

Electronic Supplementary Information

A facile synthesis for BeCl₂, BeBr₂ and BeI₂

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Powder X-ray diffraction

Powder X-ray diffraction was performed with a STOE StadiMP powder diffractometer using Cu-K α_1 radiation, a Ge monochromator and a Mythen1K detector. Samples were prepared under glovebox atmosphere in Lindemann capillaries of fused silica, which were flame sealed.

RIETVELD refinement

Phase analysis was performed with WinXPOW,¹ profiles were fitted with JANA2006.² RIETVELD refinement was carried out with JANA2006 using single crystal data from the literature.^{3–5}

IR spectroscopy

The IR spectra were recorded on a BRUKER Alpha II spectrometer with diamond ATR module inside an argon filled glovebox. Processing of the spectra was performed with the OPUS⁶ software package and OriginPro8.⁷

Raman spectroscopy

For the Raman spectroscopic measurements a small amount of the respective beryllium halide was collected and flames sealed in boro silicate capillaries with an inner diameter of 0.5 mm. Raman spectra were recorded on a RENISHAW inVia Qontor spectrometer at laser wavelengths of 457 nm, 532 nm, 633 nm and 785 nm and on a RENISHAW inVia Raman microscope at a laser wavelength of 785 nm

DFT calculations

All DFT calculations were carried out with the CRYSTAL14 program package using the PBE+D2 method. The k -mesh sampling was 8 x 8 x 8 for all studied compounds and the convergence criterion was set to 10^{-8} Hartree. The applied Basis sets were taken from^{8–11}. For the evaluation of the Coulomb and Exchange integrals (TOLINTEG), tolerance factors of 8 8 8 8 and 16 were used. After all full structural optimization including all lattice and atomic site parameters, the IR and Raman active vibrational modes including intensities were calculated. No imaginary frequencies were obtained.

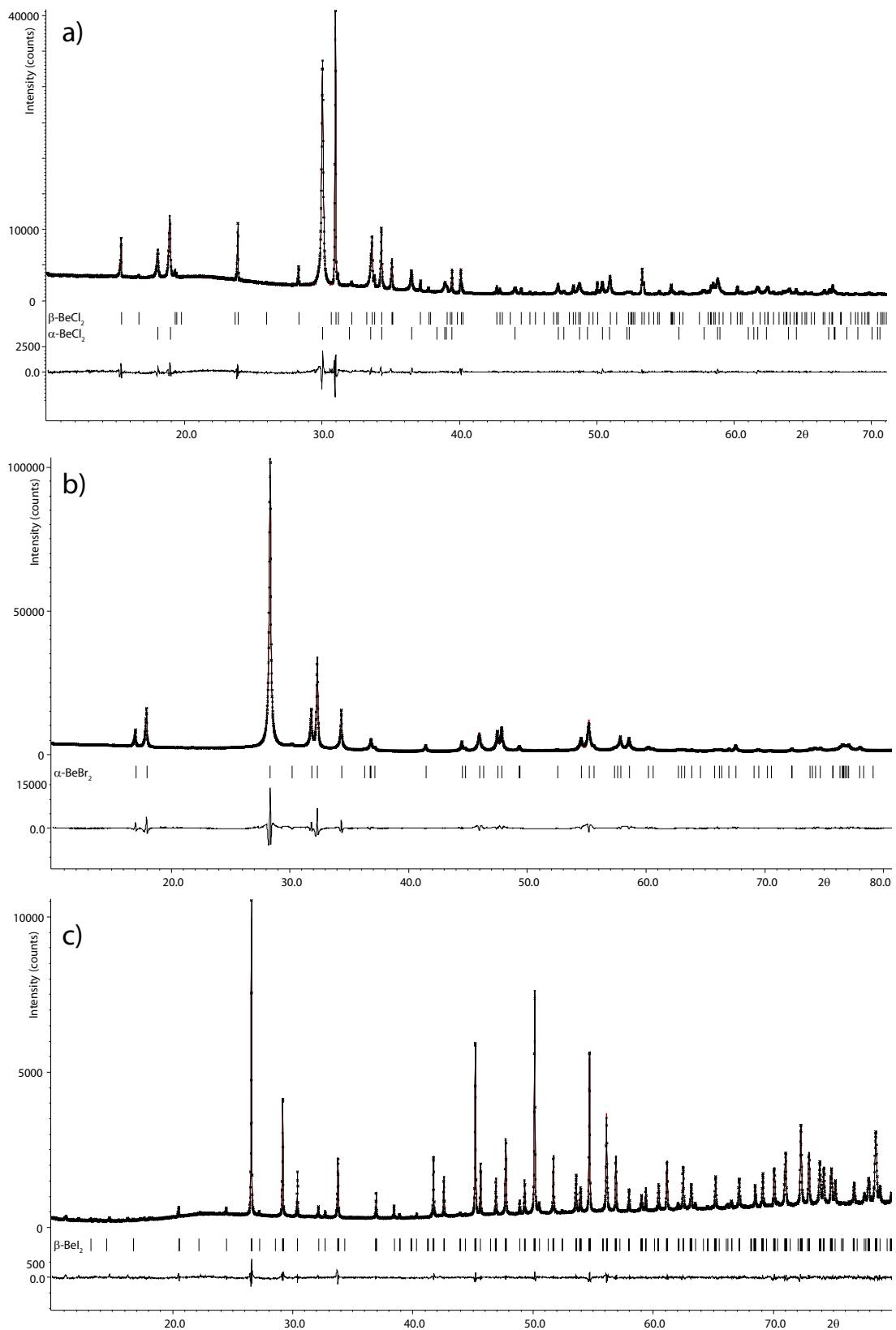


Fig. 1 X-ray powder diffractograms with LeBail profile fit, difference plot and related reference for a) $\text{BeCl}_2^{3,4}$, b) BeBr_2^5 and c) Bel_2^5 .



Fig. 2 BeCl₂ prior to fractionated sublimation.



Fig. 3 Unknown lint in the lower part of the ampoule in which BeCl₂ was synthesised.



Fig. 4 BeBr₂ crystals after sublimation.

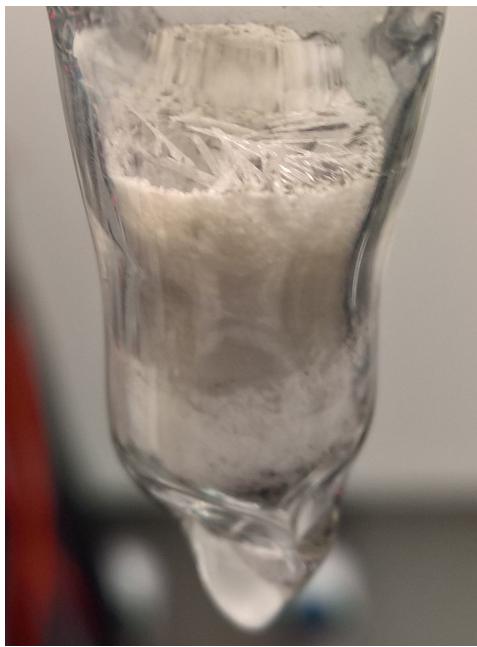


Fig. 5 Crystals of BeI_2 at the flame sealed end of the reaction ampule.

Table 1 Experimental and calculated IR and Raman data for BeF_2 , BeCl_2 , BeBr_2 and BeI_2 .

$\alpha-\beta-\text{BeF}_2$		$\alpha-\beta-\text{BeCl}_2$		$\alpha-\text{BeBr}_2$		$\beta-\text{BeI}_2$		
		IR data / cm^{-1} ; experiment (calculated) [mode]						
403	(β : 380) [E_1]	450	(α : 449) [B_{1u}]	404	(400) [B_{1u}]	450	(462) [E_u]	
	(α : 388) [E]		(β : 451) [A_{2u}]	416		516		
	(β : 393) [A_2]		(β : 462) [E_u]	462	(492) [B_{2u}]	581		
	(α : 412) [A_2]	583	(α : 560) [B_{3u}]	477	(505) [B_{3u}]			
	(α : 745) [A_2]		(β : 563) [Eu]					
	(α : 747) [E]		(β : 563) [Eu]					
	(β : 754) [E_1]		(β : 569) [A_{2u}]					
	(β : 756) [A_2]		(α : 572) [B_{2u}]					
Raman data / cm^{-1} ; experiment (calculated) [mode]								
734	174	(α : 169) [B_{3g}]	94	(99) [A_g]	76	(66) [A_{1g}]		
		(α : 175) [B_{2g}]	112	(111) [B_{2g}]		(68) [B_{1g}]		
	290	(β : 292) [A_{1g}]		(114) [B_{3g}]		(72) [E_g]		
	337	(α : 334) [Ag]	168	(175) [B_{1g}]		(74) [B_{2g}]		
	410	(α : 385) [B_{3g}]	203	(203) [A_g]		(76) [A_{1g}]		
	416	(α : 388) [B_{2g}]	374	(362) [B_{3g}]	92	(85) [E_g]		
			585	(606) [B_{1g}]		(88) [B_{2g}]		
						(92) [E_g]		
					109	(102) [E_g]		
						(103) [B_{1g}]		
						(109) [B_{2g}]		
						(111) [E_g]		
						(112) [E_g]		
						(118) [A_{1g}]		
						(132) [B_{1g}]		
						(136) [E_g]		
						(143) [A_{1g}]		
						(147) [E_g]		
140						(148) [B_{2g}]		
						368	(350) [B_{2g}]	
							(351) [E_g]	
							430	(416) [E_g]
							447	(441) [B_{2g}]
							473	(485) [A_{1g}]
								(486) [B_{2g}]
								(491) [B_{1g}]

Table 2 Calculated IR and Raman vibrational modes for BeF₂.

α -BeF ₂ , IR active modes		
wavenumber / cm ⁻¹	irreducible representation	intensity / km mol ⁻¹
90	E	0
187	E	0
280	A ₂	16
302	E	3
388	E	303
412	A ₂	181
596	E	3
745	A ₂	1354
747	E	2645
793	E	3
831	A ₂	6
866	E	53

α -BeF ₂ , Raman active modes		
wavenumber / cm ⁻¹	irreducible representation	intensity / a.u.
90	E	91
130	A ₁	262
187	E	47
302	E	78
317	A ₁	1000
355	A ₁	621
388	E	172
596	E	62
746	A ₁	33
747	E	42
793	E	3
866	E	31

β -BeF ₂ , IR active modes		
wavenumber / cm ⁻¹	irreducible representation	intensity / km mol ⁻¹
66	E ₁	0
380	E ₁	321
393	A ₂	228
754	E ₁	2714
756	A ₂	1352
792	E ₁	0

β -BeF ₂ , Raman active modes		
wavenumber / cm ⁻¹	irreducible representation	intensity / a.u.
66	E ₁	58
180	E ₂	30
295	E ₂	16
316	A ₁	1000
380	E ₁	101
602	E ₂	58
754	E ₁	14
792	E ₁	3
875	E ₂	6

Table 3 Calculated IR and Raman vibrational modes for α -BeCl₂.

α -BeCl ₂ , IR active modes		
wavenumber / cm ⁻¹	irreducible representation	intensity / km mol ⁻¹
170	B _{2u}	0
170	B _{3u}	0
449	B _{1u}	1228
560	B _{3u}	402
572	B _{2u}	385

α -BeCl ₂ , Raman active modes		
wavenumber / cm ⁻¹	irreducible representation	intensity / a.u.
62	B _{1g}	0
150	A _g	276
169	B _{3g}	189
175	B _{2g}	201
293	B _{1g}	355
334	A _g	1000
385	B _{3g}	71
388	B _{2g}	60
684	B _{1g}	38

Table 4 Calculated IR and Raman vibrational modes for $\beta\text{-BeCl}_2$.

$\beta\text{-BeCl}_2$, IR active modes		
wavenumber / cm^{-1}	irreducible representation	intensity / km mol^{-1}
57	E_u	0
83	E_u	0
89	E_u	0
108	E_u	1
129	A_{2u}	2
138	A_{2u}	0
153	E_u	6
176	A_{2u}	27
179	E_u	13
197	E_u	5
215	E_u	21
226	E_u	12
227	A_{2u}	14
232	E_u	2
296	E_u	1
328	E_u	16
329	A_{2u}	2
429	E_u	32
434	E_u	5
451	A_{2u}	5577
462	E_u	7503
556	A_{2u}	2
563	E_u	2432
563	E_u	1742
569	A_{2u}	825
588	E_u	60

$\beta\text{-BeCl}_2$, Raman active modes		
wavenumber / cm^{-1}	irreducible representation	intensity / a.u.
36	B_{2g}	4
41	E_g	0
46	E_g	14
80	A_{1g}	10
81	E_g	2
88	E_g	5
89	B_{1g}	6
109	E_g	6
123	B_{2g}	0
137	A_{1g}	99
139	B_{1g}	97
142	B_{2g}	10
146	A_{1g}	52
150	E_g	118
175	B_{1g}	0
179	E_g	58
180	B_{2g}	206
199	E_g	2
215	E_g	152
216	B_{1g}	29
226	B_{2g}	10
228	E_g	0
232	E_g	7
245	A_{1g}	85
291	B_{1g}	19
292	A_{1g}	1000
296	E_g	41
327	E_g	15
330	B_{2g}	8
430	E_g	1
432	E_g	0
446	A_{1g}	2
453	B_{2g}	99
462	E_g	73
555	B_{2g}	5
558	E_g	6
559	A_{1g}	11
560	E_g	4
565	B_{1g}	3
572	B_{1g}	5
593	B_{1g}	10
605	E_g	60
608	B_{2g}	6
610	A_{1g}	220

Table 5 Calculated IR and Raman vibrational modes for BeBr₂.

BeBr ₂ , IR active modes		
wavenumber / cm ⁻¹	irreducible representation	intensity / km mol ⁻¹
108	B _{2u}	0
109	B _{3u}	0
400	B _{1u}	1160
492	B _{2u}	349
505	B _{3u}	347

BeBr ₂ , Raman active modes		
wavenumber / cm ⁻¹	irreducible representation	intensity / a.u.
45	B _{1g}	0
99	A _g	336
111	B _{2g}	220
114	B _{3g}	239
175	B _{1g}	399
203	A _g	1000
358	B _{2g}	215
362	B _{3g}	220
606	B _{1g}	23

Table 6 Calculated IR and Raman vibrational modes for α -BeI₂.

α -BeI ₂ , IR active modes		
wavenumber / cm ⁻¹	irreducible representation	intensity / km mol ⁻¹
83	B _{2u}	0
87	B _{3u}	0
365	B _{1u}	1181
435	B _{2u}	360
454	B _{3u}	351

α -BeI ₂ , Raman active modes		
wavenumber / cm ⁻¹	irreducible representation	intensity / a.u.
51	B _{1g}	1
82	B _{2g}	263
87	A _g	433
88	B _{3g}	309
119	B _{1g}	497
147	A _g	1000
340	B _{2g}	241
343	B _{3g}	217
519	B _{1g}	20

Table 7 Calculated IR and Raman vibrational modes for $\beta\text{-BeI}_2$.

$\beta\text{-BeI}_2$, IR active modes		
wavenumber / cm ⁻¹	irreducible representation	intensity / km mol ⁻¹
41	E _u	0
47	E _u	0
53	E _u	0
58	E _u	0
68	A _{2u}	0
71	A _{2u}	0
78	E _u	0
83	A _{2u}	0
84	E _u	1
93	E _u	0
103	E _u	1
108	E _u	8
109	A _{2u}	6
113	E _u	0
136	E _u	1
148	A _{2u}	0
148	E _u	2
349	A _{2u}	3926
351	E _u	4080
372	E _u	271
376	E _u	28
427	E _u	842
451	A _{2u}	157
456	E _u	204
462	E _u	4306
463	A _{2u}	1624

$\beta\text{-BeI}_2$, Raman active modes		
wavenumber / cm ⁻¹	irreducible representation	intensity / a.u.
29	E _g	0
37	B _{2g}	1
39	E _g	8
46	E _g	0
46	A _{1g}	42
53	E _g	8
56	E _g	12
58	B _{2g}	5
59	B _{1g}	7
66	A _{1g}	154
68	B _{1g}	244
72	E _g	275
74	B _{2g}	8
76	A _{1g}	93
85	B _{1g}	0
85	E _g	149
88	B _{2g}	428
93	E _g	3
102	E _g	334
103	B _{1g}	57
109	B _{2g}	19
111	E _g	1
112	E _g	15
118	A _{1g}	85
132	B _{1g}	53
136	E _g	100
143	A _{1g}	1000
147	E _g	170
148	B _{2g}	54
350	B _{2g}	222
351	E _g	176
372	E _g	0
375	E _g	42
379	A _{1g}	2
416	E _g	65
419	B _{1g}	0
441	B _{2g}	41
447	E _g	5
458	A _{1g}	8
462	B _{1g}	28
471	E _g	4
485	A _{1g}	87
486	B _{2g}	37
491	B _{1g}	106

Table 8 Optimized and experimental^{5,12,13} (in brackets) lattice parameters in Å and cell volumes in Å³.

compound	a	b	c	V per formula unit
$\alpha\text{-BeF}_2$	4.821 (4.7390(5))		5.294 (5.1875(8))	53.27 (50.45(3))
$\beta\text{-BeF}_2$	4.910 (4.8060(1))		5.407 (5.2404(1))	56.45 (52.41(0))
$\alpha\text{-BeCl}_2$	5.305 (5.285(3))	10.309 (9.807(3))	5.336 (5.227(3))	72.95 (67.7(2))
$\beta\text{-BeCl}_2$	10.883 (10.595(5))		18.457 (18.036(7))	68.31 (63(2))
BeBr ₂	5.621 (5.569(4))	10.662 (10.405(6))	5.601 (5.543(3))	83.91 (80.3(3))
$\alpha\text{-BeI}_2$	5.867 (6.025(3))	11.494 (11.316(4))	5.994 (6.035(3))	101.05 (102.9(3))
$\beta\text{-BeI}_2$	11.962 (12.190(6))		21.796 (21.325(8))	97.46 (99(3))

Notes and references

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