 °C)



**Fig. S26.**  $^1\text{H}$  NMR spectrum of complex 8 ( $\text{C}_6\text{D}_6$ , 400 MHz, 25 °C)



**Fig. S27.**  $^{13}\text{C}$  NMR spectrum of complex **8** ( $\text{C}_6\text{D}_6$ , 100 MHz, 25 °C)

**Table S1.** Crystallographic data and structure refinement details for **HL<sup>4</sup>**

Parameter	<b>HL<sup>4</sup></b>
Empirical formula	$\text{C}_{19}\text{H}_{25}\text{N}_3\text{O}_2$
Formula weight	387.47
Temperature/K	296(2)
Wavelength/ Å	0.71073
Crystal system	Triclinic
space group	P-1
<i>a</i> (Å)	8.965(5)
<i>b</i> (Å)	8.972(5)
<i>c</i> (Å)	13.985(5)
$\alpha$ (deg)	93.281(5)
$\beta$ (deg)	102.433(5)
$\gamma$ (deg)	106.766(5)
<i>V</i> (Å <sup>3</sup> )	1043.2(9)
<i>Z</i>	2
$D_{\text{calcd}}$ [g cm <sup>-3</sup> ]	1.234
$\mu$ [mm <sup>-1</sup> ]	0.080
<i>F</i> [000]	412
$2\theta_{\text{max}}$ [deg]	50.00
Reflections collected	6049
Independent reflections	3658
	[R(int) = 0.0346]

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GOF	1.079
Final <i>R</i> indices[ $I > 2\sigma(I)$ ]	$R_1 = 0.0555$ , $wR_2 = 0.1665$

**Table S2.** Crystallographic data and structure refinement details for amidinate complexes **3**, **5-6**

Parameter	<b>3</b>	<b>5</b>	<b>6</b>
Empirical formula	$C_{53}H_{84}CaN_4O_2Si_2$	$C_{42}H_{76}CaN_4O_2Si_2$	$C_{44}H_{80}CaN_4O_2Si_2$
Formula weight	905.50	765.32	793.38
Temperature/K	184.5(2)	184.5(2)	184.5(2)
Wavelength/ Å	0.71073	0.71073	0.71073
Crystal system	Triclinic	Monoclinic	Orthorhombic
space group	P-1	P2(1)/n	Pbca
<i>a</i> (Å)	10.9102(10)	11.670(5)	20.9710(11)
<i>b</i> (Å)	13.5063(13)	20.804(5)	19.1804(10)
<i>c</i> (Å)	19.3671(19)	20.145(5)	24.3516(12)
$\alpha$ (deg)	92.792(2)	90.000(5)	90.00
$\beta$ (deg)	102.919(2)	103.268(5)	90.00
$\gamma$ (deg)	91.739(2)	90.000(5)	90.00
<i>V</i> (Å <sup>3</sup> )	2775.8(5)	4760.0(3)	9795.0(9)
<i>Z</i>	2	4	8
D <sub>calcd</sub> [g cm <sup>-3</sup> ]	1.082	1.068	1.076
$\mu$ [mm <sup>-1</sup> ]	0.196	0.217	0.213
<i>F</i> [000]	986	1680	3488
2θ <sub>max</sub> [deg]	52.20	50.24	50.08
Reflections			
collected	14670	27710	55853
Independent	10692	17804	8642
reflections	[R(int) = 0.0894]	[R(int) = 0.0802]	[R(int) = 0.1163]
GOF	1.026	1.037	1.019
Final <i>R</i>	$R_1 = 0.0789$ ,	$R_1 = 0.0753$ ,	$R_1 = 0.0706$ ,
indices[ $I > 2\sigma(I)$ ]	$wR_2 = 0.2577$	$wR_2 = 0.2162$	$wR_2 = 0.1734$

**Table S3** Crystallographic data and structure refinement details for guanidinate complexes **7-8**

Parameter	<b>7</b>	<b>8</b>
Empirical formula	$C_{23}H_{38}CaN_4O_2Si_2$	$C_{32}H_{50}CaN_4O_2Si_2$
Formula weight	498.83	619.02
Temperature/K	184.5(2)	184.5(2)
Wavelength/ Å	0.71073	0.71073
Crystal system	Orthorhombic	Orthorhombic
space group	Pccn	Pccn
<i>a</i> (Å)	14.8204(14)	14.9714(12)
<i>b</i> (Å)	15.7062(15)	16.1922(13)

<i>c</i> (Å)	29.189(3)	29.027(2)
$\alpha$ (deg)	90.00	90.00
$\beta$ (deg)	90.00	90.00
$\gamma$ (deg)	90.00	90.00
<i>V</i> (Å <sup>3</sup> )	6794.3(11)	7036.8(10)
<i>Z</i>	8	8
D <sub>calcd</sub> [g cm <sup>-3</sup> ]	0.975	1.169
$\mu$ [mm <sup>-1</sup> ]	0.276	0.279
<i>F</i> [000]	2144	2672
2 <i>θ</i> <sub>max</sub> [deg]	52.80	50.08
Reflections collected	35714	39332
Independent reflections	6960	6229
	[R(int) = 0.0580]	[R(int) = 0.0784]
GOF	1.029	1.014
Final <i>R</i> indices[ <i>I</i> >2σ( <i>I</i> )]	R <sub>1</sub> = 0.0645, wR <sub>2</sub> = 0.2047	R <sub>1</sub> = 0.0535, wR <sub>2</sub> = 0.1413

**Table S4** ROP of *rac*-LA catalyzed by calcium complexes **1-8**<sup>a</sup>

entry	Cat.	[M] <sub>0</sub> /[cat] <sub>0</sub> /[CTA] <sub>0</sub>	Conv. <sup>b</sup> (%)	<i>M</i> <sub>n,calcd</sub> <sup>c</sup> (10 <sup>4</sup> )	<i>M</i> <sub>n,exp</sub> <sup>d</sup> (10 <sup>4</sup> )	<i>M</i> <sub>w</sub> / <i>M</i> <sub>n</sub> <sup>d</sup>
1	<b>1</b>	200/1/0	63	1.82	1.68	1.92
2	<b>2</b>	200/1/0	51	1.47	1.20	1.31
3	<b>3</b>	200/1/0	78	2.25	2.49	1.60
4	<b>4</b>	200/1/0	54	1.56	1.37	1.80
5	<b>5</b>	200/1/0	76	2.19	1.94	2.11
6	<b>6</b>	200/1/0	66	1.90	1.89	1.98
7 <sup>e</sup>	<b>7</b>	200/1/0	62	0.96	1.78	1.40
8 <sup>e</sup>	<b>8</b>	200/1/0	64	1.09	1.84	1.64

<sup>a</sup> Conditions: Cat., 10 μmol, [rac-LA]<sub>0</sub> = 0.5 M, THF as solvent, 25 °C. Polymerization time was 12 h and not optimized. <sup>b</sup> Determined by <sup>1</sup>H NMR spectrum.

<sup>c</sup> *M*<sub>n,calcd</sub> = [LA]<sub>0</sub>/[Cat]<sub>0</sub> × 144.13 × conv. (%). <sup>d</sup> Determined by GPC against polystyrene standard, *M*<sub>n</sub> using a correcting factor for polylactides (0.58). <sup>e</sup>cat., 5 μmol.