

Supporting Information for

Comparative Study of the Strongest Solid Lewis Acids Known: ACF and $HS-AlF_3$

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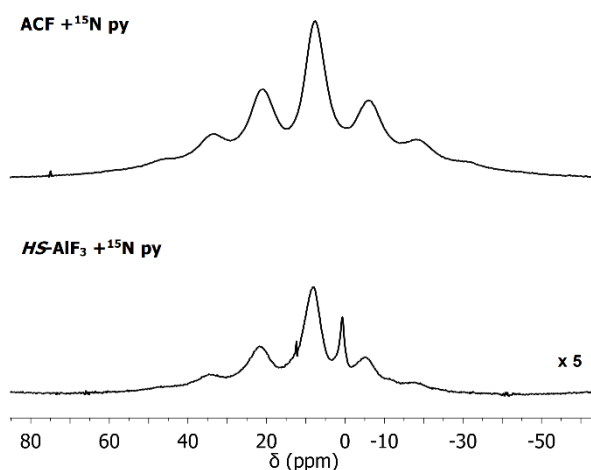


Figure S1. 1H MAS NMR spectra of ACF plus pyridine (top) and $HS-AlF_3$ plus pyridine (bottom); rotation frequency of 5 kHz, $p1 = 6.5 \mu s$ @ 8,5 db

Simulation by dmfit

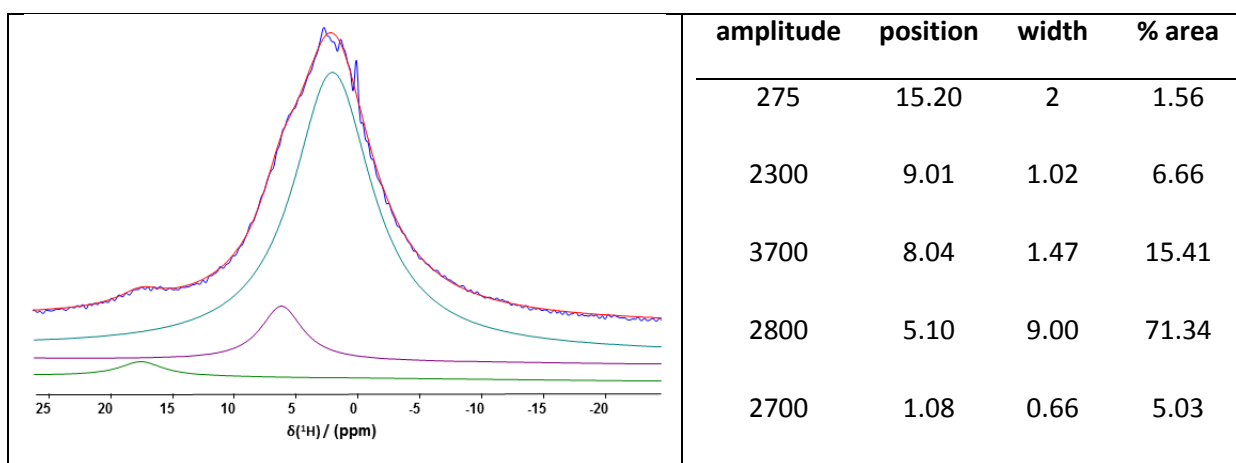


Figure S2. Top blue trace corresponds to the measured 1H MAS NMR spectrum of ACF; subsequent traces are “dmfit” simulated spectra: in red the simulation of the full spectrum; in purple, green and blue the different contributions of the material.

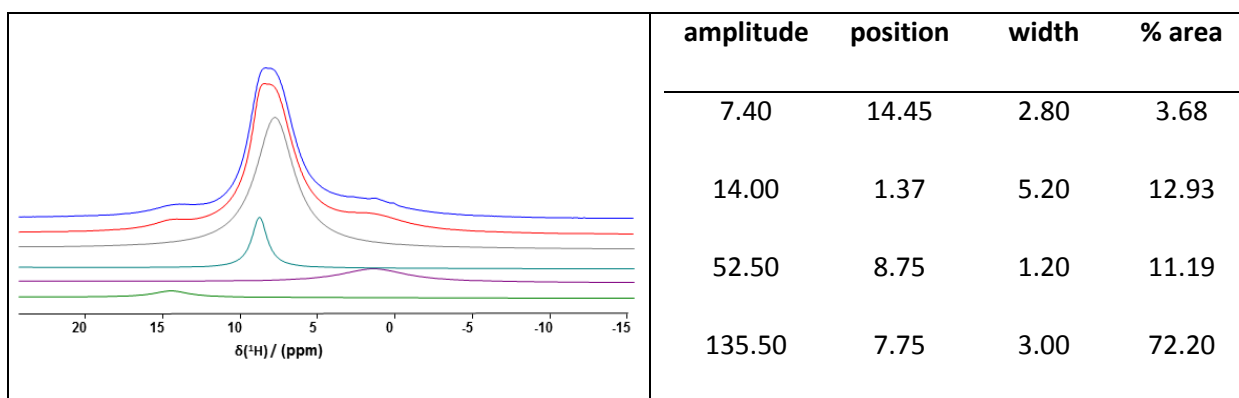


Figure S3. Top blue trace corresponds to the measured ^1H MAS NMR spectrum of $\text{ACF}+^{15}\text{N}$ -pyridine; subsequent traces are “dmfit” simulated spectra: in red the simulation of the full spectrum; in purple the different contributions of the ACF; in green contribution of the Brønsted acid sites ($^+\text{H-NC}_5\text{H}_5$), blue and grey the contributions of the aromatic protons of pyridine and pyridinium.

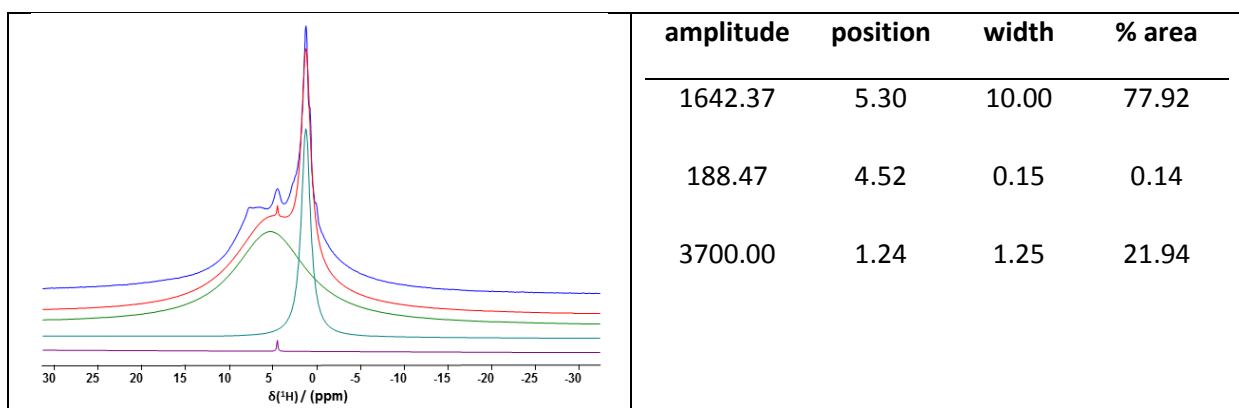


Figure S4. top blue trace corresponds to the measured ^1H MAS NMR spectrum of HS-AlF_3 ; subsequent traces are “dmfit” simulated spectra: in red the simulation of the full spectrum; in purple, green and blue the different contributions of the material.

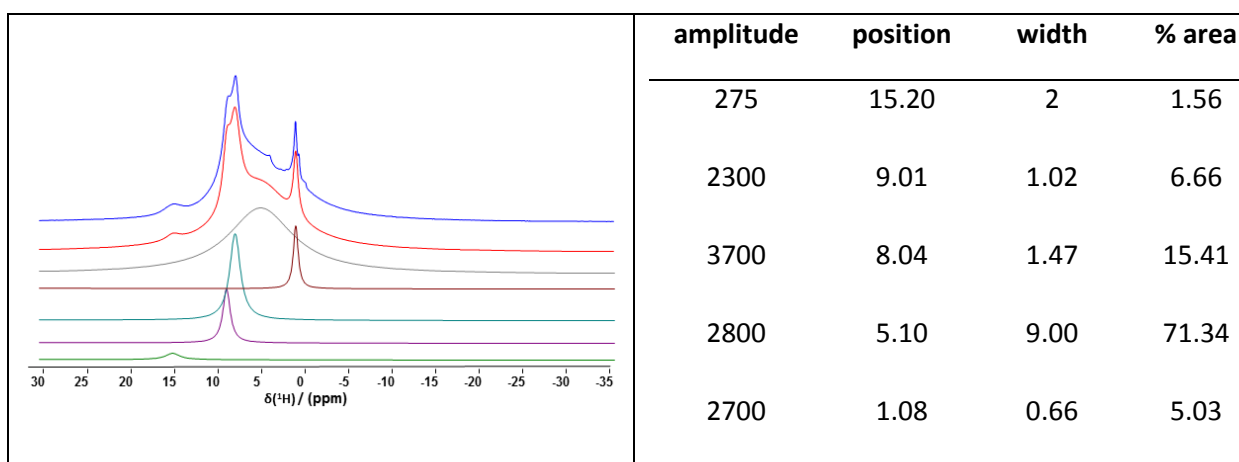


Figure S5. top blue trace corresponds to the measured ^1H MAS NMR spectrum of $\text{HS-AlF}_3+^{15}\text{N}$ -pyridine; subsequent traces are “dmfit” simulated spectra: in red the simulation of the full spectrum; in purple the different contributions of the ACF; in green contribution of the Brønsted acid sites ($^+\text{H-NC}_5\text{H}_5$), blue and grey the contributions of the aromatic protons of pyridine and pyridinium.

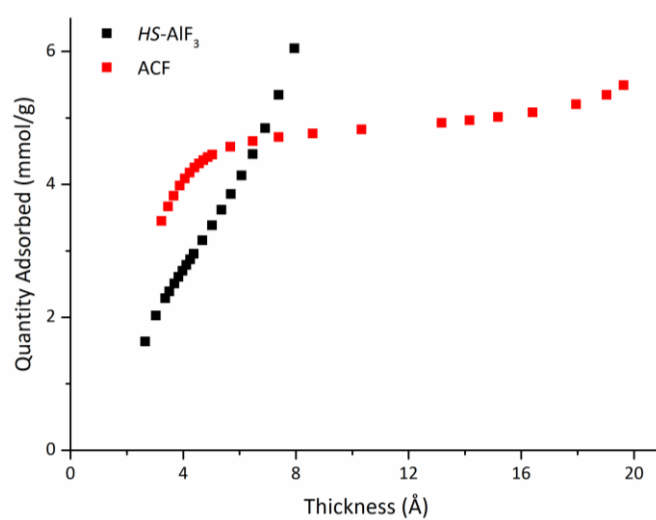


Figure S6. t-plots for ACF and *HS-AIF₃*, considering the calculations of Harkins and Jura (see References).