

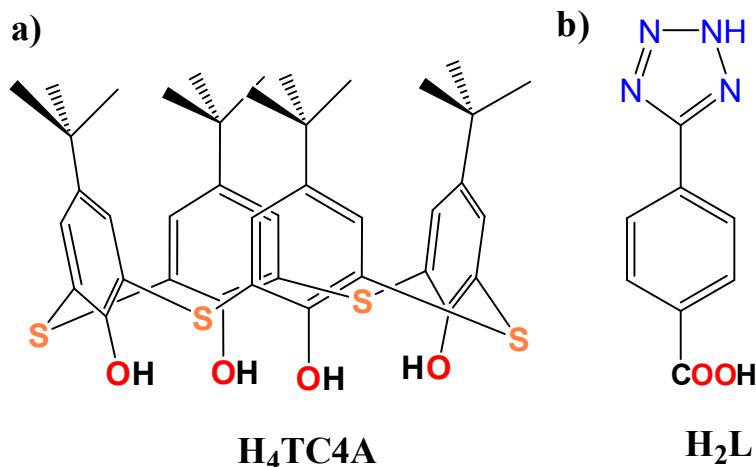
Electronic Supplementary Information

A 2D Metal-thiacalix[4]arene Porous Coordination Polymer with 1D Channels: Gas Absorption/Separation and Frequency Response

Dantong Geng,^{‡a} Min Zhang,^{‡a} Xinxin Hang,^{*a} Wenjie Xie,^a Yucai, Qin,^{*a} Qiang Li,^a Yanfeng Bi,^{*a} and Zhiping Zheng^{ab}

^aCollege of Chemistry, Chemical Engineering and Environmental Engineering, Liaoning Shihua University, Fushun 113001, P. R. China.

^b University of Arizona, Department of Chemistry, Tucson AZ 85721, USA



Scheme S1. (a) *p*-*tert*-Butylthiacalix[4]arene ($\text{H}_4\text{TC4A}$); (b) 4-(2*H*-tetrazol-5-yl)benzoic acid (H_2L).

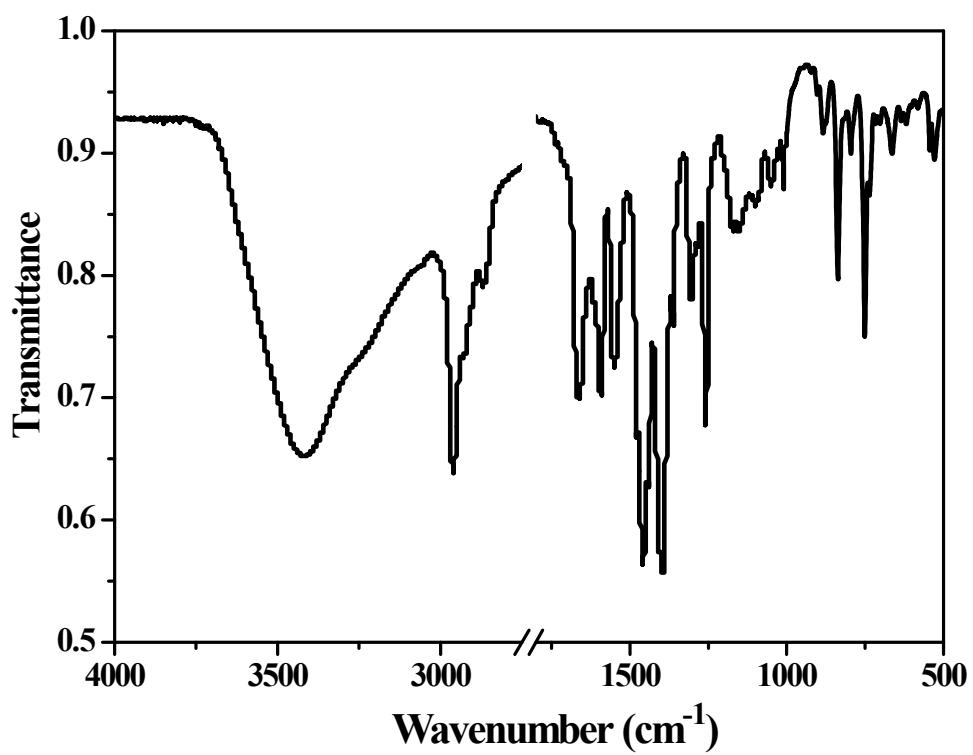


Fig. S1 FT-IR spectra of **1**

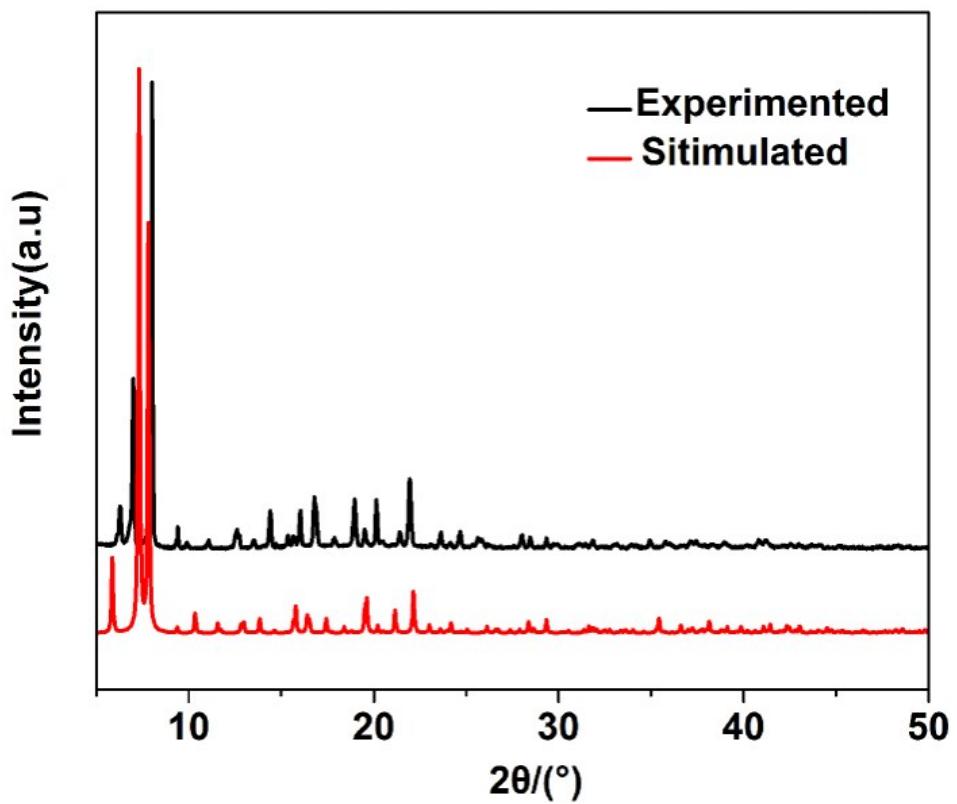


Fig. S2 PXRD patterns of **1**

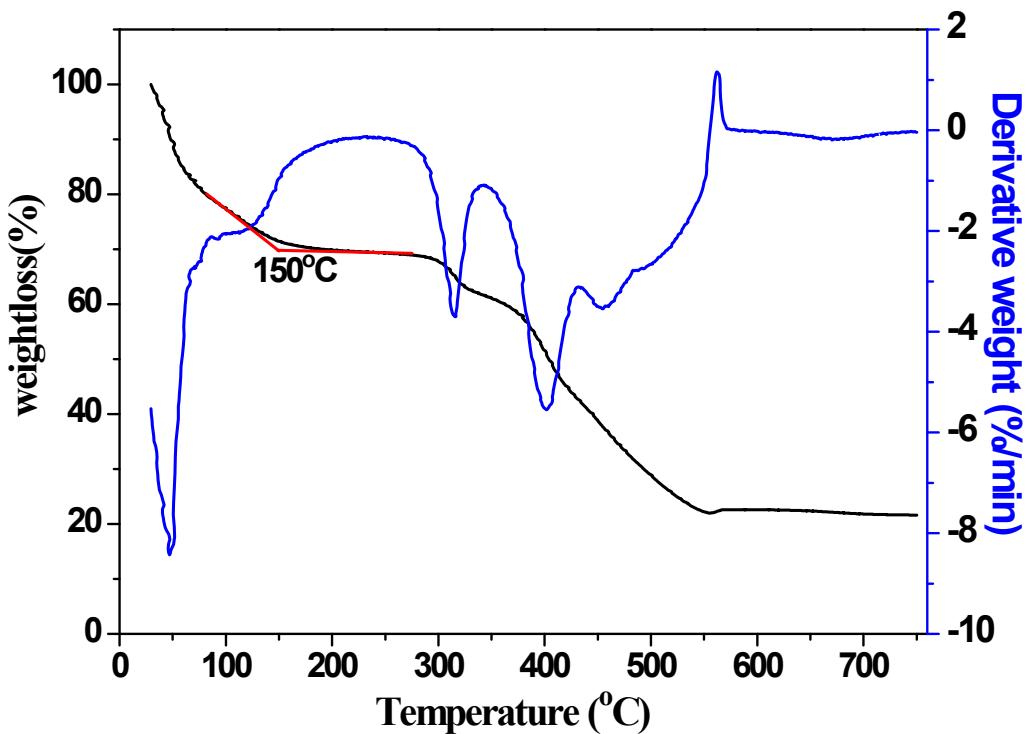


Fig. S3 TGA experiment of **1**

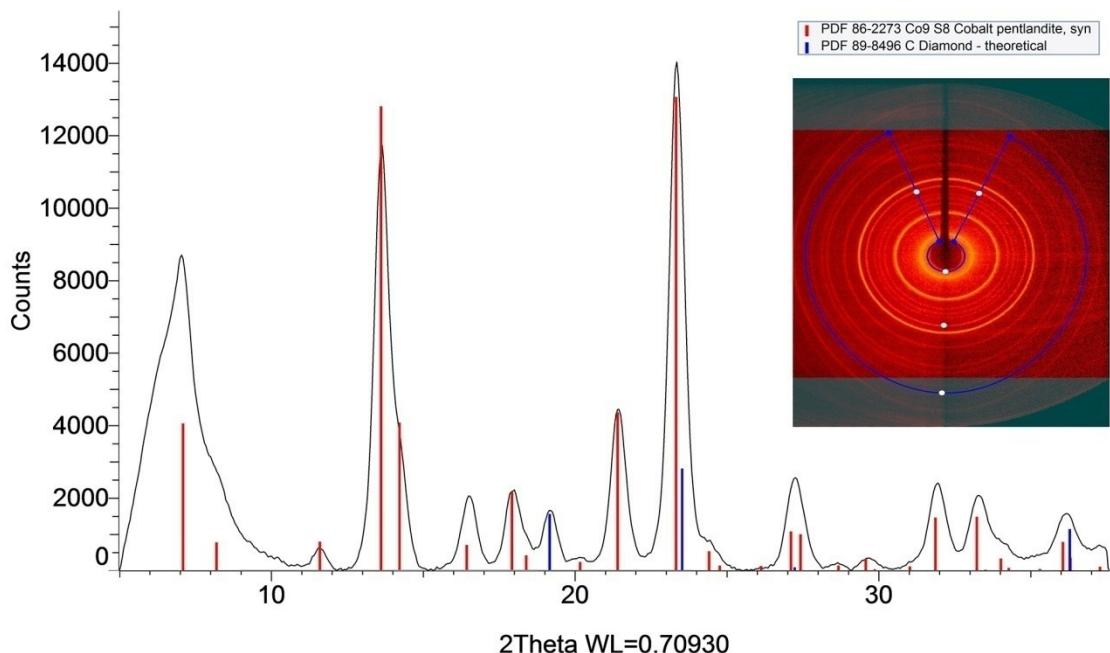


Fig. S4 PXRD analysis of residue after thermogravimetry measurement for **1**

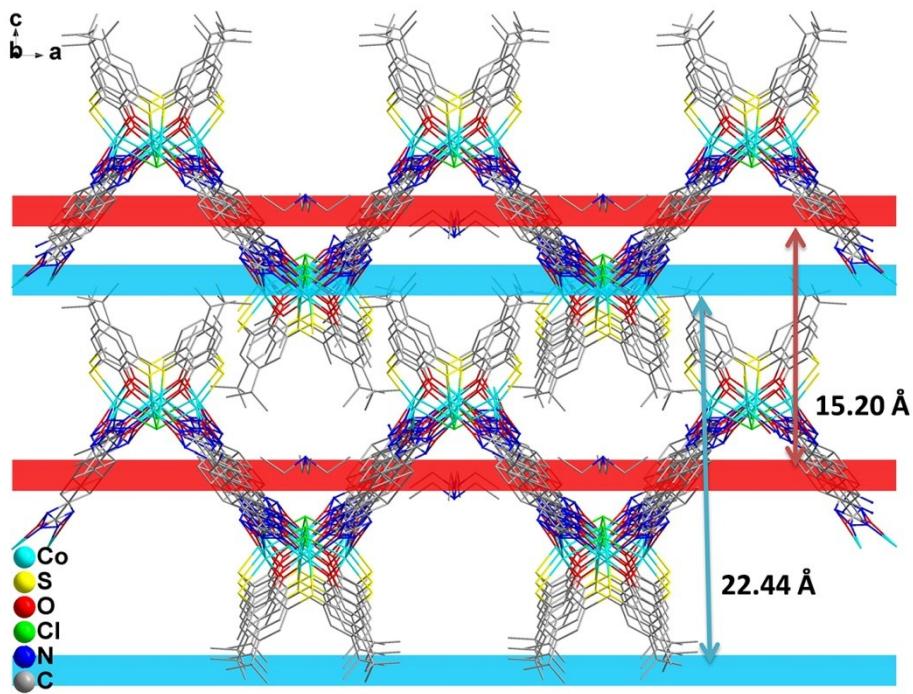


Fig. S5 Representation of the thickness of the layers and the centre distance between adjacent two layer.

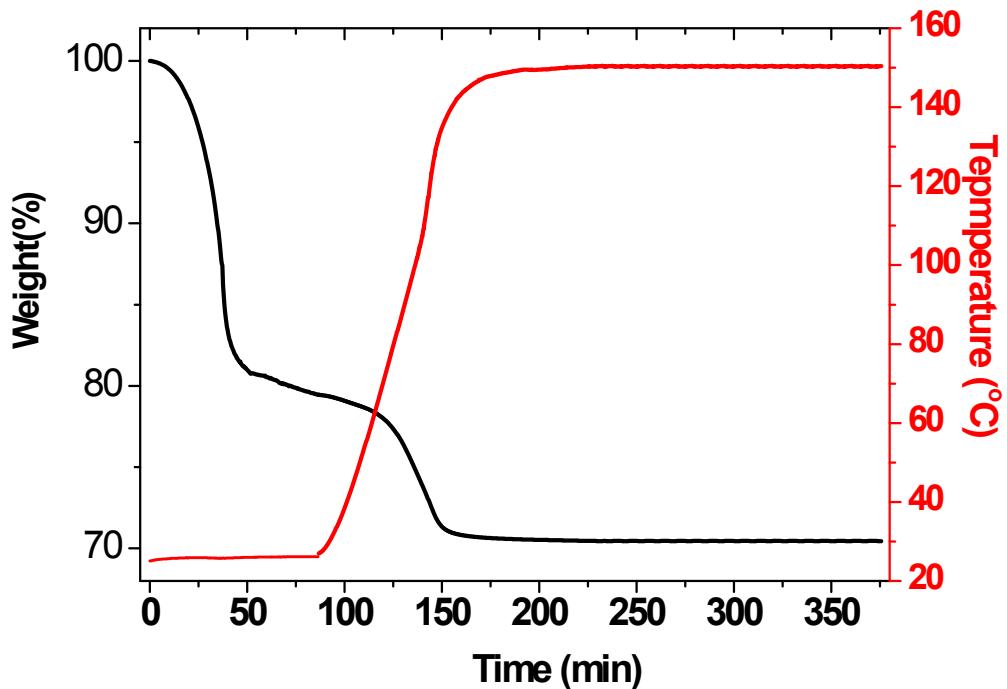


Fig. S6 Temperature programmed desorption (TPD, heating rate 3 °C/min, in vacuum for 90min, maintained at 150 °C for 220 min) of **1**.

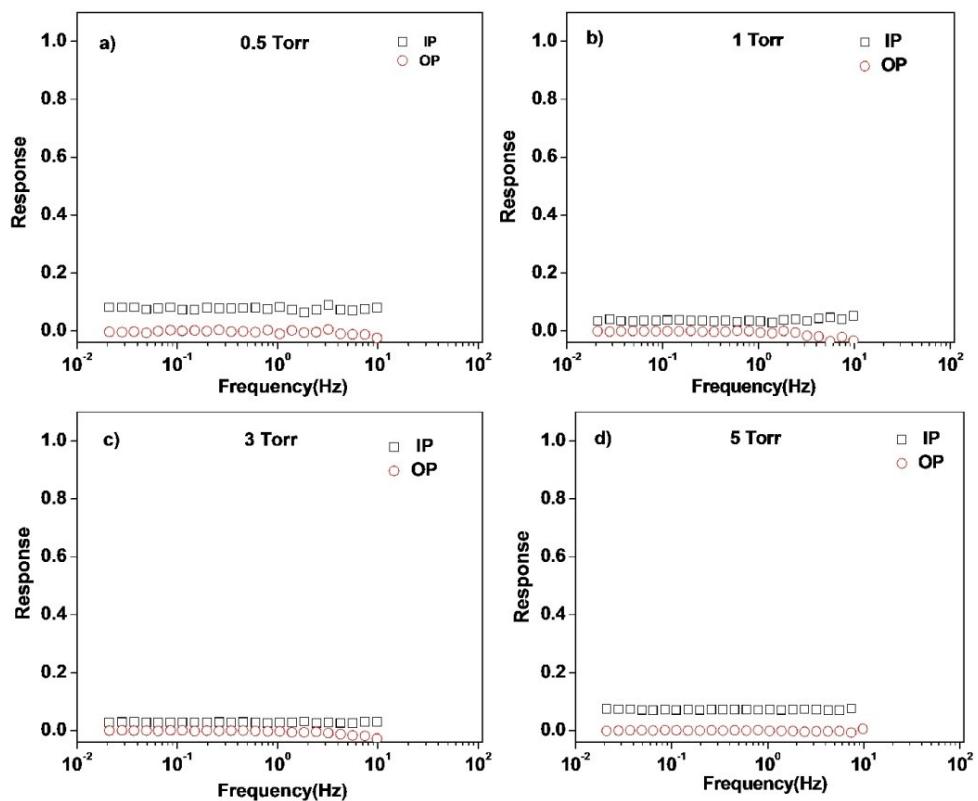


Fig. S7 Frequency response (FR) spectra of C_2H_6 for $\mathbf{1}'$ at different pressure (0.5, 1, 3, 5 Torr) at 273K, 0–100Hz.

Table S1 Crystal data and structural refinement parameters for **1**

Formula	C ₈₃ H ₁₁₈ ClCo ₄ N ₁₅ O ₁₅ S ₄
Mr	1965.33
Crystal system	tetragonal
space group	P4 ₂ /2 (No.90)
Temperature (K)	150
<i>a</i> (Å)	17.0886(5)
<i>b</i> (Å)	17.0886(5)
<i>c</i> (Å)	15.0638(5)
α (°)	90
β (°)	90
γ (°)	90
Volume(Å ³)	4398.9(3)
<i>Z</i>	2
<i>D</i> _c (g/cm ³)	1.484
μ (mm ⁻¹)	0.939
Reflections collected	46014
Unique data(<i>R</i> _{int})	3888
<i>GOF</i> on <i>F</i> ²	1.08
<i>R</i> ₁ [<i>I</i> >2sigma(<i>I</i>)]	0.0377
w <i>R</i> ₂	0.1061

$$^aR_1 = \Sigma ||F_0| - |F_c|| / \Sigma |F_0|; ^bW R_2 = \{\Sigma [w(F_0^2 - F_c^2)^2] / \Sigma [w(F_0^2)^2]\}^{1/2}*.$$

Table S2 Selected bond lengths (Å) for **1**

Co(1)-O(1)	1.994(3)
Co(1)-O(2)2#	2.025(10)
Co(1)-N(3)	2.135(13)
Co(1)-N(4)#1	2.097(12)
Co(1)-S(1))#1	2.5450(14)
Co(1)-Cl(1)	2.5973(12)
Co(1)-O(1)#1	2.002(3)
Co(1)-O(3) 3#	2.019(11)
Co(1)-N(4)4#	2.097(12)

#1: -1/2+y,3/2-x,z; #2: y,x,-z; #3: -1/2+x,3/2-y,-z; #4: -1/2+y,3/2-x,z.