

## Thermoelectric properties of (Ba,K)Cd<sub>2</sub>As<sub>2</sub> crystallized in the CaAl<sub>2</sub>Si<sub>2</sub>-type structure

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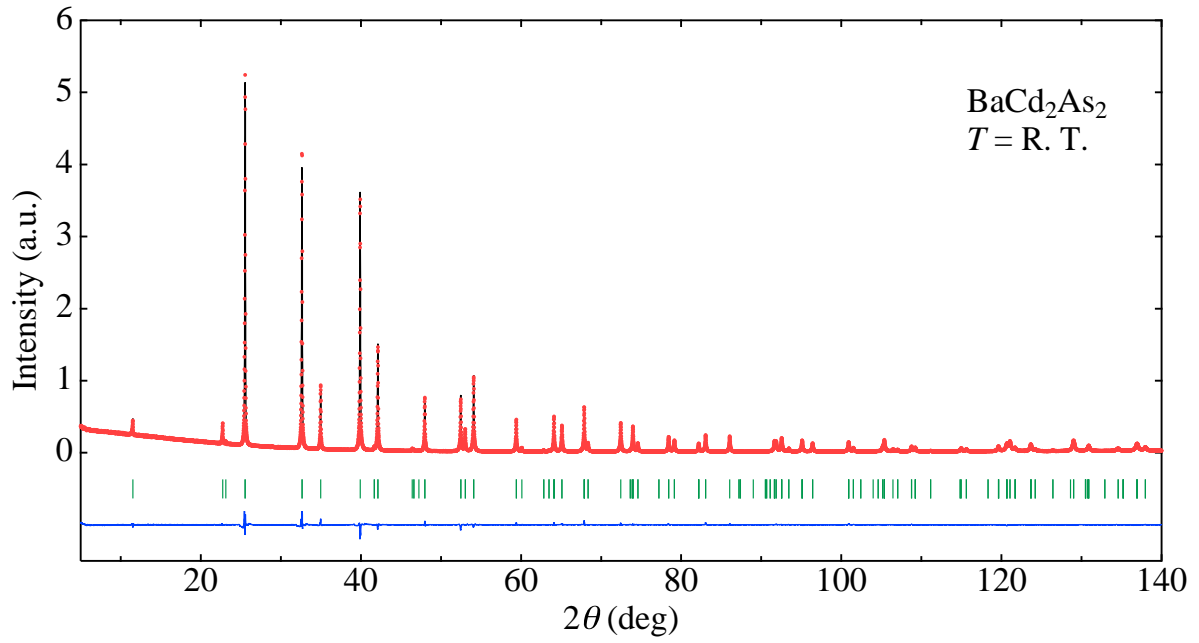


FIG. S1. Observed (red dots) and calculated (solid black line) x-ray powder diffraction patterns for BaCd<sub>2</sub>As<sub>2</sub> at room temperature. Vertical bars show the calculated positions of the diffraction peaks for BaCd<sub>2</sub>As<sub>2</sub>. The solid line at the bottom indicates the difference between observations and calculations.

Atom	Site	$g$	$x$	$y$	$z$	$B$ ( $\text{\AA}^2$ )
<b>BaCd<sub>2</sub>As<sub>2</sub></b>						
$a = 4.5142(5) \text{ \AA}, c = 7.6905(7) \text{ \AA}, R_{\text{wp}} = 5.46 \%$						
Ba	1a	1	0	0	0	0.21(2)
Cd	2d	1	1/3	2/3	0.62892(6)	0.59(2)
As	2d	1	1/3	2/3	0.2632(1)	0.34(2)
<b>Ba<sub>0.99</sub>K<sub>0.01</sub>Cd<sub>2</sub>As<sub>2</sub></b>						
$a = 4.5146(4) \text{ \AA}, c = 7.6930(5) \text{ \AA}, R_{\text{wp}} = 5.22 \%$						
Ba	1a	0.99	0	0	0	0.10(2)
K	1a	0.01	0	0	0	$B_{\text{Ba}}$
Cd	2d	1	1/3	2/3	0.62885(6)	0.40(2)
As	2d	1	1/3	2/3	0.26271(9)	0.18(2)
<b>Ba<sub>0.98</sub>K<sub>0.02</sub>Cd<sub>2</sub>As<sub>2</sub></b>						
$a = 4.5151(4) \text{ \AA}, c = 7.6980(5) \text{ \AA}, R_{\text{wp}} = 5.40 \%$						
Ba	1a	0.98	0	0	0	0.16(2)
K	1a	0.02	0	0	0	$B_{\text{Ba}}$
Cd	2d	1	1/3	2/3	0.62895(6)	0.44(2)
As	2d	1	1/3	2/3	0.26266(9)	0.23(2)
<b>Ba<sub>0.97</sub>K<sub>0.03</sub>Cd<sub>2</sub>As<sub>2</sub></b>						
$a = 4.5143(4) \text{ \AA}, c = 7.7021(5) \text{ \AA}, R_{\text{wp}} = 5.45 \%$						
Ba	1a	0.97	0	0	0	0.25(2)
K	1a	0.03	0	0	0	$B_{\text{Ba}}$
Cd	2d	1	1/3	2/3	0.62910(6)	0.57(2)
As	2d	1	1/3	2/3	0.26270(9)	0.36(2)
<b>Ba<sub>0.96</sub>K<sub>0.04</sub>Cd<sub>2</sub>As<sub>2</sub></b>						
$a = 4.5138(4) \text{ \AA}, c = 7.7061(5) \text{ \AA}, R_{\text{wp}} = 5.77 \%$						
Ba	1a	0.96	0	0	0	0.38(2)
K	1a	0.04	0	0	0	$B_{\text{Ba}}$
Cd	2d	1	1/3	2/3	0.62890(6)	0.72(2)
As	2d	1	1/3	2/3	0.2626(1)	0.45(2)
<b>Ba<sub>0.92</sub>K<sub>0.08</sub>Cd<sub>2</sub>As<sub>2</sub></b>						
$a = 4.5119(4) \text{ \AA}, c = 7.7129(5) \text{ \AA}, R_{\text{wp}} = 6.56 \%$						
Ba	1a	0.92	0	0	0	0.35(2)
K	1a	0.08	0	0	0	$B_{\text{Ba}}$
Cd	2d	1	1/3	2/3	0.62966(7)	0.80(2)
As	2d	1	1/3	2/3	0.2637(1)	0.62(2)

FIG. S2. Atomic parameters for Ba<sub>1-x</sub>K<sub>x</sub>Cd<sub>2</sub>As<sub>2</sub> (space group  $P\bar{3}m1$ ) determined by Rietveld refinements. The quantities  $g$  and  $B$  are the site occupancy and the isotropic atomic displacement parameters. The isotropic atomic displacement parameters of K atoms are constrained to be those of Ba atoms ( $B_{\text{Ba}}$ ).

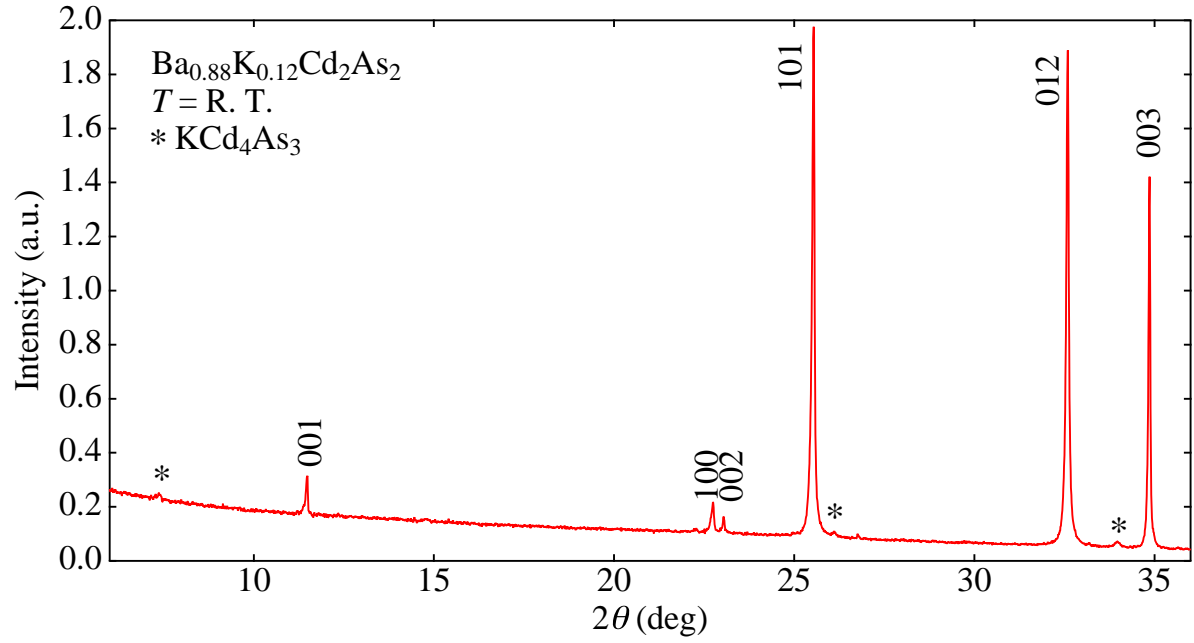


FIG. S3. X-ray diffraction pattern for  $\text{Ba}_{0.88}\text{K}_{0.12}\text{Cd}_2\text{As}_2$  at room temperature.  $\text{KCd}_4\text{As}_3$  appears as an impurity phase.

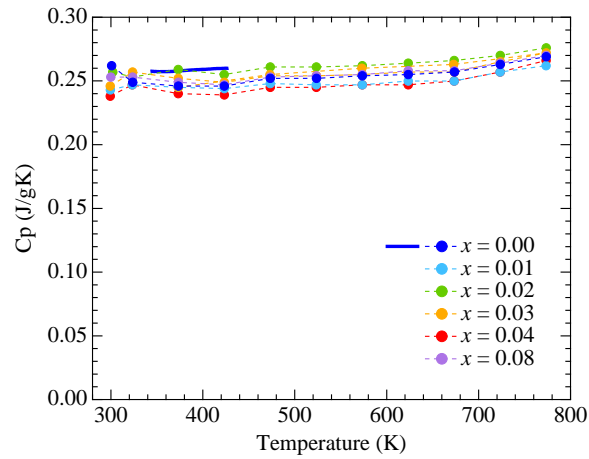


FIG. S4. Temperature dependence of the specific heat for  $\text{Ba}_{1-x}\text{K}_x\text{Cd}_2\text{As}_2$  measured by the laser-flash method (circles) and the differential scanning calorimetry (solid line).