

## Electronic Supplementary Information

# A versatile chemosensor for detection of Al<sup>3+</sup> and Picric acid (PA) in aqueous solution

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**Figure S13.** Emission spectra of **H<sub>2</sub>Vm** in the presence of [Al<sup>3+</sup>] in HEPES buffer (pH = 7.4) solution ( $\lambda_{\text{ex}} = 450$  nm,  $\lambda_{\text{em}} = 488$  nm).

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**Table S5.** Selected parameters for the vertical excitation (UV-vis absorptions) and the emission of **Al<sub>2</sub>-Vm<sub>2</sub>**, electronic excitation energies (eV) and oscillator strengths (*f*) and contributions of the lowest lying excited state.

**Figure S29.** Theoretical UV spectrum of **Al<sub>2</sub>-Vm<sub>2</sub>**.

### 4. Cell Study

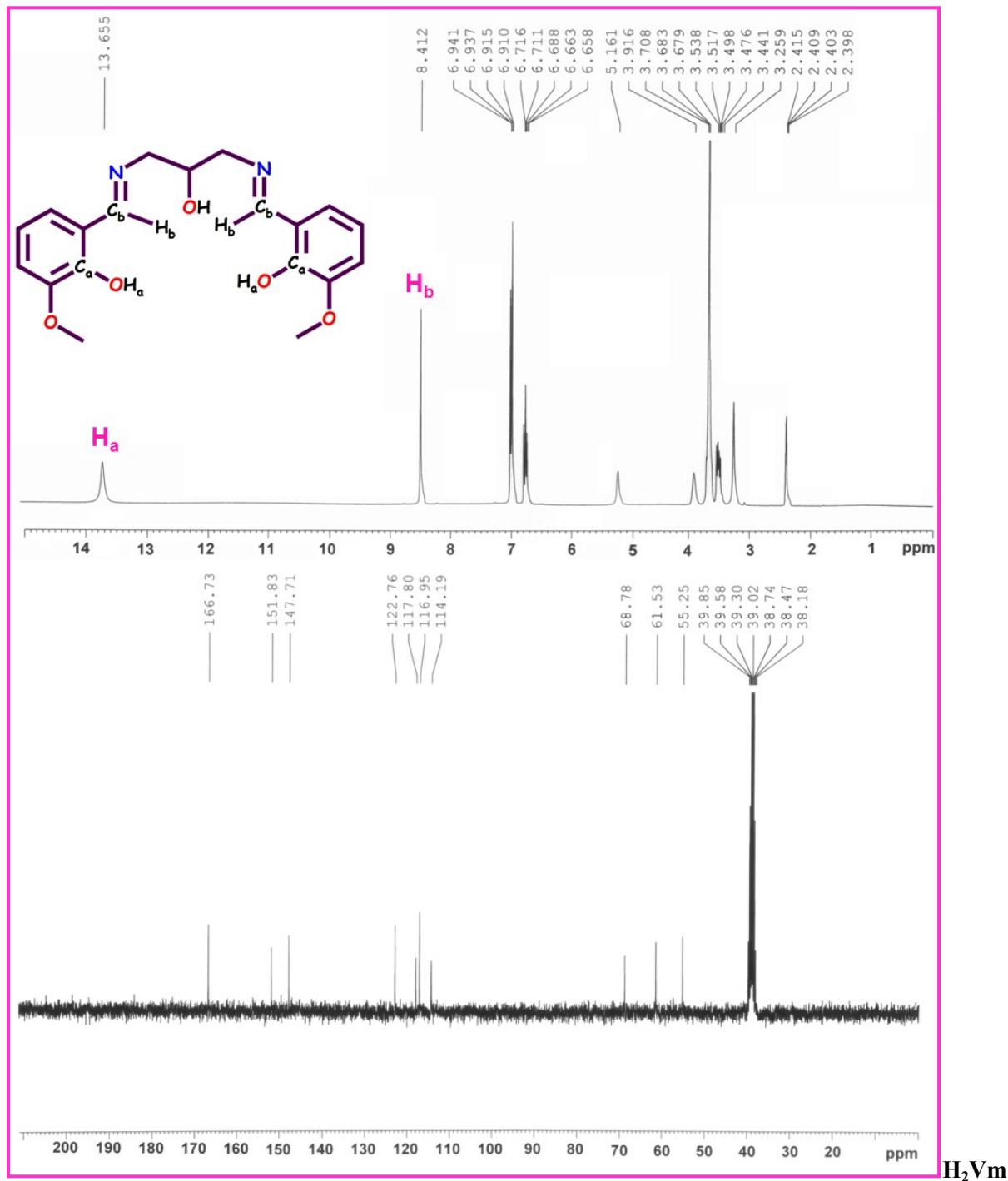
**Figure S30.** Cell viability curve of ligand **H<sub>2</sub>Vm**.

### 5. Cartesian Coordinates

Cartesian coordinates of the enol, Keto and Keto-enol tautomers and the complexes **1** and **2**

## 1. Characterization and Structural Data.

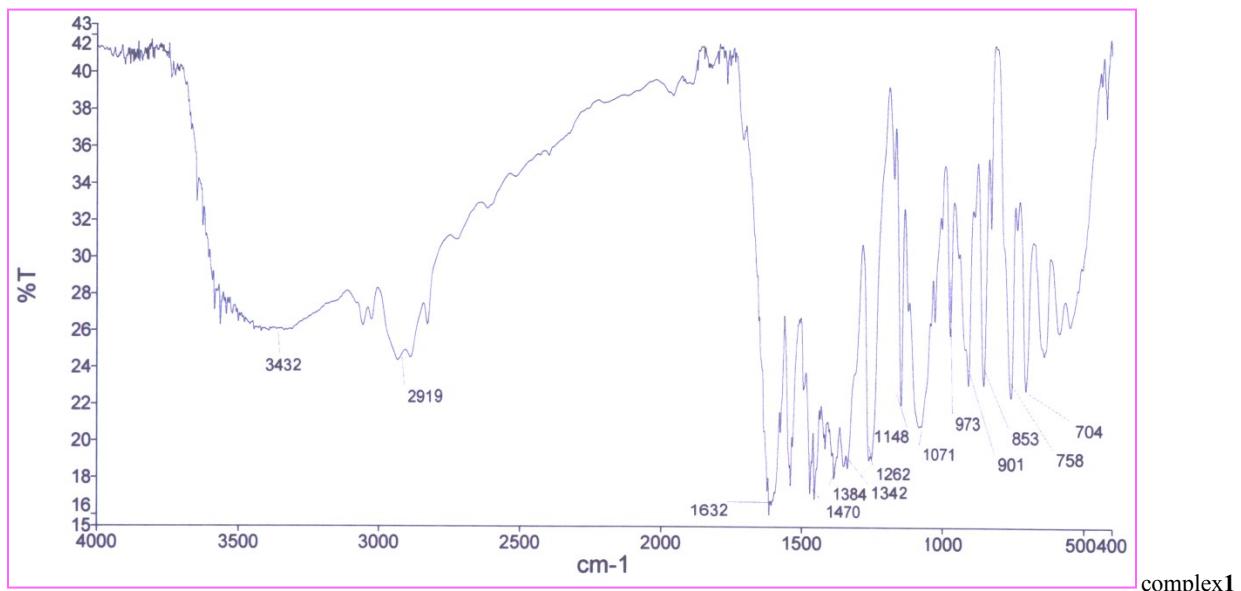
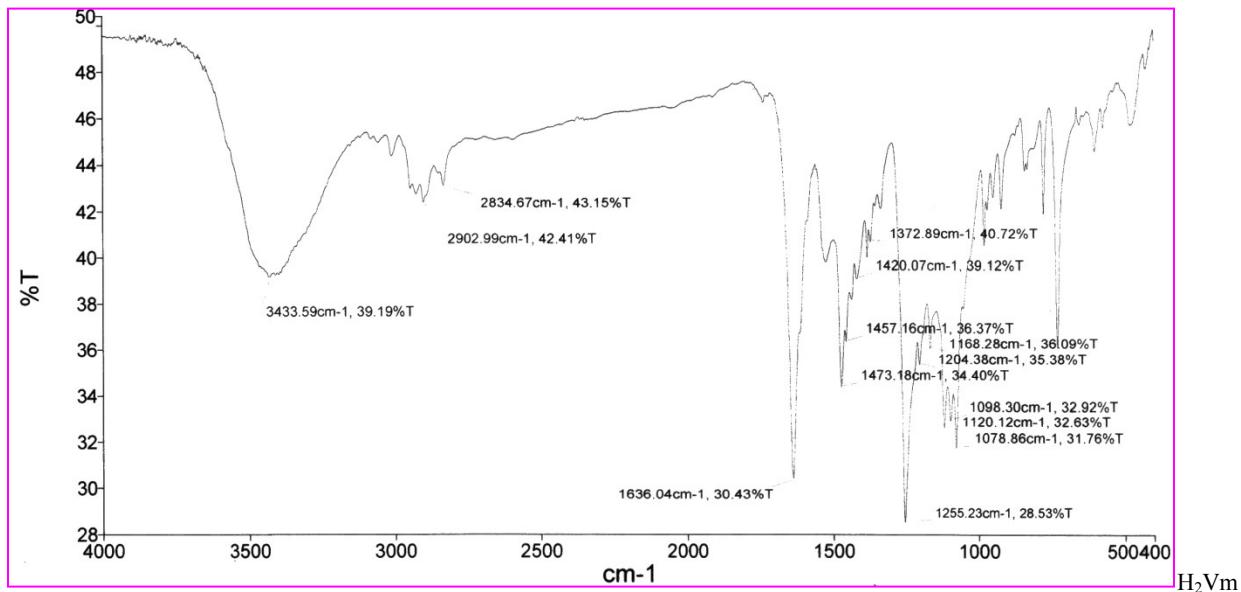
**$^1\text{H}$  and  $^{13}\text{C}$ -NMR spectra:**  $\text{H}_2\text{Vm}$  were dissolved in  $d_6$ -DMSO and recorded with TMS as internal standard on a Bruker, AV 300 Supercon Digital NMR system.

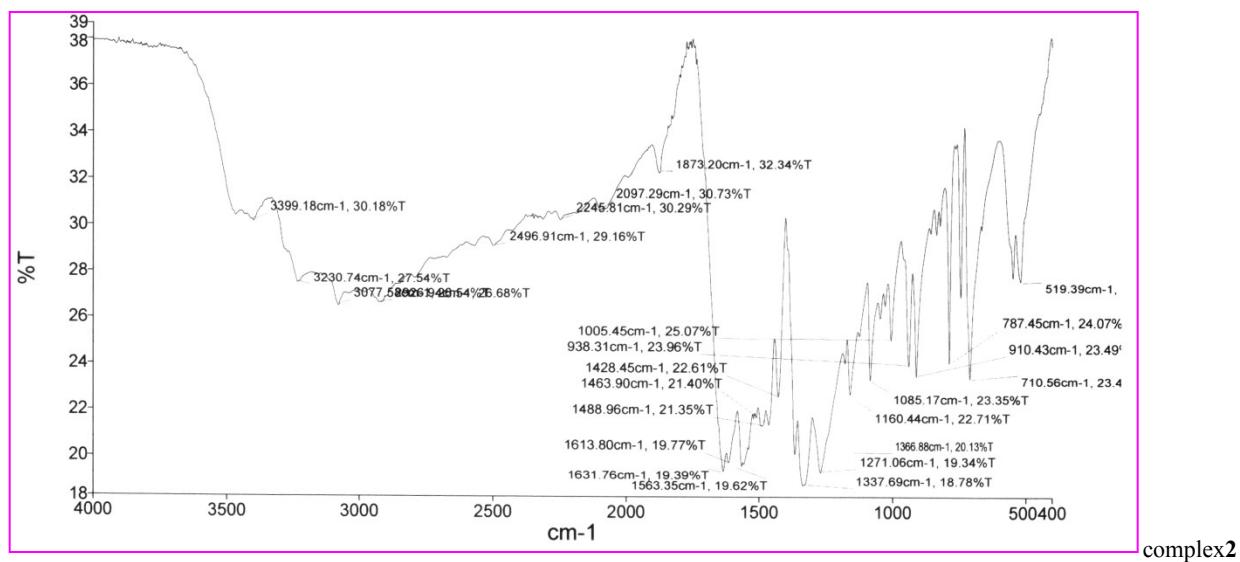


**Figure S1.**  $^1\text{H}$  and  $^{13}\text{C}$ -NMR spectrum of chemosensor  $\text{H}_2\text{Vm}$ .

## 1. Characterization and Structural Data.

**FT-IR spectroscopy:** Fourier transform infrared (FT-IR) spectra were recorded with a Perkin–Elmer RXI FT-IR spectrophotometer using the reflectance technique (4000–400 cm<sup>-1</sup>). Samples were prepared as KBr disks.

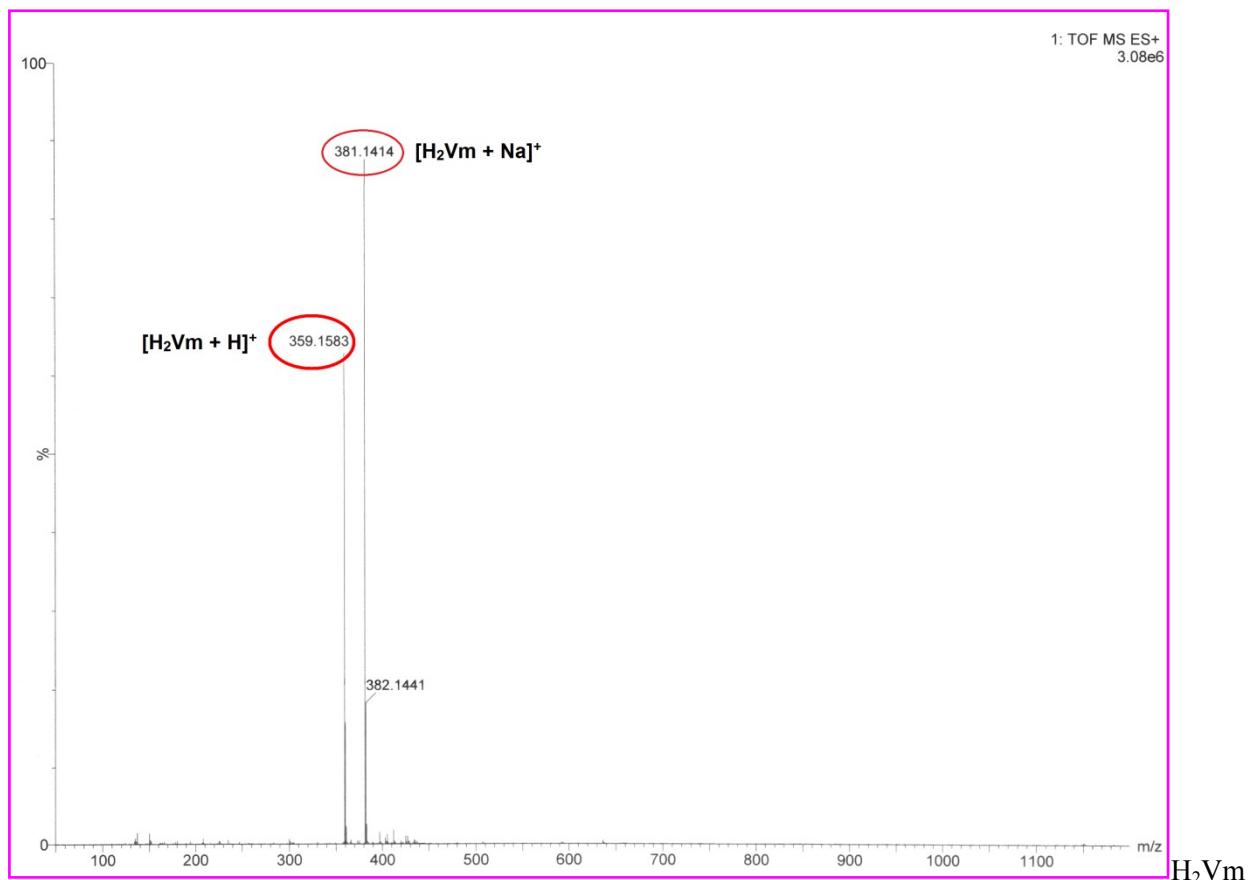


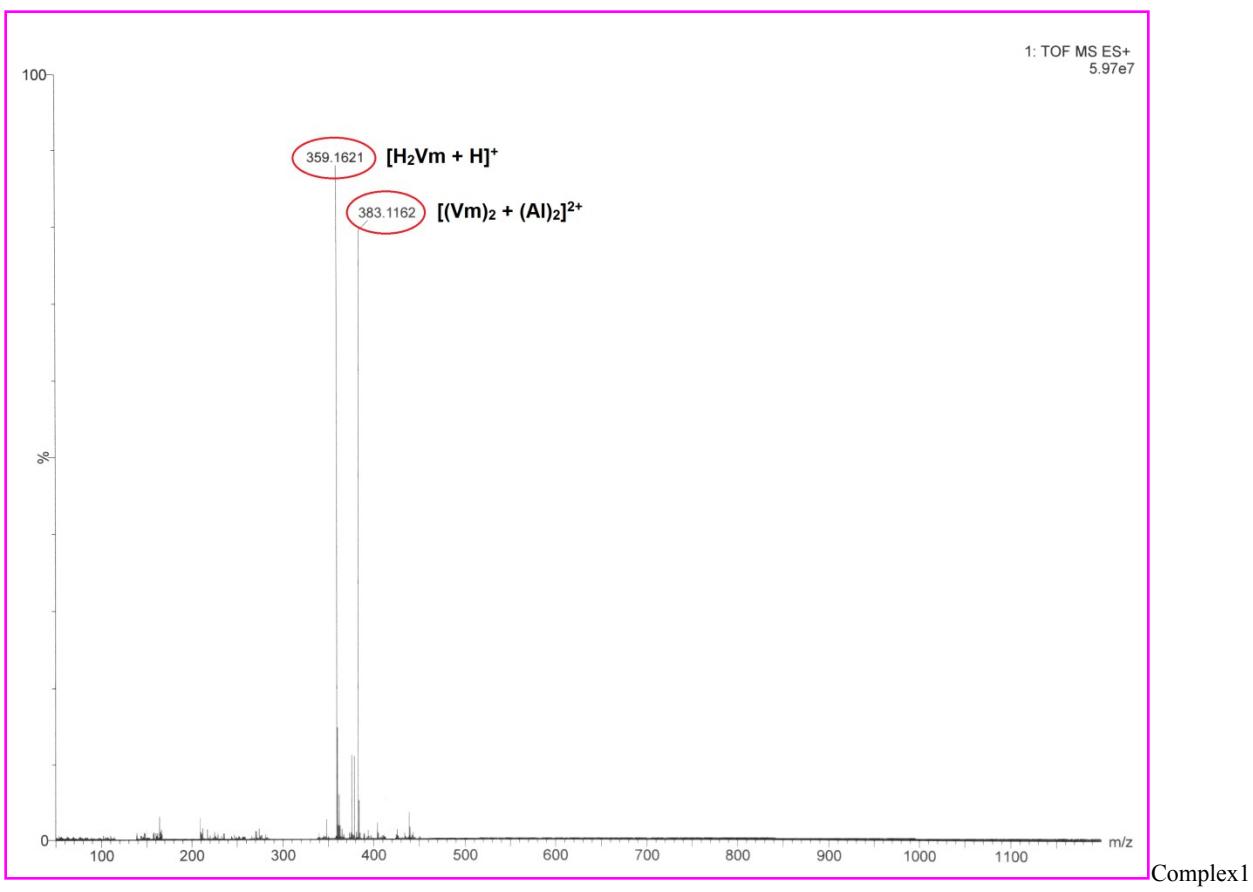


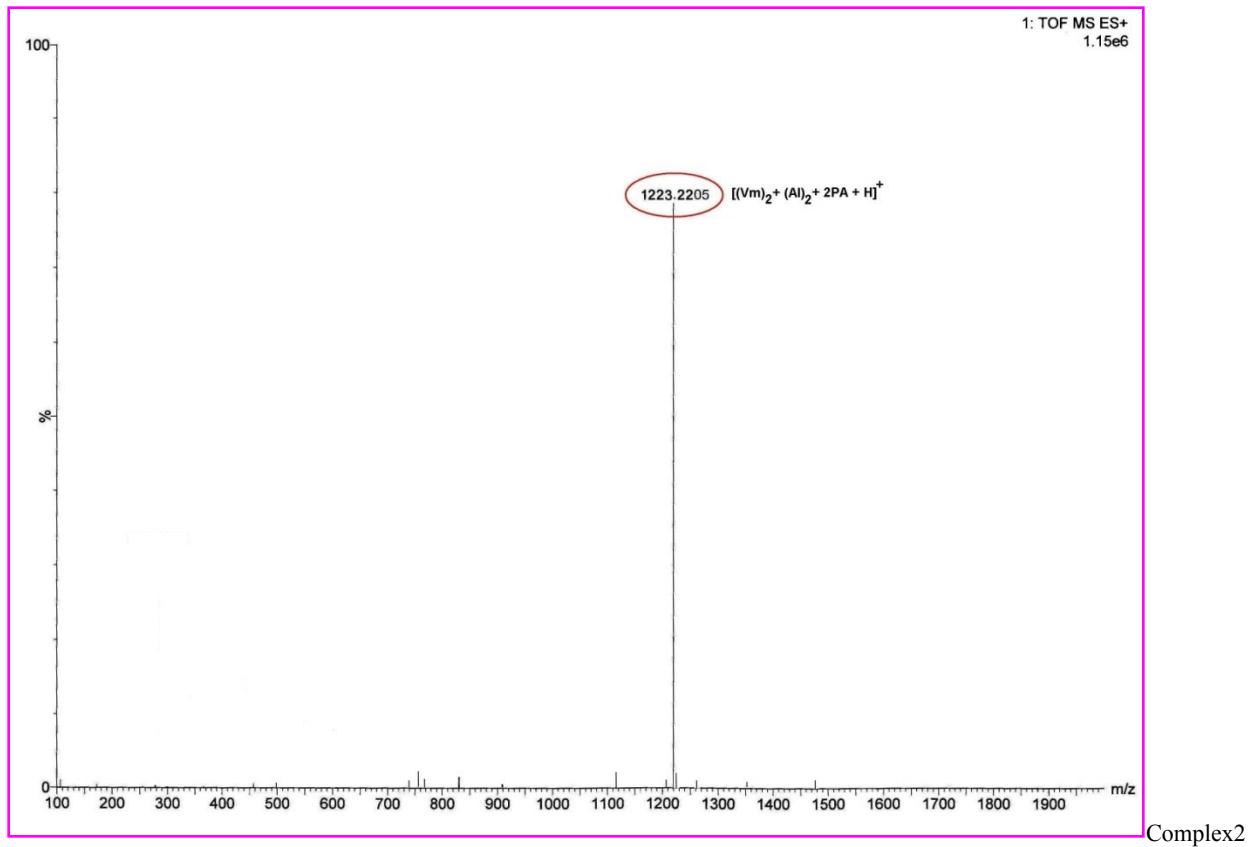
**Figure S2.** FT-IR Spectrum of **H<sub>2</sub>Vm**, **[Al<sub>2</sub>-Vm<sub>2</sub>]** complex **1** and **[Al<sub>2</sub>-Vm<sub>2</sub>].PA** complex **2**.

## 1. Characterization and Structural Data.

**Electrospray ionization mass spectra** (HR–ESI–MS) were recorded on Qtof Micro YA263 mass spectrometer dissolving the samples in LC–MS quality MeOH.

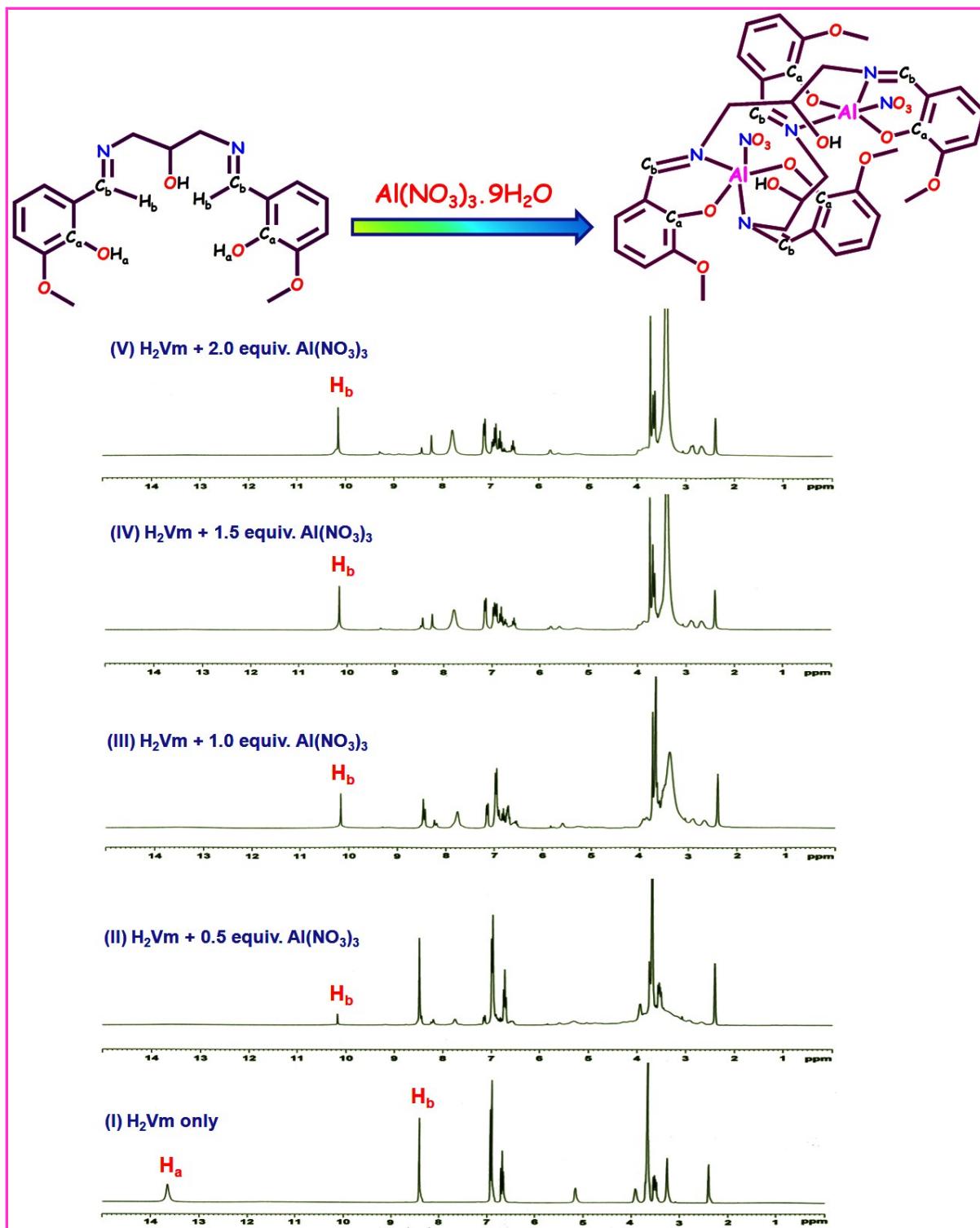




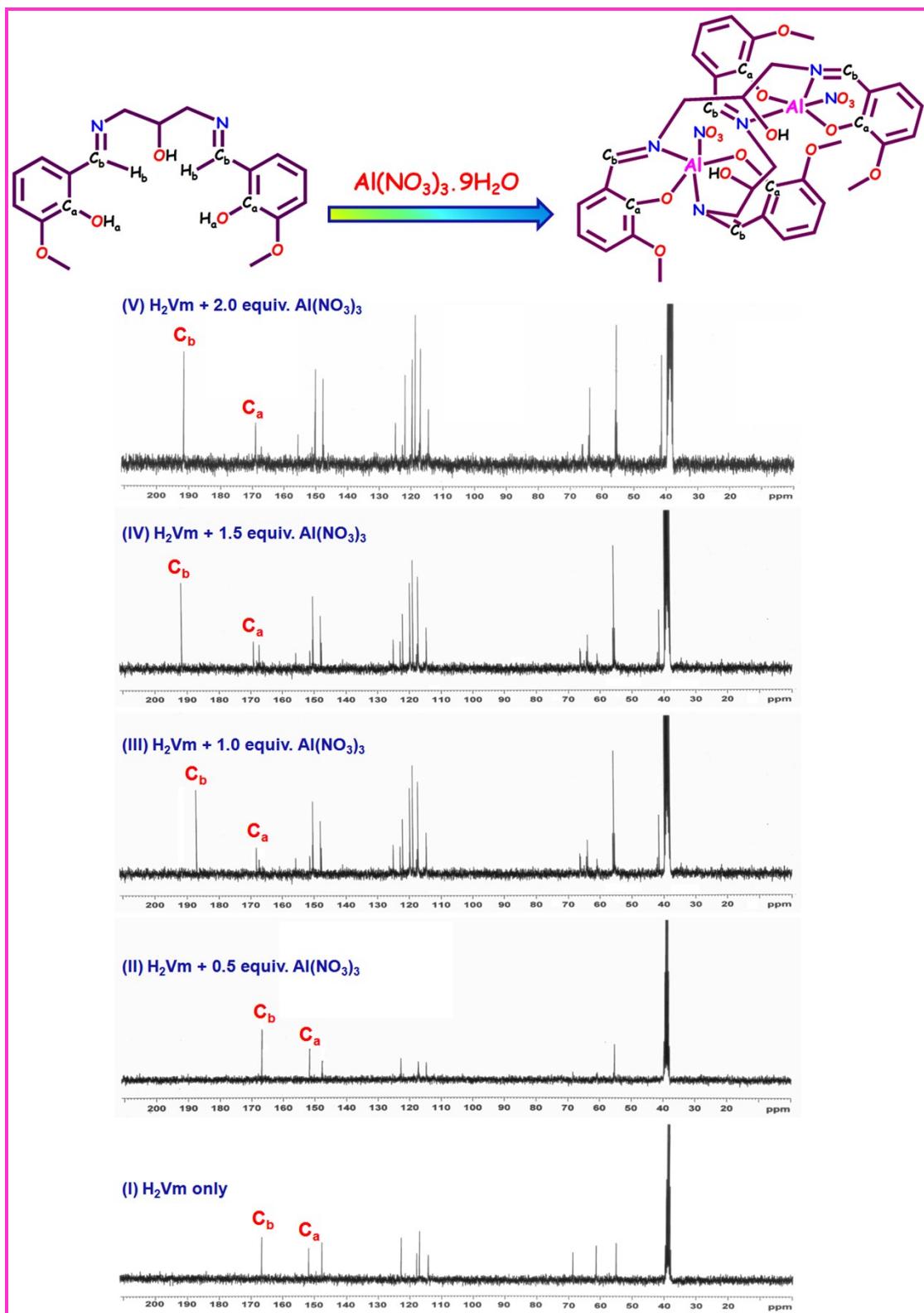


**Figure S3.** ESI-MS of ligand ( $H_2Vm$ ),  $[Al_2-Vm_2]$  complex **1** and  $[Al_2-Vm_2]$ . PA complex **2**.

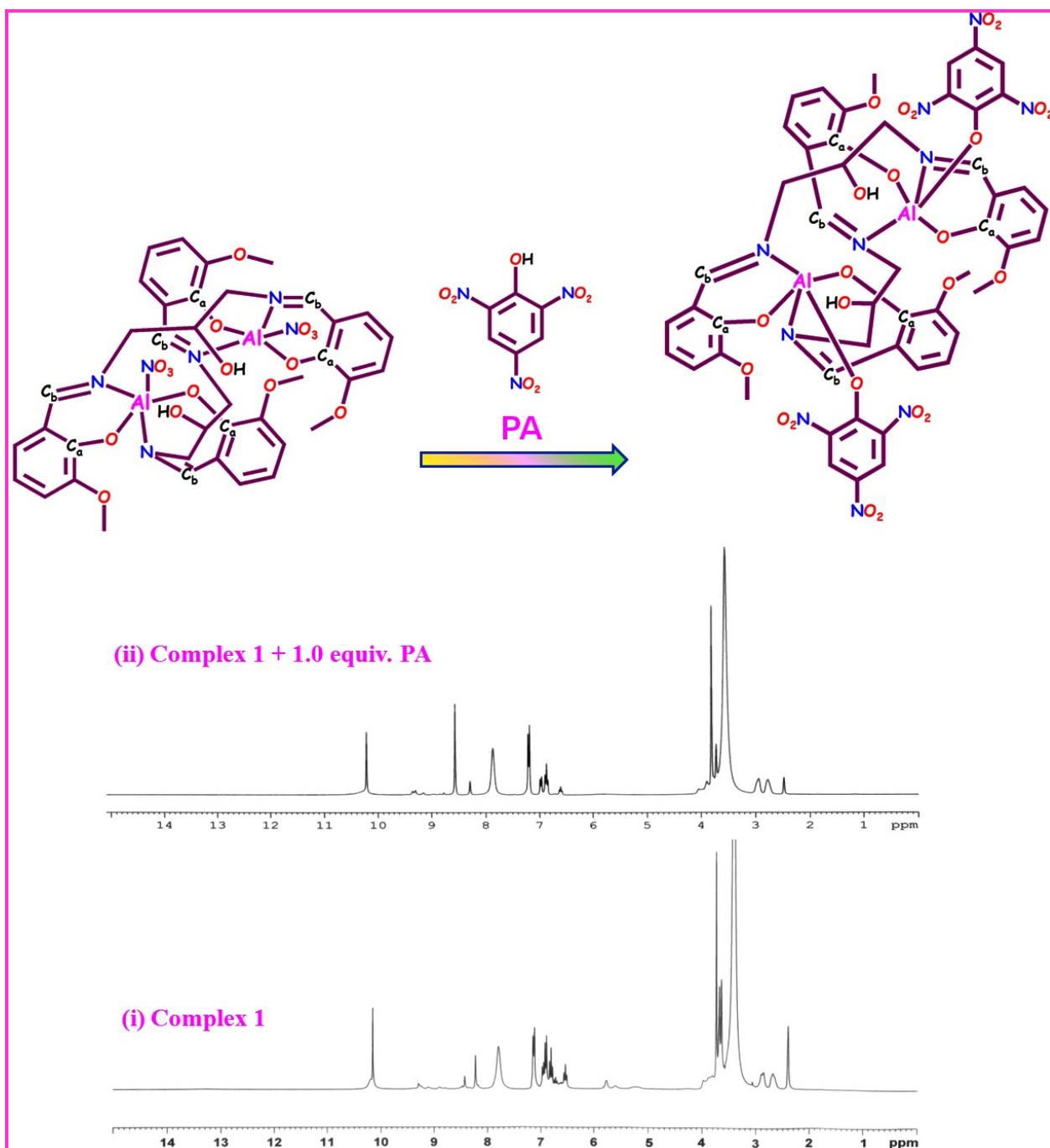
## 1. Characterization and Structural Data.



**Figure S4.**  $^1\text{H}$ -NMR titration of  $\text{H}_2\text{Vm}$  with  $\text{Al}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$  in  $\text{DMSO}-\text{d}_6$  solution.



**Figure S5.**  ${}^{13}\text{C}$ -NMR titration of  $\text{H}_2\text{Vm}$  with  $\text{Al}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$  in  $\text{DMSO}-\text{d}_6$  solution.



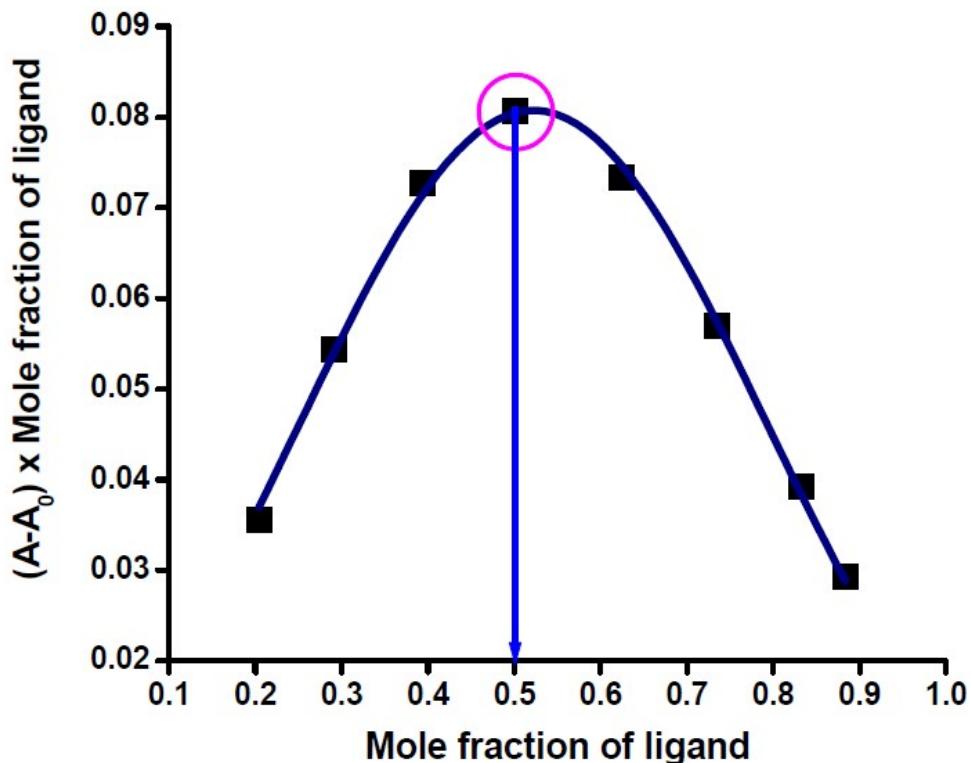
**Figure S6.** <sup>1</sup>H-NMR titration of complex 1 with Picric acid (PA) in DMSO-d<sub>6</sub> solution.

## 1. Characterization and Structural Data.

**Table S1.**  $^1\text{H}$  and  $^{13}\text{C}$ -NMR shift data of NMR titration experiment of  $\text{H}_2\text{Vm}$ .

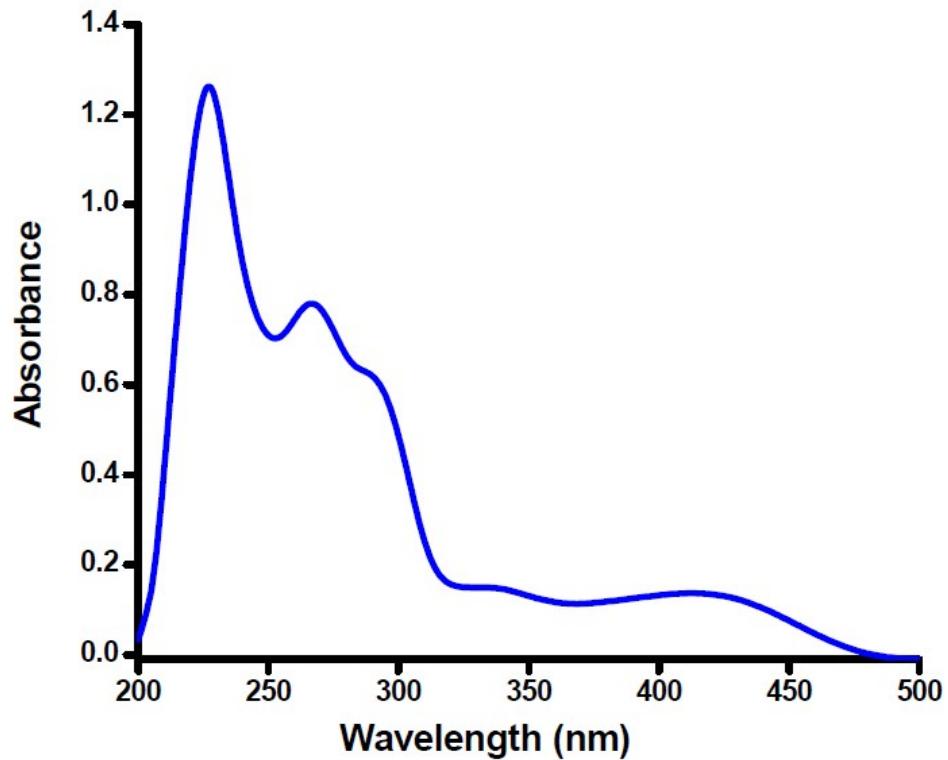
$^1\text{H}$						
Proton label	Free $\text{H}_2\text{Vm}$ (ppm) (A)	$\text{H}_2\text{Vm} + 0.5 \text{ equiv. Al}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$ (ppm) (B)	$\text{H}_2\text{Vm} + 1.0 \text{ equiv. Al}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$ (ppm) (C)	$\text{H}_2\text{Vm} + 1.5 \text{ equiv. Al}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$ (ppm) (D)	$\text{H}_2\text{Vm} + 2.0 \text{ equiv. Al}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$ (ppm) (E)	Shift (E-A) (ppm)
$\text{H}_a$	13.655	-	-	-	-	-
$\text{H}_b$	8.412	10.161	10.164	10.170	10.176	1.764 <b>downfield</b>
$^{13}\text{C}$						
$C_a$	151.83	151.97	168.89	169.06	169.09	17.26 <b>downfield</b>
$C_b$	166.73	166.96	187.56	191.63	191.67	24.94 <b>downfield</b>

## 2. Photophysical Characterization.



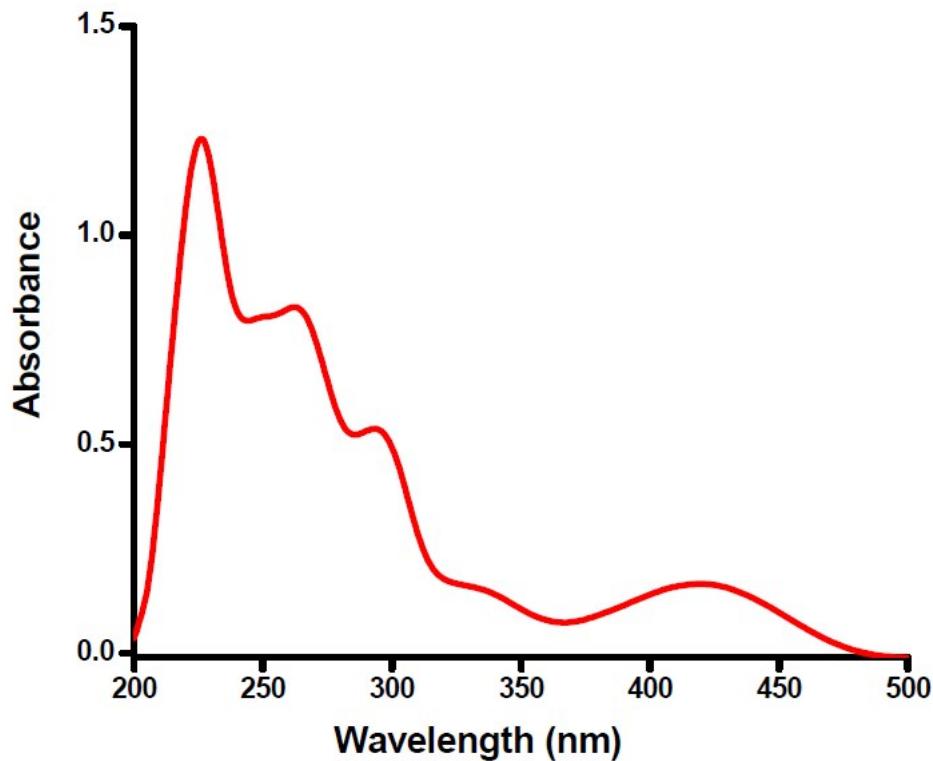
**Figure S7.** Job's plot for the identification of (1:1) complex stoichiometry between  $\text{H}_2\text{Vm}$  and  $\text{Al}^{3+}$  using absorbance values at 368 nm.

## 2. Photophysical Characterization.



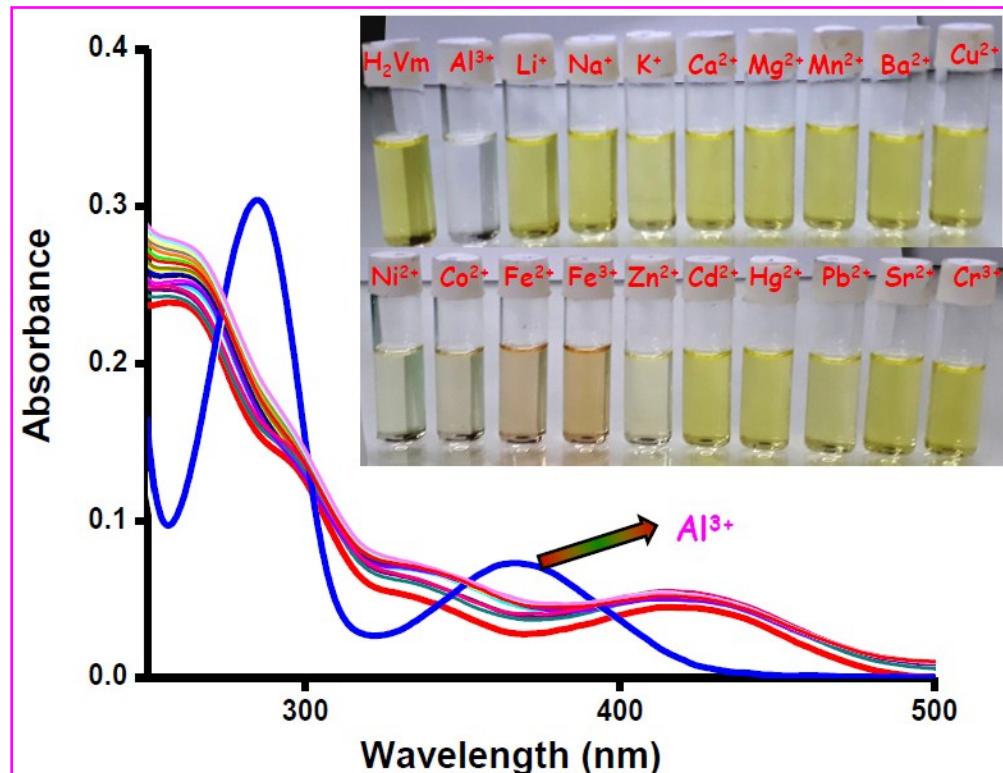
**Figure S8.** Absorbance spectra of  $\text{H}_2\text{Vm}$  in methanol solution.

## 2. Photophysical Characterization.



**Figure S9.** Absorbance spectra of  $\mathbf{H}_2\mathbf{Vm}$  in HEPES buffer (pH = 7.4) solution.

## 2. Photophysical Characterization.



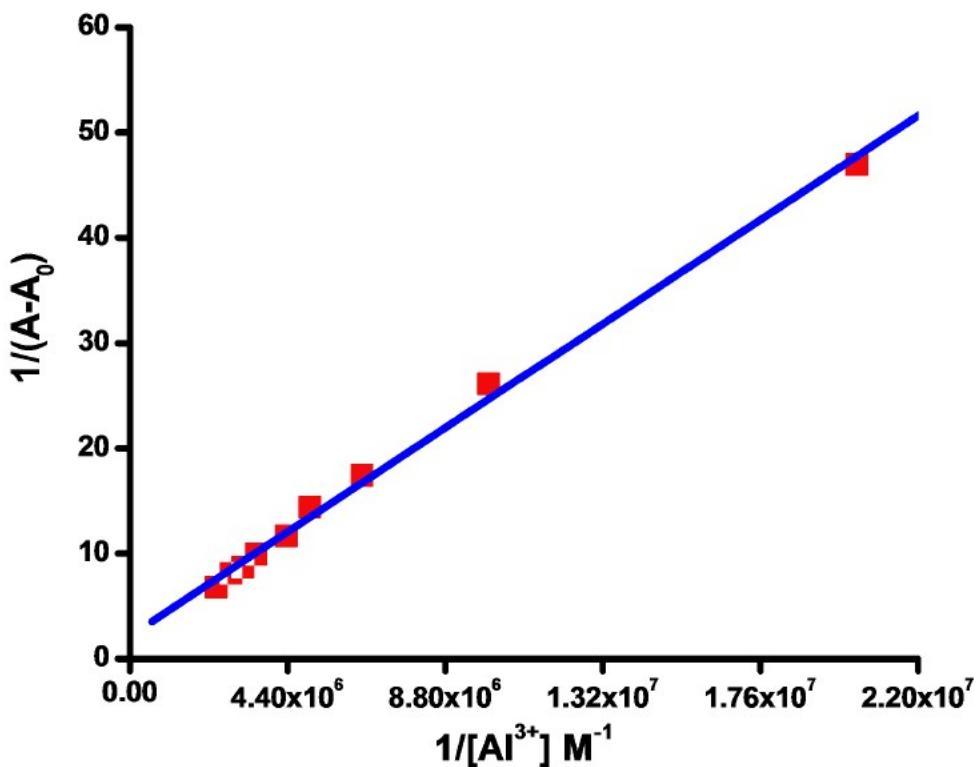
**Figure S10.** Absorbance spectra of  $\text{H}_2\text{Vm}$  ( $5 \times 10^{-7}$  M) in HEPES buffer (pH = 7.4) solution in the presence different metal ions like  $\text{Al}^{3+}$ ,  $\text{Li}^+$ ,  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Ca}^{2+}$ ,  $\text{Mg}^{2+}$ ,  $\text{Mn}^{2+}$ ,  $\text{Ba}^{2+}$ ,  $\text{Cu}^{2+}$ ,  $\text{Ni}^{2+}$ ,  $\text{Co}^{2+}$ ,  $\text{Fe}^{2+}$ ,  $\text{Fe}^{3+}$ ,  $\text{Zn}^{2+}$ ,  $\text{Cd}^{2+}$ ,  $\text{Hg}^{2+}$ ,  $\text{Pb}^{2+}$ ,  $\text{Sr}^{2+}$  and  $\text{Cr}^{3+}$ . **Inset:** Visual color change observed with addition of different metal ions to  $\text{H}_2\text{Vm}$  solution.

## 2. Photophysical Characterization.

The binding constant ( $K$ ) determined by the Benesi–Hildebrand expression was found to be  $5.23 \times 10^5 \text{ M}^{-1}$ .

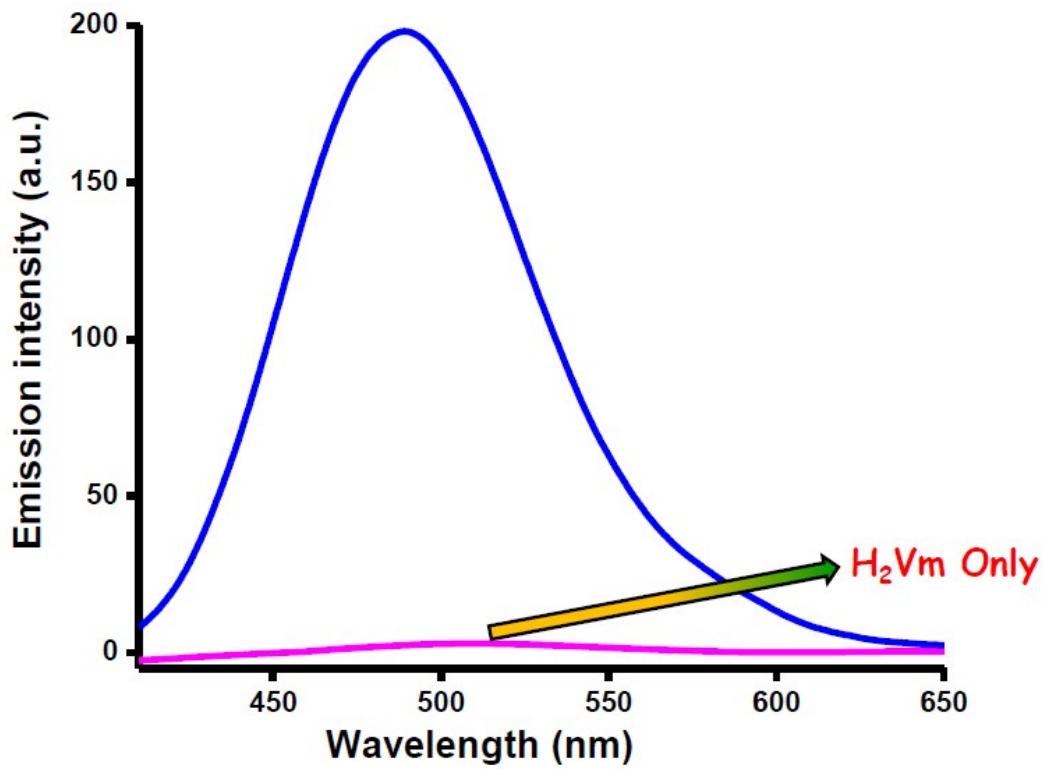
$$\frac{1}{(A - A_0)} = \frac{1}{\{K(A_{max} - A_0)[C]\}} + \frac{1}{(A_{max} - A_0)}$$

Where  $A_0$  is the absorbance of free ligand,  $A$  is the observed absorbance at that particular wavelength in the presence of a certain concentration of the metal ion  $[C]$ ,  $A_{max}$  is the maximum absorbance value of the complex formed.  $K$  is the association constant ( $\text{M}^{-1}$ ) and was determined from the slope of the linear plot and  $[C]$  is the concentration of the  $\text{Al}^{3+}$  ion added during titration studies. The goodness of the linear fit of the B–H plot of  $1/(A - A_0)$  vs.  $1/[\text{Al}^{3+}]$  for 1:1 complex formation confirms the binding stoichiometry between  $\text{H}_2\text{Vm}$  and  $\text{Al}^{3+}$ .



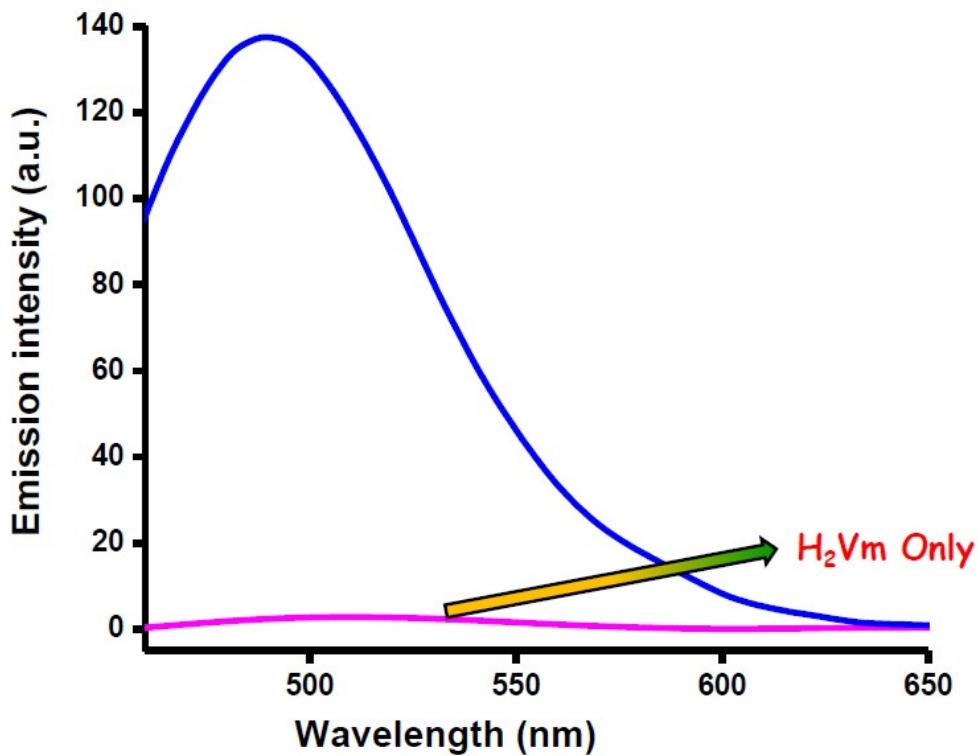
**Figure S11.** Benesi-Hildebrand plot of absorbance titration curve of  $\text{H}_2\text{Vm}$  and  $[\text{Al}^{3+}]$ .

## 2. Photophysical Characterization.



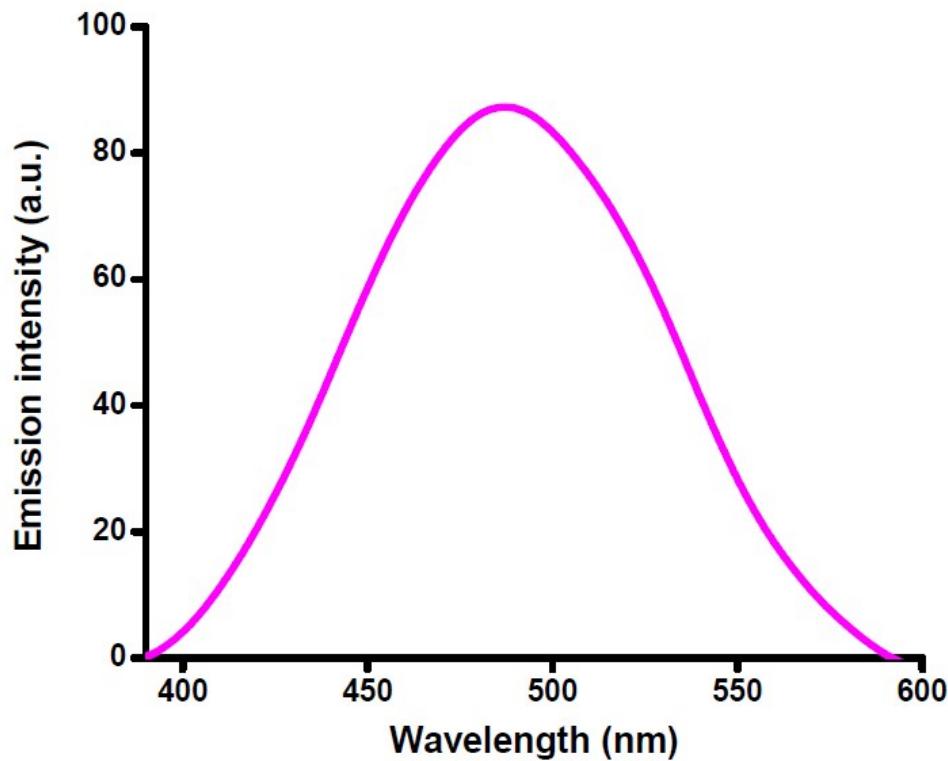
**Figure S12.** Emission spectra of **H<sub>2</sub>Vm** in the presence of [Al<sup>3+</sup>] in HEPES buffer (pH = 7.4) solution ( $\lambda_{\text{ex}} = 400$  nm,  $\lambda_{\text{em}} = 488$  nm).

## 2. Photophysical Characterization.



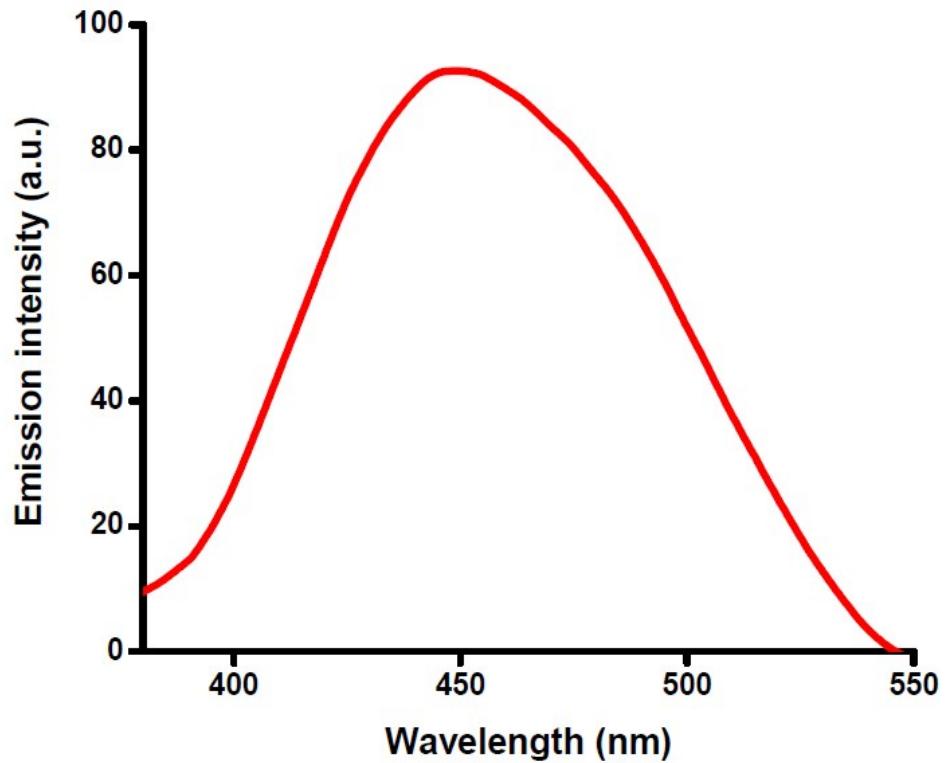
**Figure S13.** Emission spectra of **H<sub>2</sub>Vm** in the presence of [Al<sup>3+</sup>] in HEPES buffer (pH = 7.4) solution ( $\lambda_{\text{ex}} = 450$  nm,  $\lambda_{\text{em}} = 488$  nm).

## 2. Photophysical Characterization.



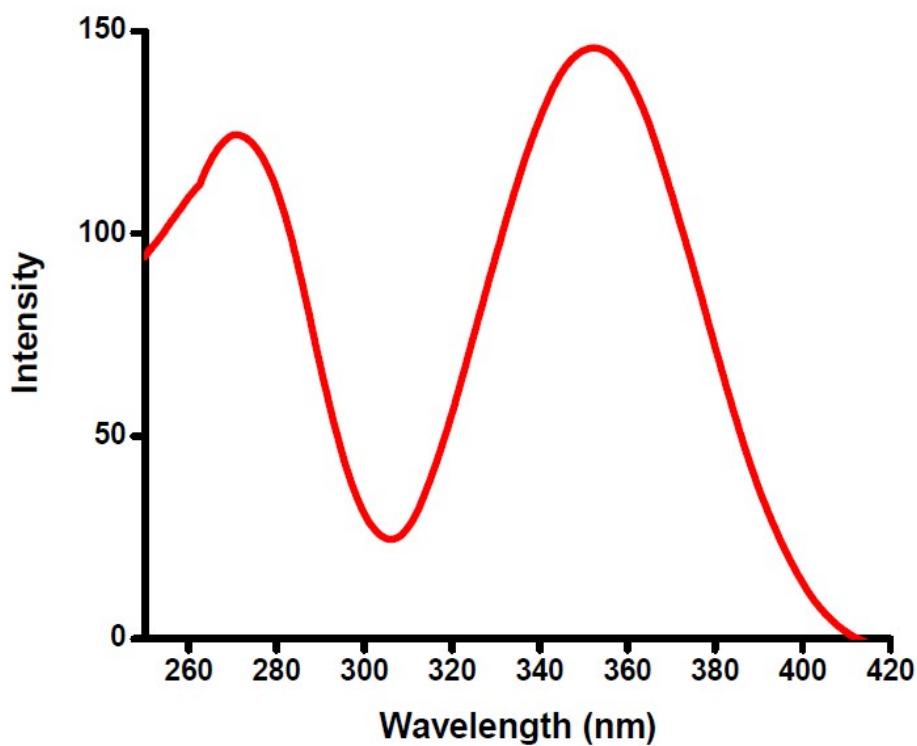
**Figure S14.** Emission spectra of  $\text{H}_2\text{Vm}$  in methanol solution.

## 2. Photophysical Characterization.



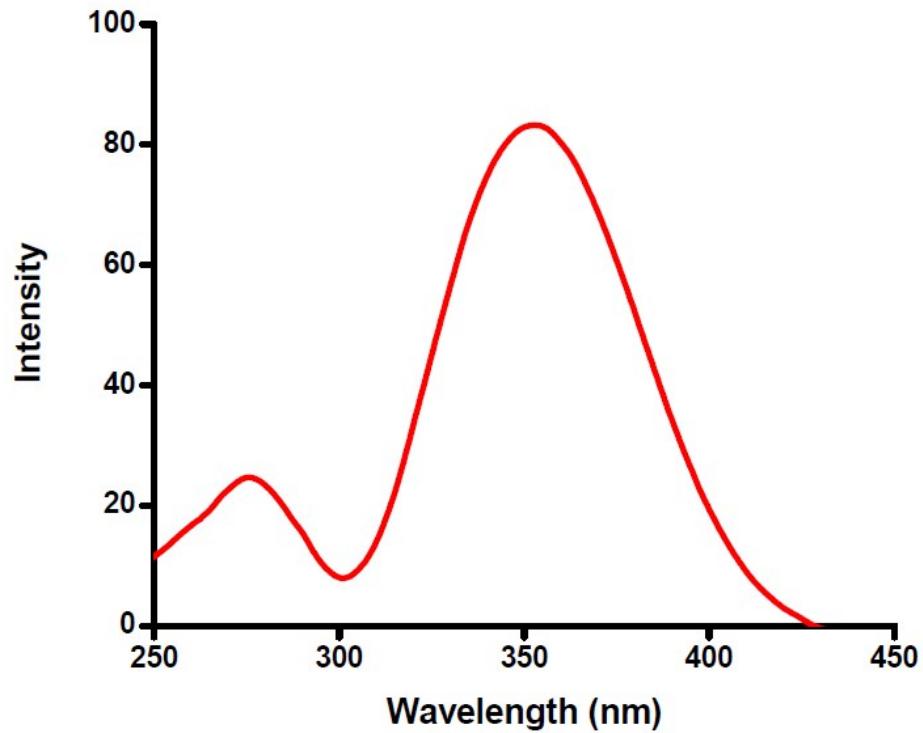
**Figure S15.** Emission spectra of  $\text{H}_2\text{Vm}$  in HEPES buffer ( $\text{pH} = 7.4$ ) solution.

## 2. Photophysical Characterization.



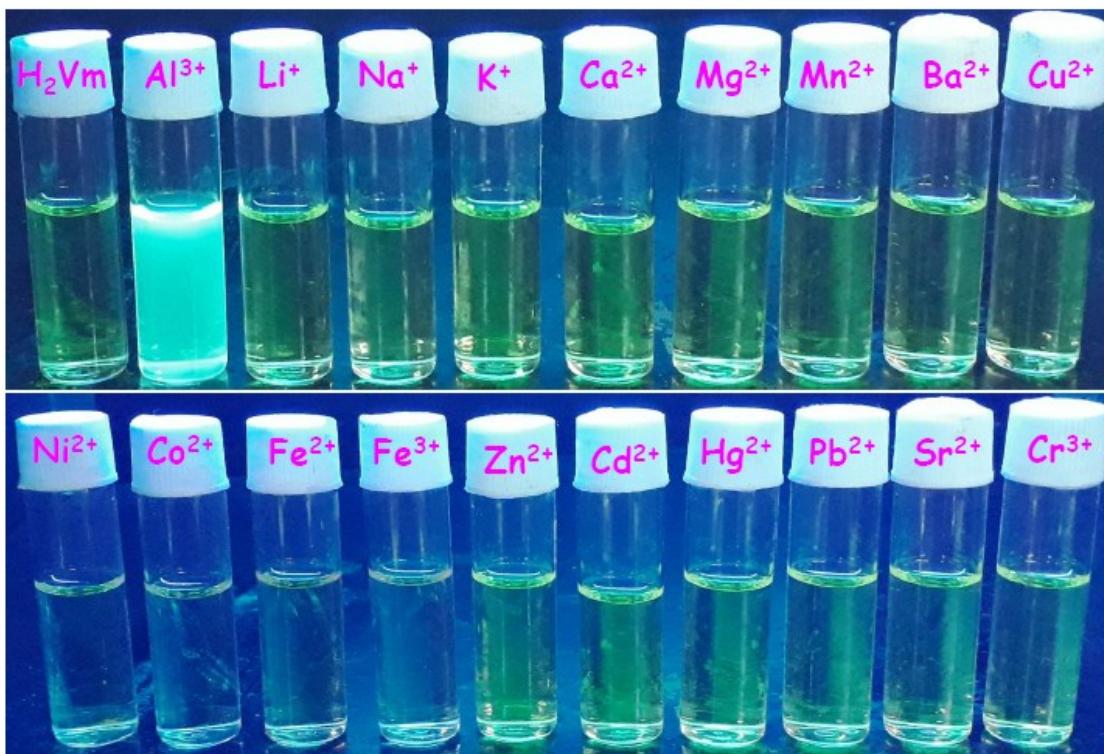
**Figure S16.** Excitation spectra of  $\text{H}_2\text{Vm}$  in methanol solution.

## 2. Photophysical Characterization.



**Figure S17.** Excitation spectra of  $\mathbf{H}_2\mathbf{Vm}$  in HEPES buffer ( $\text{pH} = 7.4$ ) solution.

## 2. Photophysical Characterization.



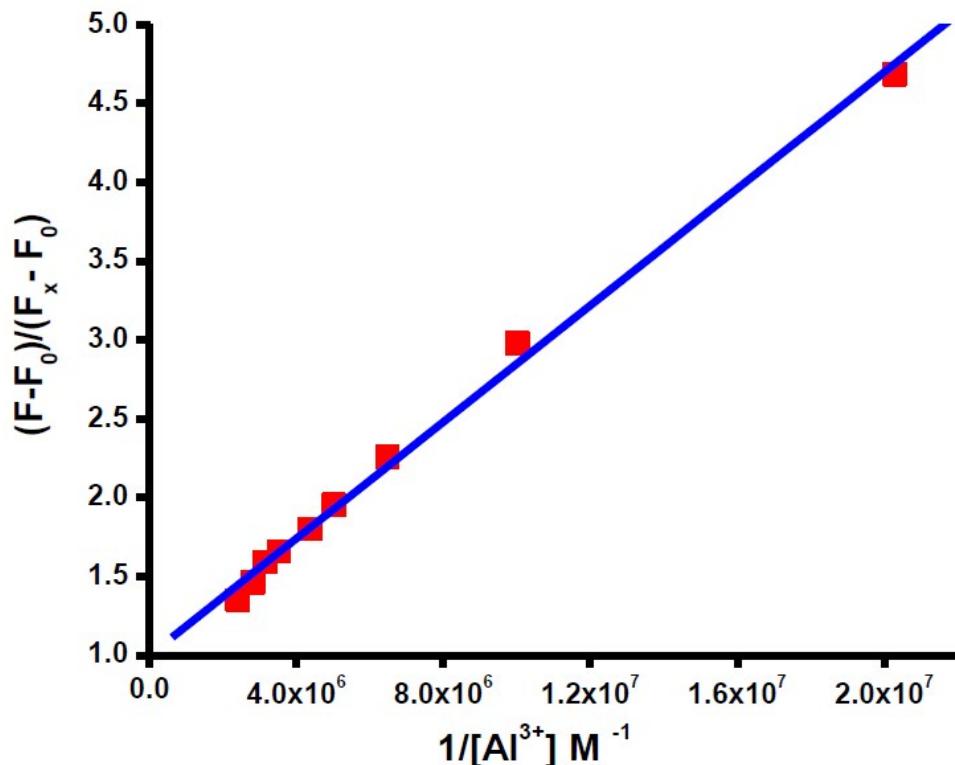
**Figure S18.** Visual color change observed with addition of different metal ions to **H<sub>2</sub>Vm** as seen under UV light ( $\lambda = 365$  nm).

## 2. Photophysical Characterization.

According to the linear Benesi–Hildebrand expression, the measured fluorescence intensity ( $F - F_0$ )/( $F_x - F_0$ ) at 488 nm varied as a function of  $1/[Al^{3+}]$  in a linear relationship, which indicates the formation of 1 : 1 stoichiometry between  $Al^{3+}$  and **H<sub>2</sub>Vm** in the complex.

$$\frac{1}{F_x - F_0} = \frac{1}{F_{max} - F_0} + \frac{1}{K[C]} \left( \frac{1}{F_{max} - F_0} \right)$$

where  $F_0$ ,  $F_x$  and  $F_{max}$  are the emission intensities of organic moiety considered in the absence of  $Al^{3+}$  ions, at an intermediate  $Al^{3+}$  concentration and at a concentration of complete interaction, respectively,  $K$  is the binding constant and  $[C]$  is the concentration of  $Al^{3+}$  ions.



**Figure S19.** Benesi-Hildebrand plot  $[(F - F_0)/(F_x - F_0)]$  vs.  $1/[Al^{3+}]$  for complexation between **H<sub>2</sub>Vm** and  $Al^{3+}$  derived from emission titration curve.

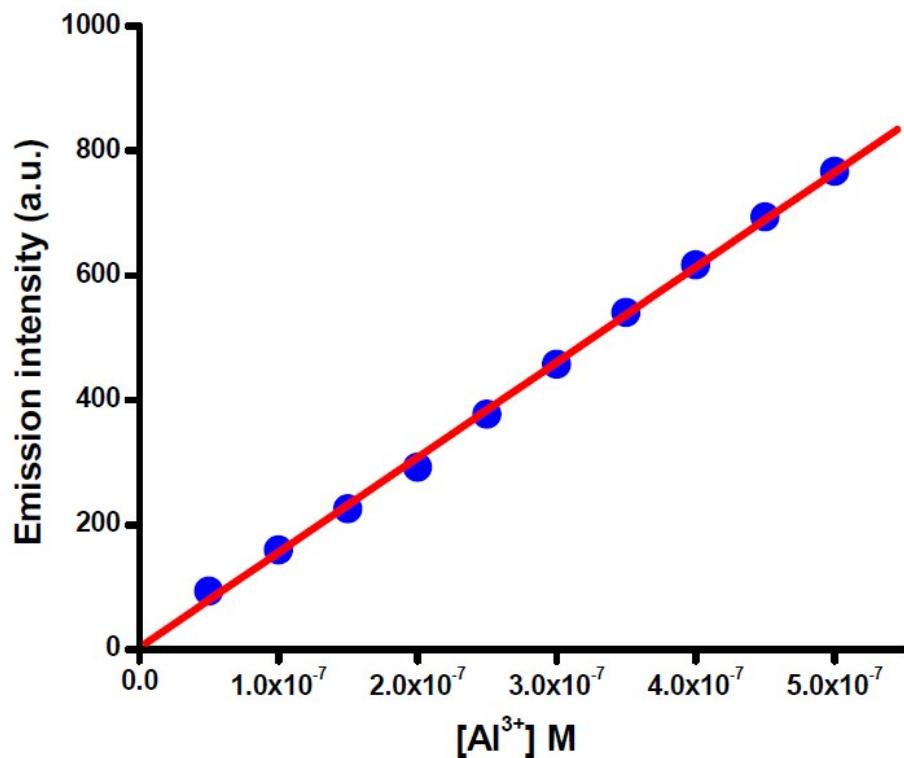
## 2. Photophysical Characterization.

**Detection limit** calculation in emission spectroscopy.

The limit of detection (**LOD**) of **Al<sub>2</sub>-Vm<sub>2</sub>** was measured on the basis of fluorescence titration measurement. The detection limit was calculated using the following equation:

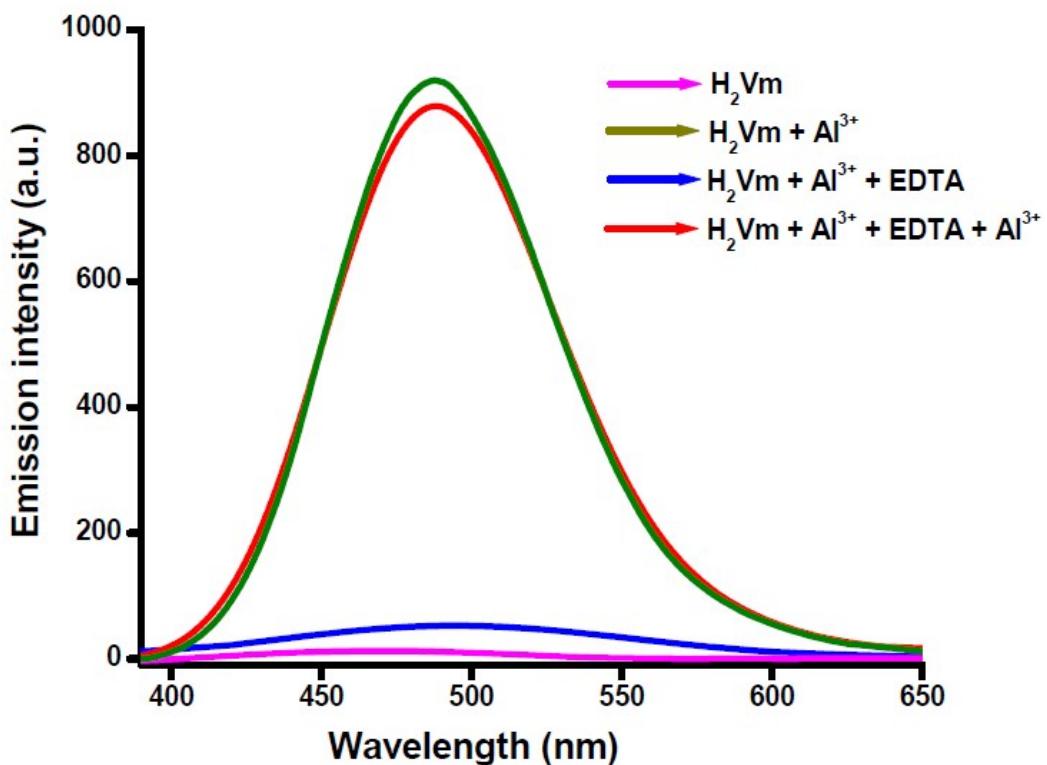
$$LOD = K \times \frac{\sigma}{S}$$

where  $K = 2$  or  $3$  (we take  $3$  in this case), ' $\sigma$ ' is the standard deviation of the blank solution and ' $S$ ' is the slope between the ratio of emission intensity *versus*  $[Al^{3+}]$ .



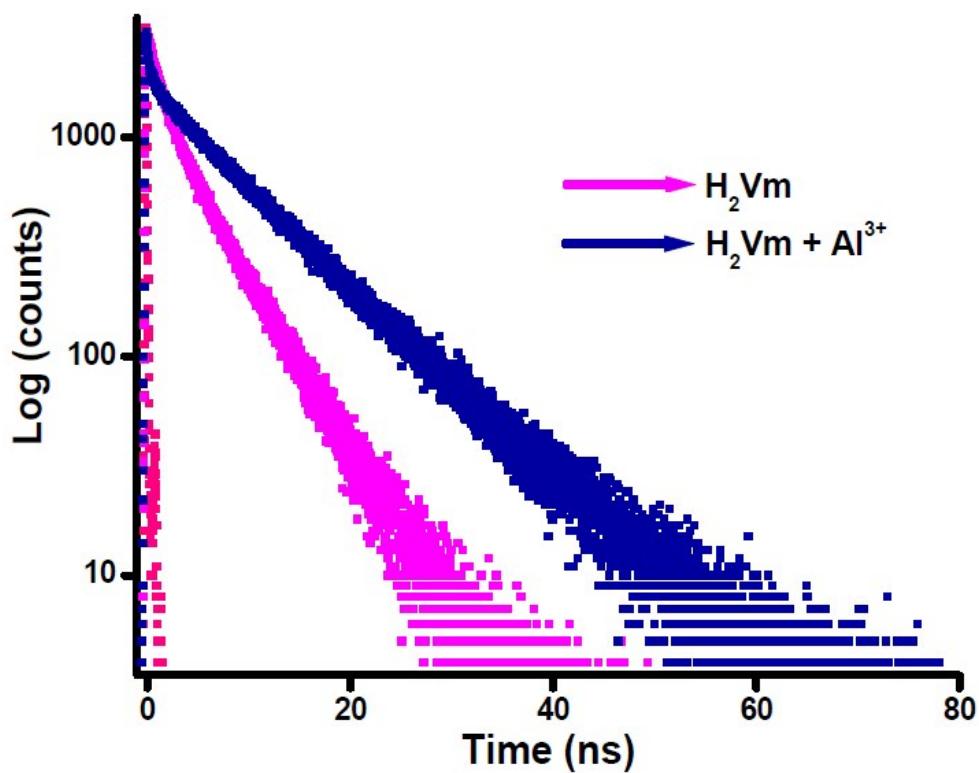
**Figure S20.** The limit of detection (LOD) of **H<sub>2</sub>Vm** for Al<sup>3+</sup> fluorescence responses ( $\lambda_{em} = 488$  nm) as a function of Al<sup>3+</sup> concentration.

## 2. Photophysical Characterization.



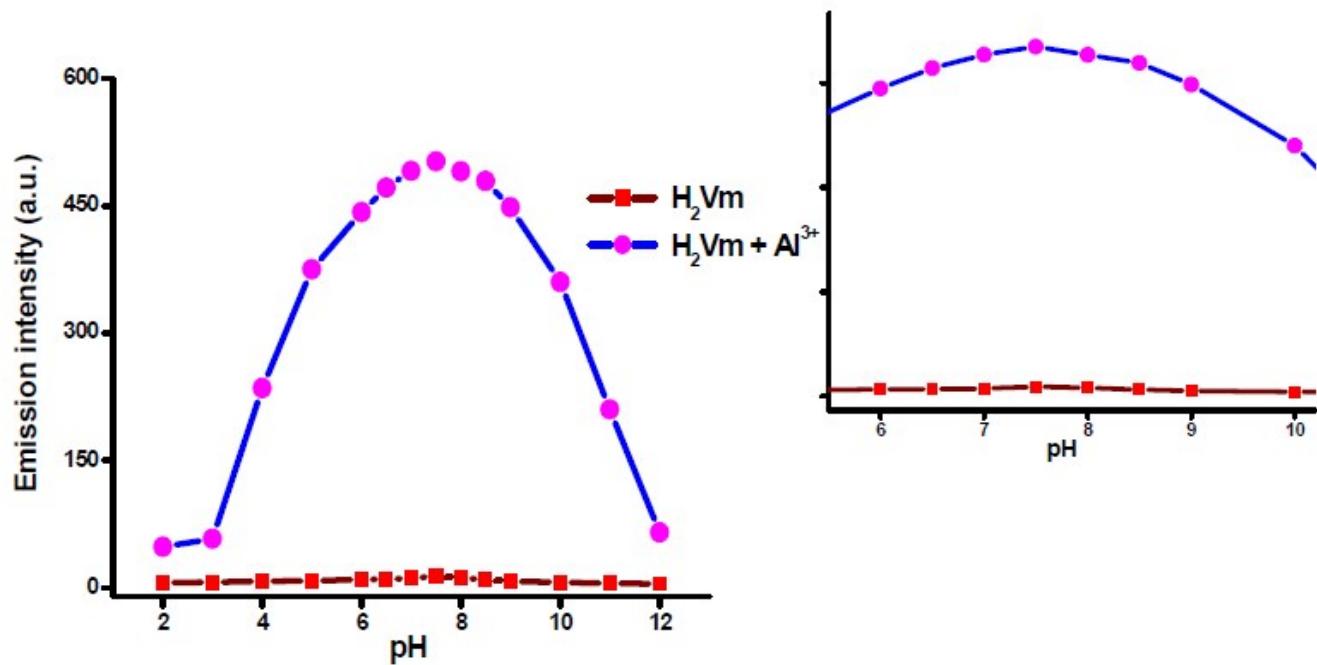
**Figure S21.** Fluorescence emission spectra of  $\text{H}_2\text{Vm}$  in the presence of  $\text{Al}^{3+}$  ion followed by addition of EDTA.

## 2. Photophysical Characterization.



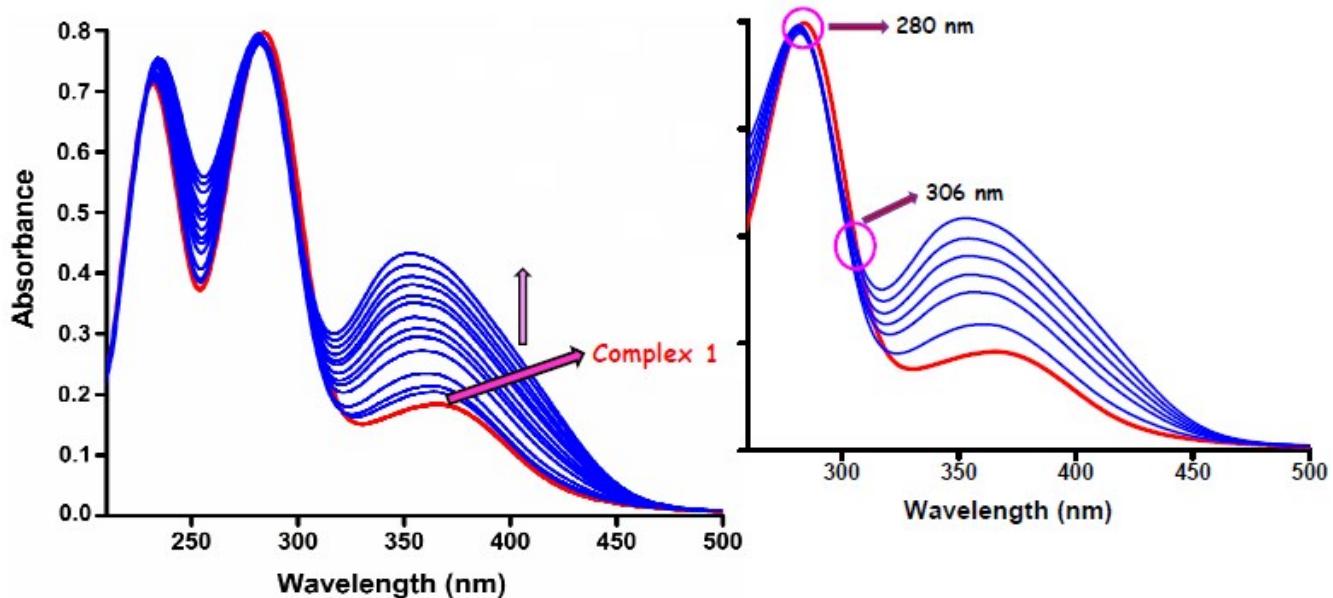
**Figure S22.** Time-resolved fluorescence decay of  $\text{H}_2\text{Vm}$  in the absence and presence of added  $\text{Al}^{3+}$  solution at 375 nm.

## 2. Photophysical Characterization.



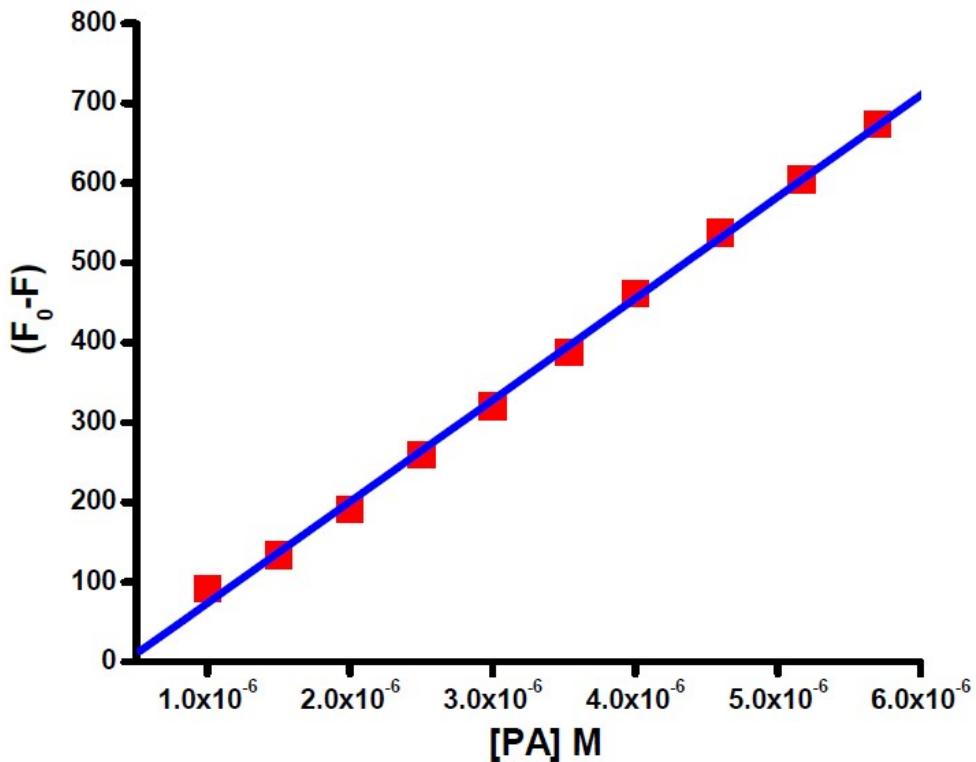
**Figure S23.** Emission intensity of probe **H<sub>2</sub>Vm** ( $5 \times 10^{-7}$  M) in absence and in presence of  $\text{Al}^{3+}$  as a function of pH values in aqueous solution at 488 nm. **Inset:** pH plot of zoomed graphic of **H<sub>2</sub>Vm** and in presence of  $\text{Al}^{3+}$ .

## 2. Photophysical Characterization.



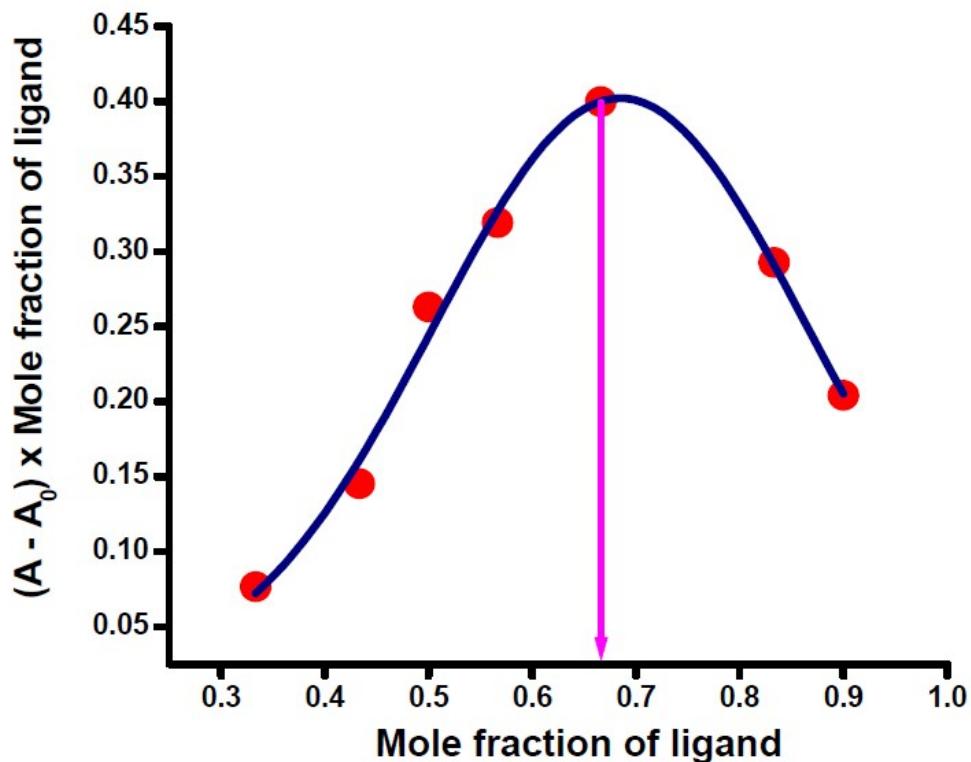
**Figure S24.** UV-vis spectra of complex **1** ( $5 \times 10^{-7}$  M) in HEPES buffer (pH = 7.4) solution in the presence of various concentration of **PA** (0, 0.5, 1, 1.5, 2, 2.5, 3, 4, 5, 6, 7, 8, 9 and 10)  $\times 10^{-7}$  M. **Inset:** UV-vis spectra of zoomed graphic of complex **1**.

## 2. Photophysical Characterization.



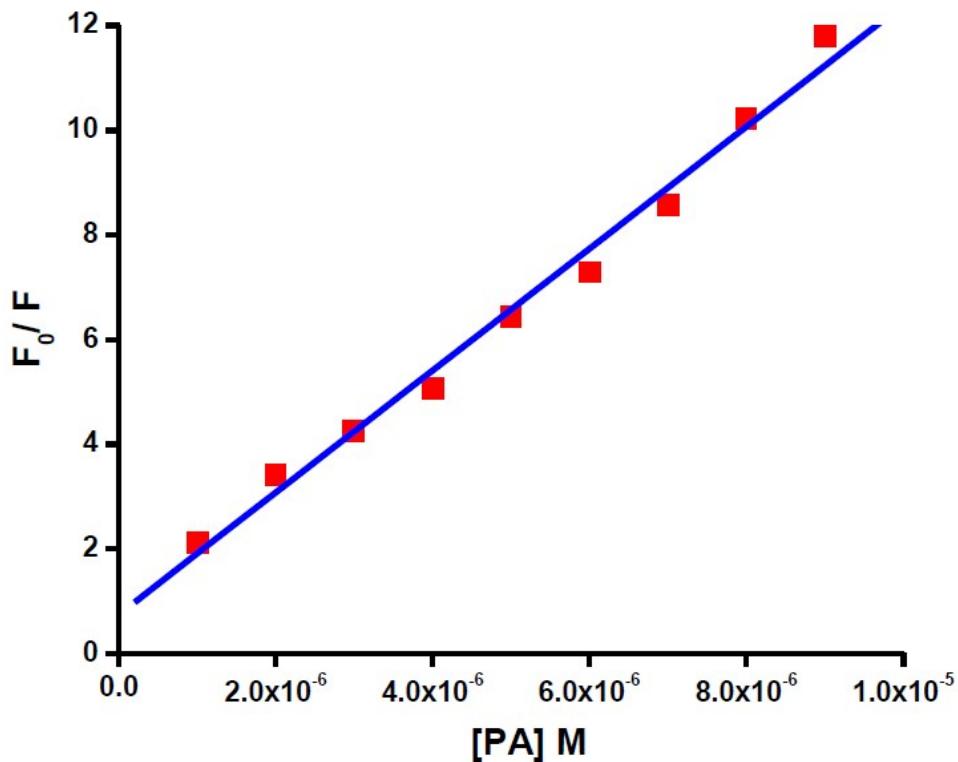
**Figure S25.** The limit of detection (LOD) of complex 1 for PA fluorescence responses ( $\lambda_{\text{em}} = 484$  nm) as a function of PA concentration.

## 2. Photophysical Characterization.



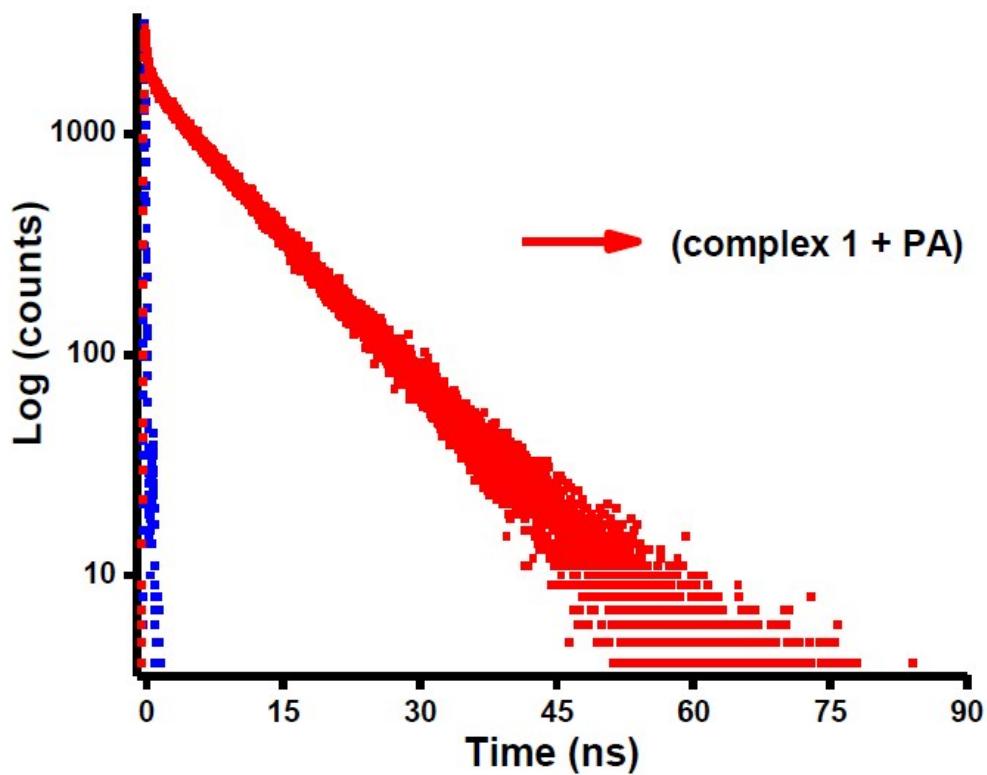
**Figure S26.** Job's plot for the identification of complex **1**-PA (1:2) complex stoichiometry using absorbance values at 364 nm.

## 2. Photophysical Characterization.



**Figure S27.** Stern–Volmer plot of fluorescence quenching for complexation between complex **1** and PA derived from emission titration curve.

## 2. Photophysical Characterization.



**Figure S28.** Time–resolved fluorescence decay of complex **1** with PA in solution at 375 nm.

**Table S2.** Fluorescence lifetime measurement of chemosensor **H<sub>2</sub>Vm**, complex **1** and complex **2** in aqueous solution.

	$\Phi_f$	$\tau_{av}$ (ns)	$K_r(\times 10^9)$ ( $S^{-1}$ )	$K_{nr}(\times 10^9)$ ( $S^{-1}$ )	$\chi^2$
<b>H<sub>2</sub>Vm</b>	0.036	0.074	0.486	13.027	1.010
<b>H<sub>2</sub>Vm + Al<sup>3+</sup> (complex 1)</b>	0.511	0.579	0.883	0.845	1.028
<b>Al<sub>2</sub>-Vm<sub>2</sub> complex + PA (complex 2)</b>	0.095	0.528	0.179	1.714	1.019

## 2. Photophysical Characterization.

**Table S3.** Stability constant was compared with complexation properties of the other metal ions.

Different metal ions	Stability constant from absorption data (M <sup>-1</sup> )	Stability constant from fluorescence data (M <sup>-1</sup> )
Al <sup>3+</sup>	$5.23 \times 10^5$	$4.19 \times 10^5$
Li <sup>+</sup>	$1.021 \times 10^4$	$1.104 \times 10^4$
Na <sup>+</sup>	$1.137 \times 10^4$	$1.149 \times 10^4$
K <sup>+</sup>	$1.276 \times 10^4$	$1.257 \times 10^4$
Ca <sup>2+</sup>	$1.314 \times 10^4$	$1.379 \times 10^4$
Mg <sup>2+</sup>	$1.576 \times 10^4$	$1.681 \times 10^4$
Mn <sup>2+</sup>	$1.823 \times 10^4$	$1.885 \times 10^4$
Ba <sup>2+</sup>	$1.199 \times 10^4$	$1.205 \times 10^4$
Cu <sup>2+</sup>	$1.079 \times 10^4$	$1.104 \times 10^4$
Fe <sup>2+</sup>	$1.753 \times 10^4$	$1.767 \times 10^4$
Zn <sup>2+</sup>	$2.803 \times 10^4$	$2.911 \times 10^4$
Cd <sup>2+</sup>	$2.158 \times 10^4$	$2.219 \times 10^4$
Hg <sup>2+</sup>	$1.812 \times 10^4$	$1.915 \times 10^4$
Ni <sup>2+</sup>	$1.865 \times 10^4$	$1.891 \times 10^4$
Pb <sup>2+</sup>	$1.943 \times 10^4$	$1.939 \times 10^4$
Sr <sup>2+</sup>	$1.087 \times 10^4$	$1.109 \times 10^4$
Co <sup>2+</sup>	$1.342 \times 10^4$	$1.351 \times 10^4$
Cr <sup>3+</sup>	$1.265 \times 10^4$	$1.278 \times 10^4$

### 3. Theoretical Data.

**Table S4.** Selected parameters for the vertical excitation (UV-vis absorption) of **H<sub>2</sub>Vm** (bis-keto form), electronic excitation energy (eV) and oscillator strength (*f*), and composition of the low-lying excited state of **H<sub>2</sub>Vm**; calculation of the S<sub>0</sub>–S<sub>n</sub> energy gaps based on optimized ground-state geometries (UV-vis absorption, H<sub>2</sub>O used as solvent). Only those transitions that contribute higher than 10% are given

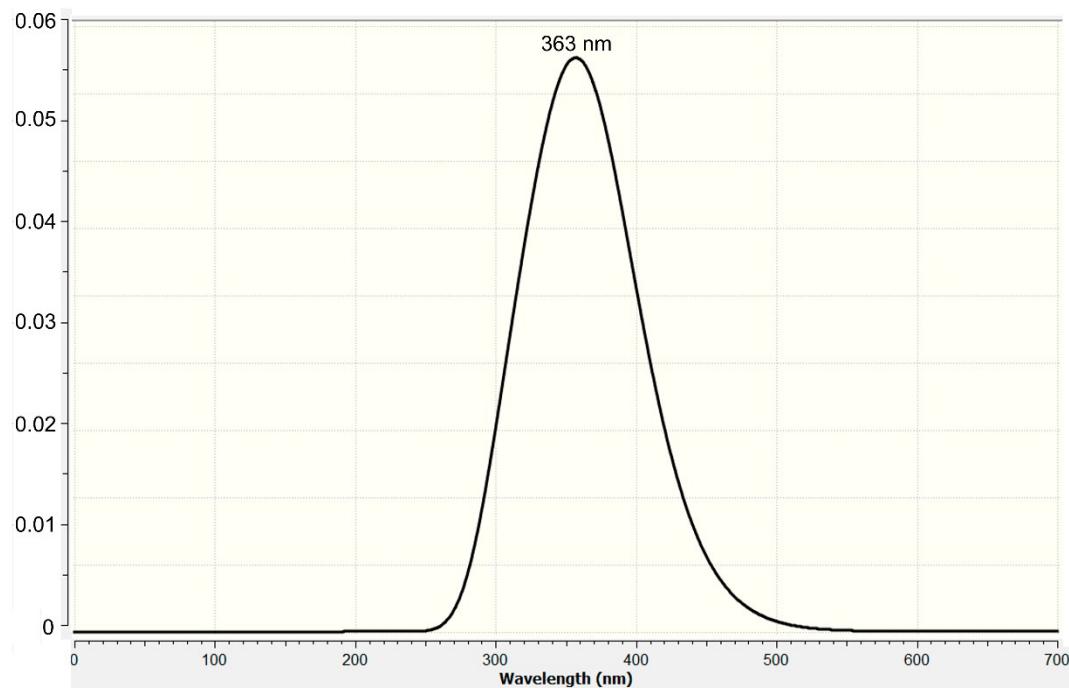
Process	Electronic transitions	Composition	Excitation energy	Oscillator strength (f)	contribution	$\lambda_{\text{exp}}$ (nm)
Absorption	S <sub>0</sub> → S <sub>1</sub>	HOMO → LUMO HOMO-1 → LUMO	3.0108 eV (412 nm)	0.1111	87% 13%	420

**Table S5.** Selected parameters for the vertical excitation (UV-vis absorptions) and the emission of **Al<sub>2</sub>-Vm<sub>2</sub>**, electronic excitation energies (eV) and oscillator strengths (*f*) and contributions of the lowest lying excited state; calculation of the S<sub>0</sub>–S<sub>n</sub> energy gap is based on optimized ground-state geometries (UV-vis absorption) (H<sub>2</sub>O used as solvent).

Process	Electronic transitions <sup>1</sup>	Composition	Excitation energy	Oscillator strength (f)	contribution	$\lambda_{\text{exp}}$ (nm)
Absorption	S <sub>0</sub> → S <sub>8</sub>	HOMO-3 → LUMO HOMO-2 → LUMO+1 HOMO-1 → LUMO+2	3.4147 eV (363 nm)	0.0822	36% 36% 28%	368

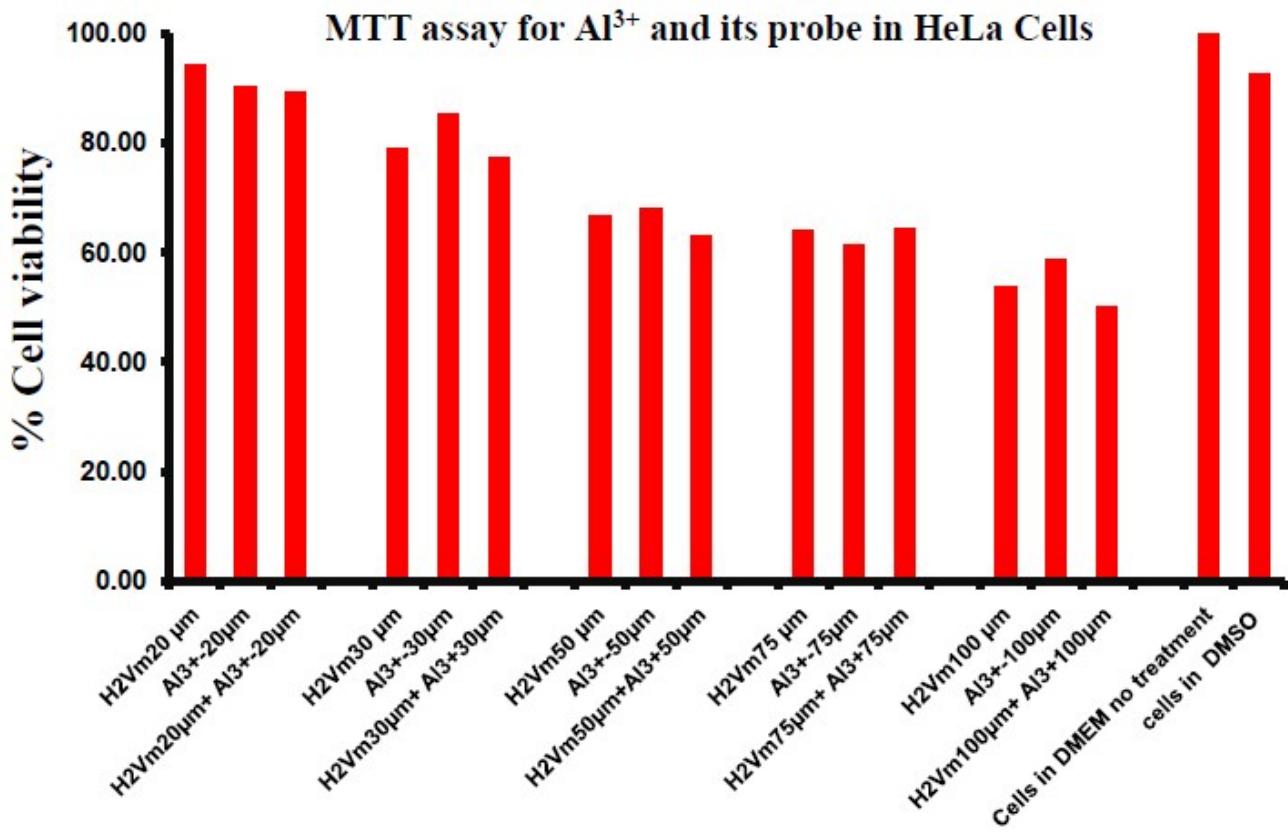
<sup>1</sup>S<sub>0</sub> → S<sub>1</sub>*f* = 0.0001; S<sub>0</sub> → S<sub>2</sub>*f* = 0.0005; S<sub>0</sub> → S<sub>3</sub>*f* = 0.0000; S<sub>0</sub> → S<sub>4</sub>*f* = 0.0003; S<sub>0</sub> → S<sub>5</sub>*f* = 0.0007; S<sub>0</sub> → S<sub>6</sub>*f* = 0.0005; S<sub>0</sub> → S<sub>7</sub>*f* = 0.0003.

### 3. Theoretical Data.



**Figure S29.** Theoretical UV-vis spectrum of  $\text{Al}_2\text{-Vm}_2$ .

#### 4. Cell study.



**Figure S30.** Percentage (%) cell viability of Hela cells treated with different concentrations of H<sub>2</sub>Vm for 24 hours determined by MTT assay.

## 5. Cartesian Coordinates.

Enol

C	1.70649300	-1.12030300	5.47524000
C	0.97424100	-0.31509500	4.57345700
C	-0.35273500	0.06388600	4.88641100
C	-0.92779300	-0.37196200	6.10961600
C	-0.18242200	-1.16366200	6.98042000
C	1.13657300	-1.53967800	6.66377300
H	2.72504900	-1.40402100	5.22008600
H	-0.61707800	-1.49839300	7.91587100
H	1.69762100	-2.15781800	7.35851100
O	-1.10423900	0.82373200	4.07546200
H	-0.56156600	1.03699400	3.26600900
O	-2.20724300	0.04232200	6.32644100
C	1.59982200	0.11878600	3.33213400
H	2.63074400	-0.23136600	3.16026200
N	1.00653800	0.86882400	2.47355600
C	-2.85332800	-0.36439700	7.52305500
H	-2.32846100	0.01402900	8.41083200
H	-3.85333800	0.07054300	7.48012900
H	-2.93289300	-1.45834700	7.58319100
C	1.70247200	1.25290900	1.26336100
H	2.69504900	0.77358500	1.19738300
H	1.86589800	2.34199400	1.29107200
C	0.88459100	0.91790400	0.00000000
C	1.70649300	-1.12030300	-5.47524000
C	0.97424100	-0.31509500	-4.57345700
C	-0.35273500	0.06388600	-4.88641100
C	-0.92779300	-0.37196200	-6.10961600
C	-0.18242200	-1.16366200	-6.98042000
C	1.13657300	-1.53967800	-6.66377300
H	2.72504900	-1.40402100	-5.22008600
H	-0.61707800	-1.49839300	-7.91587100
H	1.69762100	-2.15781800	-7.35851100
O	-1.10423900	0.82373200	-4.07546200
H	-0.56156600	1.03699400	-3.26600900
O	-2.20724300	0.04232200	-6.32644100
C	1.59982200	0.11878600	-3.33213400
H	2.63074400	-0.23136600	-3.16026200
N	1.00653800	0.86882400	-2.47355600
C	-2.85332800	-0.36439700	-7.52305500
H	-2.93289300	-1.45834700	-7.58319100
H	-3.85333800	0.07054300	-7.48012900
H	-2.32846100	0.01402900	-8.41083200
C	1.70247200	1.25290900	-1.26336100
H	1.86589800	2.34199400	-1.29107200
H	2.69504900	0.77358500	-1.19738300

H	0.65448400	-0.15342200	0.00000000
O	-0.38551900	1.55120600	0.00000000
H	-0.26762800	2.51575100	0.00000000
Keto			
C	1.74319400	-1.03868300	5.62238300
C	1.00073200	-0.29309700	4.64566600
C	-0.39063300	0.09094600	4.88735600
C	-0.95573700	-0.33082000	6.17334600
C	-0.20302400	-1.04601400	7.07424100
C	1.15698700	-1.40426500	6.80009600
H	2.77627800	-1.30642300	5.40736500
H	-0.63270400	-1.35475500	8.02199000
H	1.71044400	-1.96906400	7.54464900
O	-1.06588600	0.74551200	4.05034600
O	-2.24759500	0.05265300	6.34890900
C	1.61567500	0.08983900	3.45101100
H	2.66156700	-0.17247700	3.28412100
N	1.01615900	0.76969100	2.47647700
C	-2.89565600	-0.31598500	7.55507100
H	-2.39207900	0.11994600	8.42926800
H	-3.90901300	0.08297100	7.48208600
H	-2.93871600	-1.40829700	7.66759200
C	1.66798500	1.22426600	1.26573800
H	2.69313000	0.83373900	1.25285600
H	1.73834000	2.32392800	1.27705100
C	0.91648500	0.77762100	0.00000000
C	1.74319400	-1.03868300	-5.62238300
C	1.00073200	-0.29309700	-4.64566600
C	-0.39063300	0.09094600	-4.88735600
C	-0.95573700	-0.33082000	-6.17334600
C	-0.20302400	-1.04601400	-7.07424100
C	1.15698700	-1.40426500	-6.80009600
H	2.77627800	-1.30642300	-5.40736500
H	-0.63270400	-1.35475500	-8.02199000
H	1.71044400	-1.96906400	-7.54464900
O	-1.06588600	0.74551200	-4.05034600
O	-2.24759500	0.05265300	-6.34890900
C	1.61567500	0.08983900	-3.45101100
H	2.66156700	-0.17247700	-3.28412100
N	1.01615900	0.76969100	-2.47647700
C	-2.89565600	-0.31598500	-7.55507100
H	-2.93871600	-1.40829700	-7.66759200
H	-3.90901300	0.08297100	-7.48208600
H	-2.39207900	0.11994600	-8.42926800
C	1.66798500	1.22426600	-1.26573800
H	1.73834000	2.32392800	-1.27705100
H	2.69313000	0.83373900	-1.25285600
H	0.82997200	-0.31409200	0.00000000

O	-0.42786100	1.23725800	0.00000000
H	-0.45202100	2.20964600	0.00000000
H	0.02337300	0.99048600	2.67779100
H	0.02337300	0.99048600	-2.67779100
Keto-enol			
C	5.56533200	-1.75001200	0.96656900
C	4.62897900	-0.97929000	0.23962900
C	4.91608800	0.37223800	-0.06526400
C	6.14760200	0.93423000	0.36363800
C	7.05196300	0.15093800	1.07724800
C	6.76143900	-1.19272800	1.38028400
H	5.33037800	-2.78719200	1.19452300
H	7.99397500	0.57484600	1.40722800
H	7.48246300	-1.78249500	1.93856200
O	4.07174300	1.16038700	-0.74849600
O	6.33671900	2.23870800	0.02323300
C	3.38083300	-1.59247800	-0.18850000
H	3.23478500	-2.64528100	0.10246300
N	2.48521700	-0.96844400	-0.86867100
C	7.53958800	2.87417500	0.43036300
H	8.41942400	2.38563400	-0.00983400
H	7.47046500	3.89868200	0.06103300
H	7.63516400	2.88559800	1.52447000
C	1.27418200	-1.66414800	-1.25023700
H	1.24391800	-2.68848000	-0.83863000
H	1.26062400	-1.75527900	-2.34848700
C	0.01109200	-0.90426500	-0.80916200
C	-5.57138900	-1.71399800	1.13701800
C	-4.61776000	-0.99168900	0.34421500
C	-4.88432200	0.37821100	-0.09493900
C	-6.17182000	0.94284100	0.32439400
C	-7.05084800	0.20993100	1.08597800
C	-6.75197000	-1.12939800	1.49740800
H	-5.33718800	-2.73177100	1.44474200
H	-7.99953300	0.64005800	1.39147300
H	-7.47924700	-1.66859100	2.09750700
O	-4.06788700	1.03574600	-0.79119600
O	-6.37263000	2.21537900	-0.11076600
C	-3.42052200	-1.60913300	-0.03113100
H	-3.23874800	-2.64035900	0.27579700
N	-2.46285200	-1.03485100	-0.75218100
C	-7.58219600	2.86070500	0.24867800
H	-7.67927700	2.94959200	1.33987300
H	-7.53134200	3.85696600	-0.19461400
H	-8.45570300	2.32628700	-0.15073800
C	-1.25140700	-1.69803200	-1.19236200
H	-1.28025200	-1.82234500	-2.28606800
H	-1.22143700	-2.70015200	-0.74713400

H	0.03038200	-0.77227200	0.27853700
O	-0.05137200	0.41606500	-1.33061300
H	0.19109700	0.41241700	-2.27144600
H	-2.66923200	-0.05594900	-1.01929500
H	3.26095300	0.62427200	-0.96425800

Complex 1

C	2.29593100	-2.60401500	-2.00944800
C	3.33757800	-3.01267200	-2.88150800
H	3.11885600	-3.15585500	-3.93692000
C	4.60622300	-3.22671500	-2.38820400
H	5.40619000	-3.55366700	-3.04535700
C	4.87996400	-3.01510800	-1.01912500
H	5.88669300	-3.17867900	-0.64994300
C	3.88036400	-2.60417500	-0.14882500
C	2.55028600	-2.41787200	-0.63049200
C	1.01640300	-2.23799500	-2.55073400
H	0.95440500	-2.22024200	-3.64393200
C	-1.18078900	-1.37968600	-2.66083400
H	-2.07143200	-1.93884600	-2.37668600
H	-0.99352400	-1.52421900	-3.73321100
C	-1.47214100	0.10994900	-2.41020400
C	-0.32076900	1.08368600	-2.76320600
H	0.16895000	0.75068000	-3.68817000
H	-0.78492800	2.05002000	-2.96983200
C	1.93182700	1.06792800	-2.06396300
H	2.09523400	0.61616800	-3.04680100
C	3.13359800	1.29822100	-1.30977200
C	4.36776100	0.87607700	-1.85594200
H	4.37493700	0.41694700	-2.84120400
C	5.54433000	1.01681100	-1.14418900
H	6.48909400	0.69209400	-1.56774200
C	5.50210700	1.56619900	0.14944500
H	6.40404200	1.65827400	0.74808200
C	4.30506800	1.98609000	0.70690000
C	3.09025700	1.89675100	-0.02320900
C	4.03239600	3.79988100	2.23895900
H	3.03848400	4.08156300	1.88276400
H	4.79715900	4.40029100	1.73220500
H	4.10127000	3.95688900	3.31746600
C	-2.63464500	2.40199500	-0.08938100
C	-3.81950200	2.63416800	-0.83684800
C	-5.06790100	2.47311200	-0.25789700
H	-5.94361400	2.66552700	-0.87141400
C	-5.19812300	2.05197000	1.07744100
H	-6.18281100	1.93120400	1.51702000
C	-4.06016200	1.77254000	1.81017900
H	-4.14098500	1.40619300	2.83038300
C	-2.77454600	1.92864100	1.24188500

C	-1.62734400	1.55404100	2.02138700
H	-1.85996500	1.21656300	3.03574700
C	0.59778000	1.23502700	2.73454800
H	0.06279800	1.06944600	3.67944300
H	1.22422500	2.11958000	2.86613100
C	1.56011200	0.05122700	2.47034100
C	1.02047200	-1.33837800	2.85097900
H	1.80972100	-2.05543100	2.62511300
H	0.82063200	-1.35285900	3.93106200
C	-1.28689500	-1.83282700	2.80245600
H	-1.21193400	-1.74355300	3.89179800
C	-2.61542100	-2.01612100	2.28754000
C	-3.69954400	-2.20152400	3.18370100
H	-3.49547300	-2.31883100	4.24519800
C	-4.99094500	-2.23851800	2.70476200
H	-5.82560900	-2.39791900	3.38052700
C	-5.24287500	-2.06175600	1.32656500
H	-6.26683900	-2.08426000	0.96966300
C	-4.20047400	-1.86122600	0.43228000
C	-2.85271000	-1.86989900	0.90067100
C	-5.62581000	-1.54071900	-1.45455100
H	-5.47502600	-1.30169800	-2.50815700
H	-6.19505900	-0.73413000	-0.97335800
H	-6.18263800	-2.48303600	-1.36474700
C	5.34443400	-2.34090300	1.71864800
H	5.99252000	-1.64134300	1.17451900
H	5.79564700	-3.34224800	1.70358400
H	5.23079000	-2.00435200	2.75014200
C	-3.24748700	4.23692500	-2.51514700
H	-3.29563700	4.31090100	-3.60363000
H	-2.21760500	4.37328200	-2.17622200
H	-3.89455100	5.00077800	-2.06786000
N	-0.03870000	-1.87949600	-1.87976600
N	0.70388400	1.32692000	-1.71984600
N	-0.37525200	1.56672900	1.66666900
N	-0.19403300	-1.70691400	2.10732800
O	-1.87997700	-1.69460300	0.03497100
O	-4.32431200	-1.65099900	-0.90290000
O	1.61855000	-2.03484400	0.21193000
O	4.03375200	-2.34994400	1.17723200
O	1.97694600	2.31828100	0.52414100
O	4.28357900	2.39955800	2.02670500
O	-1.46787600	2.58266500	-0.65727600
O	-3.73380300	2.92391400	-2.18632800
H	1.83177600	0.02613400	1.41378600
H	-1.74347400	0.22600600	-1.35993200
O	2.70590800	0.21944000	3.29553500
H	3.22771500	0.96474700	2.93899000

O	-2.57143800	0.40109900	-3.26386400
H	-2.96055800	1.24921400	-2.97433100
Al	0.25731300	2.48167700	-0.06887500
Al	-0.17238700	-2.26655300	0.12143700
O	0.14358900	4.27192400	0.89343600
O	-0.25395200	-4.08412800	0.39774000
N	-1.01315700	-4.91056600	-0.35085700
O	-1.53810400	-4.44649800	-1.36400100
O	-1.10471200	-6.05896200	0.04086000
N	0.48290100	4.91921700	-0.17124800
O	0.59588800	6.11929600	-0.22213500
O	0.68772900	4.13305300	-1.17538300

Complex 2

C	-2.08921000	2.57751600	2.05799600
C	-2.41270300	3.69169500	2.85044800
H	-2.64083900	3.55595000	3.89020300
C	-2.44210500	4.93446000	2.29169500
H	-2.69902600	5.79132100	2.88029900
C	-2.13426900	5.09648500	0.93092600
H	-2.15646700	6.07898600	0.50830300
C	-1.81046900	4.02384700	0.15223300
C	-1.79920000	2.73325100	0.71313500
C	-1.96130900	1.26280300	2.65267000
H	-1.98155600	1.24672100	3.73291100
C	-1.46964100	-1.05280700	2.82996800
H	-2.12918100	-1.84802300	2.53081400
H	-1.59222200	-0.83657700	3.88622200
C	-0.05096700	-1.55052800	2.56328800
C	1.10567000	-0.58939300	2.88563100
H	0.93656400	-0.11576500	3.84724600
H	1.98891100	-1.19937200	2.96068800
C	1.42628000	1.68659900	2.30385700
H	1.20034500	1.84843700	3.34569300
C	1.70619500	2.89745300	1.57562800
C	1.70916700	4.12654400	2.24776100
H	1.54881200	4.14700600	3.30816300
C	1.90382800	5.28970900	1.55769900
H	1.90728100	6.23303100	2.06280200
C	2.08693500	5.23818300	0.17192800
H	2.22750800	6.14002600	-0.38861600
C	2.07741100	4.04845500	-0.49601800
C	1.90323700	2.84430500	0.20244800
C	3.56517000	3.69695800	-2.36610600
H	3.88900200	2.74583300	-1.97596500
H	4.25377800	4.47387900	-2.06446400
H	3.49125100	3.66551800	-3.44121500
C	1.83605900	-2.86846400	-0.17284400
C	1.98092500	-4.07637800	0.52581600

C	1.96318600	-5.26600900	-0.14213500
H	2.08176000	-6.17091000	0.41856600
C	1.78059600	-5.31322000	-1.52812800
H	1.76240400	-6.25636900	-2.03324000
C	1.61427200	-4.14579600	-2.21840900
H	1.45476600	-4.16248400	-3.27900500
C	1.63948500	-2.91696700	-1.54626200
C	1.38910200	-1.69985600	-2.27481400
H	1.16067600	-1.85633300	-3.31692500
C	1.12298900	0.58305800	-2.85694400
H	0.94392200	0.11353800	-3.81877000
H	2.02046700	1.17202600	-2.93092000
C	-0.01103800	1.57122100	-2.53599100
C	-1.44073700	1.10710800	-2.80440000
H	-2.08169200	1.91766900	-2.50603400
H	-1.56710300	0.89381600	-3.86080500
C	-1.98711900	-1.19625300	-2.62773200
H	-2.00566900	-1.17971400	-3.70799600
C	-2.14672600	-2.50757400	-2.03323200
C	-2.49545600	-3.61382200	-2.82609200
H	-2.71906400	-3.47274600	-3.86612100
C	-2.55485100	-4.85553900	-2.26739300
H	-2.83120300	-5.70610800	-2.85632000
C	-2.25257600	-5.02476300	-0.90625300
H	-2.29846200	-6.00646000	-0.48367200
C	-1.90450200	-3.96005200	-0.12715300
C	-1.86210400	-2.67008900	-0.68802200
C	-1.50333000	-5.23213900	1.93261900
H	-1.16579200	-4.93832700	2.91046200
H	-0.79574100	-5.91553700	1.47881300
H	-2.47245600	-5.71142300	1.99618200
C	-1.37689400	5.28608500	-1.90703200
H	-0.65392900	5.95260300	-1.45235800
H	-2.33436400	5.78810000	-1.97176500
H	-1.04519600	4.98437600	-2.88446900
C	3.47429300	-3.76005600	2.39771300
H	3.39983100	-3.72686500	3.47273300
H	3.82094900	-2.81684200	2.00797900
H	4.14474300	-4.55301200	2.09689600
N	-1.77772200	0.16388800	2.02859600
N	1.40654500	0.47969400	1.86828700
N	1.39732000	-0.49281500	-1.83925100
N	-1.77841100	-0.10196800	-2.00342000
O	-1.54628500	-1.64179900	0.07664400
O	-1.58086700	-3.99984000	1.19906600
O	-1.50680300	1.69778400	-0.05116200
O	-1.48437600	4.05596800	-1.17359100
O	1.89710000	1.69810000	-0.43860900

O	2.21855600	4.00701000	-1.87846700
O	1.85619100	-1.72242400	0.46822200
O	2.12133000	-4.03825600	1.90843600
H	0.04724900	1.88406500	-1.52675700
H	0.00114800	-1.86467500	1.55412100
O	0.19512500	2.68528000	-3.43535300
H	0.80757500	3.30084300	-2.99789900
O	0.12776100	-2.66912800	3.46288400
H	0.72604700	-3.29897600	3.02616400
Al	2.05573200	-0.01427500	0.01491500
Al	-2.07508100	0.03446500	0.01240800
O	3.96559900	-0.03681000	0.01607400
O	-3.98494700	0.05699900	0.01124900
C	7.31252700	-0.34067900	-1.20902500
C	8.02847400	-0.02953800	-0.05270200
C	7.35002900	0.26389100	1.13002100
C	5.95495000	0.24741800	1.15647500
C	5.23923400	-0.06319400	0.00027200
C	5.91805300	-0.35765300	-1.18248300
H	7.84762000	-0.57242800	-2.14128500
H	7.91428900	0.50853200	2.04164500
C	-5.35891400	0.01626900	-0.03081800
C	-6.05142500	-0.38974000	1.11026000
C	-7.44611300	-0.39697800	1.11411700
C	-8.14893700	0.00305600	-0.02308500
C	-7.45652600	0.40940600	-1.16369200
C	-6.06140800	0.41564900	-1.16771900
H	-7.99202500	-0.71758300	2.01327700
H	-8.01014600	0.72474300	-2.06009700
N	5.24017300	0.55680800	2.40318000
O	4.04108000	0.53435500	2.39373400
O	5.88465500	0.81932500	3.37997900
N	9.49810900	-0.01276900	-0.08085400
O	10.08482200	0.25843700	0.92941100
O	10.05172500	-0.27096400	-1.11296400
N	5.16348700	-0.68522400	-2.40077100
O	3.96544700	-0.69135500	-2.34524500
O	5.77604000	-0.93326500	-3.40160200
N	-9.61891000	-0.00461000	-0.01867800
O	-10.19403300	0.34477400	-1.01144400
O	-10.18438000	-0.35994400	0.97751000
N	-5.31067700	-0.81154600	2.30787000
O	-5.93460200	-1.15500600	3.27284200
O	-4.11198100	-0.79537600	2.27215600
N	-5.33196200	0.84370300	-2.37004200
O	-5.96495400	1.18119800	-3.33120600
O	-4.13297000	0.83834800	-2.34179000