

Supporting Information

Donor-stabilized molecular Mg/Al-bimetallic hydrides

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Table S1. Crystallographic data for compounds **1**, **2**, **3**, **4**, and **5**

	1	2	3	4	5
formula	C ₂₆ H ₆₀ MgAl ₂ O ₅	C ₁₈ H ₅₀ MgAl ₂ O ₃	C ₁₄ H ₄₀ MgAl ₂ O ₄	C ₃₀ H ₈₄ Al ₄ Mg ₂ O ₂	C ₂₁ H ₄₁ AlMgO ₂
M _r [g mol ⁻¹]	531.01	392.85	350.73	665.51	376.83
CCDC	1853436	1853437	1853438	1853439	1853440
color	colorless/block	colorless/block	colorless/block	colorless/plate	colorless/plate
crystal dimensions [mm]	0.433 x 0.166 x 0.157	0.718 x 0.506 x 0.456	0.819 x 0.608 x 0.483	0.347 x 0.136 x 0.094	0.287 x 0.282 x 0.218
cryst. syst.	triclinic	monoclinic	orthorhombic	monoclinic	orthorhombic
space group	<i>P</i> $\bar{1}$	<i>P</i> 2 ₁ / <i>c</i>	<i>Pna</i> 2 ₁	<i>P</i> 2 ₁ / <i>n</i>	<i>P</i> 2 ₁ <i>P</i> 2 ₁ <i>P</i> 2 ₁
<i>a</i> [Å]	9.554(3)	22.886(3)	17.8880(4)	10.4747(7)	9.746(5)
<i>b</i> [Å]	12.015(4)	12.4234(17)	8.6400(2)	14.0615(9)	15.225(8)
<i>c</i> [Å]	15.069(5)	18.993(3)	14.9623(3)	15.7685(10)	15.880(9)
α [°]	90.875(6)	90	90	90	90
β [°]	92.171(6)	91.203(5)	90	95.423(4)	90
γ [°]	104.872(6)	90	90	90	90
<i>V</i> [Å ³]	1670.1(10)	5398.9(13)	2312.46(9)	2312.1(3)	2357(2)
<i>Z</i>	2	8	4	2	4
<i>T</i> [K]	100(2)	100(2)	100(2)	100(2)	150(2)
ρ_{calc} [g cm ⁻³]	1.056	0.967	1.007	0.956	1.062
μ [mm ⁻¹]	0.134	0.142	0.162	0.153	0.123
F (000)	588	1760	776	744	832
θ range [°]	1.353/26.498	2.0105/11.34	2.277/28.270	3.416/22.49	2.4495/29.485
unique reflns	6875	11724	5493	5730	5419
observed reflns (<i>I</i> >2 σ)	4676	9507	5337	4074	5219
R1/ <i>w</i> R2 (<i>I</i> >2 σ) ^[a]	0.0859/0.1946	0.0649/0.1533	0.0243/0.0645	0.0449/0.1028	0.0313/0.0837
R1/ <i>w</i> R2 (all data) ^[a]	0.1286/0.2143	0.0890/0.1693	0.0253/0.0651	0.10742/0.1173	0.0331/0.0854
GOF ^[a]	1.064	1.053	1.020	1.027	1.075

[a] $R1 = \sum(|F_o| - |F_c|) / \sum |F_o|$, $F_o > 4\sigma(F_o)$. $wR2 = \{\sum[w(F_o^2 - F_c^2)^2] / \sum[w(F_o^2)]\}^{1/2}$.

General procedure for the hydromagnesation of 1-hexene. The metal hydrides were weighed into a vial and dissolved or suspended in 0.3 ml toluene- d_8 . To increase the solubility of the hydrides **3** and **4** two drops of THF- d_8 were added. 1-Hexene was dissolved in a second vial in 0.3 ml toluene- d_8 . The solutions were transferred and combined into a *J. Young* valve NMR sample tube and monitored by ^1H NMR spectroscopy. The reactions were heated carefully in steps of 10 °C until a reaction took place at 130 °C. After 16 h the reactions were stopped and in case of **1** heated again for additional 24 h. All reactions produced $[\text{MgH}_2]_n$ as a white precipitate.

Table S2. Reaction conditions for hydromagnesation reactions

Substrate	1-hexene	T [° C]	time [h]	Yield $\text{Mg}(n\text{-hex})_2$
1 , 20.0 mg, 43.6 μmol	7.34 mg, 87.2 μmol	130	16 h (40 h)	14% (19%)
3 , 20.0 mg, 57.1 μmol	9.61 mg, 114 μmol	130	16 h	6%
4 , 20.0 mg, 30.1 μmol	10.1 mg, 120 μmol	130	16 h	2%

Notes on NMR spectroscopic characterisation.

Due to the limited solubility of **3** in toluene- d_8 two drops of THF- d_8 were added.

* residual solvent peak

~ reduced peak intensity

NMR Spectra

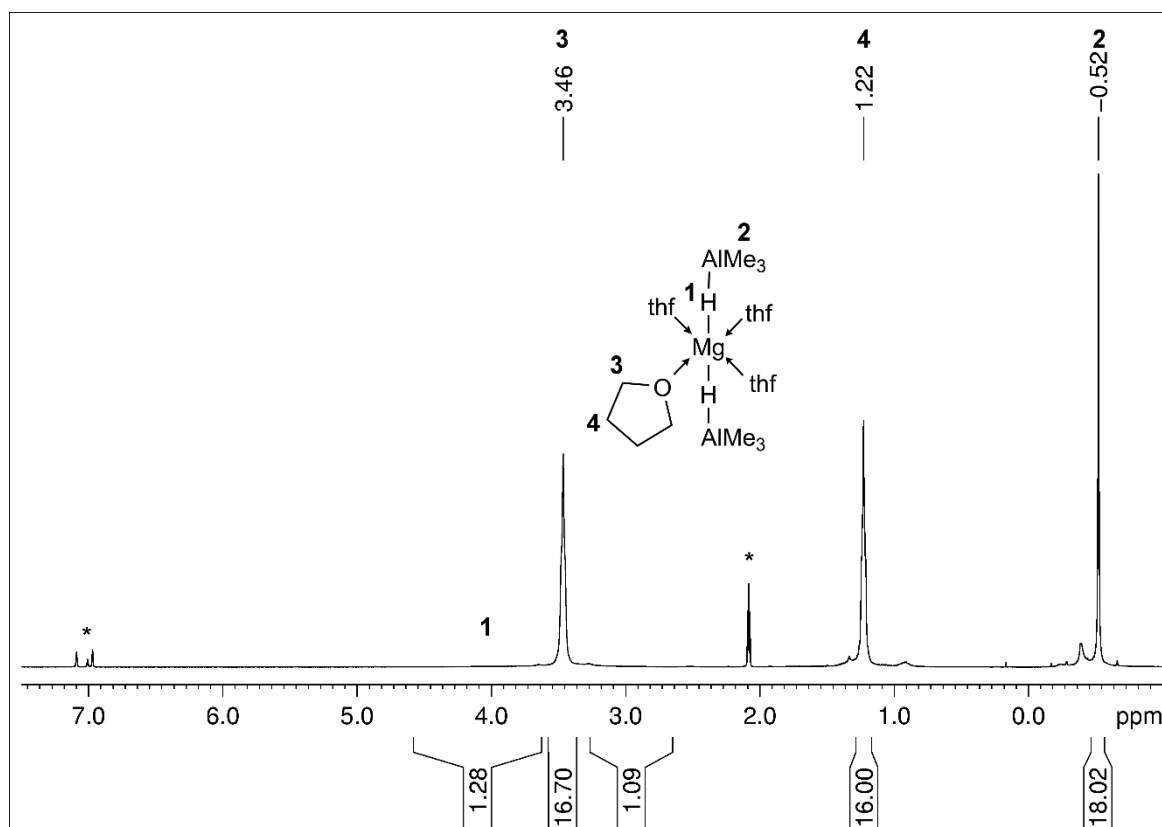


Figure S1. ^1H NMR spectrum (400 MHz) of **1** in $\text{toluene-}d_8$ at 26°C .

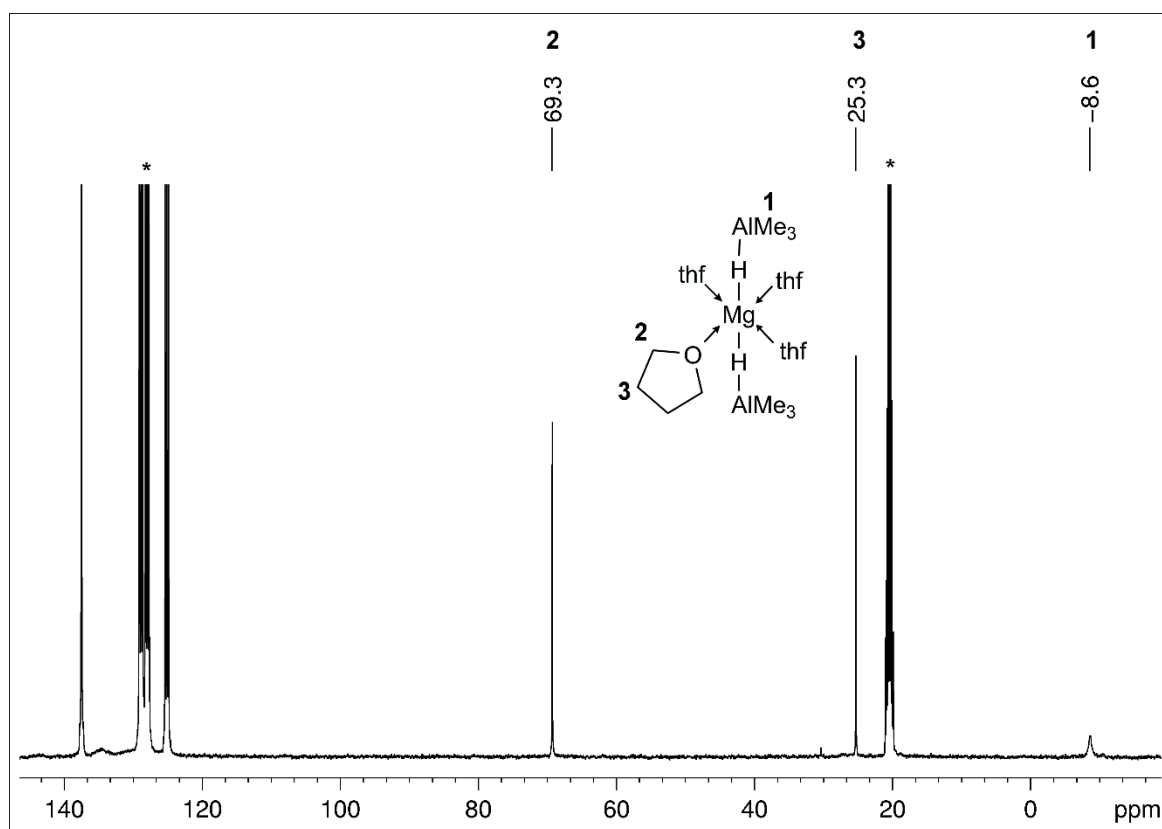


Figure S2. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum (101 MHz) of **1** in $\text{toluene-}d_8$ at 26°C .

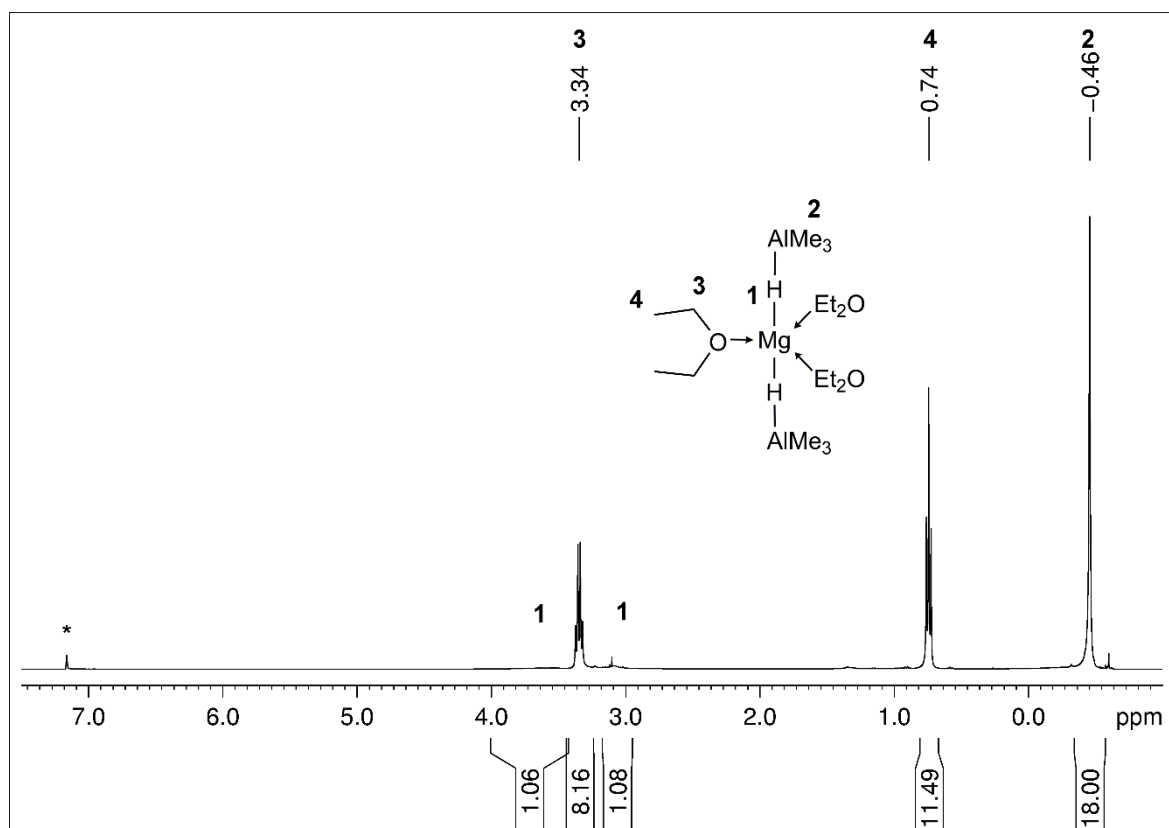


Figure S3. ^1H NMR spectrum (400 MHz) of **2** in $\text{toluene-}d_8$ at $26\text{ }^\circ\text{C}$.

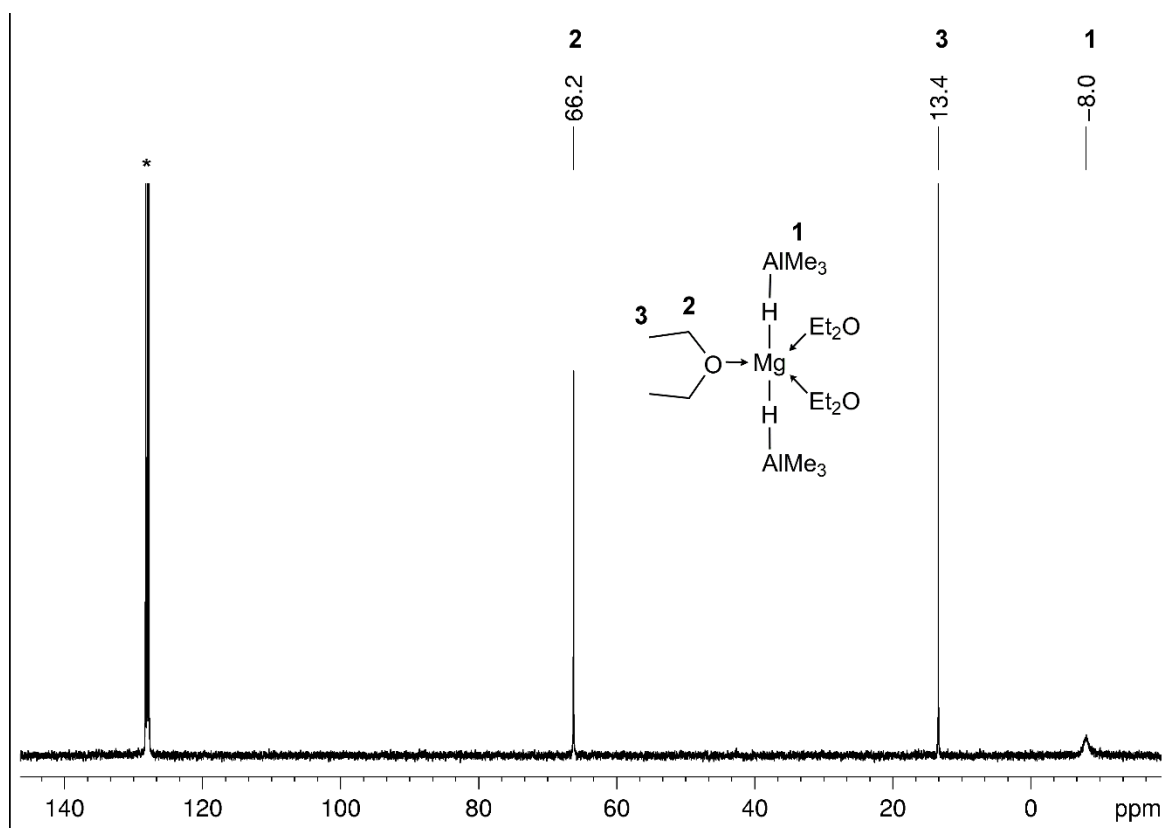


Figure S4. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum (101 MHz) of **2** in $\text{toluene-}d_8$ at $26\text{ }^\circ\text{C}$.

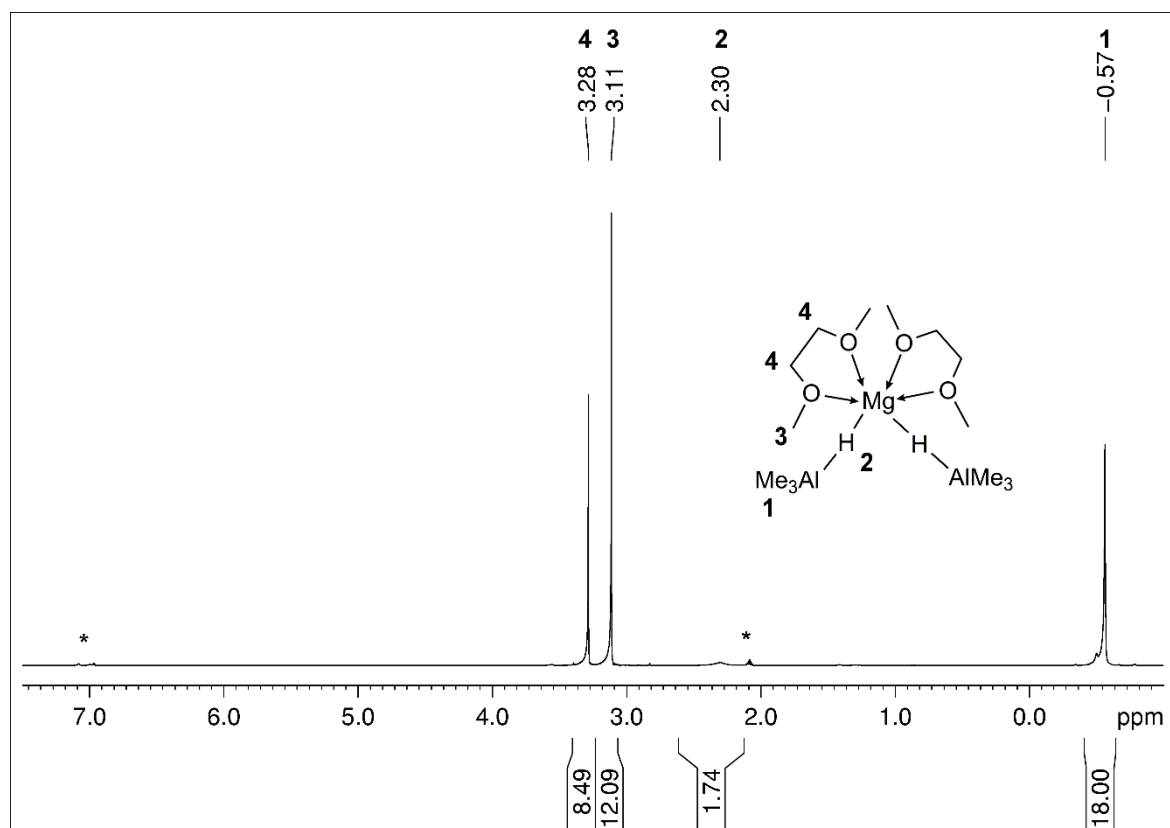


Figure S5. ^1H NMR spectrum (400 MHz) of **3** in toluene- d_8 at 26 °C.

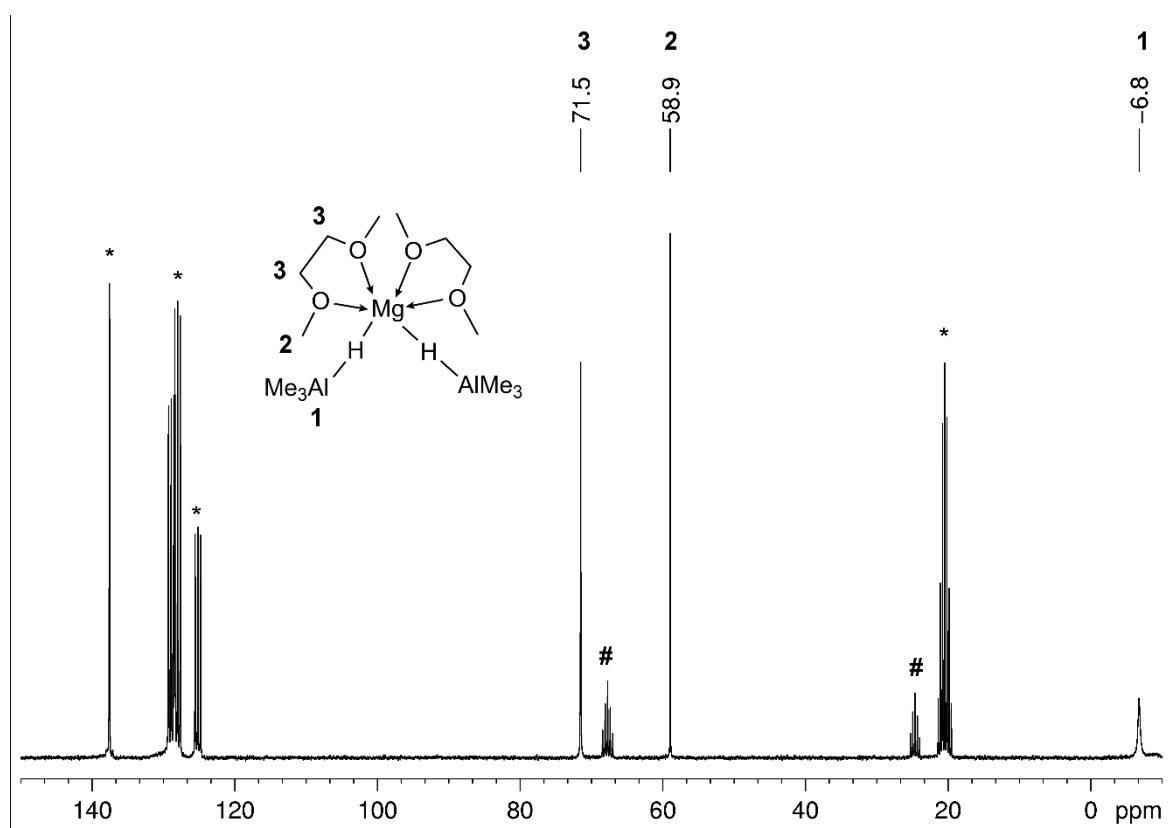


Figure S6. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum (101 MHz) of **3** in toluene- d_8 at 26 °C.

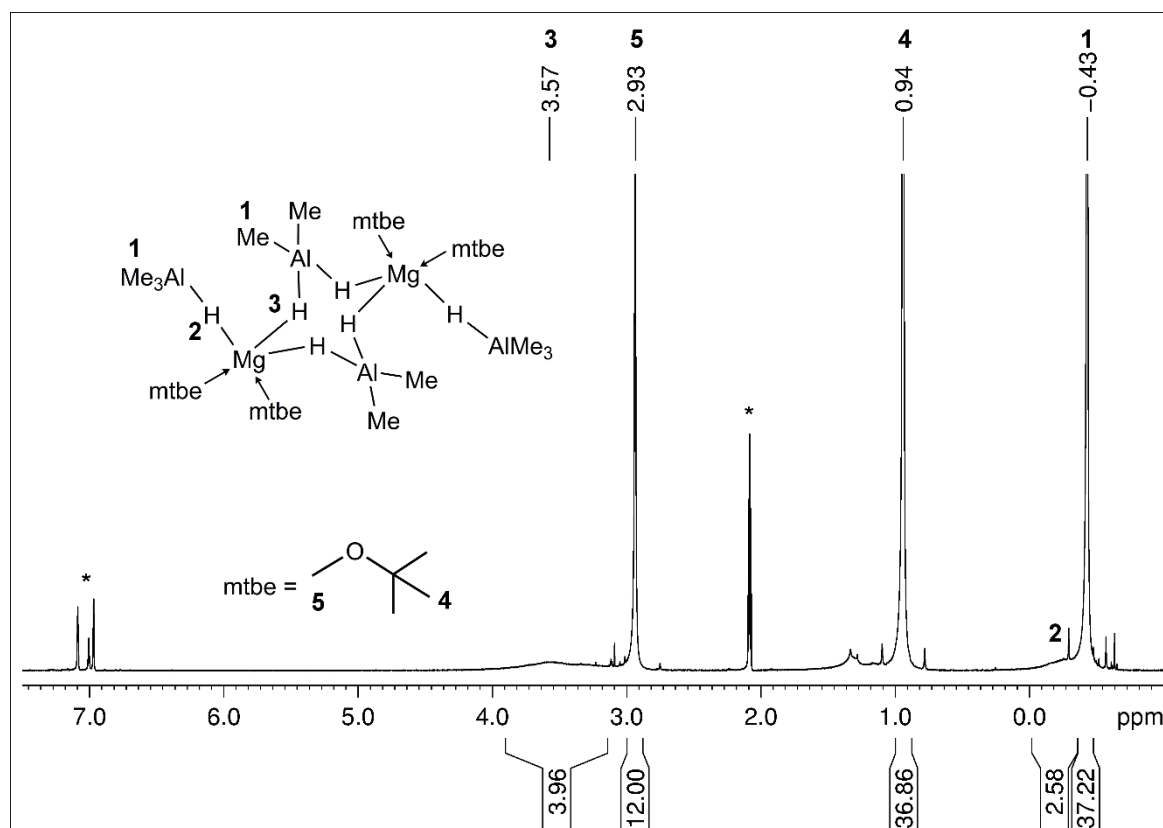


Figure S7. ^1H NMR spectrum (400 MHz) of **4** in toluene- d_8 at 26 °C.

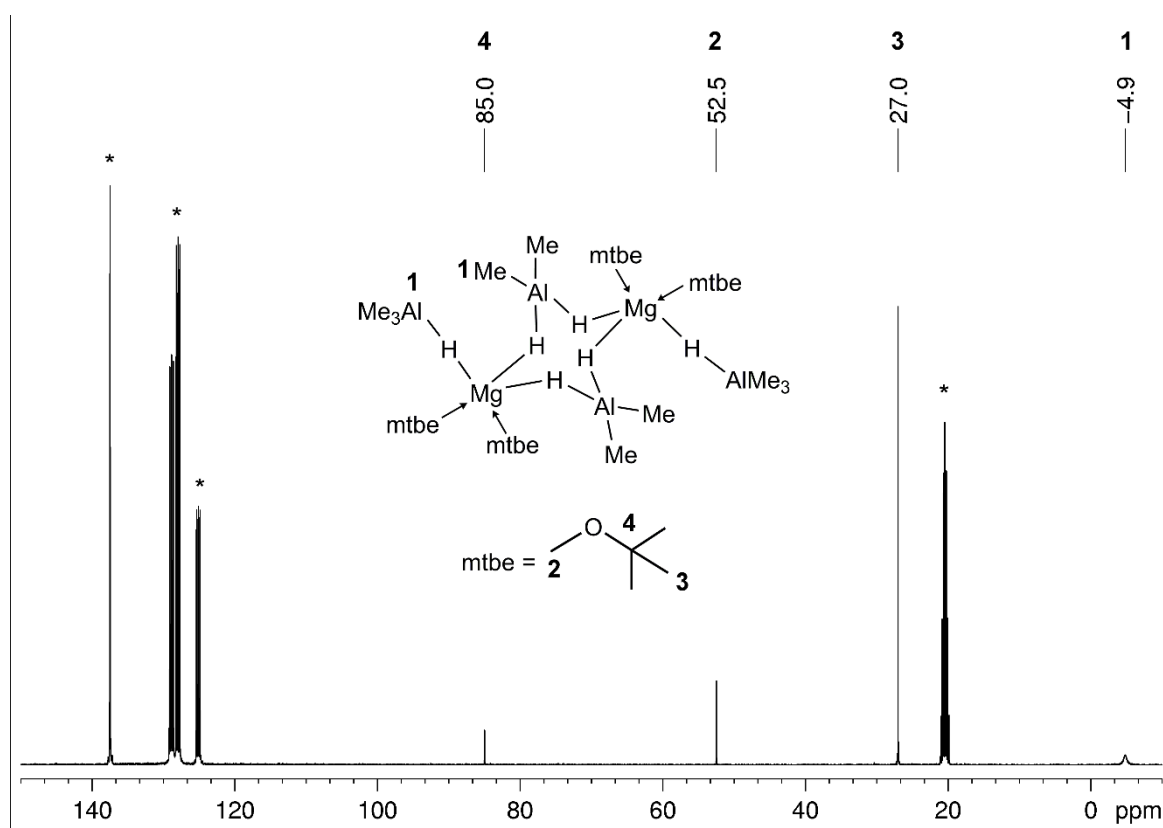


Figure S8. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum (101 MHz) of **4** in toluene- d_8 at 26 °C.

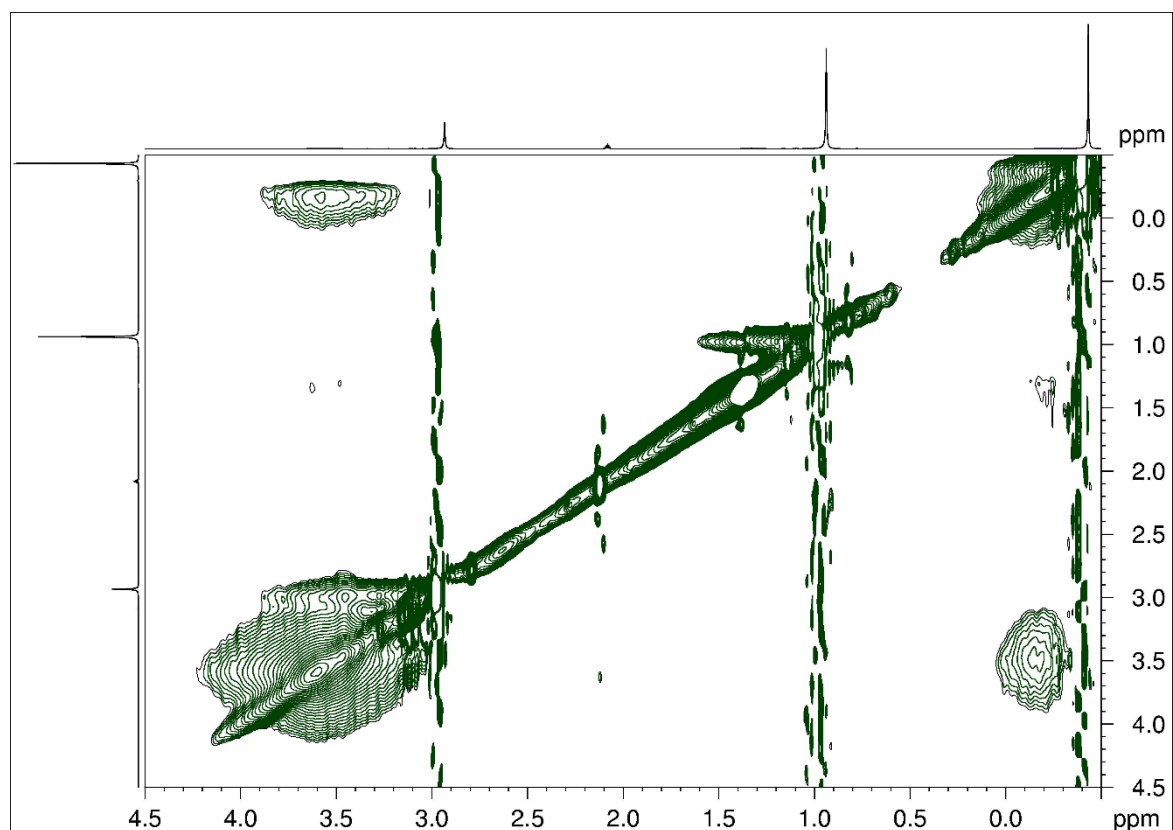


Figure S9. ^1H EXSY NMR spectrum (400 MHz) of **4** in toluene- d_8 at 26 °C indicating hydrido exchange.

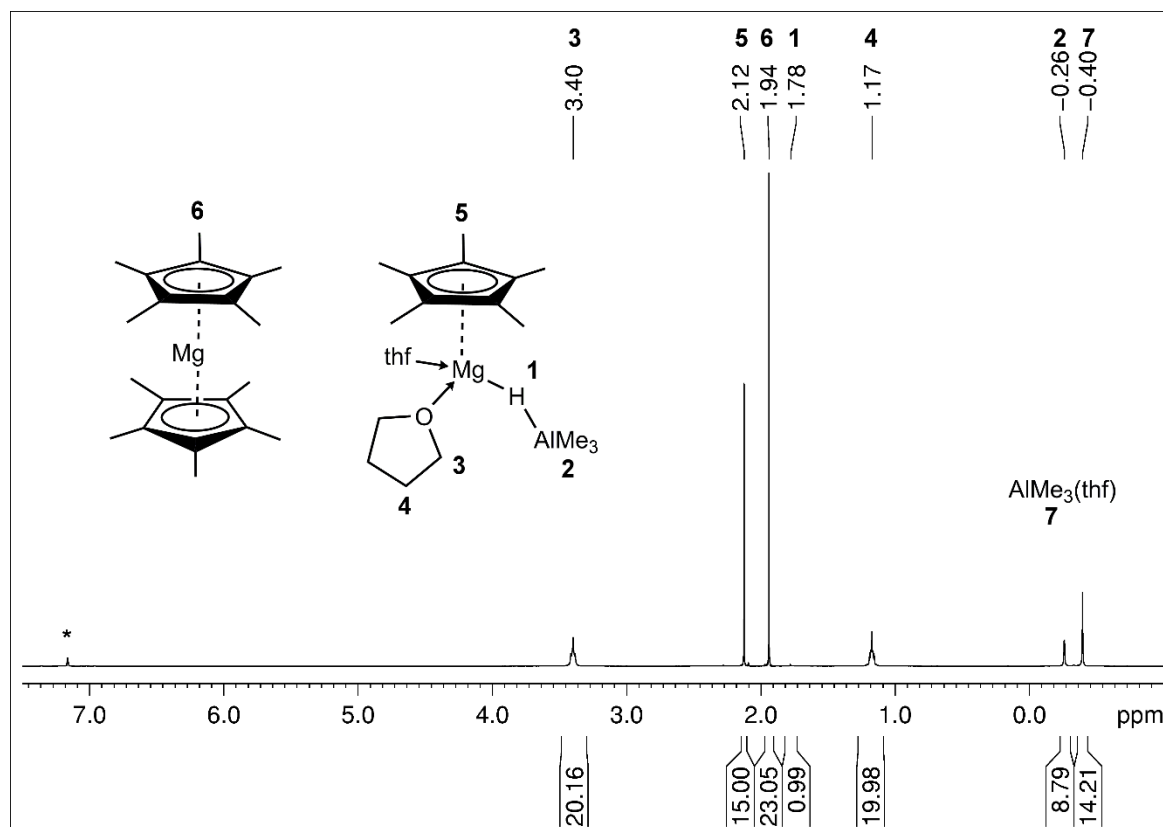


Figure S10. ^1H NMR spectrum (400 MHz) of **5** in benzene- d_6 after 1 h at 26 °C.

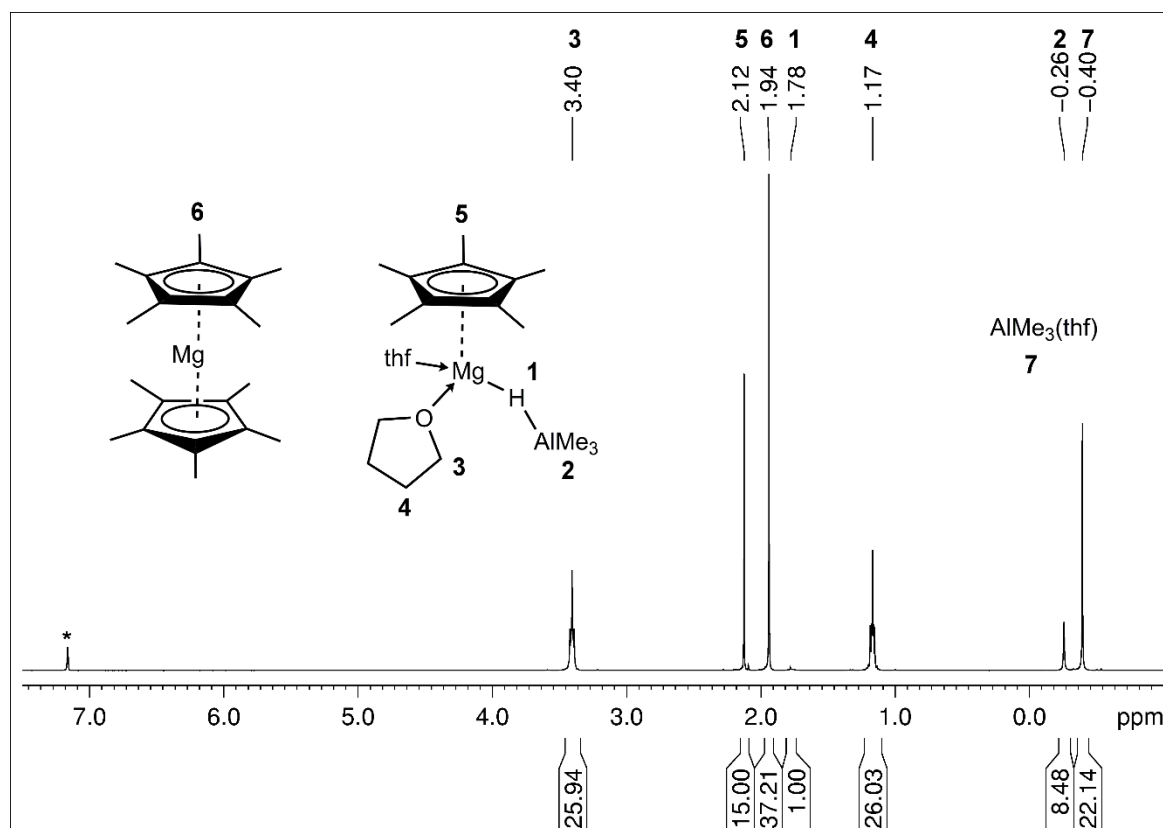


Figure S11. ^1H NMR spectrum (400 MHz) of **5** in benzene- d_6 after 16 h at 26 °C.

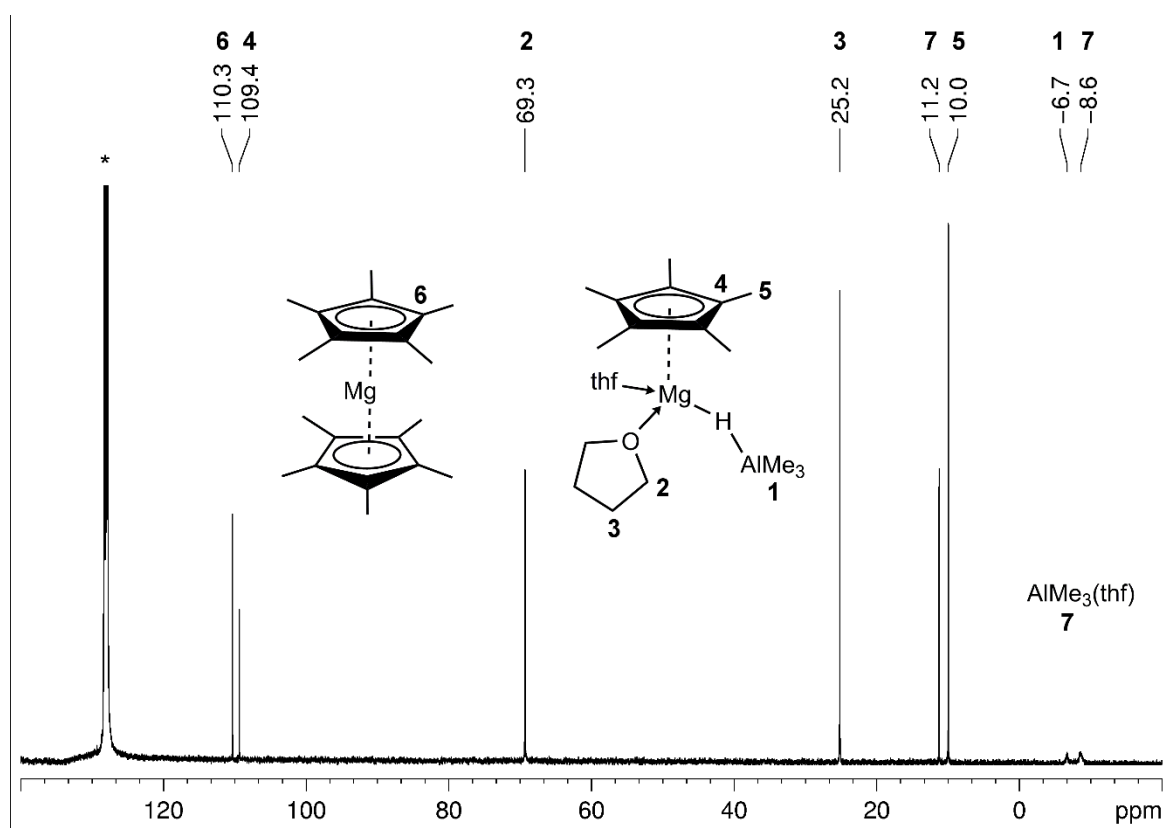


Figure S12. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum (101 MHz) of **5** in benzene- d_6 overnight at 26 °C.

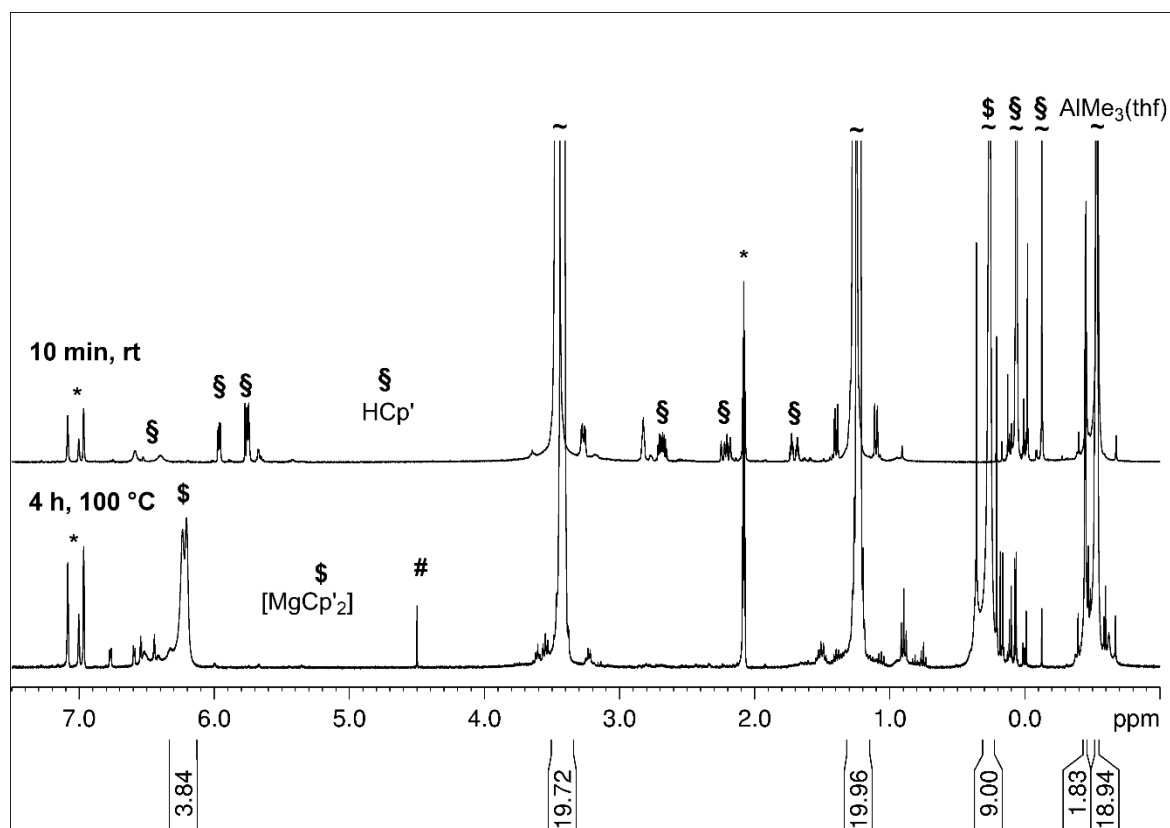


Figure S13. ¹H NMR spectra (400 MHz) of the protonolysis of **1** with HCp' in toluene-*d*₈ illustrating H₂ evolution (#).

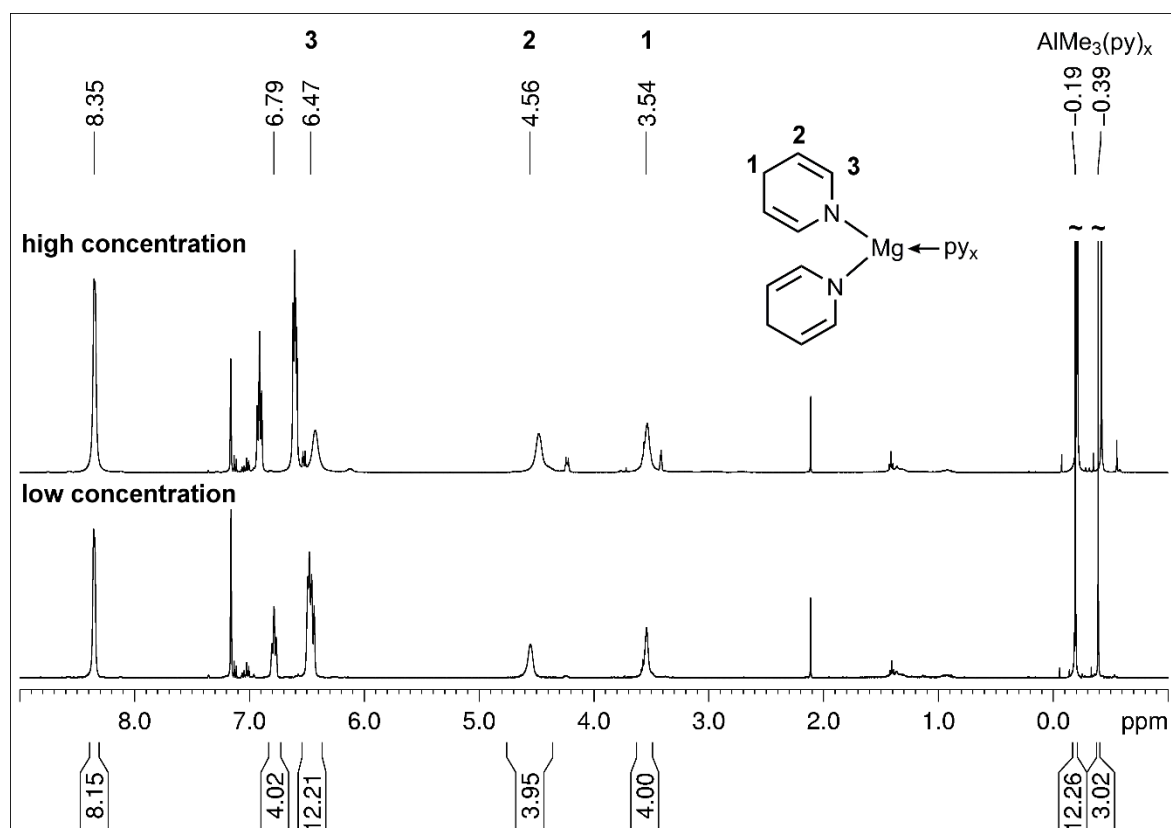


Figure S14. ¹H NMR spectrum (400 MHz) of [(py)_xMg(1,4-dihydropyridide)₂] in benzene-*d*₆ at 26 °C.

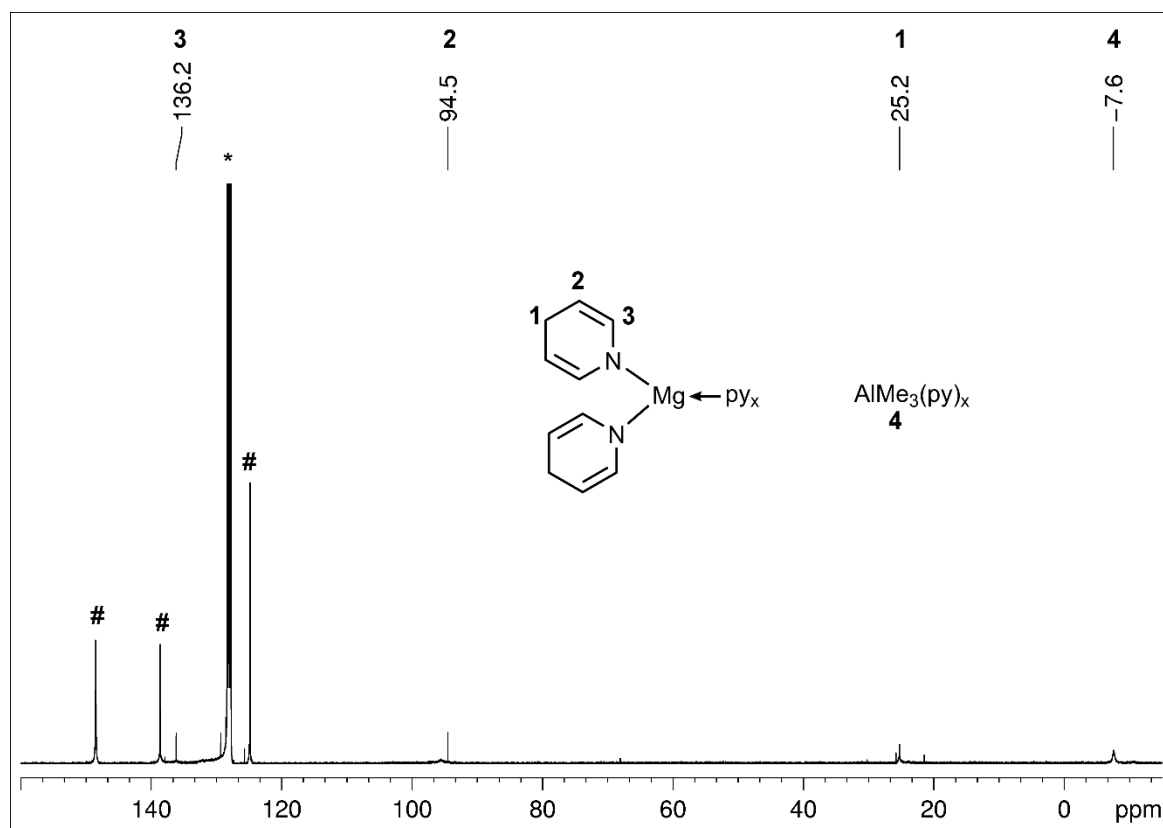


Figure S15. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum (101 MHz) of $[(\text{py})_x\text{Mg}(1,4\text{-dihydropyridide})_2]$ in benzene- d_6 at 26 °C. (# pyridine).

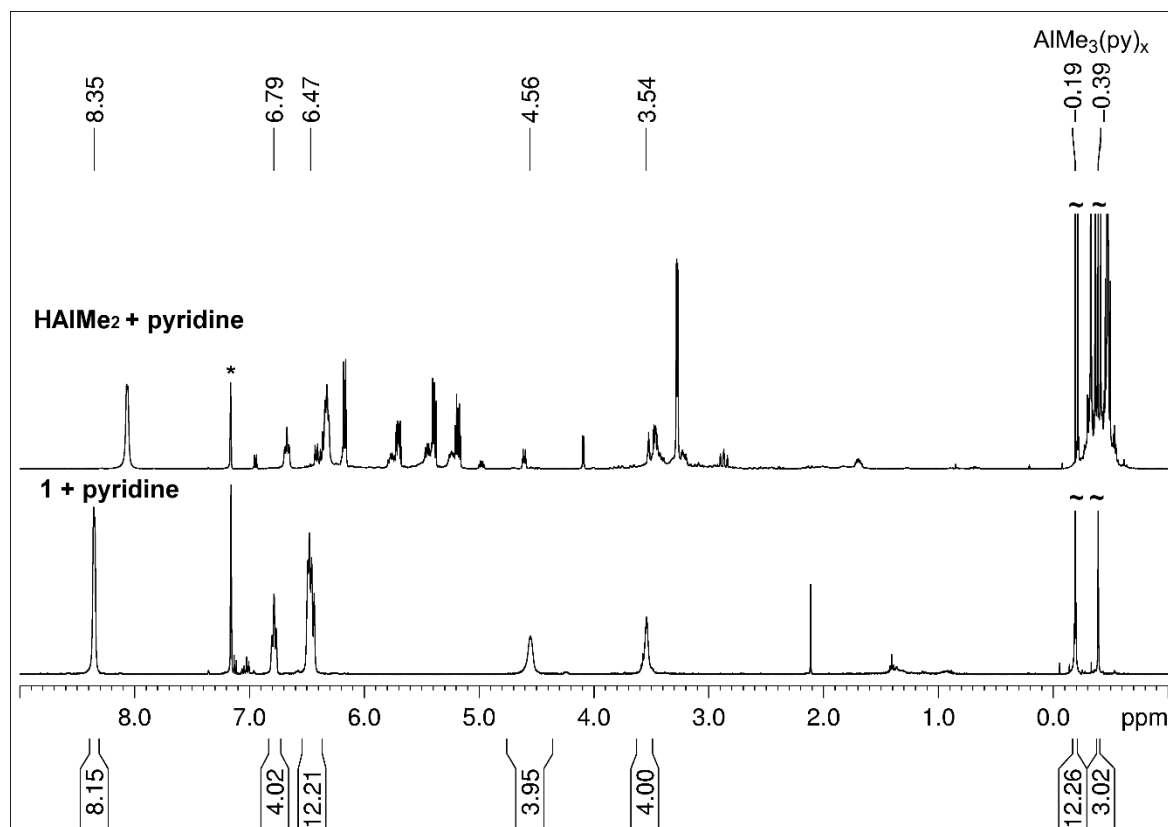


Figure S16. ^1H NMR spectra (400 MHz) of pyridine reduction by **1** and HAIme_2 in benzene- d_6 at 26 °C.

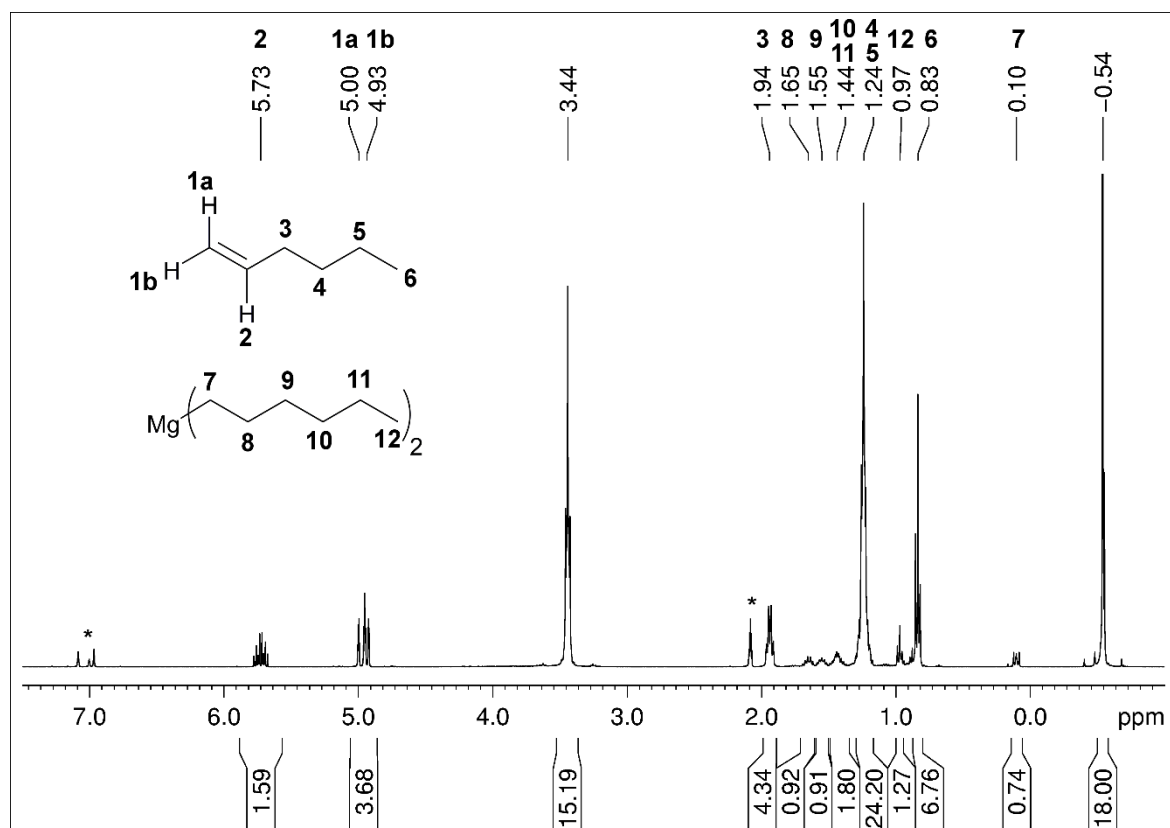


Figure S17. ^1H NMR spectrum (400 MHz) of the hydromagnesation of 1-hexene with **1** in toluene- d_8 at 26°C (40 h, 130 °C).

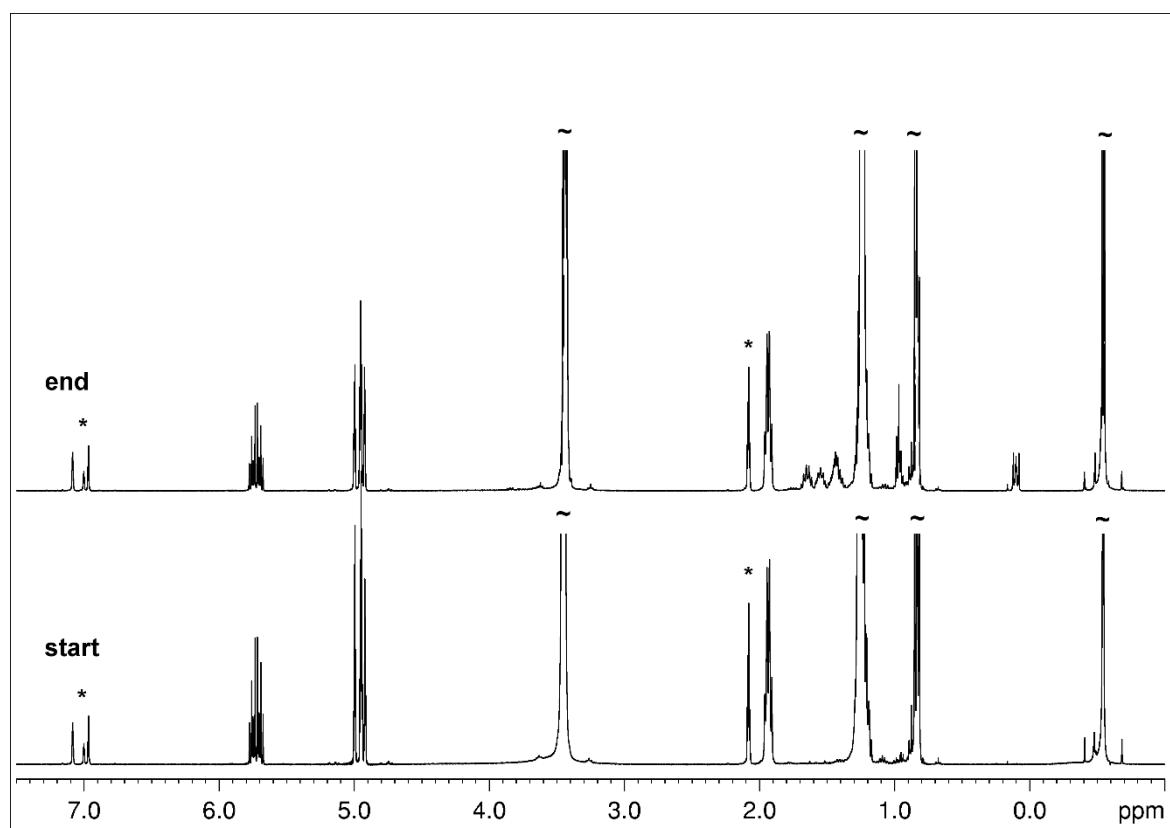


Figure S18. ^1H NMR spectra (400 MHz) of the hydromagnesation of 1-hexene with **1** in toluene- d_8 at 26 °C.

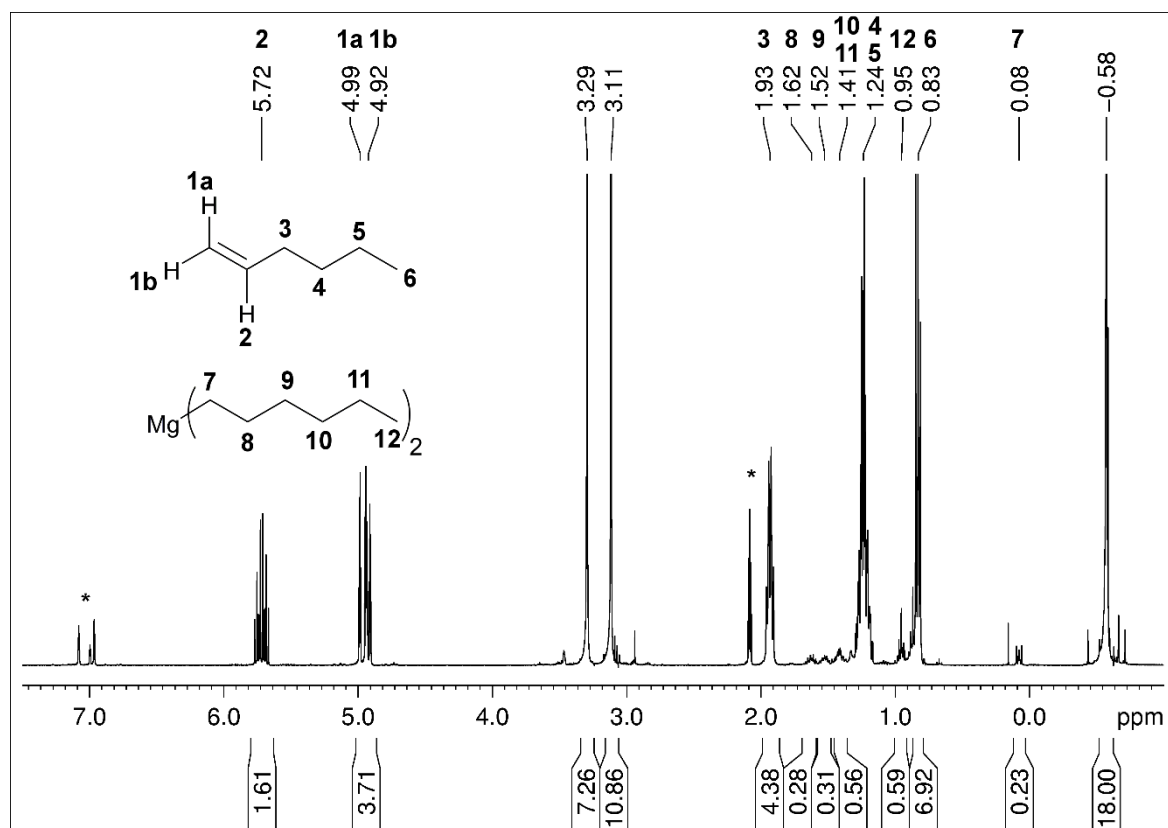


Figure S19. *In situ* ^1H NMR spectrum (400 MHz) of the hydromagnesation of 1-hexene with **3** in $\text{toluene-}d_8$ at $26\text{ }^\circ\text{C}$ after 16 h at $130\text{ }^\circ\text{C}$.

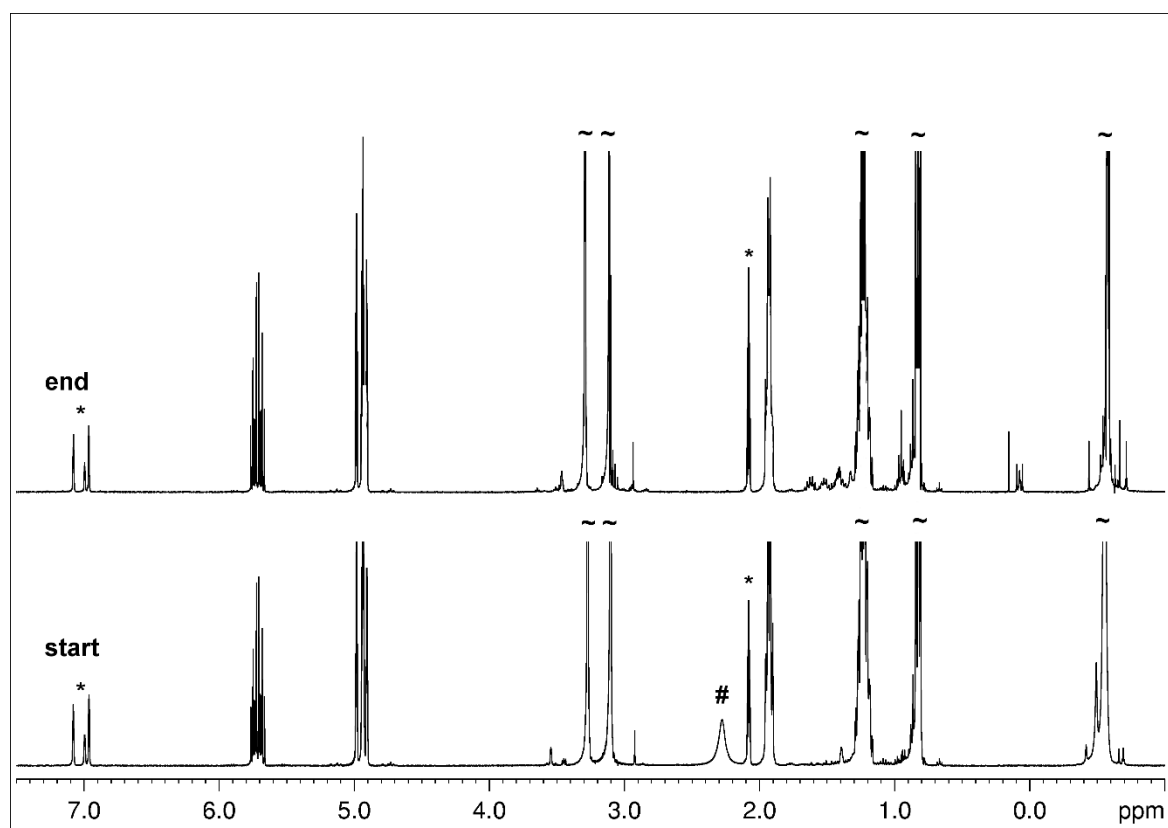


Figure S20. ^1H NMR spectra (400 MHz) of the hydromagnesation of 1-hexene with **3** in $\text{toluene-}d_8$ at $26\text{ }^\circ\text{C}$ illustrating the decrease of Mg-H-Al species (#) and the product peaks.

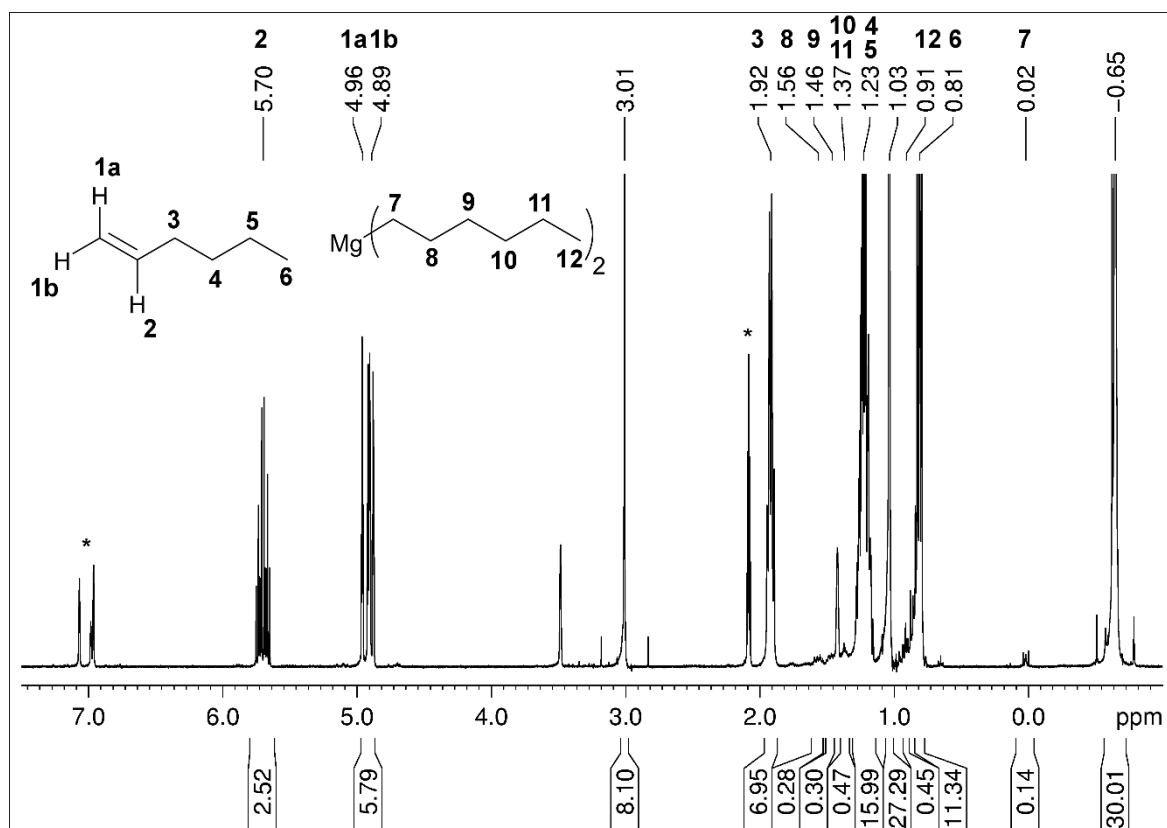


Figure S21. *In situ* ¹H NMR spectrum (400 MHz) of the hydromagnesation of 1-hexene with 4 in toluene-*d*₈ at 26 °C after 16 h at 130 °C.

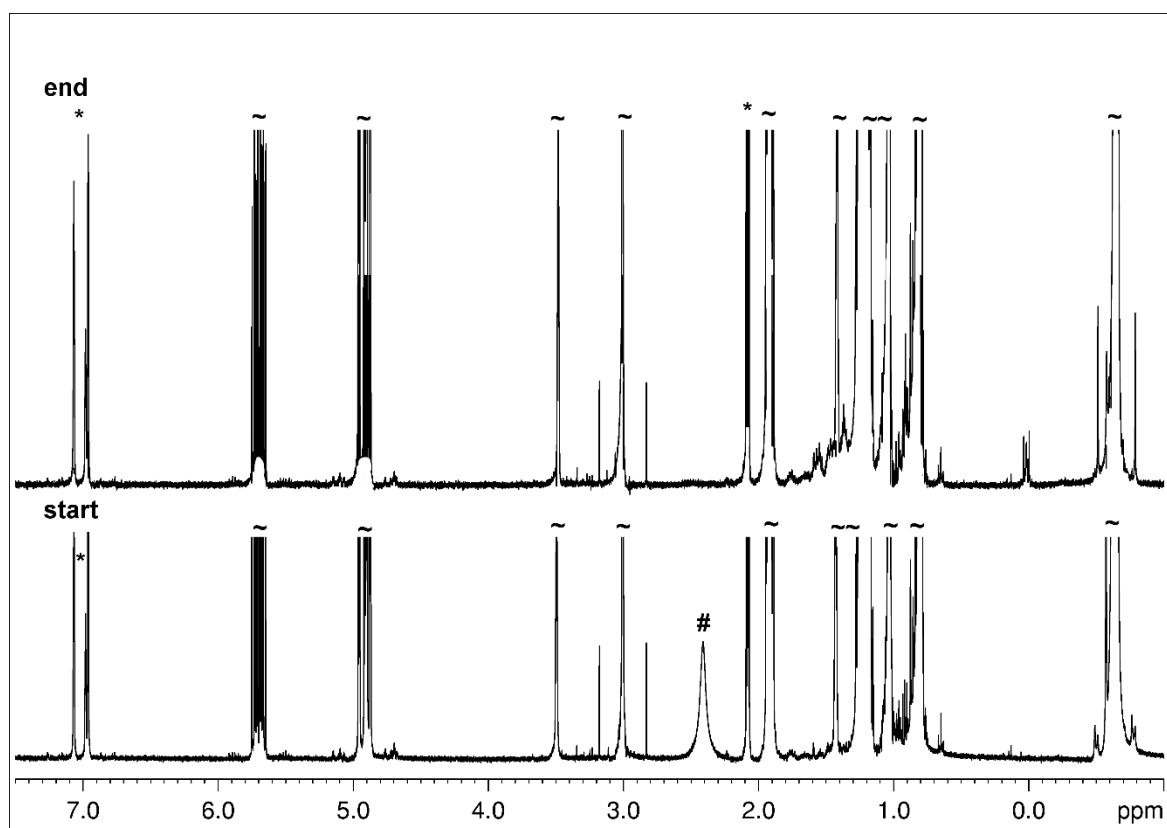


Figure S22. ¹H NMR spectra (400 MHz) of the hydromagnesation of 1-hexene with 4 in toluene-*d*₈ at 26 °C illustrating the decrease of Mg-H-Al species (#) and the product peaks.