

Syntheses and characterization of hepta-coordinated Group 4 amidinate complexes

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Supplementary Information

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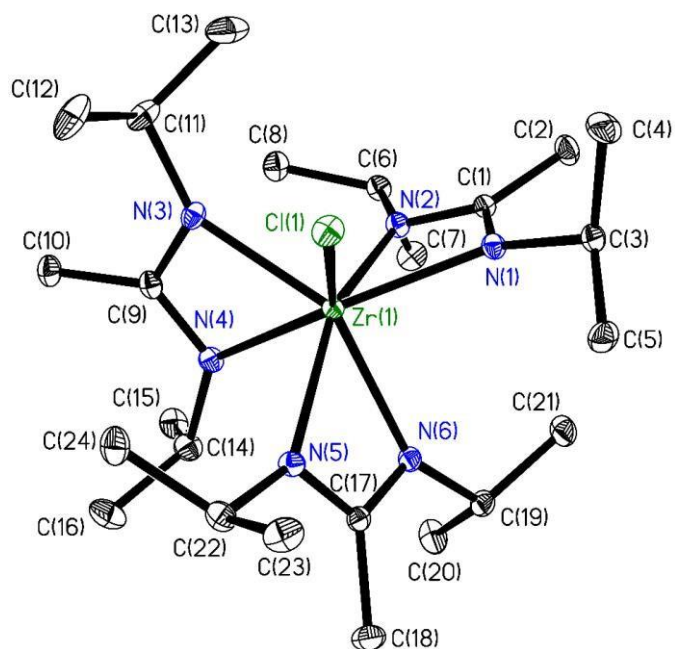


Fig. S1. ORTEP of Zr-Cl (1) at 100(2) K. Thermal ellipsoids are 30% probability level. H atoms were removed for clarity.

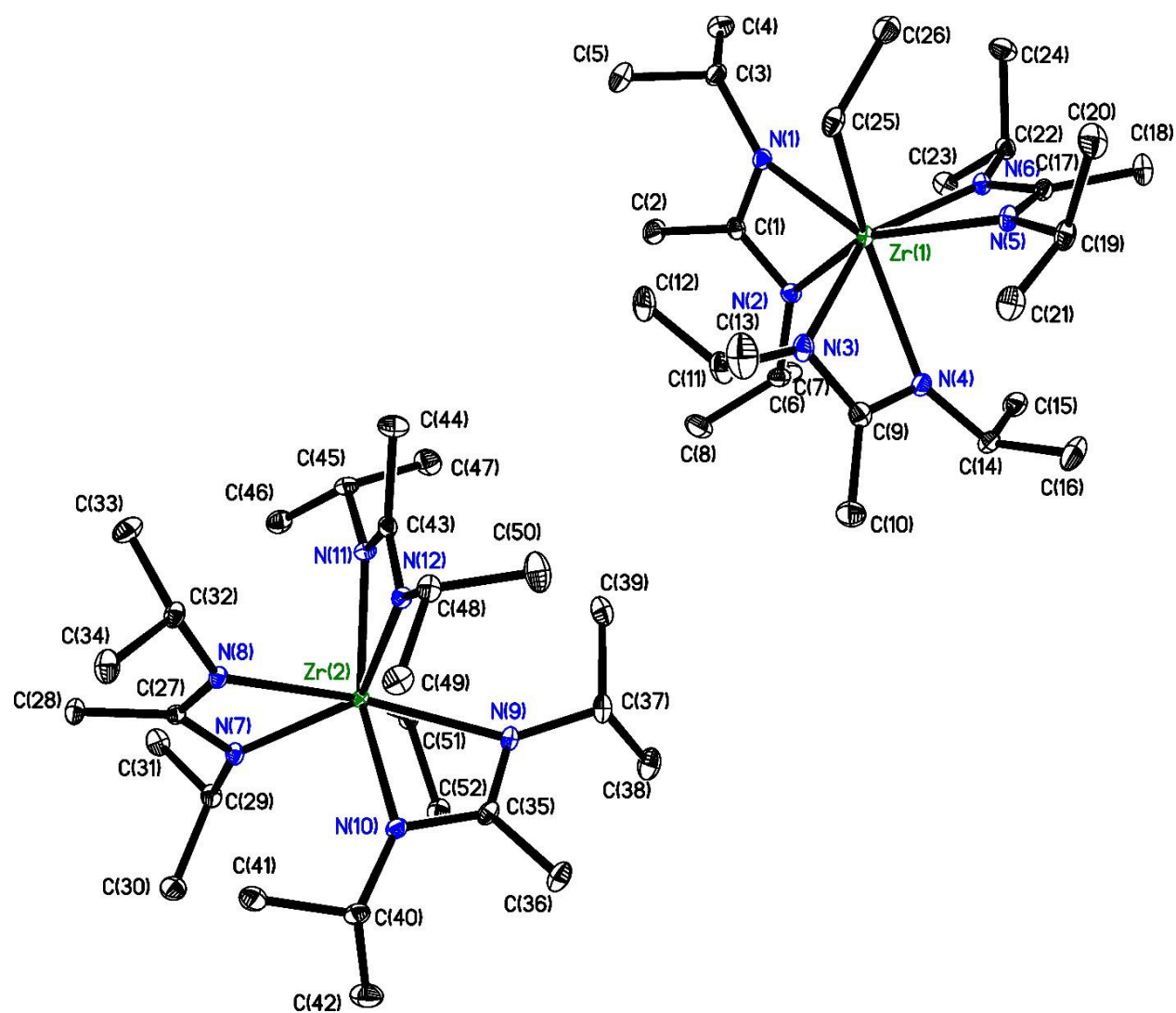


Fig. S2. ORTEP of Zr-Et (**5**) at 100(2) K showing two slightly different molecules of Zr-Et (**5**). Thermal ellipsoids are 30% probability level. H atoms were removed for clarity.

Table S1. Crystal data and structure refinement for Zr-Cl (**1**) and Hf-Cl (**2**)

Complex	Zr-Cl (1)	Hf-Cl (2)
Empirical formula	C ₂₄ H ₅₁ N ₆ ClZr	C ₂₄ H ₅₁ N ₆ ClHf
Formula weight	550.37	637.64
Temperature	100(2) K	100(2) K
Wavelength	0.71073 Å	0.71073 Å
Crystal System	Monoclinic	Monoclinic
Space Group	<i>P2₁/n</i>	<i>P2₁/n</i>
Unit cell dimensions	<i>a</i> = 10.447(2) Å	<i>a</i> = 10.4324(15) Å
	$\alpha = 90^\circ$	$\alpha = 90^\circ$
	<i>b</i> = 16.453(3) Å	<i>b</i> = 16.447(2) Å
	$\beta = 93.334^\circ$	$\beta = 93.053(2)^\circ$
	<i>c</i> = 17.100(3) Å	<i>c</i> = 17.091(2) Å
	$\gamma = 90^\circ$	$\gamma = 90^\circ$
Volume	2934.5(8) Å ³	2928.3(7) Å ³
<i>Z</i>	4	4
Density (calculated)	1.246 g/cm ³	1.446 g/cm ³
Absorption coefficient	0.487 mm ⁻¹	3.675 mm ⁻¹
<i>F</i> (000)	1176	1304.0
Crystal size	0.358 x 0.219 x 0.203 mm ³	0.271 × 0.251 × 0.233 mm ³
Theta range for data collection	2.23 to 27.91°	2.24 to 27.94°
Index ranges	-13 ≤ <i>h</i> ≤ 13, 0 ≤ <i>k</i> ≤ 21, 0 ≤ <i>l</i> ≤ 22	-13 ≤ <i>h</i> ≤ 13, 0 ≤ <i>k</i> ≤ 21, 0 ≤ <i>l</i> ≤ 22
Reflections collected	6921	6975
Independent reflections	6921 [<i>R</i> _{int} = 0.000]	6975 [<i>R</i> _{int} = 0.000]
Completeness	98.4%	93.0%
Max. and min. transmission	0.746 and 0.658	0.745 and 0.491

Refinement method	Full-matrix least-squares on F^2	Full-matrix least-squares on F^2
Absorption correction	Semi-empirical from equivalents	Semi-empirical from equivalents
Data / restraints / parameters	6921 / 0 / 305	6975 / 0 / 305
Goodness-of-fit on F^2	1.031	1.186
Final R indices [$I > 2\sigma(I)$]	$R1 = 0.0269$, $wR2 =$ 0.0639	$R1 = 0.0445$, $wR2 =$ 0.1177
R indices (all data)	$R1 = 0.0307$, $wR2 =$ 0.0662	$R1 = 0.0525$, $wR2 =$ 0.1213
Largest diff. peak and hole	0.51 and -0.40 eÅ ⁻³	5.33 and -2.24 eÅ ⁻³

^a $R1 = \sum ||F_o| - |F_c|| / \sum |F_o|$; $wR2 = [\sum w(F_o^2 - F_c^2)^2 / \sum w(F_o^2)^2]^{1/2}$;

$w = 1/[\sigma^2(F_o) + (aP)^2 + bP]$; $P = [2Fc^2 + \text{Max}(F_o^2, 0)]/3$

Table S2. Crystal data and structure refinement for Zr-Et (**5**) and Hf-Et (**6**)

Complex	Zr-Et (5)	Hf-Et (6)
Empirical formula	C ₂₆ H ₅₆ N ₆ Zr	C ₂₆ H ₅₆ N ₆ Hf
Formula weight	549.98	631.25
Temperature	100(2) K	100(2) K
Wavelength	0.71073 Å	0.71073 Å
Crystal System	Monoclinic	Monoclinic
Space Group	<i>P2₁/c</i>	<i>P2₁/c</i>
Unit cell dimensions	<i>a</i> = 21.699(7) Å	<i>a</i> = 21.658(7) Å
	<i>α</i> = 90°	<i>α</i> = 90°
	<i>b</i> = 15.619(5) Å	<i>b</i> = 15.566(5) Å
	<i>β</i> = 103.401(4)°	<i>β</i> = 103.681(4)°
	<i>c</i> = 18.597(6) Å	<i>c</i> = 18.594(6) Å
	<i>γ</i> = 90°	<i>γ</i> = 90°
Volume	6131(3) Å ³	6091(3) Å ³
<i>Z</i>	8	8
Density (calculated)	1.179 g/cm ³	1.377 g/cm ³
Absorption coefficient	0.381 mm ⁻¹	3.448 mm ⁻¹
<i>F</i> (000)	2352.0	2608.0
Crystal size	0.552 × 0.382 × 0.152 mm ³	0.404 × 0.381 × 0.32 mm ³
Theta range for data collection	1.62 to 27.74°	1.67 to 27.83°
Index ranges	-28 ≤ <i>h</i> ≤ 28, -20 ≤ <i>k</i> ≤ 20, -24 ≤ <i>l</i> ≤ 24	-27 ≤ <i>h</i> ≤ 27, -20 ≤ <i>k</i> ≤ 20, -24 ≤ <i>l</i> ≤ 24
Reflections collected	69813	69357
Independent reflections	14282 [<i>R</i> _{int} = 0.0330]	14157 [<i>R</i> _{int} = 0.0333]
Completeness	99.0%	97.8%
Max. and min. transmission	0.746 and 0.658	0.746 and 0.485
Refinement method	Full-matrix least-squares on <i>F</i> ²	Full-matrix least-squares on <i>F</i> ²

Absorption correction	Semi-empirical from equivalents	Semi-empirical from equivalents
Data / restraints / parameters	14282 / 0 / 627	14162 / 0 / 627
Goodness-of-fit on F^2	1.046	1.040
Final R indices [$I > 2\sigma(I)$]	$R1 = 0.0273$, $wR2 =$ 0.0643	$R1 = 0.0199$, $wR2 =$ 0.0467
R indices (all data)	$R1 = 0.0332$, $wR2 =$ 0.0679	$R1 = 0.0223$, $wR2 =$ 0.0477
Largest diff. peak and hole	0.51 and -0.44 eÅ ⁻³	0.83 and -1.08 eÅ ⁻³

^a $R1 = \sum ||F_o| - |F_c|| / \sum |F_o|$; $wR2 = [\sum w(F_o^2 - F_c^2)^2 / \sum w(F_o^2)^2]^{1/2}$;

$w = 1/[\sigma^2(F_o) + (aP)^2 + bP]$; $P = [2Fc^2 + \text{Max}(F_o^2, 0)]/3$

Table S3. Selected distances and angles for Zr-Cl (1)

Distances (Å)			
Zr(1)-Cl(1)	2.5125(5)	Zr(1)-N(4)	2.2814(13)
Zr(1)-N(1)	2.2141(14)	Zr(1)-N(5)	2.2172(13)
Zr(1)-N(2)	2.3154(14)	Zr(1)-N(6)	2.3175(13)
Zr(1)-N(3)	2.2015(14)		
Angles (°)			
N(1)-Zr(1)-Cl(1)	82.71(4)	N(3)-Zr(1)-N(4)	58.96(5)
N(2)-Zr(1)-Cl(1)	123.82(4)	N(3)-C(9)-N(4)	111.82(14)
N(1)-Zr(1)-N(2)	58.64(5)	N(5)-Zr(1)-Cl(1)	84.72(3)
N(1)-C(1)-N(2)	112.78(14)	N(6)-Zr(1)-Cl(1)	125.68(4)
N(3)-Zr(1)-Cl(1)	84.96(4)	N(5)-Zr(1)-N(6)	58.57(5)
N(4)-Zr(1)-Cl(1)	132.45(4)	N(5)-C(17)-N(6)	112.79(14)

Table S4. Selected distances and angles for Hf-Cl (2)

Distances (Å)			
Hf(1)-Cl(1)	2.5014(16)	Hf(1)-N(4)	2.306(6)
Hf(1)-N(1)	2.183(6)	Hf(1)-N(5)	2.202(6)
Hf(1)-N(2)	2.259(5)	Hf(1)-N(6)	2.293(6)
Hf(1)-N(3)	2.199(6)		
Angles (°)			
N(1)-Hf(1)-Cl(1)	84.30(16)	N(3)-Hf(1)-N(4)	58.9(2)
N(2)-Hf(1)-Cl(1)	132.08(16)	N(3)-C(9)-N(4)	112.6(6)
N(1)-Hf(1)-N(2)	59.2(2)	N(5)-Hf(1)-Cl(1)	82.36(16)
N(1)-C(1)-N(2)	110.3(6)	N(6)-Hf(1)-Cl(1)	123.58(15)
N(3)-Hf(1)-Cl(1)	84.52(15)	N(5)-Hf(1)-N(6)	59.2(2)
N(4)-Hf(1)-Cl(1)	125.58(17)	N(5)-C(17)-N(6)	112.6(6)

Table S5. Selected distances and angles for Zr-Et (5)

Distances (Å)			
Zr(1)-C(25)	2.3207(17)	Zr(2)-C(51)	2.3154(15)
Zr(1)-N(1)	2.2600(13)	Zr(2)-N(7)	2.2552(13)
Zr(1)-N(2)	2.2856(13)	Zr(2)-N(8)	2.2956(12)
Zr(1)-N(3)	2.2281(15)	Zr(2)-N(9)	2.3221(14)
Zr(1)-N(4)	2.3678(14)	Zr(2)-N(10)	2.2422(13)
Zr(1)-N(5)	2.3278(14)	Zr(2)-N(11)	2.2278(13)
Zr(1)-N(6)	2.2265(13)	Zr(2)-N(12)	2.3468(13)
Angles (°)			
N(1)-Zr(1)-C(25)	79.85(5)	N(7)-Zr(2)-C(51)	81.14(5)
N(2)-Zr(1)-C(25)	134.13(5)	N(8)-Zr(2)-C(51)	134.57(5)
N(1)-Zr(1)-N(2)	57.96(5)	N(7)-Zr(2)-N(8)	57.67(5)
N(1)-C(1)-N(2)	111.55(13)	N(7)-C(27)-N(8)	110.85(12)
N(3)-Zr(1)-C(25)	92.90(5)	N(9)-Zr(2)-C(51)	79.39(5)
N(4)-Zr(1)-C(25)	136.88(5)	N(10)-Zr(2)-C(51)	105.78(5)
N(3)-Zr(1)-N(4)	57.72(4)	N(9)-Zr(2)-N(10)	58.31(5)
N(3)-C(9)-N(4)	112.87(13)	N(9)-C(35)-N(10)	113.24(13)
N(5)-Zr(1)-C(25)	78.31(5)	N(11)-Zr(2)-C(51)	90.77(5)
N(6)-Zr(1)-C(25)	103.99(5)	N(12)-Zr(2)-C(51)	134.53(5)
N(5)-Zr(1)-N(6)	58.31(4)	N(11)-Zr(2)-N(12)	58.07(4)
N(5)-C(17)-N(6)	112.99(12)	N(11)-C(43)-N(12)	113.27(12)

Table S6. Selected distances and angles for Hf-Et (6)

Distances (Å)			
Hf(1)-C(25)	2.295(2)	Hf(2)-C(51)	2.298(2)
Hf(1)-N(1)	2.2379(18)	Hf(2)-N(7)	2.2405(18)
Hf(1)-N(2)	2.2831(17)	Hf(2)-N(8)	2.2795(18)
Hf(1)-N(3)	2.2057(17)	Hf(2)-N(9)	2.2041(19)
Hf(1)-N(4)	2.3305(18)	Hf(2)-N(10)	2.3487(19)
Hf(1)-N(5)	2.3072(18)	Hf(2)-N(11)	2.3105(19)
Hf(1)-N(6)	2.2831(17)	Hf(2)-N(12)	2.2047(17)
Angles (°)			
N(1)-Hf(1)-C(25)	81.50(7)	N(7)-Hf(2)-C(51)	80.42(7)
N(2)-Hf(1)-C(25)	135.50(7)	N(8)-Hf(2)-C(51)	135.01(7)
N(1)-Hf(1)-N(2)	57.93(6)	N(7)-Hf(2)-N(8)	58.12(6)
N(1)-C(1)-N(2)	110.80(17)	N(7)-C(27)-N(8)	111.35(18)
N(3)-Hf(1)-C(25)	91.12(7)	N(9)-Hf(2)-C(51)	93.06(8)
N(4)-Hf(1)-C(25)	134.68(7)	N(10)-Hf(2)-C(51)	136.99(7)
N(3)-Hf(1)-N(4)	58.58(6)	N(9)-Hf(2)-N(10)	58.25(6)
N(3)-C(9)-N(4)	113.36(18)	N(9)-C(35)-N(10)	112.66(19)
N(5)-Hf(1)-C(25)	79.20(7)	N(11)-Hf(2)-C(51)	78.04(7)
N(6)-Hf(1)-C(25)	105.19(7)	N(12)-Hf(2)-C(51)	103.40(7)
N(5)-Hf(1)-N(6)	58.63(6)	N(11)-Hf(2)-N(12)	58.66(6)
N(5)-C(17)-N(6)	113.05(18)	N(11)-C(43)-N(12)	112.76(18)

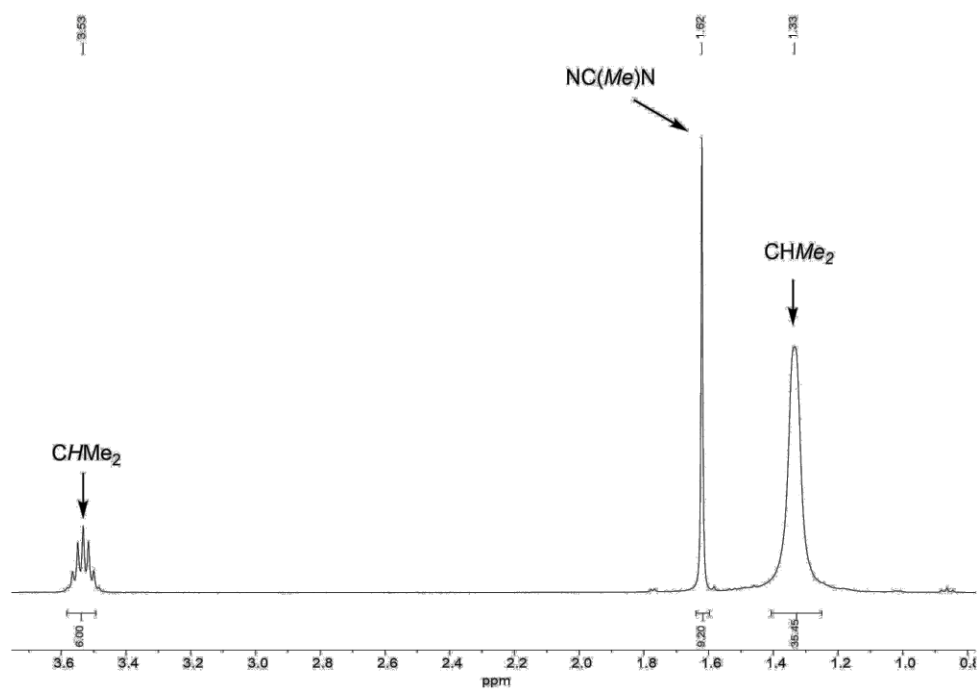


Fig. S3. ^1H NMR spectrum of Zr-Cl (**1**) in benzene- d_6 at 23 °C.

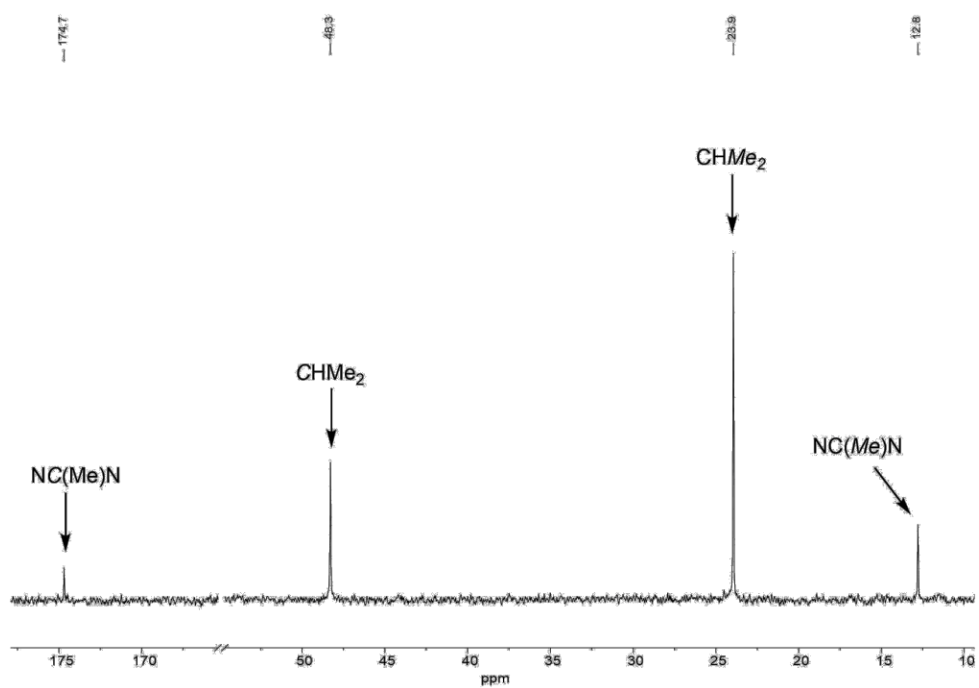


Fig. S4. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of Zr-Cl (**1**) in benzene- d_6 at 23 °C.

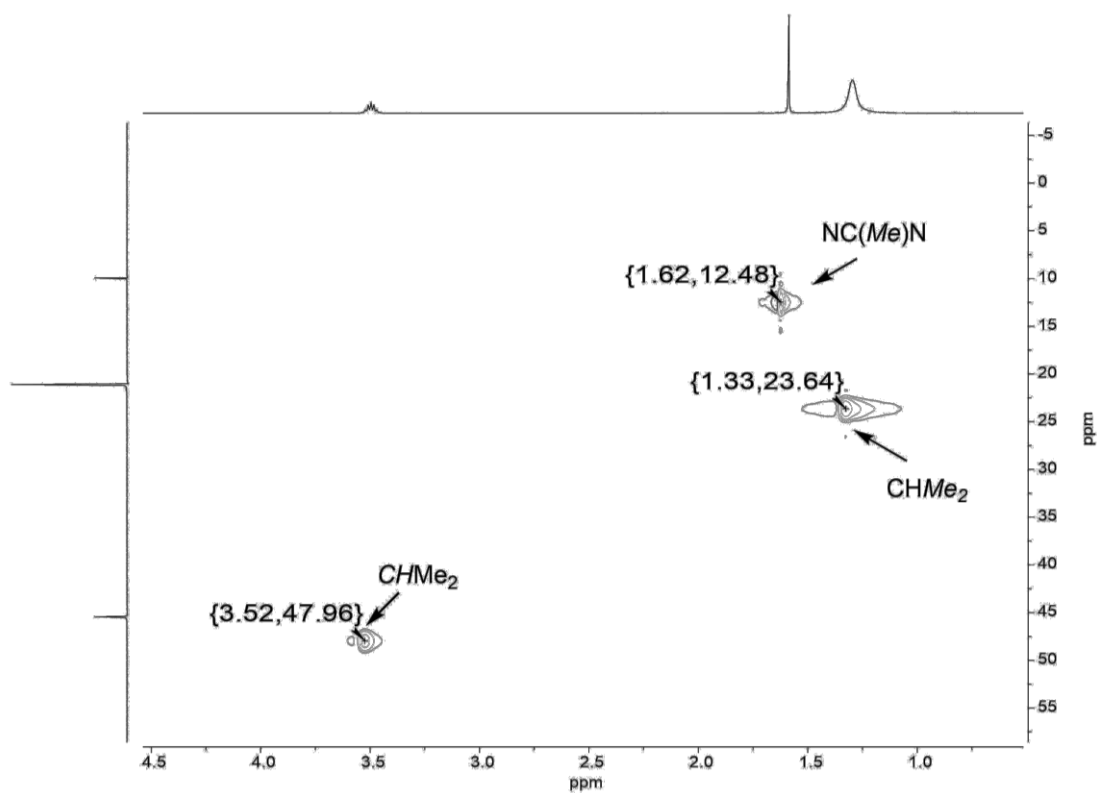


Fig. S5. ^1H - ^{13}C HSQC NMR spectrum of Zr-Cl (**1**) in benzene- d_6 at 23 °C.

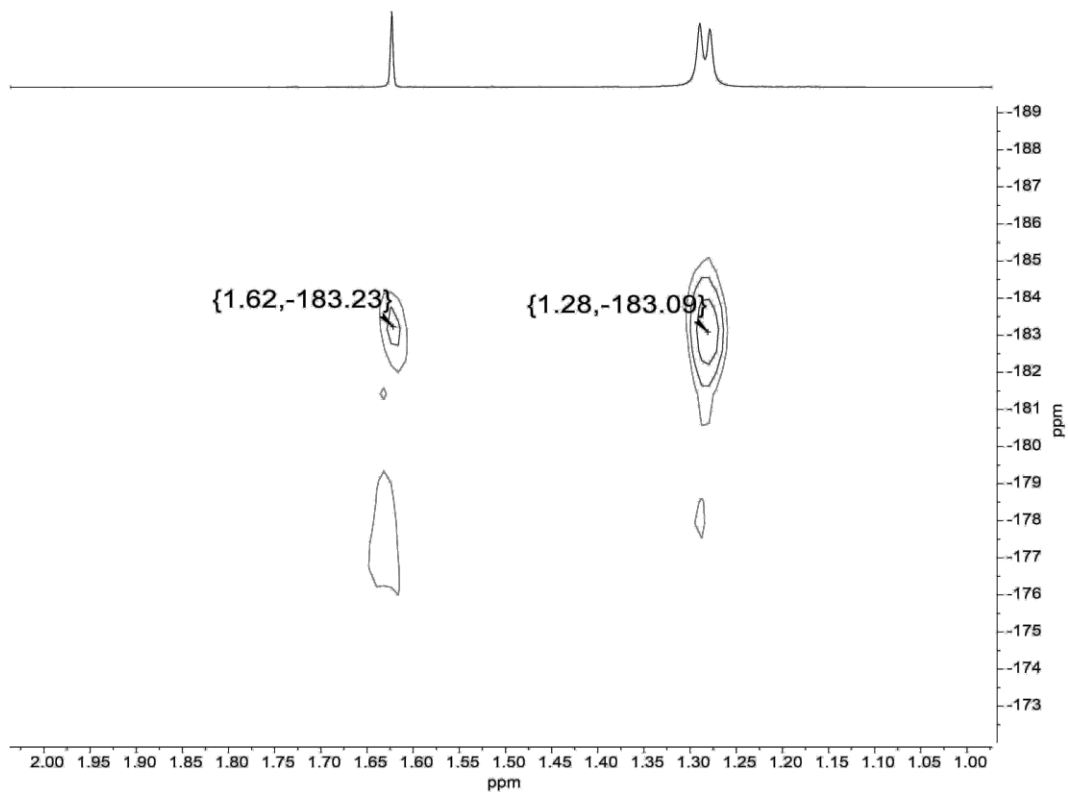


Fig. S6. ^1H - ^{15}N gHMBC NMR spectrum of Zr-Cl (**1**) in benzene- d_6 at 60 °C.

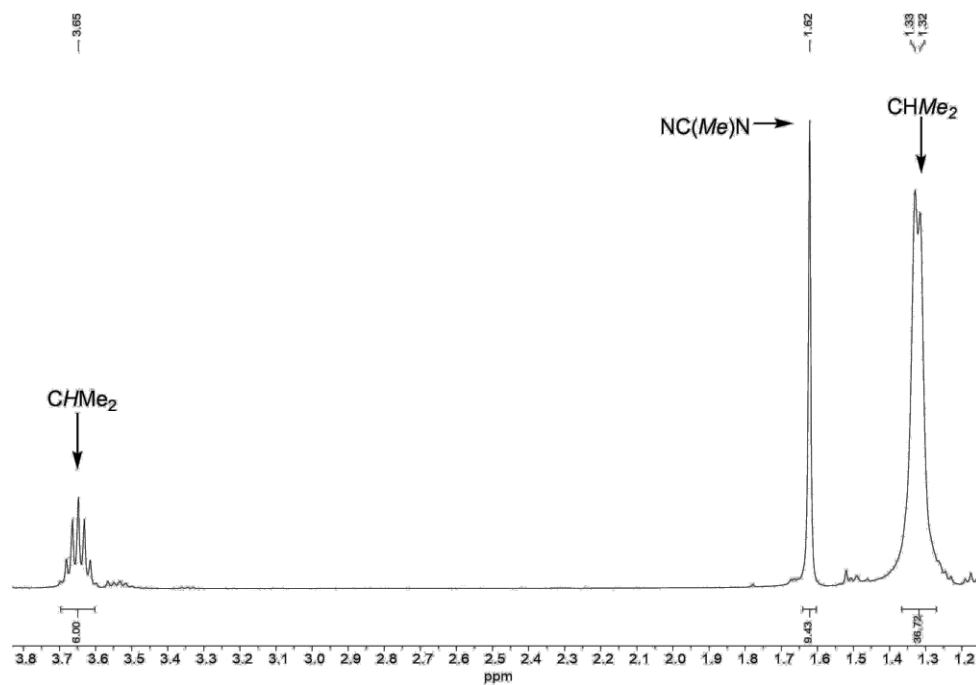


Fig. S7. ^1H NMR spectrum of Hf-Cl (**2**) in benzene- d_6 at 23 °C.

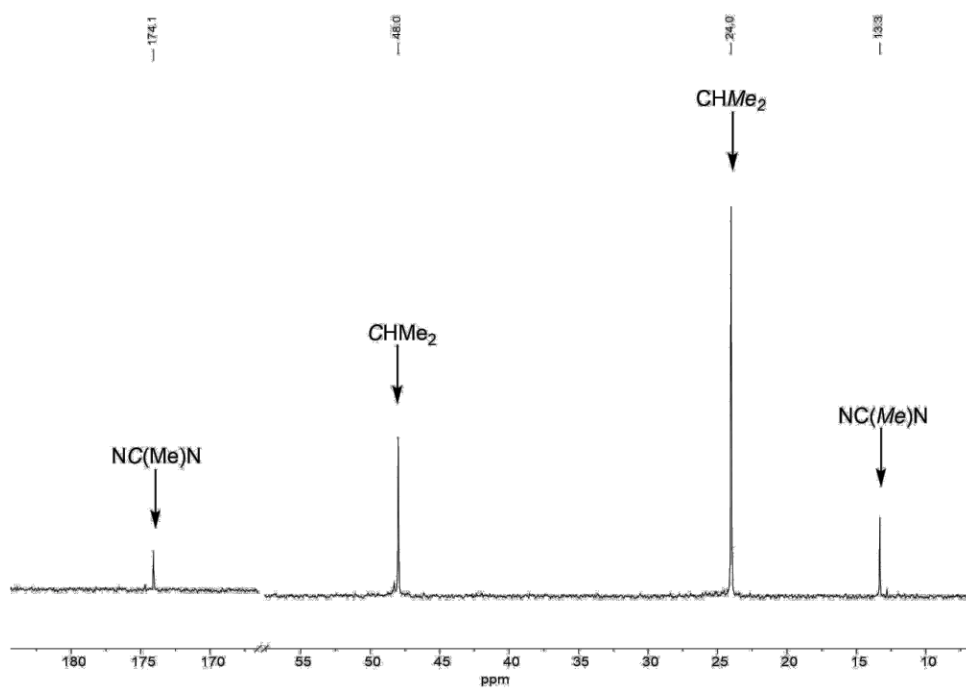


Fig. S8. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of Hf-Cl (**2**) in benzene- d_6 at 23 °C.

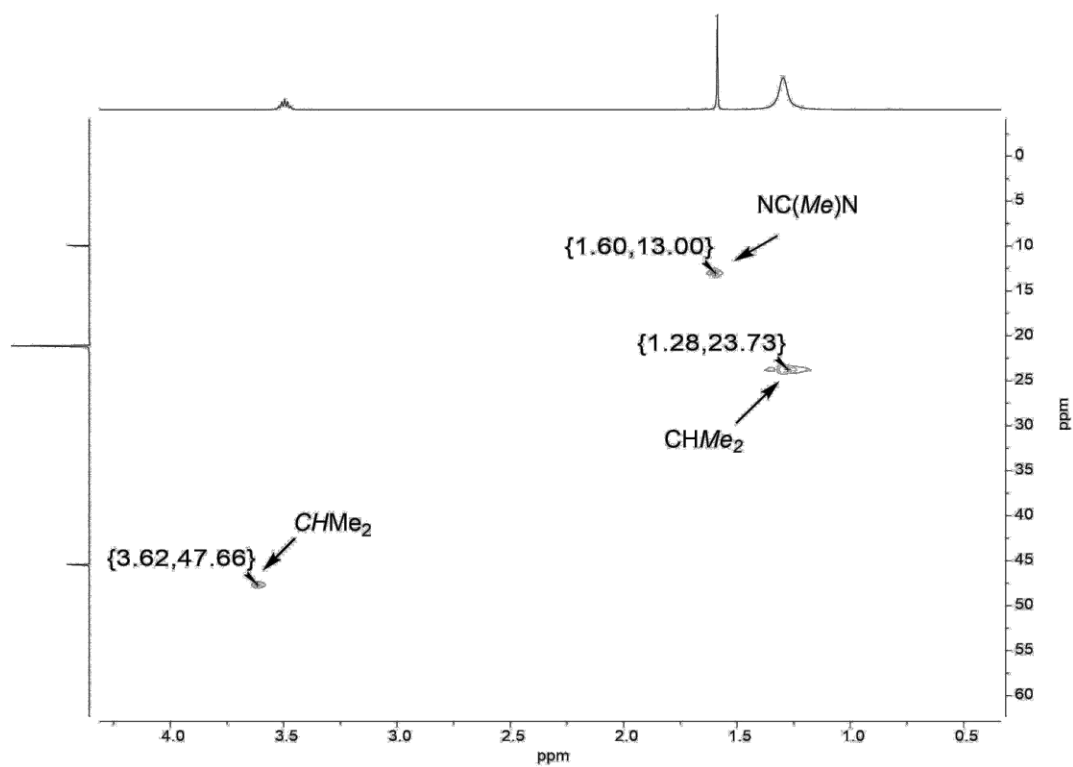


Fig. S9. ^1H - ^{13}C HSQC NMR spectrum of Hf-Cl (**2**) in benzene- d_6 at 23 °C.

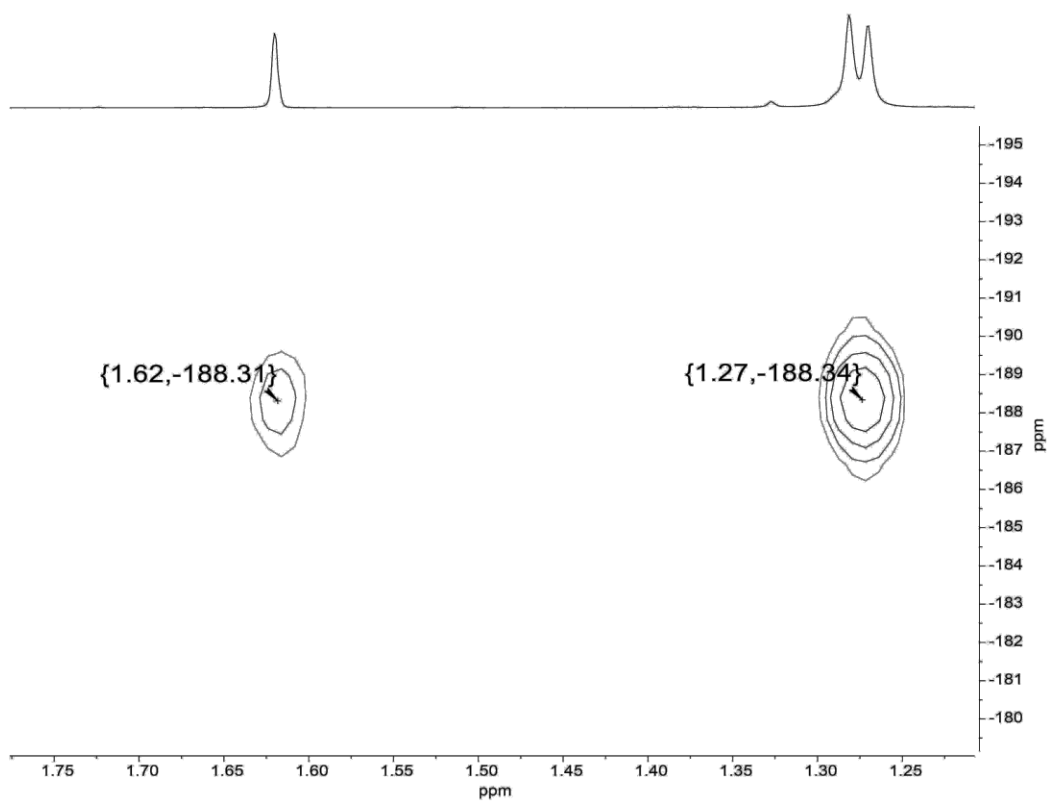


Fig. S10. ^1H - ^{15}N gHMBC NMR spectrum of Hf-Cl (**2**) in benzene- d_6 at 60 °C.

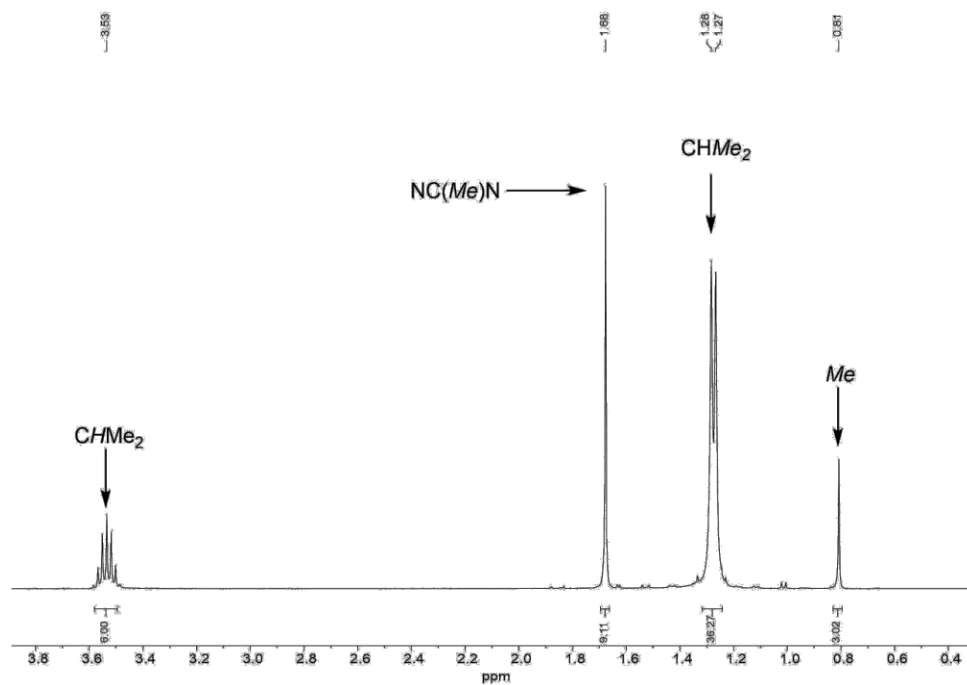


Fig. S11. ^1H NMR spectrum of Zr-Me (**3**) in benzene- d_6 at 23 °C.

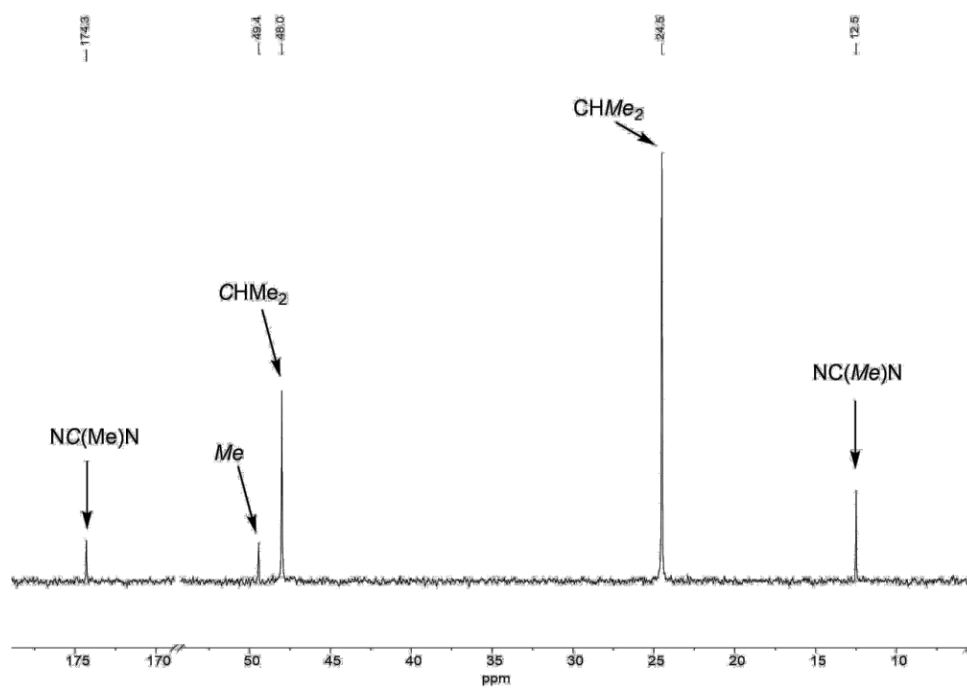


Fig. S12. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of Zr-Me (**3**) in benzene- d_6 at 23 °C.

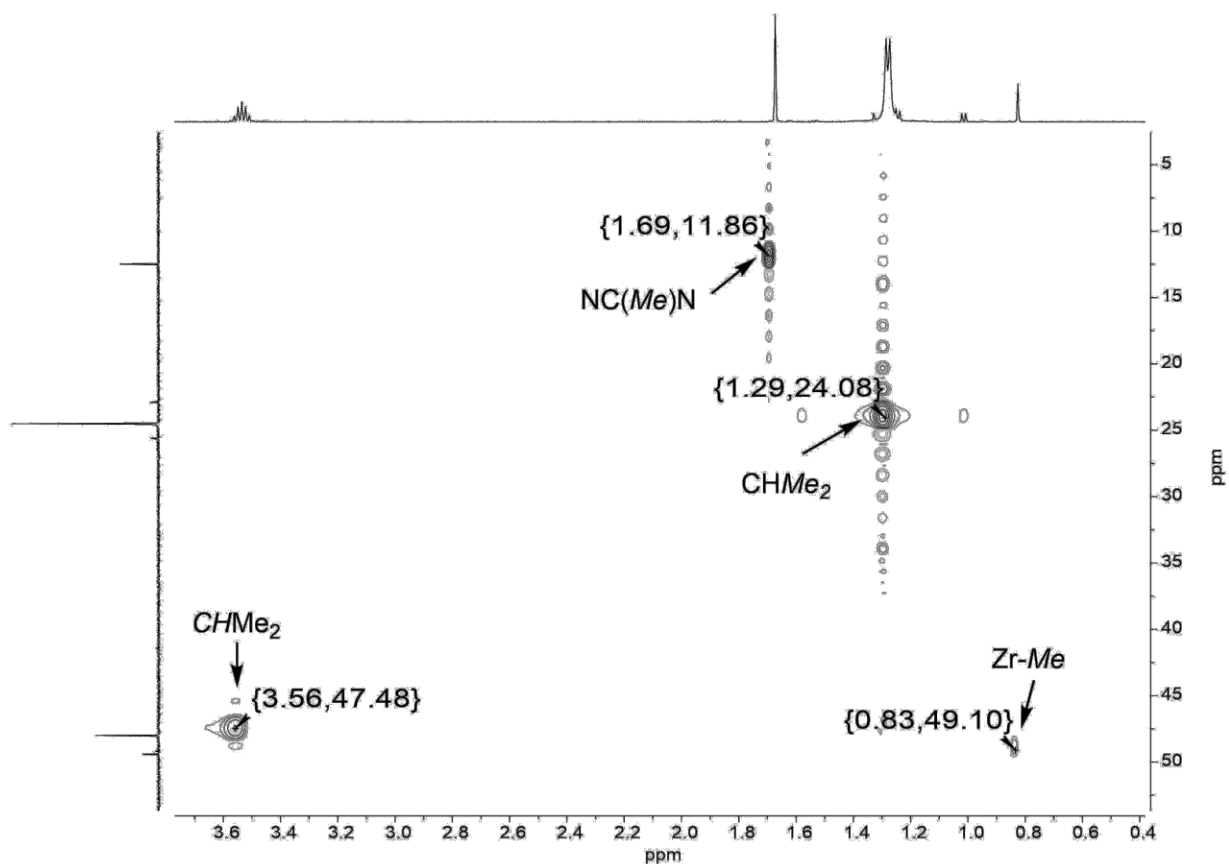


Fig. S13. ^1H - ^{13}C HSQC NMR spectrum of Zr-Me (**3**) in benzene- d_6 at 23 °C.

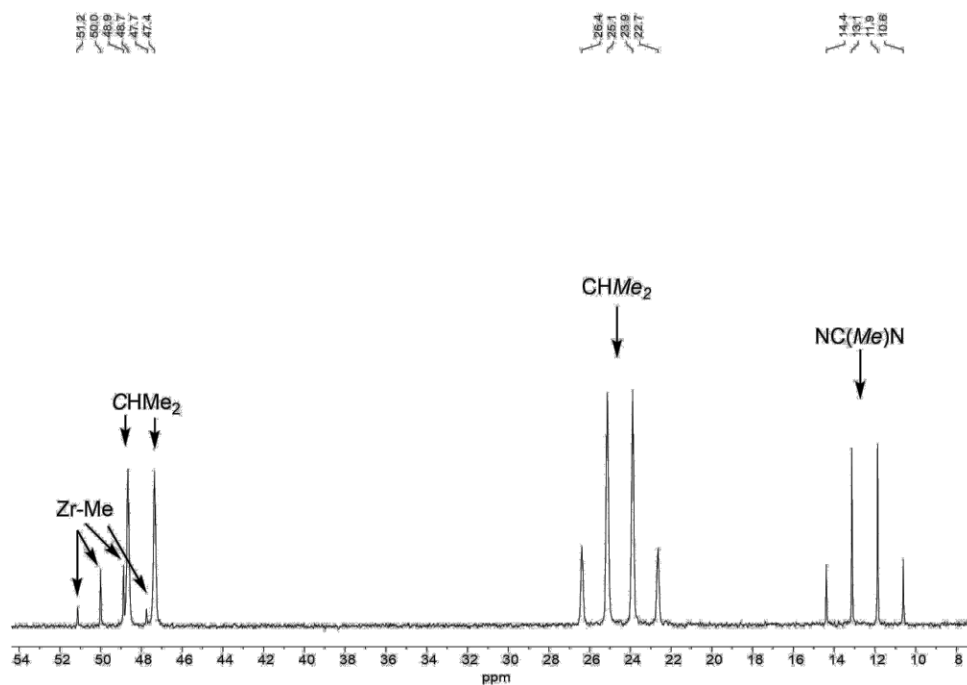


Fig. S14. Gated $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of Zr-Me (**3**) in benzene- d_6 at 23 °C.

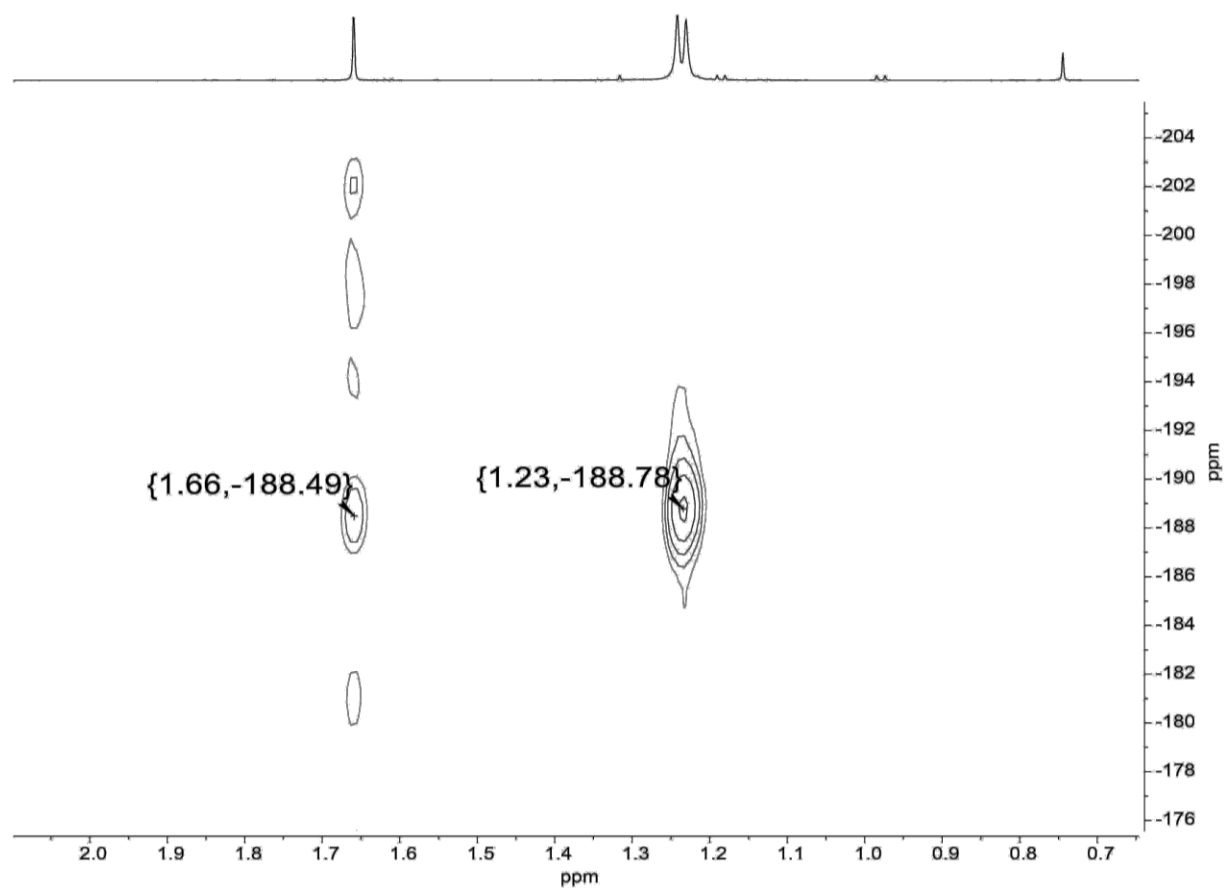


Fig. S15. ^1H - ^{15}N gHMBC NMR spectrum of Zr-Me (**3**) in benzene- d_6 at 45 °C.

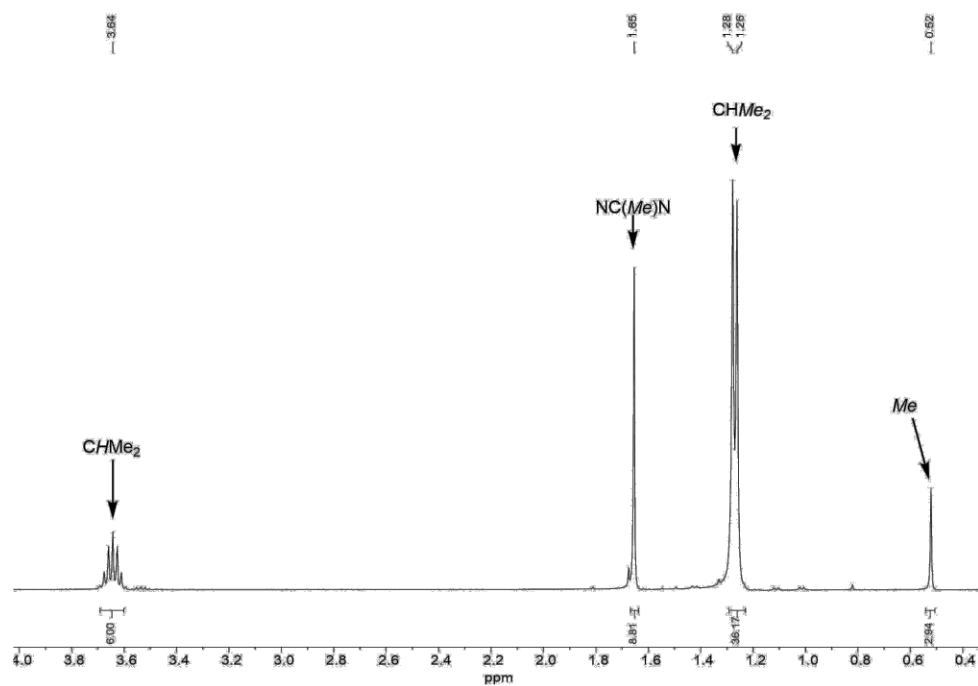


Fig. S16. ^1H NMR spectrum of Hf-Me (**4**) in benzene- d_6 at 23 °C.

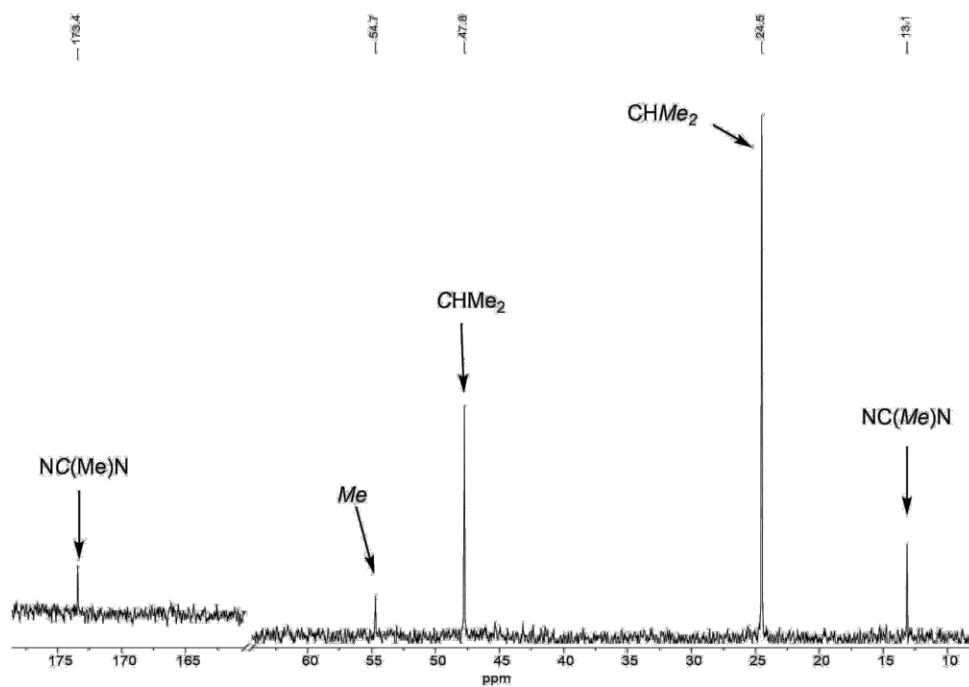


Fig. S17. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of Hf-Me (**4**) in benzene- d_6 at 23 °C.

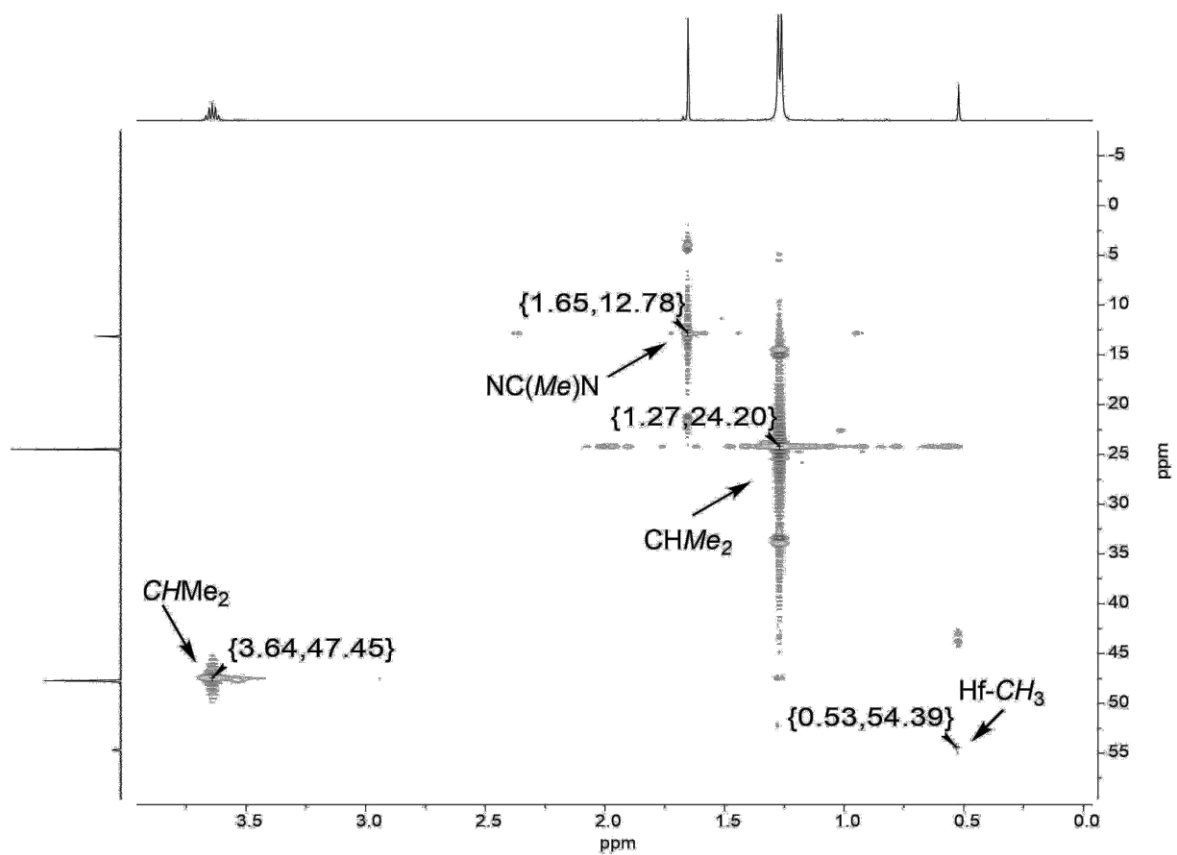


Fig. S18. ^1H - ^{13}C HSQC NMR spectrum of Hf-Me (**4**) in benzene- d_6 at 23 °C.

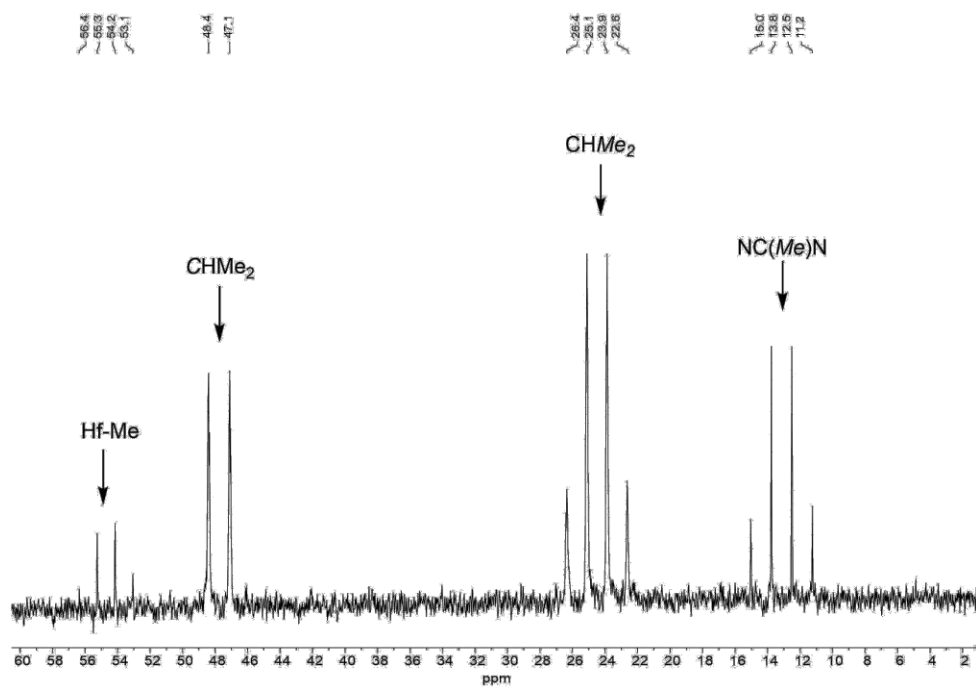


Fig. S19. Gated ^{13}C NMR spectrum of Hf-Me (**4**) in benzene- d_6 at 23 °C.

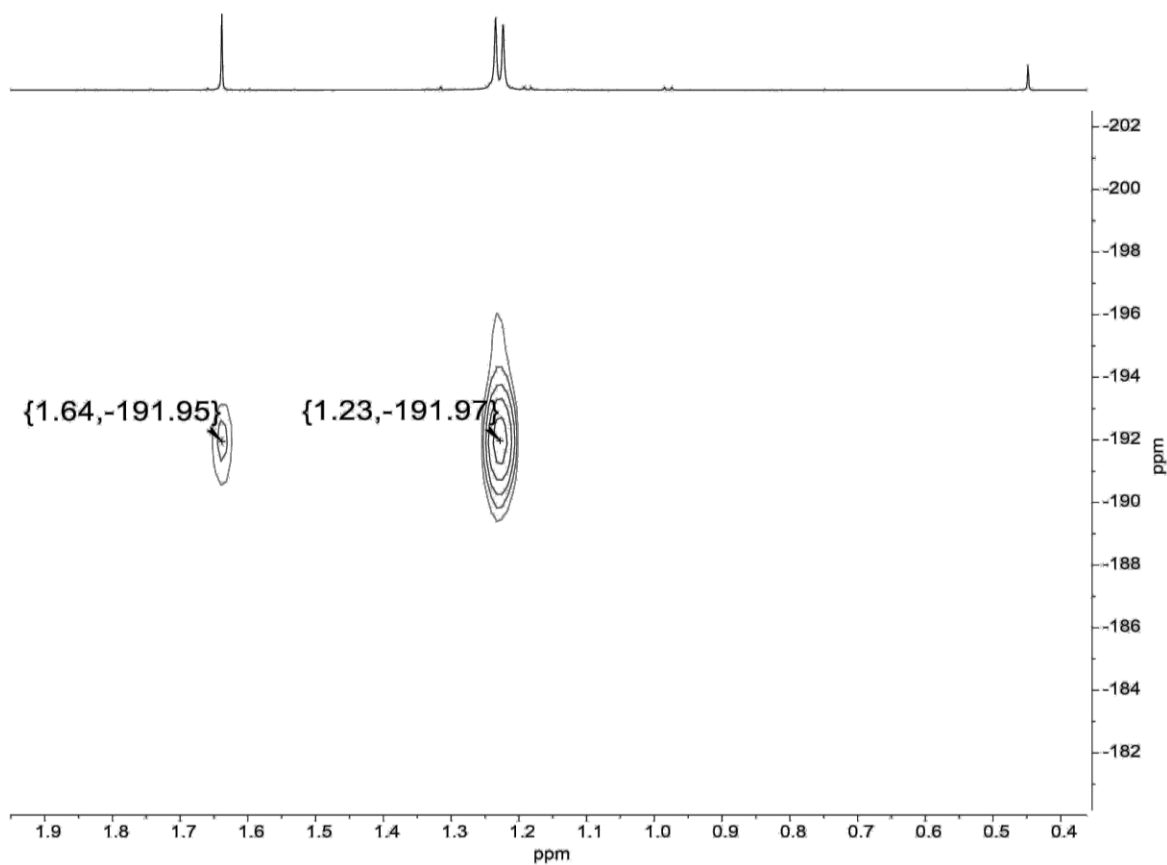


Fig. S20. ^1H - ^{15}N gHMBC NMR spectrum of Hf-Me (**4**) in benzene- d_6 at 45 °C.

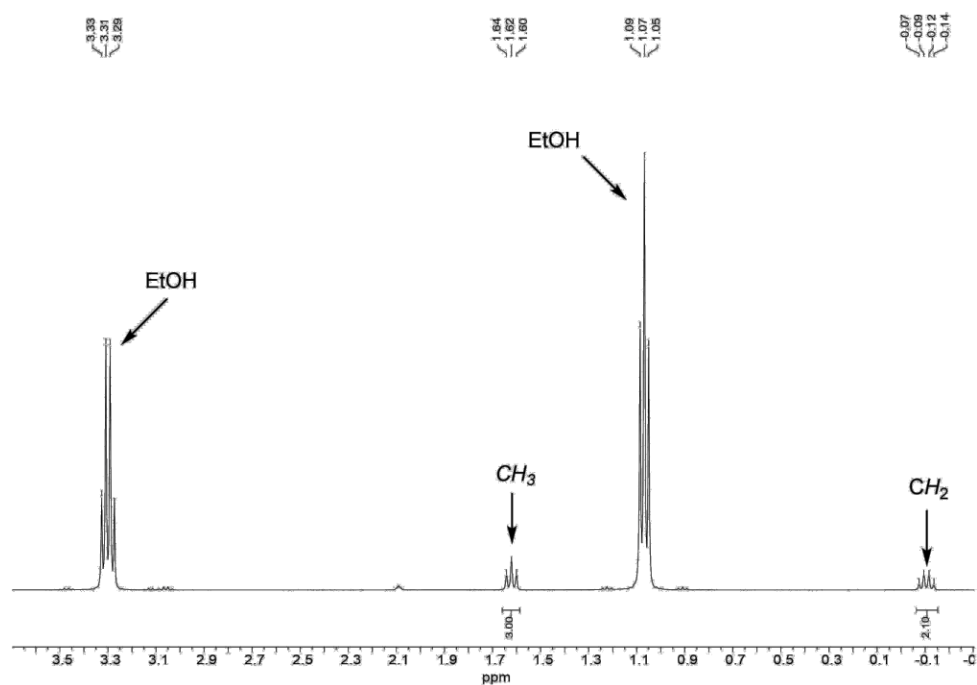


Fig. S21. Partial ^1H NMR spectrum of EtMgCl in a benzene/ Et_2O solution in toluene- d_8 at 23 $^\circ\text{C}$.

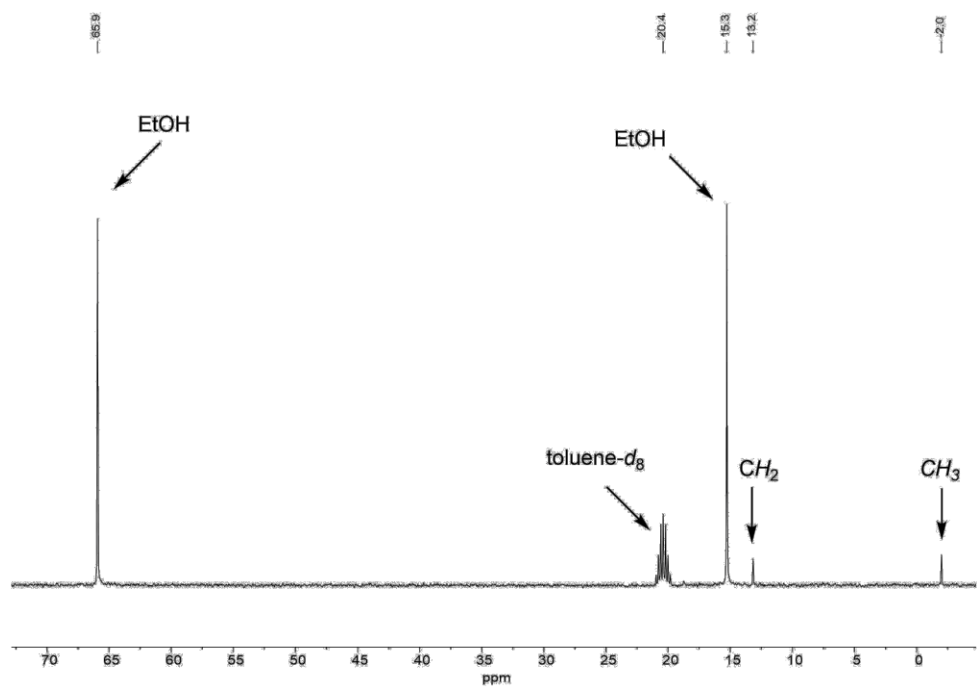


Fig. S22. Partial $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of EtMgCl in a benzene/ Et_2O solution in toluene- d_8 at 23 $^\circ\text{C}$.

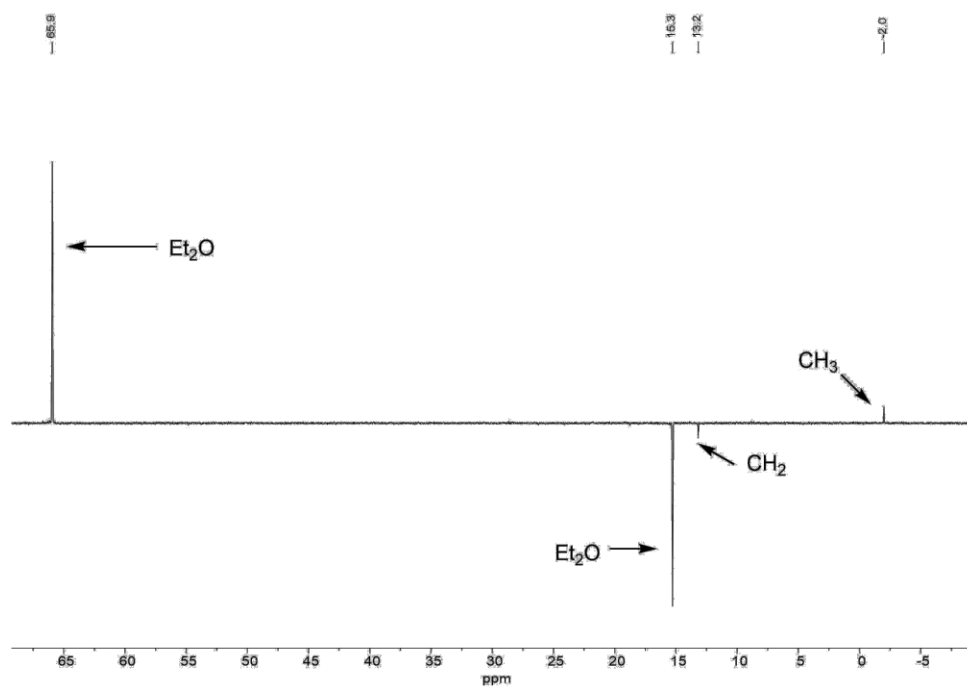


Fig. S23. DEPT-135 NMR spectrum of EtMgCl in a benzene/Et₂O solution in toluene-*d*₈ at 23 °C.

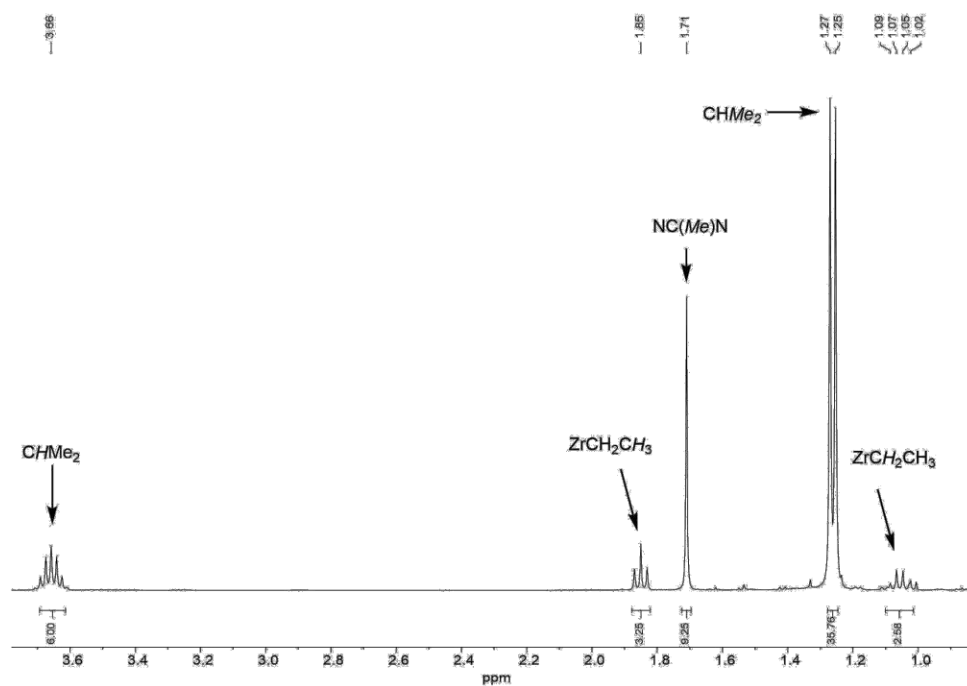


Fig. S24. ¹H NMR spectrum of Zr-Et (**5**) in benzene-*d*₆ at 23 °C.

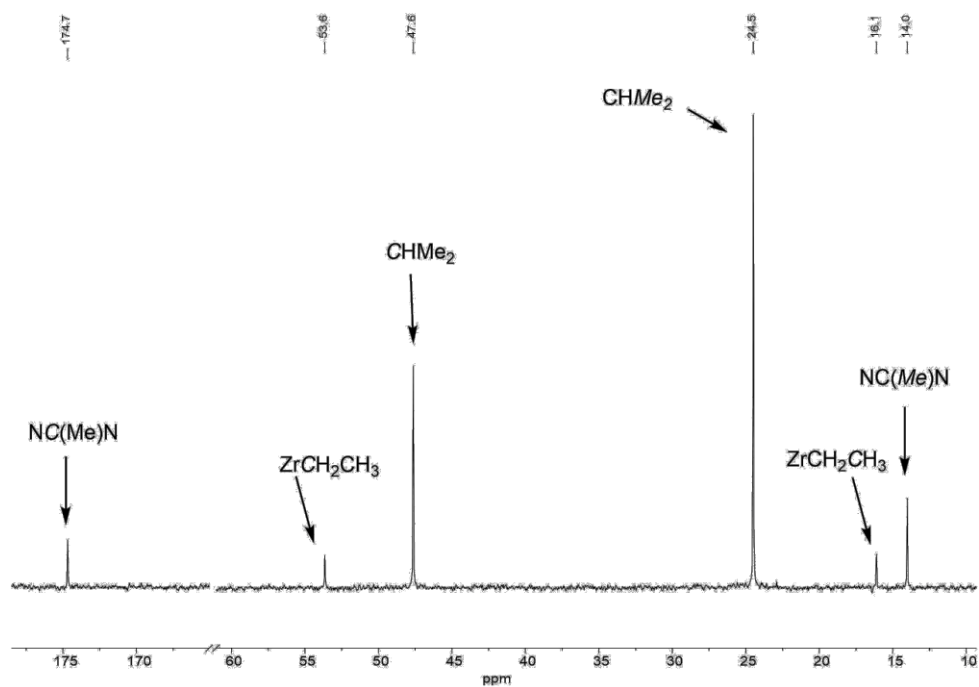


Fig. S25. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of Zr-Et (**5**) in benzene- d_6 at 23 °C.

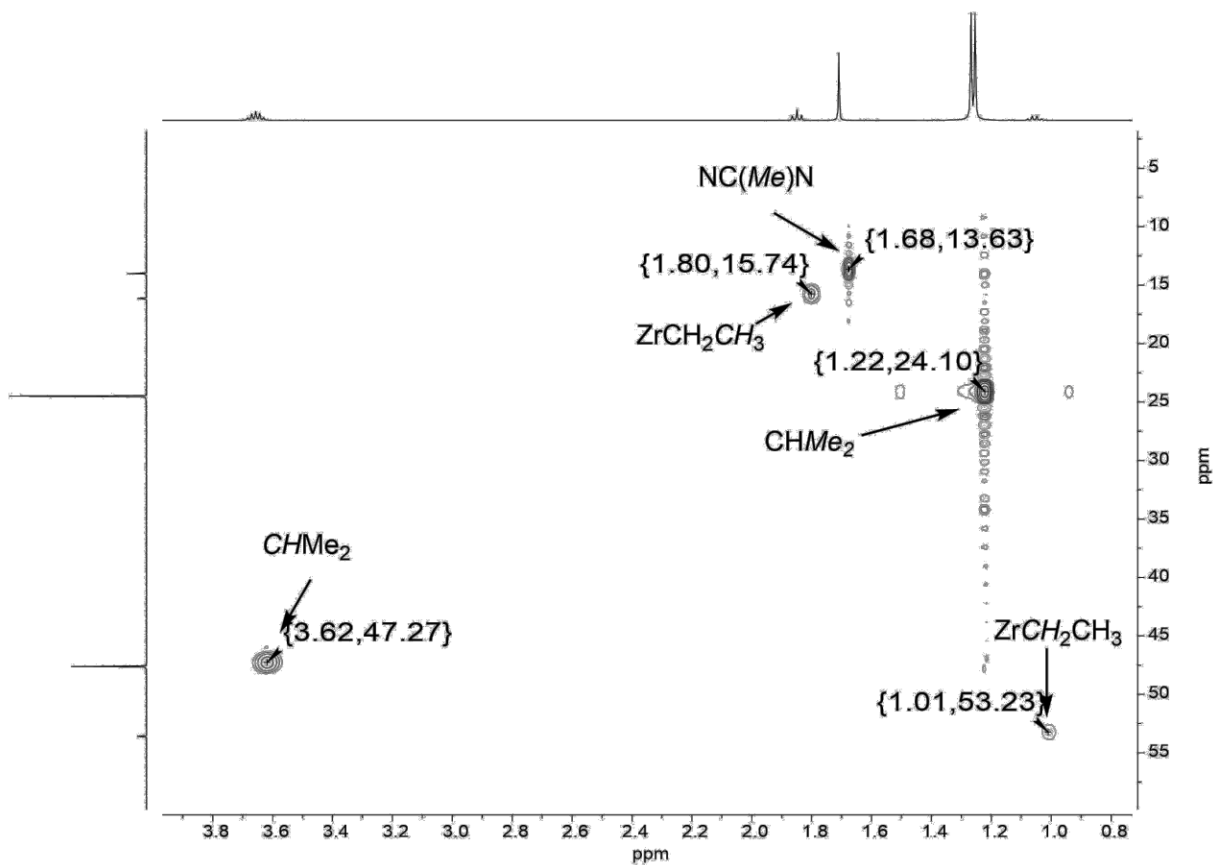


Fig. S26. ^1H - ^{13}C HSQC NMR spectrum of Zr-Et (**5**) in benzene- d_6 at 23 °C.

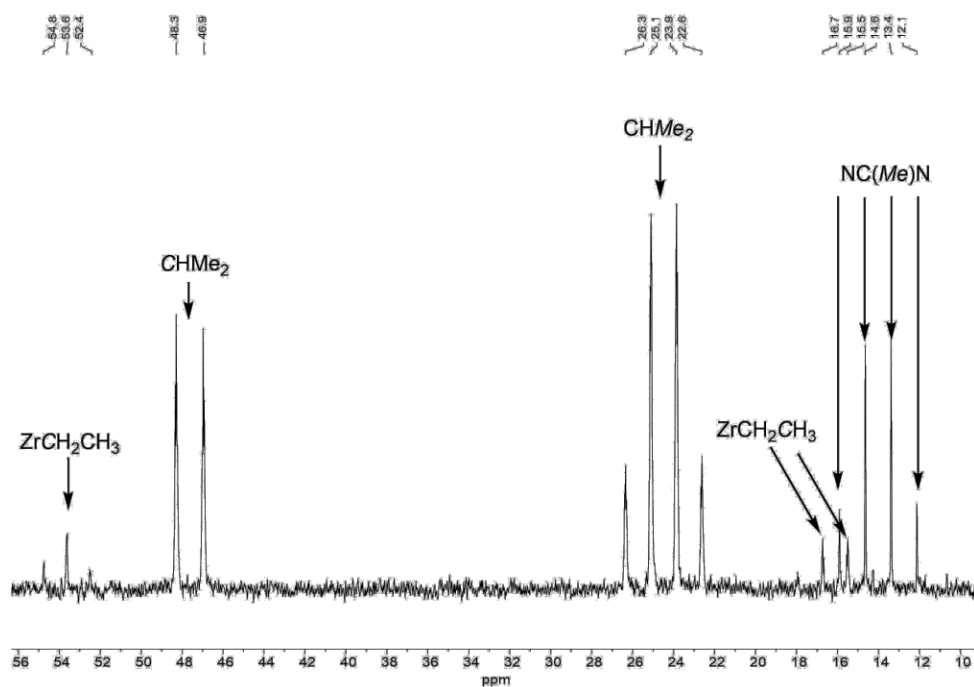


Fig. S27. Gated ^{13}C NMR spectrum of Zr-Et (**5**) in benzene- d_6 at 23 °C.

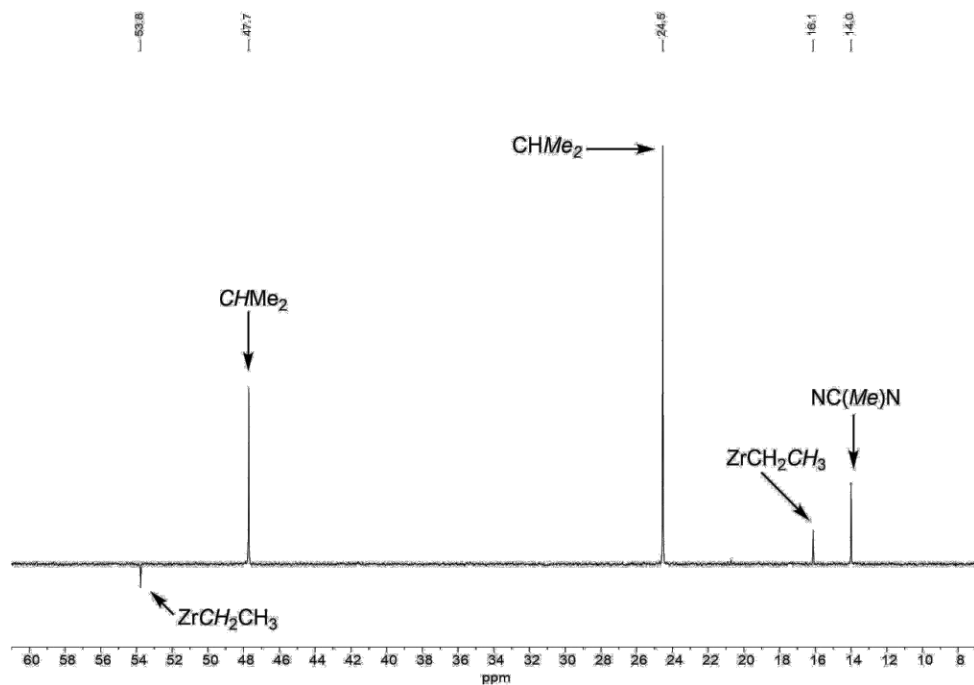


Fig. S28. DEPT-135 NMR spectrum of Zr-Et (**5**) in toluene- d_8 at 23 °C.

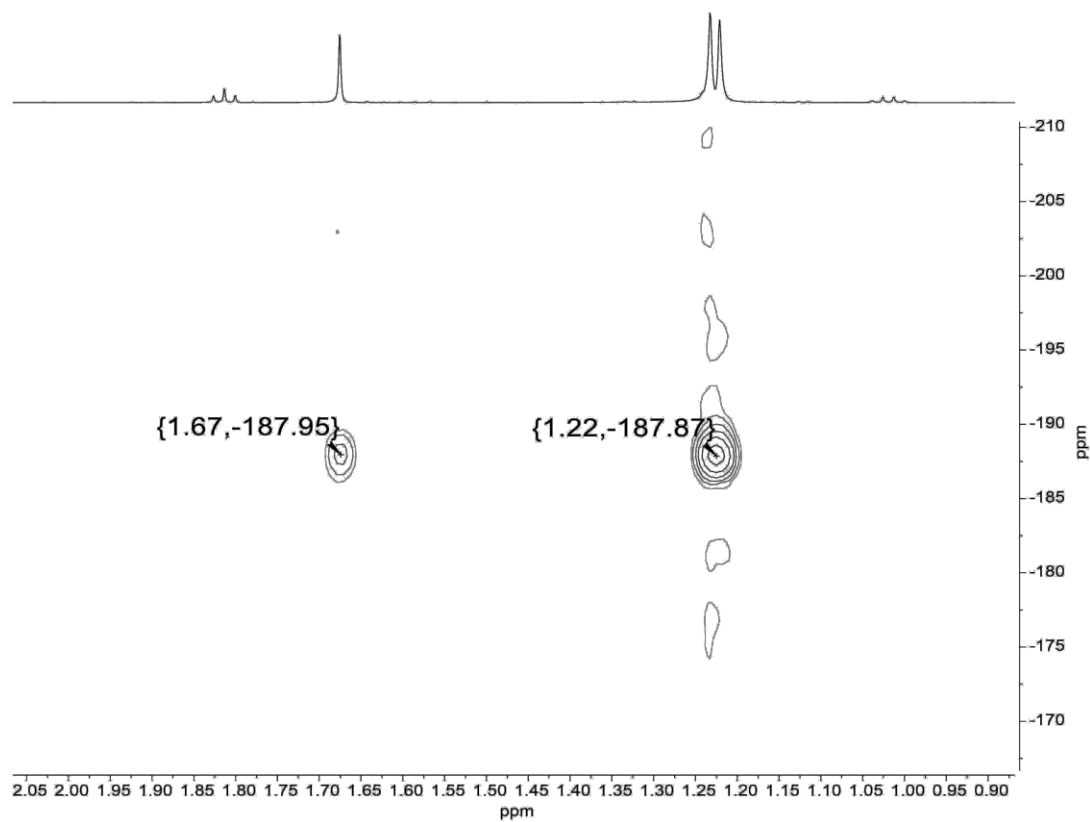


Fig. S29. ^1H - ^{15}N gHMBC NMR spectrum of Zr-Et (**5**) in benzene- d_6 at 25 °C.

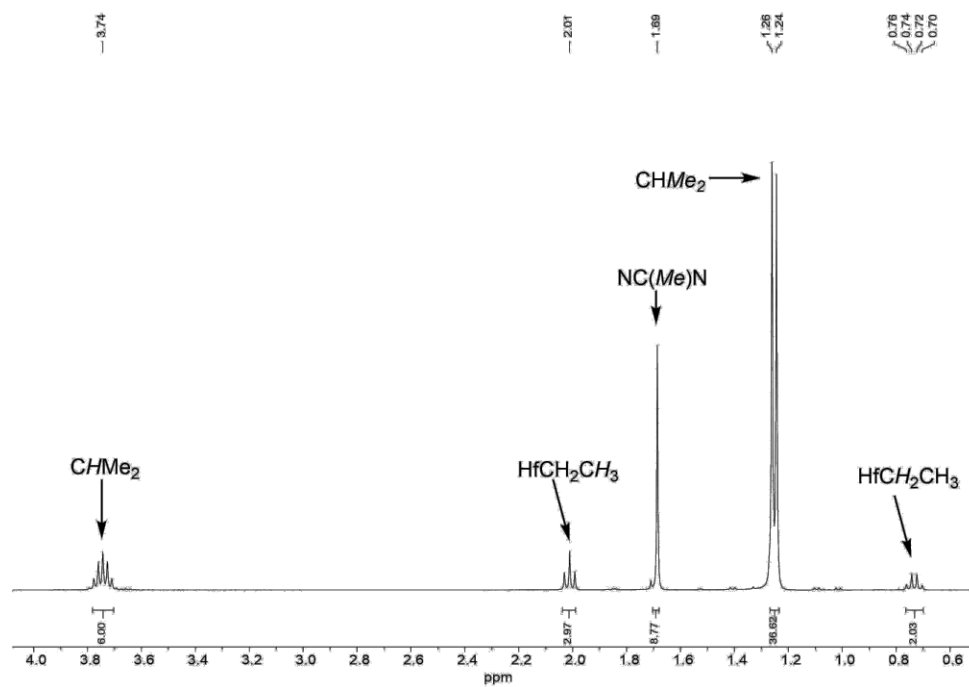


Fig. S30. ^1H NMR spectrum of Hf-Et (**6**) in benzene- d_6 at 23 °C.

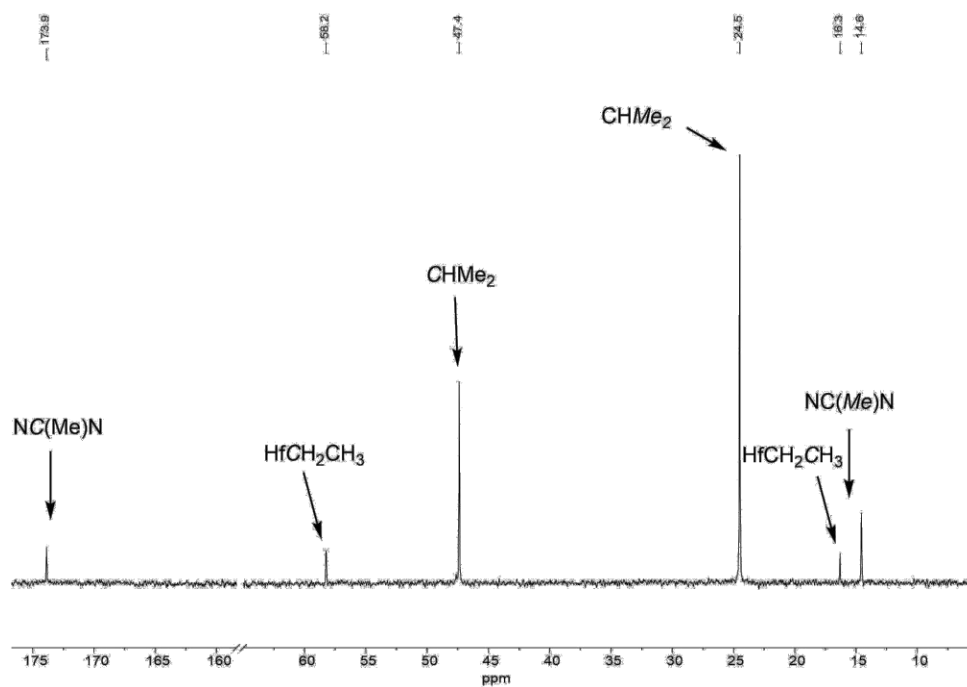


Fig. S31. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of Hf-Et (**6**) in benzene- d_6 at 23 °C.

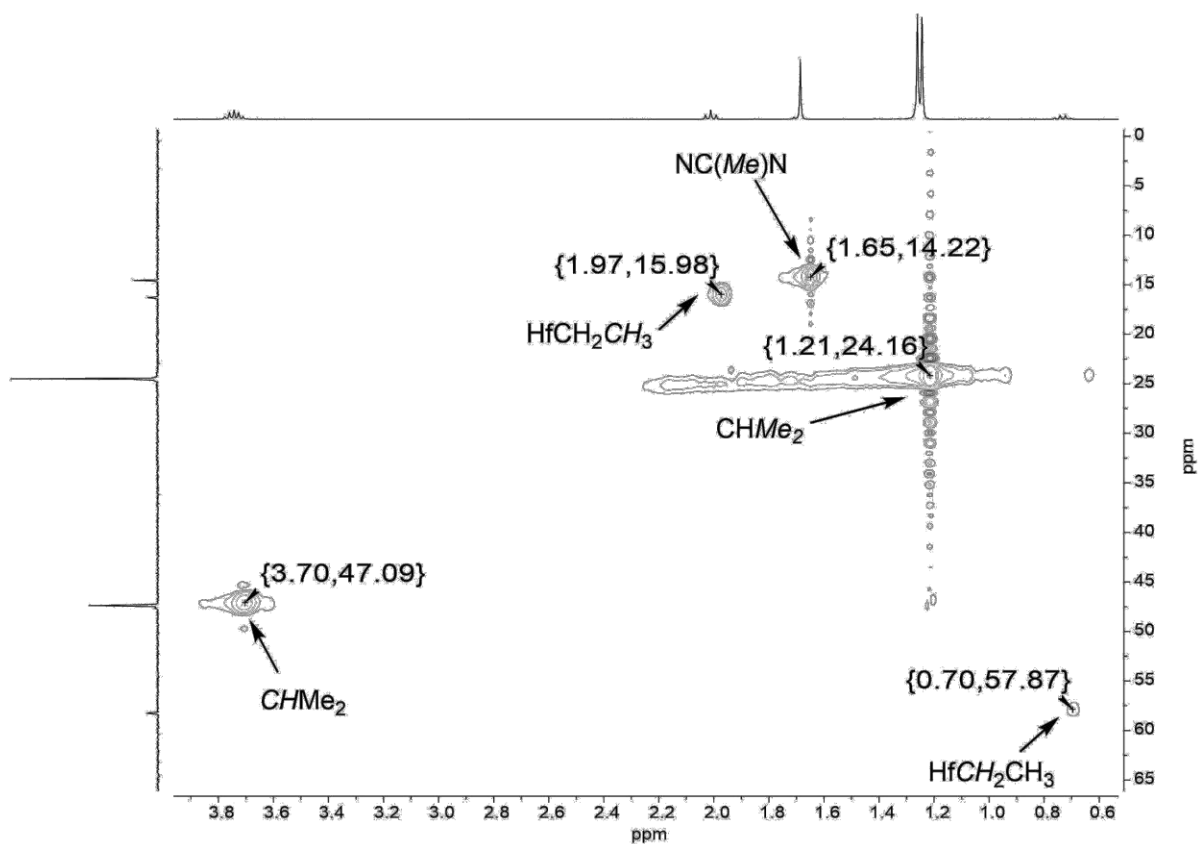


Fig. S32. ^1H - ^{13}C HSQC NMR spectrum of Hf-Et (**6**) in benzene- d_6 at 23 °C.

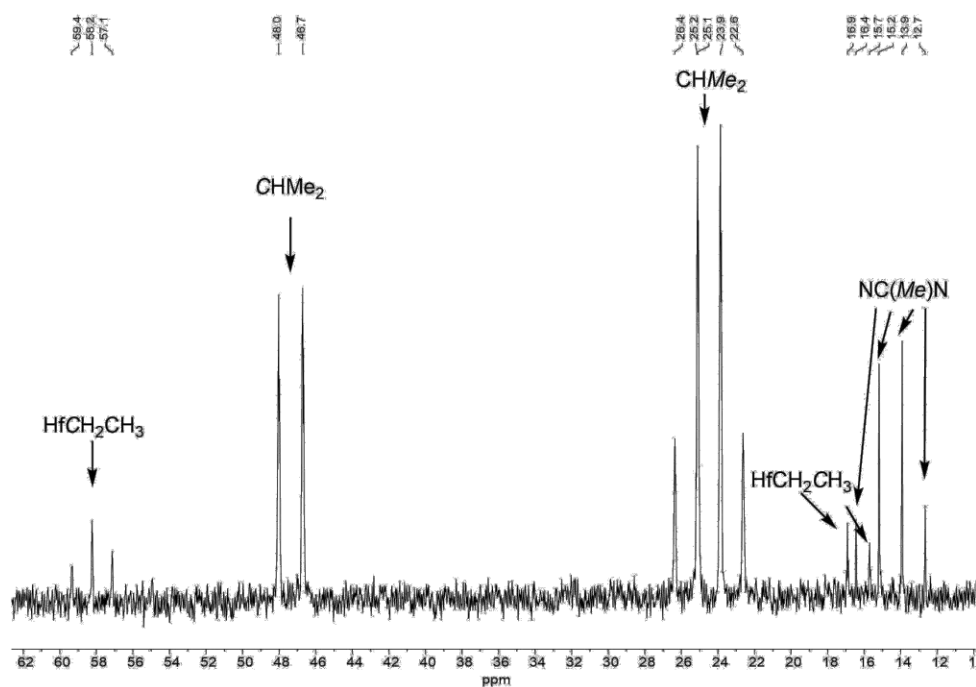


Fig. S33. Gated $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of Hf-Et (**6**) in benzene- d_6 at 23 °C.

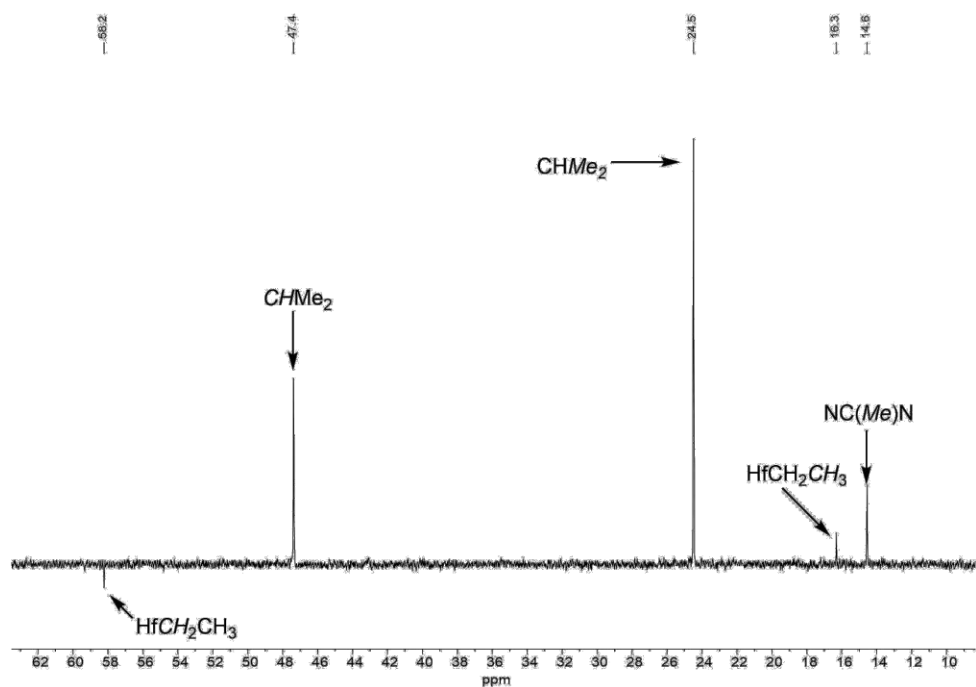


Fig. S34. DEPT-135 NMR spectrum of Hf-Et (**6**) in benzene- d_6 at 23 °C.

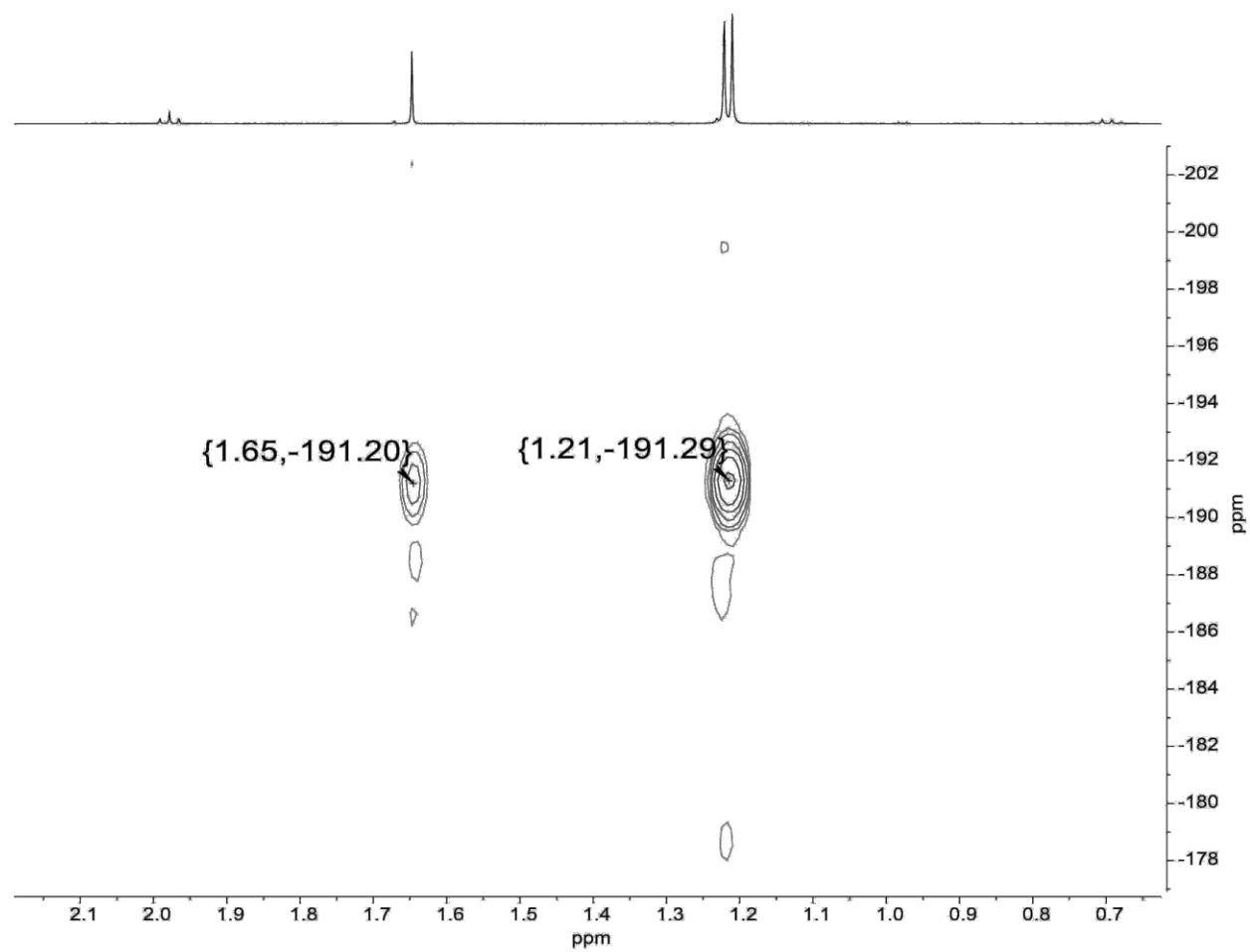


Fig. S35. ^1H - ^{15}N gHMBC NMR spectrum of Hf-Et (**6**) in benzene- d_6 at 25 °C.

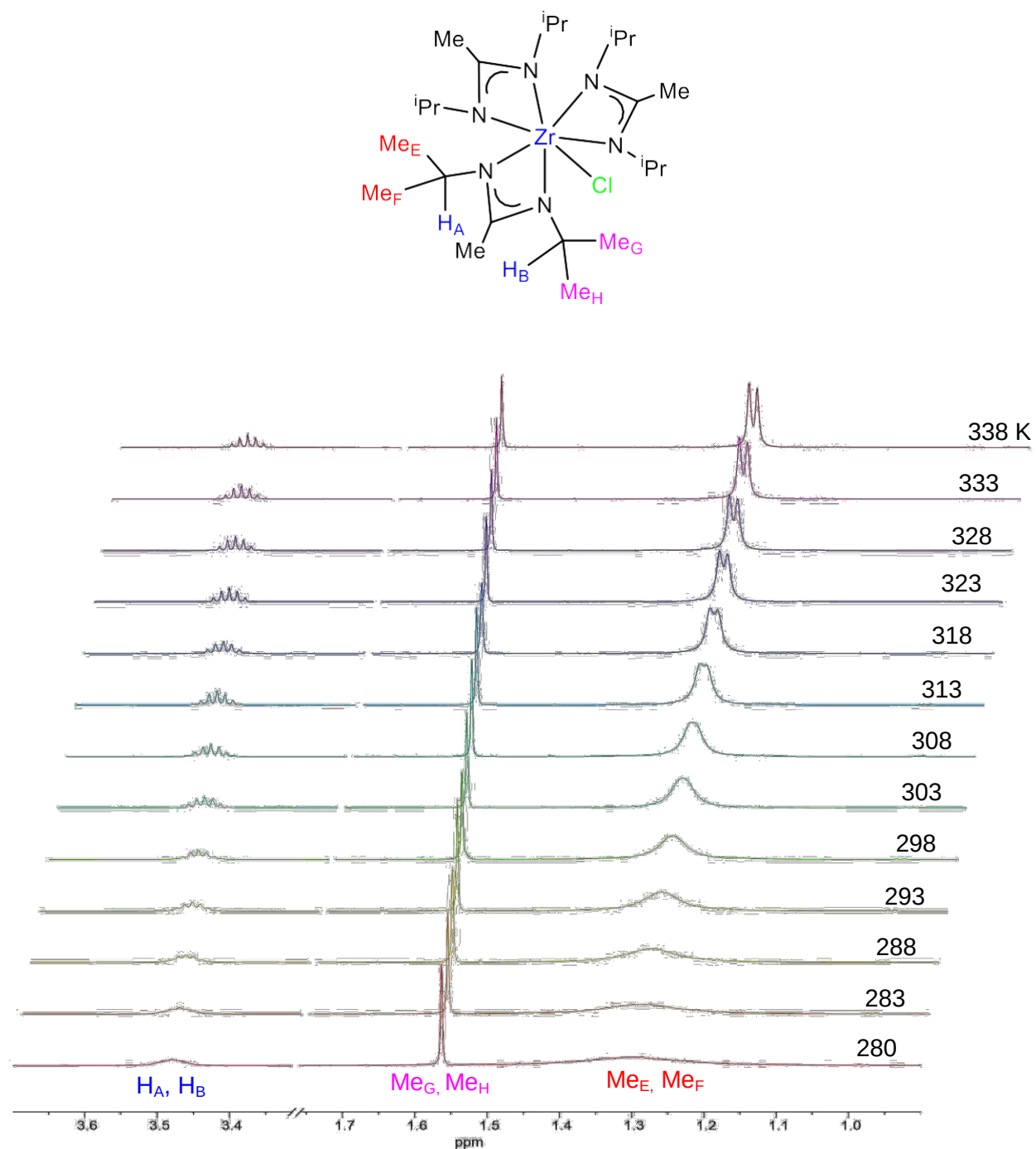


Fig. S36. Partial ^1H NMR spectra of Zr-Cl (1) in benzene- d_6 at several temperatures recorded on a 500 MHz NMR spectrometer.

Table S7. Rate constants of the interconversions in Zr-Cl (**1**)

<i>T</i> (K)	<i>k</i> (s ⁻¹) for <i>H_AH_B</i> ^b	<i>k</i> (s ⁻¹) for <i>Me_EMe_F</i>	<i>k</i> (s ⁻¹) for <i>Me_GMe_H</i>	<i>k</i> (s ⁻¹) for <i>Me_{E-F}Me_{G-H}</i>
203(1)		7.86(7)		
208(1)		10.7(4)	28.0(1)	
213(1)		14.51(7)	37.4(7)	
218(1)		18.3(4)	48(2)	
223(1)		25.2(6)	55(1)	
228(1)	131(2)	34(1)	69(3)	
233(1)	143(3)	46(1)	84.8(2)	
238(1)	154(3)	52.3(9)	112(5)	297.9(6)
243(1)	177(3)		152(2)	333(3)
248(1)	211(6)			365(16)
253(1)	249(5)			431(8)
258(1)	279(1)			482(11)
263(1)	301(1)			573(15)
268(1)				688(3)
273(1)				745(3)

^a The total uncertainties $\delta k/k$ were calculated from $\delta k_{\text{ran}}/k$ from each column and $\delta k_{\text{sys}}/k = 0.050$. (1) *H_AH_B*: $\delta k_{\text{ran}}/k = 0.027$, $\delta k/k = 0.057$; (2) *Me_EMe_F*: $\delta k_{\text{ran}}/k = 0.039$, $\delta k/k = 0.063$; (3) *Me_GMe_H*: $\delta k_{\text{ran}}/k = 0.045$, $\delta k/k = 0.067$; (4) *Me_{E-F}Me_{G-H}*: $\delta k_{\text{ran}}/k = 0.043$, $\delta k/k = 0.066$.

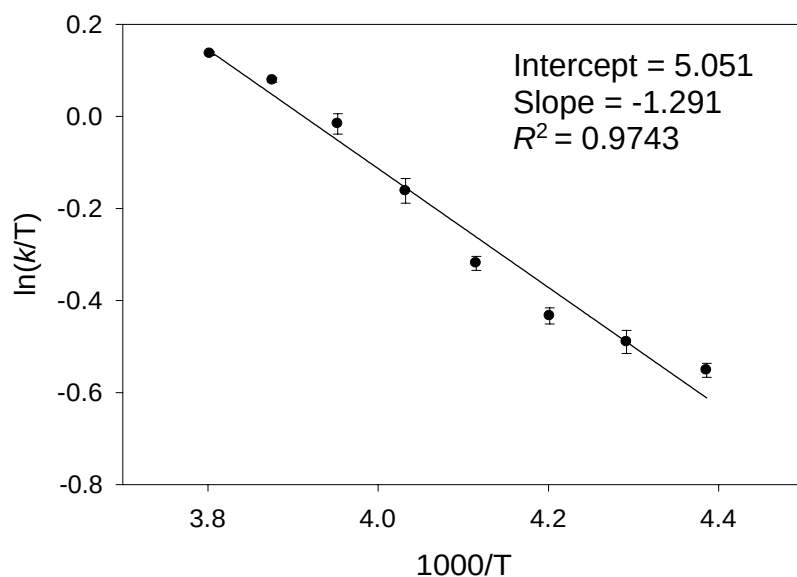


Fig. S37. Eyring plot of the $H_A H_B$ exchange in Zr-Cl (**1**) yielding activation parameters for the exchange.

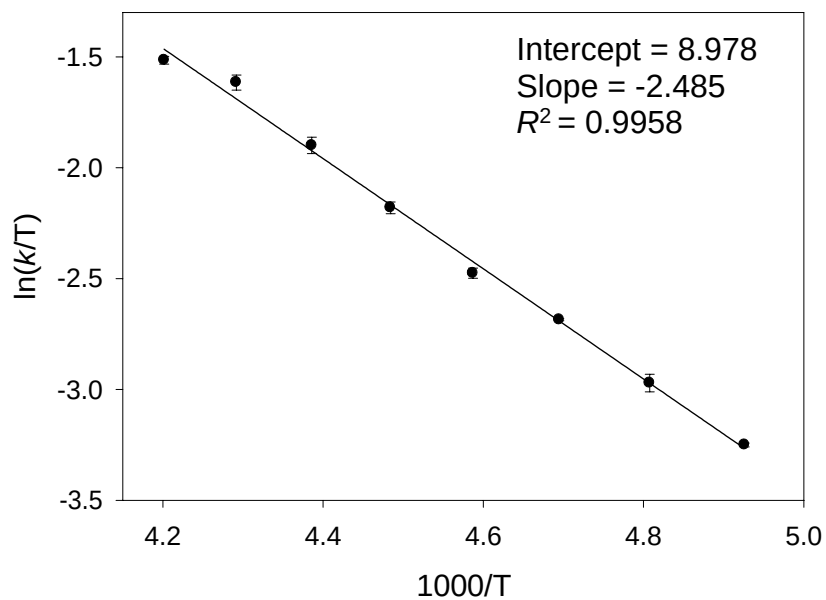


Fig. S38. Eyring plot of the $Me_E Me_F$ exchange in Zr-Cl (**1**) yielding activation parameters for the exchange.

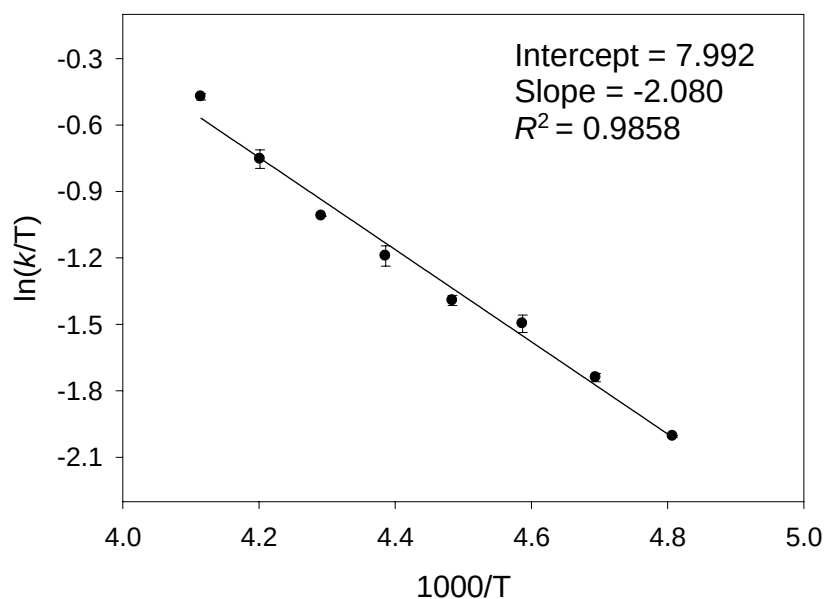


Fig. S39. Eyring plot of the Me_GMe_H exchange in Zr-Cl (**1**) yielding activation parameters for the exchange.

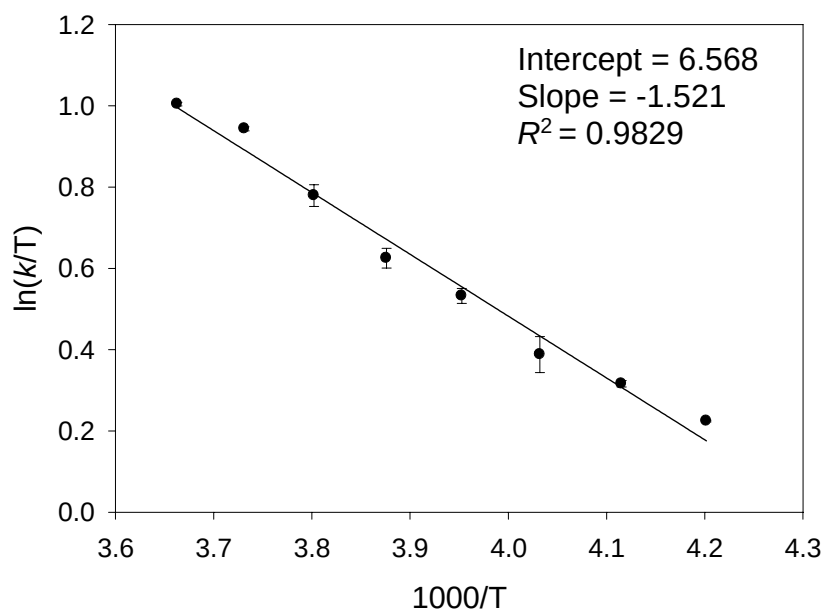


Fig. S40. Eyring plot of the $Me_{E-F}Me_{G-H}$ exchange in Zr-Cl (**1**) yielding activation parameters for the exchange.

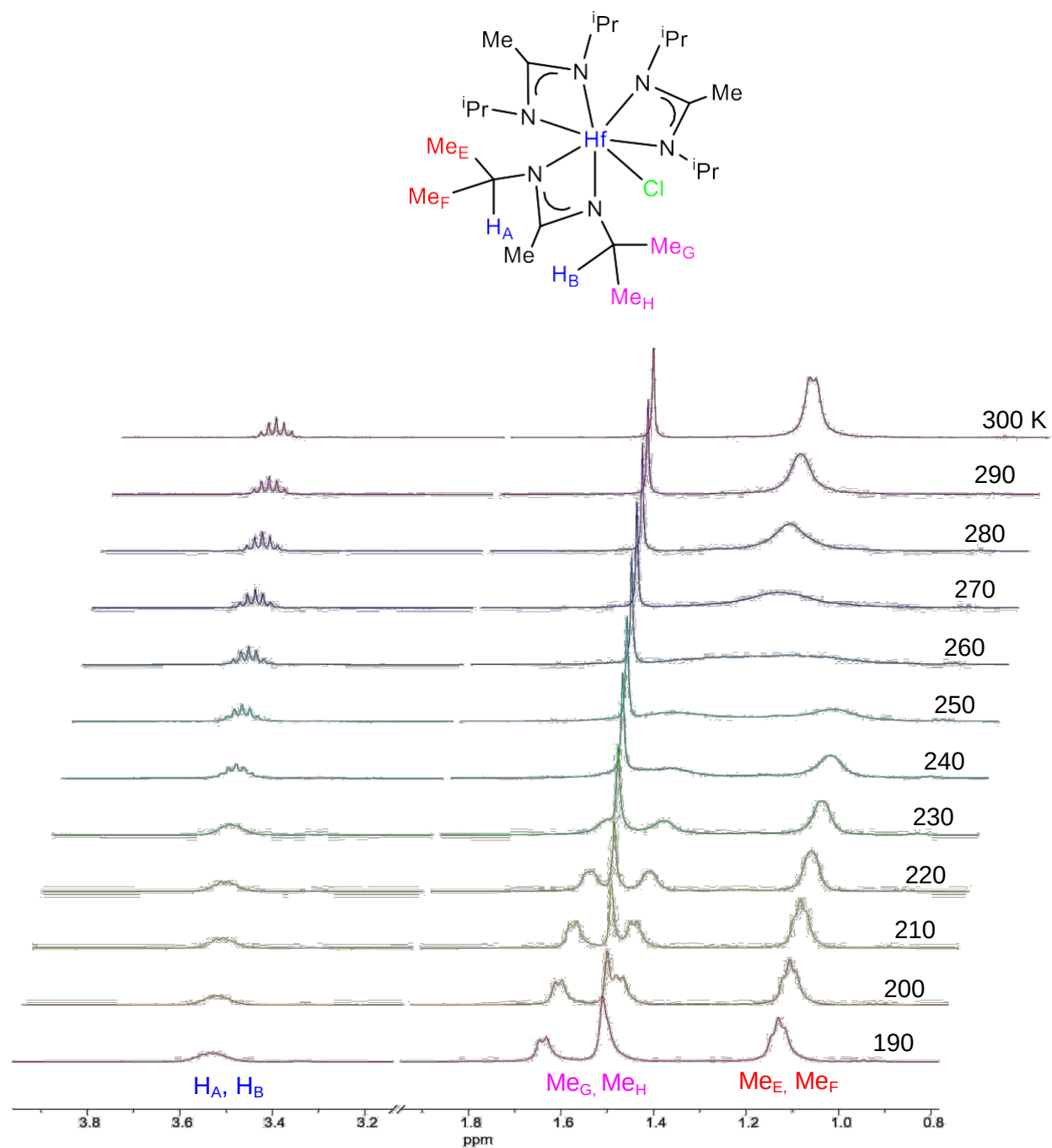


Fig. S41. Partial ^1H NMR spectra of Hf-Cl (2) at several temperatures in toluene- d_8 recorded on a 400 MHz NMR spectrometer.

Table S8. Rate constants for the interconversions in Hf-Cl (2)

<i>T</i> (K)	<i>k</i> (s ⁻¹) for <i>Me_EMe_F</i>	<i>k</i> (s ⁻¹) for <i>Me_GMe_H</i>	<i>k</i> (s ⁻¹) for <i>Me_{E-F}Me_{G-H}</i>
200(1)	7.4(1)		
205(1)	8.44(4)		
210(1)	10.54(0)		
215(1)	13.0(2)	51.3(9)	
220(1)	15.0(4)	61(3)	
225(1)	16.9(5)	76.9(6)	
230(1)	20.3(5)	102(2)	287(6)
235(1)	26.01(0)	141(6)	318.4(6)
240(1)		169(5)	373(2)
245(1)		222.0(6)	425(13)
250(1)		236.6(7)	485(18)
255(1)			610(17)
260(1)			702(6)
265(1)			765(6)

^a The total uncertainties $\delta k/k$ were calculated from $\delta k_{\text{ran}}/k$ from each column and $\delta k_{\text{sys}}/k = 0.050$. (1) *Me_EMe_F*: $\delta k_{\text{ran}}/k = 0.029$, $\delta k/k = 0.058$; (2) *Me_GMe_H*: $\delta k_{\text{ran}}/k = 0.065$, $\delta k/k = 0.081$; (3) *Me_{E-F}Me_{G-H}*: $\delta k_{\text{ran}}/k = 0.038$, $\delta k/k = 0.063$.

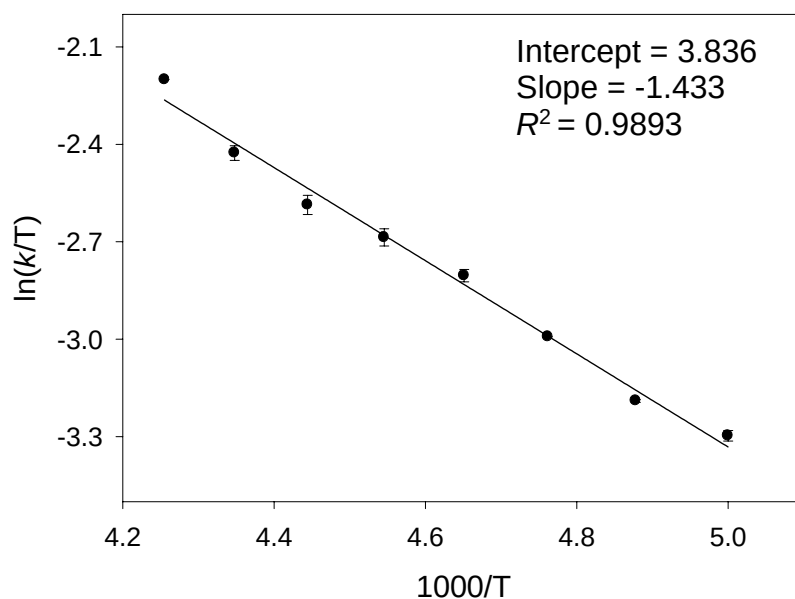


Fig. S42. Eyring plot of the $Me_E Me_F$ exchange in Hf-Cl (2) yielding activation parameters for the exchange.

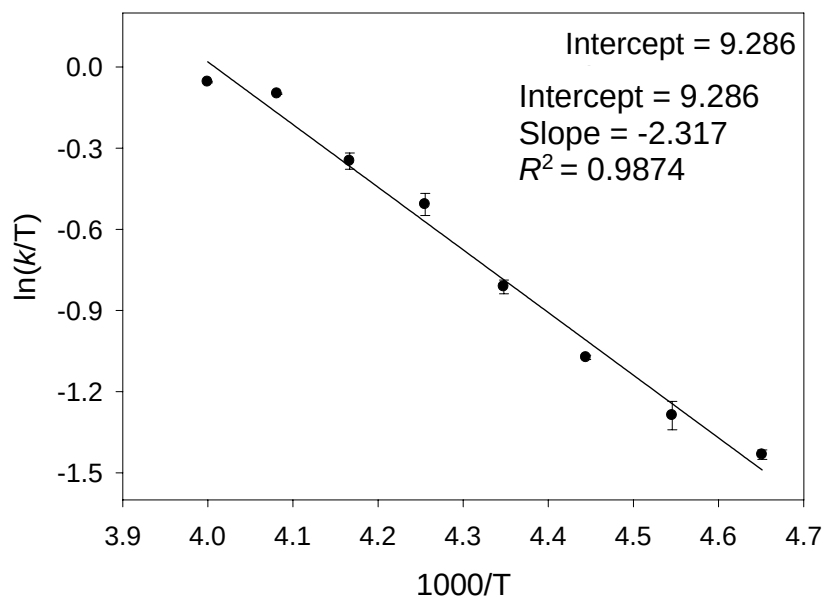


Fig. S43. Eyring plot of the $Me_G Me_H$ exchange in Hf-Cl (2) yielding activation parameters for the exchange.

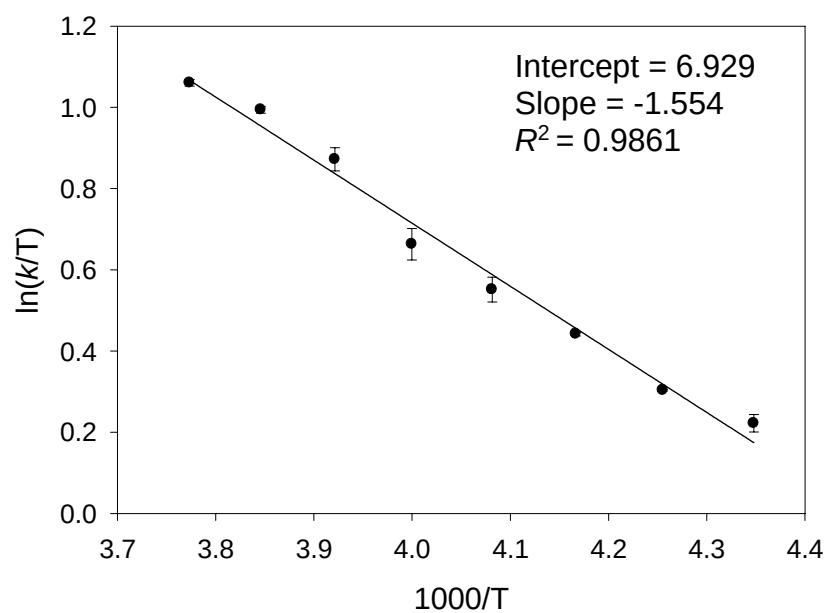


Fig. S44. Eyring plot of the $Me_{E-F}Me_{G-H}$ exchange in Hf-Cl (2) yielding activation parameters for the exchange.

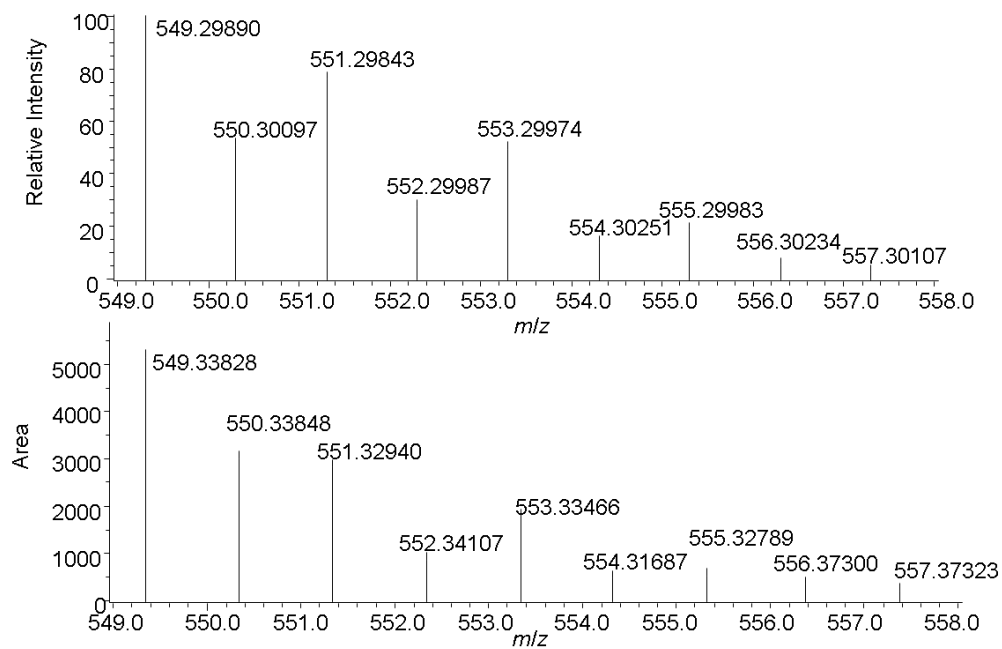


Fig. S45. (Top) Calculated and (Bottom) Observed MS for $[\text{Zr-Cl (1)}+\text{H}^+]$.

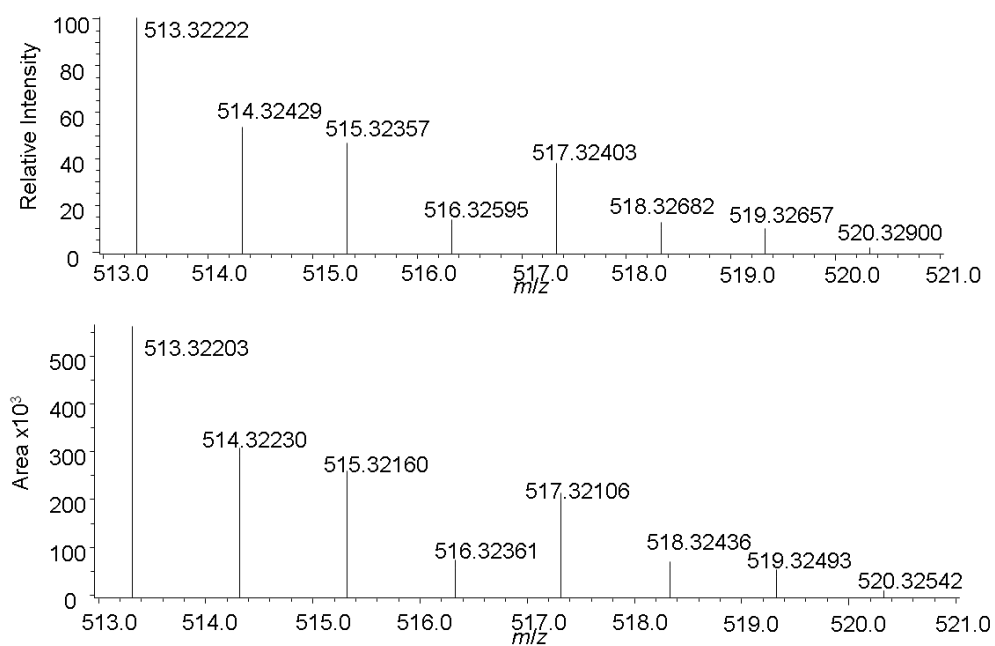


Fig. S46. (Top) Calculated and (Bottom) Observed MS for $\text{Zr}[\text{MeC}(\text{N}^i\text{Pr})_2]_3^+$ (**7**).

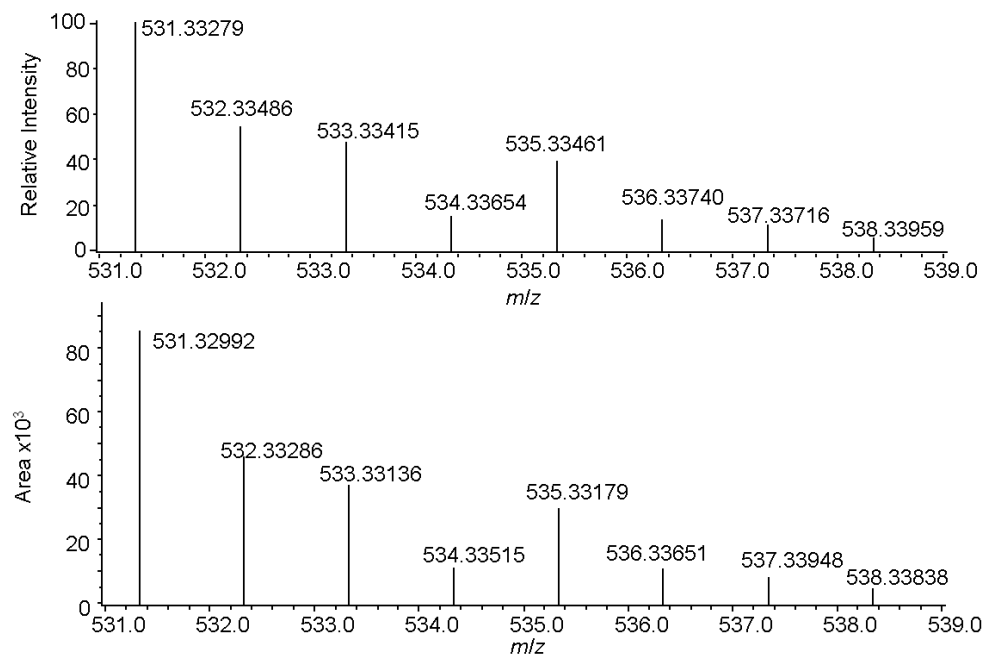


Fig. S47. (Top) Calculated and (Bottom) Observed MS of $[\text{Zr-OH (9)}+\text{H}^+]$.

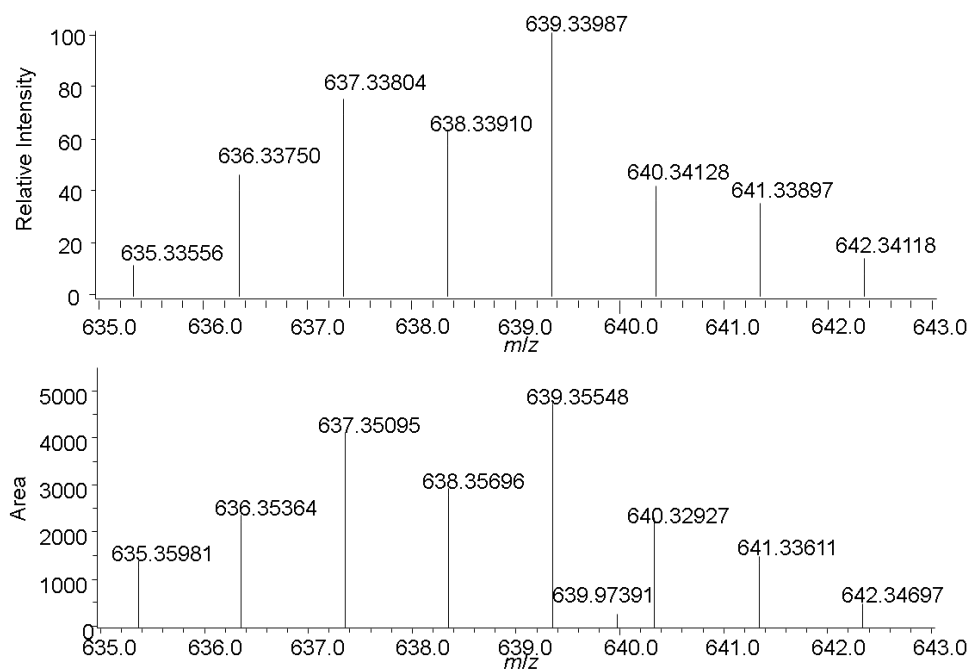


Fig. S48. (Top) Calculated and (Bottom) Observed MS for $[\text{Hf-Cl (2)}+\text{H}^+]$.

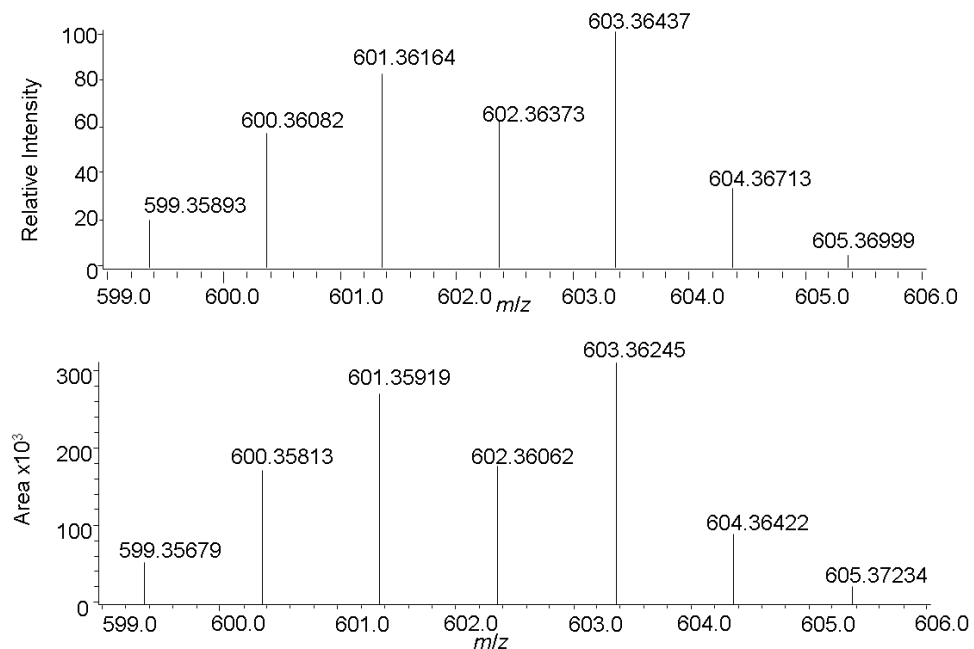


Fig. S49. (Top) Calculated and (Bottom) Observed MS for $\text{Hf}[\text{MeC}(\text{N}^i\text{Pr})_2]_3^+$ (**8**).

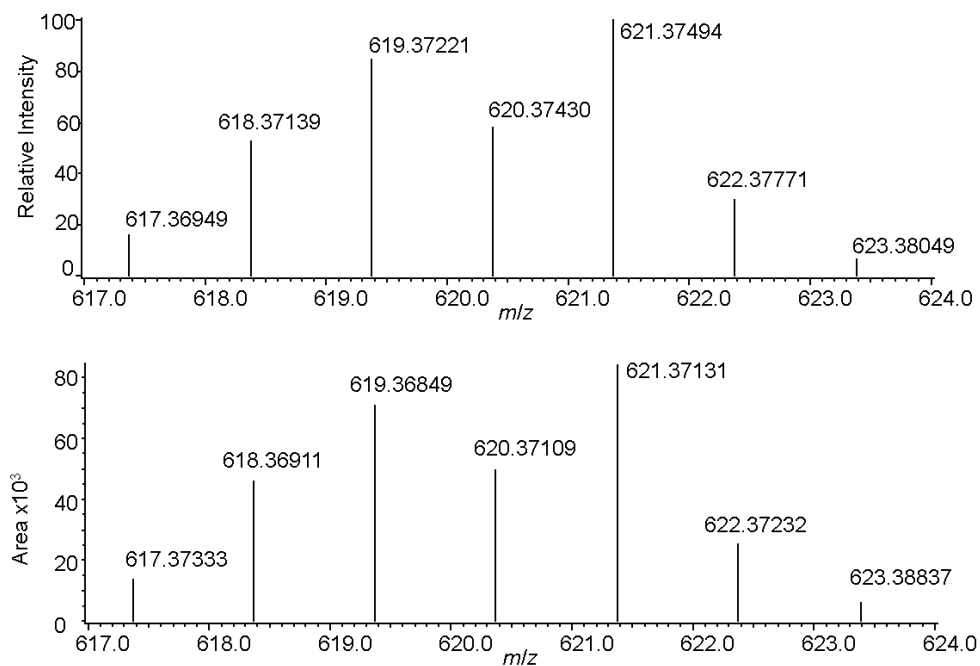


Fig. S50. (Top) Calculated and (Bottom) Observed MS of $[\text{Hf}-\text{OH}(\textbf{10})+\text{H}^+]$.