

Supporting Information

First Investigation of Europium Silicate Melilite for Second-order Nonlinear Optical Application: Experimental and Theoretical Studies

Yang Chi, Huai-Guo Xue and Sheng-Ping Guo*

School of Chemistry & Chemical Engineering, Yangzhou University, Yangzhou, Jiangsu

225002, P. R. China

Corresponding author: spguo@yzu.edu.cn

Table S1. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters (U_{eq}^a , $\text{\AA}^2 \times 10^3$) for **1**.

atom	x	y	z	$U_{\text{eq}}/\text{\AA}^2$
Eu(1)	6649.5(5)	1649.5(5)	5084.2(11)	9.9(3)
Mg(1)	10000	0	0	9.1(10)
Si(1)	8618(3)	3618(3)	-540(7)	7.7(6)
O(1)	5000	0	1570(20)	10(2)
O(2)	8608(8)	3608(8)	2547(19)	12.2(17)
O(3)	9209(8)	1902(9)	-1951(12)	12.0(13)

^a U_{eq} is defined as one third of the trace of the orthogonalized U_{ij} tensor.

Table S2. Important bond lengths (Å) for **1**.

Bond	Dist./Å	Bond	Dist./Å
Eu(1)–O(1)	2.610(8)	Mg(1)–O(3)	1.938(7)
Eu(1)–O(2)#4	2.740(7)	Si(1)–O(1)#11	1.657(5)
Eu(1)–O(2)	2.583(10)	Si(1)–O(2)	1.596(11)
Eu(1)–O(3)#1	2.572(7)	Si(1)–O(3)	1.630(7)
Eu(1)–O(3)#3	2.797(6)		

Symmetry transformations used to generate equivalent atoms: #1 +x, +y, 1+z; #2 +y, 1-x, -z; #3 1-y, -1+x, -z; #4 1-y, -1+x, 1-z. #5 +y, 1-x, 1-z; #6 1/2+y, -1/2+x, 1+z; #7 -1/2+x, 1/2-y, -z; #8 +x, +y, -1+z; #9 1+y, 1-x, 1-z; #10 2-x, -y, -1+z; #11 1+y, 1-x, -z;

Table S3. Calculated directions and magnitudes of dipole moments for structural units in **1**.

Unit	Vector Analysis of Dipole Moments				
	x	y	z	Magnitude (Debye)	esu·cm/Å ³ (× 10 ⁴)
EuO ₈	21.4292	-8.4201	-3.1783	23.2425	0.1394
MgO ₄	-102.81427	50.6032	100.4658	152.3969	0.9139
SiO ₄	-65.5268	-65.5268	119.6869	151.3686	0.9077

Table S4. Calculated valence state of ions in **1** by bond valence model.

Atom	BVS	Atom	BVS
Eu(1)	1.98	O1	-2.40
Mg(1)	2.06	O2	-1.78
Si(1)	3.96	O3	-1.99

Table S5. The SHG intensities of silicates.

Compound	SHG intensity	Compound	SHG intensity
$\text{Sr}_2\text{ZnSi}_2\text{O}_7$ ¹	$\sim 35 \times \alpha\text{-SiO}_2$ ($1.75 \times \text{KDP}$)	$\text{Li}_2\text{K}_4[(\text{TiO})\text{Si}_4\text{O}_{12}]^2$	$\sim 4.5 \times \text{KDP}$
$\text{Sr}_2\text{MgSi}_2\text{O}_7$ ¹	$\sim 5 \times \alpha\text{-SiO}_2$ ($0.25 \times \text{KDP}$)	$\text{Li}_2\text{Rb}_4[(\text{TiO})\text{Si}_4\text{O}_{12}]^2$	$\sim 4.5 \times \text{KDP}$
$\text{Rb}_2\text{ZnSi}_3\text{O}_8$ ³	$\sim 2.5 \times \alpha\text{-SiO}_2$ ($0.13 \times \text{KDP}$)	$\text{Ba}_4(\text{BO}_3)_3(\text{SiO}_4) \cdot \text{Ba}_3\text{Cl}^4$	$\sim 1 \times \text{KDP}$
$\text{Ba}_2\text{TiOSi}_2\text{O}_7$ ⁵	$\sim 5.5 \times \text{KDP}$	$\text{Ba}_4(\text{BO}_3)_3(\text{SiO}_4) \cdot \text{Ba}_3\text{Br}^4$	$\sim 1 \times \text{KDP}$,
$\text{Sr}_2\text{TiOSi}_2\text{O}_7$ ⁶	$\sim 8 \times \text{KDP}$	$\text{Eu}_2\text{MgSi}_2\text{O}_7$	$\sim 0.7 \times \text{KTP}$ ($10.5 \times \text{KDP}$)
$\text{Cs}_2\text{B}_4\text{SiO}_9$ ⁷	$\sim 4.6 \times \text{KDP}$		

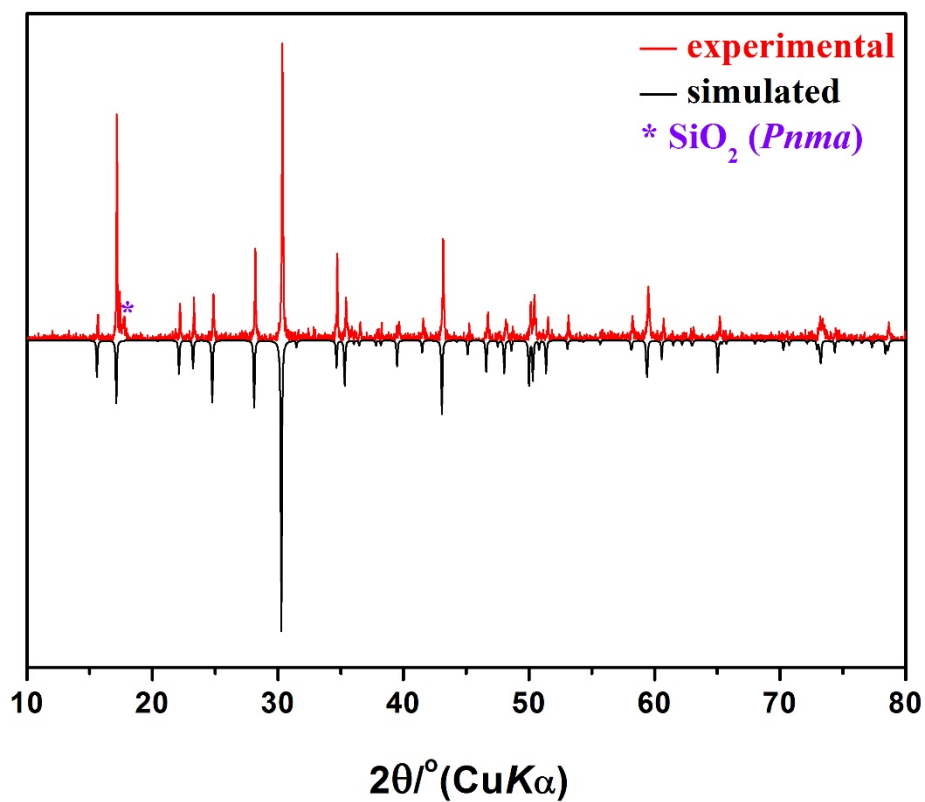


Fig. S1 The X-ray powder diffraction pattern of **1**.

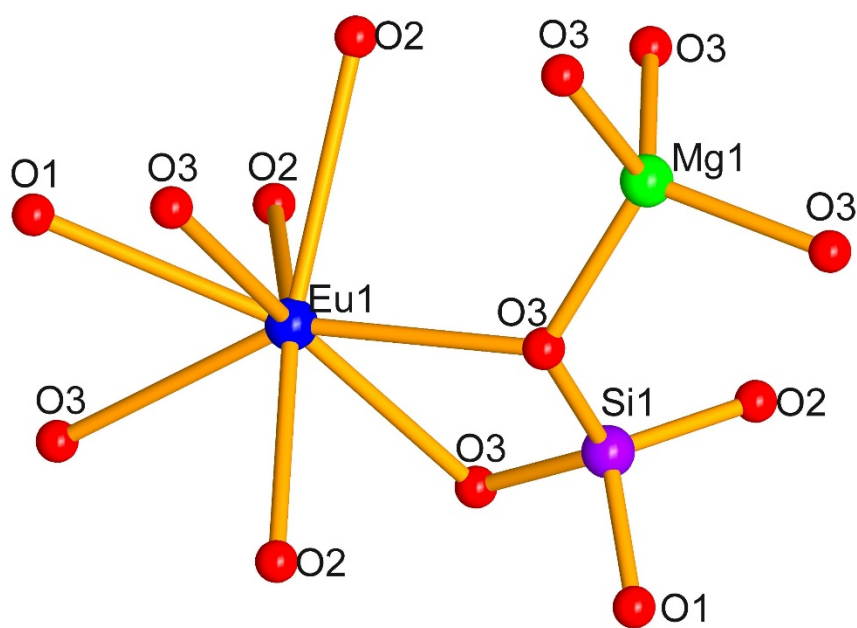
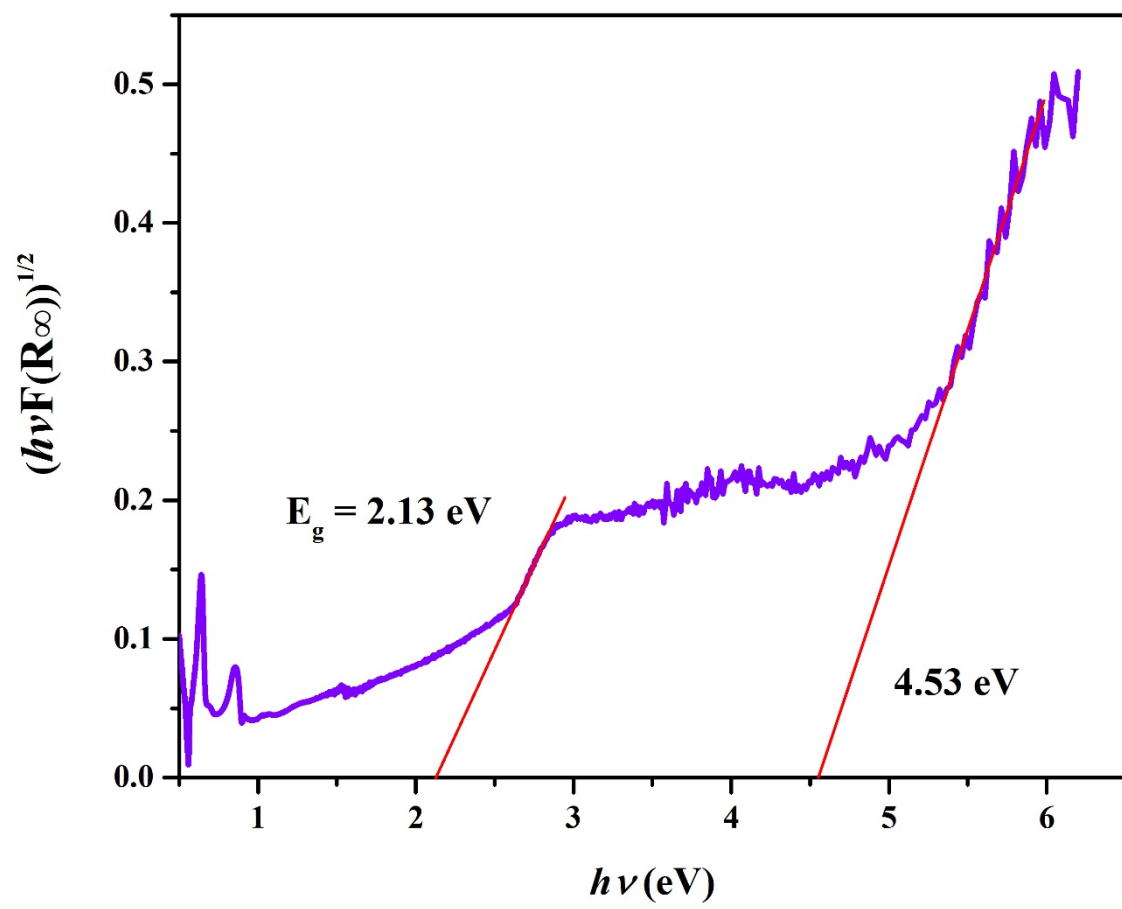
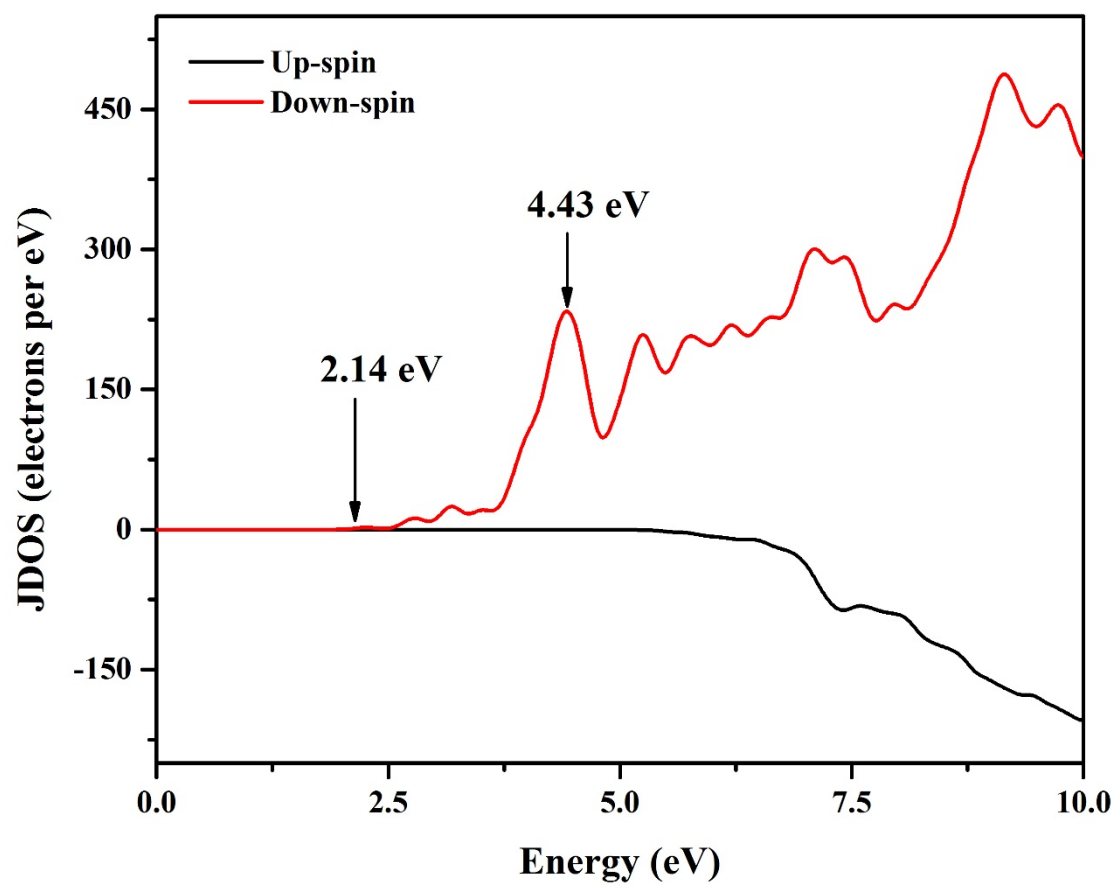


Fig. S2 Coordination geometry of **1**.



(a)



(b)

Fig. S3 (a) Tauc plot of **1**. (b) The joint density of states (JDOS) from DFT for **1**.

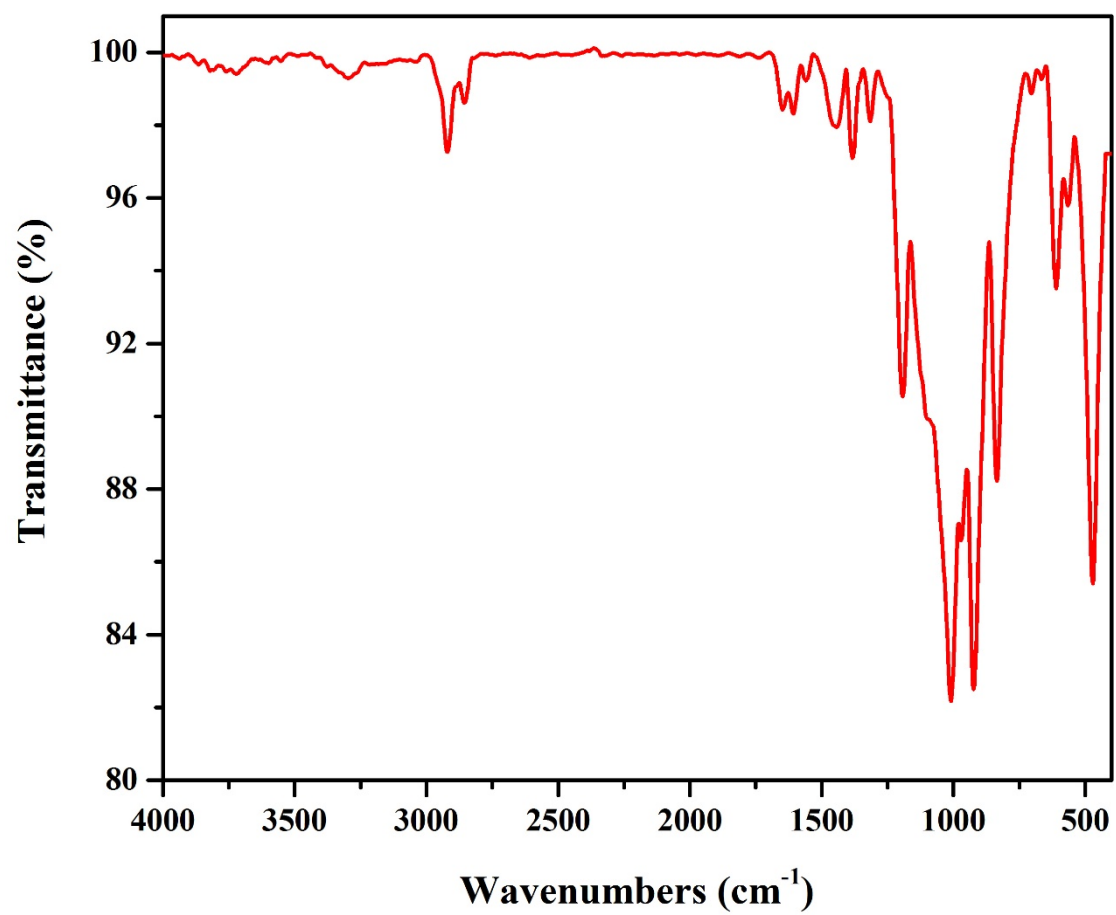
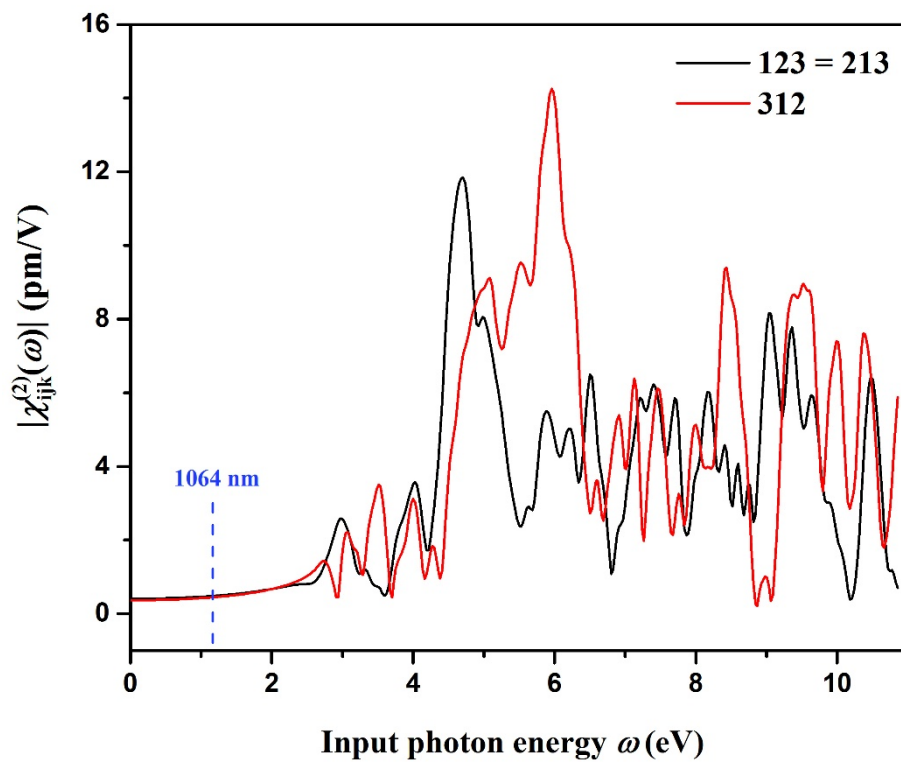
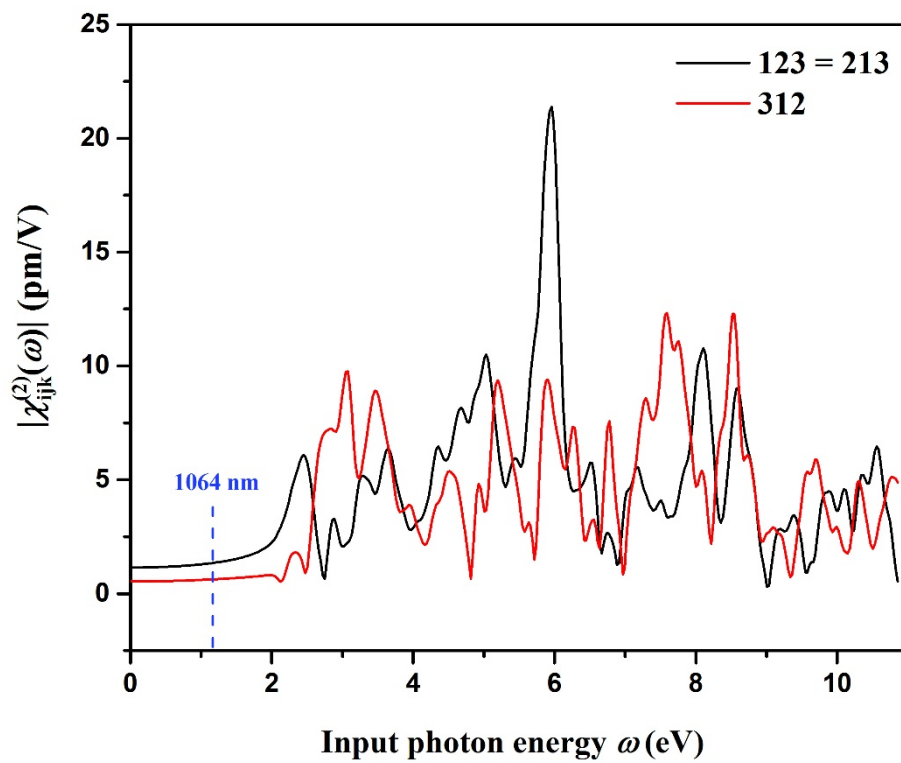


Fig. S4 FT-IR of **1**.



(a)



(b)

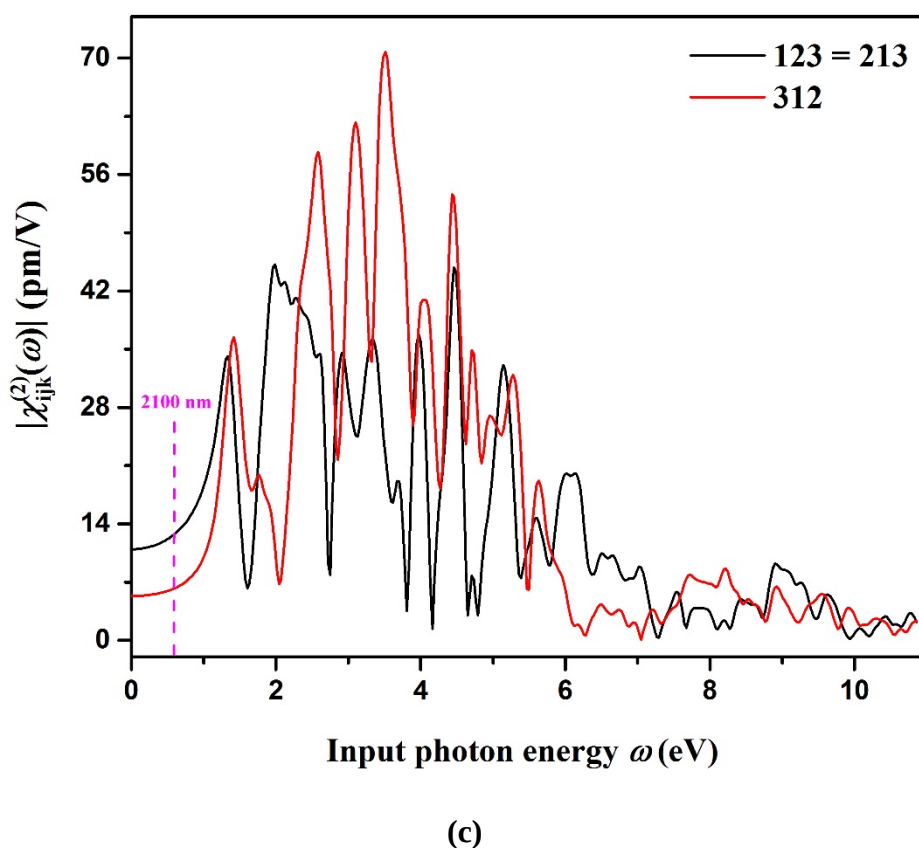


Fig. S5 Calculated absolute values of frequency dependence second-order susceptibilities $|\chi_{ijk}^{(2)}(\omega)|$ for $\text{Sr}_2\text{MgSi}_2\text{O}_7$ (a), $\text{Sr}_2\text{MnSi}_2\text{O}_7$ (b), and **1** (c).

References

- 1 M. Mutailipu, Z. Li, M. Zhang, D. W. Hou, Z. H. Yang, B. B. Zhang, H. P. Wu, and S. L. Pan, *Phys. Chem. Chem. Phys.*, 2016, **18**, 32931–32936.
- 2 T.-L. Chao, W. J. Chang, S.-H. Wen, Y. Q. Lin, B.-C. Chang, and K.-H. Lii., *J. Am. Chem. Soc.*, 2016, **138**, 9061.
- 3 B. Q. Zhao, Y. Yang, S. G. Zhao, Y. G. Shen, X. F. Li, L. N. Li, C. M. Ji, Z. S. Lin, and J. H. Luo, *J. Mater. Chem. C*, 2017, **5**, 11025–11029.

-
- 4 X. X. Lin, F. F. Zhang, S. L. Pan, H. W. Yu, F. Y. Zhang, X. Y. Dong, S. J. Han, L. Y. Dong, C. Y. Bai, and Z. Wang, *J. Mater. Chem. C*, 2014, **2**, 4257–4264.
- 5 W. W. Zhao, F. Y. Zhang, J. Liu, B. Hao, S. L. Pan, F. F. Zhang, and L. Liu, *J. Cryst. Growth*, 2015, **413**, 46.
- 6 J. K. Yuan, P. Z. Fu, J. X. Wang, F. Guo, Z. P. Yang, and Y. C. Wu, *Prog. Crystal Growth and Charact.*, 2000, **40**, 103.
- 7 H. Wu, H. Yu, S. L. Pan, Z. Huang, Z. Yang, X. Su, and K. R. Poeppelmeier, *Angew. Chem. Int. Ed.*, 2013, **52**, 3406–3410.