# Neuroglobin is capable of self-oxidation of methionine64 introduced at the heme axial position

Hai-Xiao Liu,<sup>a</sup> Lianzhi Li,<sup>b</sup> Bo He,<sup>a</sup> Shu-Qin Gao,<sup>c</sup> Ge-Bo Wen,<sup>c</sup> and Ying-Wu Lin\*<sup>a,c</sup>

<sup>a)</sup> School of Chemistry and Chemical Engineering, University of South China, Hengyang 421001, China; E-mail:linlinying@hotmail.com

<sup>b)</sup> School of Chemistry and Chemical Engineering, Liaocheng University, Liaocheng 252059,

China.

<sup>c)</sup> Laboratory of Protein Structure and Function, University of South China, Hengyang 421001,

China.

## **Supporting Information**

### Contents

#### **1. Experimental Section**

1.1 Protein preparation	p. S2
1.2 UV-Vis and stopped-flow studies	p. S2
1.3 EPR spectroscopy	p. S3
1.4 ESI-MS mass spectrometry	p. S3
2. Fig. S1. ESI-MS spectra of H64M Ngb.	p. S4
3. Fig. S2. ESI-MS spectra of H64M/C120S Ngb.	p. S5
4. Fig. S3. ESI-MS spectra of trypsin-digested H64M/C120S Ngb.	p. S6
5. Fig. S4. Amino acid sequence and ESI-MS spectrum of trypsin-digeste	d H64M
Ngb with DTT.	p. S7
6. Fig. S5. ESI-MS spectra of H64M/C120S Ngb with DTT for 5 h.	p. S8
<b>7. Fig. S6</b> ESI-MS spectra of H64M/C120S Ngb after incubation with DTT.	p. S9
<b>8. Fig. S7</b> UV-Vis spectral changes of H64M/C120S Ngb in presence of DTT. p. S10	
<b>9. Fig. S8</b> EPR study of H64M/C120S Ngb in reaction with $H_2O_2$ .	p. S10
10. Fig. S9 ESI-MS spectra of H64M/C120S Ngb and after addition of $H_2O_2$ ,	
$K_3[Fe(CN)_6]$ , or <i>m</i> -CPBA for 10 min.	p. S11
11. Fig. S10 Original multiply-charged series in ESI-MS spectra of H64M	M/C120S
Ngb addition of $H_2O_2$ or <i>m</i> -CPBA for 10 min.	p. S12
12. Fig. S11 Stopped-flow UV-Vis spectra of H64M/C120S Ngb in reaction with	
$H_2O_2$ for 0.5 sec.	p. S13

#### **1. Experimental Section**

#### **1.1 Protein preparation**

The pET3a plasmid DNA containing the gene of wild-type (WT) human Ngb was a gift from Prof. T. Burmester, Gutenberg University of Mainz, Germany. The H64M Ngb and H64M/C120S Ngb genes were constructed using the QuikChange Site Directed Mutagenesis Kit (Stratagene) with two primers: H64M, forward, 5'-GAG TTC CTG GAC ATG ATC AGG AAG GTG-3', and reverse, 5'-CAC CTT CCT GAT CAT GTC CAG GAA CTC-3'; C120S, 5'- ATG CTG GAG AAG TCT CTG GGC CCT GCC-3', and reverse, 5'-GGC AGG GCC CAG AGA CTT CTC CAG CAT-3'. The mutations were confirmed by DNA sequencing assay. Both mutants were expressed in BL21(DE3) and purified using a similar procedure for WT Ngb.<sup>1</sup> After eluted from DEAE column with 200 mM NaCl, the protein was concentrated by Amicon filtration (PM10) and oxidized with K<sub>3</sub>[Fe(CN)<sub>6</sub>]. It was then passed though a Sephacryl S-100 column, followed by further purification using a Mono Q column and eluted with a linear gradient of 1 M NaCl.

#### 1.2 UV-Vis and stopped-flow studies

UV-Vis spectrum of ferric Ngbs were recorded in 100 mM potassium phosphate buffer (pH 7.0) on a Agilent 8453 diode array spectrometer. The ferrous protein samples were obtained by addition of a small amount of sodium dithionite. The protein concentration was determined with an extinction coefficient of  $\varepsilon_{405} = 160 \pm 5$ mM<sup>-1</sup> cm<sup>-1</sup> for both H64M Ngb and H64M/C120S Ngb mutants, as calculated using the standard hemochromagen method.<sup>2</sup> In another study, after addition of DTT to a final concentration of 0.5 mM, the spectral changes of H64M/C120S Ngb (10  $\mu$ M) were collected at different time courses at 25 °C.

Kinetic UV-Vis studies of H64M/C120S Ngb in reaction with dithiothreitol (DTT), *m*-chloroperbenzoic acid (*m*-CPBA) or H<sub>2</sub>O<sub>2</sub>, were performed with a SF-61DX2 Hi-Tech KinetAsyst<sup>TM</sup> dual mixing stopped-flow spectrophotometer. Typically, one syringe contains 20  $\mu$ M protein (100 mM potassium phosphate buffer, pH 7.0), and the second syringe contains 1 mM DTT, 0.2 mM *m*-CPBA or H<sub>2</sub>O<sub>2</sub>. The reaction was stated with mixing of equal volume of solutions from the both syringes. 500 (or 50) time-dependent spectra were collected over 30 s (or 0.5 s) from 350 to 700 nm at 25 °C.

#### 1.3 EPR studies

Electron Paramagnetic Resonance (EPR) spectra of WT and H64M Ngb (0.3 mM) in 100 mM potassium phosphate buffer (pH 7.0), containing 10% glycerol, were collected at the high magnetic field laboratory of Chinese Academy of Science (Hefei, China). The samples were analyzed by X-band EPR on a Bruker EMX plus 10/12 spectrometer. A standard Bruker cavity (ER4119hs TE011) was used in conjunction with an Oxford Instrument EPR910 liquid helium continuous-flow cryostat for low-temperature analysis. The spectra were measured at 10 K, with frequency of 9.43 GHz, center field 2200 G and sweep width 3600 G, microwave power 2 mW and modulation amplitude 2.0 G. The EPR spectra H64M/C120S Ngb (0.5 mM) and in reaction with DTT (50 mM) for 15, 30, 45 min, 1, 2 and 12 h, or in reaction with  $H_2O_2$  (0.5 mM) for 30 s, 1, 5 and 10 min, were recorded on a Bruker A300 spectrometer (X-band), equipped with Bruker ER4141VTM liquid nitrogen system, as available in the authors' lab. The spectra were measured at 100 K, with frequency of 9.43 GHz, center field 2200 G and sweep width 3600 G, microwave power 2.0 mW and modulation amplitude 3.0 G.

#### **1.4 Mass spectrometry**

Protein mass spectrum measurement was carried out on G2-XS QToF mass spectrometer (Waters). The desalted protein solution (~20  $\mu$ M) was mixed with 1% formic acid and transferred into the mass spectrometer chamber for measurement under positive mode. The multiple m/z peaks were transformed to the protein molecular weight by using software MaxEnt1. For trypsin digestion studies, 50  $\mu$ L H64M Ngb and H64M/C120S Ngb (~0.1 mM) in ammonium carbonate buffer (50 mM, pH 7.8) was mixed with the same volume of trypsin solution (1 mg/mL) in the same buffer, and the mixture was heated at 37 °C for 12 h, followed by addition of DTT (10 mM) and treatment at 37 °C for 3 h, before determining the MS of the digested protein fragments. The mass spectra of the protein after reaction with 10 eq. *m*-CPBA, H<sub>2</sub>O<sub>2</sub> or K<sub>3</sub>[Fe(CN)<sub>6</sub>] for 10 min were determined used the same procedure.

#### **References:**

S. Dewilde, L. Kiger, T. Burmester, et al., J. Biol. Chem., 2001, 276, 38949-38955.
M. Morrison, S. Horie, Anal. Biochem., 1965, 12, 77-82.



**Fig. S1** ESI-MS spectra of H64M Ngb as purified. (A) Original multiply-charged series. (B) The MaxEnt survey spectrum showing the major components. Note that the calculated mass of H64M Ngb with an intramolecular disulfide bond of Cys46-Cys55 is 16925.5 Da.



**Fig. S2** ESI-MS spectra of H64M/C120S Ngb as purified. (A) Original multiply-charged series. (B) The MaxEnt survey spectrum showing the major components. Note that the calculated mass of H64M/C120S Ngb with an intramolecular disulfide bond of Cys46-Cys55 is 16909.4 Da.



**Fig. S3** ESI-MS spectra of trypsin-digested (12 h) H64M/C120S Ngb without (A) and with (B) treated by DTT at 37  $^{\circ}$ C for 3 h. The mass spectra in the 1500–2100 m/z region are shown as insets.



**Fig. S4** (A) Amino acid sequence of H64M Ngb and the expected peptide fragments with molecular weight larger than 400 Da obtained by trypsin digestion. (B) ESI-MS spectrum of trypsin-digested (12 h) H64M Ngb treated by DTT (10 mM) at 37 °C for 3 h. The mass spectrum in the 1500–2100 m/z region is shown as an inset.



**Fig. S5** ESI-MS spectra of H64M/C120S Ngb after digestion by trypsin with DTT (10 mM) at 37  $^{\circ}$ C for 5 h before fully digested. (A) Original multiply-charged series. (B) The MaxEnt survey spectrum showing the major components. Note that the calculated mass of H64M/C120S Ngb with an intramolecular disulfide bond of Cys46-Cys55 is 16909.4 Da, which was almost converted to SO-Met (+16 Da) and SO<sub>2</sub>-Met (+32 Da) species (inset).



**Fig. S6** ESI-MS spectra of H64M/C120S Ngb (10  $\mu$ M) after incubation with 1 mM DTT in air for 10 min. (A) Original multiply-charged series and (B) the MaxEnt survey spectrum showing the major components.



Fig. S7 UV-Vis spectral changes of H64M/C120S Ngb (10  $\mu$ M) in presence of DTT (0.5 mM) and in 100 mM potassium phosphate buffer (pH 7.0).



Fig. S8 EPR spectra of ferric H64M/C120S Ngb (0.5 mM) and in reaction with  $H_2O_2$  (0.5 mM) in air for 30 sec, 1, 5 and 10 min.



**Fig. S9** ESI-MS spectra of H64M/C120S Ngb (A) and after addition of 10 eq.  $H_2O_2$  (B),  $K_3[Fe(CN)_6]$  (C), and *m*-CPBA (D), respectively, for 10 min at room temperature.



**Fig. S10** Original multiply-charged series in ESI-MS spectra of H64M/C120S Ngb after addition of *m*-CPBA (A) or  $H_2O_2$  (B) or for 10 min at room temperature.



Fig. S11 Stopped-flow UV-Vis spectra of H64M/C120S Ngb (10  $\mu$ M) in reaction with H<sub>2</sub>O<sub>2</sub> (0.1 mM) for 0.5 sec at 25 °C.