

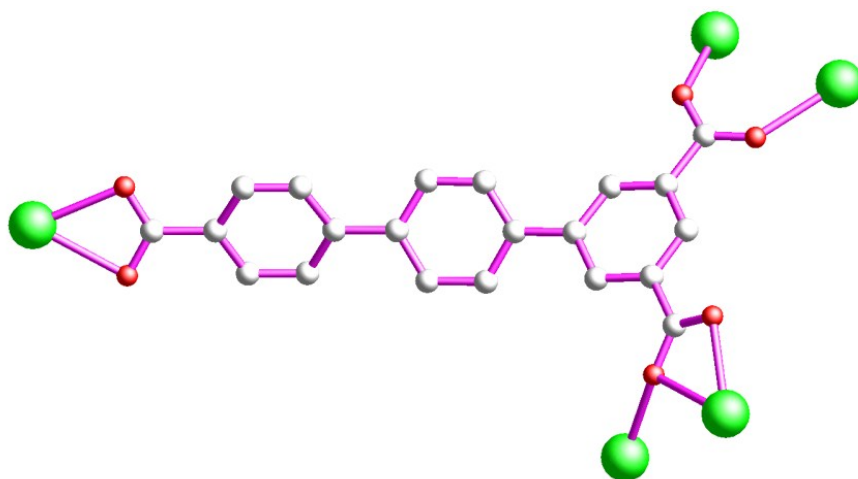
Supporting Information

Highly efficient fluorescence sensing of phosphate by dual-emissive lanthanide MOFs

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Scheme S1 The full de-protonated L^{3-} ligand adopts $\mu_7-\eta_1, \eta_1, \eta_2, \eta_1, \eta_1$ and η_1 coordination mode.

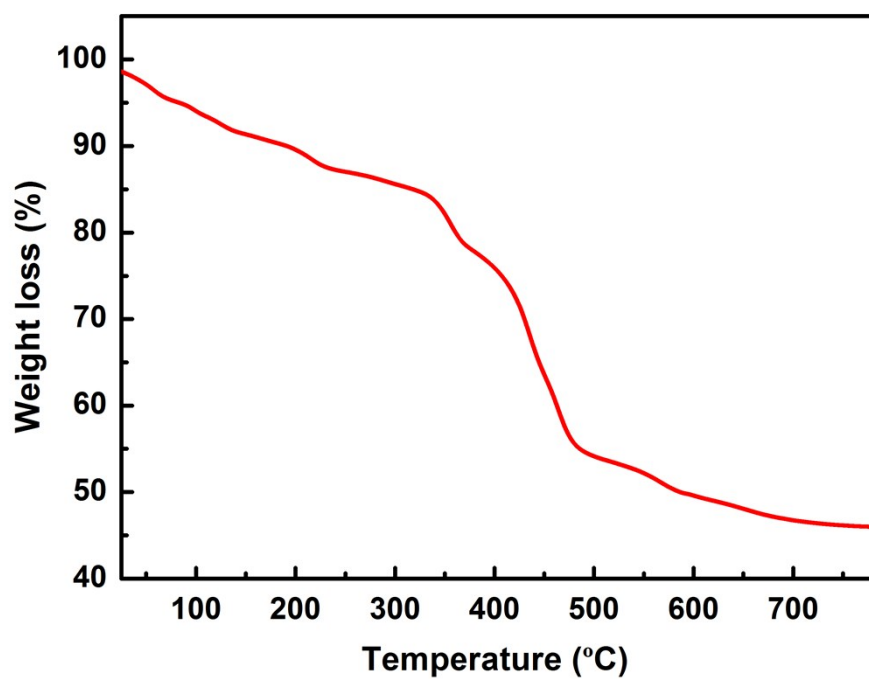


Fig. S1 TGA profile for the as-synthesized **1**.

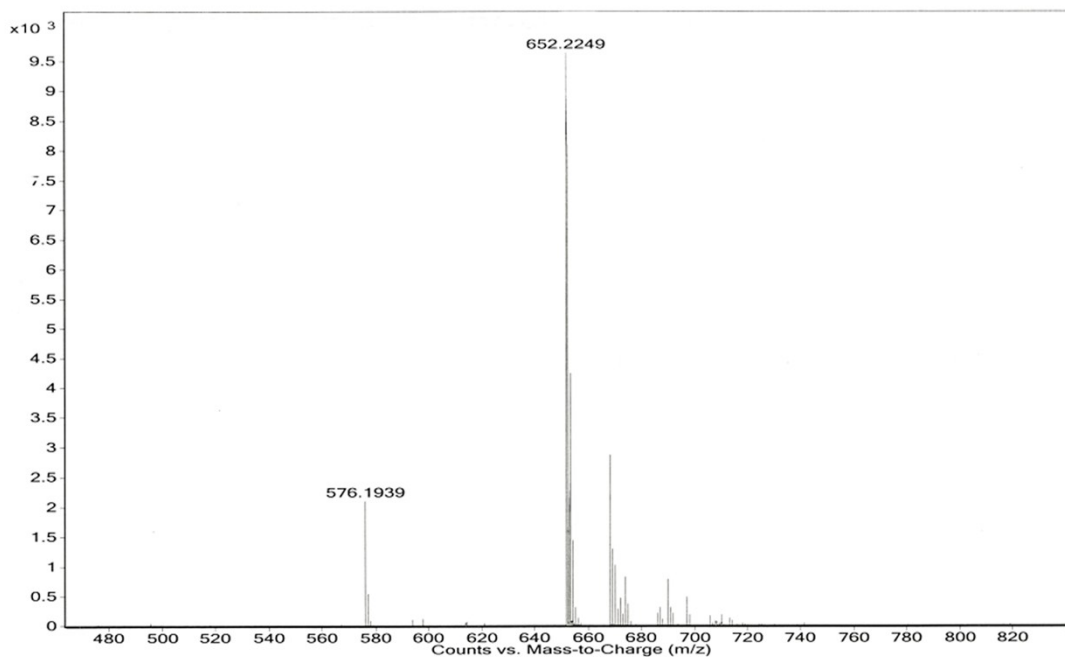


Fig. S2 High resolution mass spectrum of probe 1.

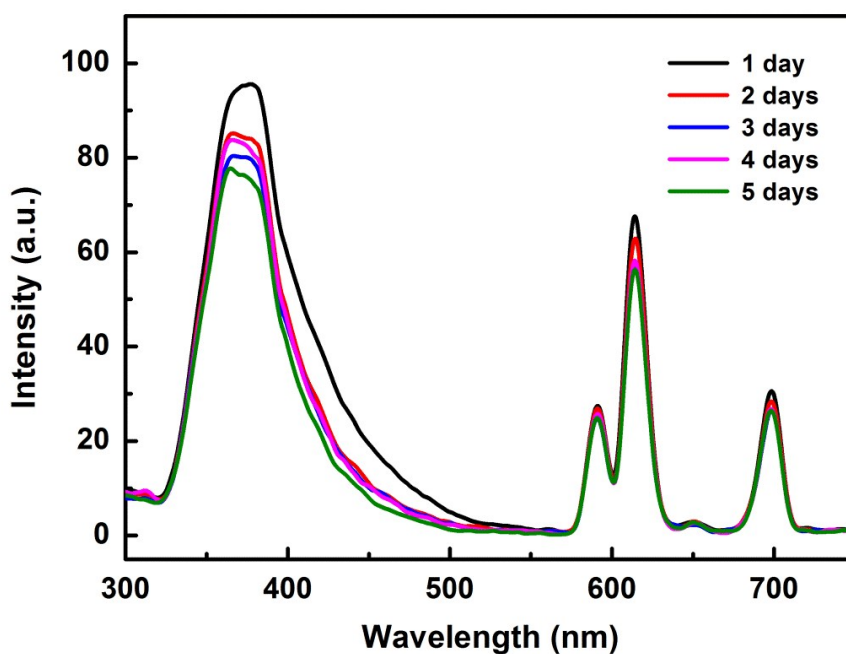


Fig. S3 Fluorescence emission spectra of **1** dispersed into water solutions as the suspensions within 5 days.

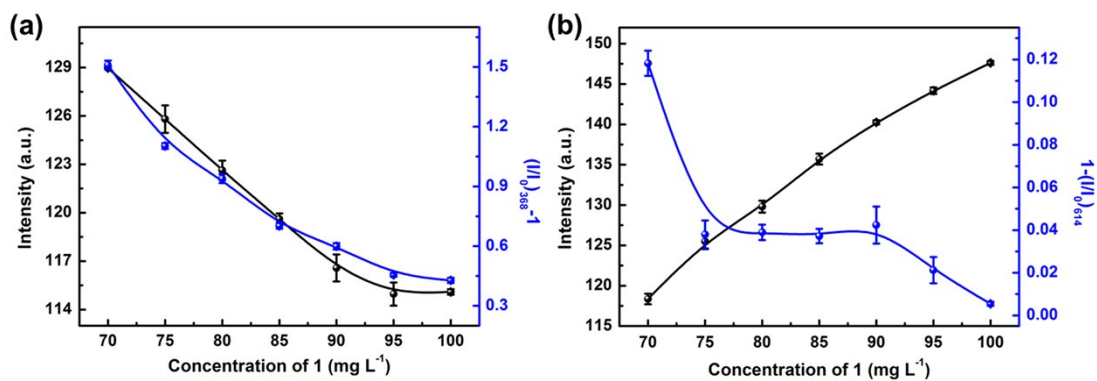


Fig. S4 (a) The effect of concentration of 1 on the fluorescence intensity and enhancing efficiency after addition of 10 μM Pi. (b) The effect of concentration of 1 on the fluorescence intensity and quenching efficiency after addition of 10 μM Pi.

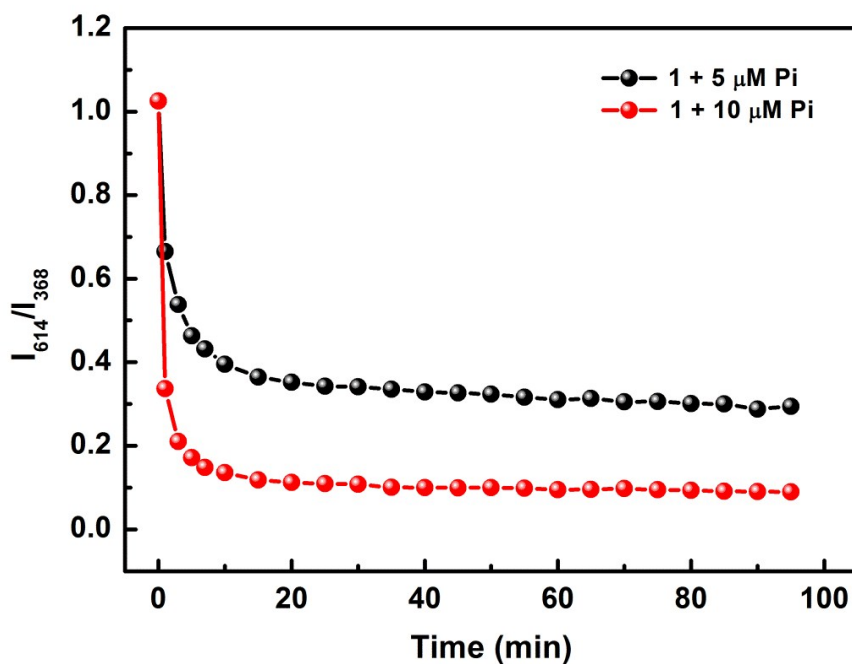


Fig. S5 Effect of reaction time of 1 on the fluorescence intensity ratio (I_{614}/I_{368}) in the presence of 5 μM (black) and 10 μM (red) Pi.

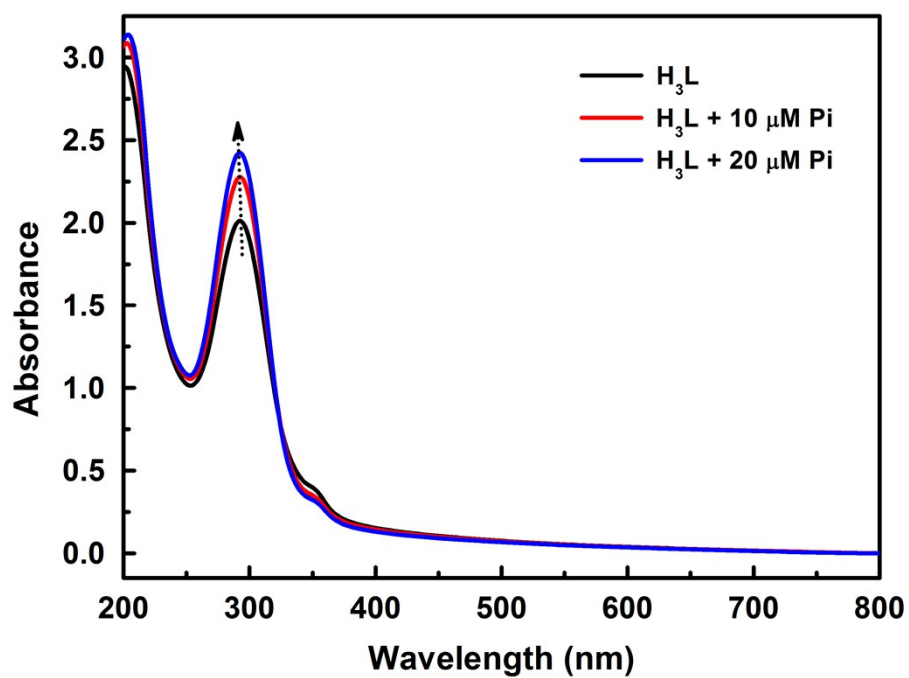


Fig. S6 UV-vis absorption spectra of free ligand H₃L and H₃L in the presence of various concentrations of Pi.

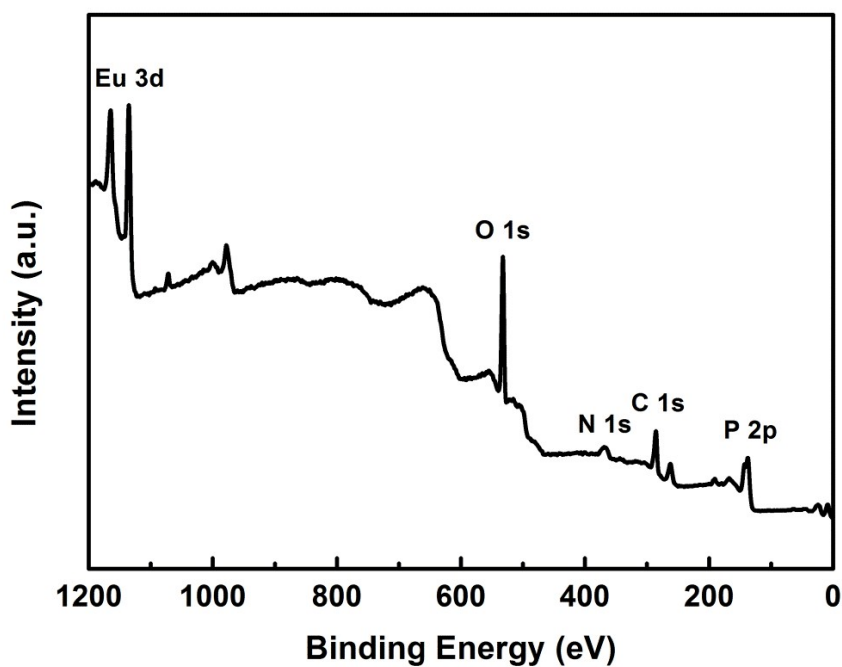


Fig. S7 Full-scan XPS spectrum of **1** after incubation with Pi.

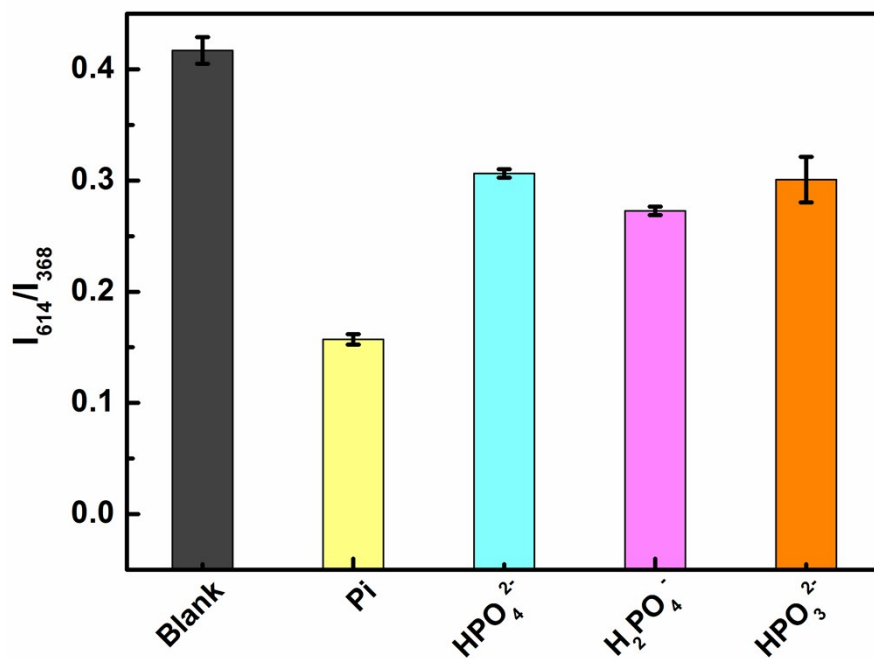


Fig. S8 Fluorescence intensity ratio (I_{614}/I_{368}) of **1** after addition of different phosphorus containing species.

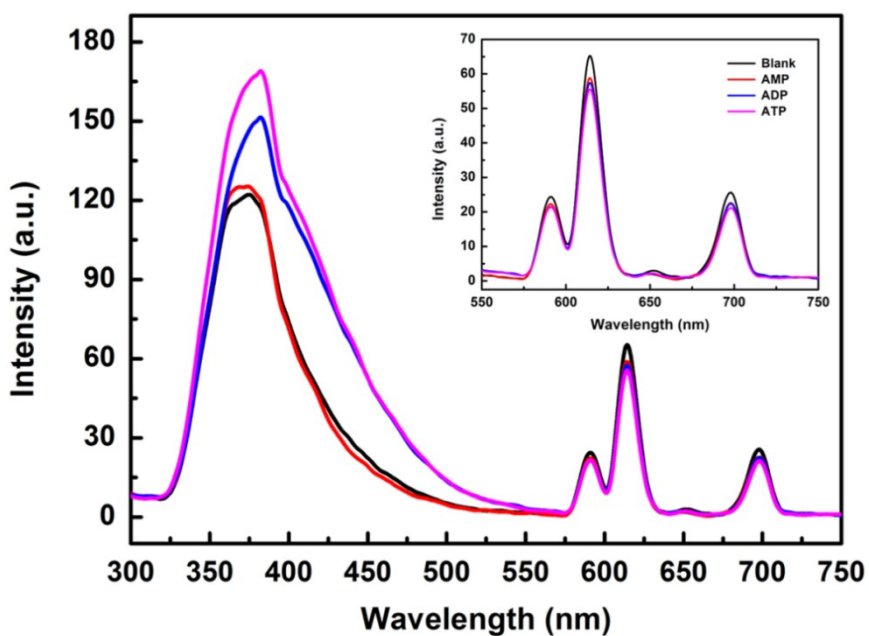


Fig. S9 Fluorescence emission spectra of **1** with 10 μM of AMP, ADP and ATP nucleoside phosphates added, respectively. Inset: The magnification of fluorescence emission spectra from 550-750 nm.

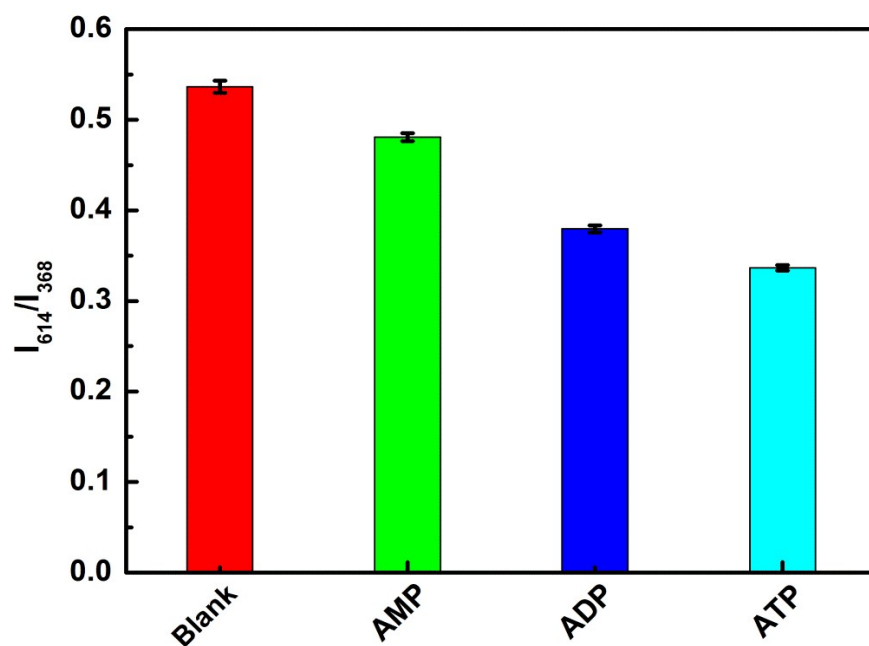


Fig. S10 Fluorescence intensity ratio (I_{614}/I_{368}) of **1** in the presence of 10 μM nucleoside phosphates including AMP, ADP and ATP.

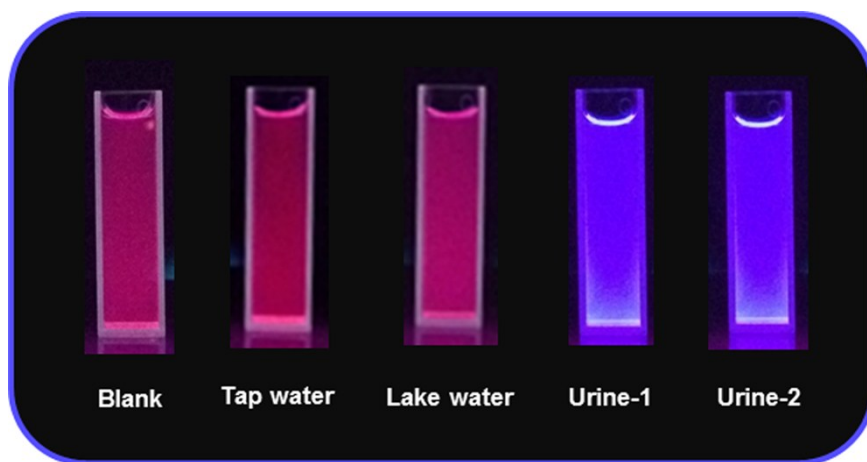


Fig. S11 Photographs of **1** with the addition of different original (undiluted) real samples under 302 nm UV light irradiation.

Table S1 Crystallographic data and details of refinements for **1**.^{a,b}

Complex	1
Empirical formula	C _{22.95} H _{16.25} EuN _{0.65} O ₈
Formula weight	581.09
Crystal system	Monoclinic
Space group	C ₂ /c
<i>a</i> /Å	33.0877(8)
<i>b</i> /Å	13.6008(4)
<i>c</i> /Å	14.1080(3)
α /°	90
β /°	102.249(2)
γ /°	90
<i>V</i> /Å ³	6204.3(3)
<i>Z</i>	8
<i>D</i> /g·cm ⁻³	1.231
μ /mm ⁻¹	14.779
<i>F</i> (000)	2300
Crystal size /mm ³	0.25 × 0.22 × 0.20
θ range for data collection /°	9.136 - 147.26
Limiting indices	-31 ≤ <i>h</i> ≤ 40, -13 ≤ <i>k</i> ≤ 13, -14 ≤ <i>l</i> ≤ 17, -16 ≤ <i>k</i> ≤ 9, -17 ≤ <i>l</i> ≤ 15
Reflections collected / unique (<i>R</i> _{int}) unique (<i>R</i> _{int})	12348 / 6102(0.0343)
Completeness	97.5%
Max. and min. transmission	0.062 and 0.052
Data / restraints / parameters	6102 / 82 / 312
Goodness-of-fit on <i>F</i> ²	1.043
Final <i>R</i> indices [<i>I</i> > 2σ(<i>I</i>)]	<i>R</i> ₁ = 0.0399, <i>wR</i> ₂ = 0.1019
<i>R</i> indices (all data)	<i>R</i> ₁ = 0.0472, <i>wR</i> ₂ = 0.1081 <i>wR</i> ₂ = 0.2069
Largest diff. peak and hole /e·Å ⁻³	1.43 / -0.77

^a $R_1 = \sum ||F_o| - |F_c|| / |F_o|$. ^b $\omega R_2 = [\sum w(|F_o|^2 - |F_c|^2)^2 / w|F_o|^2]^1/2$.

Table S2 Selected bond lengths /Å and bond angles /° for complex **1**.

Eu(1)-O(1)	2.393(3)	Eu(1)-O(3)	2.447(3)
Eu(1)-O(4)	2.368(3)	Eu(1)-O(7)	2.461(4)
Eu(1)-O(2)	2.388(3)	Eu(1)-O(5)	2.531(3)
Eu(1)-O(6)	2.435(3)	Eu(1)-O(7)	2.461(4)
O(1)- Eu (1) -O(3)	72.78(12)	O(1)- Eu (1) -O(4)	68.85(9)
O(1)- Eu (1) -O(5)	70.65(11)	O(1)- Eu (1) -O(6)	80.19(12)
O(1)- Eu (1) -O(7)	137.54(14)	O(3)- Eu (1) -O(8)	135.5(2)
O(2)- Eu (1) -O(3)	91.58(11)	O(2)- Eu (1) -O(4)	67.86(9)
O(2)- Eu (1) -O(5)	142.70(11)	O(2)- Eu (1) -O(6)	142.01(13)
O(2)- Eu (1) -O(7)	71.48(12)	O(2)- Eu (1) -O(8)	75.9(2)
O(3)- Eu (1) -O(4)	50.53(9)	O(3)- Eu (1) -O(5)	125.36(10)
O(3)- Eu (1) -O(7)	73.24(15)	O(3)- Eu (1) -O(8)	143.5(6)
O(4)- Eu (1) -O(5)	82.65(10)	O(4)- Eu (1) -O(6)	133.78(11)
O(4)- Eu (1) -O(7)	147.06(12)	O(4)- Eu (1) -O(8)	81.9(4)
O(5)- Eu (1) -O(6)	52.32(11)	O(6)- Eu (1) -O(7)	70.91(13)
O(6)- Eu (1) -O(8)	91.6(7)	O(7)- Eu (1) -O(8)	76.0(4)

Table S3 Comparison of various methods for the determination of Pi.

Analytic methods	Systems	Linear range (μM)	Detection limit (μM)	Ref.
Colorimetric method	MA-AuNPs	0.5-30	0.076	[1]
Colorimetric method	Fe_3O_4 MNPs-TMB- H_2O_2	0.2-200	0.11	[2]
Electrochemical method	Al-Cu/Si-p/SiO ₂ /Si ₃ N ₄ / Cu(II)Pc-PAA	0.0001-1	0.001	[3]
Electrochemical method	Three-electrode	1-20	0.3	[4]
Electrochemical method	Ni-BPE	40-1000	0.3	[5]
Fluorescent method Signal ^a	ZnO QDs-MOF-5 Fluorescence enhancing	0.5-12	0.053	[6]
Fluorescent method Signal	UiO-66-NH ₂ Fluorescence enhancing	5-150	1.25	[7]
Fluorescent method Signal	Mn: ZnTe/ZnSe QDs Fluorescence quenching	0.67-50	0.2	[8]
Fluorescent method Signal	{[EuL(H ₂ O) _{1.35} (DMF) _{0.65}] \cdot 1.9DMF} _n (1) Ratio fluorescence	0.1-15	0.052	This work

^aFluorescence signal changing mode.

Table S4 Analytical results (mean \pm σ , n=3) for the detection of Pi in real samples.

Samples	Spiked (μM)	Measured (μM)	Recovery (%)	R.S.D. (n=3, %)
Tap water	0	3.83 ± 0.35		
	5	8.86 ± 0.04	100.6	1.7
Lake water	0	9.91 ± 0.01		
	5	14.55 ± 0.03	92.8	2.3
Urine-1 ^a	0	10.06 ± 0.01		
	5	14.95 ± 0.13	97.8	4.3
Urine-2 ^a	0	5.28 ± 0.01		
	5	10.25 ± 0.06	99.4	1.4

^aDiluted urine sample.

Notes and references

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