#### SUPPORTING INFORMATION

# 15-Copper(II)-Containing 36-Tungsto-4-silicates(IV) $[Cu_{15}O_2(OH)_{10}X(A-\alpha-SiW_9O_{34})_4]^{25-}$ (X = Cl, Br): Synthesis, Structure, Magnetic Properties, and Electrocatalytic CO<sub>2</sub> Reduction

Bassem S. Bassil,<sup>ab</sup> Ali Haider,<sup>a¥</sup> Masooma Ibrahim,<sup>a#</sup> Ali S. Mougharbel,<sup>a</sup> Saurav Bhattacharya,<sup>a</sup> Jonathan H. Christian,<sup>c</sup> Jasleen K. Bindra,<sup>c</sup> Naresh S. Dalal,<sup>c</sup> Meng Wang,<sup>d,e</sup> Guangjin Zhang,<sup>d</sup> Bineta Keita,<sup>f,‡</sup> Iwona A. Rutkowska,<sup>g</sup> Pawel Kulesza,<sup>g</sup> and Ulrich Kortz<sup>\*a</sup>



Figure S1. Infrared spectra of NaK-1 (black) and NaK-2 (red).



Figure S2. Thermogram of KNa-1.



Figure S3. Thermogram of KNa-2.



**Figure S4**. Polyhedral and ball-and-stick representation of Mialane's  $[Cu_{14}(OH)_{12}X(A-\alpha-SiW_9O_{34})_2(A-\alpha-SiW_9O_{33}(OH))_2]^{23-}$  (top) versus our  $[Cu_{15}O_2(OH)_{10}X(A-\alpha-SiW_9O_{34})_4]^{25-}$  (bottom) Color code: red octahedra (WO<sub>6</sub>), blue balls (Si), turquoise balls (Cu), red balls (O), light green ball (Cl).



Figure S5. Ball-and-stick representation of Mialane's  $\{Cu_{14}(OH)_{12}X\}^{15+}$  (top) versus our  $\{Cu_{15}O_2(OH)_{10}X\}^{15+}$  core (bottom). Color code same as in Figure S4.



Figure S6. ORTEP representation of polyanions 1 (top) and 2 (bottom).



Figure S7. Normalized magnetization of KNa-1 and KNa-2 at 1.8 K plotted with the Brillouin function for  $S = \frac{1}{2}$ , 1, and  $\frac{3}{2}$ . The curvature of both compounds matches closely with the  $S = \frac{1}{2}$  function up to 6 T.



**Figure S8.** Cyclic voltammograms of 0.1 mM solution of **1** restricted to the Cu<sup>II</sup>-to-Cu<sup>I</sup> (black) and Cu<sup>II</sup>-to-Cu<sup>0</sup> (red) redox processes, run in 1M CH<sub>3</sub>COOLi /CH<sub>3</sub>COOH (pH 5), at 100 mV s<sup>-1</sup>.



**Figure S8.** Cyclic voltammograms of 0.1 mM solution of **2** restricted to the Cu<sup>II</sup>-to-Cu<sup>I</sup> (black) and Cu<sup>II</sup>-to-Cu<sup>0</sup> (red) redox processes, run in 1M CH<sub>3</sub>COOLi /CH<sub>3</sub>COOH (pH 5), at 100 mV s<sup>-1</sup>.



**Figure S10.** Comparison of the cyclic voltammograms of 0.1 mM solutions of **1** (black) and **2** (red) restricted to the redox processes associated with the W<sup>VI</sup> centers, run in 1M CH<sub>3</sub>COOLi /CH<sub>3</sub>COOH (pH 5), at 100 mV s<sup>-1</sup>.

Compound	KNa-1	KNa-2
Formula	$Na_{3.5}K_{21.5}[Cu_{15}O_2(OH)_{10}(SiW_9O_{34})_4Cl]$	Na <sub>3</sub> K <sub>22</sub> [Cu <sub>15</sub> O <sub>2</sub> (OH) <sub>10</sub> (SiW <sub>9</sub> O <sub>34</sub> ) <sub>4</sub> Br]
	·55H <sub>2</sub> O	·60H <sub>2</sub> O
Formula weight, g/mol	12009.58	12152.17
Crystal system	Triclinic	Triclinic
Space group	p1	pl
<i>a</i> , Å	19.2429(6)	19.2051(12)
<i>b</i> , Å	20.9195(5)	20.8590(12)
<i>c</i> , Å	26.4944(8)	26.3973(16)
α, °	106.8890(10)	106.961(2)
$\beta, \circ$	101.894(2)	102.573(2)
γ, °	90.921(2)	90.741(2)
Volume, Å <sup>3</sup>	9953.4(5)	9839.0(10)
Ζ	2	2
D <sub>calc</sub> , g/cm <sup>3</sup>	4.007	4.102
Absorption coefficient	22.884	23.352
F(000)	10726	10870
Crystal size, mm	0.26 x 0.23 x 0.12	0.12 x 0.09 x 0.04
Theta range for data	3.403 - 26.372	3.410 - 26.372
Completeness to $\Theta_{max}$ ,	98.9	99.4
Index ranges	-24 >h>24 , -26>k>26 , -33>l>33	-19 <h<24, -26<k<26,="" -32<l<32<="" td=""></h<24,>
Total Reflections	272948	326448
Independent	40215	39975
Calculated Reflections	24967	24085
R(int)	0.1094	0.1294
Data / restraints /	40215 / 0 / 1362	39975 / 0 / 1372
Goodness-of-fit on F2	1.002	1.003
$R_1^{[a]}$	0.0788	0.0985
wR2 <sup>[b]</sup>	0.2662	0.2860
e Density max / min, e.	4.455 / -3.287	4.374 / -3.696

## Table S1. Crystallographic data for KNa-1 and KNa-2.

<sup>[a]</sup>  $R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|$ . <sup>[b]</sup>  $wR_2 = [\sum w (F_o^2 - F_c^2)^2 / \sum w (F_o^2)^2]^{1/2}$ 

Table S2. Selected bond valence sum (BVS) values for 1 (upper) and 2 (lower).

μ <sub>2</sub> -Ο (3Cu-Ο)	BVS Value	μ <sub>2</sub> -Ο (3Cu-Ο)	BVS Value
O34C	-0.90	O67C	-0.73
μ <sub>3</sub> -Ο (3Cu-O)	BVS Value	μ <sub>3</sub> -Ο (3Cu-Ο)	BVS Value
O23C	-1.04	059C	-1.05
O24C	-1.04	O91C	-1.05
O14C	-1.19	O81C	-1.10
O15C	-1.13	O82C	-1.12
μ <sub>4</sub> -Ο (4Cu-Ο)	BVS Value	μ <sub>4</sub> -Ο (4Cu-Ο)	BVS Value
O45C	-1.61	O55C	-1.60

## $[Cu_{15}O_2(OH)_{10}(A-\alpha-SiW_9O_{34})_4Cl]^{25-}(1)$

## $[Cu_{15}O_2(OH)_{10}(A-\alpha-SiW_9O_{34})_4Br]^{25-}(2)$

μ <sub>2</sub> -Ο (3Cu-Ο)	BVS Value	μ <sub>2</sub> -Ο (3Cu-Ο)	BVS Value
O34C	-0.91	O67C	-0.75
μ <sub>3</sub> -Ο (3Cu-Ο)	BVS Value	µ3-O (3Cu-O)	BVS Value
O93C	-1.08	O26C	-1.13
O81U	-1.09	O14U	-1.32
O91C	-1.09	O24C	-1.05
O81C	-1.09	014C	-1.27
μ <sub>4</sub> -Ο (4Cu-Ο)	BVS Value	μ <sub>4</sub> -Ο (4Cu-Ο)	BVS Value
015C	-1.61	O45C	-1.57