

Supporting Information for:

Design and Synthesis of a Family of 1D-Lanthanide-Coordination Polymers Showing Luminescence and Slow Relaxation of the Magnetization

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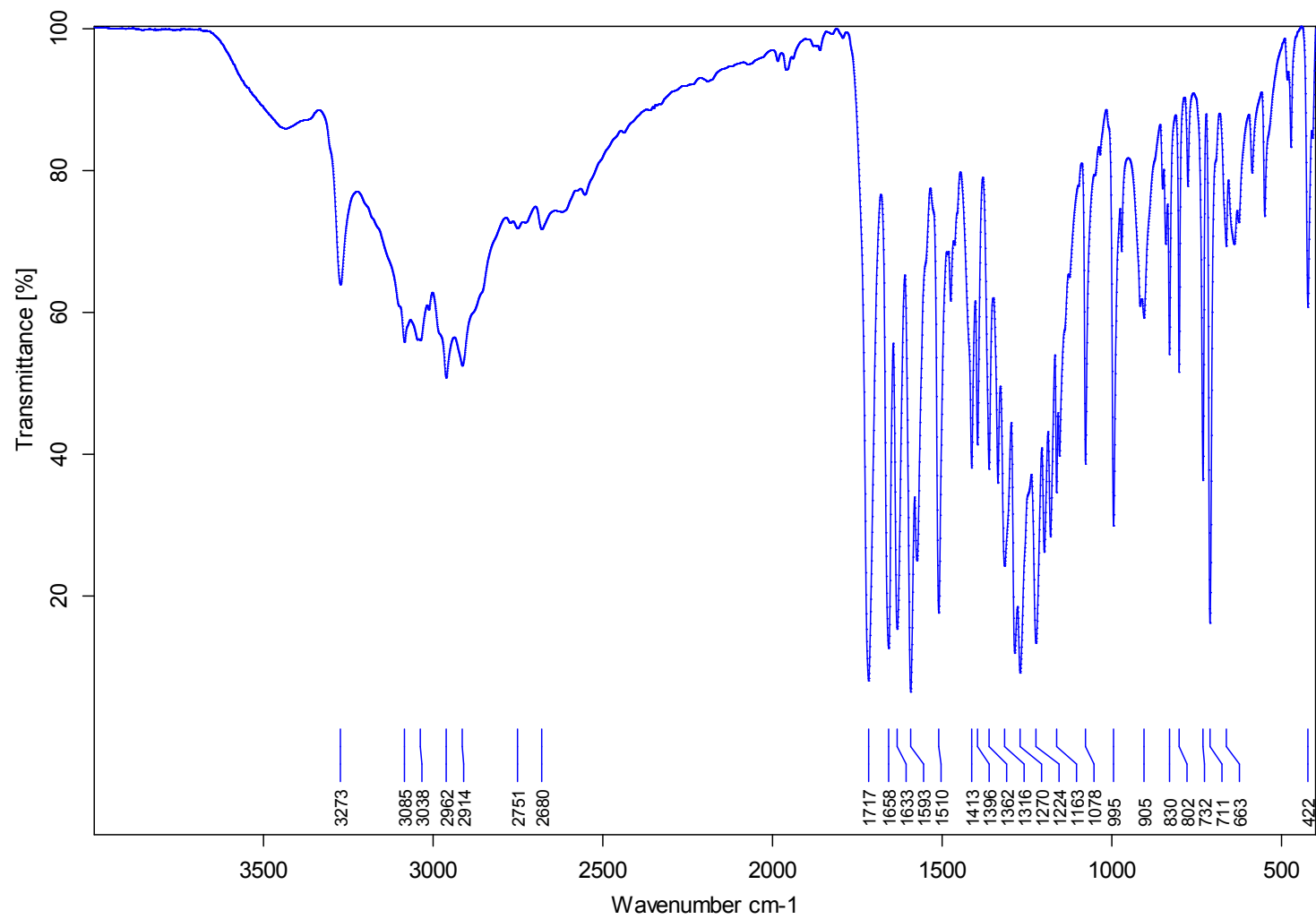


Figure S1. FT-IR (KBr) spectrum of **1**.

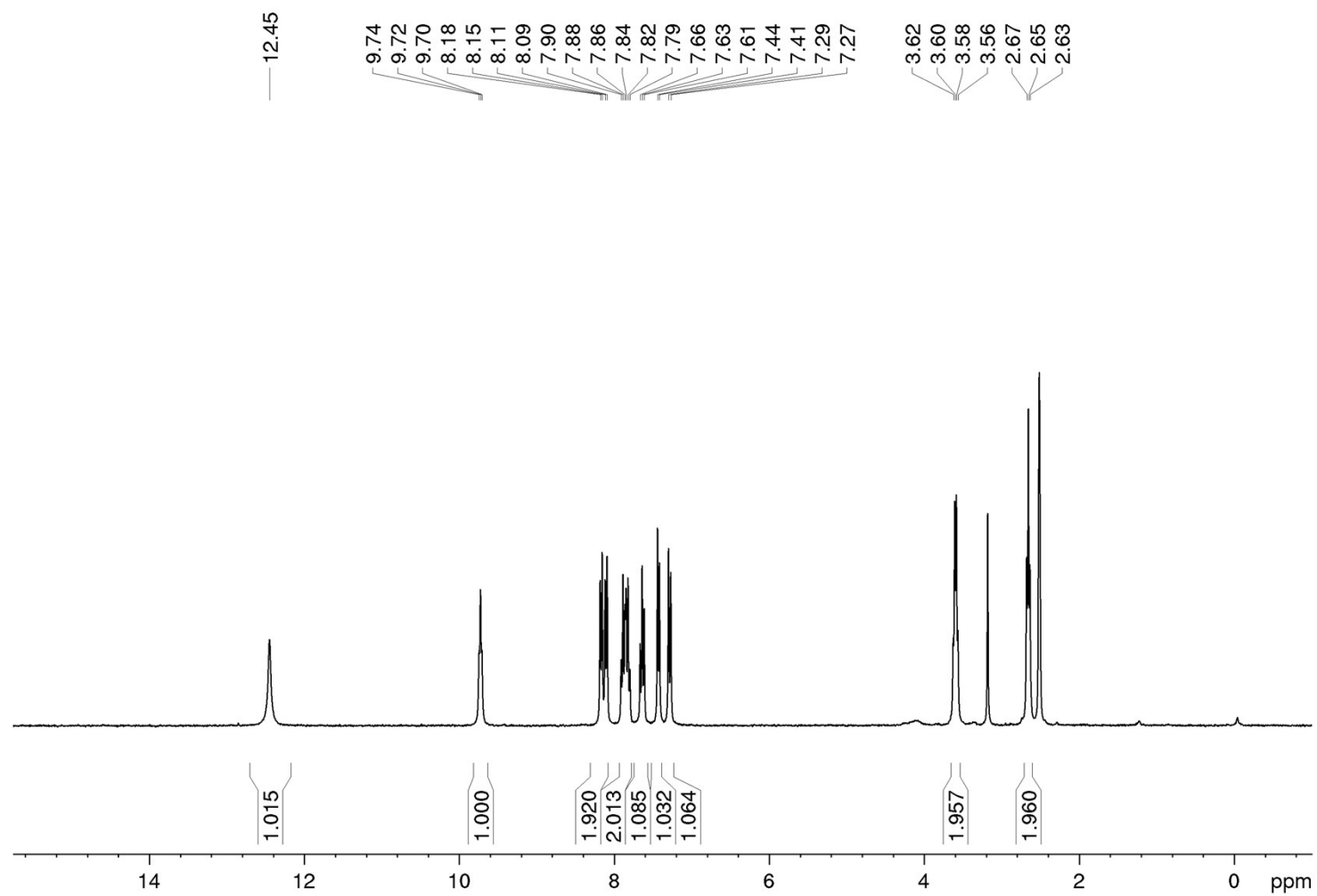


Figure S2. ¹H NMR spectrum of **1** in DMSO-*d*₆ at room temperature (300.13 MHz).

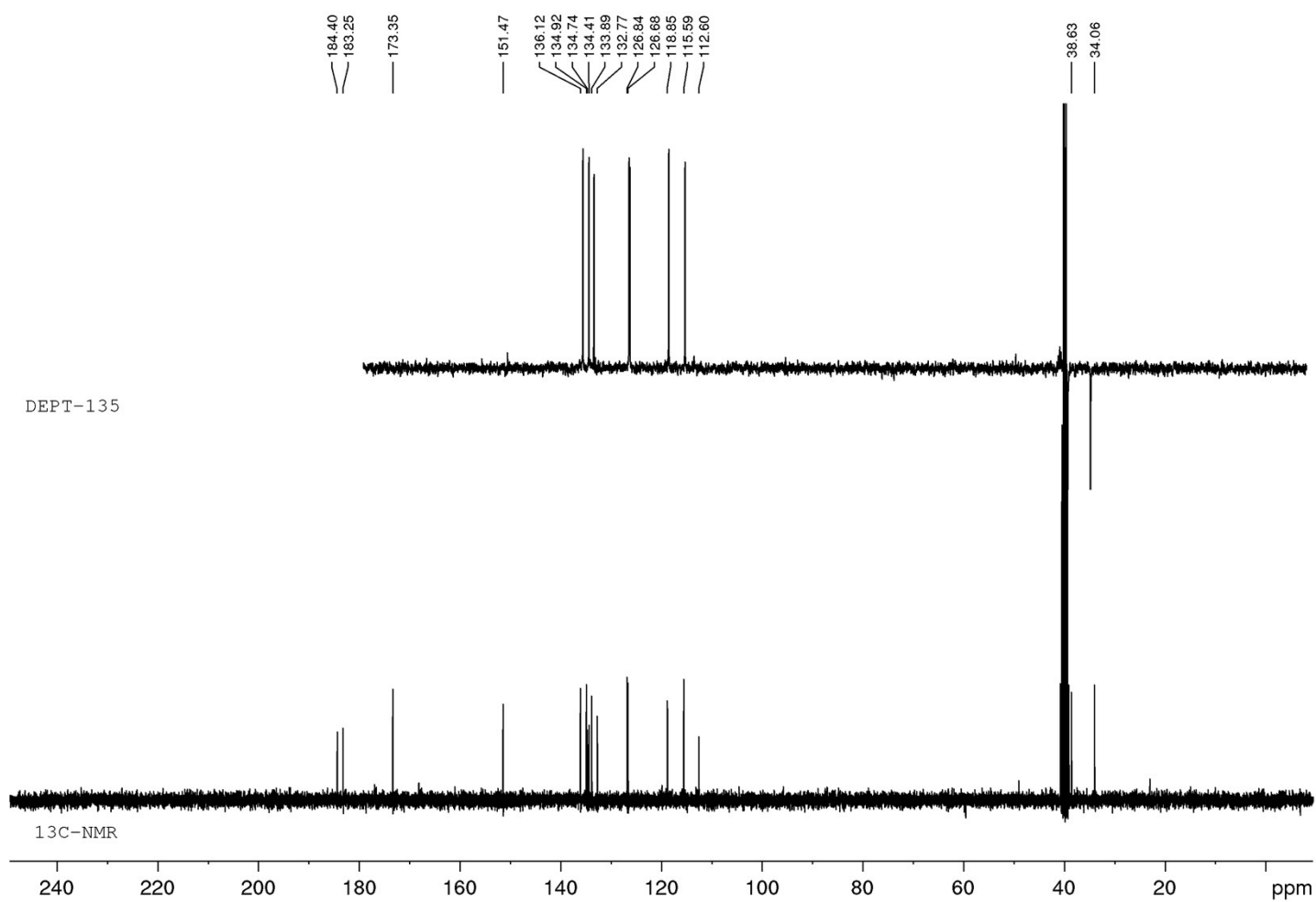


Figure S3. ¹³C NMR (bottom) and DEPT-135 (top) NMR spectra of **1** in DMSO-*d*₆ at room temperature (75.47 MHz).

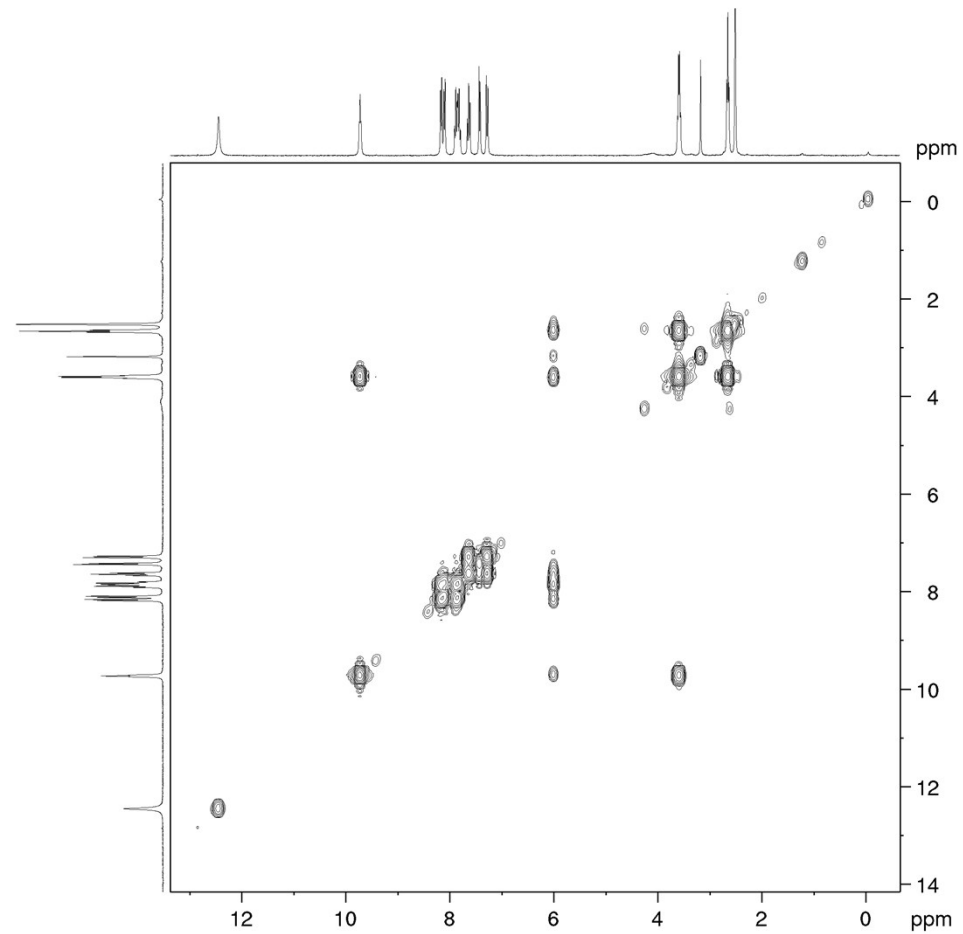


Figure S4. COSY spectrum of **1** in DMSO- d_6 at room temperature.

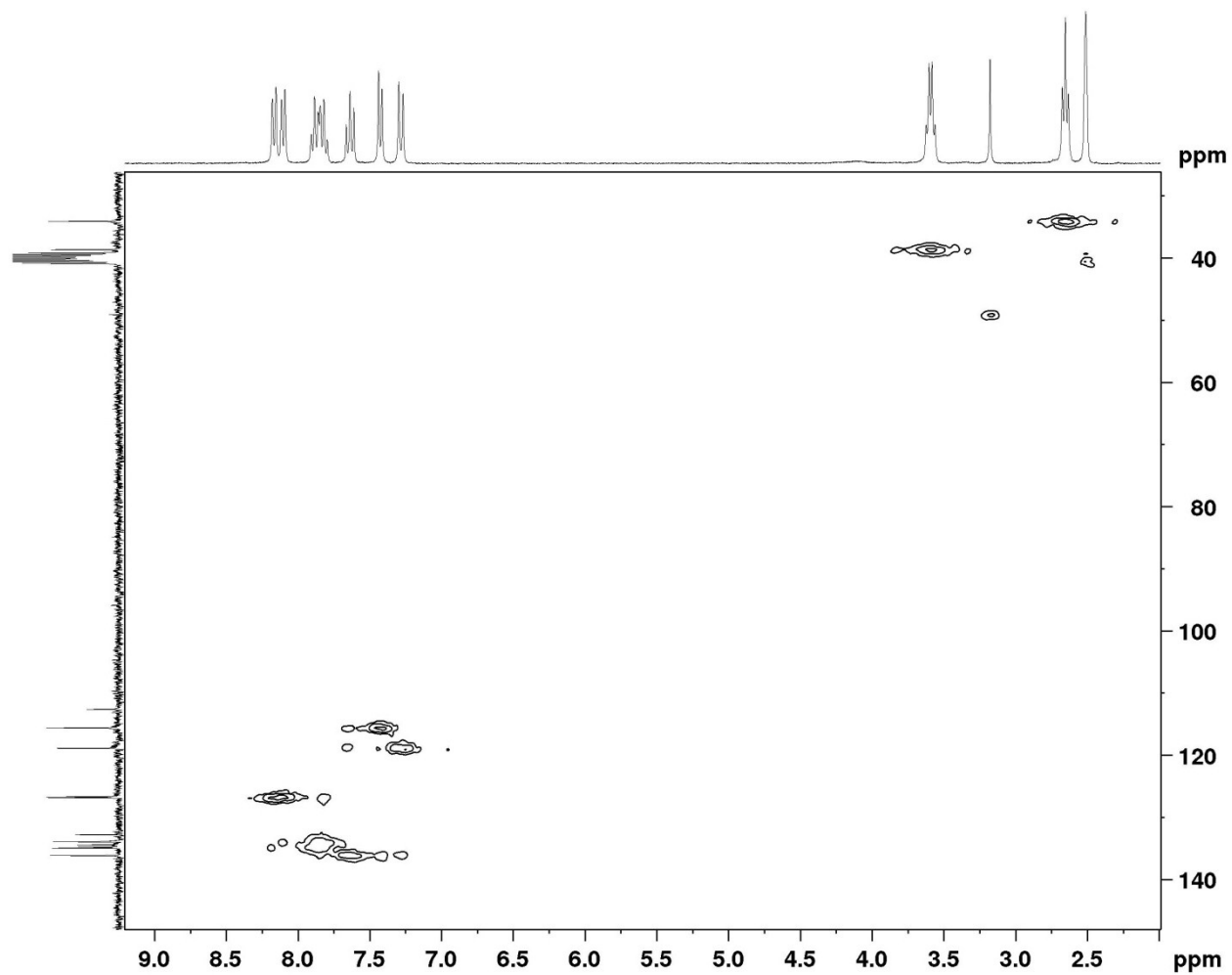


Figure S5. ^1H , ^{13}C gHMQC spectrum of **1** in $\text{DMSO}-d_6$ at room temperature.

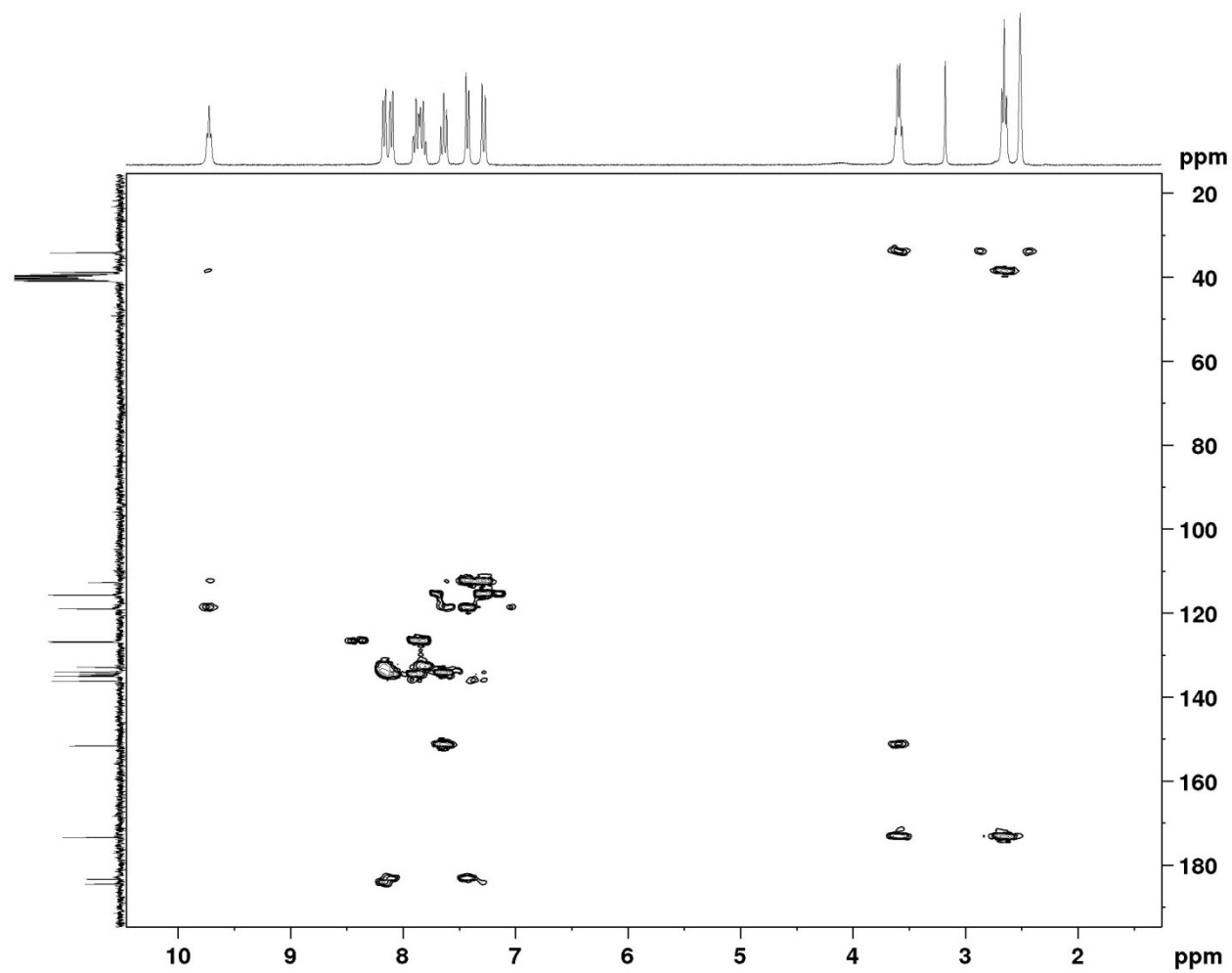


Figure S6. ^1H , ^{13}C gHMBC spectrum of **1** in $\text{DMSO}-d_6$ at room temperature.

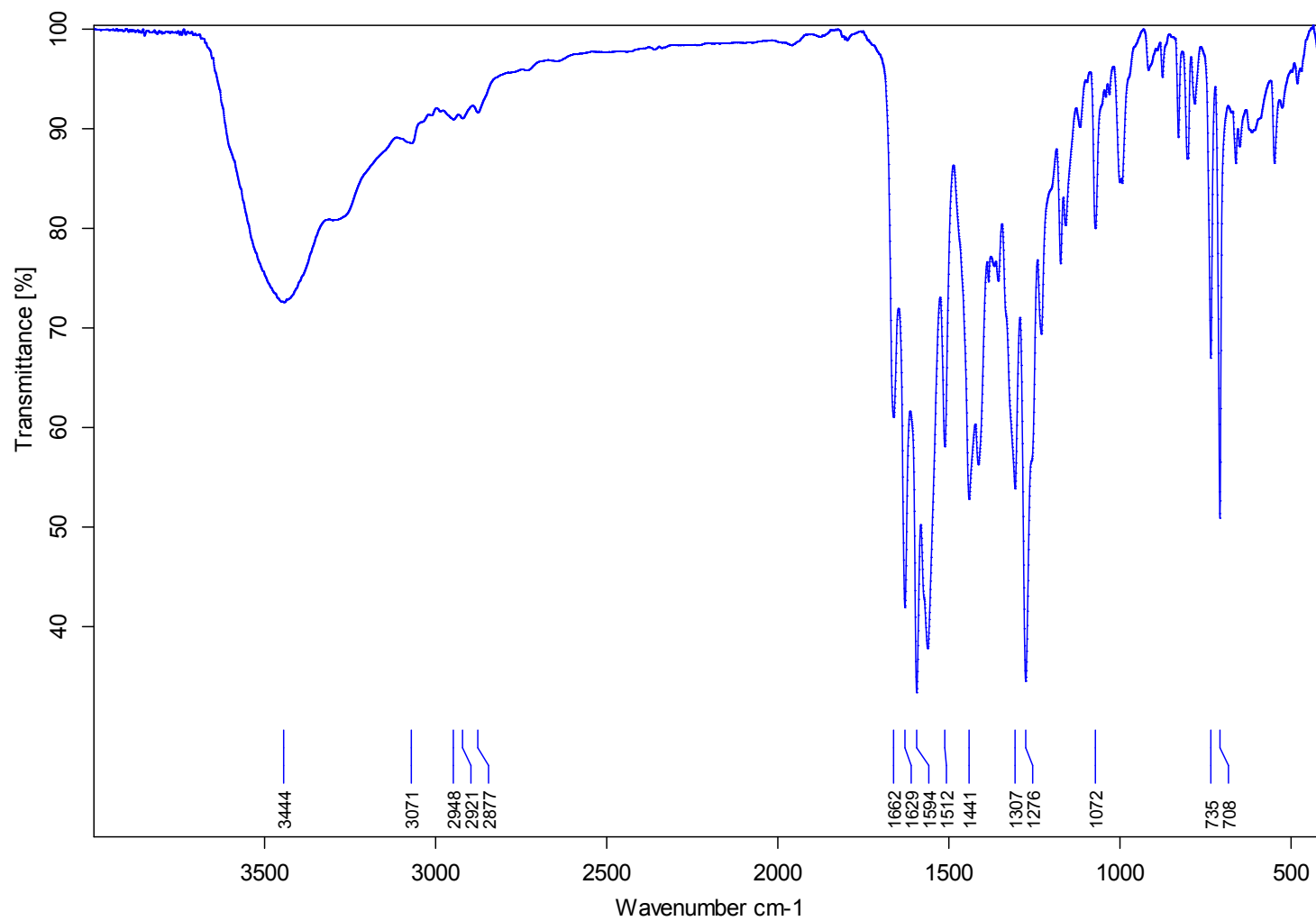


Figure S7. FT-IR (KBr) spectrum of complex 2.

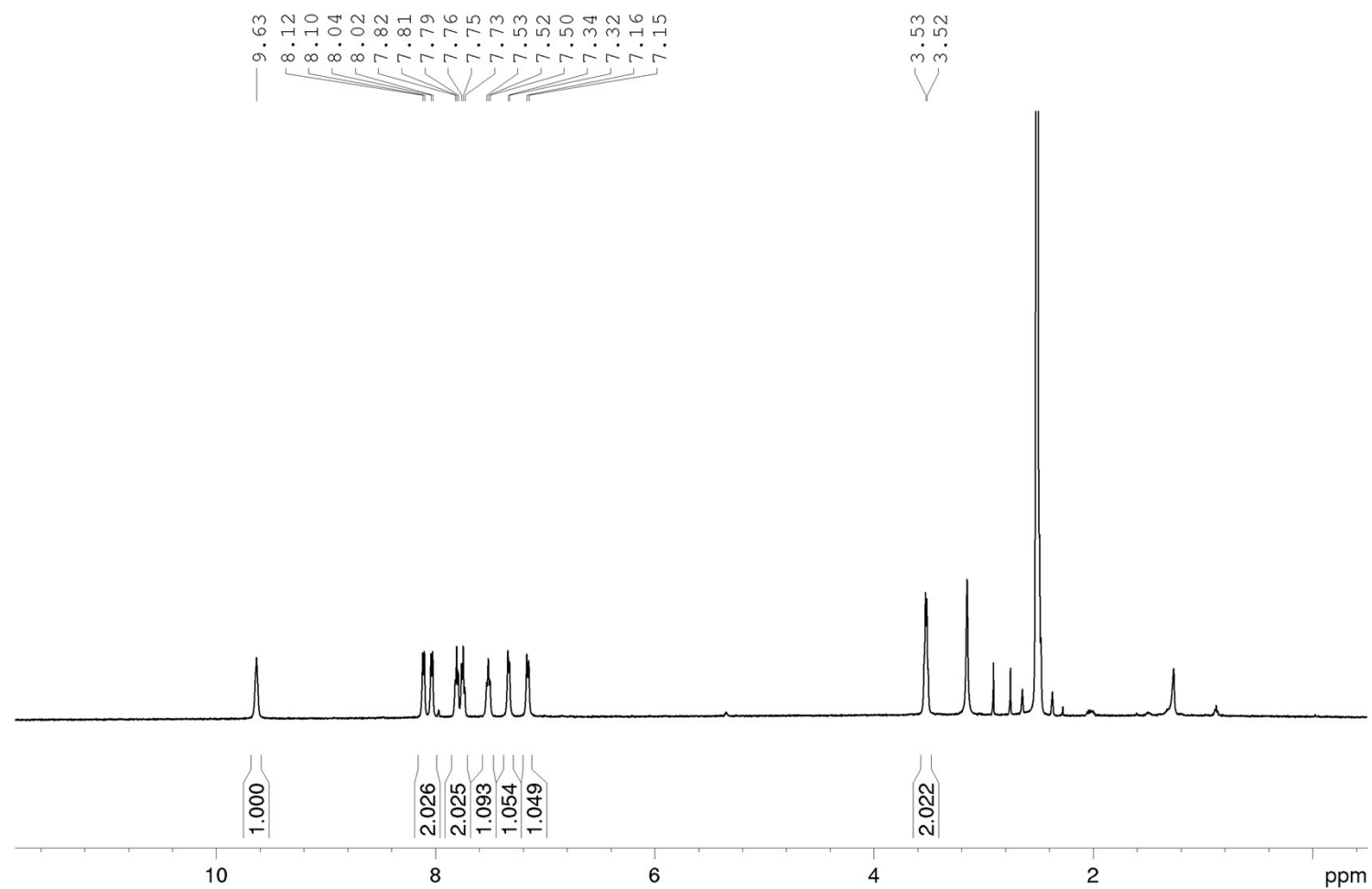


Figure S8. ^1H NMR spectrum of complex **2** in $\text{DMSO-}d_6$ at 333 K (500.13 MHz).

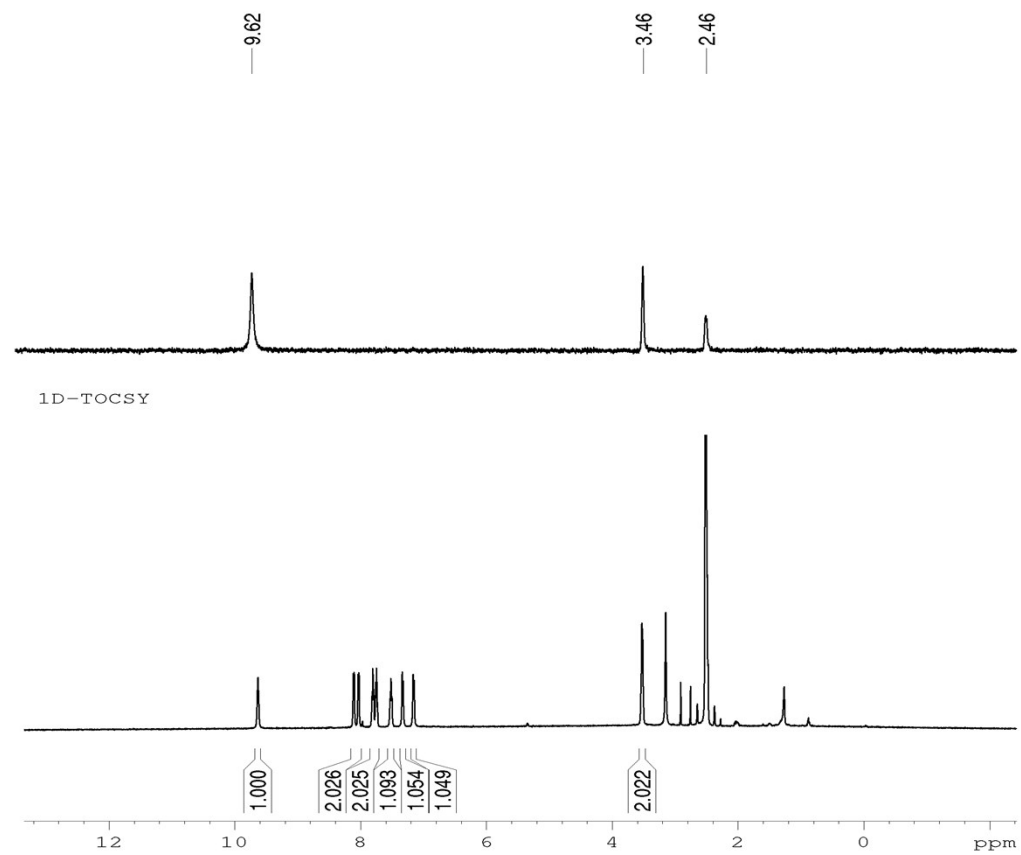


Figure S9. 1D-gTOCSY spectrum of complex **2** in DMSO- d_6 at 333 K.

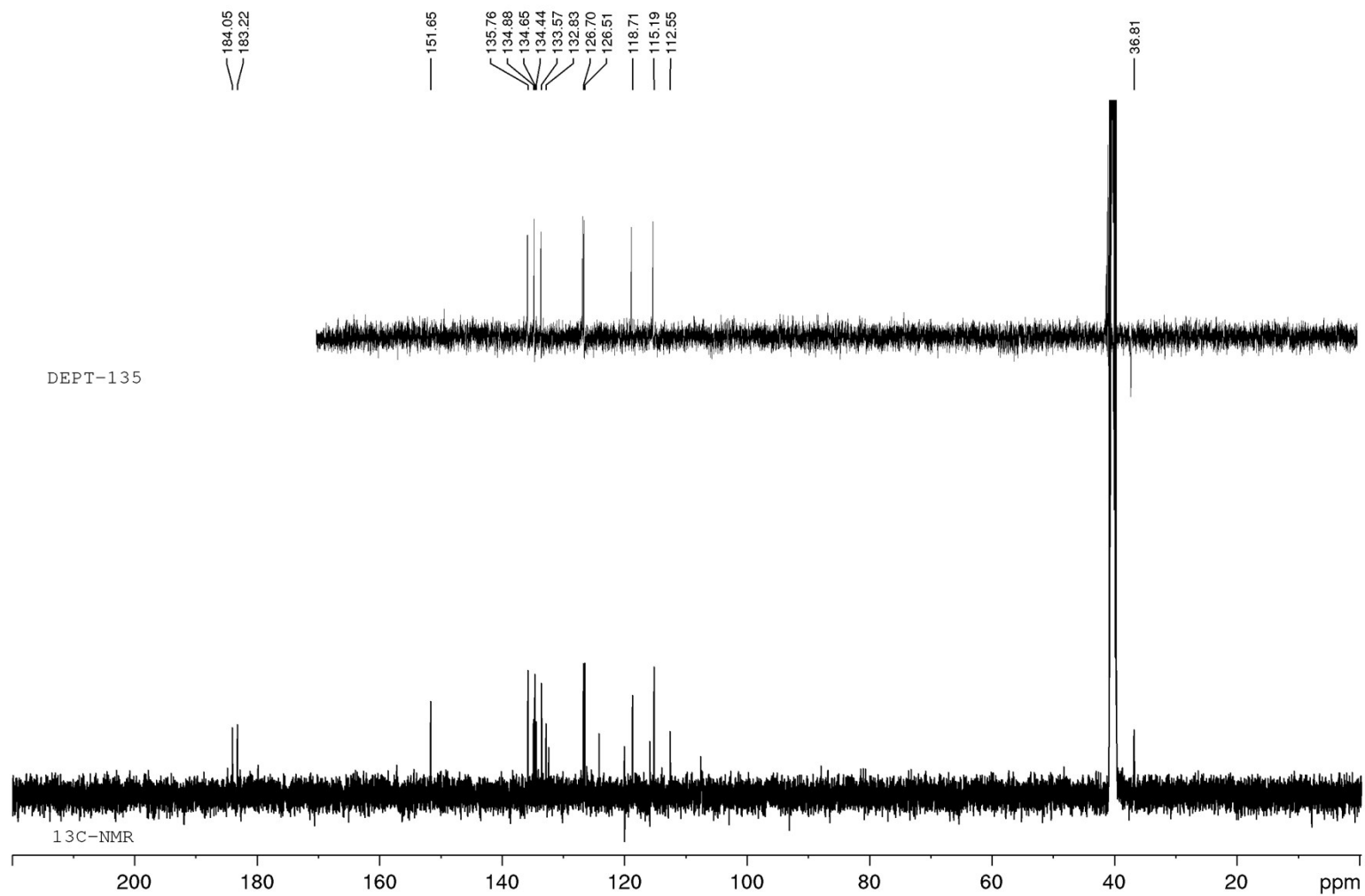


Figure S10. ^{13}C NMR (bottom) and DEPT-135 (top) NMR spectra of complex **2** in $\text{DMSO}-d_6$ at 333 K (126 MHz).

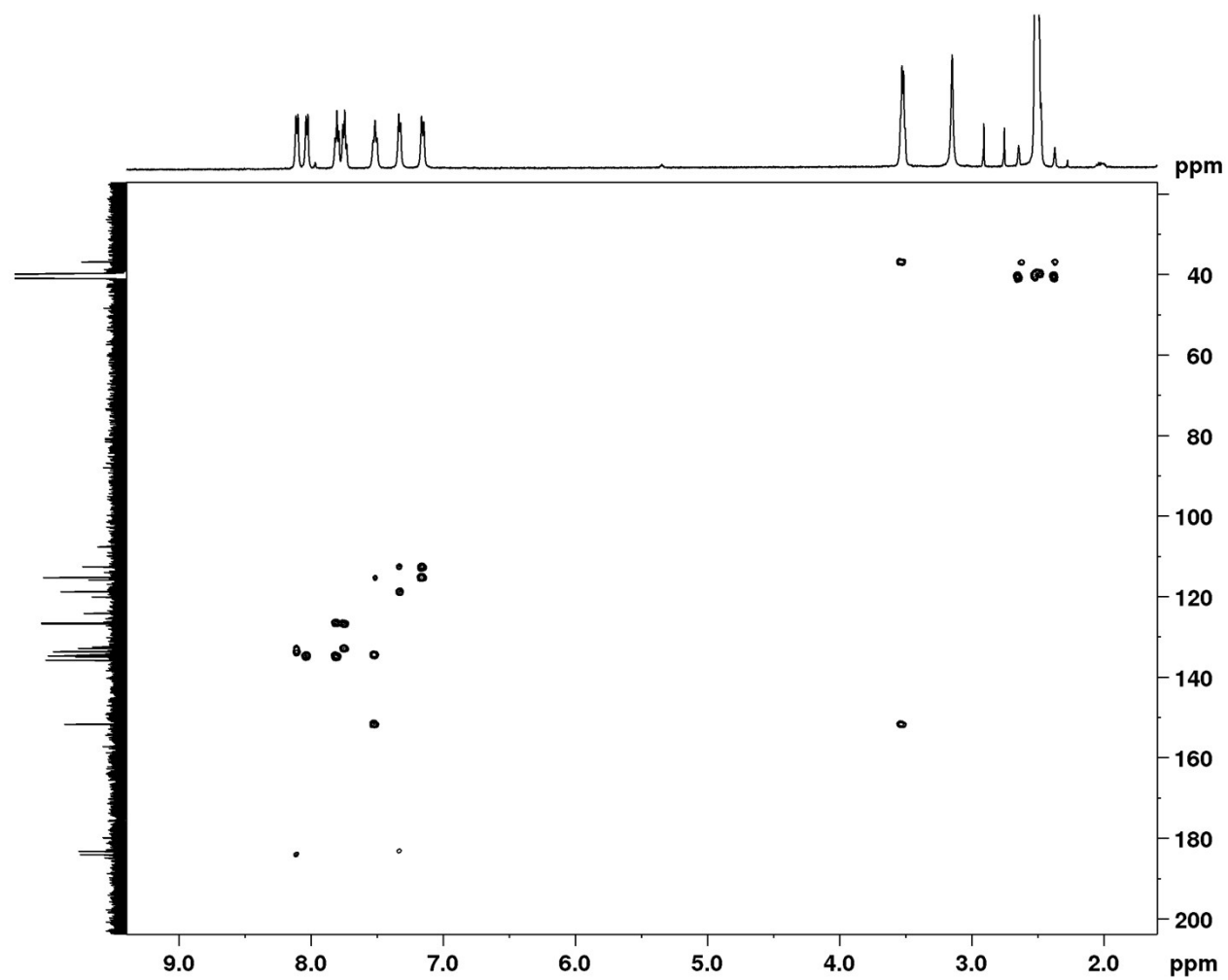


Figure S11. ^1H , ^{13}C gHMBC spectrum of complex **2** in $\text{DMSO}-d_6$ at 333 K.

S12. Diffusion NMR Details studies.

PGSE NMR diffusion measurements were carried out using the stimulated echo sequence and monopolar pairs pulses (stegp).¹ A sine shape was used for the gradient pulses and their strength varied automatically in the course of the experiments. The D values were determined from the slope of the regression line $\ln(I/I_0)$ versus G^2 , according to the following equation, and employing the DiffAtOnce package.² I/I_0 = observed spin echo intensity/intensity without gradients, G = gradient strength, Δ = delay between the midpoints of the gradients, D = diffusion coefficient, δ = gradient length.

$$\ln\left(\frac{I}{I_0}\right) = -(\gamma\delta)^2 \left(\Delta - \frac{\delta}{4}\right) D \frac{4}{\pi^2} G^2$$

The measurements were carried out without spinning. Gradient calibration was carried out by means of a diffusion measurement of HDO in D₂O ($D(\text{HDO}) = 1.902 \cdot 10^{-9} \text{ m}^2 \text{ s}^{-1}$).³ To check reproducibility, three different measurements with different diffusion parameters (δ and/or Δ) were always carried out. The experimental error in D values was estimated to be smaller than $\pm 2\%$. All of the data leading to the reported D values afforded lines whose correlation coefficients were above 0.999 and 8 to 12 points were used for regression analysis. To check reproducibility, three different measurements with different diffusion parameters (δ and/or Δ) were always carried out. The gradient strength was incremented in 4–8% steps from 2–10% to 98% so that, depending on signal: noise, 12–25 points could be used for regression analysis. The recovery delay was set to 5 times T_1 . All diffusion processing and molecular size estimation were performed by using the DiffAtOnce software package available at www.diffatonce.com.

S13. Structural details of compound 8.

Table S1. Structural parameters (Å, °) of π - π interactions of compound 8.^a

Ring...Ring ^b	α	DC	β	DZ	Dist.
1A-1A(v)	0.02	3.582(3)	21.94	3.32	3.17-3.56

[a] Symmetry: (v) $-y - 1, -x + 2, -z + 1/2$; (vi) $-y + 3/2, x + 3/2, z + 3/4$. α : dihedral angle between mean planes of the rings (°), DC: distance between ring centroids (Å), β : angle between DC vector and normal to plane(I) (°), DZ: perpendicular distance of the centroids of ring(I) on plane of ring(II) (Å), Dist.: shorter distances between non-hydrogen atoms of rings (I) and (II). [b] Ring: 1A: C1A, C2A, C7A, C8A, C9A, C14A.

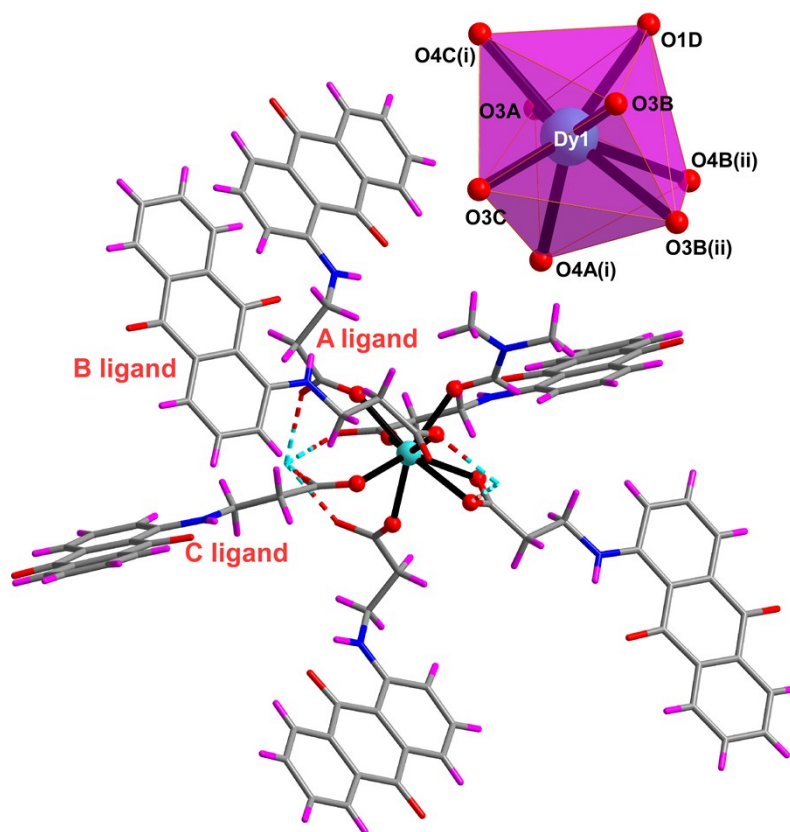


Figure S12. Excerpt of the compound 8 showing the two successive dimeric entities of the 1D chain as well as the coordination sphere.

S14. Continuous shape measures (CShMs)

Table S2. Continuous Shape Measurements for Compound **8**.

OP-8	D8h	Octagon
HPY-8	C7v	Heptagonal pyramid
HBPY-8	D6h	Hexagonal bipyramid
CU-8	Oh	Cube
SAPR-8	D4d	Square antiprism
TDD-8	D2d	Triangular dodecahedron
JGBF-8	D2d	Johnson gyrobifastigium J26
JETBPY-8	D3h	Johnson elongated triangular bipyramid J14
JBTPR-8	C2v	Biaugmented trigonal prism J50
BTPR-8	C2v	Biaugmented trigonal prism
JSD-8	D2d	Snub diphonoid J84
TT-8	Td	Triakis tetrahedron
ETBPY-8	D3h	Elongated trigonal bipyramid

OP-8	HPY-8	HBPY-8	CU-8	SAPR-8	TDD-8	JGBF-8
32.048	21.324	13.305	7.972	1.642	2.261	14.384

JETBPY-	JBTPR-8	BTPR-8	JSD-8	TT-8	ETBPY-
26.907	2.917	1.994	5.428	8.799	22.259

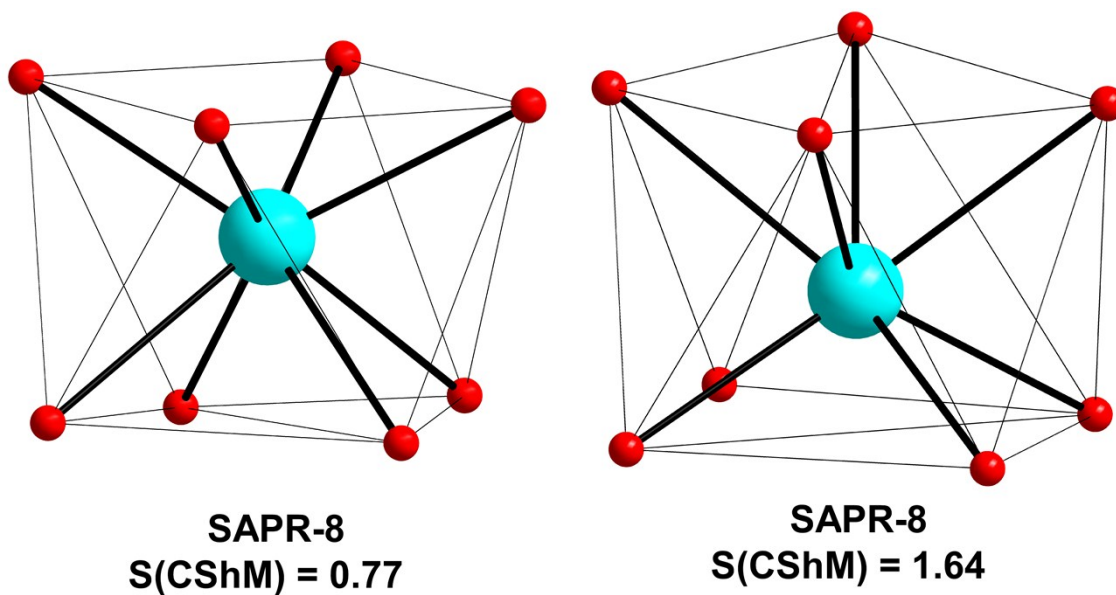


Figure S13. DyO₈ coordination environments for compound **8** and {[Dy((NH₂)₂-
 bdc)_{1.5}(DMF)₂]·DMF·H₂O}_n⁴ with their continuous shape measures.

S15. Powder X-ray Data Collection.

Table S3, S4 and Figure S13 summarize the data resulting from the PXRD pattern-matching analyses performed on compounds **2–9**, which confirm the isostructural nature and purity of all samples.

Table S3. PXRD pattern-matching results of compounds **2–5**.

	2	3	4	5
Empirical formula	C ₅₄ H ₄₃ N ₄ O ₁₃ Y	C ₅₄ H ₄₃ LaN ₄ O ₁₃	C ₅₄ H ₄₃ NdN ₄ O ₁₃	C ₅₄ H ₄₃ EuN ₄ O ₁₃
Formula weight	1044.84	1094.84	1100.18	1107.90
Crystal system	triclinic	triclinic	triclinic	triclinic
Space group	<i>P</i> – <i>1</i>	<i>P</i> – <i>1</i>	<i>P</i> – <i>1</i>	<i>P</i> – <i>1</i>
<i>a</i> (Å)	8.1575(4)	8.154(2)	8.154(2)	8.154(2)
<i>b</i> (Å)	17.4805(8)	8.154(2)	8.154(2)	8.154(2)
<i>c</i> (Å)	17.9640(6)	20.052(4)	20.052(4)	20.052(4)
α (°)	62.669(2)			
β (°)	80.922(3)			
γ (°)	88.786(2)			
V (Å ³)	1260.7(7)	1333.3(6)	1333.3(6)	1333.3(6)
Z	2	4	4	4
Chi ²	1.94	2.94	2.94	2.94

Table S4. PXRD pattern-matching results of compounds **6–9**.

	6	7	8	9
Empirical formula	C ₅₄ H ₄₃ GdN ₄ O ₁₃	C ₅₄ H ₄₃ N ₄ O ₁₃ Tb	C ₅₄ H ₄₃ DyN ₄ O ₁₃	C ₅₄ H ₄₃ ErN ₄ O ₁₃
Formula weight	1113.19	1114.86	1118.44	1123.20
Crystal system	triclinic	triclinic	triclinic	triclinic
Space group	<i>P</i> – <i>1</i>	<i>P</i> – <i>1</i>	<i>P</i> – <i>1</i>	<i>P</i> – <i>1</i>
<i>a</i> (Å)	8.1575(4)	8.154(2)	8.154(2)	8.154(2)
<i>b</i> (Å)	17.4805(8)	8.154(2)	8.154(2)	8.154(2)
<i>c</i> (Å)	17.9640(6)	20.052(4)	20.052(4)	20.052(4)
α (°)	62.669(2)			
β (°)	80.922(3)			
γ (°)	88.786(2)			
V (Å ³)	1260.7(7)	1333.3(6)	1333.3(6)	1333.3(6)
Z	2	4	4	4
Chi ²	1.94	2.94	2.94	2.94

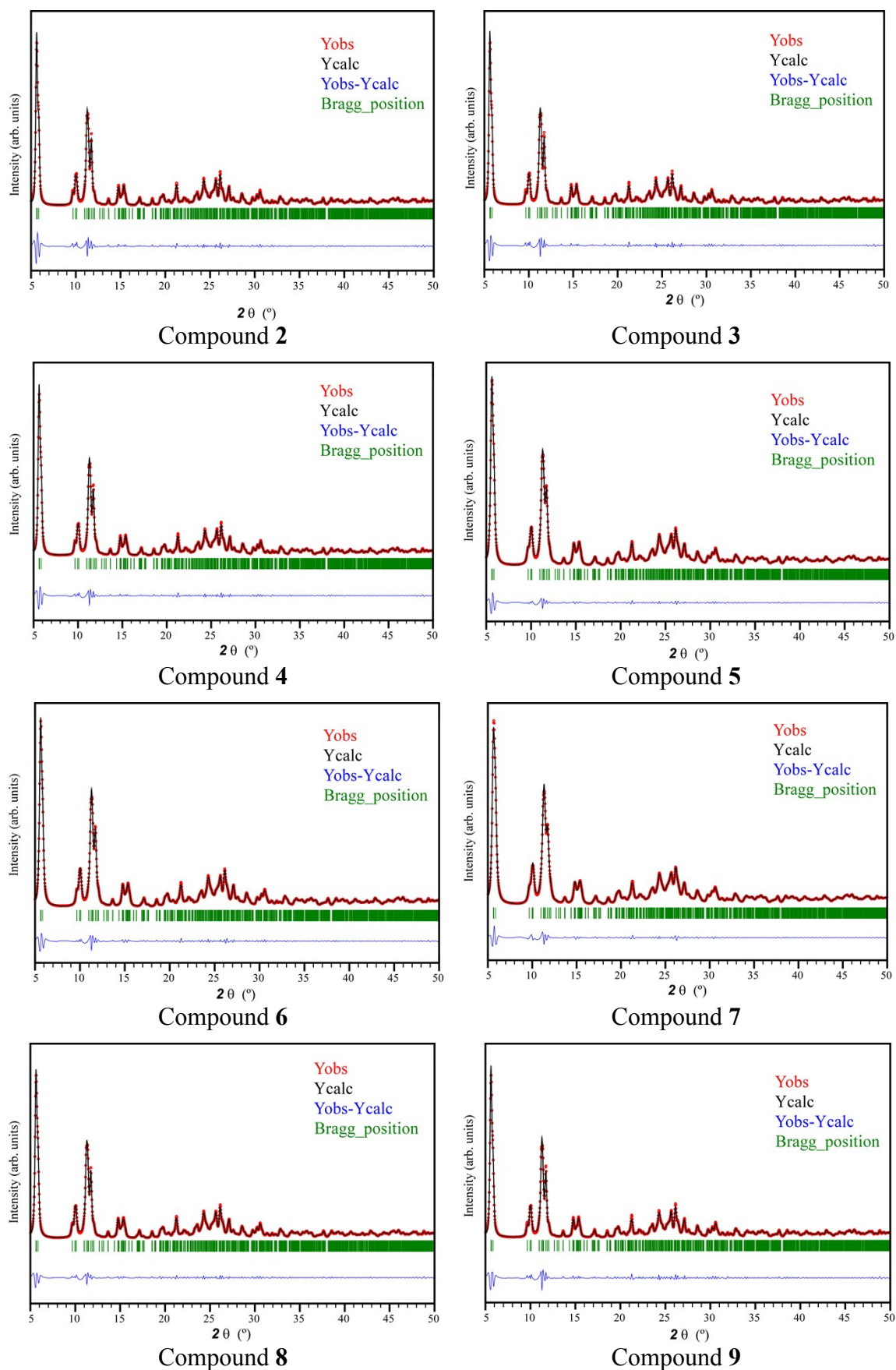


Figure S14. Comparison of experimental diffractogram and the theoretical one with their full profile pattern-matching analyses.

S16. Photoluminescence measurements.

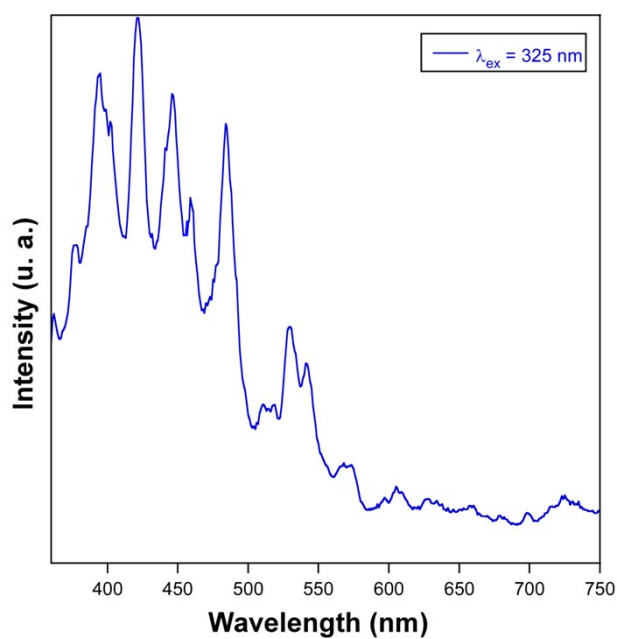


Figure S15. Room temperature emission spectra of compound **6** under excitation at 325 nm.

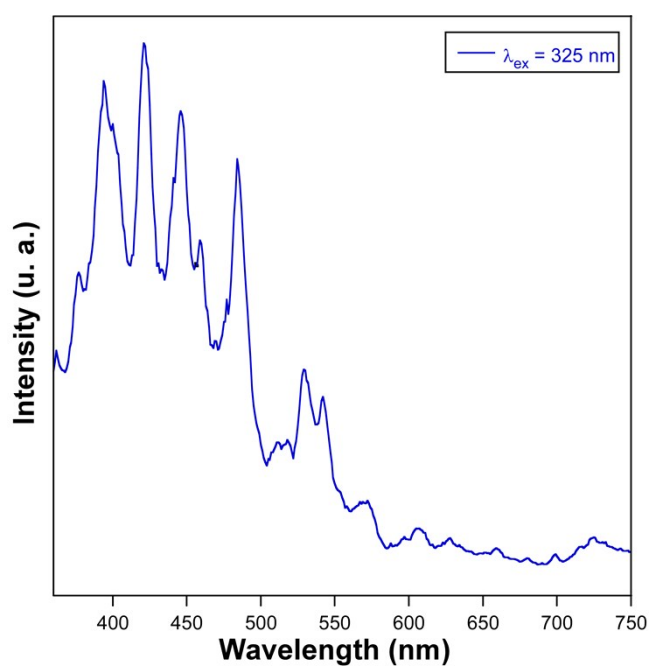


Figure S16. Room temperature emission spectra of compound **5** under excitation at 325 nm.

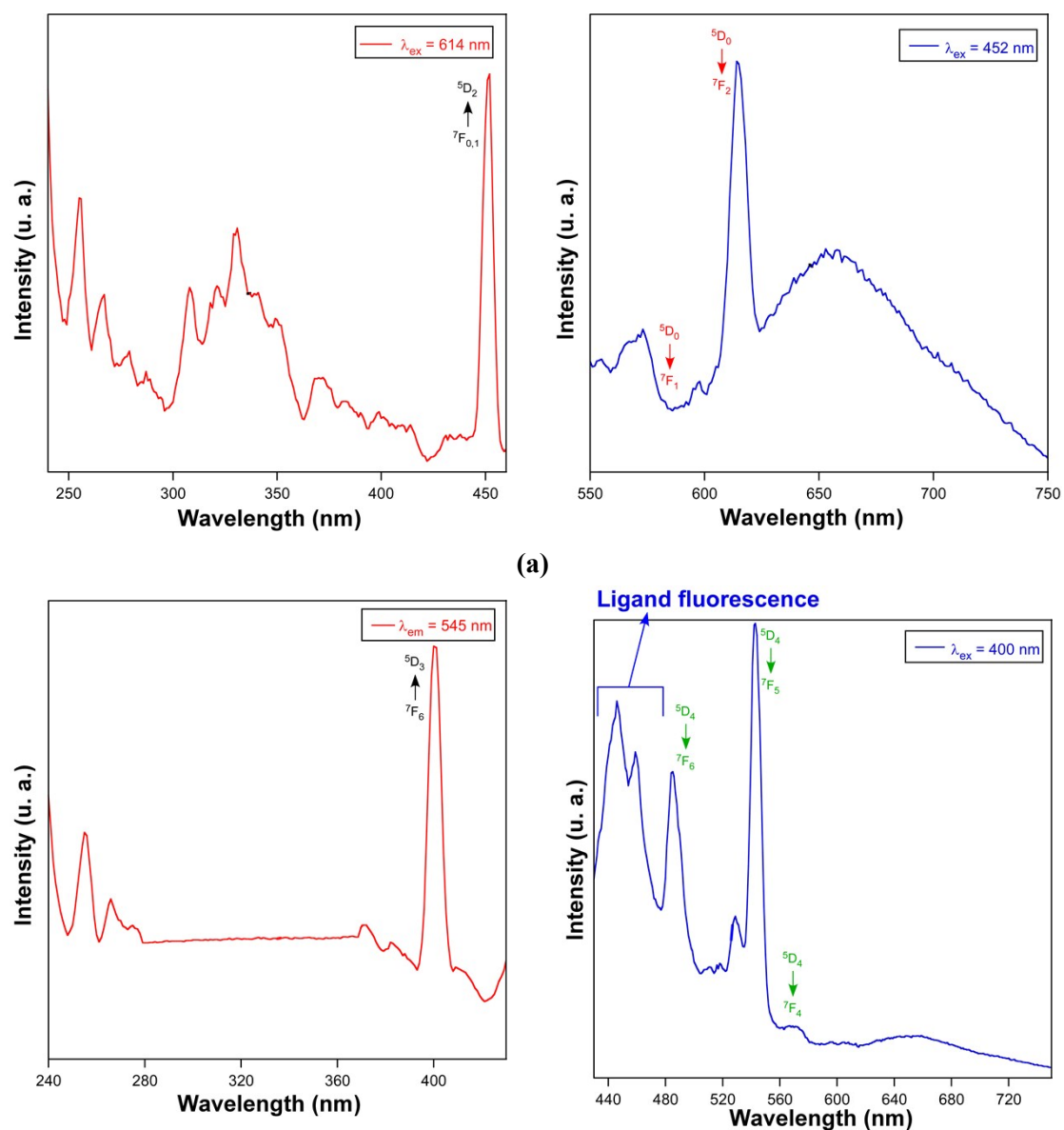


Figure S17. Room temperature excitation (red line) and emission (blue line) spectra of compounds (a) 5 and (b) 7.

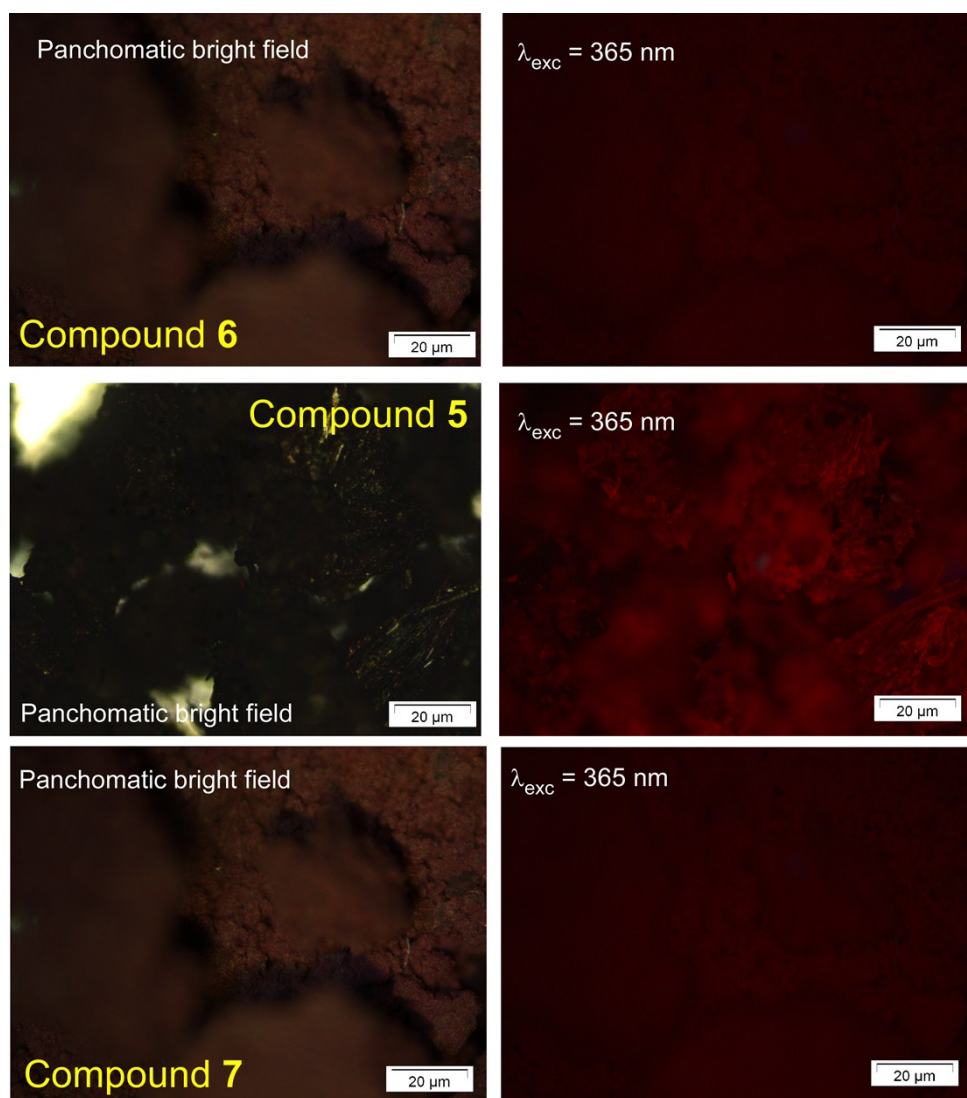


Figure S18. Micro-PL images taken on crystals or polycrystalline samples of compounds 5–7 under panchromatic field and UV irradiation ($\lambda_{\text{exc}} = 365 \text{ nm}$).

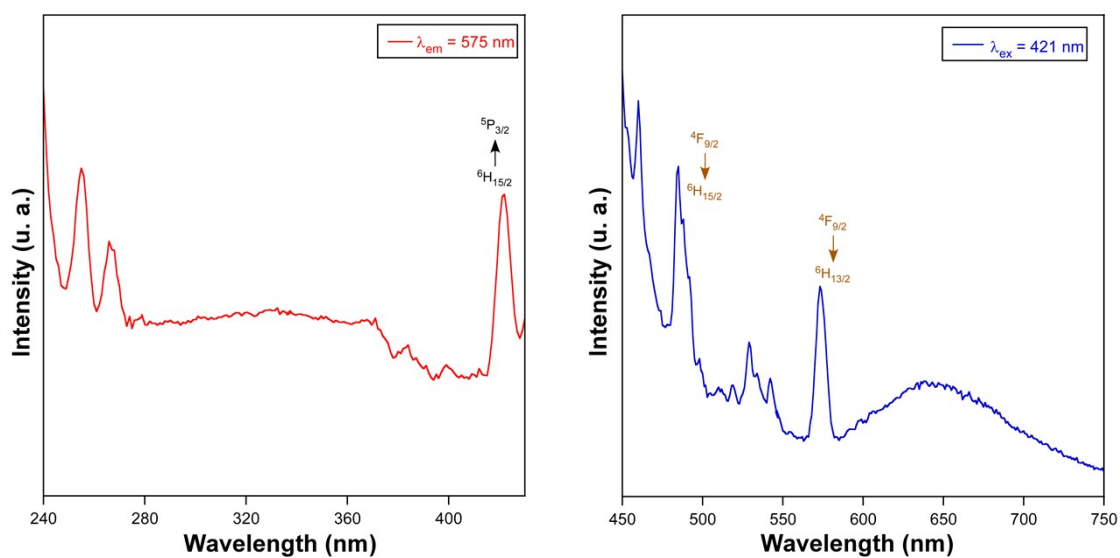


Figure S19. Room temperature excitation (red) and emission (blue) spectra of compound 8.

S17. Ac susceptibility measurements.

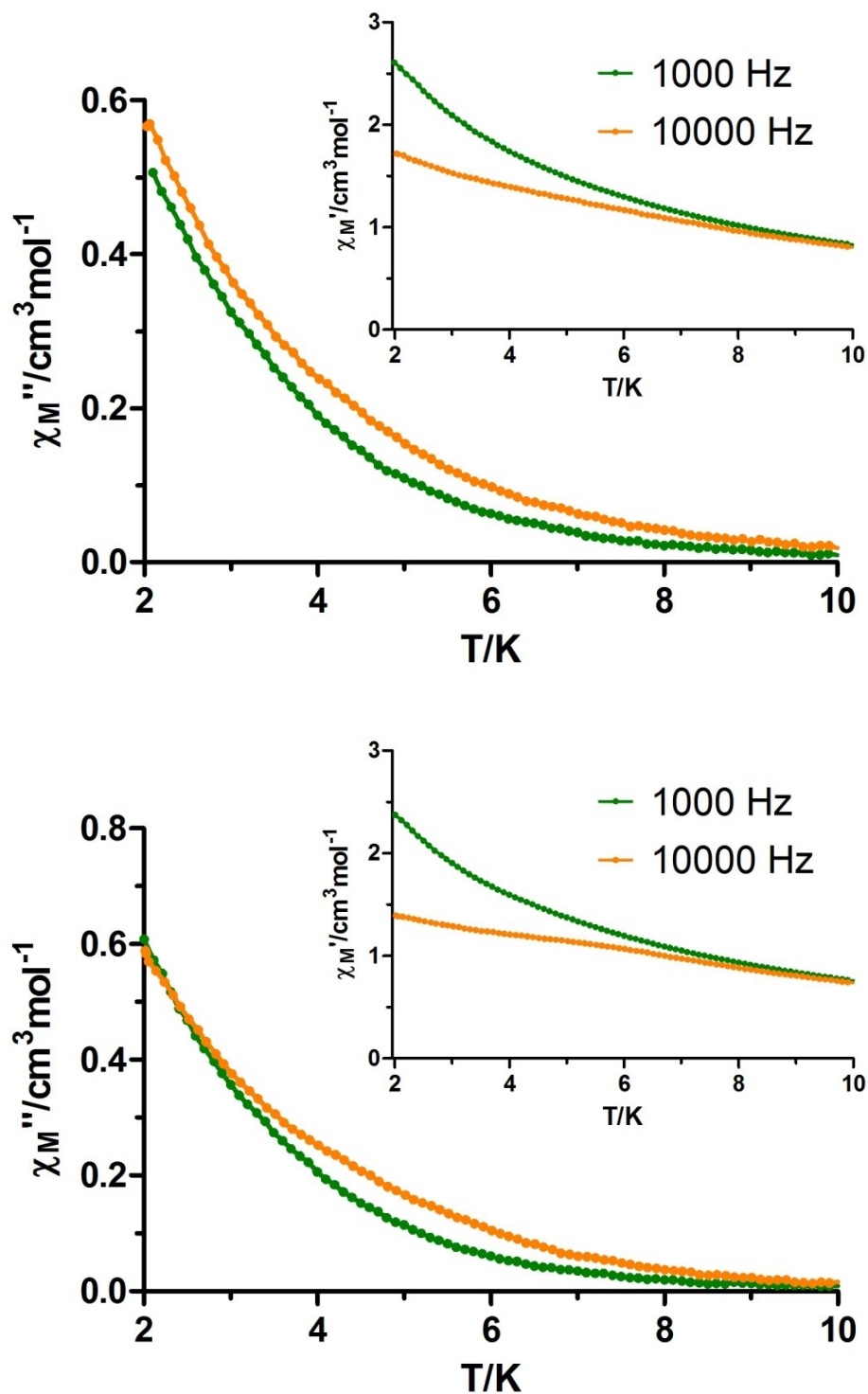


Figure S20. Temperature dependence of out-of-phase components of the *ac* susceptibility under 1000 Oe *dc* field for the diluted complexes (Dy:Y of 1:3) (top) and (Dy:Y of 1:6) (bottom). Inset: temperature dependence of in-phase components of the *ac* susceptibility under 1000 Oe *dc* field for the diluted complexes (Dy:Y of 1:3) (top) and (Dy:Y of 1:6) (bottom).

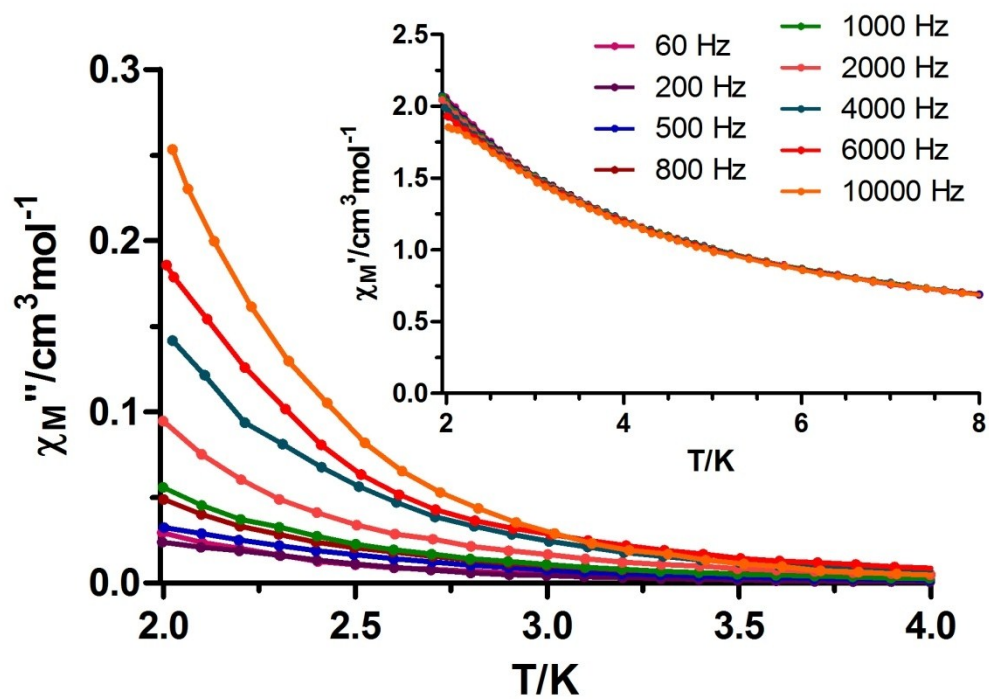


Figure S21. Temperature dependence of in-phase (inset) and out-of-phase components of the *ac* susceptibility under 1000 Oe *dc* field for the Er(III) complex.

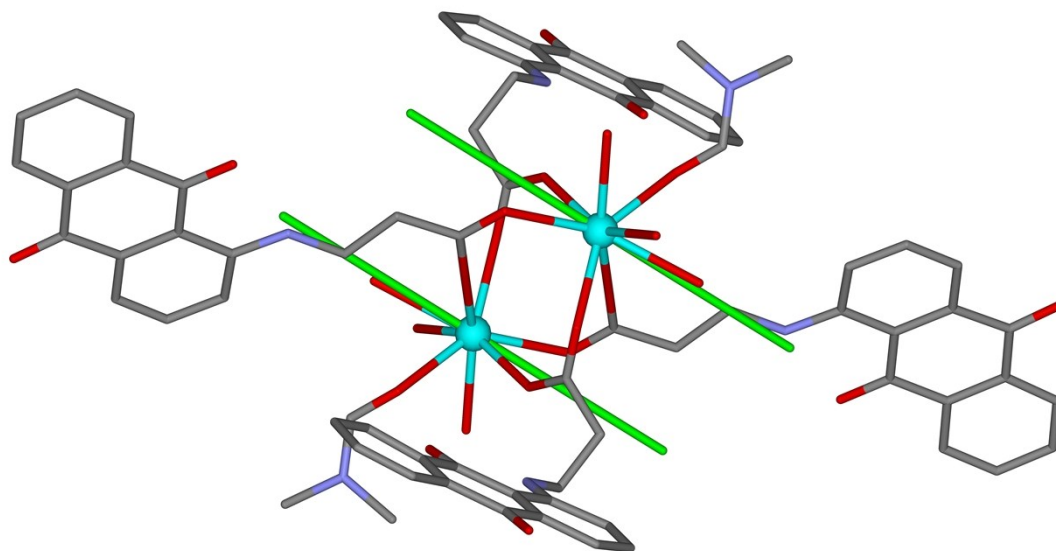


Figure S22. Orientation of the magnetic moment (Green line) for compound 8.

S18. Magneto-caloric effect for gadolinium compound.

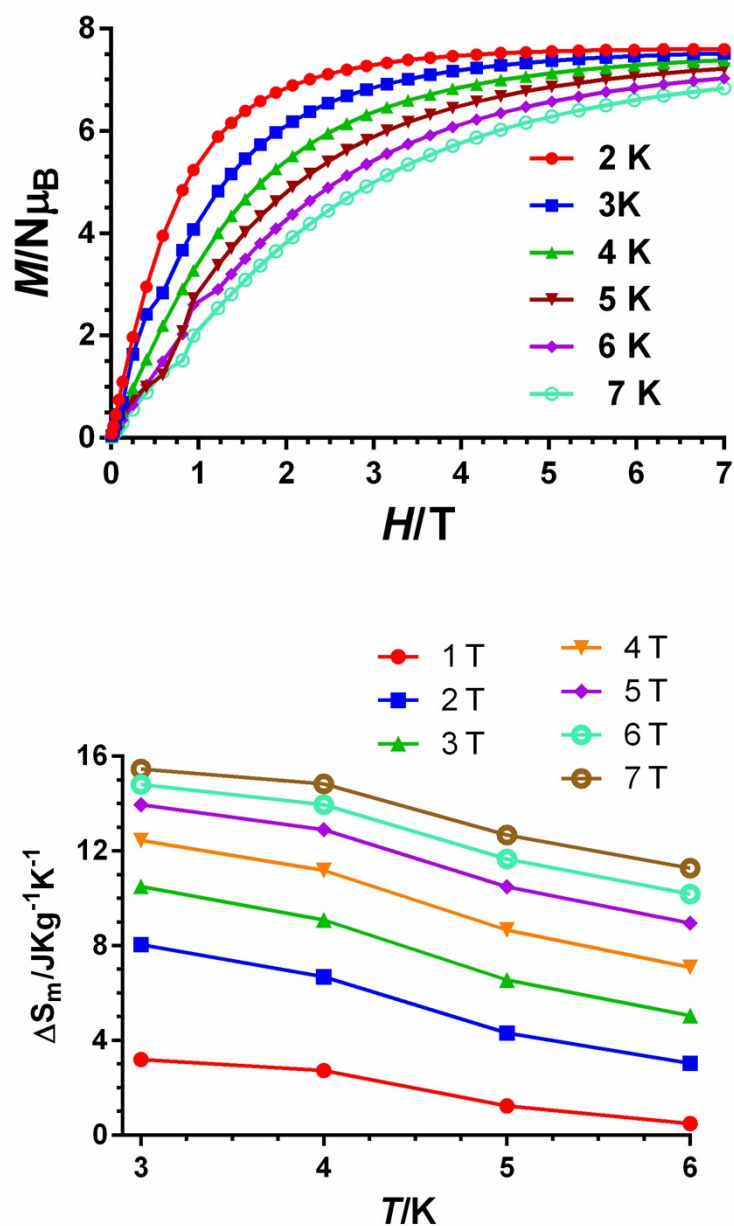


Figure S23.- (Top) The field dependence of the magnetization plots for **1** between 2 and 7 K and (bottom) magnetic entropy changes ($-\Delta S_m$) calculated from the experimental magnetization data from 1 to 7 T and temperatures from 3 to 6 K (points). Solid lines are only a guide for the eye.

REFERENCES

- 1 D. H. Wu; A. D. Chen; C. S. Johnson, *J. Magn. Reson. Ser., A* **1995**, *115*, 260.
- 2 DiffAtOnce® is a registered program developed by I. Fernández and F. M. Arrabal-Campos in the University of Almería (Spain). 2013.
- 3 H. J. V. Tyrrell; K. R. Harris, *Diffusion in Liquids*, Butterworths, London, 1984.
- 4 I. Oyarzabal, B. Fernández, J. Cepeda, S. Gómez-Ruiz, A. J. Calahorro, J. M. Seco, A. Rodríguez-Diéguez, *CrystEngComm*, 2016, **18**, 3055.