Electronic Supplementary Information (ESI) for:

Preparation and Reactivity of Rhodium and Iridium Complexes Containing a Methylborohydride Based Unit Supported by Two 7-Azaindolyl Heterocycles,

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A. Crystallographic Information for Li[^{Me}Bai] and Complexes 1, 2, 3 and 2a.

1. Crystallographic parameters for Li[^{Me}Bai] and Complexes 1, 2, 3 and 2a.

Complex	Li(NCMe)₂[^{Me} Bai]	1	2	3	2a
Formula	C ₁₉ H ₂₀ BLiN ₆	C ₂₃ H ₂₆ BN ₄ Ir	C ₂₃ H ₂₆ BN ₄ Rh	C ₂₂ H ₂₂ BN ₄ Rh	C ₂₃ H ₂₆ BN ₄ Rh
D _{calc.} / g cm ⁻³	1.228	1.781	1.502	1.604	1.553
µ/mm⁻¹	0.076	6.393	0.834	0.919	0.863
Formula Weight	350.16	561.49	472.20	456.15	472.20
Colour	colourless	yellow	yellow	yellow	colourless
Size/mm ³	0.49 × 0.36 × 0.12	$0.05 \times 0.05 \times 0.04$	0.18 × 0.12 × 0.05	0.50 × 0.10 × 0.10	0.10×0.09×0.08
T/K	100(2)	100(2)	100(2)	100(2)	100(2)
Crystal System	monoclinic	trigonal	triclinic	monoclinic	monoclinic
Space Group	P2 ₁ /n	R-3	P-1	C2/c	P2 ₁ /n
a/Å	8.3408(4)	26.9680(16)	14.9965(3)	22.5002(3)	14.5731(3)
b/Å	15.0529(6)	26.9680(16)	16.0602(3)	10.5702(2)	16.5975(4)
c/Å	15.1212(6)	14.961(2)	16.3753(5)	15.9172(2)	17.2681(3)
α /°	90	90	111.4750(10)	90	90
βľ	94.024(2)	90	107.6280(10)	93.7480(10)	104.767(2)
γ/°	90	120	106.4890(10)	90	90
V/Å ³	1893.84(14)	9423.2(18)	3132.31(13)	3777.52(10)	4038.80(15)
Z	4	18	6	8	8
Z'	1	1	3	1	2
Wavelength/Å	0.71073	0.71073	0.71073	0.71073	0.71075
Radiation type	ΜοΚα	ΜοΚα	ΜοΚα	ΜοΚα	ΜοΚα
$\Theta_{min}/^{\circ}$	1.911	2.212	1.52	3.628	2.440
Θ_{max}	27.44498	25.342	27.531	55.106	27.483
Measured Refl.	17411	18758	87639	32574	46283
Independent Refl.	4351	3824	14384	4364	9257
Rint	0.0305	0.1040	0.0327	0.0189	0.0773
Parameters	250	263	787	257	532
Restraints	0	0	0	0	0
Largest Peak	0.29	2.59	1.40	1.22	2.043
Deepest Hole	-0.21	-2.89	-0.80	-0.40	-1.404
GooF	1.045	1.129	1.064	1.032	1.043
wR ₂ (all data)	0.0984	0.1283	0.0743	0.0544	0.1559
wR ₂	0.0930	0.1165	0.0701	0.0531	0.1407
R₁ (all data)	0.0455	0.1018	0.0362	0.0223	0.0771
R ₁	0.0375	0.0661	0.0286	0.0207	0.0571
CCDC Dep. No.	1847347	1847348	1847349	1847350	1847138

B. Selected Spectroscopy for the Ligand and Complexes

1. Spectra for Li(NCMe)₂[B(Me)H(azaindolyl)₂]





Figure 1.3: ^{11}B NMR (top) and $^{11}B\{^{1}H\}$ (bottom) in CD $_{3}CN,$ 128.37 MHz, 298 K



Figure 1.4: $^{13}C\{^{1}H\}$ NMR, CD_3CN, 100.62 MHz, 298 K

2. Spectra of complex $[Ir{\kappa^3-N,N,H-B(Me)H(azaindolyl)_2}(COD)]$ (1)



Figure 2.1: ¹H NMR (C₆D₆), 400.13 MHz, 298 K (* = residual solvent)





Figure 2.3: ¹³C{¹H} NMR (C₆D₆), 100.62 MHz, 298 K (* = residual solvent)



3. Spectra of complex [Rh{k³-N,N,H-B(Me)H(azaindolyl)₂}(COD)] (2)

Figure 3.2: Partial ¹H{¹¹B} NMR, C₆D₆, 400.13 MHz, 298 K



Figure 3.3: ¹¹B NMR (top) and ¹¹B{¹H} (bottom) NMR, C₆D₆, 128.37 MHz, 298 K [Rh{ κ^3 -N,N,H-B(Me)H(azaindolyl)₂}(COD)] (**2**)



Figure 3.4: ¹³C{¹H} NMR (C₆D₆), 100.62 MHz, 298 K (* = residual solvent)

4. Spectra of complex [Rh{κ³-N,N,H-B(Me)H(azaindolyl)₂}(COD^{Me})] (2-Me)



Figure 4.1: ¹H NMR, C₆D₆, 400.13 MHz, 298 K:[Rh{κ³-N,N,H-B(Me)H(azaindolyl)₂}(COD^{Me})] (2-Me)



Figure 4.3: ¹¹B NMR (top) and ¹¹B{¹H} (bottom) NMR, C₆D₆, 128.37 MHz, 298 K [Rh{ κ^3 -N,N,H-B(Me)H(azaindolyl)_2}(COD)] (**2-Me**)



5. Spectra of complex [Rh{κ³-N,N,H-B(Me)H(azaindolyl)₂}(NBD)] (3)



Figure 5.1: ¹H NMR (C₆D₆), 400.13 MHz, 298 K (* = residual solvent)





Figure 5.2: Partial ¹H{¹¹B} NMR (C₆D₆), 400.13 MHz, 298 K (* = residual solvent)





6. Activation studies of complex [Rh{κ³-N,N,H-B(Me)H(azaindolyl)₂}(COD)] (2) to form 2a



Figure 6.1: ^{11}B NMR (top) and $^{11}B\{^{1}H\}$ (bottom) in C₆D₆, 128.37 MHz, 298 K



Figure 6.2: ¹¹B NMR (top) and ¹¹B{¹H} (bottom) NMR, 128.37 MHz, 298 K in DCM-*d*₂









Figure 6.5: ¹¹B NMR (top) and ¹¹B{¹H} (bottom) NMR, C₆D₆, 128.37 MHz, 298 K (activated product **2a** – after crystallisation)



Figure 6.6: ¹H NMR, C₆D₆, 400.13 MHz, 298 K (activated product **2a –** after crystallisation)



Figure 6.7: ¹³C{¹H} NMR (C₆D₆), 100.62 MHz, 298 K (activated product **2a** - after crystallisation)



Figure 6.8: Comparison of infrared of starting complex 2 and activated product 2a