

Supporting Information

Stannous oxide promoted charge separation in rational designed heterojunction photocatalysts with controllable mechanism

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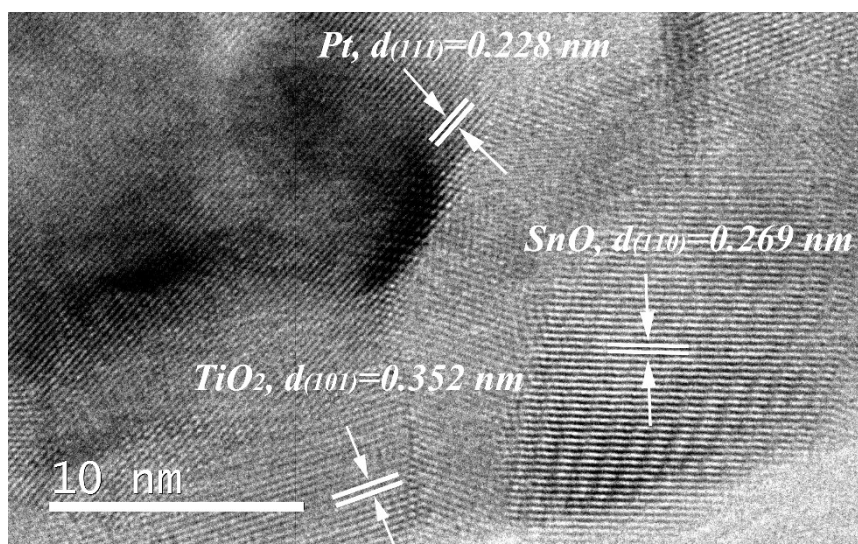


Fig. S1 HRTEM image of SnO/TiO₂ with Pt.

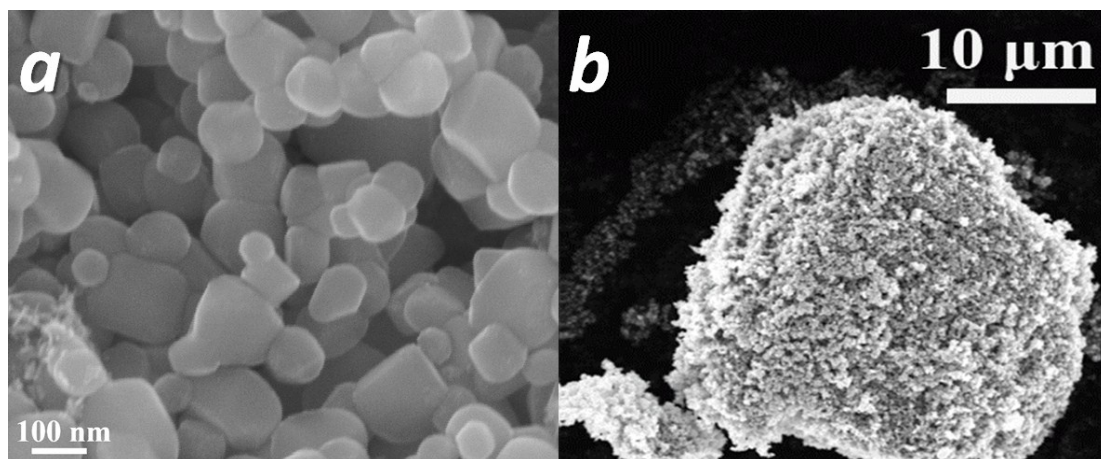


Fig. S2 SEM of a) TiO₂ nanoparticles and b) SnO/TiO₂ heterojunction.

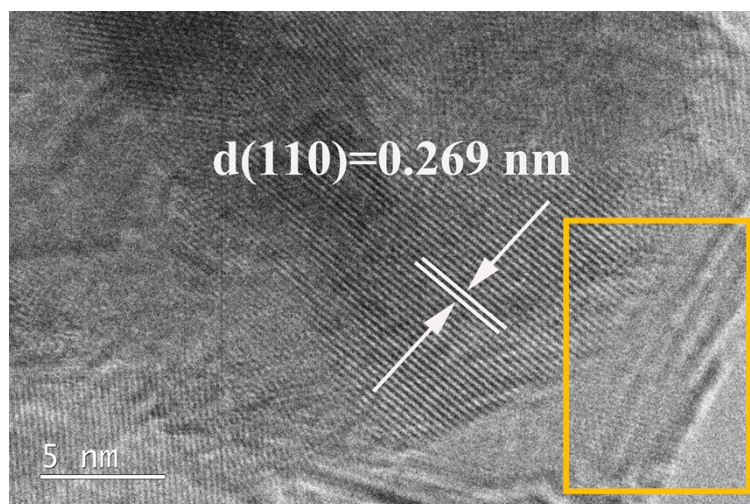


Fig. S3 HRTEM of as-prepared SnO.

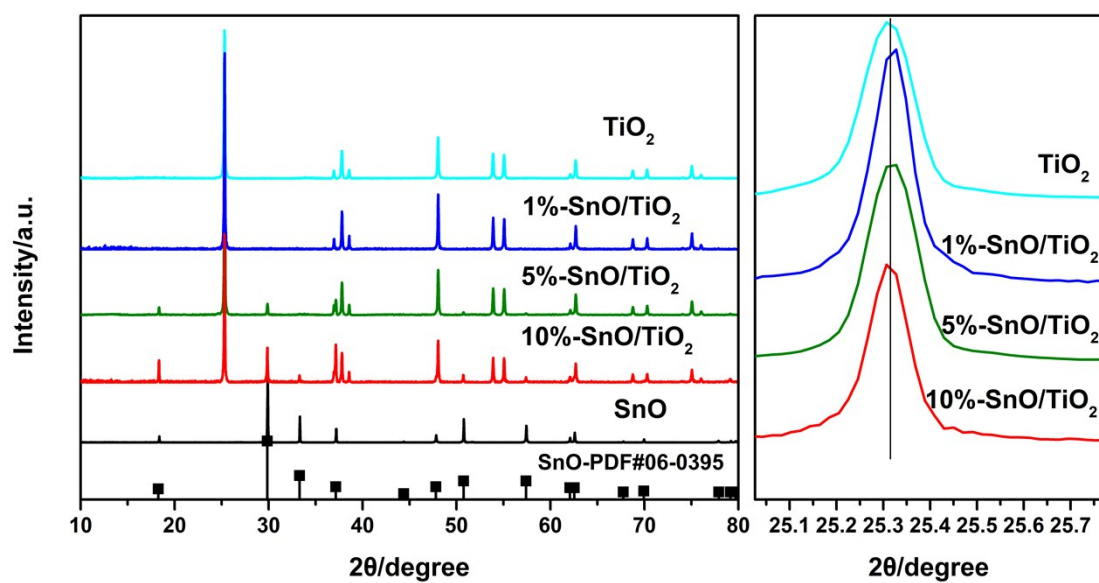


Fig. S4 XRD patterns of all samples.



Fig. S5 Optical Microscope images of all samples.

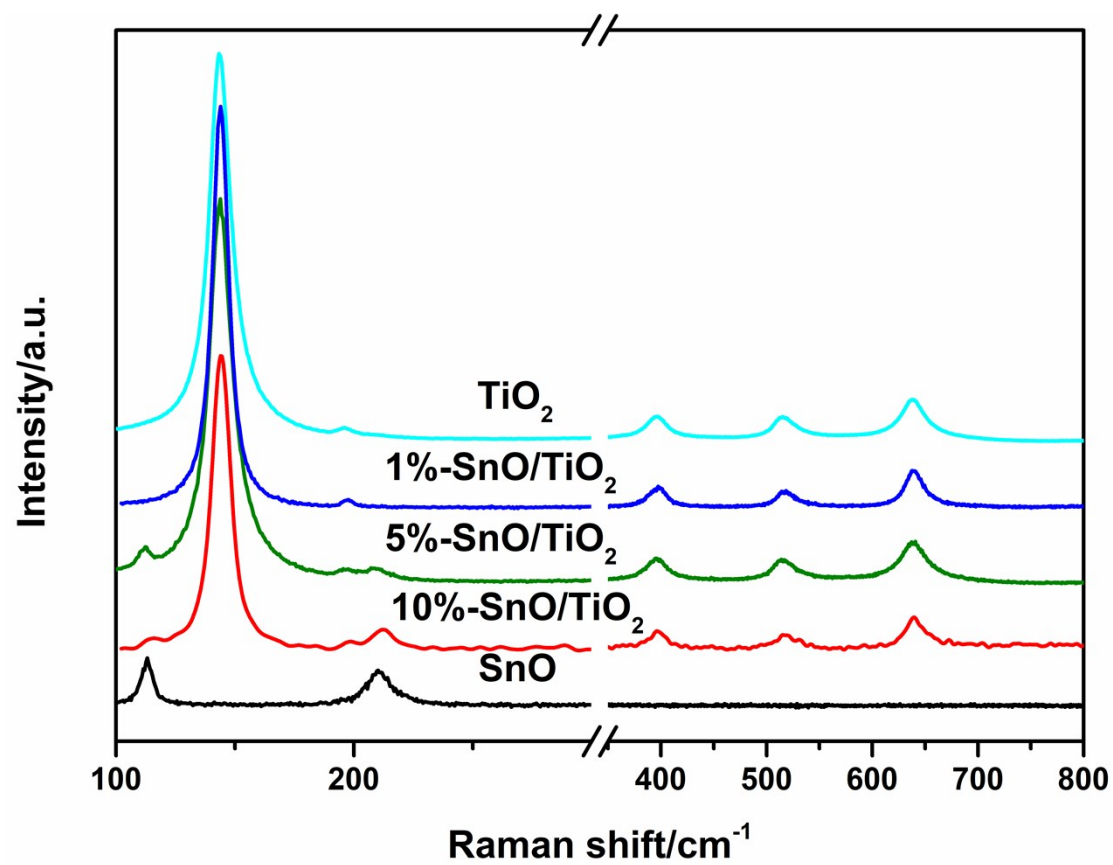


Fig. S6 Raman spectra of all samples.

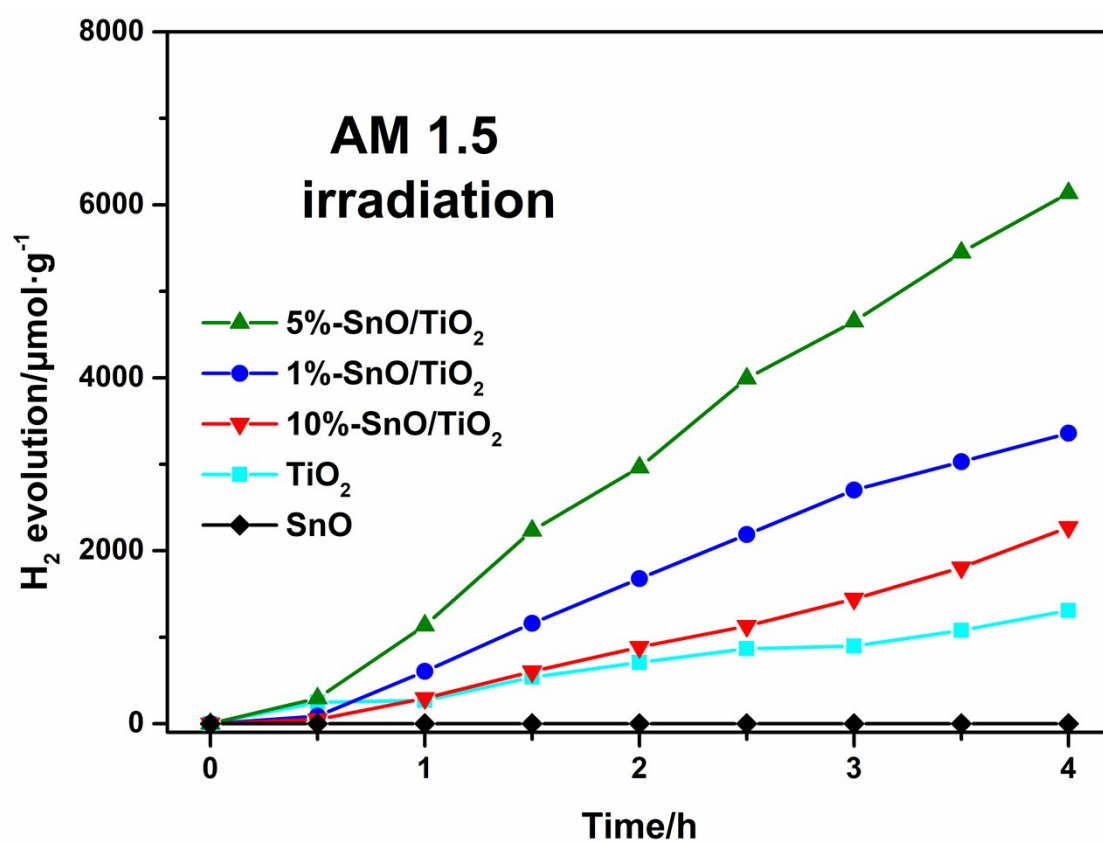


Fig. S7 H₂ evolution of all samples under AM 1.5 irradiation.

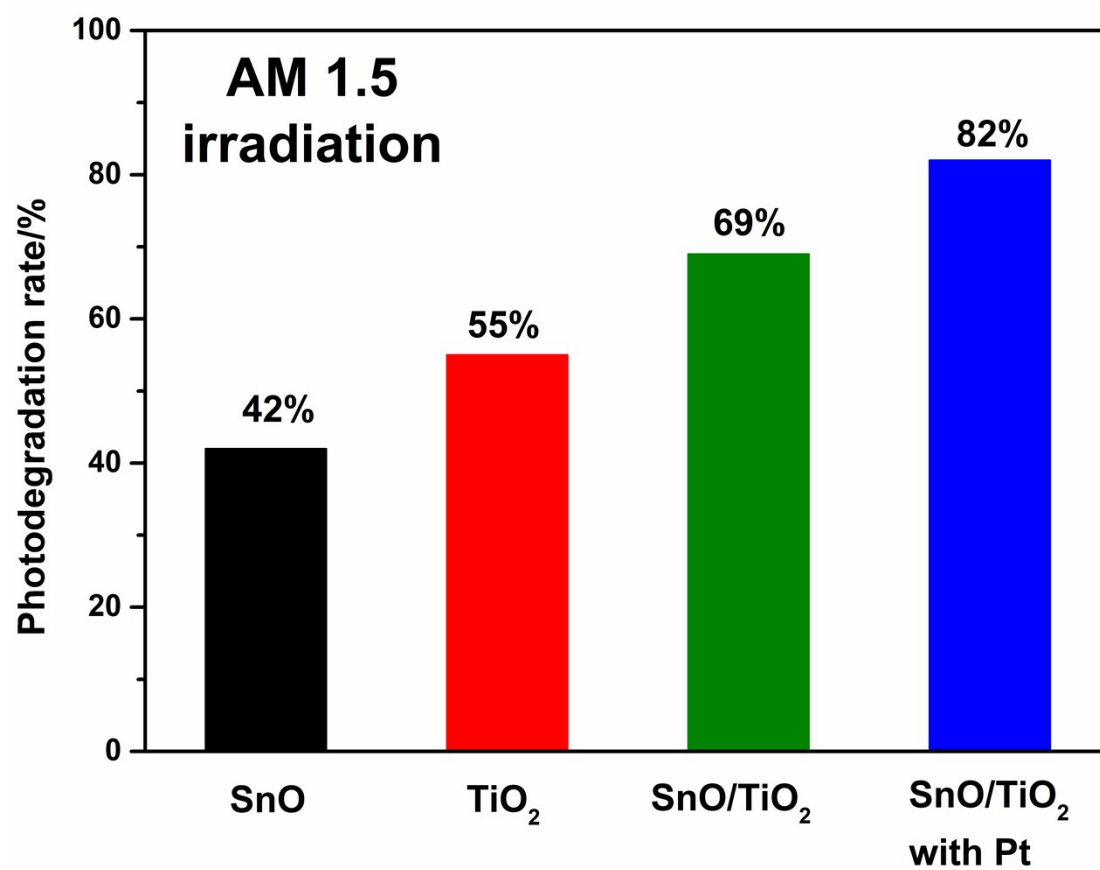


Fig. S8 Photodegradation rate of as-prepared photocatalysts.

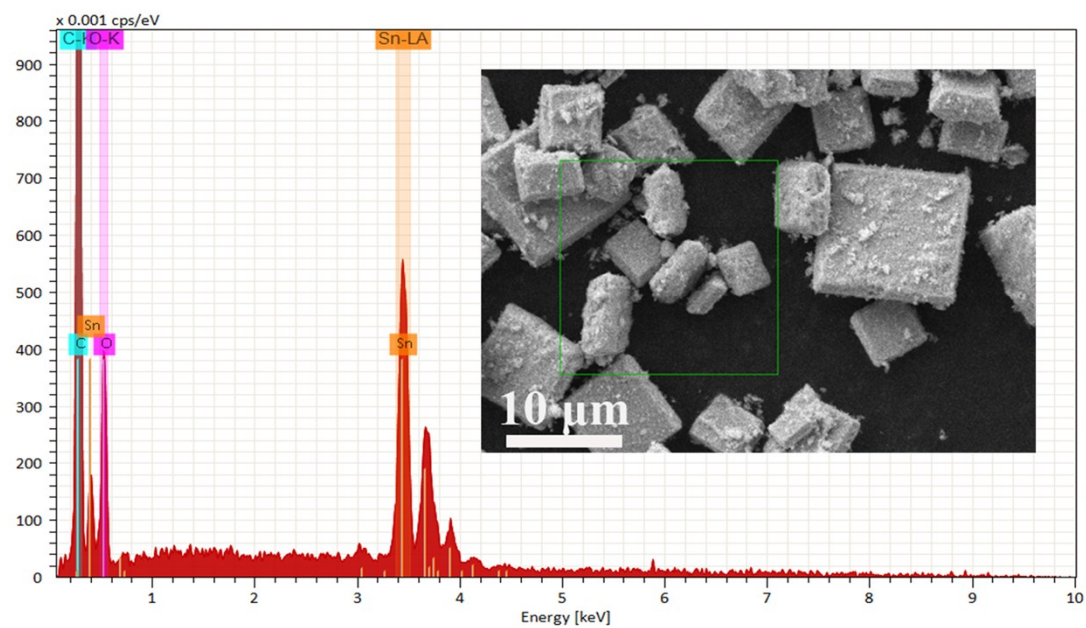


Fig. S9 EDS of SnO after H₂ evolution reaction.

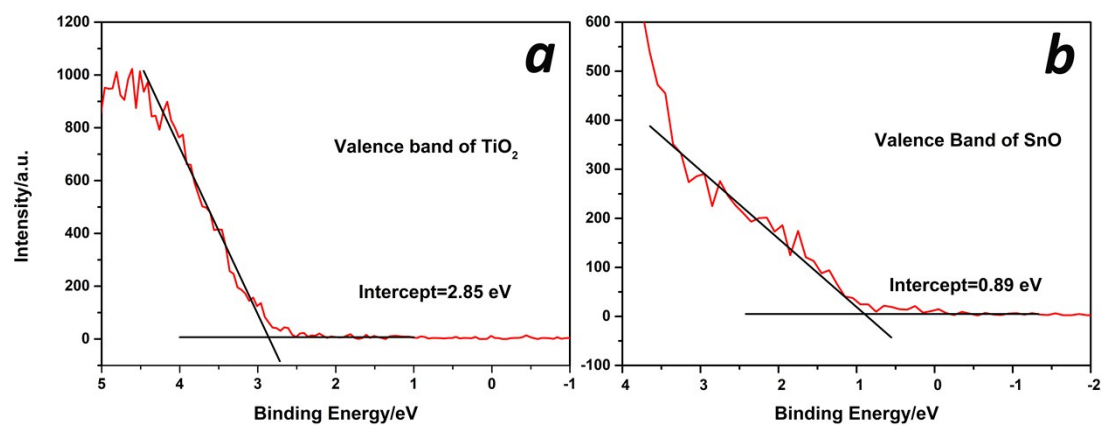


Fig. S10 Valence band of a) TiO₂ and b) SnO, respectively.

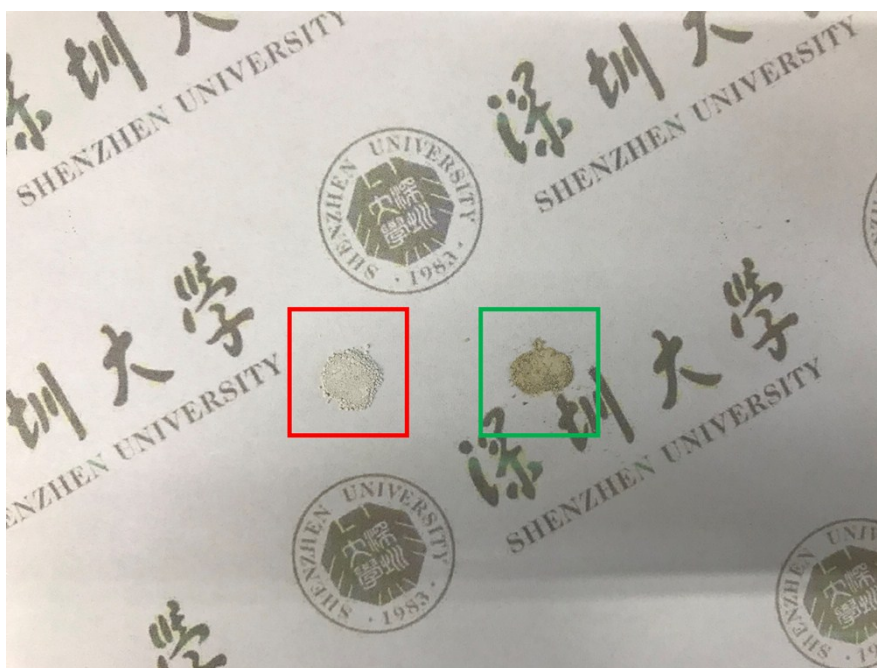


Fig. S11 Comparing the sample color between a) in-situ prepared SnO/TiO₂ heterojunction (red circles) and b) mechanical mixing TiO₂ and SnO (green circle).

Table S1. Peak position of XPS spectra.

Sample	Ti 2p (eV)		Sn 3d (eV)		O 1s (eV)		Pt 4f (eV)		
	Ti 2p _{3/2}	Ti 2p _{1/2}	Sn 3d _{5/2}	Sn 3d _{3/2}	lattice O ²⁻	OH ⁻	Pt 4f _{5/2}	Pt 4f _{7/2}	
TiO ₂	458.3	464.0	—	—	529.5		531.5	-	-
SnO	—	—	486.6	495.1	530.0		—	-	-
SnO/TiO ₂	458.5	464.2	486.5	495.0	529.5 (TiO ₂)	530.0 (SnO)	531.6	-	-
SnO/TiO ₂ with Pt	458.7	464.4	486.8	495.3	-	-	-	73.3	69.9

	A_1 (%)	τ_1 (ns)	A_2 (%)	τ_2 (ns)	τ_a (ns)
TiO_2	91.2	3.4	8.8	18.7	4.8
SnO/TiO_2	90.4	3.1	9.6	69.5	9.5
SnO/TiO_2 with Pt	90.5	3.2	9.5	83.7	10.8

Table S2. Results of carriers lifetime.