

Supplementary material for:

Heavier Pnictinidene Gold(I) Complexes.

Monika Kořenková, Vít Kremláček, Milan Erben, Robert Jirásko, Frank De Proft, Jan Turek,
Roman Jambor, Aleš Růžička, Ivana Císařová, Libor Dostál

Table of contents:

Table S1 summarizing the crystallographic data	page S2
Figure S1	page S3
Computational details	page S4
Computational results	page S5
References	page S10

Table S1. Relevant crystallographic data for the studied compounds.

	2	(3⁺)[Au₂(C≡CPh)₃]⁻	(4⁺)[Au(C≡CPh)₂]⁻
chemical formula	C ₁₆ H ₂₃ AuClN ₂ Sb.CH ₂ Cl ₂	C ₃₂ H ₄₆ As ₂ AuN ₄ .C ₂₄ H ₁₅ Au ₂	2(C ₃₂ H ₄₆ AuN ₄ Sb ₂).2(C ₁₆ H ₁₀ Au).C ₄ H ₈ O
Cryst syst	monoclinic	triclinic	monoclinic
Space group	P2 ₁ /c	P-1	P2 ₁ /c
<i>a</i> [Å]	13.2510(8)	13.9980(12)	17.4908(13)
<i>b</i> [Å]	12.8921(11)	14.9401(11)	21.2341(13)
<i>c</i> [Å]	16.0880(7)	15.6470(12)	27.3532(19)
α[°]	90	89.143(6)	90
β[°]	126.448(6)	64.451(7)	100.939(2)
γ[°]	90	72.326(7)	90
<i>Z</i>	4	2	4
μ[mm ⁻¹]	8.215	9.095	6.974
D _x [Mg m ⁻³]	2.050	1.879	1.815
Cryst size [mm]	0.27×0.25×0.18	0.48×0.22×0.09	0.19×0.09×0.02
θ range, [deg]	1-27.5	1-27.5	2.372-25.9877
T _{min} , T _{max}	0.228, 0.351	0.4178, 0.7456	0.355, 0.867
no. of reflns measd	21 379	33 182	146 576
no. of unique reflns, R _{int}	5061, 0.045	8885, 0.065	19587, 0.028
no. of obsd reflns	4360	5685	16058
no. of params	217	606	1109
S all data	1.120	0.989	1.054
final R indices [<i>I</i> >2σ(<i>I</i>)]	0.036	0.045	0.024
wR2 indices (all data)	0.090	0.075	0.048
Δρ, max., min. [e Å ⁻³]	3.203, -1.884	1.059, -0.917	0.867, -1.244

	(5⁺)[BF₄]⁻	(6⁺)[BF₄]⁻	(7⁺)[BF₄]⁻
chemical formula	2(C ₄₃ H ₅₉ AsAuN ₄).2(BF ₄) .C ₆ H ₆	C _{42.75} H _{58.75} AuN _{3.75} Sb.BF ₄	C _{43.25} H _{59.25} AuBiN ₄ .BF ₄
Cryst syst	Triclinic	monoclinic	monoclinic
Space group	P-1	P21/c	P21/c
<i>a</i> [Å]	13.3370(8)	10.9280(17)	10.9912(3)
<i>b</i> [Å]	17.4601(15)	24.306(5)	24.1964(6)
<i>c</i> [Å]	19.7050(17)	19.2240(17)	19.3350(2)
α[°]	86.893(7)	90	90
β[°]	83.441(6)	119.412(8)	119.362(4)
γ[°]	85.496(8)	90	90
<i>Z</i>	2	4	4
μ[mm ⁻¹]	4.016	3.951	7.241
D _x [Mg m ⁻³]	1.507	1.539	1.672
Crys tsize [mm]	0.59×0.36×0.22	0.45×0.24×0.20	0.59×0.26×0.26
θ range, [deg]	1-27.5	1-27.5	1-27.5
T _{min} , T _{max}	0.323, 0.551	0.382, 0.608	0.192, 0.314
no. of reflns measd	94 388	35 980	37 310
no. of unique reflns, R _{int}	20373, 0.054	10175, 0.079	10147, 0.086
no. of obsd reflns	15639	6683	6558
no. of params	1020	512	597
S all data	1.014	1.120	1.122
final R indices [<i>I</i> >2σ(<i>I</i>)]	0.043	0.066	0.072
wR2 indices (all data)	0.100	0.149	0.170
Δρ, max., min. [e Å ⁻³]	3.518, -2.592	1.821, -4.419	3.518, -2.592

Definitions: R_{int} = $\sum |F_o^2 - F_{o,\text{mean}}^2| / \sum F_o^2$, S = $[\sum (w(F_o^2 - F_c^2)^2) / (N_{\text{diffs}} - N_{\text{params}})]^{1/2}$ for all data, R(F) = $\sum |F_o| - |F_c| / \sum |F_o|$ for observed data, wR(F²) = $[\sum (w(F_o^2 - F_c^2)^2) / (\sum w(F_o^2)^2)]^{1/2}$ for all data.

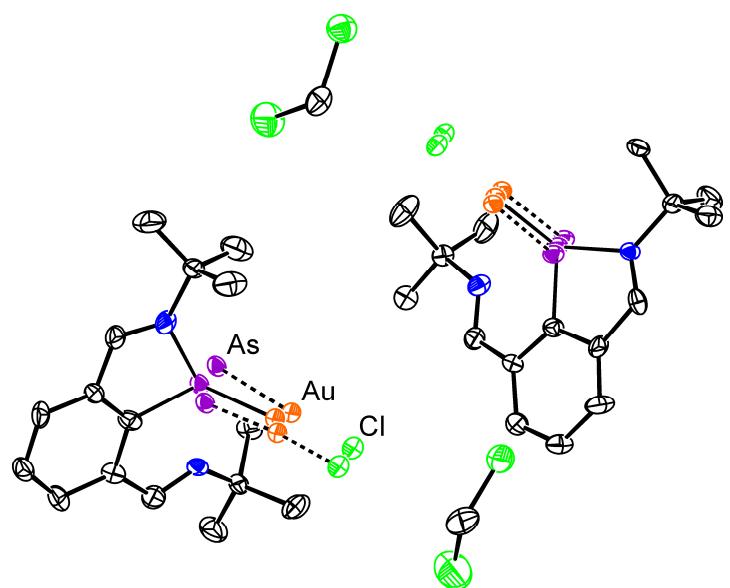


Figure S1. View on the heavily disordered structure of **1**.

Computational details

All calculations were carried out using Density Functional Theory (DFT) as implemented in the Gaussian09 quantum chemistry program.^{S1} Geometry optimizations were carried out at the M06/cc-pVDZ^{S2} level of theory (for heavier atoms - As, Sb, Bi, and Au - the cc-pVDZ-PP^{S3} basis set including small-core relativistic pseudopotentials that account also for relativistic effects was used). The electronic energies were re-evaluated by additional single point calculations on each of all optimized geometries using the triple- ζ -quality cc-pVTZ(-PP) basis set. Analytical vibrational frequencies within the harmonic approximation were computed with the cc-pVDZ basis set to confirm a proper convergence to well-defined minima or saddle points on the potential energy surface. The subsequent NBO analysis^{S4} and calculation of Wiberg bond indices^{S5} were performed at the M06/cc-pVTZ(-PP) level.

Computational results

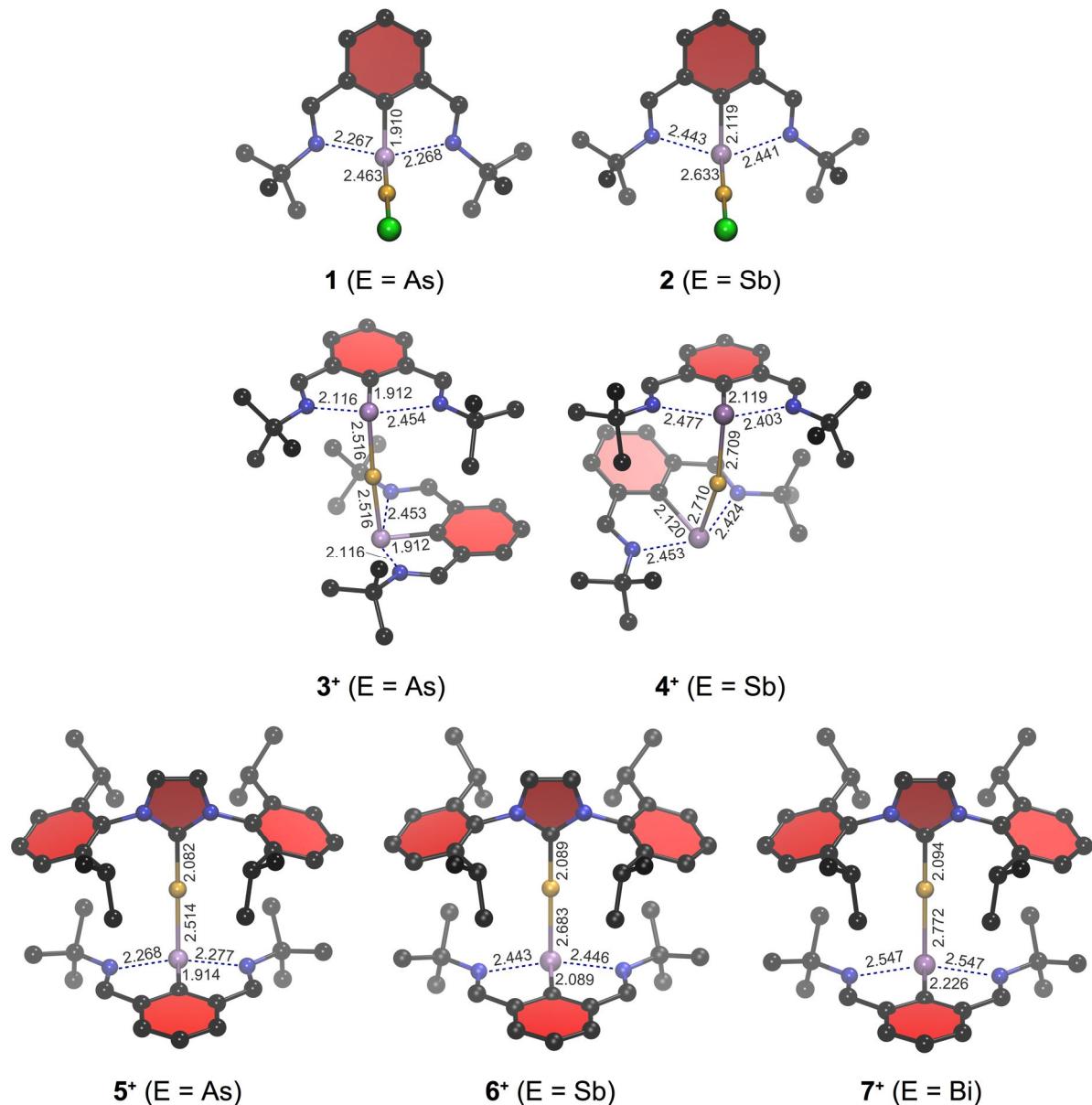


Figure S2. M06/cc-pVDZ(-PP) optimized geometries of compounds **1-7⁺** (Hydrogen atoms are omitted for clarity) along with selected distances (in Å).

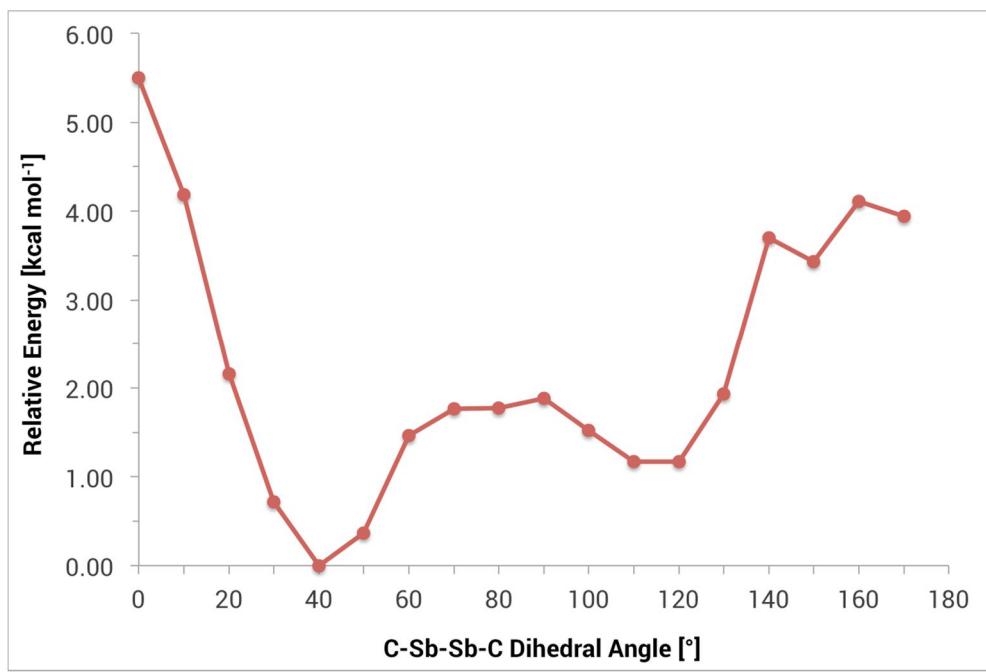


Figure S3. Relaxed potential energy scan along the C-Sb-Sb-C dihedral angle in $\mathbf{4}^+$.

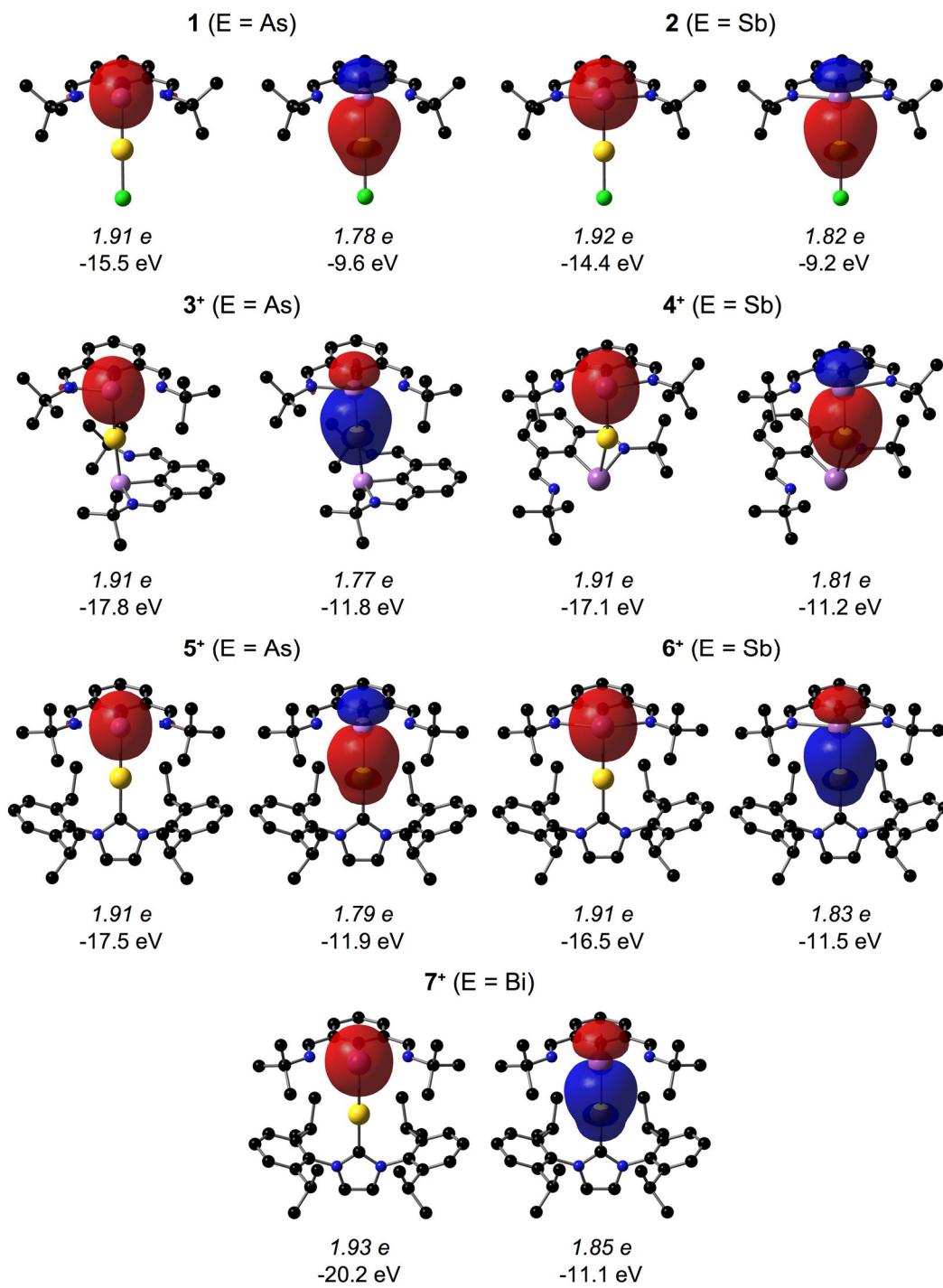


Figure S4. Relevant NBOs (isosurface 0.03 a.u.) showing the σ -type lone pair on the pnictogen center and E-Au σ -bond in **1**–**7**⁺ (Hydrogen atoms are omitted for clarity). NBO populations and orbital energies are also displayed.

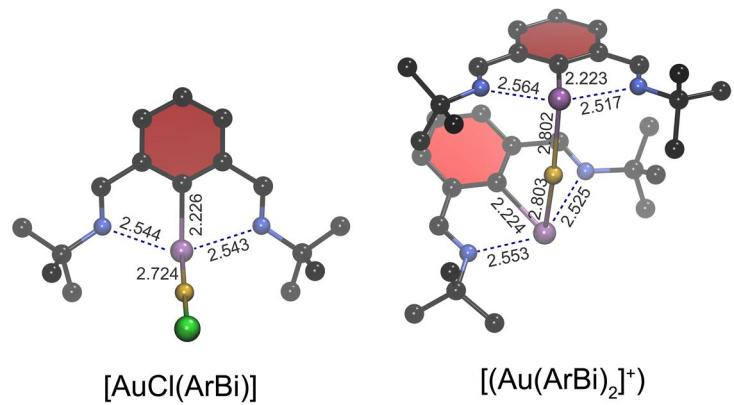


Figure S5. M06/cc-pVDZ(-PP) optimized geometries of compounds $[\text{AuCl}(\text{ArBi})]$ and $[\text{Au}(\text{ArBi})_2]^+$ (Hydrogen atoms are omitted for clarity) along with selected distances (in Å).

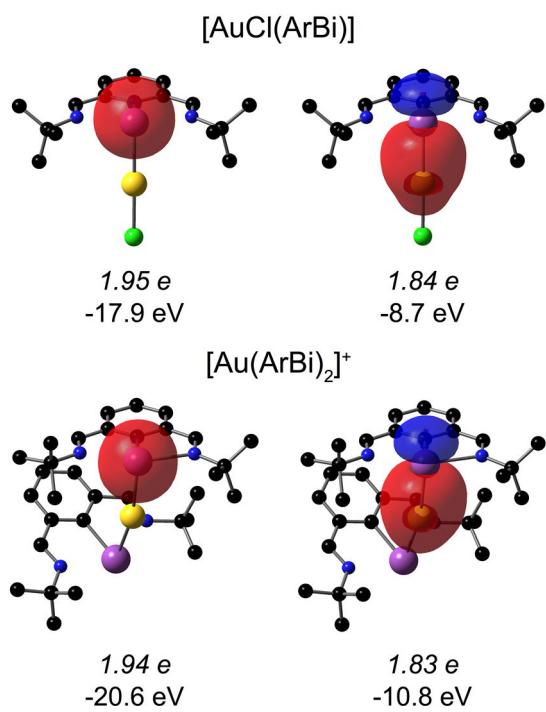


Figure S6. Relevant NBOs (isosurface 0.03 a.u.) showing the σ -type lone pair on the pnicogen center and E-Au σ -bond in $[\text{AuCl}(\text{ArBi})]$ and $[\text{Au}(\text{ArBi})_2]^+$ (Hydrogen atoms are omitted for clarity). NBO populations and orbital energies are also displayed.

Table S2. Wiberg bond indices (WBI) and NPA atomic charges (q) for the compounds $[\text{Au}(\text{IPr})(\text{ACN})]^+$, $[\text{Au}(\text{IPr})(\text{DMAP})]^+$, $[\text{Au}(\text{IPr})_2]^+$ and $[\text{Au}(\text{IPr})(\text{PPh}_3)]^+$.

Compound	WBI _{L-Au} ^[a]	WBI _{(IPr)-Au}	q_{Au}	q_L ^[a]	$q_{\text{C(IPr)}}$
$[\text{Au}(\text{IPr})(\text{ACN})]^+$	0.33	0.65	0.43	-0.46	0.16
$[\text{Au}(\text{IPr})(\text{DMAP})]^+$	0.32	0.64	0.36	-0.59	0.17
$[\text{Au}(\text{IPr})_2]^+$	0.51	0.51	0.19	0.14	0.14
$[\text{Au}(\text{IPr})(\text{PPh}_3)]^+$	0.53	0.50	0.19	0.98	0.13

^[a]L corresponds to the donor atom of the respective ligand.

References

- [S1] M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, G. A. Petersson, H. Nakatsuji, X. Li, M. Caricato, A. Marenich, J. Bloino, B. G. Janesko, R. Gomperts, B. Mennucci, H. P. Hratchian, J. V. Ortiz, A. F. Izmaylov, J. L. Sonnenberg, D. Williams-Young, F. Ding, F. Lipparini, F. Egidi, J. Goings, B. Peng, A. Petrone, T. Henderson, D. Ranasinghe, V. G. Zakrzewski, J. Gao, N. Rega, G. Zheng, W. Liang, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, K. Throssell, J. A. Montgomery, Jr., J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, T. Keith, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, J. M. Millam, M. Klene, C. Adamo, R. Cammi, J. W. Ochterski, R. L. Martin, K. Morokuma, O. Farkas, J. B. Foresman, and D. J. Fox, *Gaussian 09, Revision D.01*, Gaussian, Inc., Wallingford CT, 2009.
- [S2] a) Y. Zhao and D.G. Truhlar, *Theor. Chem. Acc.*, 2008, **120**, 215-241; b) T.H. Dunning, *J. Chem. Phys.*, 1989, **90**, 1007-1023; c) D.E. Woon and T.H. Dunning, *J. Chem. Phys.*, 1993, **99**, 1358-1371.
- [S3] a) K.A. Peterson, *J. Chem. Phys.*, 2003, **119**, 11099-11112; b) K.A. Peterson and C. Puzzarini, *Theor. Chem. Acc.*, 2005, **114**, 283-296.
- [S4] a) J.P. Foster and F. Weinhold, *J. Am. Chem. Soc.*, 1980, **102**, 7211-7218; b) F. Weinhold, *J. Comput. Chem.*, 2012, **33**, 2363-2379 and references therein. c) C.R. Landis and F. Weinhold, F. The NBO View of Chemical Bonding, in “*The Chemical Bond: Fundamental Aspects of Chemical Bonding*” (eds G. Frenking and S. Shaik), Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim, Germany, 2014.
- [S5] K.B. Wiberg, *Tetrahedron* 1968, **24**, 1083-1096.