

ELECTRONIC SUPPLEMENTARY INFORMATION

Reactions of Dicobalt Octacarbonyl with Dinucleating and Mononucleating Bis(imino)pyridine Ligands

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1. General Experimental Details

All reactions involving air-sensitive materials were carried out in a nitrogen-filled glovebox. 2,7-Di-*tert*-butyl-9,9-dimethyl-4,5-diaminoxanthene was synthesized using previously published procedures.¹ Cobalt carbonyl, ethyl propiolate, and 2,4,6-trimethylaniline were purchased from Sigma and used as received. 2,6-Pyridinedicarboxaldehyde was purchased from TCI and used as received. All non-deuterated solvents were purchased from Aldrich and were of HPLC grade. The non-deuterated solvents were purified using an MBraun solvent purification system. Acetonitrile- d_3 , benzene- d_6 , dichloromethane- d_2 , and toluene- d_8 were purchased from Cambridge Isotope Laboratories. All solvents were stored over 3 Å molecular sieves. Compounds were generally characterized by ^1H and ^{13}C NMR, high-resolution mass spectrometry and/or elemental analysis, IR spectroscopy, and UV-Vis spectroscopy. Selected compounds were characterized by X-ray crystallography. NMR spectra of the ligands and metal complexes were recorded at the Lumigen Instrument Centre (Wayne State University) on an Agilent DD2-600 MHz Spectrometer, a Varian VNMR5-500 MHz Spectrometer or an Agilent 400 MHz Spectrometer in CD_3CN , C_6D_6 , CD_2Cl_2 , or C_7D_8 at room temperature. Chemical shifts and coupling constants (J) were reported in parts per million (δ) and Hertz, respectively. High resolution mass spectra of the ligand and metal complexes were collected at the Lumigen Instrument Centre (Wayne State University) on a Thermofisher Scientific LTQ Orbitrap XL mass spectrometer. The MS survey scan was set from 200 – 2000. The resolution was set to 60000. In all cases, only one microscan was used in the analysis. Elemental analysis was performed under ambient air-free conditions by Midwest Microlab LLC. IR spectra of powdered samples were recorded on a Shimadzu IR Affinity-1 FT-IR Spectrometer outfitted with a MIRacle10 attenuated total reflectance accessory with a monolithic diamond crystal stage and pressure clamp. UV-Visible spectra were obtained on a Shimadzu UV-1800 spectrometer. X-ray structures were collected using Bruker Apex2 at the Lumigen Instrument Centre (Wayne State University). Quantitative analysis of arene formation was done by gas chromatography–mass spectrometry using Agilent 6890N spectrometer, Thermo TG5MS 30m $\times$ 0.32mm $\times$ 0.25 μm column, 7683 series injector and Agilent 5973 detector.

2. X-ray Crystallographic Details

Structures of complexes **1'**, **1**, **2**, **3**, and **4** were confirmed by X-ray structure determination. The crystals were mounted on a Bruker APEXII/Kappa three circle goniometer platform diffractometer equipped with an APEX-2 detector. A graphic monochromator was employed for wavelength selection of the Mo K α radiation ($\lambda = 0.71073$ Å). The data were processed and refined using the APEX2 software. Structures were solved by direct methods in SHELXS and refined by standard difference Fourier techniques in the SHELXTL program suite (6.10 v., Sheldrick G. M., and Siemens Industrial Automation, 2000). Hydrogen atoms were placed in calculated positions using the standard riding model and refined isotropically; all other atoms were refined anisotropically. Crystalline sample of **1** contained both needle-like crystals (**1'**) and plate-like crystals (**1**). The structure of **1'** was of low quality as indicated by its high R factors, and therefore only the overall connectivity and the cobalt-cobalt bonds are discussed. The structure of **4** contained three THF molecules per unit cell; one was found to be severely disordered. Our attempts to model the disorder met with only limited success. The structure of **1** contained one ether molecule per unit cell. Detailed crystal and structure refinement data are given in **Table S1**.

Table S1. X-ray crystallographic details for complexes **1**, **1'**, **2**, **3** and **4**.

complex	1 ×2C ₄ H ₁₀ O	1' ×2C ₄ H ₁₀ O	2a/2b	3 ×3C ₄ H ₈ O	4
formula	C ₇₈ H ₈₆ Co ₄ N ₆ O ₁₄	C ₇₈ H ₈₆ Co ₄ N ₆ O ₁₄	C ₆₀ H ₅₄ Co ₄ N ₆ O ₁₀	C ₆₆ H ₇₀ Co ₂ N ₆ O ₇	C ₇₆ H ₇₈ Co ₄ N ₁₀ O ₁₀
Fw, g/mol	1567.25	1567.24	1254.81	1177.14	1527.26
temperature	100(2)	100(2)	100(2)	100(2)	100(2)
cryst syst	monoclinic	monoclinic	monoclinic	monoclinic	triclinic
space group	C2/c	P2 ₁ /n	P2 ₁	P2 ₁ /c	P-1
color	red-purple	red-purple	purple	brown	purple
Z	4	4	2	4	2
a, Å	24.960(3)	20.268(2)	10.3616(10)	19.9171(10)	14.541(6)
b, Å	10.0567(12)	18.2285(17)	11.6941(11)	16.0470(8)	14.765(6)
c, Å	30.999(4)	21.168(2)	23.226(2)	18.9243(9)	20.123(6)
α, deg	90.00	90.00	90.00	90.00	101.991(7)
β, deg	102.7044(15)	90.683(6)	100.813(5)	99.304(3)	90.788(10)
γ, deg	90.00	90.00	90.00	90.00	118.813(5)
V, Å ³	7590.6(16)	7820.3(13)	2764.3(4)	5968.8(5)	3670(2)
d _{calcd} , g/cm ³	1.370	1.331	1.508	1.309	1.3819
μ, mm ⁻¹	0.926	0.898	1.245	0.614	0.953
2θ, deg	50.70	59.15	48.00	54.48	51.20
R ₁ ^a (all data)	0.0740	0.2285	0.0878	0.0934	0.0779
wR ₂ ^b (all data)	0.1264	0.3922	0.1019	0.1668	0.1335
R ₁ ^a [(I>2σ)]	0.0481	0.1360	0.0491	0.0583	0.0493
wR ₂ ^b [(I>2σ)]	0.1096	0.3441	0.0893	0.1454	0.1146
GOF (F ²)	1.014	1.197	1.0342	1.044	1.0188

3. Syntheses and Characterization of L^2 and 1-5

Preparation of L^2

A 25 mL methanol solution of 2,6-pyridinedicarboxaldehyde (0.383 g, 2.84 mmol) and acetic acid (0.1 mL) was added to a 25 mL methanol solution of 2,7-di-*tert*-butyl-9,9-dimethyl-9*H*-xanthene-4,5-diamine (1.000 g, 2.84 mmol). The reaction was refluxed overnight, producing a light yellow precipitate that was subsequently collected by filtration. The precipitate was washed with methanol and dried *in vacuo* to afford a light-yellow powder (1.087 g, 1.20 mmol, 85%). ^1H NMR (CD_2Cl_2 , 600 MHz) δ 8.36 (s, 4H, imino-**H**), 7.97 (d, $J = 7.7$ Hz, 4H, β -**H** on pyridine), 7.62 (t, $J = 7.7$ Hz, 2H, γ -**H** on pyridine), 7.30 (d, $J = 1.8$ Hz, 4H, ortho-**H** on xanthene), 6.92 (d, $J = 2.2$ Hz, 4H, para-**H** on xanthene), 1.75 (s, 6H, methyl-**H**), 1.69 (s, 6H, methyl-**H**), 1.34 (s, 36H, *tert*-butyl-**H**); $^{13}\text{C}\{^1\text{H}\}$ NMR (CD_2Cl_2 , 150 MHz) δ 164.29, 154.51, 146.26, 140.48, 139.73, 137.41, 131.52, 122.60, 119.26, 116.83, 35.74, 34.93, 32.08, 31.61, 30.58. HRMS (ESI) Calcd for $[\text{C}_{60}\text{H}_{66}\text{N}_6\text{O}_2\text{H}]^+$: 903.5325. Found: 903.5391.

Preparation of **1**

A 2 mL THF solution of L^2 (45.1 mg, 0.050 mmol) was added to a stirring 2 mL THF solution of $\text{Co}_2(\text{CO})_8$ (37.6 mg, 0.110 mmol). The solution color changed from dark-red to red-violet. The reaction was stirred overnight, upon which the volatiles were removed *in vacuo*. The resulting residue was washed with hexane to afford **1** as a red-violet solid (67.9 mg, 0.048 mmol, 96%). X-ray quality crystals were obtained from a concentrated ether solution at -35°C . ^1H NMR (CD_2Cl_2 , 600 MHz) δ 8.40 (s, 4H, imino-**H**), 8.11 (br s, 2H, γ -**H** on pyridine), 7.99 (br s, 4H, β -**H** on pyridine), 7.50 (d, $J = 2.2$ Hz, 4H, ortho-**H** on xanthene), 7.36 (d, $J = 2.2$ Hz, 4H, para-**H** on xanthene), 1.69 (s, 6H, methyl-**H**), 1.62 (s, 6H, methyl-**H**), 1.32 (s, 36H, *tert*-butyl-**H**); $^{13}\text{C}\{^1\text{H}\}$ NMR (CD_2Cl_2 , 150 MHz) δ 210.75, 203.81, 155.27, 150.09, 146.33, 141.61, 139.91, 131.21, 126.58, 123.89, 121.53, 120.73, 35.70, 35.27, 33.18, 31.53, 30.75. IR (Paratone): $\nu_{\text{CO}} = 2074\text{ cm}^{-1}$, 2026 cm^{-1} , 1990 cm^{-1} , 1890 cm^{-1} . $\lambda_{\text{max}}(\epsilon_{\text{M}})$: 719 (6510), 542 (13350), 487 (13010). Anal. Calcd for $\text{C}_{70}\text{H}_{66}\text{Co}_4\text{N}_6\text{O}_{12}$: C, 59.25; H, 4.69; N, 5.92. Found: C, 58.98; H, 4.90; N, 6.02.

Preparation of 2

A 2 mL THF solution of 1,1'-(pyridine-2,6-diyl)bis(*N*-mesitylmethanimine) (**L**¹, 37.0 mg, 0.100 mmol) was added to a stirring 2 mL THF solution of Co₂(CO)₈ (37.6 mg, 0.110 mmol). The solution color changed from dark-red to purple. The reaction was stirred overnight, upon which the volatiles were removed *in vacuo*. The resulting residue was washed with cold hexane to afford **2** as a purple solid (55.0 mg, 0.088 mmol, 88%). X-ray quality crystals were obtained from a concentrated pentane solution at -35 °C. ¹H NMR (C₆D₆, 600 MHz) δ 10.06 (v br s, 2H, imino-**H**), 7.28 (br s, 3H, γ-**H** and β-**H** on pyridine), 6.73 (s, 4H, meta-**H** on mesityl), 2.09 (s, 18H, methyl-**H** on mesityl); ¹³C{¹H} NMR (C₆D₆, 150 MHz) δ 151.34, 149.73, 149.28, 136.45, 130.60, 129.33, 128.25, 124.99, 122.82, 20.82, 18.48. IR (Paratone): ν_{CO} = 2029 cm⁻¹, 1957 cm⁻¹, 1880 cm⁻¹, 1802 cm⁻¹. λ_{max}(ε_M): 767 (3708), 711 (3672), 556 (5596), 495.5 (5896). Anal. Calcd for C₃₀H₂₇Co₂N₃O₅: C, 57.43; H, 4.34; N, 6.70. Found: C, 56.95; H, 4.42; N, 6.54.

Preparation of 3

A 2 mL THF solution of 1,1'-(pyridine-2,6-diyl)bis(*N*-mesitylmethanimine) (**L**¹, 80.0 mg, 0.217 mmol) was added to a stirring 2 mL THF solution of Co₂(CO)₈ (40.0 mg, 0.117 mmol). The solution color changed from dark-red to purple, then gradually to brown. The reaction was stirred overnight, upon which the volatiles were removed *in vacuo*. The resulting residue was washed with several times with hexane. Successive recrystallization from a concentrated ether solution at -35 °C afforded brown X-ray quality crystals of **3** (59.7 mg, 0.062 mmol, 57%). ¹H NMR (CD₂Cl₂, 400 MHz) δ 21.36, 6.67, 4.63, 1.19. μ_{eff} = 2.4 ± 0.2 μ_B. IR (Paratone): ν_{CO} = 1867 cm⁻¹. λ_{max}(ε_M): 762 (3077), 532.5 (3495), 482(4582).

Preparation of 4 from 1

1 (42.6 mg, 0.030 mmol) was treated with 2 mL of acetonitrile (or CD₃CN), producing a dark-green solution. The reaction was stirred for 30 minutes, upon which the volatiles were removed *in vacuo* to afford **4** as a dark-green solid (45.2 mg, 0.030 mmol, 99%). X-ray quality crystals were obtained from acetonitrile/ether vapor diffusion at -35 °C. ¹H NMR (CD₃CN, 600 MHz) δ 9.83 (s, 4H, imino-**H**), 8.75 (t, *J* = 7.8 Hz, 2H, γ-**H** on pyridine), 7.71 (d, *J* = 7.8 Hz, 4H, β-**H** on pyridine), 7.63 (d, *J* = 1.8 Hz, 4H, ortho-**H** on xanthene), 7.60 (d, *J* = 1.8 Hz, 4H, para-**H** on xanthene), 1.96 (s, 12H, acetonitrile-**H**), 1.92 (s, 6H, methyl-**H**), 1.62 (s, 6H, methyl-**H**), 1.34

(s, 36H, *tert*-butyl-**H**); $^{13}\text{C}\{^1\text{H}\}$ NMR (CD_3CN , 150 MHz) δ 157.63, 150.94, 147.49, 142.97, 141.30, 133.05, 128.66, 122.39, 121.09, 119.47, 36.68, 35.46, 34.38, 31.78, 26.85. IR (Paratone): $\nu_{\text{CO}} = 1868\text{ cm}^{-1}$. $\lambda_{\text{max}}(\epsilon_{\text{M}})$: 857.5 (2549), 676.5 (1828), 438.5 (7067).

Preparation of 5 from 2

2 (31.4 mg, 0.050 mmol) was treated with 2 mL of acetonitrile (or CD_3CN), producing a dark-green solution. The reaction was stirred for 30 minutes, upon which the volatiles were removed *in vacuo* to afford **5** as a dark-green solid (33.8 mg, 0.030 mmol, 99%). ^1H NMR (CD_3CN , 600 MHz) δ 9.73 (s, 2H, imino-**H**), 8.81 (t, $J = 6.4$ Hz, 1H, γ -**H** on pyridine), 7.95 (d, $J = 7.0$ Hz, 2H, β -**H** on pyridine), 6.98 (s, 4H, meta-**H** on mesityl), 2.30 (s, 6H, para-methyl-**H** on mesityl), 1.94 (s, 12H, ortho-methyl-**H** on mesityl); $^{13}\text{C}\{^1\text{H}\}$ NMR (CD_3CN , 150 MHz) δ 157.92, 152.21, 151.19, 136.73, 129.99, 129.75, 128.28, 122.49, 21.03, 18.25. IR (Paratone): $\nu_{\text{CO}} = 1868\text{ cm}^{-1}$. $\lambda_{\text{max}}(\epsilon_{\text{M}})$: 855 (2176), 630.5 (1104), 443.5 (6680).

4. NMR Spectra

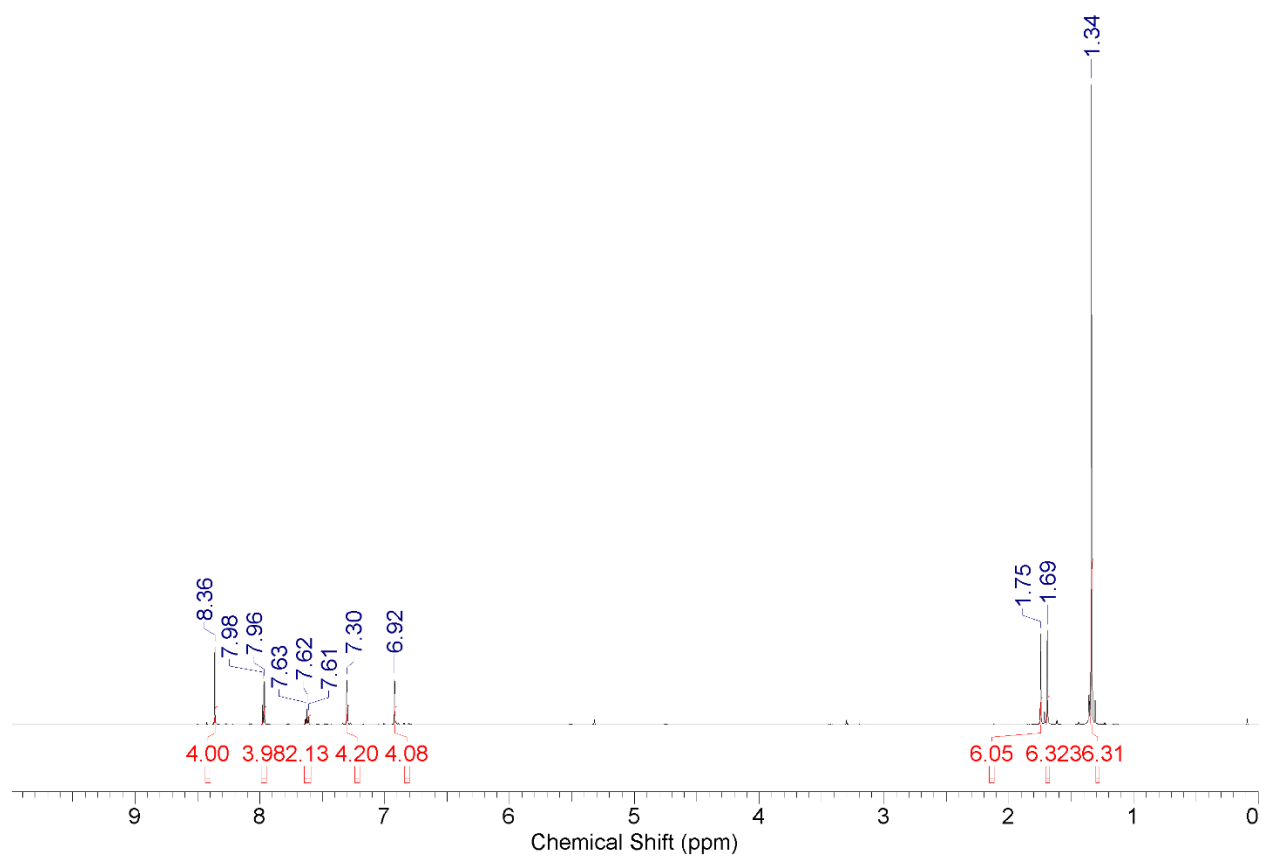


Figure S1: ^1H spectrum of L^2 (CD_2Cl_2 , 600 MHz).

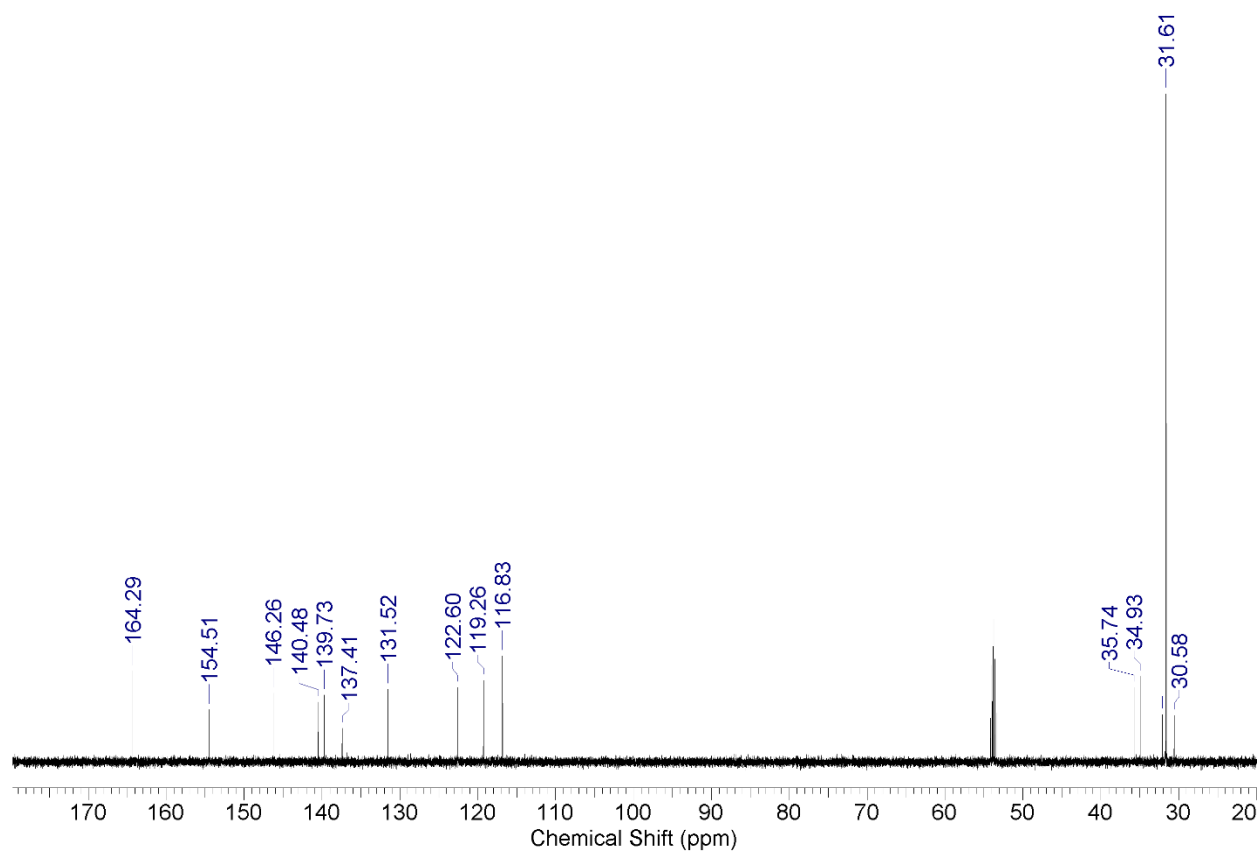


Figure S2: ¹³C spectrum of **L**² (CD₂Cl₂, 150 MHz).

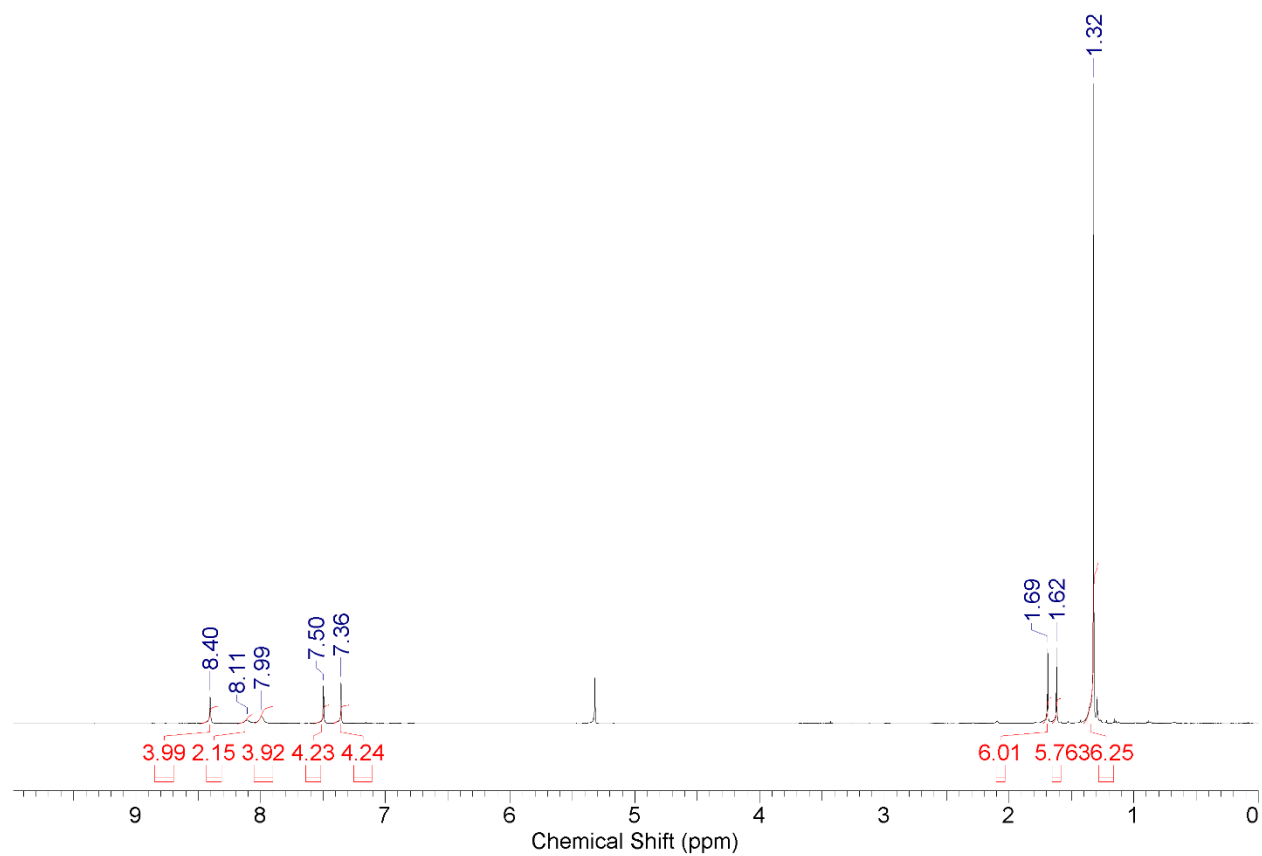


Figure S3: ^1H spectrum of **1** (CD_2Cl_2 , 600 MHz).

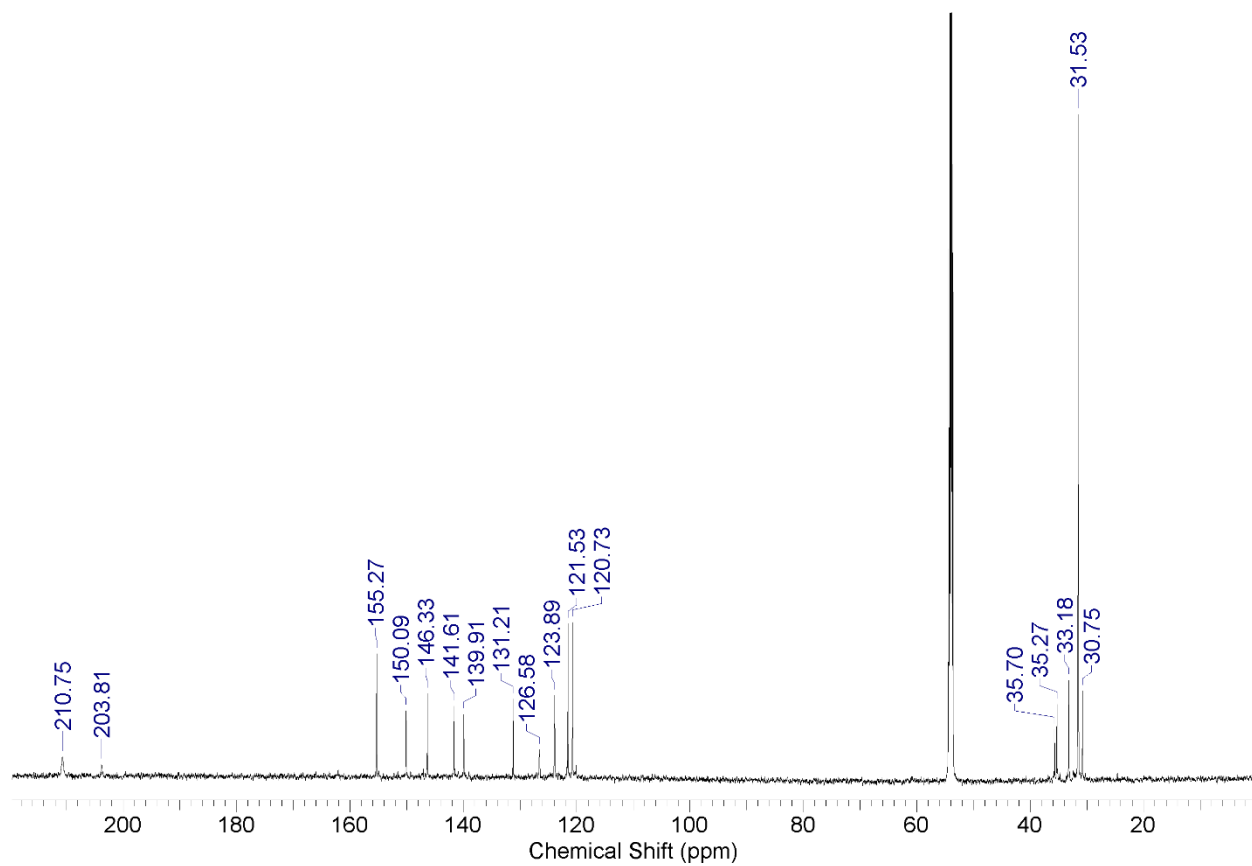


Figure S4: ¹³C spectrum of **1** (CD₂Cl₂, 150 MHz).

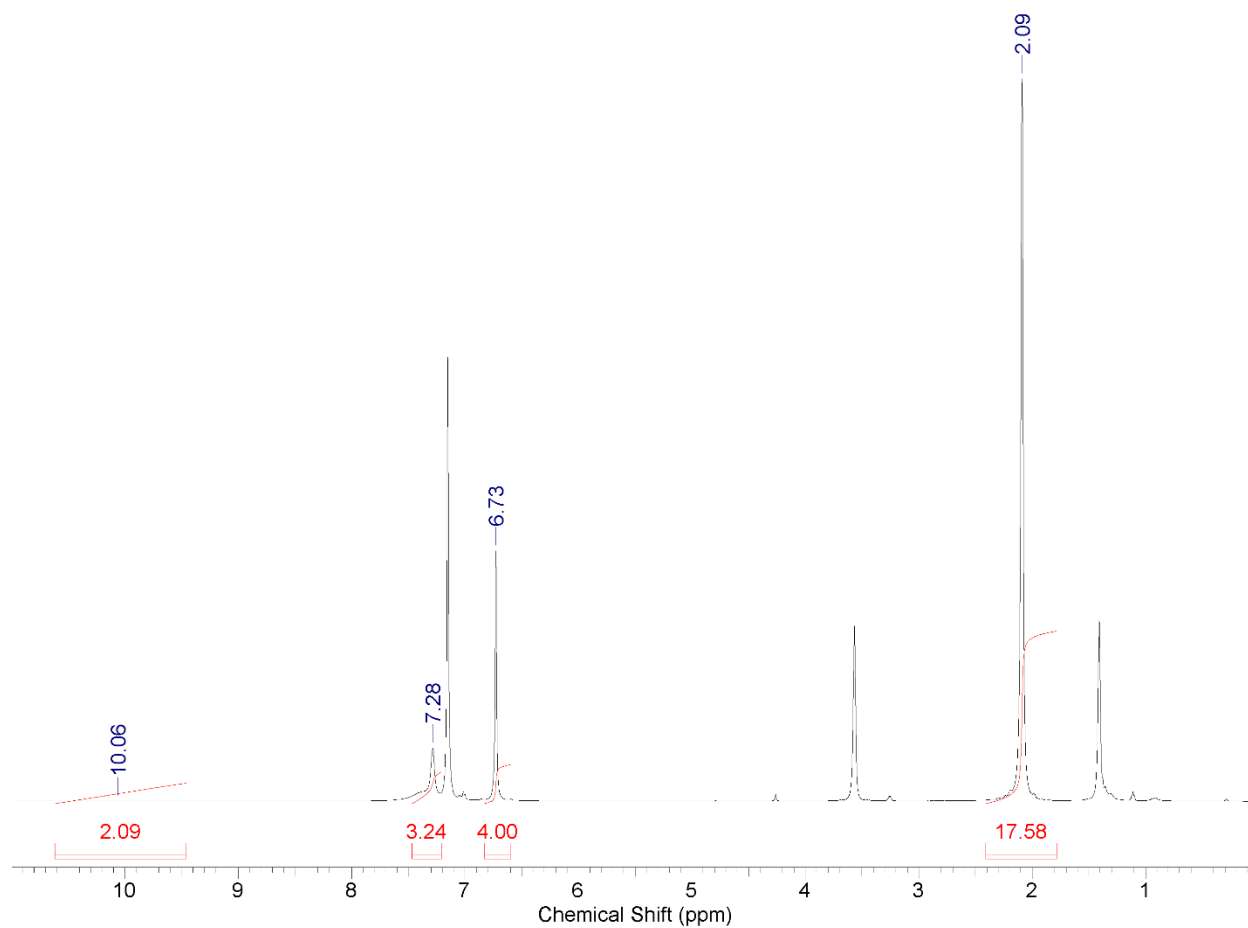


Figure S5: ^1H spectrum of **2** (C_6D_6 , 600 MHz). The unlabeled peaks correspond to THF.

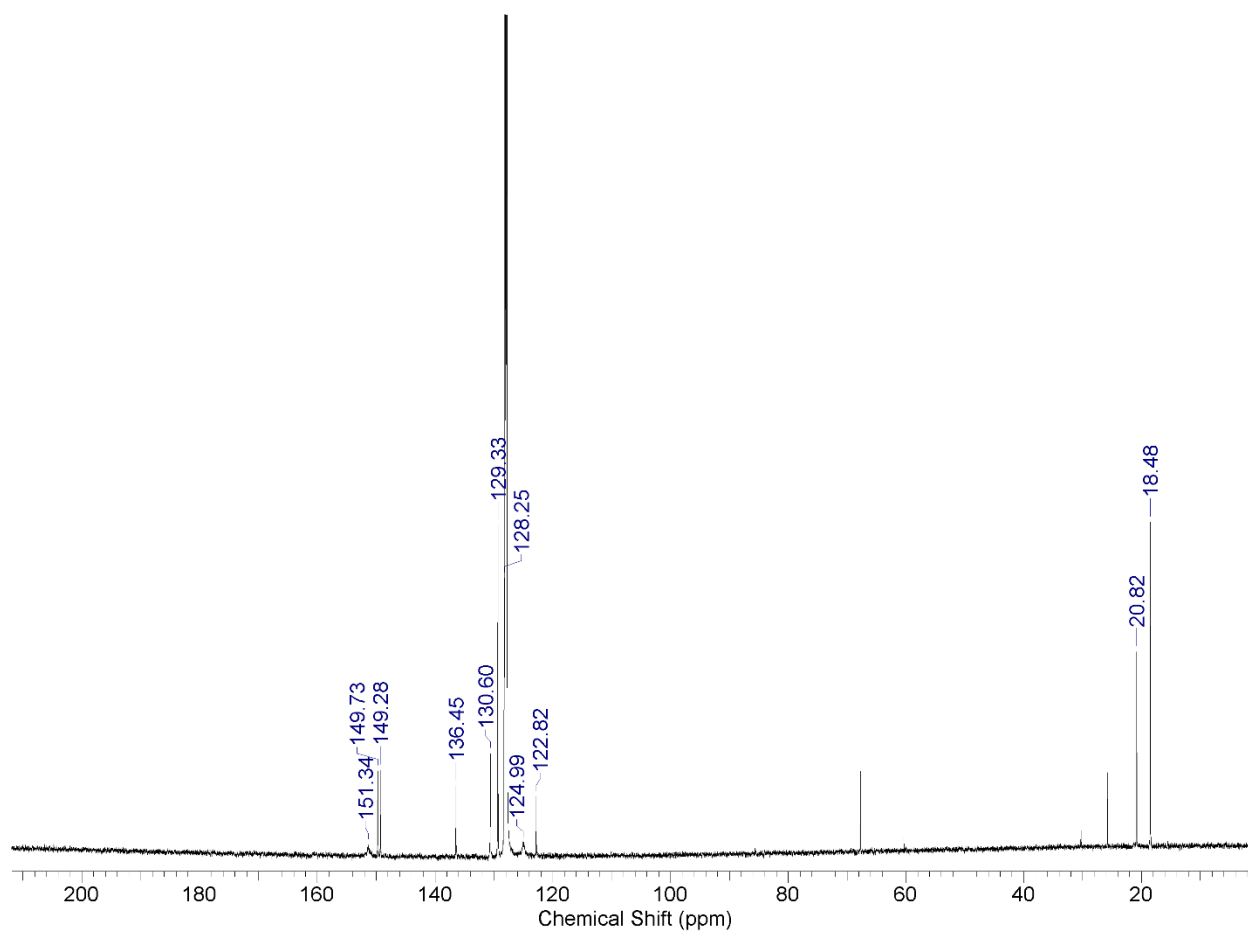


Figure S6: ^{13}C spectrum of 2 (C_6D_6 , 150 MHz). The unlabeled peaks correspond to THF.

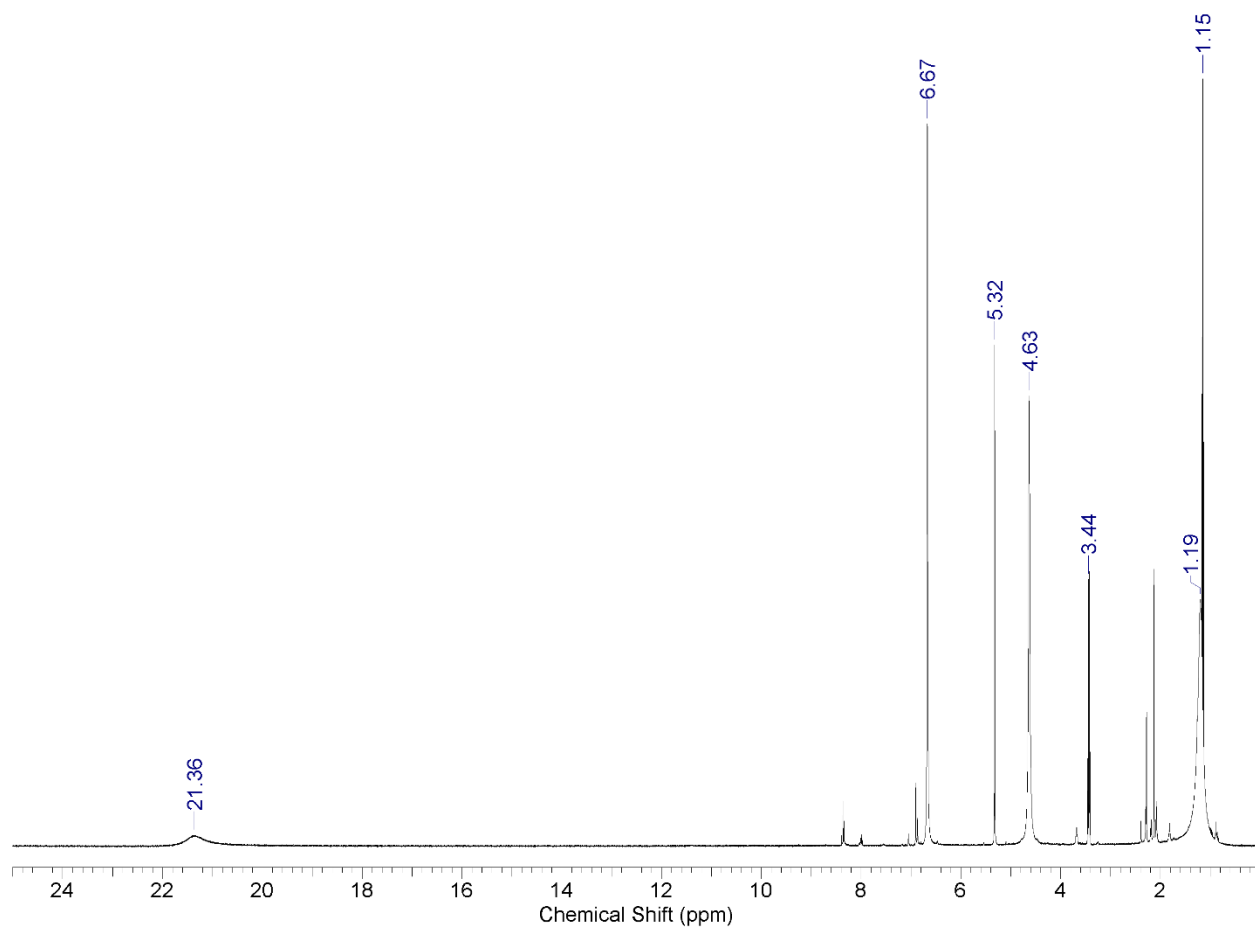


Figure S7: ^1H spectrum of **3** (CD_2Cl_2 , 400 MHz).

The peaks at 3.44 and 1.15 correspond to ether. The other unlabeled peaks correspond to small quantities of free ligand and **2**.

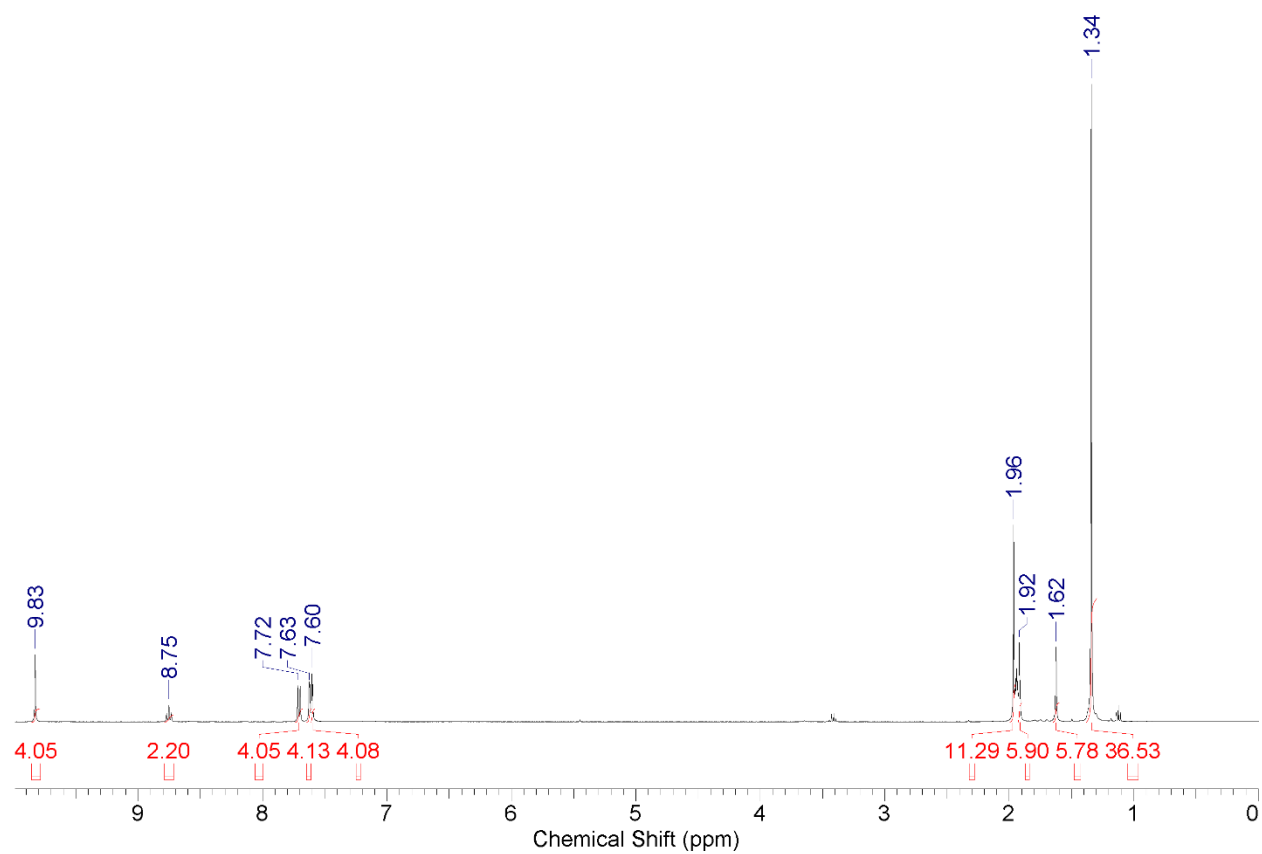


Figure S8: ¹H spectrum of **4** (CD₃CN, 600 MHz).

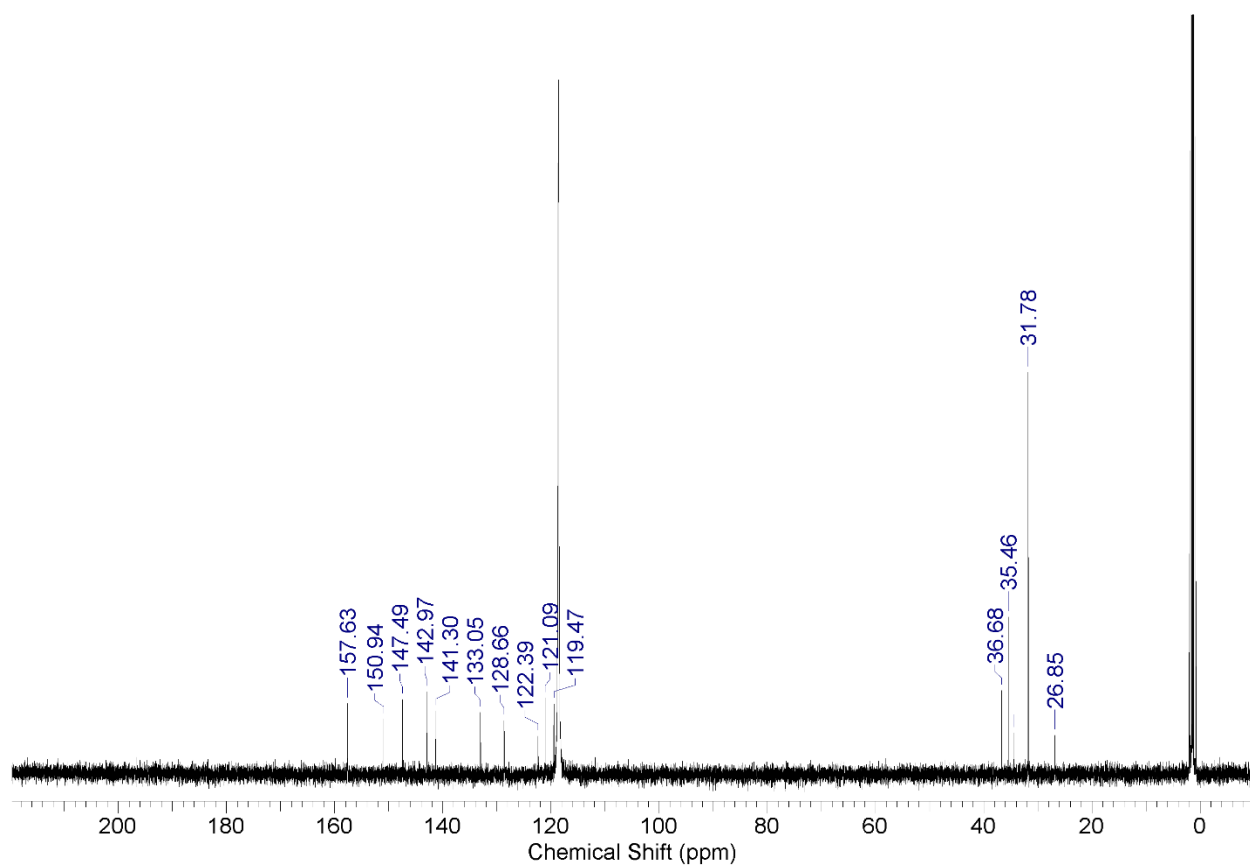


Figure S9: ¹³C spectrum of **4** (CD₃CN, 150 MHz).

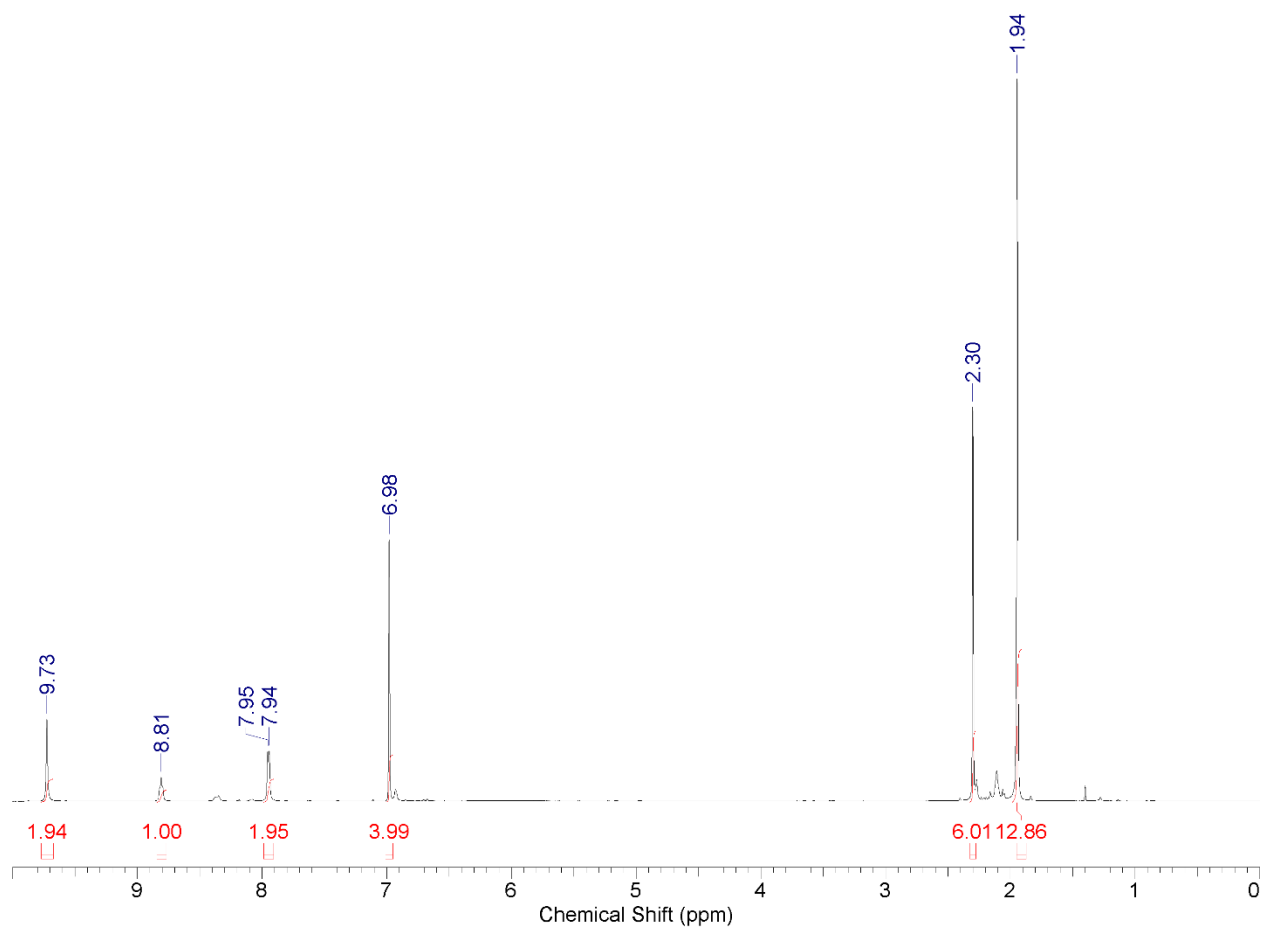


Figure S10: ^1H spectrum of **5** (CD_3CN , 600 MHz).

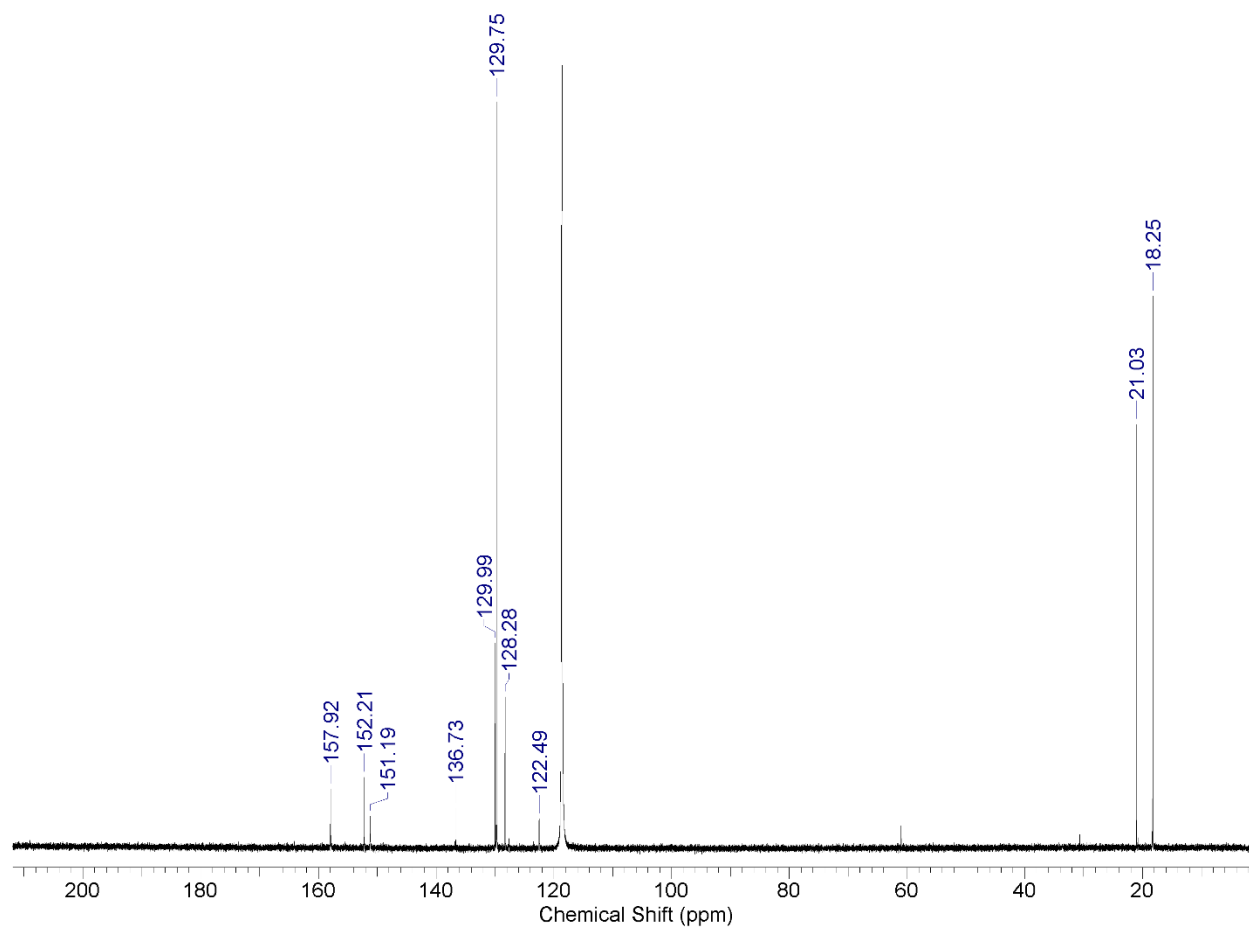


Figure S11: ¹³C spectrum of 5 (CD₃CN, 150 MHz).

5. UV-Vis Spectra

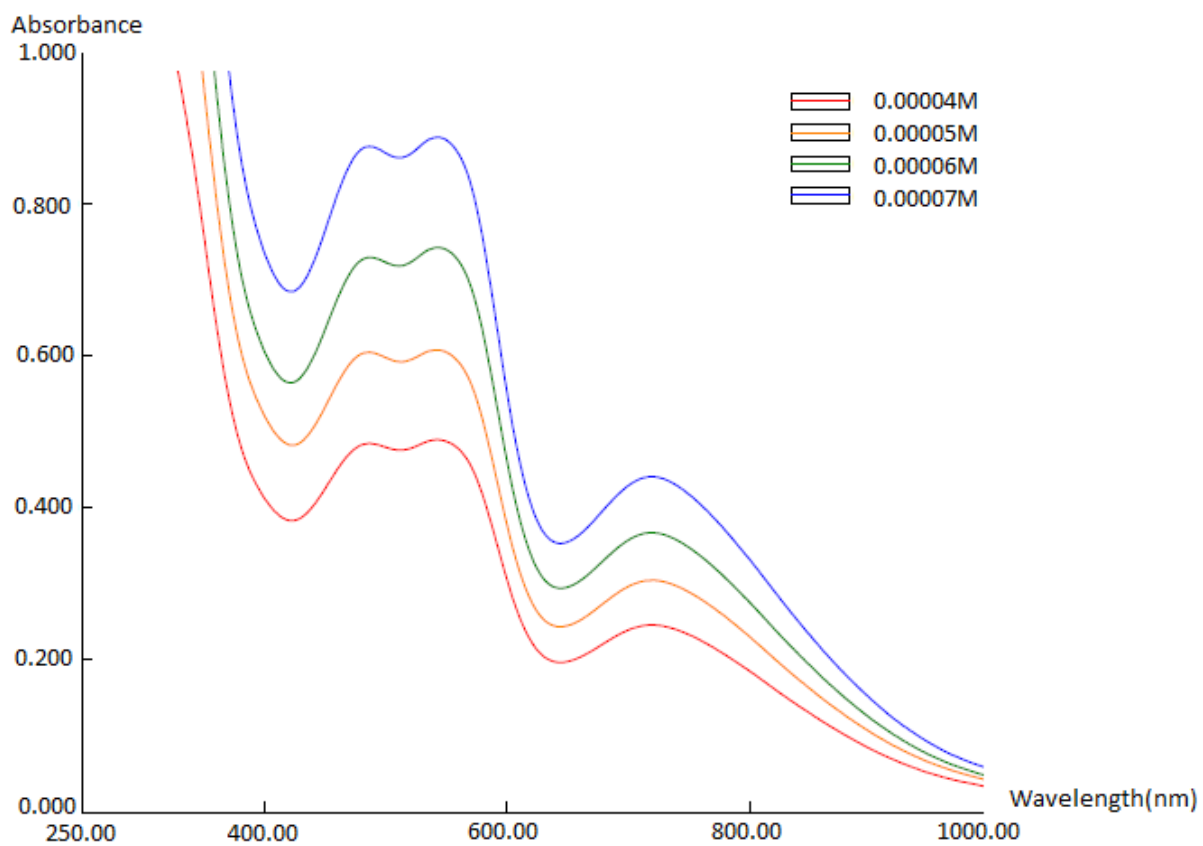


Figure S12: UV-Vis spectrum of **1** at four different concentrations.

$\lambda_{\max}(\epsilon_M)$: 719 (6510), 542 (13350), 487 (13010).

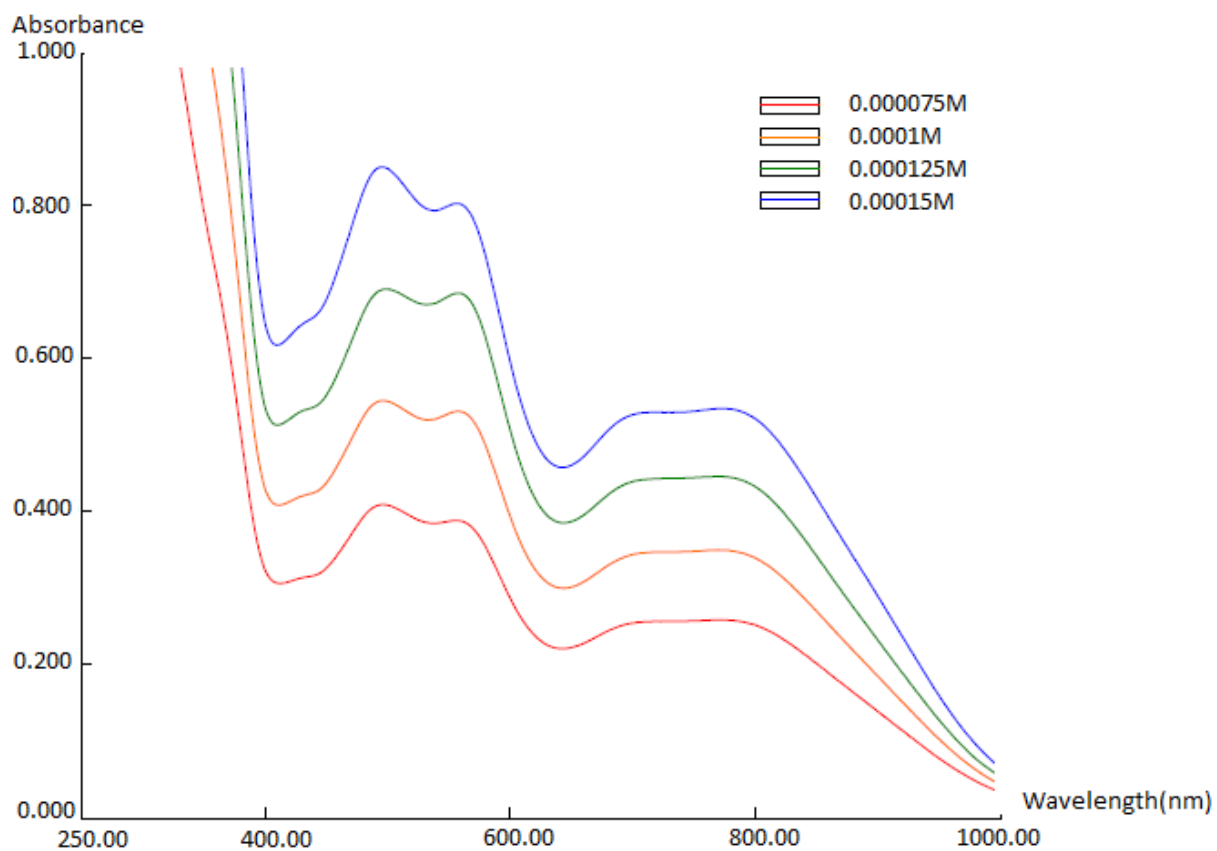


Figure S13: UV-Vis spectrum of **2** at four different concentrations.

$\lambda_{\max}(\epsilon_M)$: 767 (3708), 711 (3672), 556 (5596), 495.5 (5896).

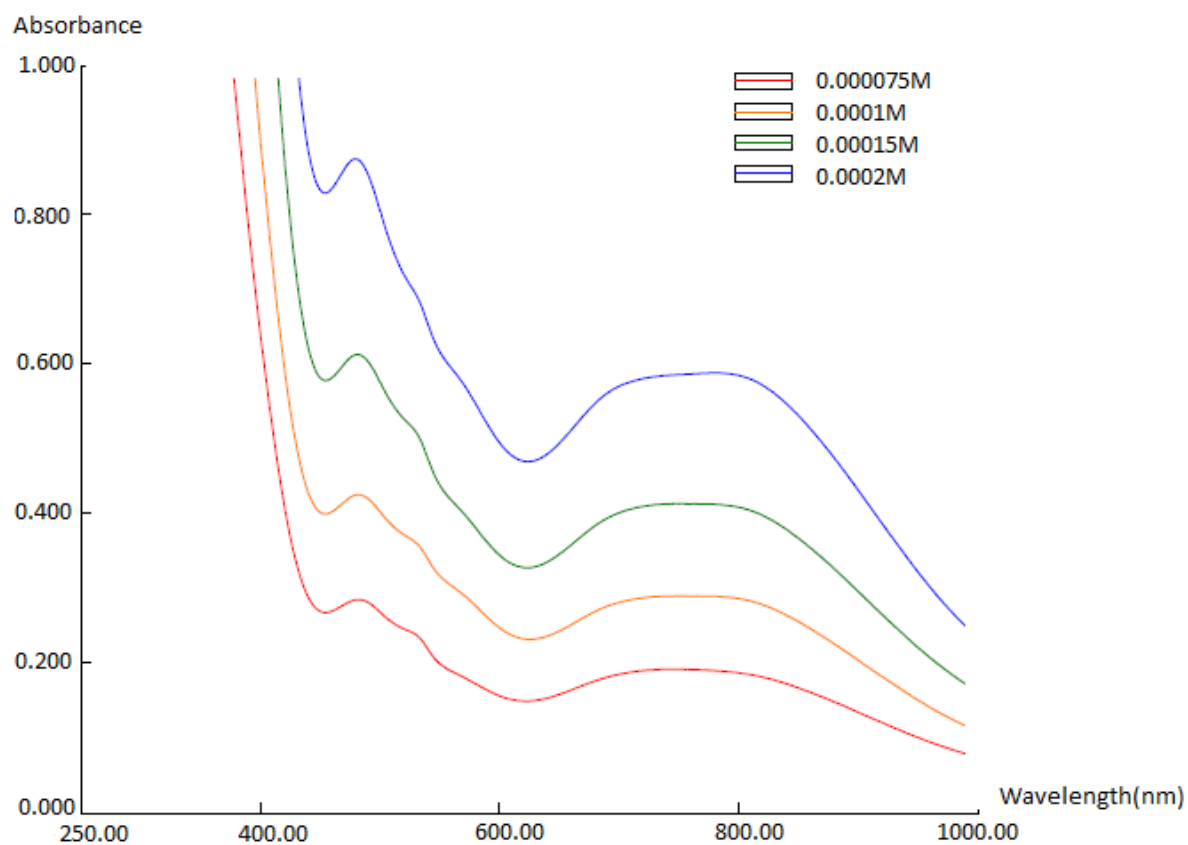


Figure S14: UV-Vis spectrum of **3** at four different concentrations.

$\lambda_{\max}(\epsilon_M)$: 762 (3080), 532.5 (3495), 482(4580).

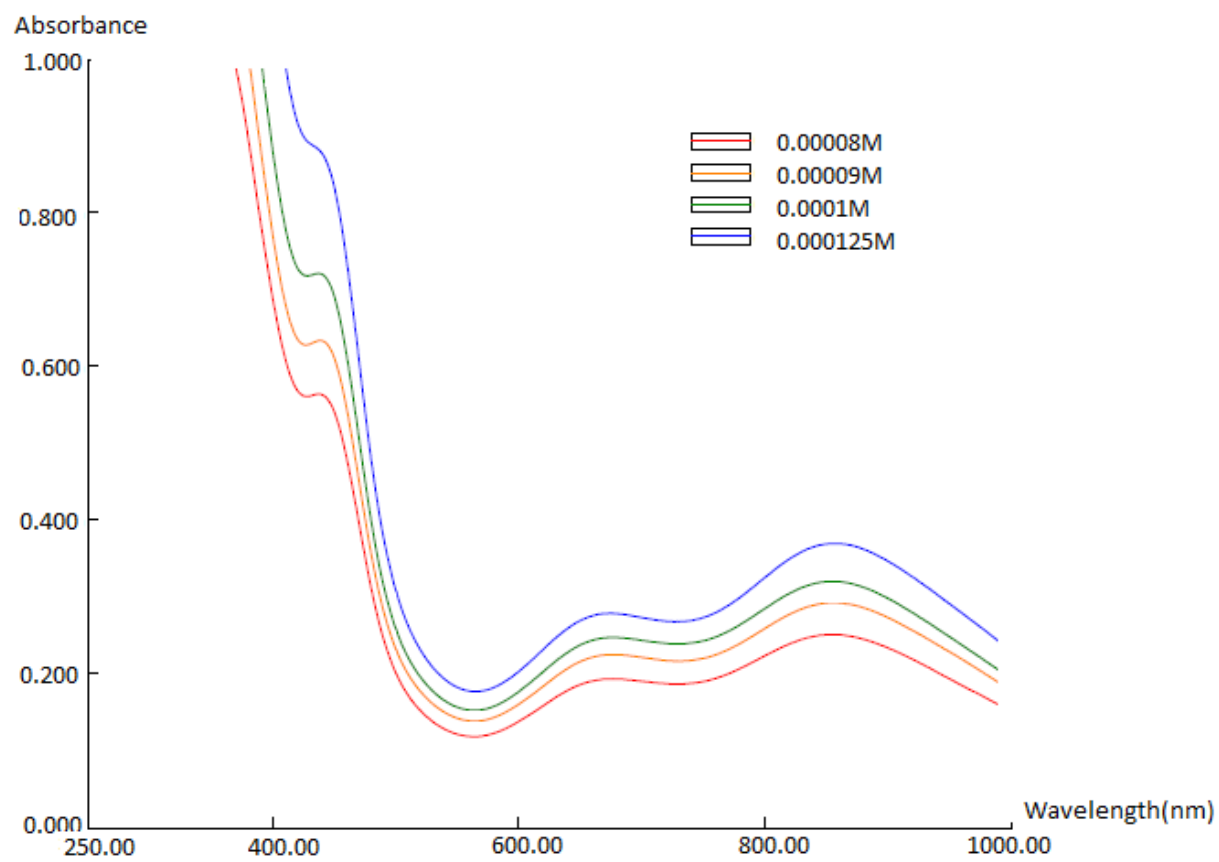


Figure S15: UV-Vis spectrum of **4** at four different concentrations.

$\lambda_{\max}(\epsilon_M)$: 858 (2550), 677 (1830), 439 (7070).

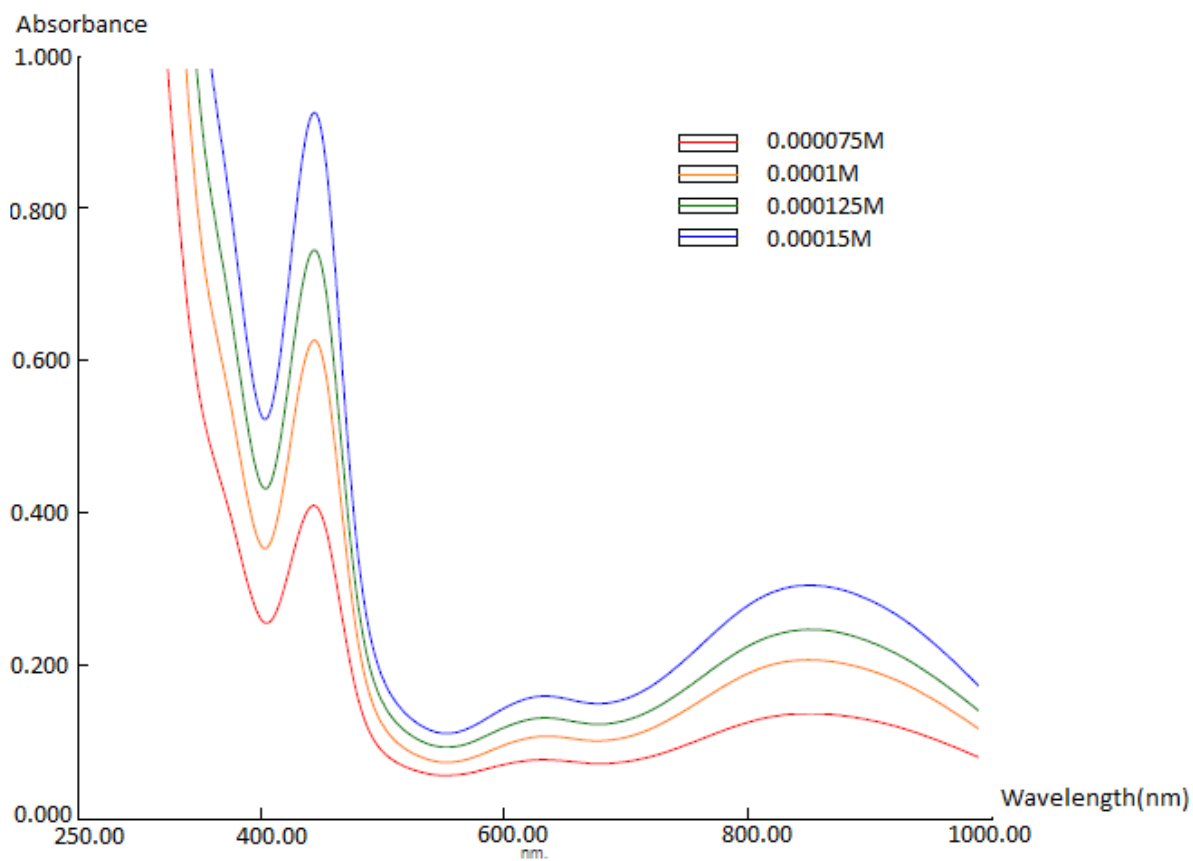


Figure S16: UV-Vis spectrum of 5 at four different concentrations.

$\lambda_{\max}(\epsilon_M)$: 855 (2180), 630.5 (1100), 443.5 (6680).

6. IR Spectra

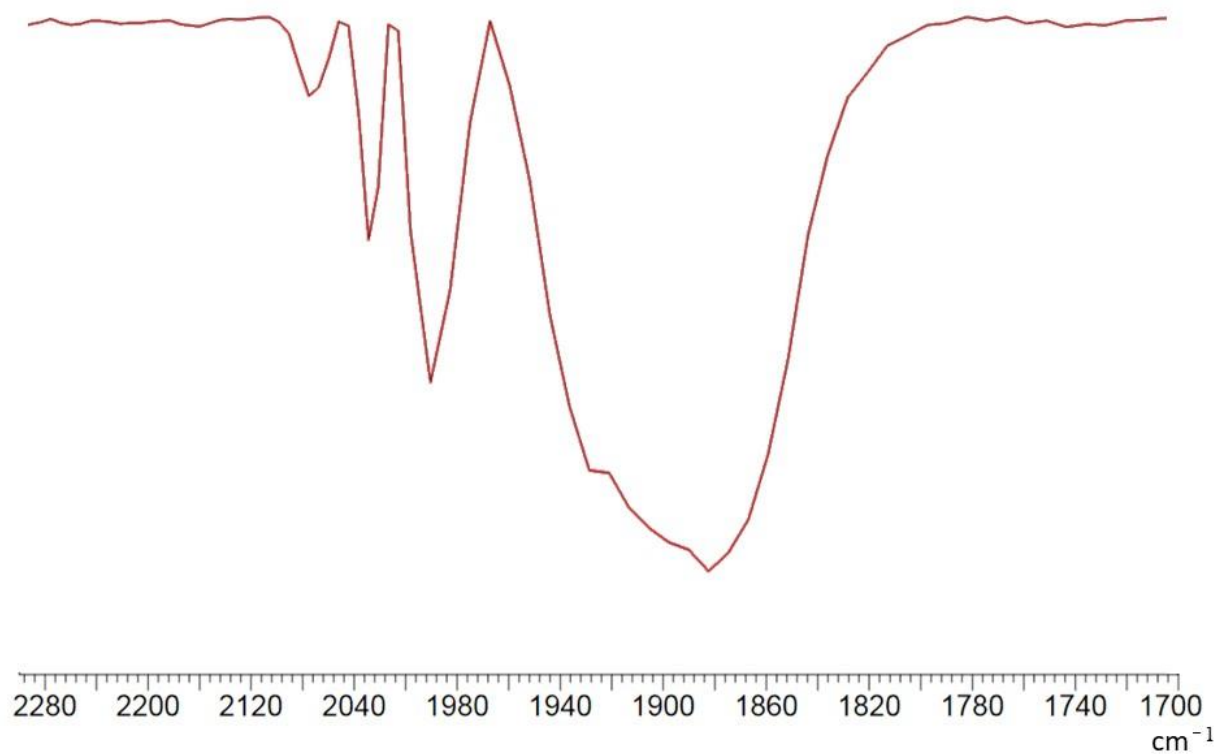


Figure S17: IR spectrum of **1** in the CO region. $\nu_{\text{C=O}}$: 2074, 2026, 1990, and 1890 cm^{-1} .

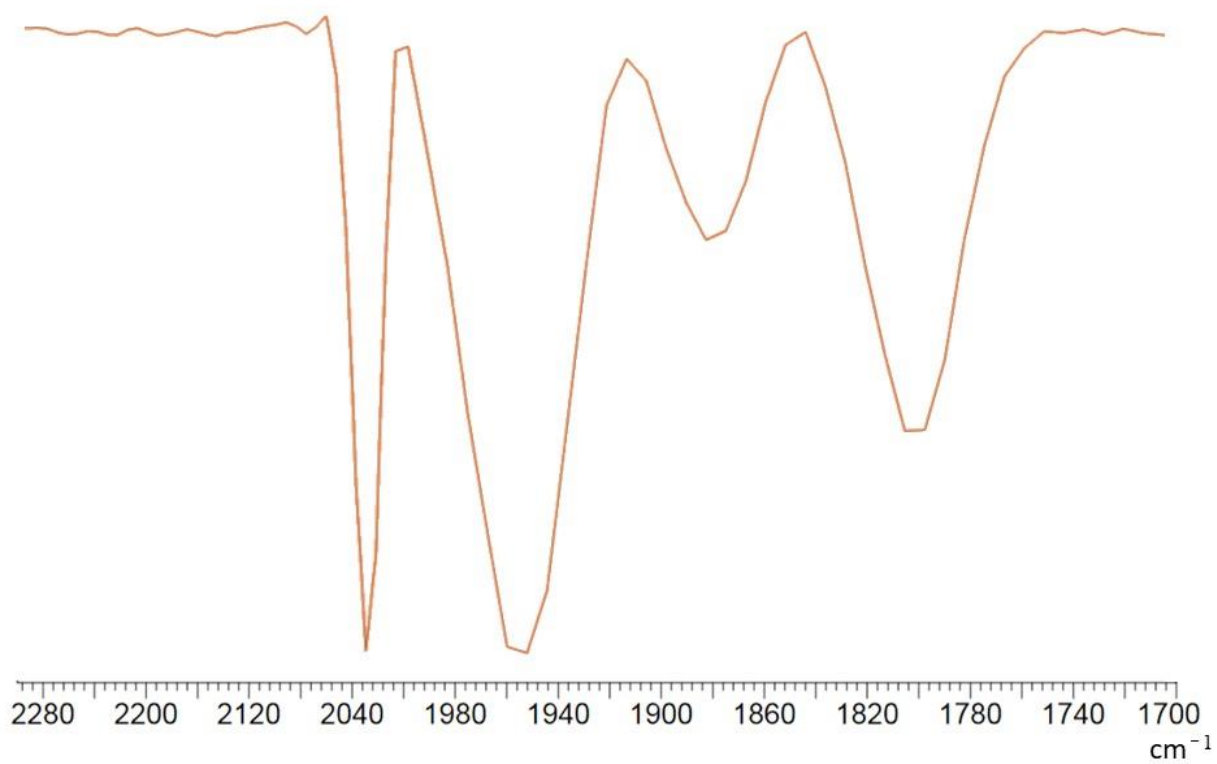


Figure S18: IR spectrum of **2** in the CO region. $\nu_{\text{C}\equiv\text{O}}$: 2029, 1957, 1880, and 1802 cm⁻¹.

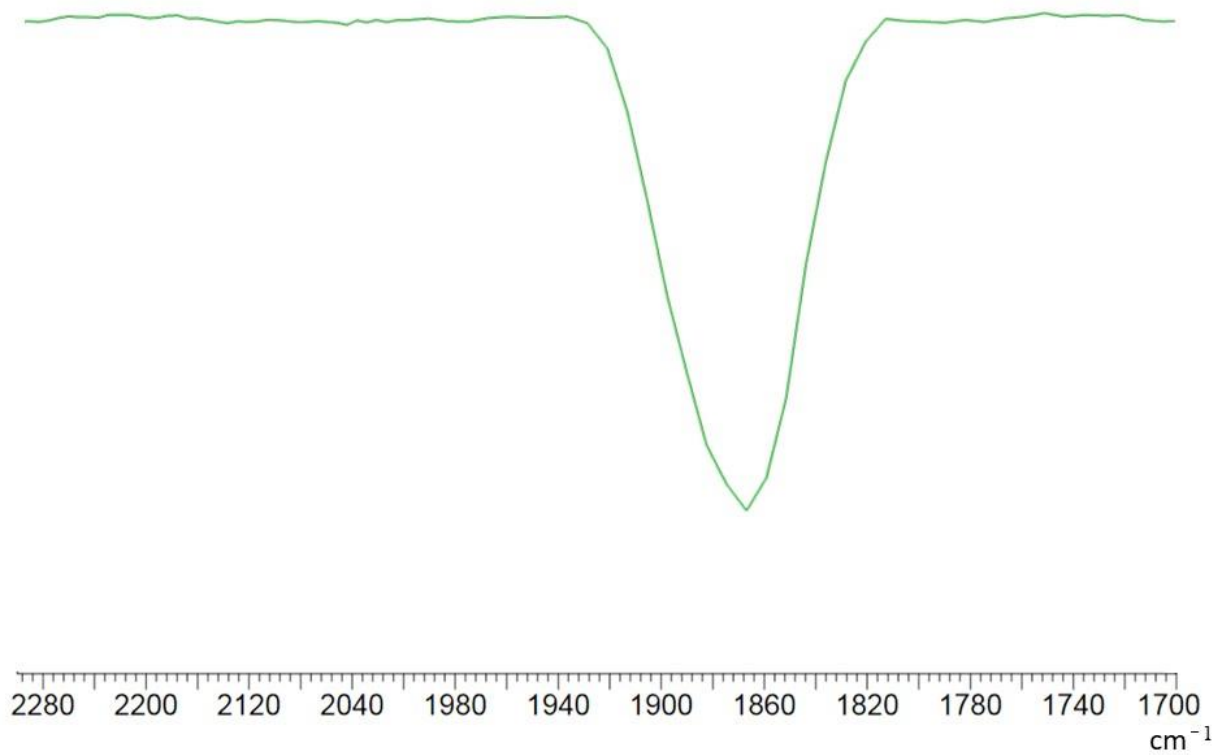


Figure S19: IR spectrum of **3** in the CO region. $\nu_{\text{C=O}}$: 1867 cm⁻¹.

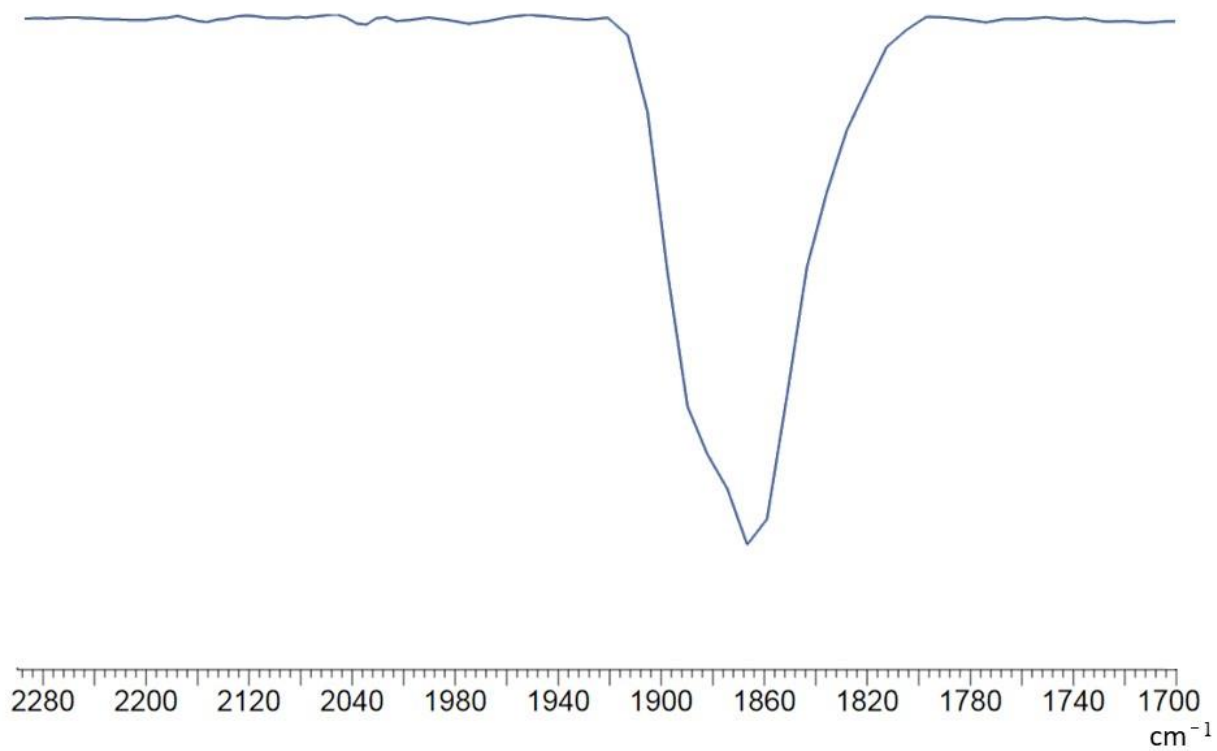


Figure S20: IR spectrum of **4** in the CO region. $\nu_{\text{C=O}}$: 1868 cm⁻¹.

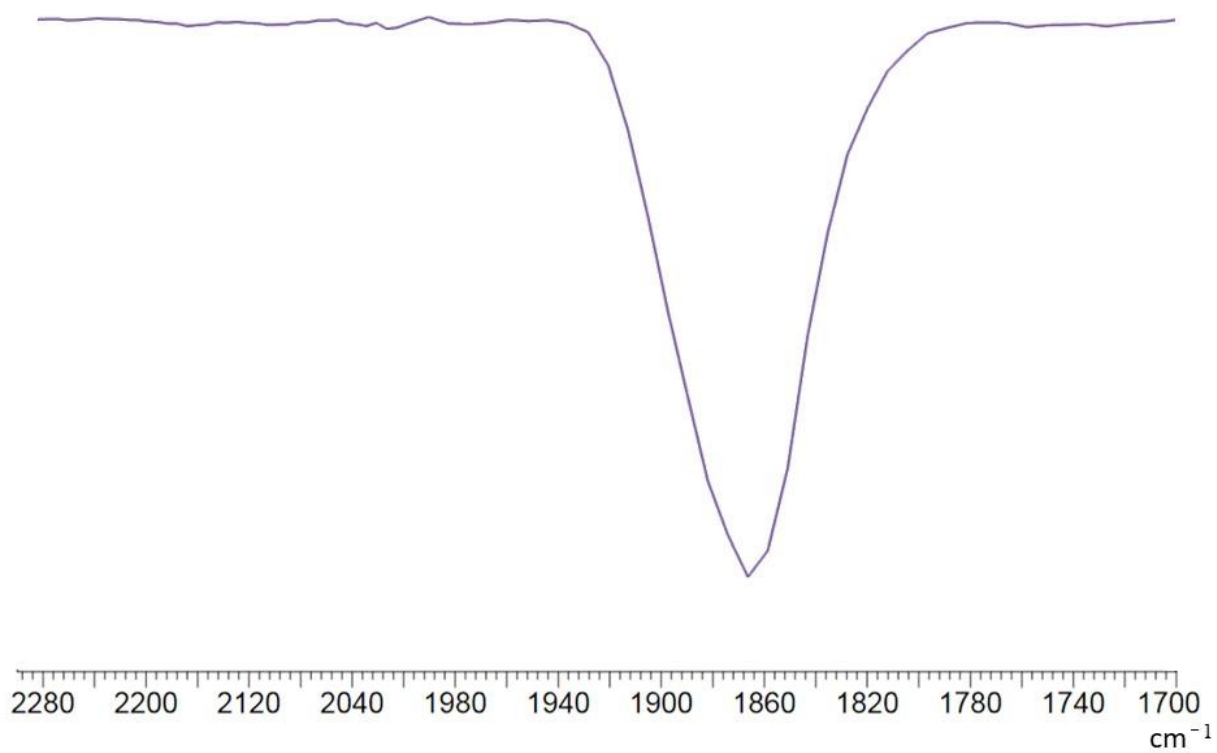


Figure S21: IR spectrum of 5 in the CO region. $\nu_{\text{C=O}}$: 1868 cm^{-1} .

7. GC-MS Catalytic Results

General Procedure: 1.00 mmol of ethyl propiolate was added to a 2 mL solution of deuterated solvent, containing 1 or 2 mol% catalyst with 2 mol% internal standard. After completion of the reaction, 0.1 mL of the reaction mixture was diluted 1000-fold in methanol (HPLC) and filtered through a silica plug. GCMS was taken, integrating both arene peaks and the internal standard peak to calculate yield.

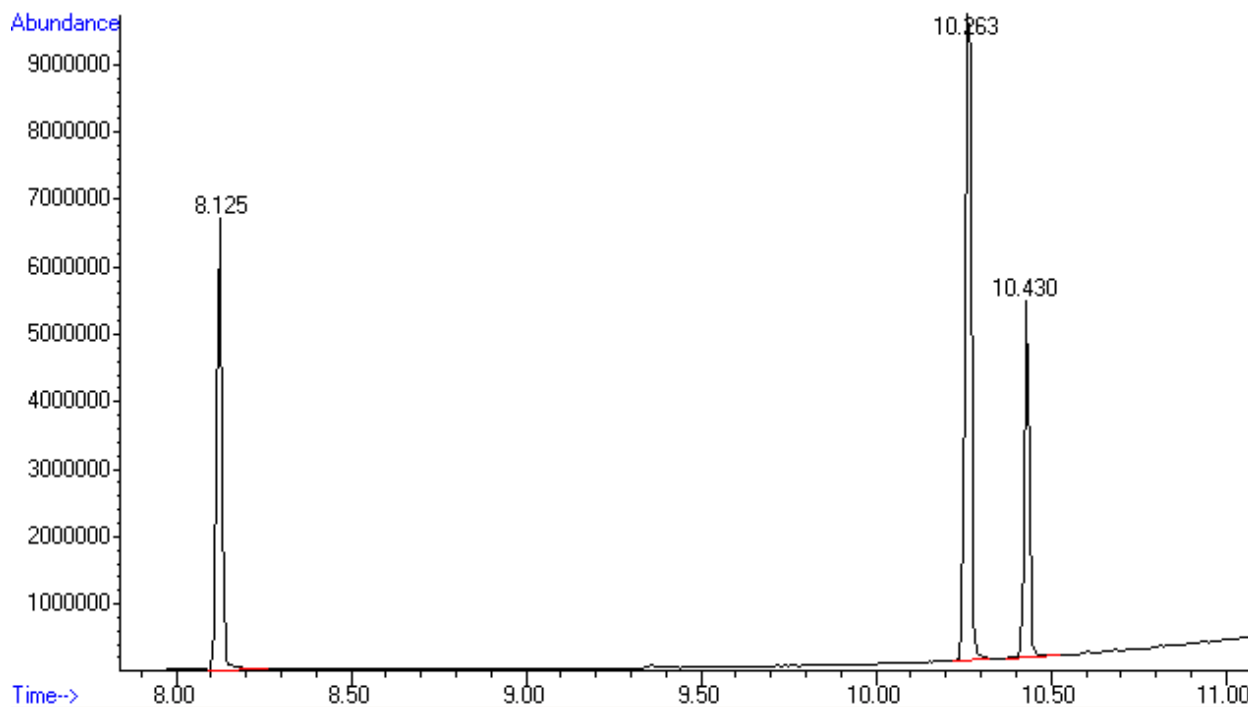


Figure S22: GC-MS spectrum demonstrating arene formation by complex **1** in CD_2Cl_2 at room temperature. The peak at 8.125 corresponds to hexamethylbenzene, the peak at 10.263 corresponds to the 1,2,4-isomer, and the peak at 10.430 corresponds to the 1,3,5-isomer.

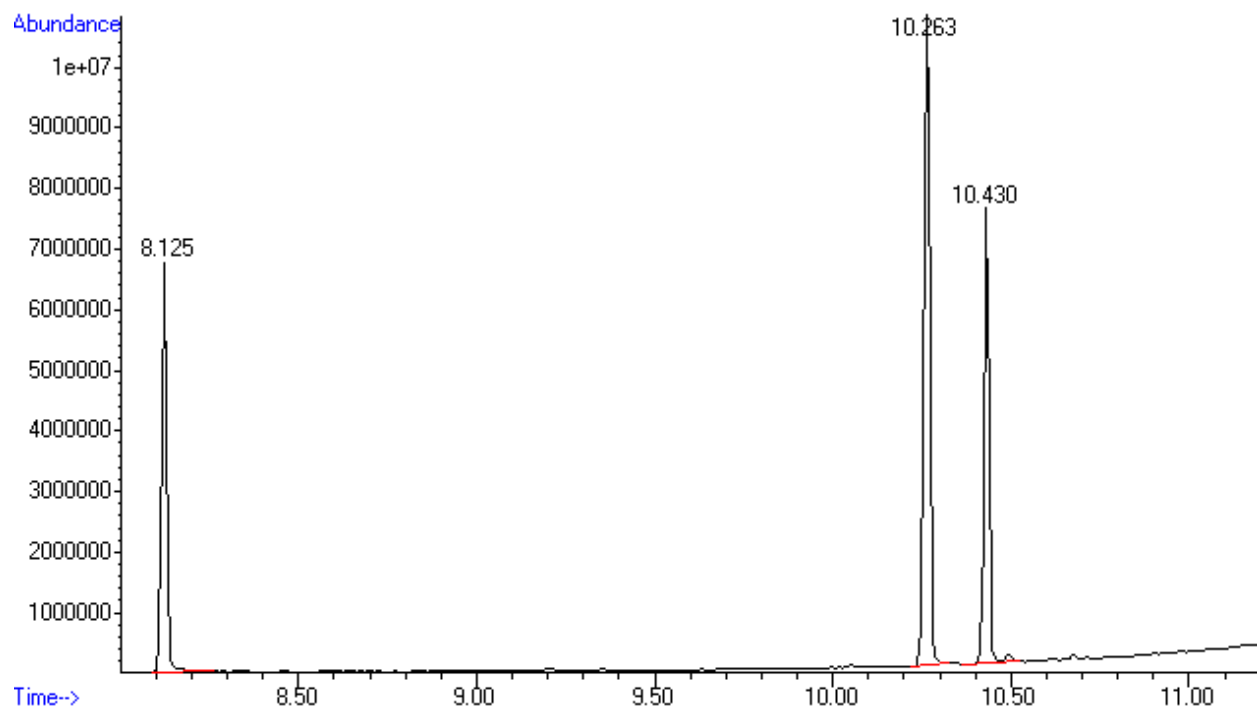


Figure S23: GC-MS spectrum demonstrating arene formation by complex **2** in CD_2Cl_2 at room temperature. The peak at 8.125 corresponds to hexamethylbenzene, the peak at 10.263 corresponds to the 1,2,4-isomer, and the peak at 10.430 corresponds to the 1,3,5-isomer.

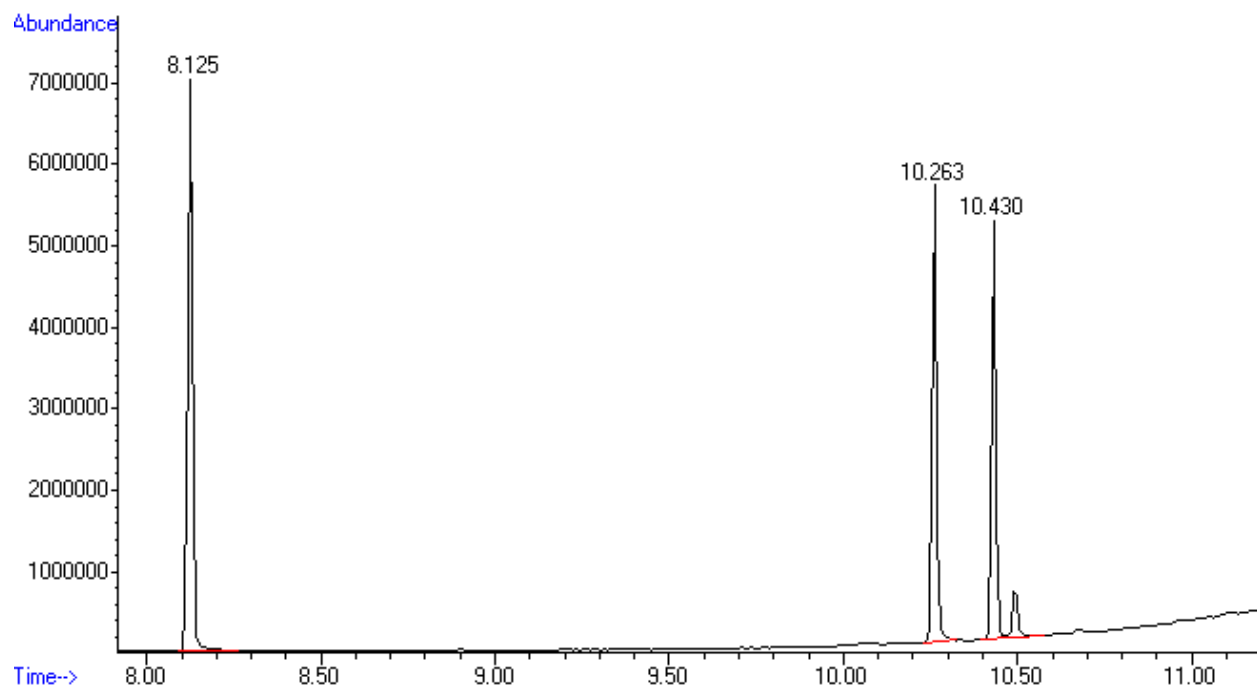


Figure S24: GC-MS spectrum demonstrating arene formation by complex **3** in CD_2Cl_2 at room temperature. The peak at 8.125 corresponds to hexamethylbenzene, the peak at 10.263 corresponds to the 1,2,4-isomer, and the peak at 10.430 corresponds to the 1,3,5-isomer.

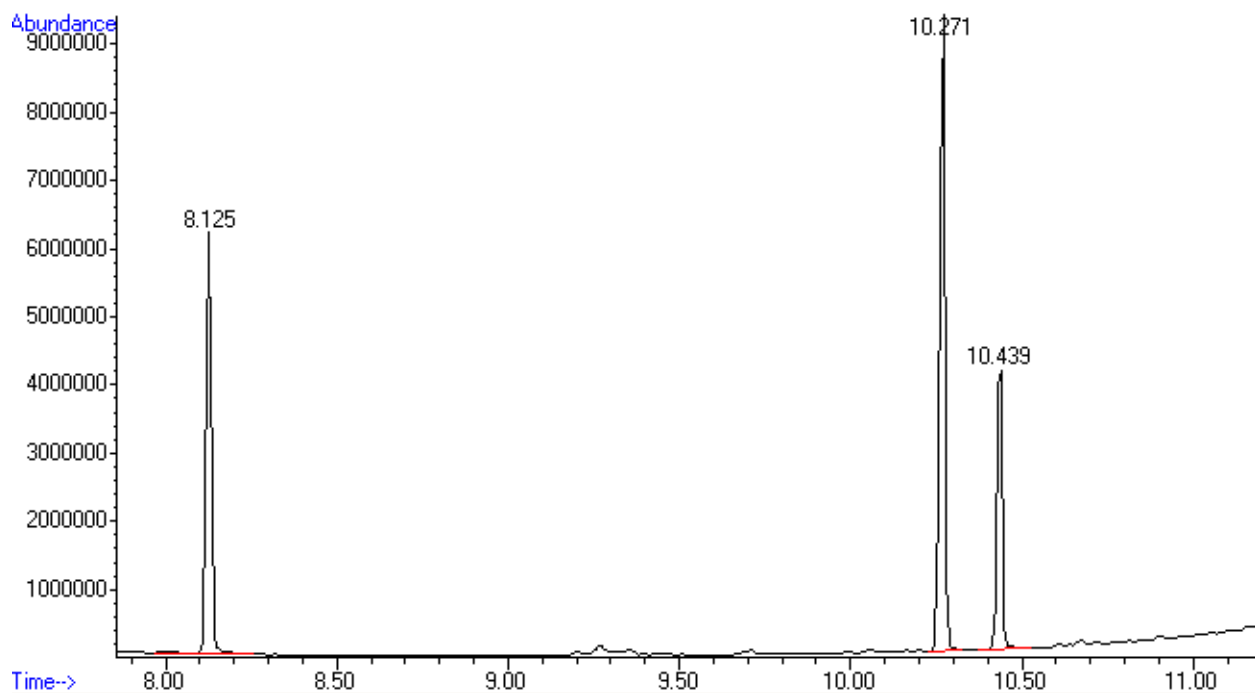


Figure S25: GC-MS spectrum demonstrating arene formation by complex **4** in CD_2Cl_2 at room temperature. The peak at 8.125 corresponds to hexamethylbenzene, the peak at 10.271 corresponds to the 1,2,4-isomer, and the peak at 10.439 corresponds to the 1,3,5-isomer.

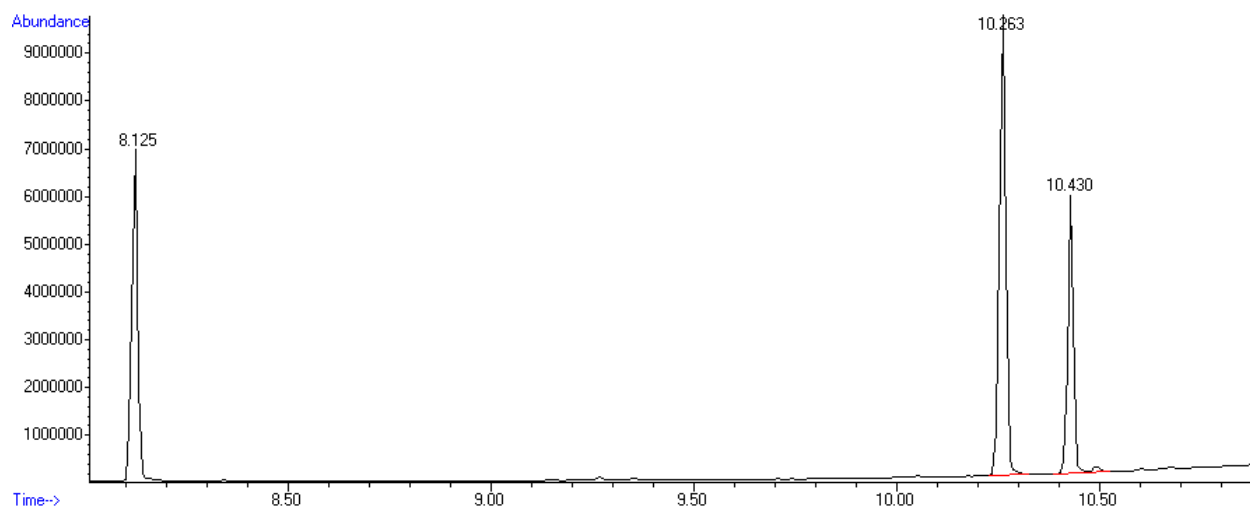


Figure S26: GC-MS spectrum demonstrating arene formation by complex **5** in CD_2Cl_2 at room temperature. The peak at 8.125 corresponds to hexamethylbenzene, the peak at 10.263 corresponds to the 1,2,4-isomer, and the peak at 10.430 corresponds to the 1,3,5-isomer.

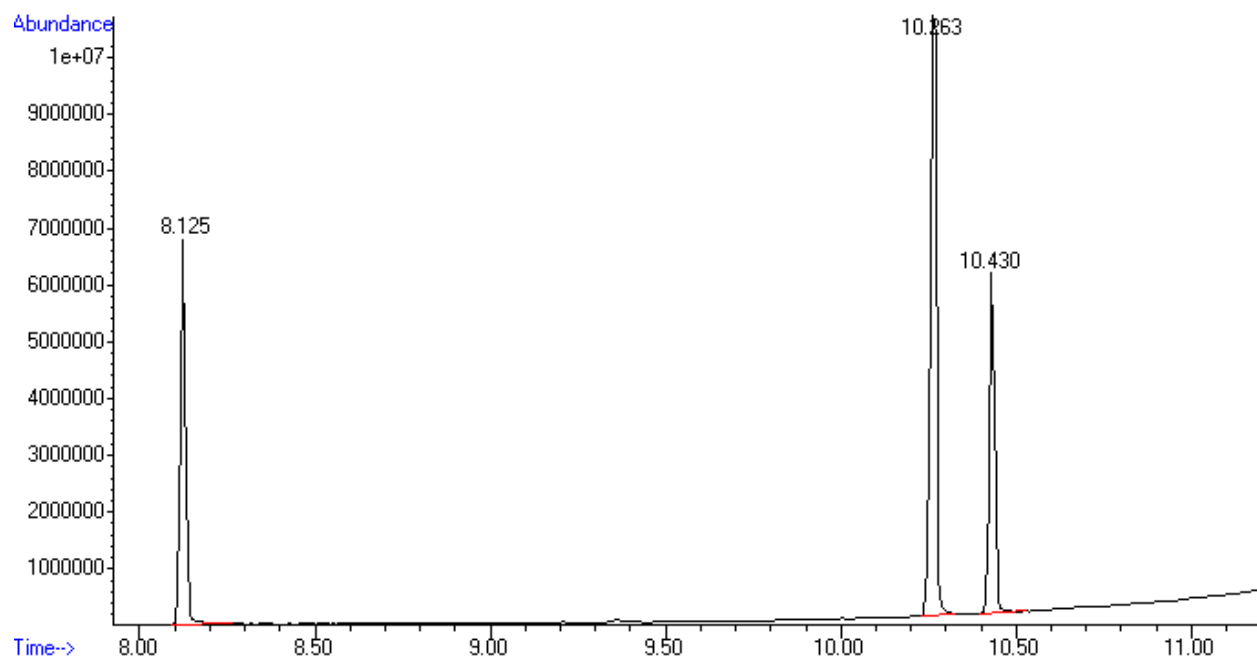


Figure S27: GC-MS spectrum demonstrating arene formation by $\text{Co}_2(\text{CO})_8$ in CD_2Cl_2 at room temperature. The peak at 8.125 corresponds to hexamethylbenzene, the peak at 10.263 corresponds to the 1,2,4-isomer, and the peak at 10.430 corresponds to the 1,3,5-isomer.

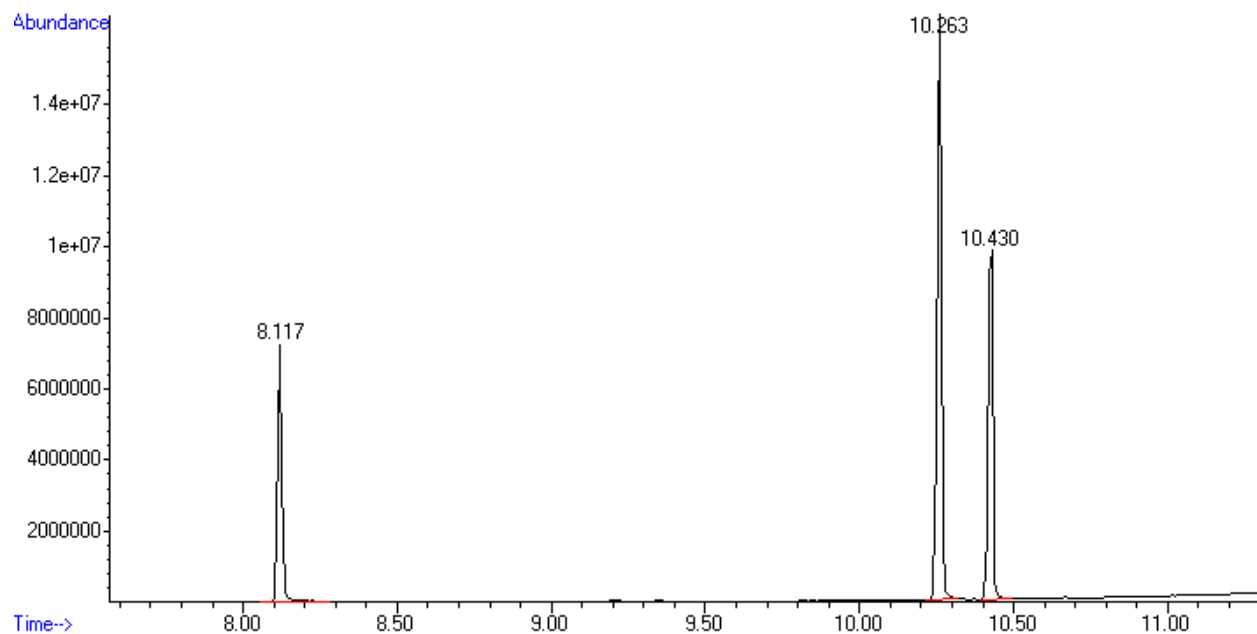


Figure S28: GC-MS spectrum demonstrating arene formation by complex **1** in CD_2Cl_2 at 40 °C.

The peak at 8.117 corresponds to hexamethylbenzene, the peak at 10.263 corresponds to the 1,2,4-isomer, and the peak at 10.430 corresponds to the 1,3,5-isomer.

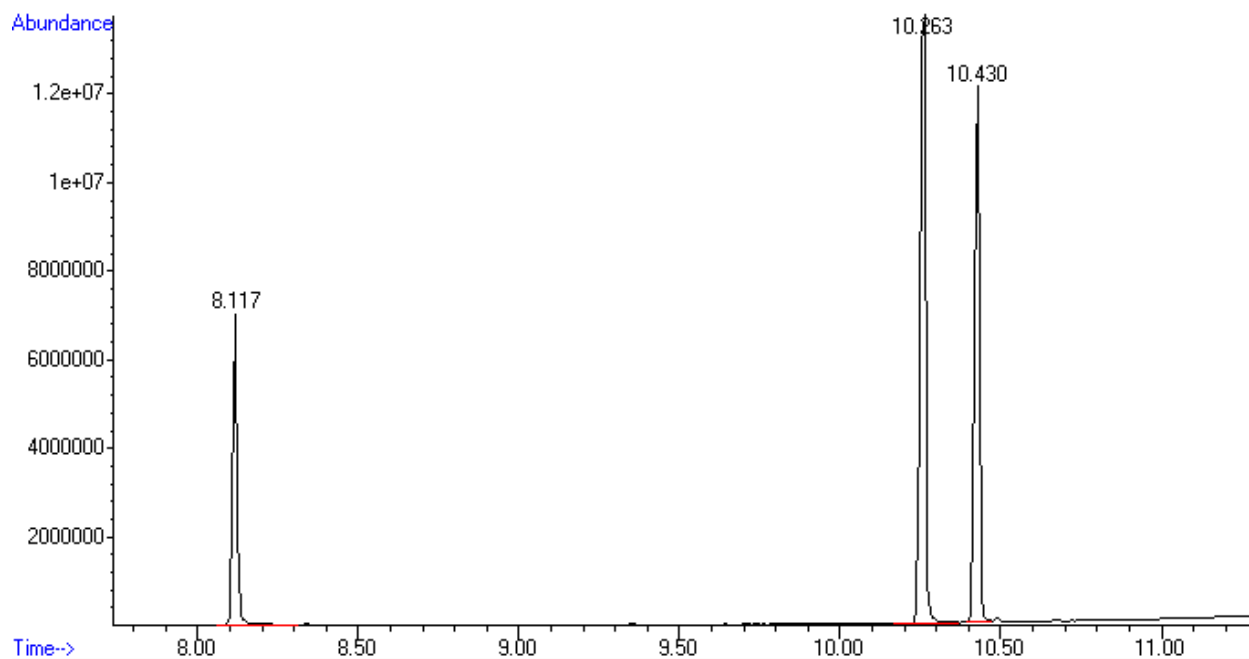


Figure S29: GC-MS spectrum demonstrating arene formation by complex **2** in CD_2Cl_2 at 40 °C.

The peak at 8.117 corresponds to hexamethylbenzene, the peak at 10.263 corresponds to the 1,2,4-isomer, and the peak at 10.430 corresponds to the 1,3,5-isomer.

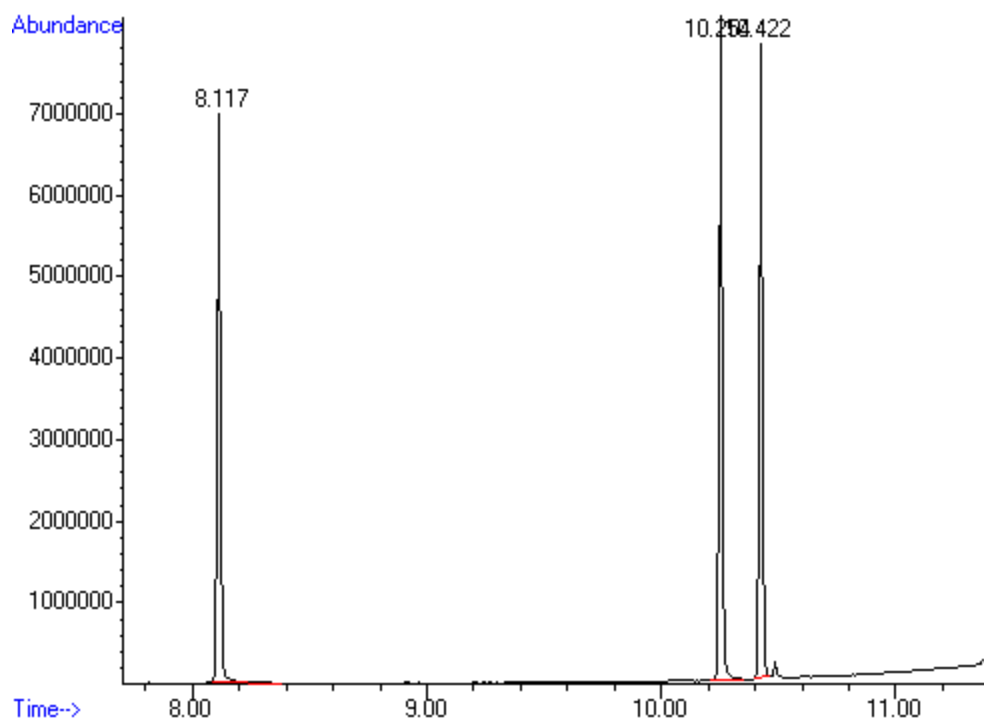


Figure S30: GC-MS spectrum demonstrating arene formation by complex **3** in CD_2Cl_2 at 40 °C.

The peak at 8.117 corresponds to hexamethylbenzene, the peak at 10.254 corresponds to the 1,2,4-isomer, and the peak at 10.422 corresponds to the 1,3,5-isomer.

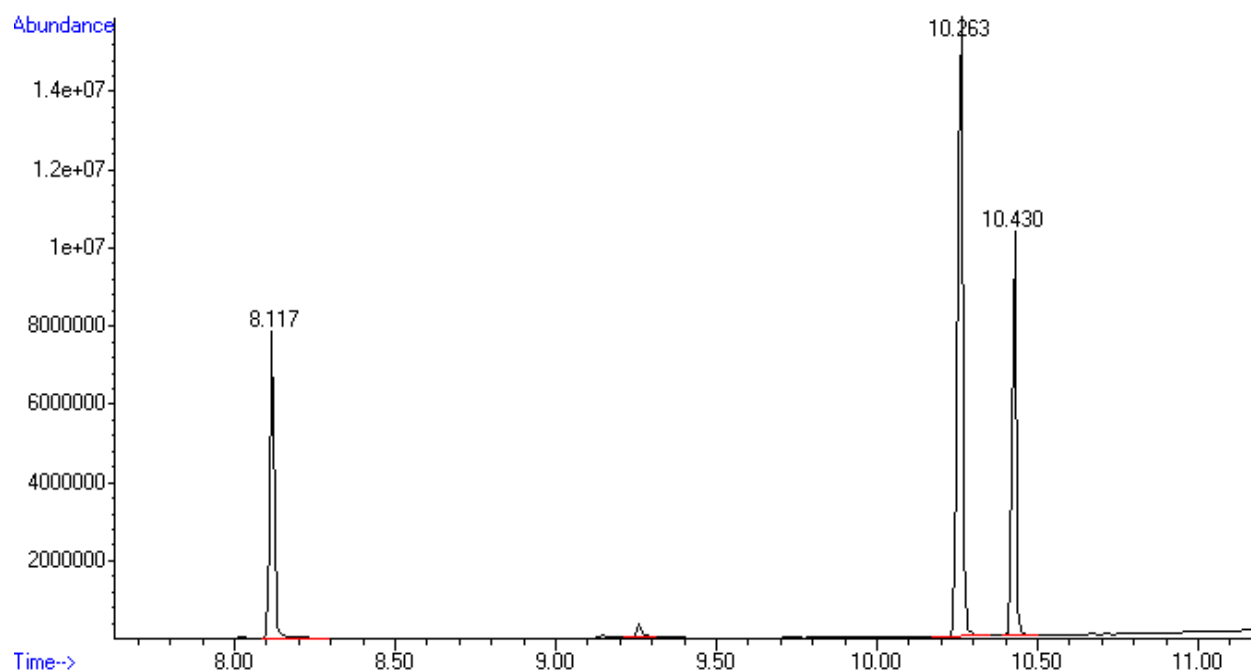


Figure S31: GC-MS spectrum demonstrating arene formation by complex **4** in CD_2Cl_2 at 40 °C.

The peak at 8.117 corresponds to hexamethylbenzene, the peak at 10.263 corresponds to the 1,2,4-isomer, and the peak at 10.430 corresponds to the 1,3,5-isomer.

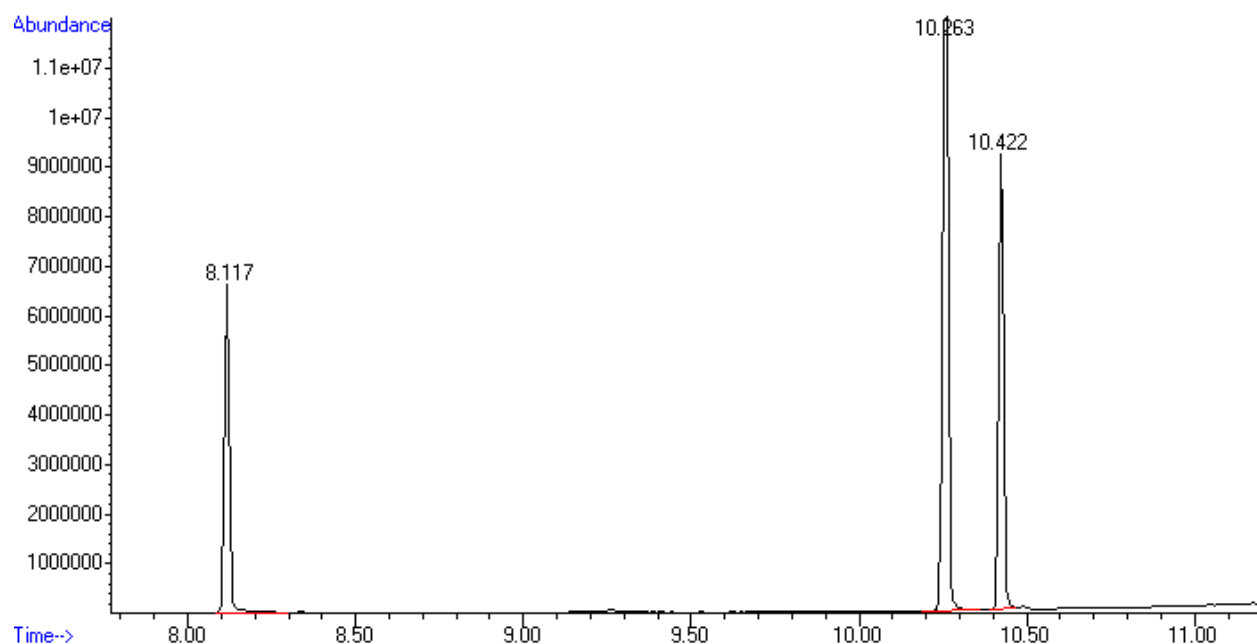


Figure S32: GC-MS spectrum demonstrating arene formation by complex **5** in CD_2Cl_2 at 40 °C.

The peak at 8.117 corresponds to hexamethylbenzene, the peak at 10.263 corresponds to the 1,2,4-isomer, and the peak at 10.422 corresponds to the 1,3,5-isomer.

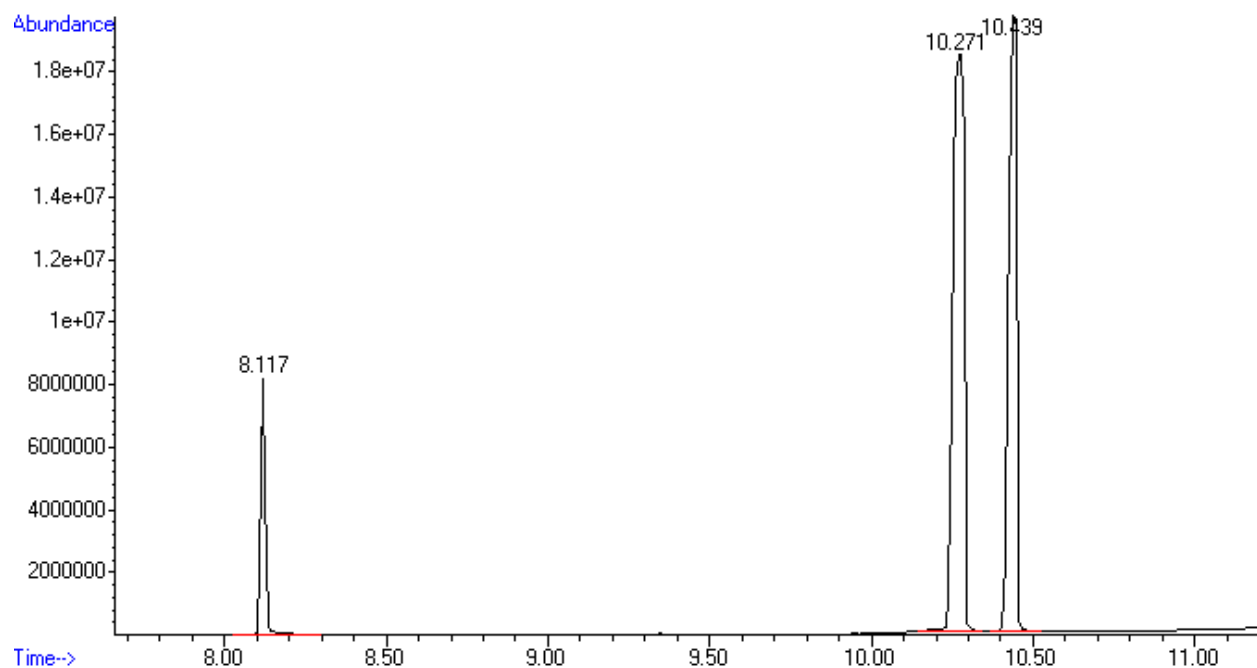


Figure S33: GC-MS spectrum demonstrating arene formation by $\text{Co}_2(\text{CO})_8$ at 40 °C. The peak at 8.117 corresponds to hexamethylbenzene, the peak at 10.271 corresponds to the 1,2,4-isomer, and the peak at 10.439 corresponds to the 1,3,5-isomer.

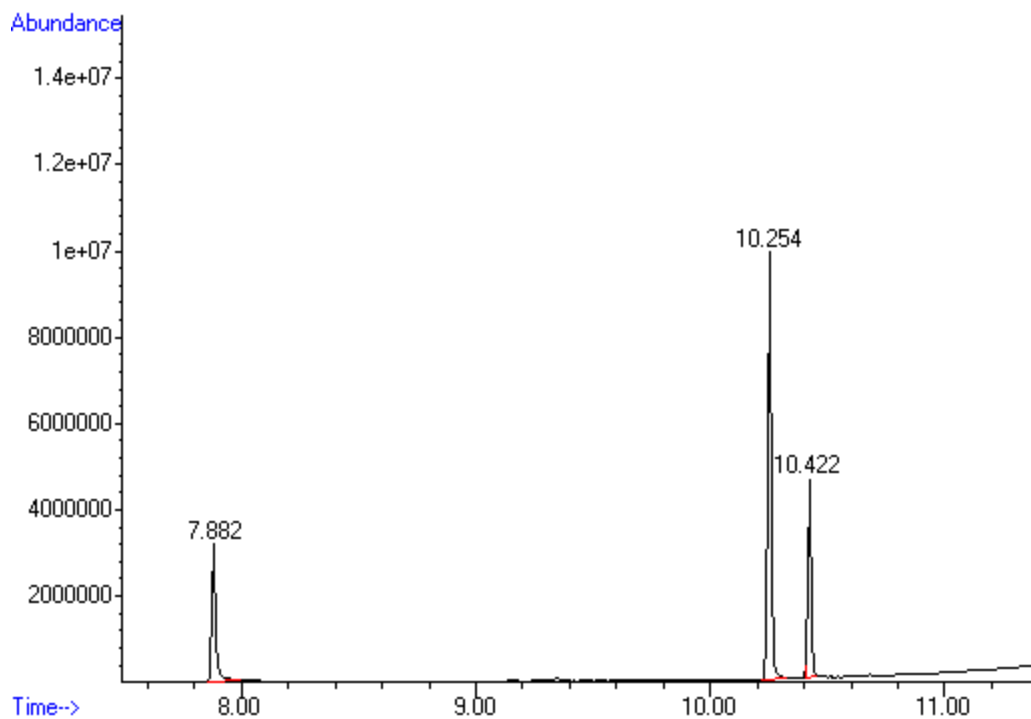


Figure S34: GC-MS spectrum demonstrating arene formation by complex **1** in C_7D_8 at room temperature. The peak at 7.882 corresponds to 1,3,5-trimethoxybenzene, the peak at 10.254 corresponds to the 1,2,4-isomer, and the peak at 10.422 corresponds to the 1,3,5-isomer.

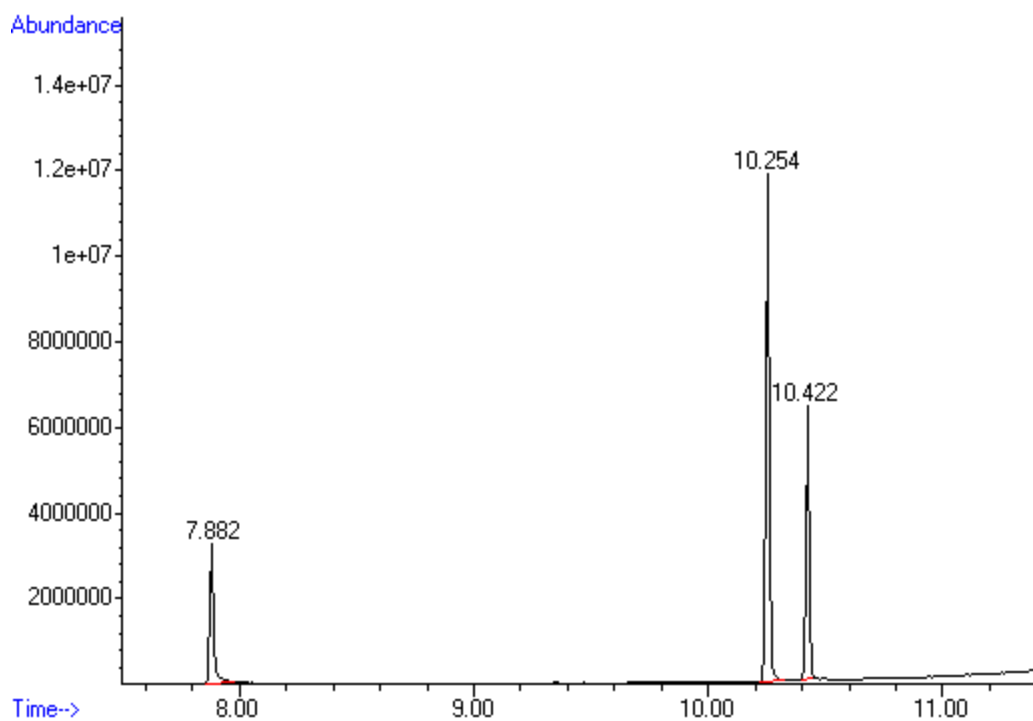


Figure S35: GC-MS spectrum demonstrating arene formation by complex **2** in C_7D_8 at room temperature. The peak at 7.882 corresponds to 1,3,5-trimethoxybenzene, the peak at 10.254 corresponds to the 1,2,4-isomer, and the peak at 10.422 corresponds to the 1,3,5-isomer.

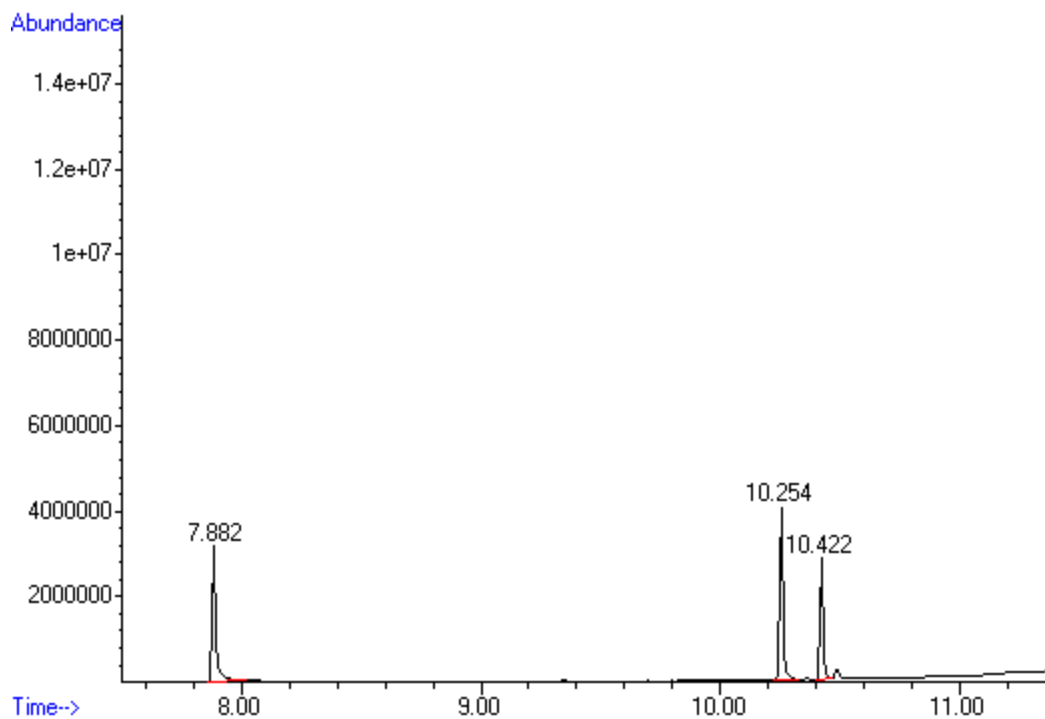


Figure S36: GC-MS spectrum demonstrating arene formation by complex **3** in C_7D_8 at room temperature. The peak at 7.882 corresponds to 1,3,5-trimethoxybenzene, the peak at 10.254 corresponds to the 1,2,4-isomer, and the peak at 10.422 corresponds to the 1,3,5-isomer.

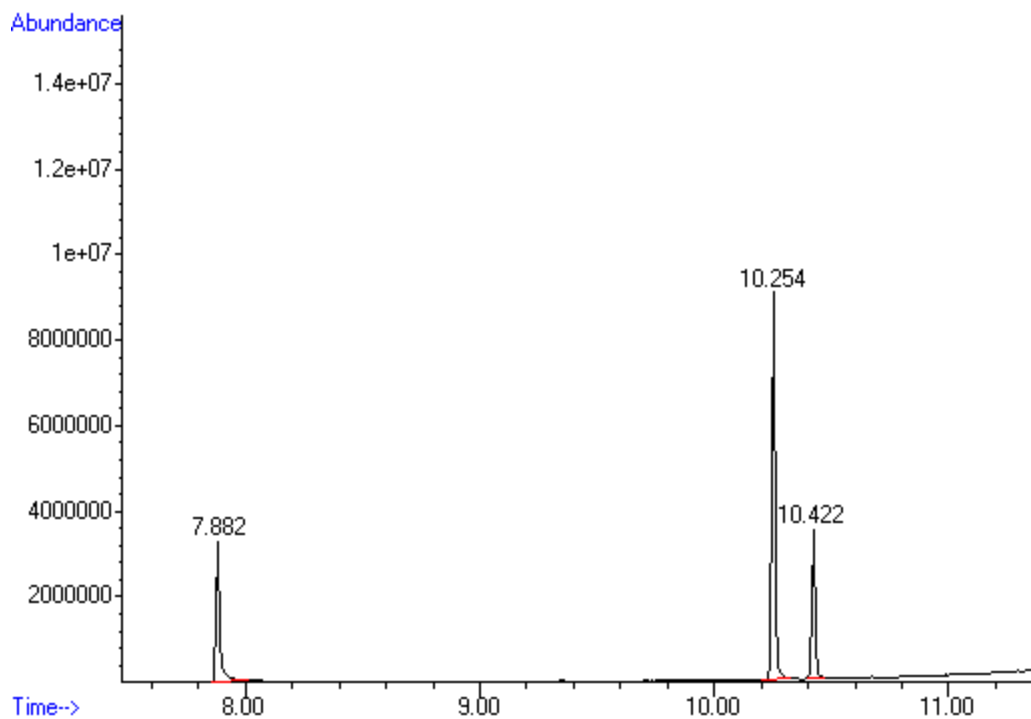


Figure S37: GC-MS spectrum demonstrating arene formation by complex **4** in C_7D_8 at room temperature. The peak at 7.882 corresponds to 1,3,5-trimethoxybenzene, the peak at 10.254 corresponds to the 1,2,4-isomer, and the peak at 10.422 corresponds to the 1,3,5-isomer.

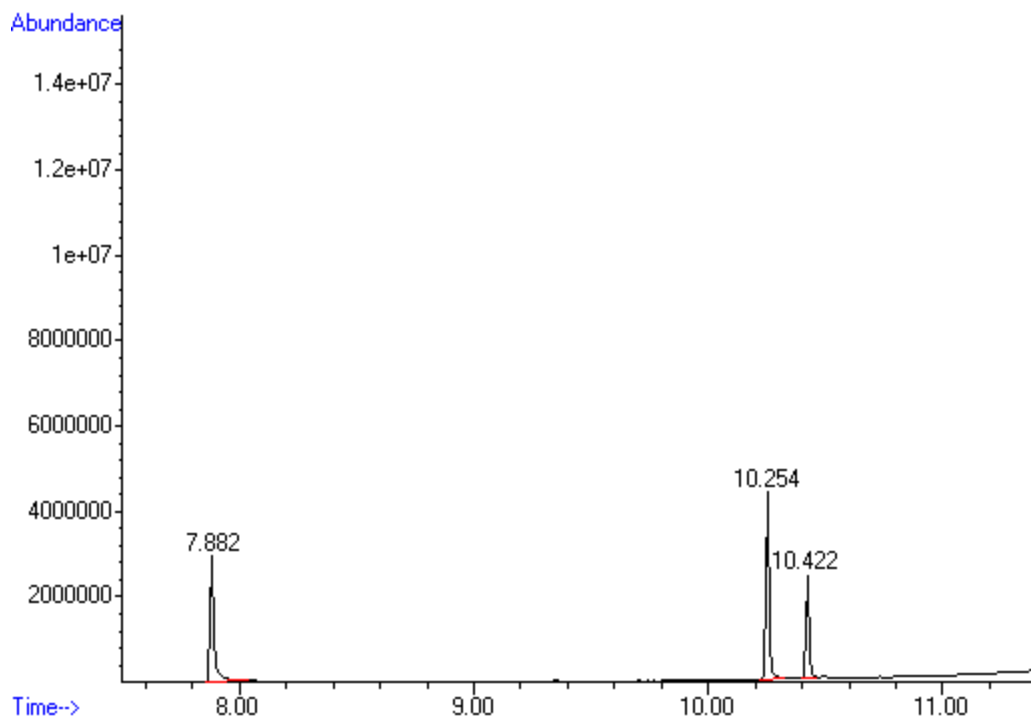


Figure S38: GC-MS spectrum demonstrating arene formation by complex **5** in C_7D_8 at room temperature. The peak at 7.882 corresponds to 1,3,5-trimethoxybenzene, the peak at 10.254 corresponds to the 1,2,4-isomer, and the peak at 10.422 corresponds to the 1,3,5-isomer.

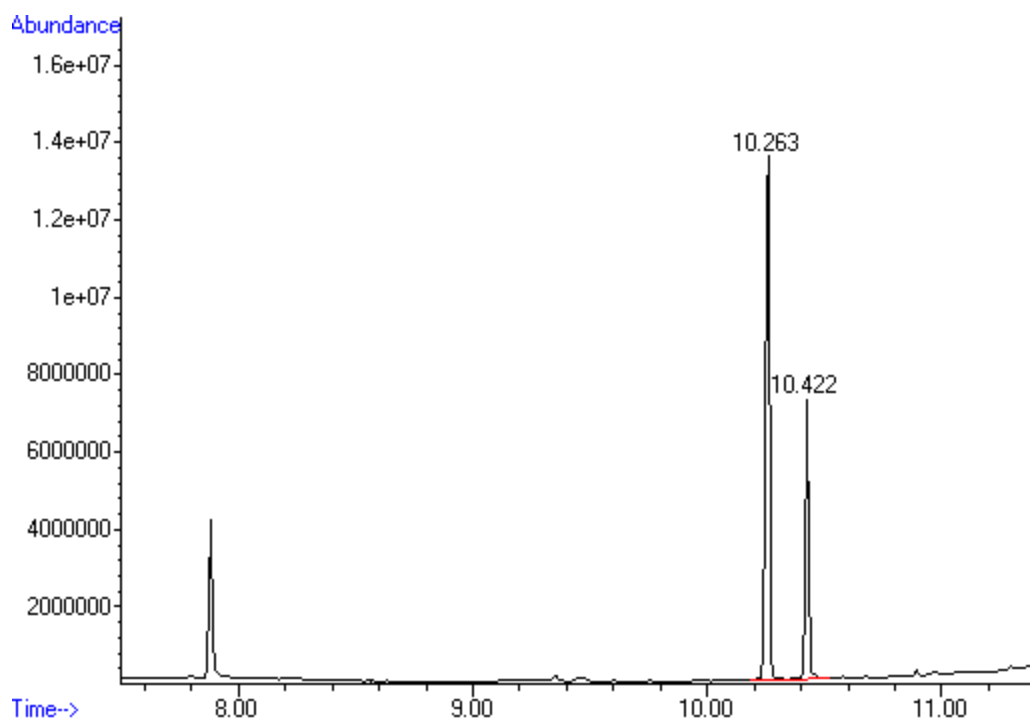


Figure S39: GC-MS spectrum demonstrating arene formation by $\text{Co}_2(\text{CO})_8$ in C_7D_8 at room temperature. The peak at 7.882 corresponds to 1,3,5-trimethoxybenzene, the peak at 10.263 corresponds to the 1,2,4-isomer, and the peak at 10.422 corresponds to the 1,3,5-isomer.

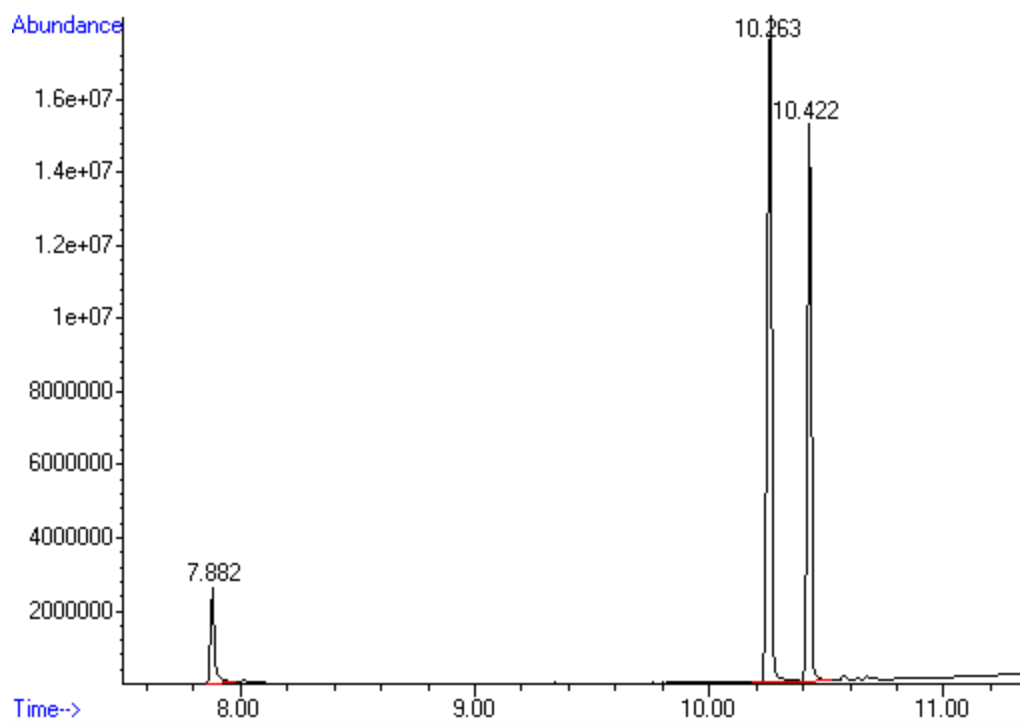


Figure S40: GC-MS spectrum demonstrating arene formation by complex **1** in C₇D₈ at 80 °C. The peak at 7.882 corresponds to 1,3,5-trimethoxybenzene, the peak at 10.263 corresponds to the 1,2,4-isomer, and the peak at 10.422 corresponds to the 1,3,5-isomer.

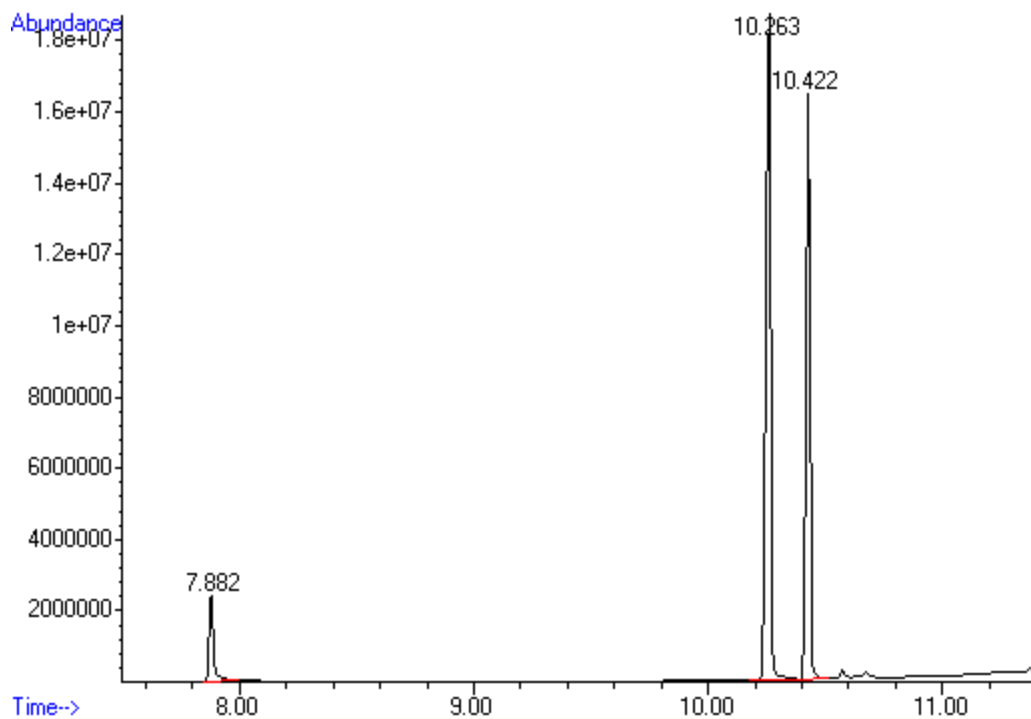


Figure S41: GC-MS spectrum demonstrating arene formation by complex **2** in C₇D₈ at 80 °C. The peak at 7.882 corresponds to 1,3,5-trimethoxybenzene, the peak at 10.263 corresponds to the 1,2,4-isomer, and the peak at 10.422 corresponds to the 1,3,5-isomer.

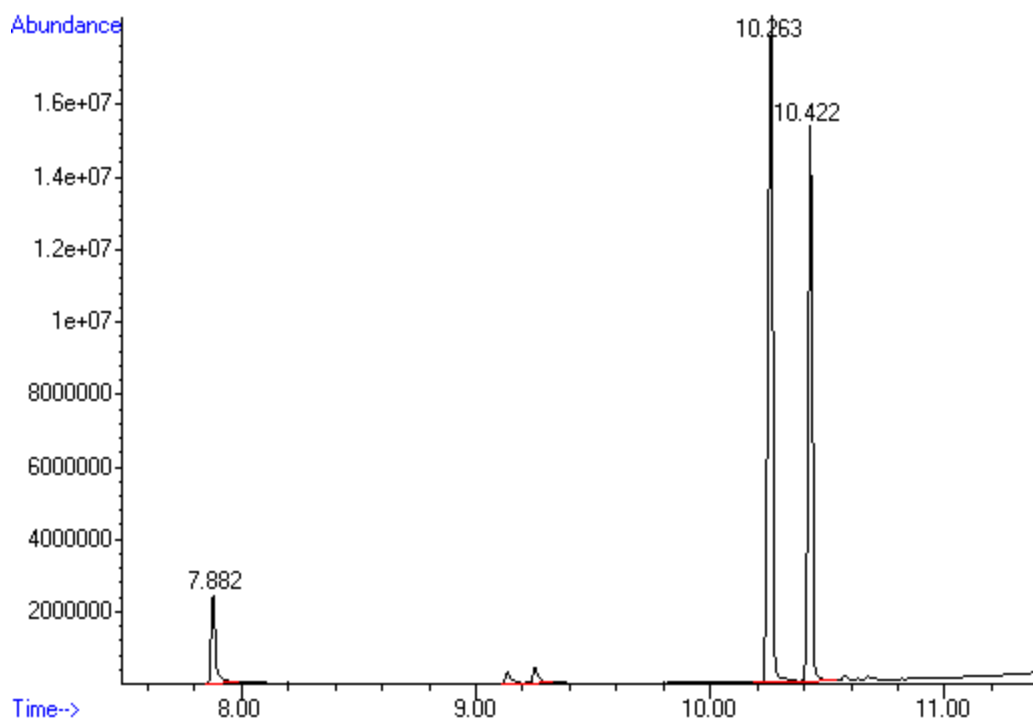


Figure S42: GC-MS spectrum demonstrating arene formation by complex **4** in C₇D₈ at 80 °C. The peak at 7.882 corresponds to 1,3,5-trimethoxybenzene, the peak at 10.263 corresponds to the 1,2,4-isomer, and the peak at 10.422 corresponds to the 1,3,5-isomer.

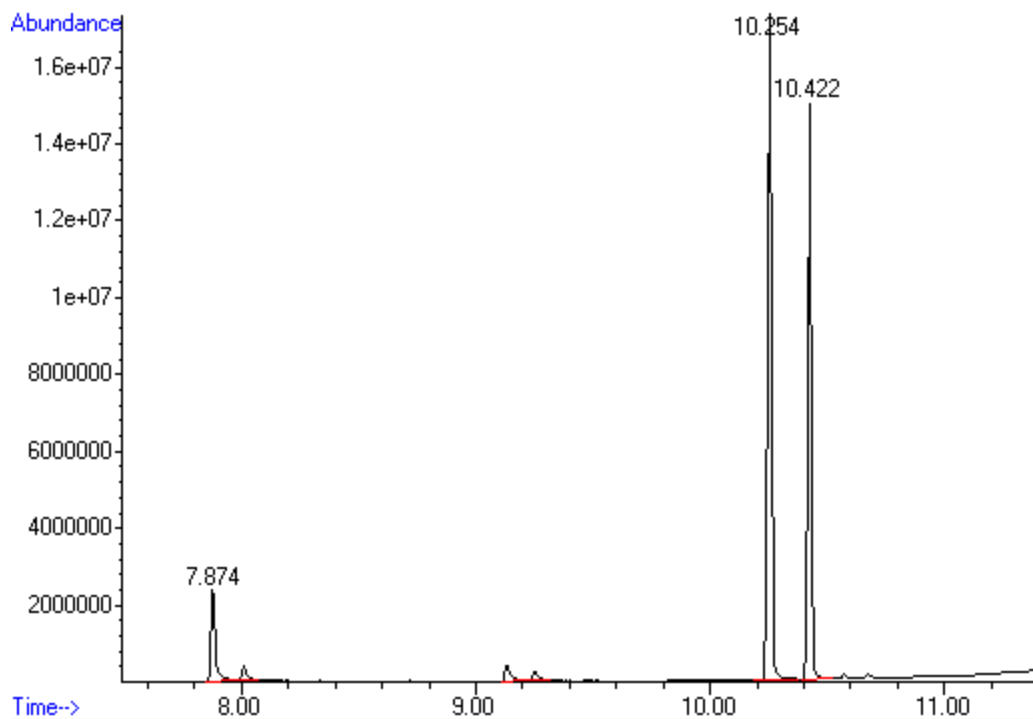


Figure S43: GC-MS spectrum demonstrating arene formation by complex **5** in C₇D₈ at 80 °C. The peak at 7.874 corresponds to 1,3,5-trimethoxybenzene, the peak at 10.254 corresponds to the 1,2,4-isomer, and the peak at 10.422 corresponds to the 1,3,5-isomer.

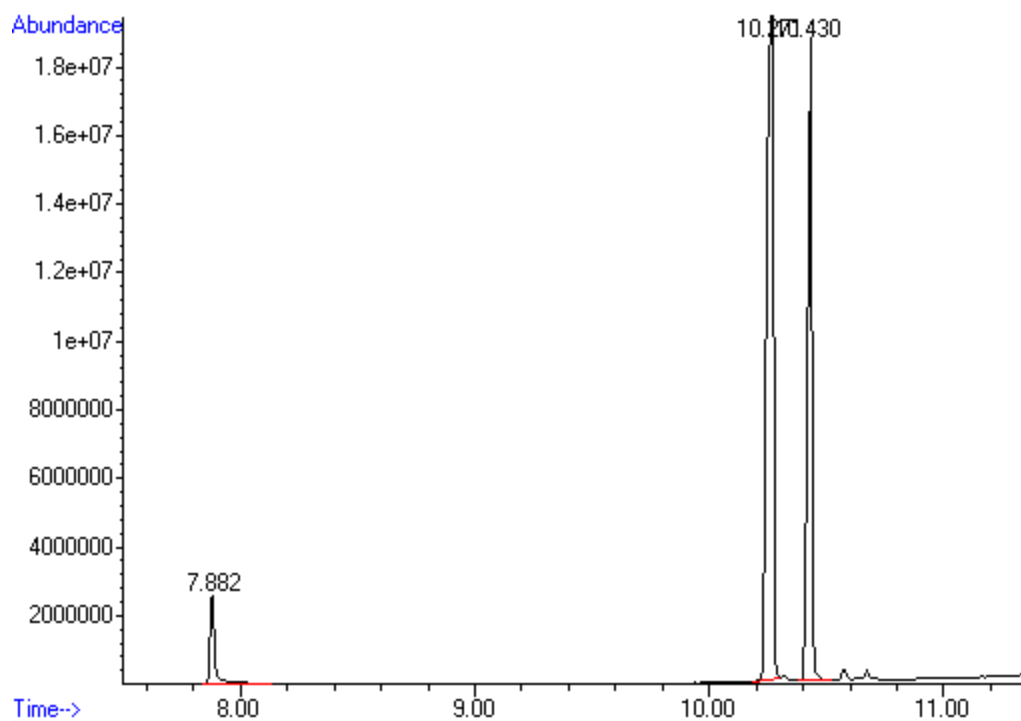


Figure S44: GC-MS spectrum demonstrating arene formation by $\text{Co}_2(\text{CO})_8$ in C_7D_8 at 80 °C. The peak at 7.882 corresponds to 1,3,5-trimethoxybenzene, the peak at 10.271 corresponds to the 1,2,4-isomer, and the peak at 10.430 corresponds to the 1,3,5-isomer.

8. Computational Details

All density functional theory calculations were performed in Gaussian 09 revision D01.² The B3LYP functional was employed in combination with the semi-empirical D3 dispersion correction developed by Grimme and co-workers, including the Becke-Johnson damping, in combination with the 6-31G(d) basis set.³ Because the spin states were known experimentally in most cases (except **3**), only the experimental spin state was explored but multiple guess wavefunctions (including broken symmetry options) were used as initial guesses. In all cases the results converged to a single wavefunction that was confirmed to be stable.⁴ All structures were confirmed to be minima by analyzing the harmonic frequencies.⁵

All topological analyses were performed using Multiwfn version 3.5.⁶ The source function analyses were performed with the bond critical point between the Co centers used as the reference point. Integration of the source function used atomic-center and uniform grids (high quality, 0.06 Bohr spacing) with exact refinement of the basin boundaries (option 2). As Tables S4-S5 demonstrate, we observe small negative contributions to the source function from the cobalt centers and major contributions from the carbonyls for the two isomers of $\text{Co}_2(\text{CO})_8$ similar to what Gatti and Lasi reported.⁷ One small difference is that the contributions from the Co centers in the bridged isomer are small and negative vs. small and positive, though it should be noted that we identified a bcp for the bridged isomer and evaluated the source function at that point rather than the midpoint of the Co-Co separation as they did. Looking at Table S6 for **i** and Table S7 for **iiB**, the story is generally the same: (i) most of the Co centers make negative contributions (all are small), (ii) the carbonyls are the primary contributors, and (iii) **i** has larger contributions from Co to the source function than **iiB**, consistent with all other indicators of less bonding in the bridged species. The primary difference compared to $\text{Co}_2(\text{CO})_8$ is unequal contributions from the two Co centers. For both species, the Co bound to the bis(imino)pyridine makes a much smaller contribution (-0.82% vs. -12% and +2.9% vs. -1.9%) which further reinforces the assignment of different oxidation states for the two Co centers.

Figure S45. Spin density isosurface plot (iso = 0.002 au) for species **v**.

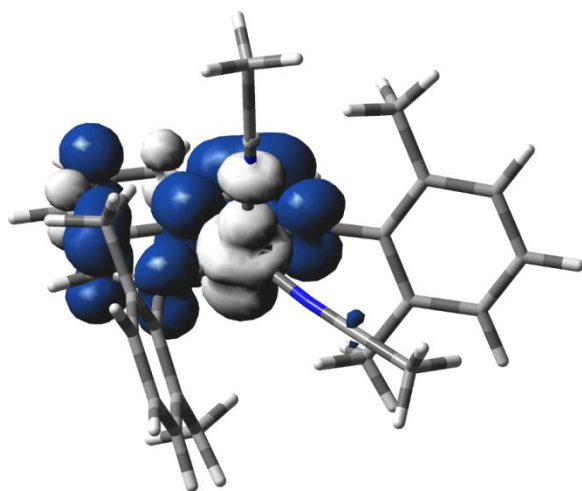


Table S2. Comparison of select bond lengths (Å) in **i** and **1**. See Figure 1 in manuscript for atom labels.

Bond	i	1
Co1---Co2	2.691	2.798
Co2-N1	1.925	1.920
Co2-N2	1.938	1.924
Co2-N3	1.831	1.833
N _{im} -C _{im}	1.319	1.32*
C _{im} -C _{py}	1.427	1.43*
C _{py} -N _{py}	1.367	1.372*
Co2-C5	1.760	1.774
Co1-C1	1.772	1.773
Co1-C2	1.755	1.754
Co1-C3	1.757	1.778
Co1-C4	1.769	1.784

im = iminopyridine, py = pyridine; *values in two arms averaged

Table S3. Comparison of select bond lengths (Å) in **iib** and **2b**. See Figure 3 in manuscript for atom labels.

Bond	iia	2a
Co3---Co4	2.568	2.594
Co3-N1	1.948	1.985
Co3-N2	1.932	1.954
Co3-N3	1.826	1.867
N _{im} -C _{im}	1.32*	1.307
C _{im} -C _{py}	1.42*	1.43*
C _{py} -N _{py}	1.373	1.36*
Co3-C6	1.876	1.960
Co3-C6	1.915	1.894
Co4-C7	1.832	1.814
Co4-C7	2.077	2.220
Co4-C8	1.788	1.796
Co4-C9	1.790	1.802
Co4-C10	1.766	1.750

im = iminopyridine, py = pyridine; *values in two arms averaged

Figure S46. Plot of bond critical point locations for unbridged $\text{Co}_2(\text{CO})_8$. Yellow spheres correspond to $(3,+1)$ bcps and orange spheres correspond to $(3,-1)$ bcps.

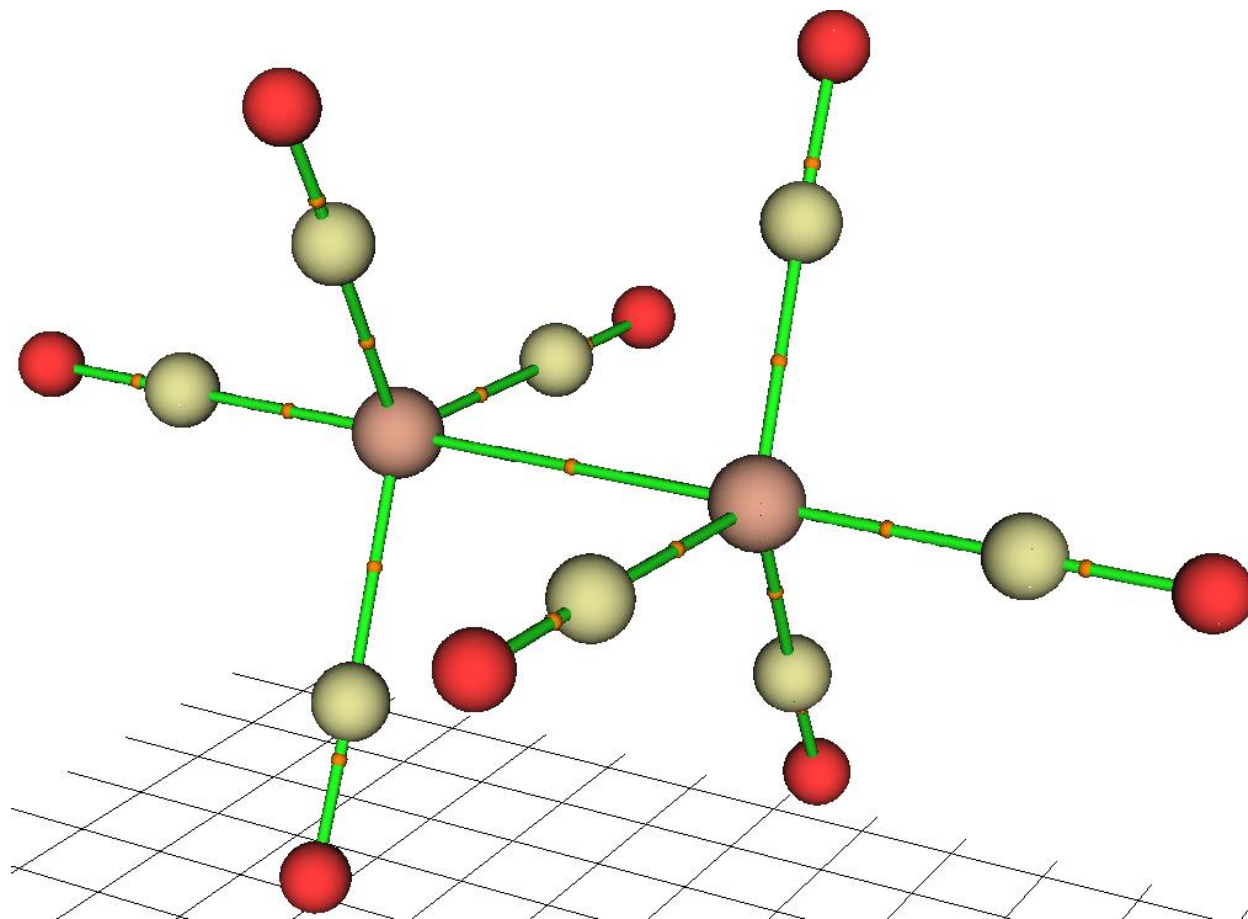


Figure S47. Plot of bond critical point locations for bridged $\text{Co}_2(\text{CO})_8$. Yellow spheres correspond to $(3,+1)$ bcps and orange spheres correspond to $(3,-1)$ bcps.

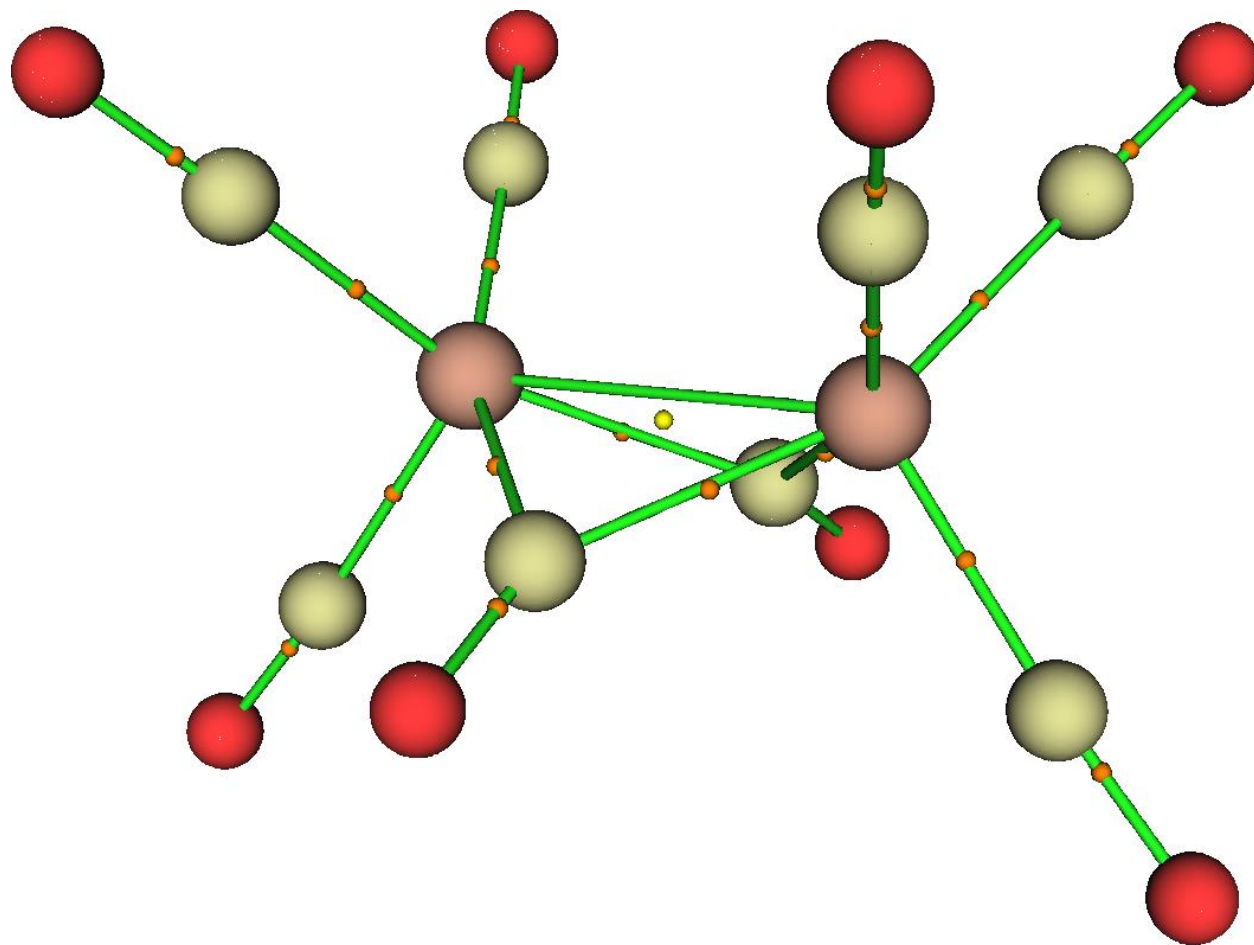


Figure S48. Plot of bond critical point locations for **i**. Yellow spheres correspond to (3,+1) bcps and orange spheres correspond to (3,-1) bcps.

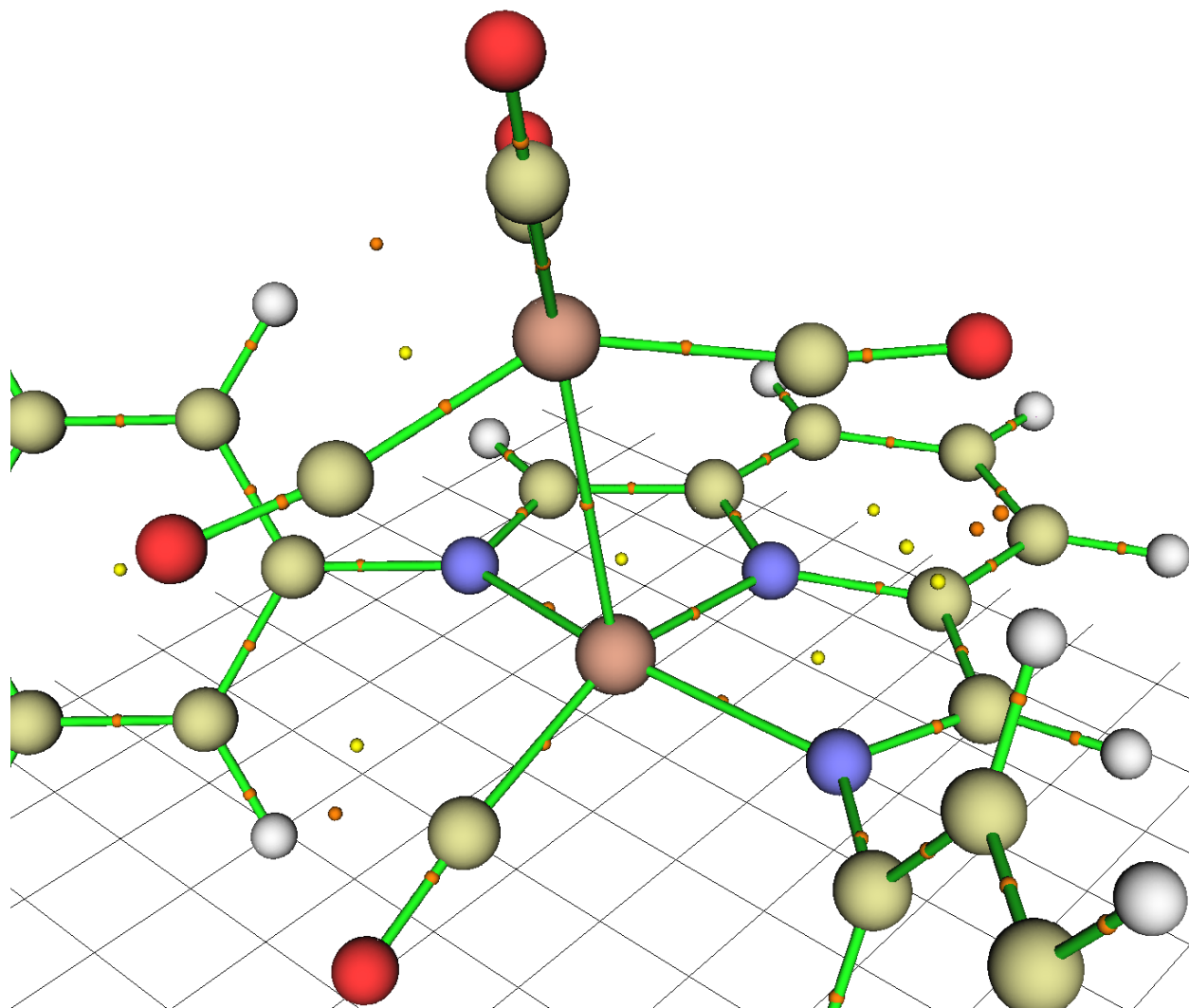


Figure S49. Plot of bond critical point locations for **iiib**. Yellow spheres correspond to (3,+1) bcps and orange spheres correspond to (3,-1) bcps.

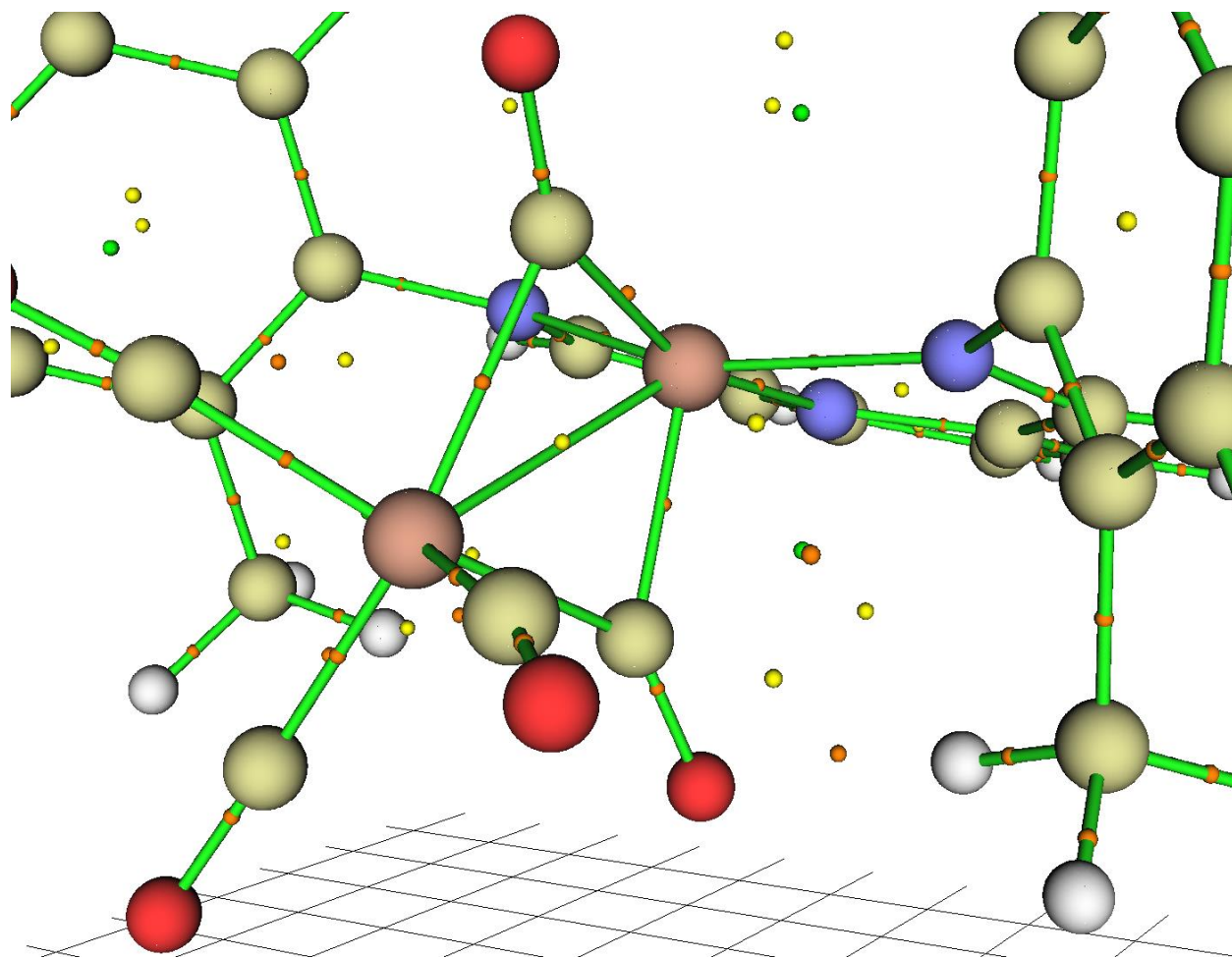


Figure S50. Schematic representation of all computed model structures.

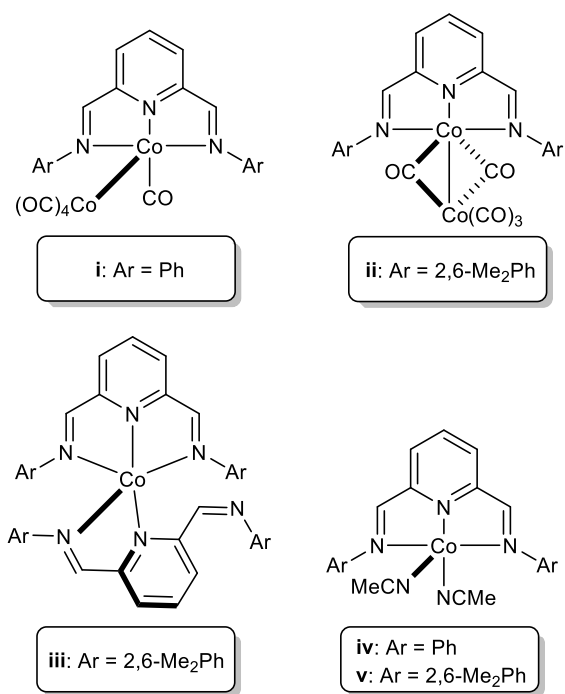


Table S4. Source function analysis for the (3,-1) bcp between the Co centers in Figure S46. The atom ordering is the same as reported in Table S8.

Basin	% Contribution
Co	-1.71
Co	-1.71
C	1.63
O	12.63
C	1.59
O	12.66
C	1.68
O	12.62
C	1.57
O	7.34
C	1.62
O	12.64
C	1.69
O	12.62
C	1.59
O	12.66
C	1.56
O	7.34

Table S5. Source function analysis for the (3,+1) bcp between the Co centers in Figure S47. The atom ordering is the same as reported in Table S8.

Basin	% Contribution
Co	-0.15
Co	-0.17
C	1.65
O	20.11
C	1.67
O	7.78
C	1.65
O	7.79
C	1.85
O	7.54
C	1.65
O	7.80
C	1.93
O	20.14
C	1.63
O	7.76
C	1.82
O	7.55

Table S6. Source function analysis for the (3,-1) bcp between the Co centers in Figure S48. The atom ordering is the same as reported in Table S8.

Basin	% Contribution
Co	-0.82
N	-2.10
C	1.55
C	1.90
C	2.04
C	1.88
C	1.27
H	1.56
H	1.55
H	1.57
C	4.09
H	3.12
C	3.91
H	2.84
N	-4.47
N	-3.20
Co	-11.94
C	2.01
O	12.52
C	3.35
O	9.44
C	1.39
O	19.19
C	2.08
O	17.67
C	3.00
O	12.97
C	-1.17
C	-0.09
C	0.36
C	1.13
H	-0.06
C	1.16
H	0.61
C	1.02
H	1.17
H	1.15
H	1.14
C	-1.14
C	0.51
C	-0.57
C	1.04
H	0.55
C	1.04
H	-0.39

C	0.90
H	1.10
H	1.06
H	1.14

Table S7. Source function analysis for the (3,+1) bcp between the Co centers in Figure S49. The atom ordering is the same as reported in Table S8.

Basin	% Contribution
C	0.84
C	1.15
C	1.25
C	1.38
C	1.15
H	0.91
H	0.88
H	0.87
N	0.12
C	2.92
C	2.87
N	-2.63
N	-2.54
Co	2.89
C	-1.98
C	-0.34
C	-0.88
C	0.78
C	0.76
C	0.69
H	1.04
H	1.04
C	-1.64
C	-0.78
C	-0.22
C	0.70
C	0.73
C	0.67
H	0.94
H	0.95
C	0.83
C	-0.06
O	21.97
O	22.31
Co	-1.88
C	1.94
C	2.46
C	1.81
O	9.01
O	8.37
O	9.45
H	1.84
H	1.85
C	-0.04
H	0.62

H	1.54
H	-0.95
C	0.10
H	1.29
H	1.64
H	-2.21
C	0.20
H	1.30
H	1.01
H	-1.53
C	0.08
H	1.55
H	1.20
H	-2.19
H	0.95
H	1.02

Table S8. Cartesian coordinates (Å) for all optimized structures.

-----				N	-1.667782	1.279528	0.460736
i S=0				Co	0.136229	0.710217	0.066710
-----				C	-2.786772	0.443523	0.775412
Co	0.099794	0.101661	-0.752323	C	-2.953744	0.019743	2.105955
N	0.143717	1.923311	-0.936113	C	-3.672550	0.052423	-0.246341
C	-1.030555	2.621815	-0.971129	C	-4.006639	-0.854845	2.389290
C	1.349028	2.562658	-1.015539	C	-4.703392	-0.833660	0.081915
C	-1.019709	4.014112	-1.030736	C	-4.866471	-1.293762	1.385574
C	1.403478	3.953946	-1.083057	H	-4.147598	-1.192677	3.412353
C	0.209380	4.680936	-1.078226	H	-5.380417	-1.164682	-0.701116
H	-1.955012	4.563919	-1.044984	C	3.022927	-0.157504	0.521167
H	2.363489	4.456693	-1.136697	C	3.698418	-0.897766	-0.462477
H	0.235402	5.764420	-1.122670	C	3.233923	-0.380683	1.896150
C	-2.130325	1.712433	-0.979429	C	4.552091	-1.923551	-0.038119
H	-3.168536	2.028729	-1.014076	C	4.099861	-1.408084	2.272270
C	2.402565	1.599880	-1.005542	C	4.745908	-2.186814	1.313154
H	3.456608	1.858367	-1.036003	H	5.060860	-2.522821	-0.788094
N	-1.797933	0.436296	-0.959392	H	4.264943	-1.596664	3.329519
N	2.001382	0.346310	-0.923672	C	0.122043	0.068272	-1.696448
Co	-0.184017	-0.107547	1.915070	C	0.111590	-0.888140	0.962468
C	0.120950	-1.546393	-1.368585	O	0.332443	0.567788	-2.743997
O	0.165016	-2.578119	-1.881276	O	0.278253	-1.464275	1.971301
C	0.306962	-0.810422	3.448914	Co	-0.413294	-1.597480	-0.917288
O	0.599217	-1.233257	4.482597	C	-1.560903	-2.570198	0.052871
C	-0.837309	-1.607208	1.232692	C	-1.404045	-1.799693	-2.365420
O	-1.330469	-2.623212	0.975585	C	1.030063	-2.626658	-1.154400
C	1.160063	1.007616	1.746799	O	-2.280055	-3.281078	0.600956
O	2.000520	1.808803	1.813310	O	-2.035102	-2.020721	-3.304786
C	-1.531446	1.026064	2.078898	O	1.824038	-3.420077	-1.414803
O	-2.363376	1.828840	2.156607	H	-2.936202	2.948214	0.222491
C	-2.866695	-0.511596	-0.947109	H	3.620275	2.198415	-0.334945
C	-3.858849	-0.449251	0.037988	C	-2.055641	0.522044	3.209509
C	-2.912364	-1.510967	-1.922829	H	-1.857078	1.593531	3.107030
C	-4.879039	-1.398452	0.053392	H	-2.521066	0.349734	4.184568
H	-3.806625	0.315920	0.802948	H	-1.088187	0.011738	3.206817
C	-3.939642	-2.450894	-1.904240	C	-3.549469	0.577253	-1.656403
H	-2.153563	-1.538789	-2.695271	H	-3.989401	-0.126101	-2.368390
C	-4.922002	-2.402610	-0.913820	H	-4.077832	1.532710	-1.770106
H	-5.635111	-1.356479	0.831871	H	-2.511463	0.748257	-1.947894
H	-3.969460	-3.223907	-2.666235	C	2.591264	0.506120	2.931854
H	-5.714176	-3.145031	-0.895105	H	2.747622	0.103001	3.936123
C	3.015624	-0.656798	-0.811950	H	3.020888	1.516014	2.902127
C	3.195289	-1.583275	-1.842071	H	1.517388	0.613472	2.765045
C	3.817228	-0.711422	0.333121	C	3.560584	-0.600897	-1.935529
C	4.175290	-2.566806	-1.722866	H	4.286060	0.160964	-2.250345
H	2.575491	-1.522007	-2.729476	H	3.755371	-1.500807	-2.525348
C	4.790646	-1.703361	0.446206	H	2.571821	-0.228821	-2.200197
H	3.658394	0.007506	1.128231	H	5.403952	-2.994269	1.621037
C	4.971939	-2.632831	-0.578464	H	-5.667214	-1.988587	1.620904
H	4.314474	-3.282823	-2.527383	-----			
H	5.403646	-1.750239	1.341477	iii S=1			
H	5.728526	-3.406260	-0.485623	-----			
-----				Co	0.540156	0.012262	0.409668
iib S=0				N	-0.023921	-0.497952	2.181280
-----				C	0.123597	-1.816331	2.536841
C	0.595066	5.213772	-0.709994	C	-0.531063	0.393146	3.086067
C	1.723764	4.384520	-0.700169	C	-0.265082	-2.266058	3.803838
C	1.564501	3.022723	-0.459113	C	-0.928246	-0.010401	4.353938
C	-0.793127	3.306340	-0.213566	C	-0.797473	-1.361814	4.719697
C	-0.667110	4.676772	-0.444693	H	-0.147903	-3.314742	4.059634
H	0.705714	6.277997	-0.888059	H	-1.332481	0.717466	5.050563
H	2.717557	4.792864	-0.853726	H	-1.102568	-1.693452	5.705799
H	-1.545241	5.313144	-0.399008	N	-0.351776	0.774047	-1.294847
N	0.311253	2.493171	-0.284603	C	-1.659547	0.759245	-1.609731
C	-1.928853	2.546849	0.165097	C	0.468786	1.663387	-1.905700
C	2.552343	2.022107	-0.240611	C	-2.195241	1.669932	-2.533966
N	2.076469	0.858323	0.163731	C	0.004315	2.587837	-2.836775

C	-1.359079	2.596029	-3.140807
H	-3.255865	1.617402	-2.747149
H	0.689963	3.292846	-3.293440
H	-1.756426	3.314057	-3.850351
C	0.714342	-2.613149	1.507530
H	0.940532	-3.664993	1.675721
C	-0.652957	1.745899	2.587066
H	-1.158952	2.492862	3.202526
C	1.871584	1.548716	-1.525295
H	2.621178	2.112667	-2.079470
C	-2.481854	-0.268713	-0.956734
H	-1.927826	-1.008437	-0.376045
N	1.003594	-2.018927	0.368153
N	2.187384	0.767676	-0.548047
N	-3.748132	-0.276934	-1.129417
N	-0.198315	2.026757	1.404519
C	-0.507555	3.272015	0.788819
C	0.533392	4.174634	0.475404
C	-1.839840	3.543861	0.405578
C	0.227080	5.313417	-0.272522
C	-2.093259	4.689149	-0.357579
C	-1.072101	5.564600	-0.709303
H	1.021569	6.018359	-0.501682
H	-3.114574	4.892846	-0.668102
H	-1.289322	6.450884	-1.297706
C	1.780836	-2.712962	-0.605920
C	3.114869	-3.068162	-0.316199
C	1.220201	-2.968000	-1.874424
C	3.880547	-3.654710	-1.330413
C	2.015722	-3.574404	-2.848930
C	3.343219	-3.908501	-2.587699
H	4.913843	-3.915275	-1.118348
H	1.582239	-3.790221	-3.821597
H	3.951132	-4.372301	-3.358635
C	3.558029	0.490012	-0.272676
C	4.029081	0.604146	1.052505
C	4.395818	0.031759	-1.316678
C	5.384770	0.364707	1.290810
C	5.734163	-0.232217	-1.014503
C	6.237685	-0.042211	0.269192
H	5.765557	0.481127	2.301497
H	6.384390	-0.598461	-1.803573
H	7.285890	-0.233388	0.477555
C	-4.620795	-1.212290	-0.563818
C	-5.765647	-1.494449	-1.353710
C	-4.462425	-1.807011	0.715807
C	-6.704806	-2.409949	-0.883635
C	-5.449887	-2.692928	1.154772
C	-6.551417	-3.009959	0.364472
H	-7.571269	-2.642009	-1.495914
H	-5.349660	-3.138617	2.140770
H	-7.298236	-3.708569	0.729387
C	3.134943	0.914123	2.229033
H	2.243285	1.475735	1.962050
H	2.801295	-0.015556	2.707575
H	3.682178	1.487407	2.983615
C	3.890107	-0.268264	-2.709815
H	2.870799	-0.664241	-2.696152
H	3.907179	0.609557	-3.366623
H	4.523114	-1.028937	-3.171990
C	1.935833	3.979735	0.994920
H	1.940837	3.812905	2.077184
H	2.439763	3.123713	0.540743
H	2.542448	4.866599	0.792861
C	-2.996301	2.658496	0.802986
H	-2.703714	1.614497	0.916302
H	-3.430024	2.979034	1.758563
H	-3.794034	2.698371	0.056230
C	-3.320813	-1.495360	1.654097
H	-2.425443	-2.088512	1.435529
H	-3.024032	-0.443231	1.618118

H	-3.608483	-1.723771	2.684200
C	-5.944618	-0.836378	-2.697446
H	-5.839808	0.250518	-2.618664
H	-5.189870	-1.175343	-3.416782
H	-6.930447	-1.065021	-3.111254
C	3.740666	-2.839893	1.037733
H	3.339135	-1.955047	1.533054
H	3.575974	-3.694229	1.706358
H	4.820416	-2.704535	0.938804
C	-0.217933	-2.632812	-2.176438
H	-0.885700	-2.947174	-1.367829
H	-0.357834	-1.556048	-2.313061
H	-0.544173	-3.128001	-3.095104

iii S=0

Co	0.482201	0.161399	0.579683
N	-0.433458	-0.056474	2.158515
C	-0.864734	-1.314134	2.506651
C	-0.657793	1.023352	2.976109
C	-1.562060	-1.509616	3.696055
C	-1.334595	0.858068	4.184093
C	-1.794291	-0.414943	4.540318
H	-1.911240	-2.501512	3.964015
H	-1.505359	1.712097	4.831499
H	-2.329016	-0.555901	5.472878
N	-0.184543	0.784491	-1.252270
C	-1.443578	0.940476	-1.689633
C	0.835325	1.311892	-1.970410
C	-1.727116	1.657595	-2.862737
C	0.628139	2.050117	-3.132102
C	-0.683699	2.228429	-3.578384
H	-2.761135	1.739078	-3.175880
H	1.471303	2.473687	-3.666190
H	-0.882208	2.798449	-4.480147
C	-0.439205	-2.262752	1.529174
H	-0.625195	-3.329828	1.617569
C	-0.160251	2.214105	2.370726
H	-0.315717	3.200399	2.800615
C	2.153207	0.960716	-1.461445
H	3.037851	1.253618	-2.021974
C	-2.496255	0.276590	-0.906174
H	-2.161174	-0.206353	0.014812
N	0.264591	-1.769125	0.524764
N	2.223743	0.250449	-0.383015
N	-3.693954	0.216587	-1.341567
N	0.440637	2.059272	1.203789
C	0.670975	3.225326	0.401243
C	1.980163	3.686729	0.153201
C	-0.443242	3.854904	-0.196999
C	2.158739	4.704467	-0.790547
C	-0.214595	4.862320	-1.138778
C	1.076952	5.270554	-1.458188
H	3.164691	5.067605	-0.982034
H	-1.067956	5.334883	-1.617242
H	1.237090	6.053376	-2.193368
C	0.903154	-2.713001	-0.353187
C	2.033192	-3.407369	0.126472
C	0.407776	-2.939714	-1.651845
C	2.701365	-4.274373	-0.743869
C	1.096951	-3.831979	-2.479590
C	2.246681	-4.482806	-2.041620
H	3.583379	-4.796893	-0.384668
H	0.710835	-4.024089	-3.476976
H	2.771420	-5.167145	-2.701528
C	3.469025	-0.358986	-0.036150
C	3.931499	-0.328582	1.299238
C	4.199354	-1.049760	-1.039657
C	5.173172	-0.903804	1.585501
C	5.417852	-1.634049	-0.688688

C	5.920758	-1.545504	0.605880
H	5.542234	-0.857001	2.606405
H	5.972974	-2.175502	-1.449053
H	6.878394	-1.993460	0.852737
C	-4.685646	-0.486145	-0.629549
C	-5.478940	-1.362407	-1.405222
C	-4.926146	-0.312676	0.753166
C	-6.475344	-2.103534	-0.770994
C	-5.955484	-1.057139	1.337453
C	-6.715092	-1.956530	0.594039
H	-7.076524	-2.791957	-1.358239
H	-6.167521	-0.915894	2.393996
H	-7.506428	-2.525433	1.072691
C	3.166097	0.264048	2.457668
H	2.257676	0.769622	2.149069
H	2.884282	-0.526085	3.163877
H	3.789137	0.977685	3.006447
C	3.699452	-1.265373	-2.451060
H	2.610324	-1.332565	-2.499307
H	4.024846	-0.476397	-3.139610
H	4.094706	-2.208900	-2.834007
C	3.169605	3.190360	0.934277
H	2.923739	3.069580	1.992154
H	3.541845	2.226625	0.580107
H	3.991989	3.907029	0.859680
C	-1.864836	3.508750	0.172419
H	-1.979872	2.468290	0.472727
H	-2.208730	4.124037	1.013763
H	-2.542093	3.698316	-0.664799
C	-4.172967	0.687513	1.599695
H	-3.224579	0.291446	1.976515
H	-3.948344	1.601040	1.041749
H	-4.771465	0.966848	2.471608
C	-5.212711	-1.519424	-2.879959
H	-5.135978	-0.543692	-3.370707
H	-4.263801	-2.039060	-3.064362
H	-6.008745	-2.095245	-3.359786
C	2.510596	-3.290375	1.552517
H	2.249024	-2.333214	2.001313
H	2.065731	-4.077262	2.175389
H	3.595930	-3.405111	1.603612
C	-0.865315	-2.310798	-2.156848
H	-1.558623	-2.088922	-1.344045
H	-0.673060	-1.377014	-2.694036
H	-1.369980	-2.986657	-2.853779

iv S=0

Co	0.026163	0.283897	-0.234859
N	0.122619	2.093169	-0.514898
C	1.353707	2.687126	-0.557978
C	-1.034981	2.822210	-0.517803
C	1.450311	4.074360	-0.653049
C	-0.976409	4.212642	-0.611234
C	0.274951	4.836306	-0.681090
H	2.423564	4.551342	-0.703303
H	-1.890609	4.796793	-0.628715
H	0.335062	5.916303	-0.757725
C	2.380881	1.694044	-0.517935
H	3.440977	1.929948	-0.559394
C	-2.164932	1.951240	-0.446529
H	-3.193236	2.302833	-0.455649
N	1.951284	0.444899	-0.461047
N	-1.873961	0.661163	-0.405896
N	0.072184	0.133629	1.759025
C	0.128711	0.135403	2.914001
N	-0.128657	-1.585064	-0.459898
C	-0.331951	-2.717122	-0.565874
C	0.200909	0.131893	4.369032
H	-0.692367	-0.341929	4.787384

H	1.084313	-0.423895	4.697984
H	0.266963	1.157905	4.743764
C	-0.638227	-4.129872	-0.715249
H	-0.253280	-4.694999	0.138991
H	-1.725347	-4.251904	-0.766911
H	-0.187703	-4.521529	-1.632564
C	-2.935916	-0.274980	-0.273137
C	-3.125386	-1.249301	-1.260277
C	-3.769351	-0.246753	0.850012
C	-4.147461	-2.186324	-1.121669
H	-2.486029	-1.243740	-2.137008
C	-4.786265	-1.192297	0.985598
H	-3.603919	0.506337	1.614848
C	-4.976812	-2.164592	0.003430
H	-4.309696	-2.923766	-1.902763
H	-5.429878	-1.167285	1.859729
H	-5.771880	-2.896159	0.108049
C	2.906744	-0.598984	-0.337171
C	2.893314	-1.659005	-1.250965
C	3.847683	-0.581384	0.699437
C	3.824907	-2.687209	-1.131795
H	2.172637	-1.648123	-2.061472
C	4.771916	-1.618925	0.817095
H	3.836711	0.236263	1.414245
C	4.763899	-2.672999	-0.097029
H	3.826535	-3.496964	-1.855516
H	5.498159	-1.602484	1.624135
H	5.486888	-3.477614	-0.006847

v S=0

Co	-0.026556	0.370483	-0.055958
N	-0.123231	2.121651	-0.588495
C	-1.353746	2.680755	-0.802492
C	1.033676	2.846016	-0.695856
C	-1.452743	4.032290	-1.124246
C	0.970850	4.201863	-1.015704
C	-0.280503	4.794638	-1.222654
H	-2.424167	4.480790	-1.304501
H	1.882857	4.782010	-1.110993
H	-0.342565	5.847155	-1.475980
C	-2.370927	1.676828	-0.704968
H	-3.425978	1.877086	-0.873202
C	2.165704	1.991120	-0.510494
H	3.193796	2.329801	-0.609134
N	-1.930168	0.464719	-0.433135
N	1.871010	0.727761	-0.263836
N	0.173498	-1.512046	-0.061799
C	0.454020	-2.631860	-0.096388
N	-0.195175	0.425950	1.939372
C	-0.363050	0.511182	3.080565
C	0.868060	-4.022487	-0.173337
H	1.953047	-4.054668	-0.319806
H	0.611250	-4.550400	0.749867
H	0.373331	-4.516491	-1.015245
C	-0.589387	0.615124	4.515045
H	-1.661775	0.551251	4.724745
H	-0.075749	-0.198274	5.036480
H	-0.210420	1.571767	4.887398
C	2.918617	-0.237945	-0.144666
C	3.622593	-0.369300	1.064024
C	3.159561	-1.087946	-1.240206
C	4.566738	-1.396874	1.167996
C	4.112287	-2.101065	-1.094101
C	4.808058	-2.262331	0.103841
H	5.118838	-1.512003	2.096573
H	4.318658	-2.756438	-1.936274
H	5.546121	-3.052612	0.202422
C	-2.851583	-0.622693	-0.331443
C	-3.695488	-0.737320	0.787185

C	-2.843898	-1.589304	-1.356170
C	-4.515034	-1.868597	0.879638
C	-3.682108	-2.698124	-1.224868
C	-4.507785	-2.845289	-0.111138
H	-5.168132	-1.975212	1.741585
H	-3.693831	-3.446786	-2.012321
H	-5.150459	-3.715774	-0.021379
C	-3.762435	0.326472	1.856918
H	-2.811576	0.848004	1.978632
H	-4.517957	1.085199	1.615839
H	-4.049338	-0.112947	2.817579
C	-2.005148	-1.399258	-2.594849
H	-2.411056	-0.598162	-3.225230
H	-0.978468	-1.113875	-2.351374
H	-1.982987	-2.312782	-3.195495
C	3.386224	0.576337	2.215285
H	3.810038	1.567974	2.014275
H	2.319514	0.719314	2.404563
H	3.856610	0.199207	3.127471
C	2.431122	-0.890558	-2.545530
H	1.347702	-0.840218	-2.399264
H	2.724358	0.050511	-3.026137
H	2.651754	-1.702776	-3.243295

unbridged Co2(CO)8

Co	-1.320629	-0.000367	0.000099
Co	1.320621	0.000359	-0.000086
C	-1.155710	-0.870402	-1.559917
O	-1.082264	-1.429445	-2.561636
C	1.156369	0.919015	-1.532080
O	1.083112	1.509540	-2.515578
C	-1.155726	1.785653	0.028195

O	-1.083190	2.932739	0.046070
C	3.083963	-0.000979	0.000334
O	4.230766	-0.002122	0.000846
C	1.155711	0.870323	1.559960
O	1.082631	1.429619	2.561563
C	1.155724	-1.785661	-0.028105
O	1.083207	-2.932740	-0.046393
C	-1.156397	-0.918993	1.532098
O	-1.083435	-1.509632	2.515548
C	-3.083970	0.000985	-0.000378
O	-4.230773	0.002112	-0.000547

bridged Co2(CO)8

Co	1.255352	-0.002743	0.002935
Co	-1.255338	0.002334	0.003274
C	-0.003344	-1.297381	-0.676045
O	-0.005306	-2.343448	-1.205446
C	-1.884778	-1.301578	1.083762
O	-2.317967	-2.123659	1.754608
C	1.872718	-1.305898	1.091348
O	2.296310	-2.127153	1.769337
C	-2.318576	-0.001793	-1.443305
O	-3.068804	-0.004367	-2.310397
C	-1.863139	1.321862	1.077368
O	-2.278695	2.154277	1.746702
C	0.002979	1.284545	-0.698923
O	0.004852	2.320853	-1.247267
C	1.874567	1.316923	1.070162
O	2.298501	2.149414	1.734116
C	2.319456	-0.016672	-1.442904
O	3.071149	-0.024542	-2.308708

Table S9. Frequencies (cm^{-1}) for all optimized structures. Some small ($< 10 \text{ cm}^{-1}$) imaginary frequencies corresponding to breathing motions were ignored.

-----				341.5936	359.3150	384.1354
i S=0				398.8903	412.7890	420.3373
-----				428.2372	436.4206	461.1922
10.7498	21.6465	23.6173		467.2561	472.6616	475.7521
34.8526	47.2050	52.5003		486.0711	489.4084	504.8159
58.7676	65.1113	71.5491		509.4458	510.7993	517.2259
80.2367	90.0683	97.1271		524.3828	528.0234	531.6558
99.7425	107.7153	113.5465		540.8711	551.1254	560.7543
125.2558	137.3234	146.6721		565.3480	574.9054	592.2354
158.9062	187.4721	189.7537		595.4075	606.0386	611.6816
207.2773	209.9958	236.7945		620.0627	639.4042	660.3645
253.3842	269.5767	302.6724		686.1783	707.0705	731.1910
325.8770	340.6824	350.4969		738.6568	758.3439	777.5657
354.8451	379.8190	389.3906		778.0886	787.0950	787.8940
421.4720	425.0470	430.1244		825.6086	847.8664	862.9828
443.6784	449.7950	470.9494		891.5678	903.2211	905.6982
480.3334	497.1534	504.7633		919.0942	926.9709	943.0251
513.3335	522.7326	527.0560		944.9633	963.1298	971.9910
529.1519	553.4961	557.6565		972.3031	1018.8748	1019.7790
563.8072	569.1997	576.8657		1057.7968	1064.1662	1066.2742
583.6752	592.4548	598.4086		1072.7853	1074.1358	1077.6797
610.3712	623.4934	624.7797		1081.3925	1100.1681	1103.9078
638.5320	639.2461	684.2258		1129.8068	1130.0144	1196.2647
693.2402	713.9073	715.8661		1202.2353	1202.5736	1223.0412
720.1014	775.7409	785.9568		1228.9283	1241.2681	1274.6433
788.0447	818.6396	853.1239		1289.1533	1290.3896	1291.8274
855.3336	869.8433	881.6908		1293.1591	1336.7742	1342.2484
894.3447	921.1888	924.4834		1343.2508	1390.2526	1400.5758
936.7735	947.3526	971.3231		1417.6121	1443.6708	1445.2105
971.4983	972.3368	996.2606		1446.1603	1448.1516	1449.2471
996.8624	1022.6557	1023.0488		1476.4473	1479.4379	1506.7564
1058.2620	1060.0947	1061.2178		1510.0028	1510.7128	1514.3960
1102.8260	1106.2207	1115.6200		1515.4345	1521.3721	1523.2959
1119.3482	1193.8779	1194.4859		1530.8275	1531.4192	1534.5132
1202.5226	1210.5432	1214.1231		1536.4669	1539.6384	1585.3225
1237.4619	1249.1317	1257.1260		1646.0551	1647.0712	1651.1798
1281.1095	1333.6296	1342.8720		1653.3527	1672.5996	1897.3579
1351.0151	1365.8825	1369.7448		1937.6625	2077.8113	2086.9772
1395.1626	1402.0129	1422.1569		2136.6502	3046.8504	3047.7188
1480.9106	1499.2153	1501.0406		3049.2066	3065.5320	3116.2548
1521.7658	1541.1553	1544.9757		3118.2276	3119.9466	3129.6720
1550.8842	1594.8224	1644.3654		3138.6553	3155.4196	3161.4496
1645.0100	1657.0632	1659.7248		3183.9574	3184.5369	3191.4043
1672.3056	2009.4271	2034.7378		3193.0401	3196.0464	3199.4826
2050.7217	2089.0450	2119.3720		3205.1388	3206.9867	3209.4711
3192.0121	3193.2182	3202.1905		3210.3947	3214.5555	3227.8886
3203.4496	3207.9193	3209.9972				
3210.9304	3213.8666	3215.7667		-----		
3219.6422	3227.1345	3230.3012		iii S=1		
3237.1341	3239.1961	3242.7188		-----		
-----				-8.9350	18.8664	24.3547
iib S=0				32.6931	38.1400	46.0139
-----				47.5540	50.1817	56.6246
14.9671	31.4686	38.2635		63.3100	65.9388	75.5850
43.8557	51.8994	62.6645		78.4019	84.9926	88.1242
65.1936	71.0813	73.6002		94.6753	107.1836	112.4276
81.3943	90.5218	93.3068		117.7107	120.4553	123.8477
95.8792	101.4699	105.8782		136.3279	142.2820	144.6262
107.2259	116.6312	124.0561		148.1714	153.9193	158.6681
134.4262	146.0761	150.1385		160.4021	167.1237	173.4105
161.9787	166.8863	172.0600		175.4467	183.8170	186.2007
185.9031	188.7494	201.6123		190.2816	195.6214	202.4845
205.7022	218.5411	236.1824		217.7515	220.9238	225.4809
243.3687	254.2569	297.1282		227.7333	234.6465	235.3314
299.7446	308.6762	323.7695		247.5112	251.7340	267.3316
328.1348	336.8311	339.6822		274.1069	284.4827	305.5196
				314.8060	322.2928	328.3444

332.8137	336.4093	340.8684	3140.0677	3141.1647	3151.8457
348.3051	357.3404	364.1100	3152.9635	3166.9254	3169.8554
374.7448	379.5538	381.6038	3170.1574	3171.2888	3186.8767
393.1278	417.2884	426.1334	3187.4655	3187.5926	3192.8932
446.3110	457.3726	463.4484	3193.0740	3193.9848	3197.9844
472.7445	507.3035	509.2164	3199.3366	3200.4897	3209.3881
512.2128	513.8392	515.5878	3213.3871	3213.7425	3213.7995
517.6631	518.8663	521.2931	3215.8374	3217.4748	3220.4532
525.0545	528.9038	532.9074	3233.3524	3234.1231	3250.1779
542.6225	547.3131	561.5859			
563.7770	565.4945	574.0830	-----		
581.0417	593.3817	599.4705	iii S=0		
618.6049	627.2912	638.2630	-----		
641.1824	642.8884	651.0923	-7.6954	20.6649	25.8132
708.9700	715.7443	717.9588	35.1758	39.2491	48.2105
732.9053	734.5567	745.7378	53.7542	57.5954	64.3516
755.8278	757.2928	765.9191	65.8178	68.8380	76.1615
773.9675	789.1644	791.3405	84.9233	91.8112	106.0931
792.8985	793.8071	796.1425	107.3920	121.3407	128.3222
818.1743	818.9375	823.8100	135.8937	141.2460	149.6515
868.5228	873.4760	892.9879	154.7579	163.1313	167.7507
897.9591	906.0658	910.0678	170.7775	174.8528	176.4481
911.3465	911.8740	914.8949	177.9658	184.7220	186.8474
924.2427	935.9933	940.5649	188.2624	191.3379	201.2780
942.5505	947.0525	949.7373	207.1030	209.0770	216.1824
964.0894	974.9867	975.6306	217.6180	222.8563	227.7928
976.0548	976.8087	983.0336	232.6402	243.7033	245.1935
985.1445	985.7190	1015.6417	254.3641	269.5646	288.5034
1017.0643	1017.5267	1018.7956	291.9989	306.9674	312.5228
1019.7385	1027.2068	1034.9712	320.3850	336.8027	342.1866
1036.8848	1060.6100	1062.7105	344.1128	348.3015	354.4530
1064.3309	1066.8162	1070.1192	364.6800	377.1377	383.2802
1071.4630	1073.9329	1075.0908	406.6044	421.0890	427.3702
1077.4867	1077.7950	1078.2778	434.0070	438.1933	455.8353
1081.2336	1081.9123	1101.3556	462.4662	464.8602	469.5491
1116.4052	1130.1934	1130.9040	496.1054	507.2562	512.2598
1131.6538	1132.7698	1162.2823	513.8067	517.0734	521.3376
1194.6148	1203.4212	1204.1576	522.4292	524.4492	525.0058
1205.6784	1205.8315	1212.9488	527.8277	530.4895	532.7761
1219.6348	1223.0422	1230.9840	549.4453	555.0004	569.3807
1235.6901	1260.3006	1280.4648	578.4986	580.0524	583.8127
1282.2800	1287.1152	1287.7729	599.0599	606.0353	610.7851
1289.8354	1292.5148	1292.9638	612.7254	630.6232	637.4966
1293.7168	1294.8117	1313.6522	641.9843	649.4999	657.8841
1325.7069	1336.9522	1337.4826	702.2907	714.4712	728.1764
1340.4114	1342.1386	1349.4972	732.3370	739.3186	746.4724
1379.3127	1390.4412	1399.5743	752.5564	758.9263	768.2028
1411.7926	1415.5753	1441.9650	777.6580	785.0996	791.3153
1443.5576	1444.2356	1444.7925	795.4144	796.4254	800.4169
1445.9083	1447.6143	1448.9288	819.4618	821.4492	824.3719
1449.5282	1450.6151	1460.0756	873.8834	880.0424	887.5181
1465.8688	1473.8893	1476.2678	892.4605	893.3333	910.5248
1479.2637	1503.8786	1506.9601	911.6802	913.6656	917.9585
1508.0460	1509.3876	1510.9982	919.8081	929.7301	933.3300
1511.8423	1512.3135	1513.8088	944.0985	947.0221	947.7314
1514.6144	1517.2327	1519.0910	949.3390	973.7935	977.5063
1520.3001	1523.1364	1523.5149	978.9057	981.7135	982.3396
1526.8372	1530.0655	1530.7650	988.0983	1003.5578	1015.6604
1531.6354	1532.7220	1532.9704	1016.1156	1017.4265	1020.2490
1535.0379	1547.3003	1579.4975	1022.0360	1026.0251	1034.6396
1607.5117	1615.1315	1629.7312	1048.0221	1057.1293	1059.7290
1634.6346	1637.4889	1639.9909	1062.4247	1068.7602	1071.8897
1640.9810	1641.3072	1644.1385	1073.4343	1074.2484	1074.8454
1650.0731	1652.8057	1656.0766	1077.3031	1077.5040	1077.9433
1669.6241	1701.3630	3054.0181	1078.3968	1091.4952	1094.3550
3054.5842	3054.9581	3055.1842	1115.5734	1128.1800	1132.7311
3055.5212	3056.3916	3063.8272	1133.9409	1136.3563	1181.3652
3070.8207	3110.7006	3113.3026	1193.3465	1204.3978	1204.5124
3113.5616	3120.9894	3121.5151	1205.2897	1206.7687	1215.1136
3124.4569	3126.7743	3130.1991	1215.3967	1219.1158	1231.2861
3131.9428	3135.1208	3138.2527	1240.8199	1259.6707	1281.3375

1283.3557	1284.4669	1288.9885
1291.6503	1292.9356	1293.0822
1294.0616	1294.7294	1311.5443
1316.5843	1334.9519	1337.8458
1338.4380	1342.3199	1346.3045
1377.4588	1383.0900	1400.4849
1410.5059	1415.4565	1424.4555
1440.0903	1440.5583	1443.4366
1445.1767	1446.0863	1447.3039
1448.5165	1451.8765	1463.2098
1465.3196	1472.9647	1476.0968
1480.6699	1503.6292	1507.8127
1508.9563	1510.2229	1511.6901
1512.9517	1514.2657	1515.9524
1516.9688	1517.6016	1518.2024
1519.7340	1522.9774	1525.6014
1527.3689	1528.9817	1531.8667
1532.3280	1534.6394	1535.9499
1536.5138	1545.1589	1559.3638
1579.6439	1608.6311	1631.9154
1635.5368	1638.9279	1639.8965
1644.3830	1645.0050	1647.5417
1650.1735	1650.8773	1658.6955
1666.7129	1711.5293	3048.5861
3051.9204	3054.2922	3055.4048
3059.9726	3067.3118	3070.8889
3081.6924	3104.9392	3109.2126
3120.2375	3121.5760	3125.3403
3125.9403	3128.8995	3131.1480
3139.7545	3141.4515	3146.4325
3154.9650	3155.4666	3158.7498
3180.8832	3184.4422	3186.3428
3187.4238	3188.4811	3189.1479
3190.8910	3195.0868	3195.3626
3195.4906	3196.5029	3198.8043
3200.4794	3213.9795	3214.4166
3214.6194	3215.0958	3217.8442
3218.2633	3220.7894	3232.4558
3232.6199	3235.5103	3246.4320

iv S=0

-8.2143	7.8886	18.4529
28.4829	29.1239	39.7168
46.2490	55.5719	64.7793
67.0926	77.7068	87.0585
115.7231	140.1478	156.6443
177.5982	184.8118	197.2840
210.2033	225.0521	229.7153
246.5349	250.6628	270.9966
311.8627	336.1176	342.7681
377.4563	385.9497	388.6973
411.9665	422.5925	423.0579
426.3435	443.7895	451.6884
454.6486	488.1220	517.3321
527.5474	560.7442	584.4299
604.5418	610.3326	620.0232
636.1238	636.7836	680.2983
690.8885	697.5935	713.3499
714.9660	777.8109	787.3103
787.5238	815.5378	850.3796
853.8310	863.3305	876.2759
894.0815	919.5176	929.4067
932.9501	943.4639	950.6259
970.4866	971.5475	972.2686
979.1914	1002.1073	1002.2411
1020.7659	1021.4408	1056.5931
1056.9101	1059.6687	1064.3542
1067.0383	1067.5608	1068.2232
1091.8266	1102.1564	1110.4042

1112.2181	1187.4802	1195.0970
1195.6317	1204.2878	1205.7184
1229.8512	1246.6004	1251.0726
1280.8626	1319.6147	1339.7839
1341.7423	1362.5127	1364.1749
1387.3236	1404.3935	1419.3579
1429.7028	1433.6016	1435.5957
1487.8638	1488.8396	1489.6161
1490.0524	1498.3820	1499.7644
1514.2818	1538.1671	1541.5710
1560.3594	1582.5105	1640.9782
1641.1310	1650.9074	1656.6760
1663.7046	2394.9206	2410.6493
3066.6260	3071.4607	3144.3618
3148.1148	3149.8114	3149.8629
3193.0950	3194.6114	3197.4498
3199.3980	3202.2146	3202.7071
3209.2211	3209.8929	3214.8830
3216.4391	3218.8652	3222.8812
3223.0520	3224.8781	3235.3568

v S=0

7.5851	13.0702	25.6837
33.4680	34.5494	39.2830
41.1133	51.6313	62.5942
70.0900	75.4225	81.6503
96.4620	106.1073	111.3075
124.1828	132.3170	133.6959
147.9890	150.6843	159.0864
177.4506	185.2218	195.5174
198.5050	221.6668	231.7857
233.4350	235.2580	244.0635
265.6348	297.5121	304.2501
311.9237	321.1392	334.4717
346.4993	378.4001	386.1520
387.6624	407.9971	411.8960
425.3476	435.5277	450.8637
465.6994	470.4383	488.1845
506.5945	510.0376	522.5081
523.9310	530.8926	559.7716
576.4919	578.6841	598.8311
607.6299	610.0413	637.8964
660.9911	695.9154	729.0548
738.3306	761.2364	779.4291
785.5026	794.0137	794.2668
820.2598	865.9477	880.5632
891.6133	911.1975	912.6194
925.7955	930.1844	942.4856
945.4415	950.4734	971.6117
977.4149	980.2842	982.3576
1018.7128	1020.9635	1058.2777
1064.5507	1065.9062	1066.0607
1068.1098	1069.0776	1069.7726
1071.3218	1073.3169	1075.9114
1077.2468	1085.8288	1100.5478
1128.0640	1130.6394	1184.9309
1203.6025	1204.9693	1221.3110
1227.7964	1234.5116	1277.9618
1288.0096	1290.1359	1293.9713
1294.9512	1318.1239	1341.4219
1341.7403	1381.7742	1400.4620
1413.4368	1428.6822	1431.6017
1433.0441	1440.3025	1442.1784
1444.9548	1447.3995	1475.6284
1480.3375	1487.4368	1488.4871
1488.8559	1489.2787	1504.9242
1509.0691	1510.9240	1512.1937
1516.8052	1521.1898	1525.5741
1530.7408	1532.5879	1534.0954

1535.5256	1567.8043	1582.0260
1644.1364	1645.2575	1649.5794
1651.6553	1658.7636	2391.2338
2411.2114	3051.4171	3055.3763
3056.9598	3057.6559	3066.3201
3070.3007	3107.3628	3111.0451
3114.8683	3119.5381	3141.8456
3144.7423	3147.1327	3148.1052
3149.2940	3150.3265	3155.1499
3161.8947	3187.2378	3189.0824
3195.2656	3195.3055	3198.9753
3199.5937	3214.3900	3215.5011
3216.1503	3222.5107	3235.3096

unbridged Co2 (CO) 8

22.8490	24.4557	54.0847
58.2247	59.1891	73.2400
73.8847	82.2795	82.4373
94.4677	94.5925	102.5125
102.6188	103.3758	105.4345
170.6639	314.2522	324.7259
326.8377	326.9815	335.4918
335.6166	403.5475	428.6094
428.6526	437.8580	466.5484
467.2250	470.3213	496.3478

499.2270	499.5802	508.5809
508.9239	551.3650	551.3803
551.4581	551.5003	554.0748
572.6726	2092.0770	2092.3461
2099.3683	2110.6941	2110.7414
2127.8491	2150.3726	2195.3547

bridged Co2 (CO) 8

23.6912	38.4660	38.8260
63.2649	68.2206	75.6973
76.1940	86.8009	88.2932
92.4025	94.8356	109.1090
111.4334	223.2486	235.1909
255.1524	315.2740	315.6078
329.7798	353.5114	356.9798
361.7722	417.6901	431.4739
435.2898	436.4138	446.3864
449.7584	464.4920	477.6771
478.3903	479.9080	516.5748
521.8719	525.9032	538.7023
565.3561	577.8573	604.1686
701.8616	1943.6026	1958.1314
2120.8279	2121.5715	2128.6342
2131.5245	2153.1688	2195.5688

Table S10. Energies (E_h) for all optimized structures.

Species	E(SCF)	H(gas)	G(gas)
i	-4229.517222	-4229.143570	-4229.240930
iib	-4386.810610	-4386.319646	-4386.427241
iii _{S=1}	-3491.957048	-3491.089139	-3491.220039
iii _{S=0}	-3491.954578	-3491.085076	-3491.210498
iv	-2545.494020	-2545.073675	-2545.163084
v	-2702.793091	-2702.253493	-2702.359224
unbridged Co ₂ (CO) ₈	-3672.090550	-3672.001451	-3672.078294
bridged Co ₂ (CO) ₈	-3672.098101	-3672.008981	-3672.083200

9. References

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