

SUPPORTING INFORMATION FOR:

Versatile organoaluminium catalysts based on heteroscorpionate ligands for the preparation of polyesters

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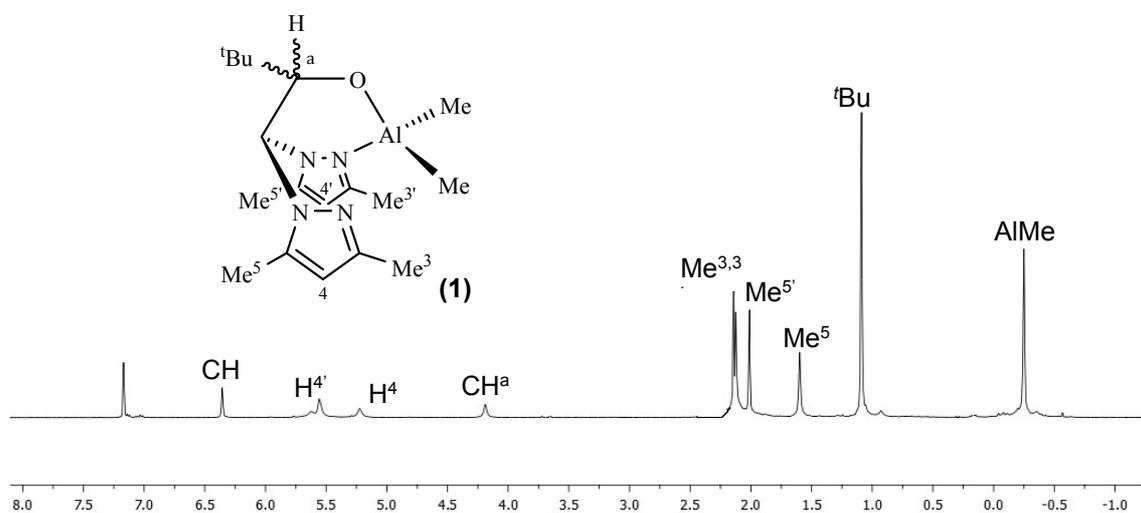


Figure S1. $^1\text{H-NMR}$ Spectrum for complex **1** in C_6D_6 .

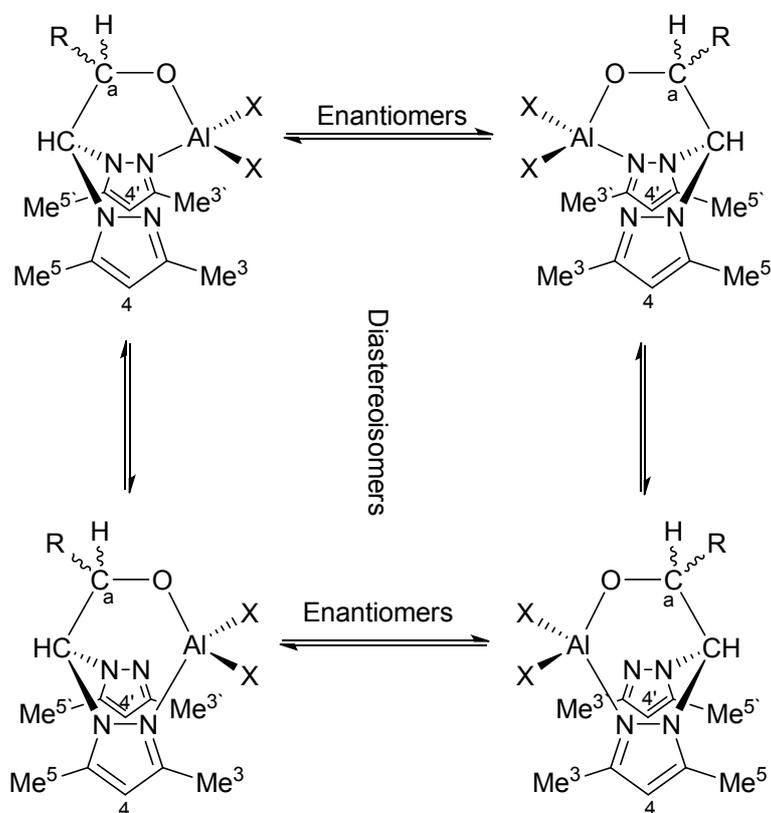


Figure S2. Fluxional behaviour as a result of a slow exchange process between the coordinated and non-coordinated pyrazole rings for complexes **1-4**.

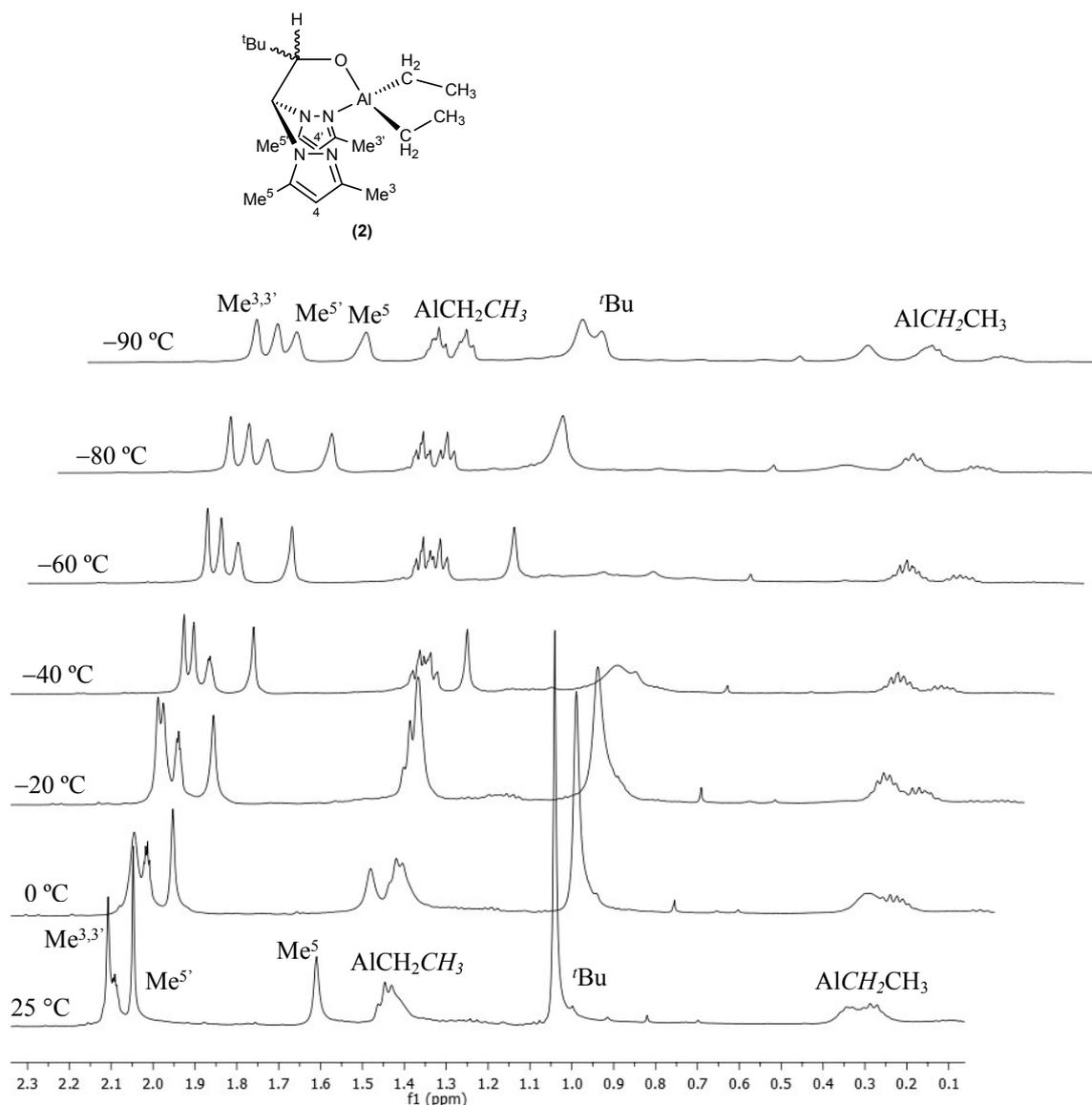


Figure S3. VT-NMR study in the region from 2.4 to 0.0 ppm for compound **2** in toluene- d_8 . The VT NMR analysis showed that the resonances of the pyrazole rings and the alkyl groups broaden and become resolved indicating the presence of two diastereoisomers at -90 °C. Therefore, there is an exchange process between the coordinated and the non-coordinated pyrazole rings, involving an interconversion from one diastereoisomer to the other. It is worth mentioning that the methine carbon bridging the two pyrazole rings is a stereocentre due to the coordination mode of heteroscorpionate ligand.

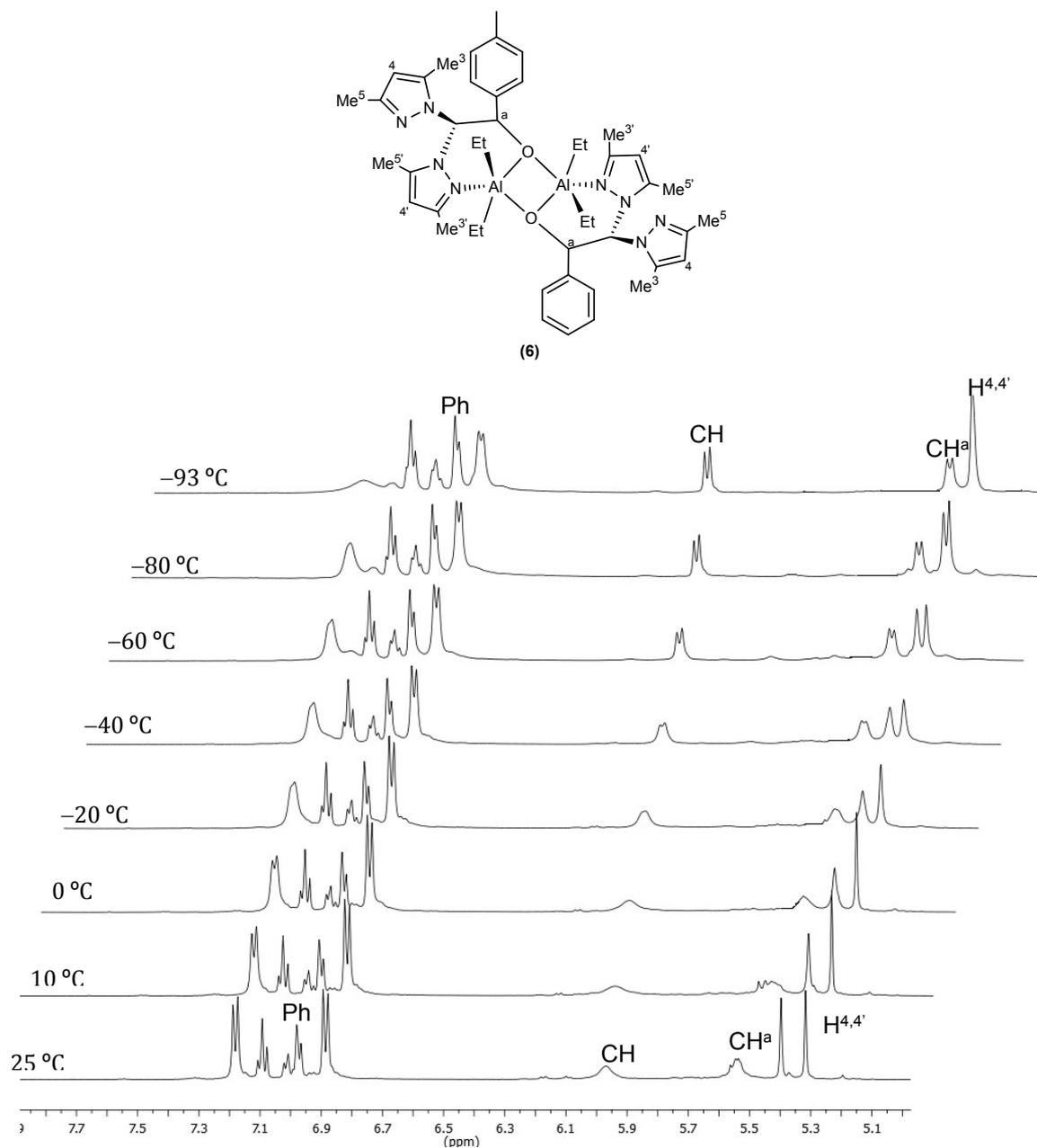


Figure S4. VT-NMR study in the region from 8.0 to 5.1 ppm for compound **6** in toluene- d_8 . The VT NMR analysis showed that the resonances of the pyrazole rings become resolved indicating the presence of ondiastereoisomers at $-90\text{ }^\circ\text{C}$. Therefore, there is not an exchange process between the coordinated and the non-coordinated pyrazole rings.

Table S1. Crystal data and structure refinement for compounds **1** and **5**

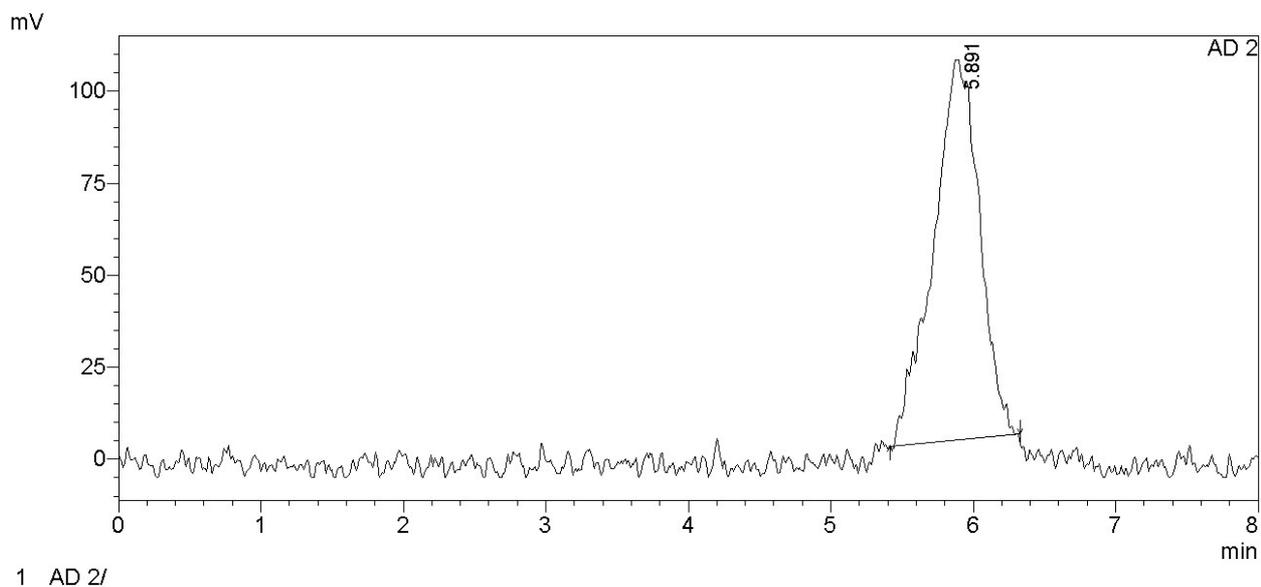
	1	5
Empirical formula	C ₁₈ H ₃₁ Al N ₄ O	C ₄₉ H ₆₆ Al ₂ N ₈ O ₂
Formula weight	346.45	853.05
Temperature (K)	230(2)	100(2)
Wavelength (Å)	0.71073	0.71073
Crystal system	Triclinic	Triclinic
Space group	P $\bar{1}$	P $\bar{1}$
a(Å)	8.913(5)	10.3817(6)
b(Å)	9.995(6)	10.7672(7)
c(Å)	12.184(8)	10.8726(7)
α (°)	93.893(13)	102.164(2)
β (°)	97.531(13)	95.020(2)
γ (°)	101.566(16)	91.554(2)
Volume(Å ³)	1049.2(11)	1182.22(13)
Z	2	1
Density (calculated) (g/cm ³)	1.097	1.198
Absorption coefficient (mm ⁻¹)	0.108	0.109
F(000)	376	458
Crystal size (mm ³)	0.23 x 0.17 x 0.12	0.20 x 0.20 x 0.14
Index ranges	-11 ≤ h ≤ 9 -7 ≤ k ≤ 12 -15 ≤ l ≤ 15	-12 ≤ h ≤ 12 -13 ≤ k ≤ 13 -13 ≤ l ≤ 13
Reflections collected	6894	15102
Independent reflections	4512 [R(int) = 0.0227]	4644 [R(int) = 0.0493]
Data / restraints / parameters	4512 / 0 / 226	4644 / 0 / 315
Goodness-of-fit on F ²	1.063	1.038
Final R indices [<i>I</i> > 2σ(<i>I</i>)]	R1 = 0.0559, wR2 = 0.1363	R1 = 0.0418, wR2 = 0.0936
Largest diff. peak / hole, e.Å ⁻³	0.208 / -0.175	0.349 / -0.214

^a $R = \sum ||F_o| - |F_c| / \sum |F_o|$. ^b $wR = \{ \sum w(F_o^2 - F_c^2)^2 / \sum w(F_o^2) \}^{1/2}$. ^c $GOF = \{ \sum [w((F_o^2 - F_c^2)^2) / (n-p)] \}^{1/2}$, where *n* = number of reflections and *p* = total number of parameters refined.

Table S2. Bond distances (Å) and angles (°) for compound **1** and **5**

Bond Distances			
1		5	
Al(1)-O(1)	1.7575(19)	Al(1)-O(1)	1.8388(12)
Al(1)-C(17)	1.964(3)	Al(1)-C(1)	1.9883(18)
Al(1)-C(18)	1.970(3)	Al(1)-C(2)	1.9841(17)
Al(1)-N(1)	2.004(2)	Al(1)-N(1)	2.1436(15)
		Al(1)-O(1A) ^a	1.9936(12)
Bond Angles			
O(1)-Al(1)-C(17)	112.55(11)	O(1)-Al(1)-C(2)	129.92(7)
O(1)-Al(1)-C(18)	114.13(10)	O(1)-Al(1)-C(1)	114.48(7)
C(17)-Al(1)-C(18)	118.62(12)	C(2)-Al(1)-C(1)	115.43(8)
O(1)-Al(1)-N(1)	92.25(7)	O(1)-Al(1)-O(1A) ^a	74.44(5)
C(17)-Al(1)-N(1)	108.80(11)	C(2)-Al(1)-O(1A) ^a	95.34(6)
C(18)-Al(1)-N(1)	106.86(11)	C(1)-Al(1)-O(1A) ^a	97.49(6)
		O(1)-Al(1)-N(1)	86.37(5)
		C(2)-Al(1)-N(1)	92.39(7)
		C(1)-Al(1)-N(1)	95.89(7)
		O(1A) ^a -Al(1)-N(1)	159.89(6)

^a Symmetry transformations used to generate equivalent atoms: -x+1,-y+1,-z+1



GPC Summary

Chromatogram AD 2						
#	Title	Mn	Mw	Mz	Mz1	Mv
1	40.lcd	18797	19983	21746	24287	0
	Average	18797	19983	21746	24287	0
	%RSD	0.000	0.000	0.000	0.000	0.000
	Maximum	18797	19983	21746	24287	0
	Minimum	18797	19983	21746	24287	0
	SD	0	0	0	0	0

Figure S5. GPC profile of PCL with $M_n = 18797$ Da and $M_w/M_n = 1.06$ (entry 6 in Table 1)

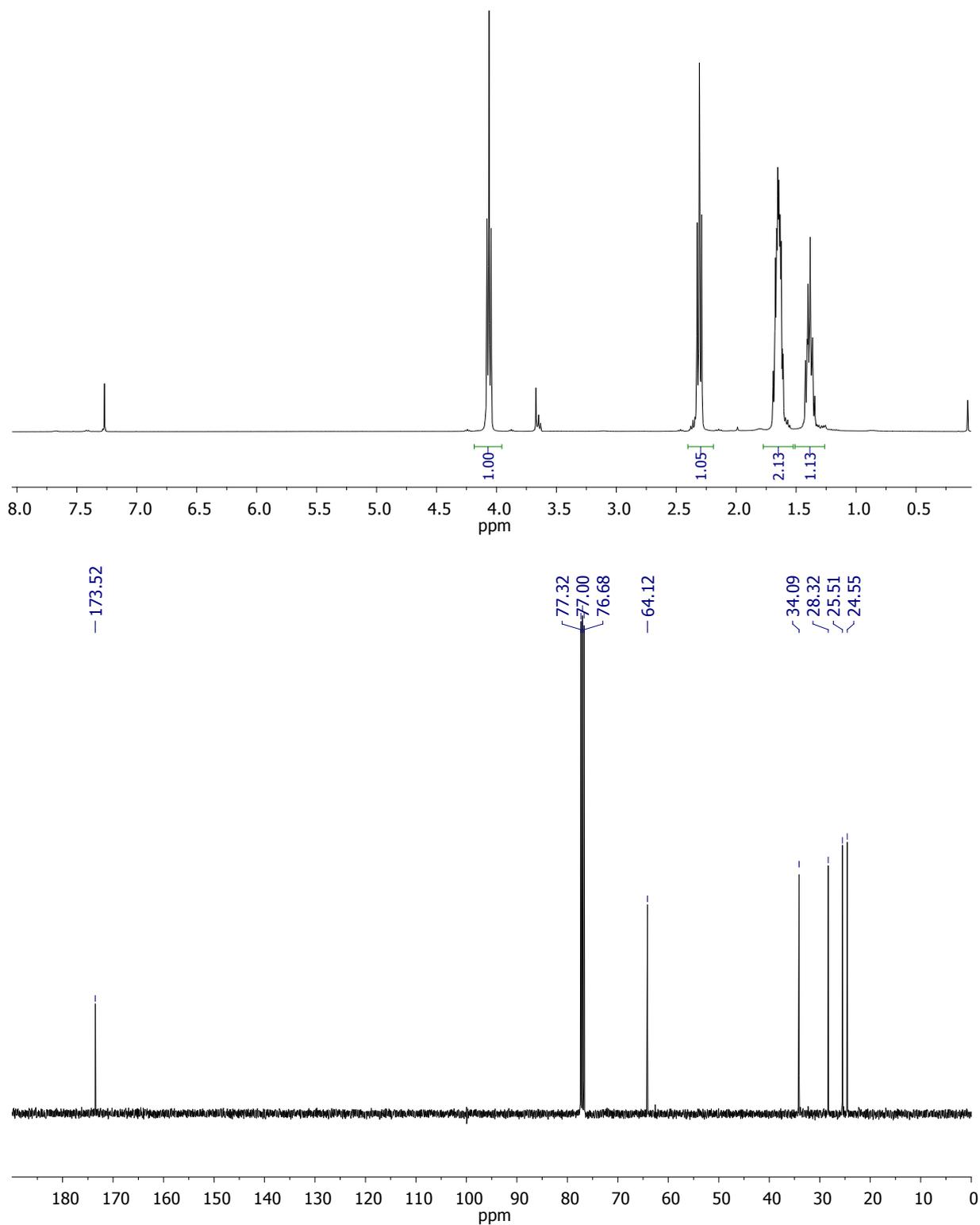


Figure S6. NMR spectra of PCL with $M_n = 18797$ Da and $M_w/M_n = 1.06$ prepared with complex **6**

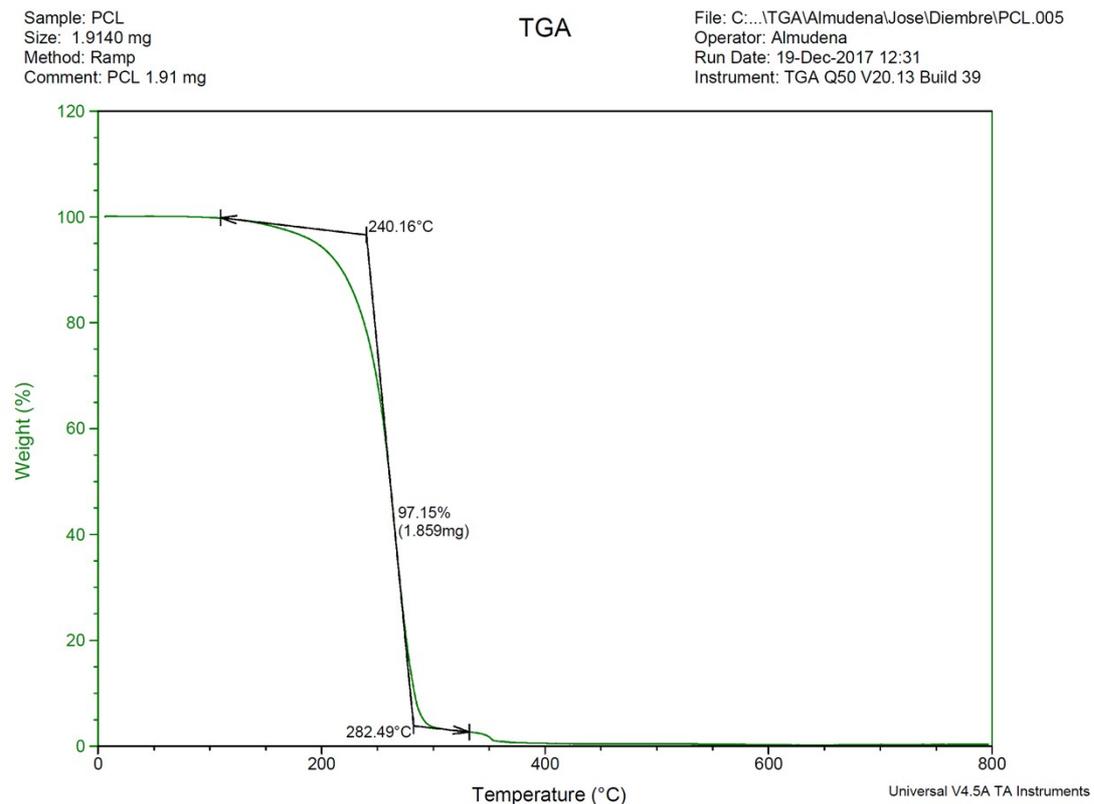


Figure S7. TGA analysis of PCL with $M_n = 18797$ Da and $M_w/M_n = 1.06$ prepared with complex **6**

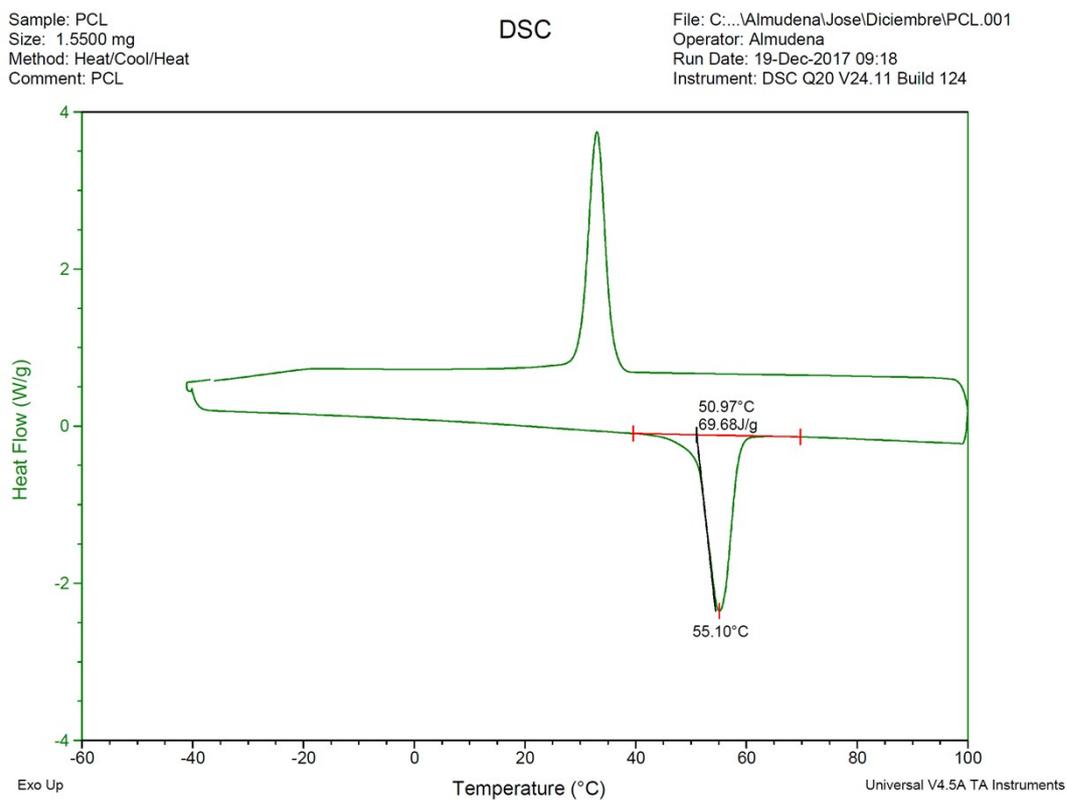
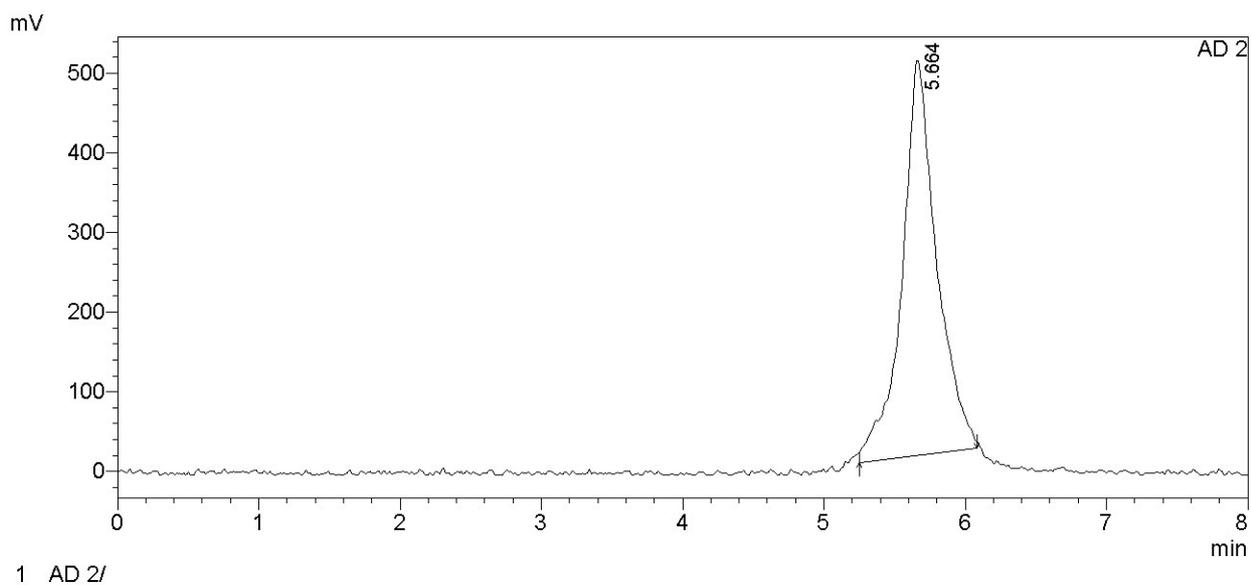


Figure S8. DSC analysis of PCL with $M_n = 18797$ Da and $M_w/M_n = 1.06$ prepared with complex **6**



GPC Summary

Chromatogram AD 2

#	Title	Mn	Mw	Mz	Mz1	Mv
1	41.lcd	25869	29550	36399	48564	0
	Average	25869	29550	36399	48564	0
	%RSD	0.000	0.000	0.000	0.000	0.000
	Maximum	25869	29550	36399	48564	0
	Minimum	25869	29550	36399	48564	0
	SD	0	0	0	0	0

Figure S9. GPC profile of PLA with $M_n = 25869$ Da and $M_w/M_n = 1.14$ (entry 2 in Table 2)

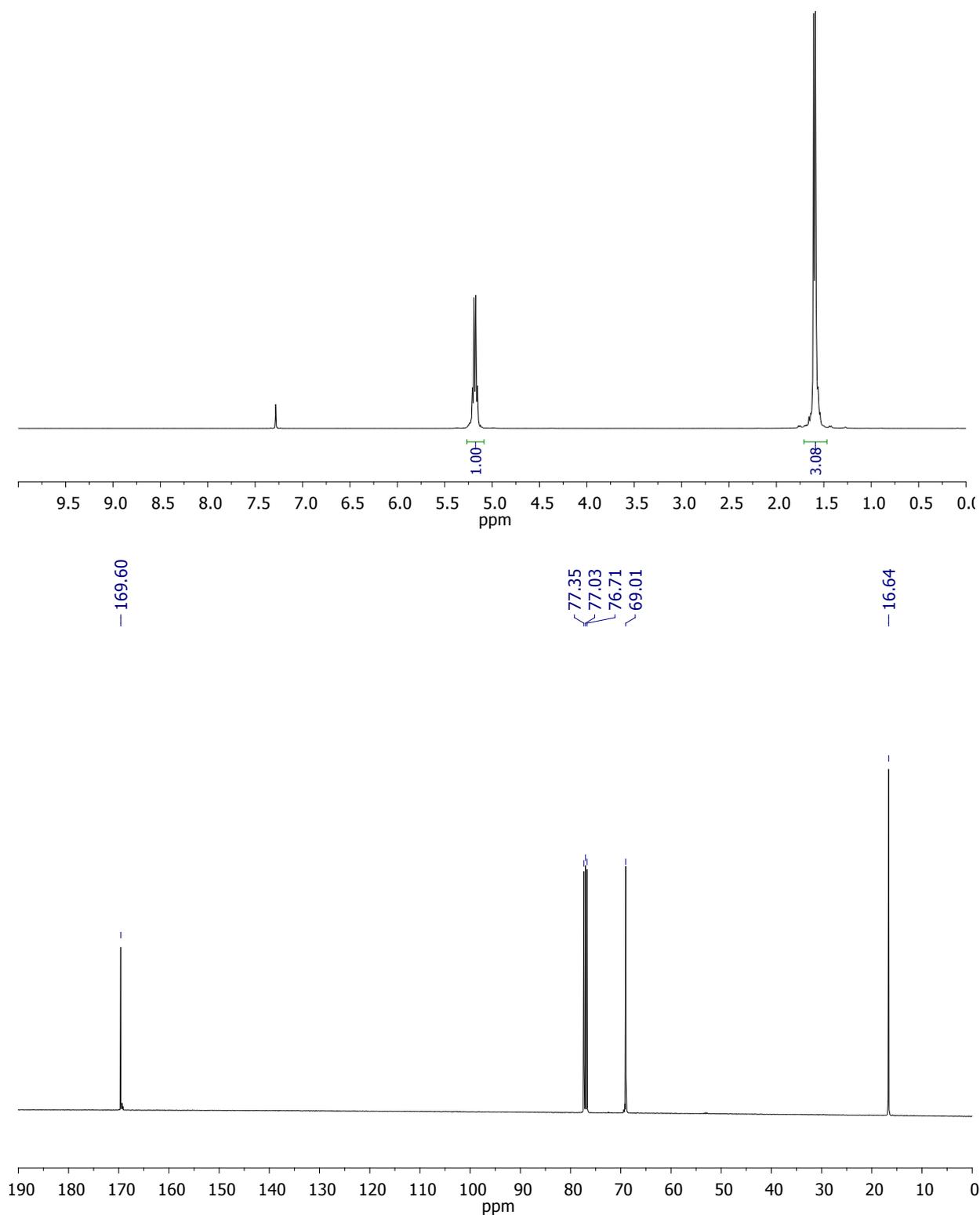


Figure S10. NMR spectra of PLA with $M_n = 25869$ Da and $M_w/M_n = 1.14$ prepared with complex **6**

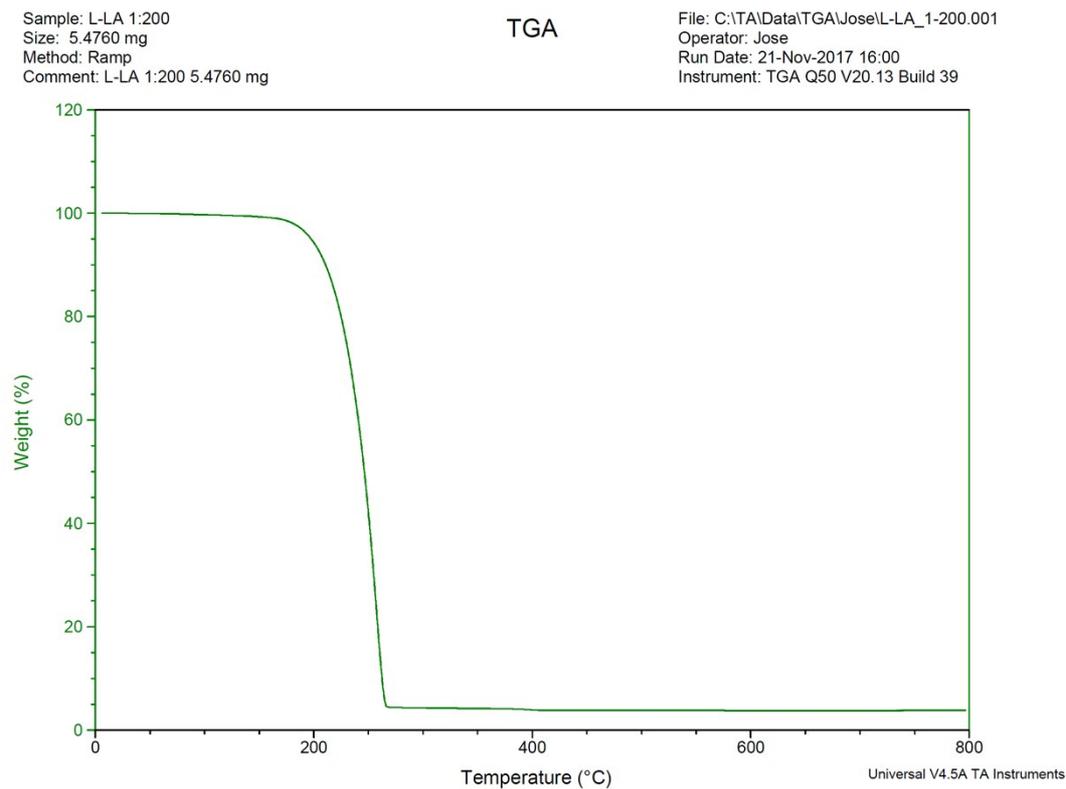


Figure S11. TGA analysis of PLA with $M_n = 25869$ Da and $M_w/M_n = 1.14$ prepared with complex **6**

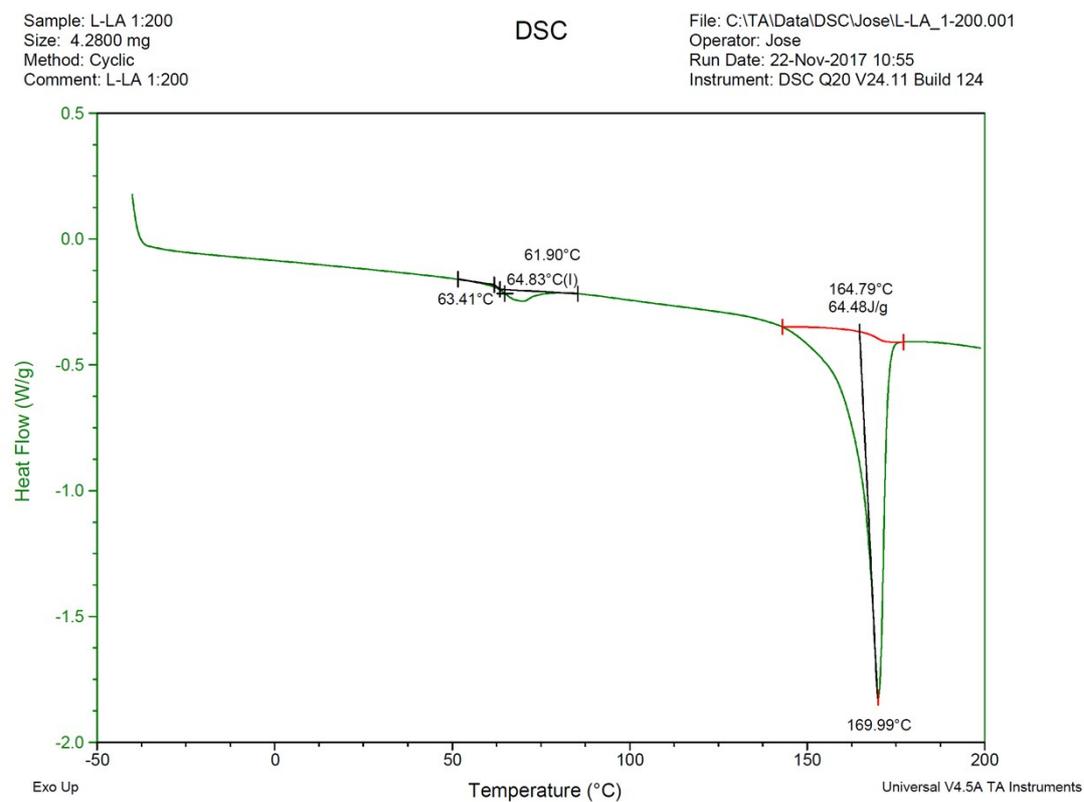
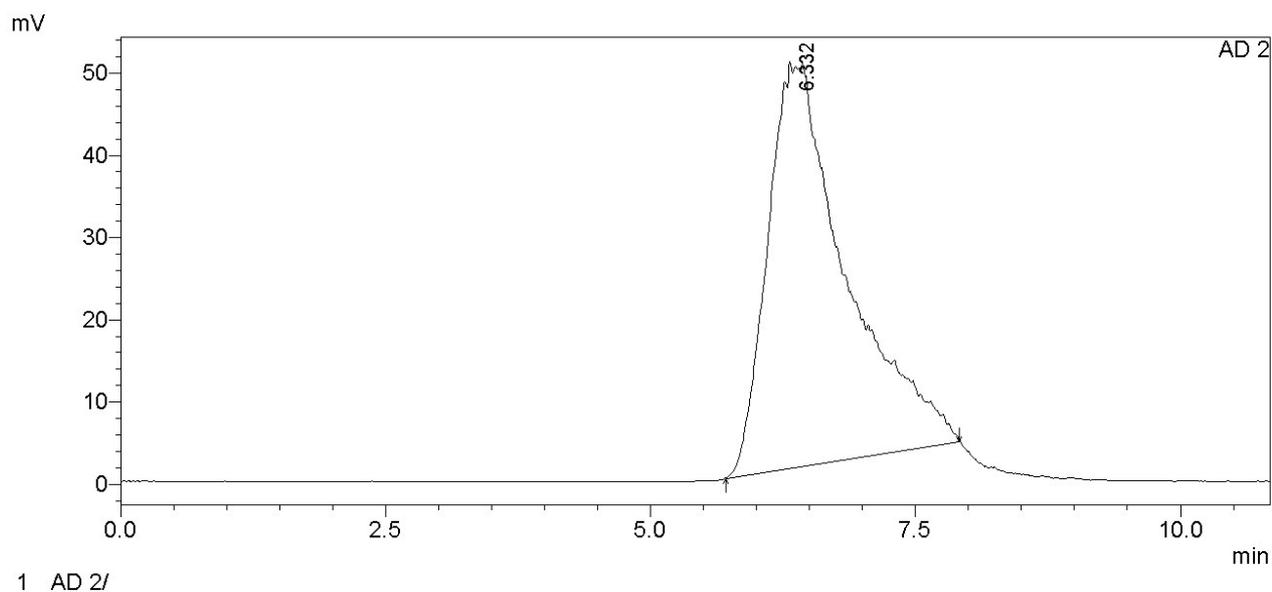


Figure S12. DSC analysis of PLA with $M_n = 25869$ Da and $M_w/M_n = 1.14$ prepared with complex **6**



GPC Summary

Chromatogram AD 2						
#	Title	Mn	Mw	Mz	Mz1	Mv
1	75.lcd	14744	15323	16172	17489	0
	Average	14744	15323	16172	17489	0
	%RSD	0.000	0.000	0.000	0.000	0.000
	Maximum	14744	15323	16172	17489	0
	Minimum	14744	15323	16172	17489	0
	SD	0	0	0	0	0

Figure S13. GPC profile of copolymer of cyclohexene oxide and phthalic anhydride with $M_n = 14744$ Da and $M_w/M_n = 1.04$ (entry 4 in Table 4)

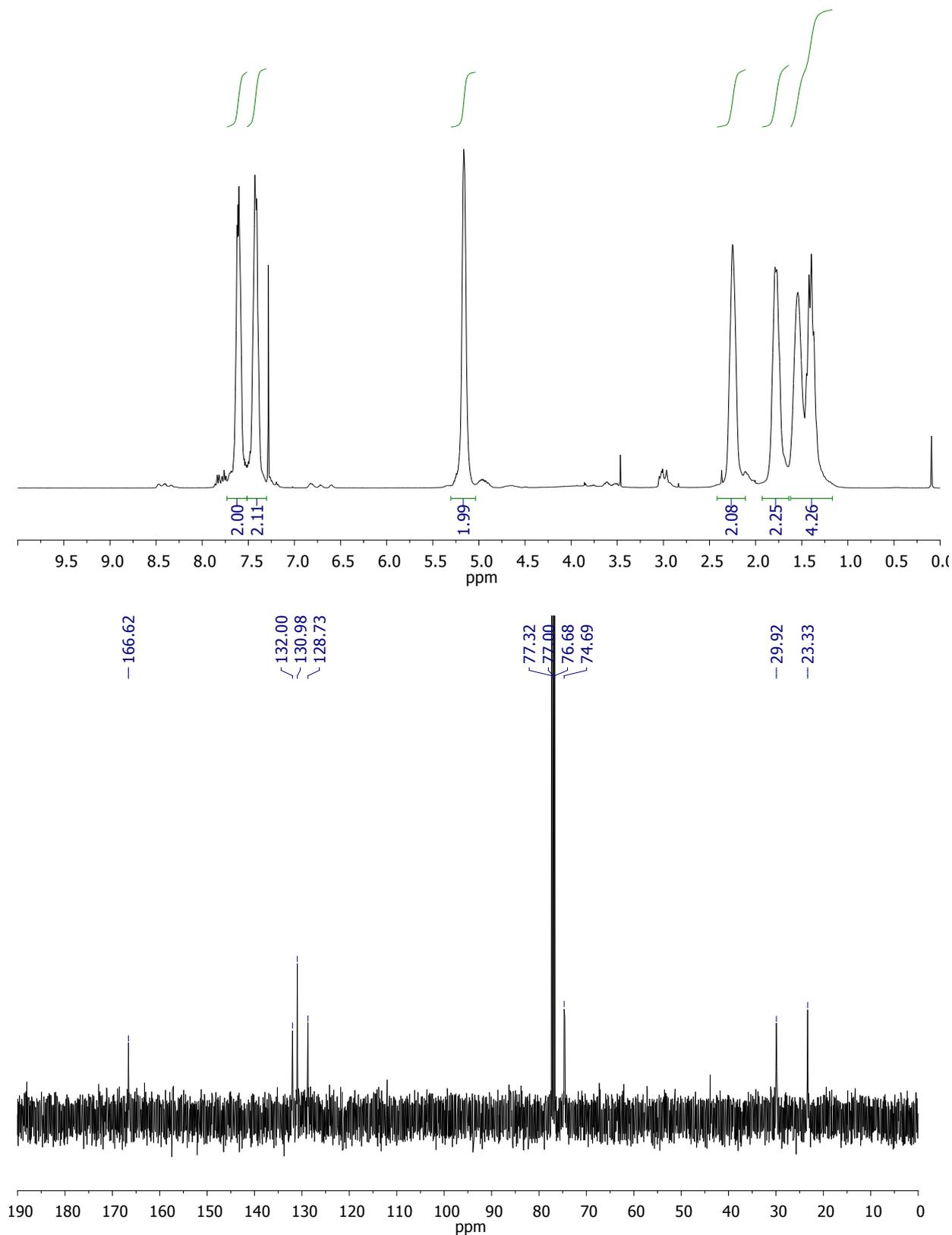


Figure S14. NMR spectra for copolymer of cyclohexene oxide and phthalic anhydride with $M_n = 14744$ Da and $M_w/M_n = 1.04$ prepared with complex **6**

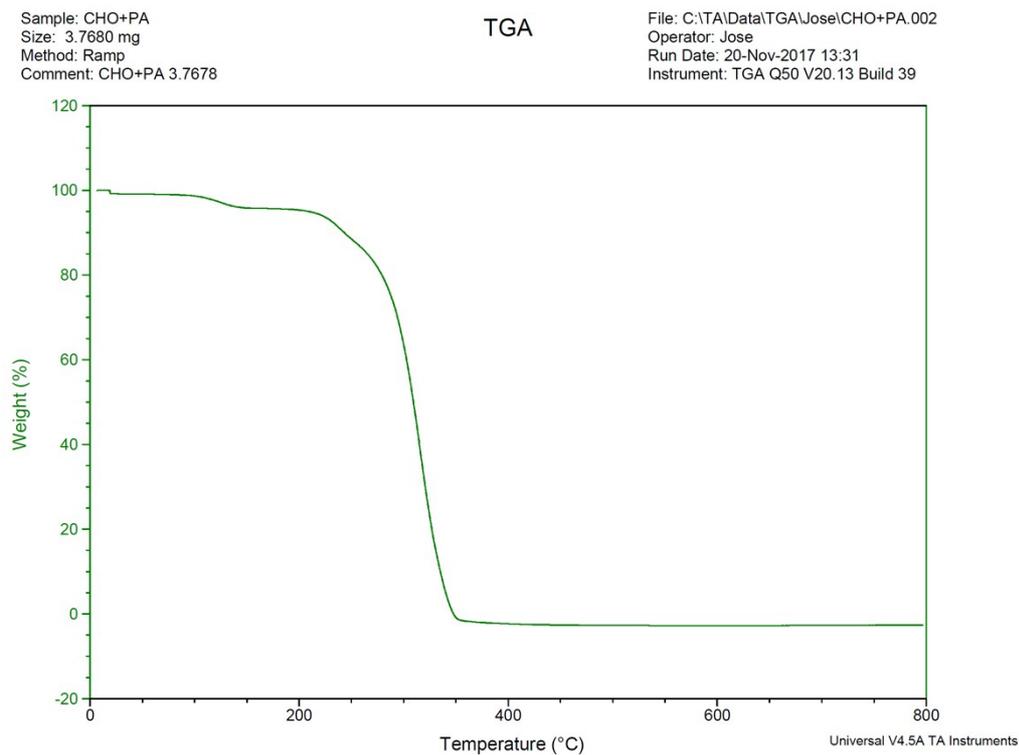


Figure S15. TGA analysis for copolymer of cyclohexene oxide and phthalic anhydride with $M_n = 14744$ Da and $M_w/M_n = 1.04$ prepared with complex **6**

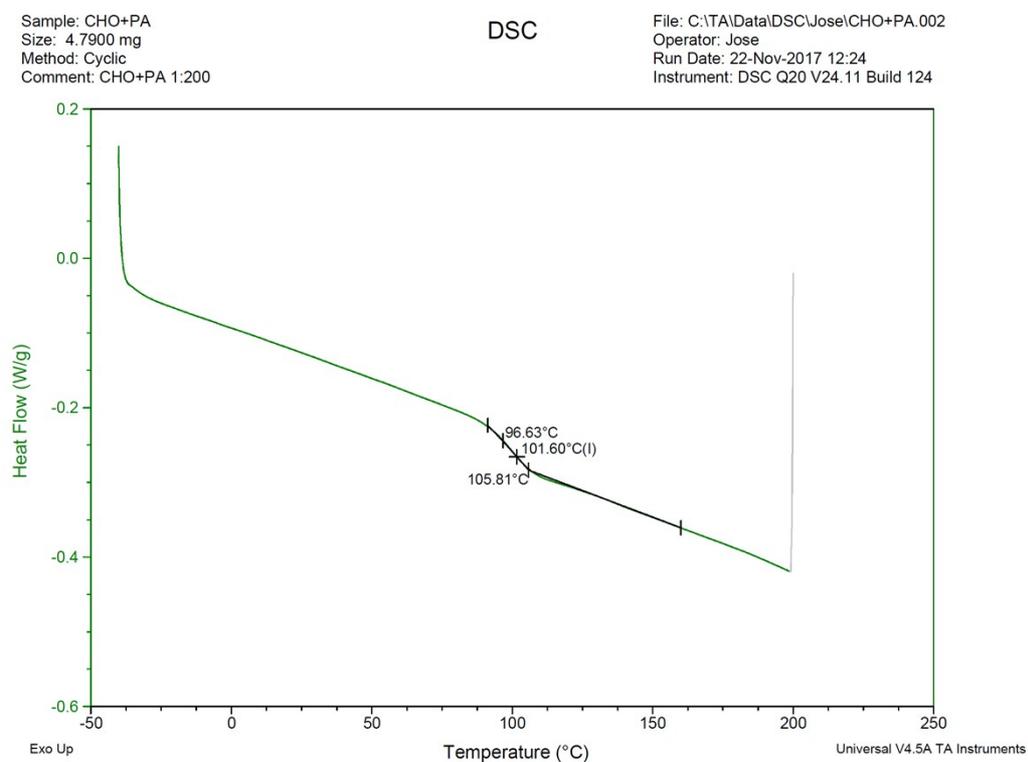


Figure S16. DSC analysis for copolymer of cyclohexene oxide and phthalic anhydride with $M_n = 14744$ Da and $M_w/M_n = 1.04$ prepared with complex **6**