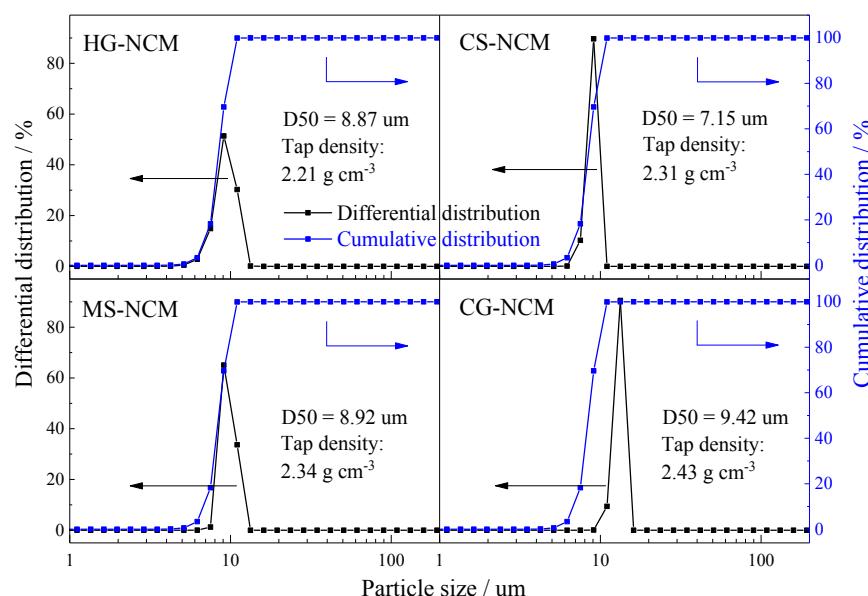


## Supplementary Information

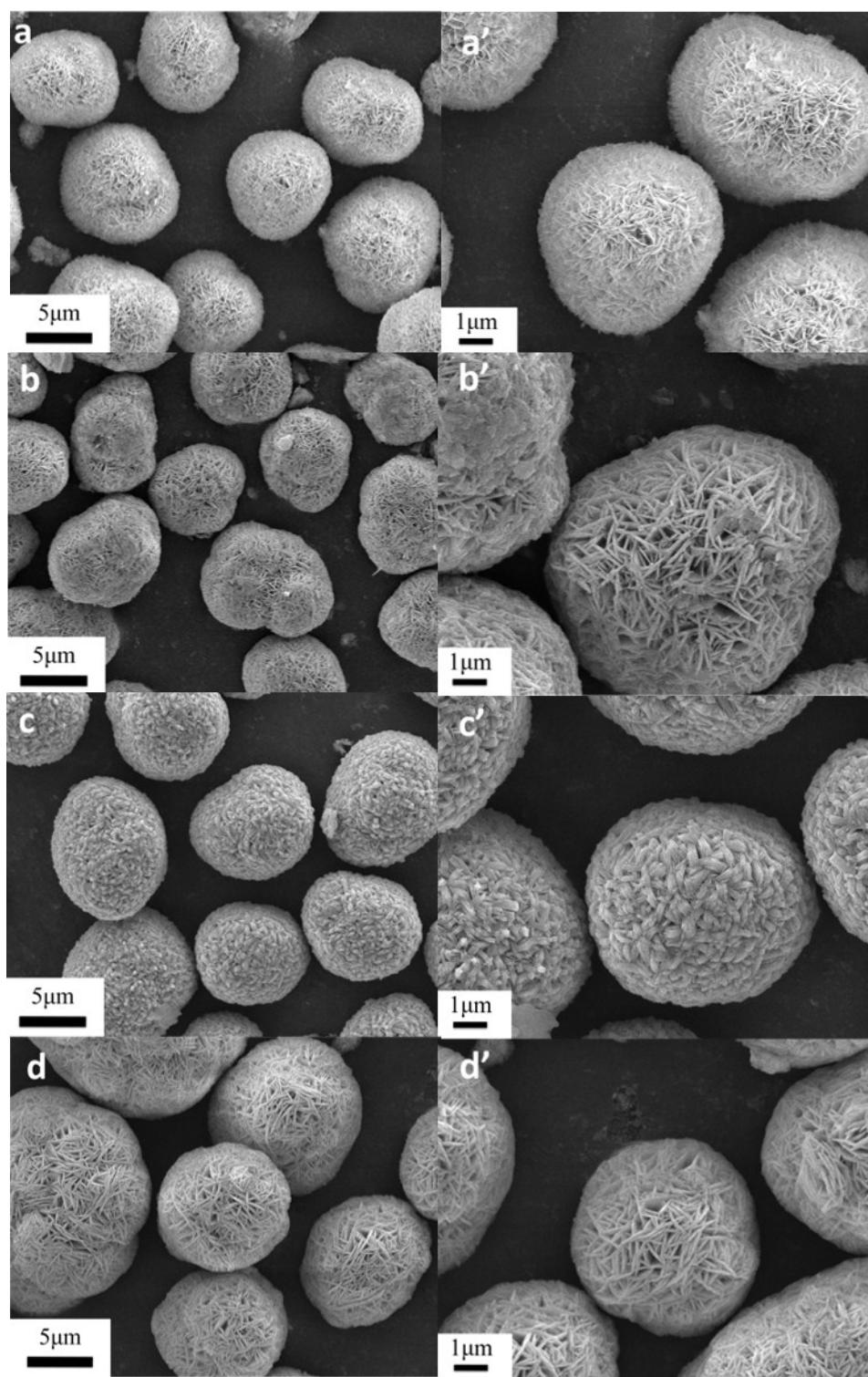
### Influence of Ni/Mn distributions on the structure and electrochemical properties of Ni-rich cathode material

Yiming Sun, Zhikun Zhang, Huanhuan Li, Tao Yang, Hongzhou Zhang, Xixi Shi,

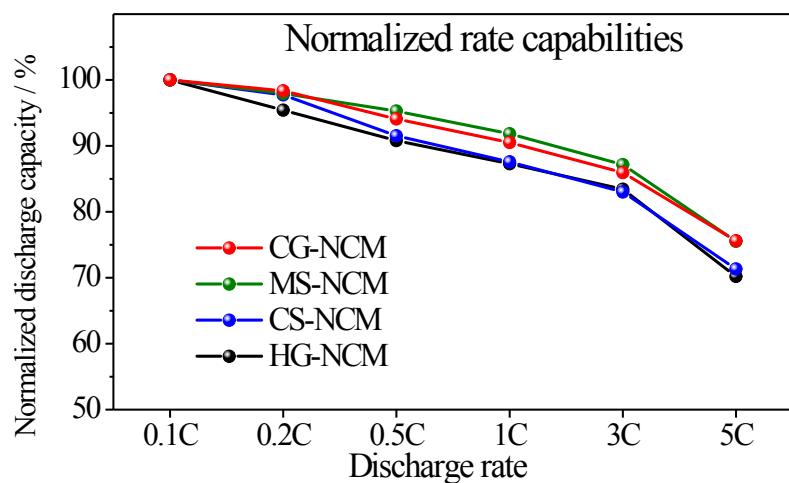
Dawei Song and Lianqi Zhang



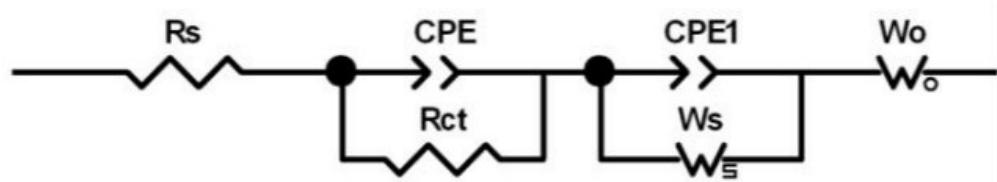
**Fig. S1** Particle size distributions of NCM materials with different Ni/Mn distributions.



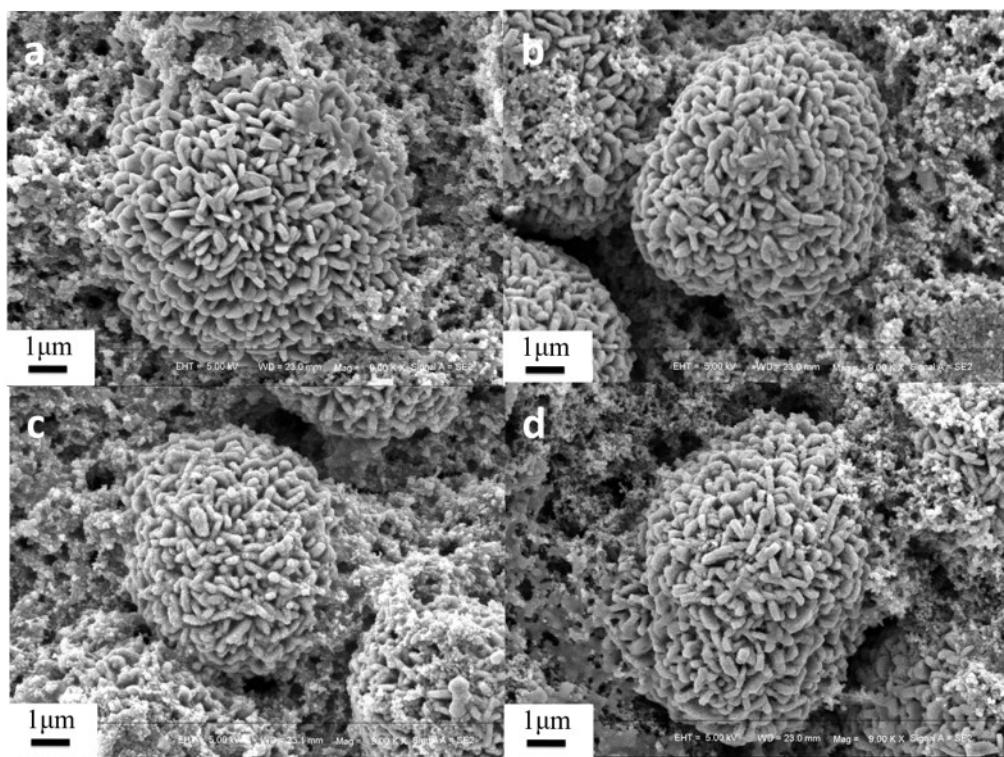
**Fig. S2** SEM images of (a, a') HG-NCM, (b, b') CS-NCM, (c, c') MS-NCM and (d, d') FCG-NCM precursors.



**Fig. S3** Normalized rate capacity vs. discharge rate of NCM cathodes between 3.0 and 4.4 V.



**Fig. S4** Equivalent circuits used to fit the experimental data.  $R_s$  is solution resistance,  $R_{ct}$  is interfacial charge-transfer resistance, CPE is a constant phase element,  $W_o$  is assigned to the semi-infinite Warburg diffusion impedance in the bulk.



**Fig. S5** SEM images of (a) HG-NCM, (b) CS-NCM, (c) MS-NCM and (d) FCG-NCM cathodes

after 200 cycles.

**Table S1** The simulated results from electrochemical impedance spectra and Li<sup>+</sup> diffusion coefficients obtained at the 10th cycle.

	Re ( $\Omega$ )	Rct ( $\Omega$ )	$D_{Li^+}$ ( $cm^2 s^{-1}$ )
HG-NCM	2.128	48.16	$2.53 \times 10^{-13}$
CS-NCM	2.061	31.04	$3.52 \times 10^{-1}$
MS-NCM	1.243	22.27	$1.95 \times 10^{-12}$
CG-NCM	5.607	10.06	$2.97 \times 10^{-12}$

**Table S2** The simulated results of Ni 2p<sub>3/2</sub> spectra of NCM cathode materials with different Ni/Mn distributions.

		Ni <sup>3+</sup>		Ni <sup>2+</sup>		Ni <sup>3+</sup> /(Ni <sup>3+</sup> +Ni <sup>2+</sup> ) (%)
		Peak (eV)	FWHM (eV)	Peak (eV)	FWHM (eV)	
Fresh samples	HG-NCM	856.1	1.83	854.8	1.42	75.0
	CS-NCM	856.0	1.83	855.0	1.42	71.4
	MS-NCM	856.1	1.83	855.0	1.42	70.6
	CG-NCM	856.1	1.83	855.0	1.42	72.6
After 200 cycles	HG-NCM	856.1	1.83	855.0	1.42	35.3
	CS-NCM	856.1	1.83	855.0	1.42	47.5
	MS-NCM	856.1	1.83	855.0	1.42	54.3
	CG-NCM	856.1	1.83	855.0	1.42	62.0