

Supporting Information

Top-Down Fabrication of Hierarchical Nanocubes on Nanosheets Composite for High-Rate Lithium Storage

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This file includes Figure S1-S9 and Table S1.

Figure S1.

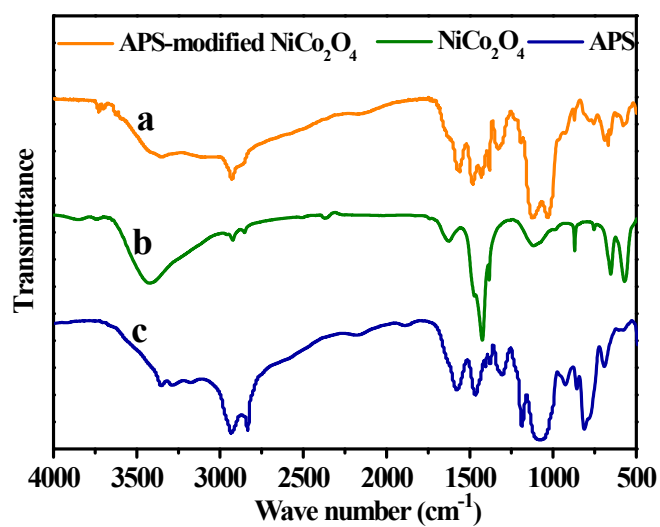


Figure S1. FTIR spectra of (a) APS-modified NiCo_2O_4 spheres, (b) pure NiCo_2O_4 spheres and (c) pure APS.

Figure S2.

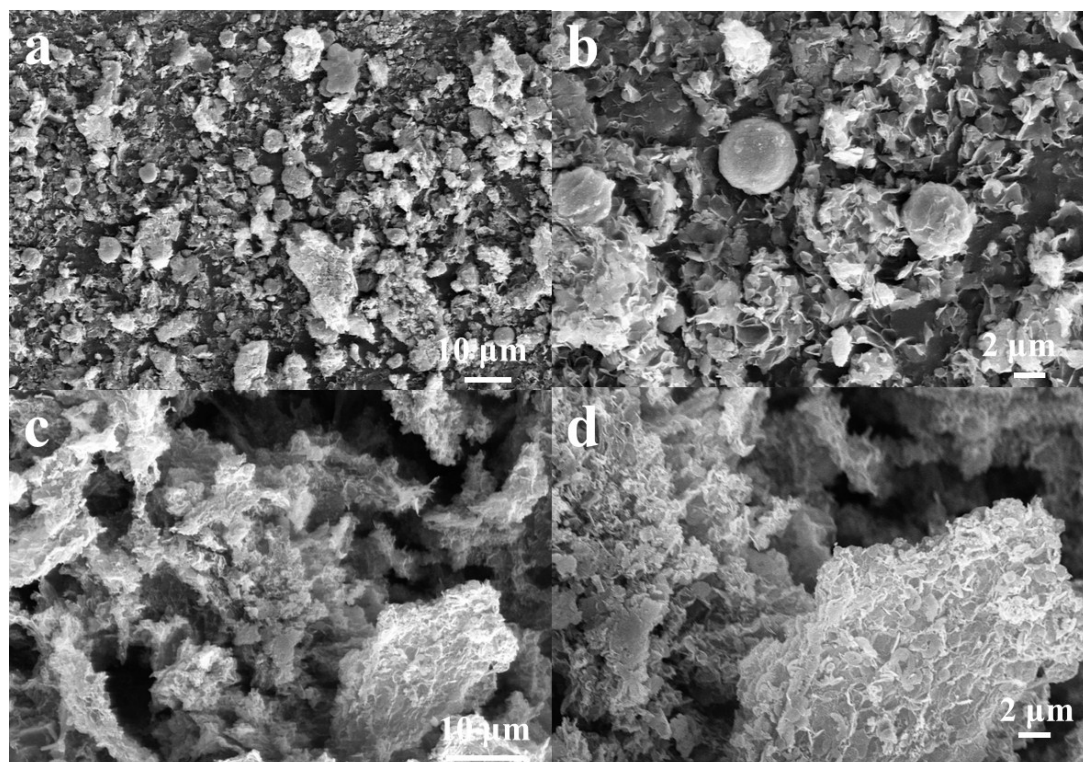


Figure S2. Low- and high-magnification SEM images of composites with (a,b) a small amount of GO and (c,d) excessive GO.

Figure S3.

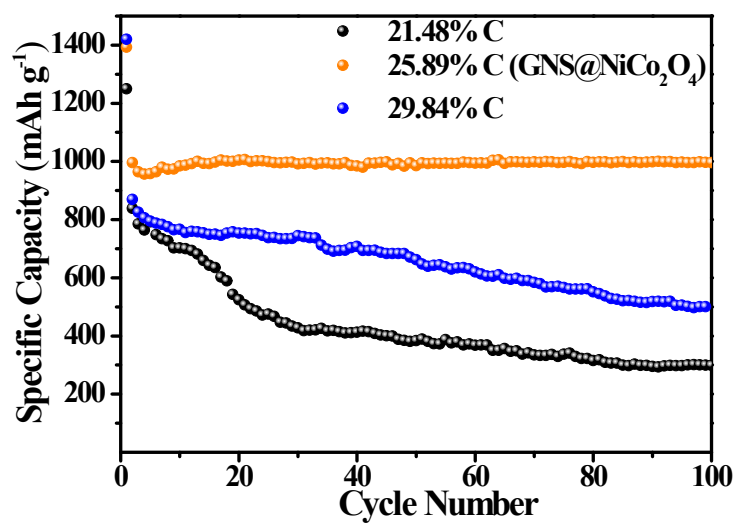


Figure S3. Cycling performance of the GNS@NiCo₂O₄ composites with different carbon contents cycled at a current density of 100 mA g⁻¹ for 100 cycles.

Figure S4.

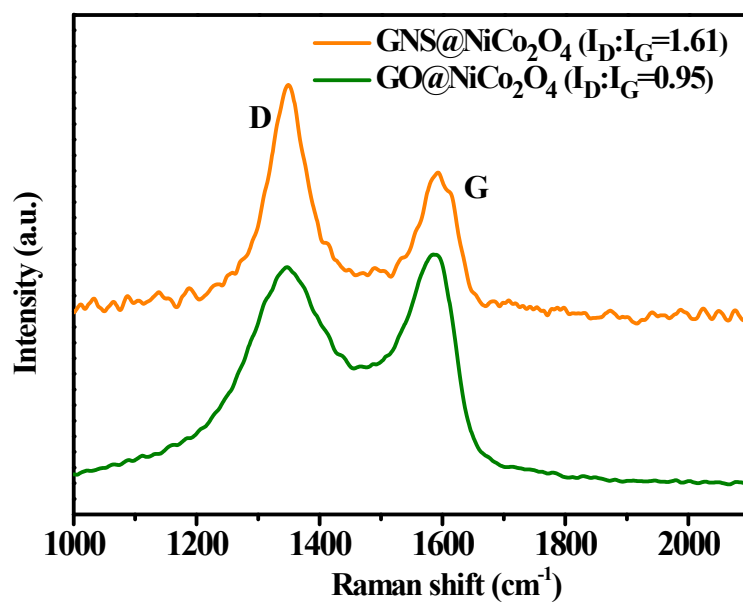


Figure S4. Raman spectra of GNS@NiCo₂O₄ and GO@NiCo₂O₄.

Figure S5.

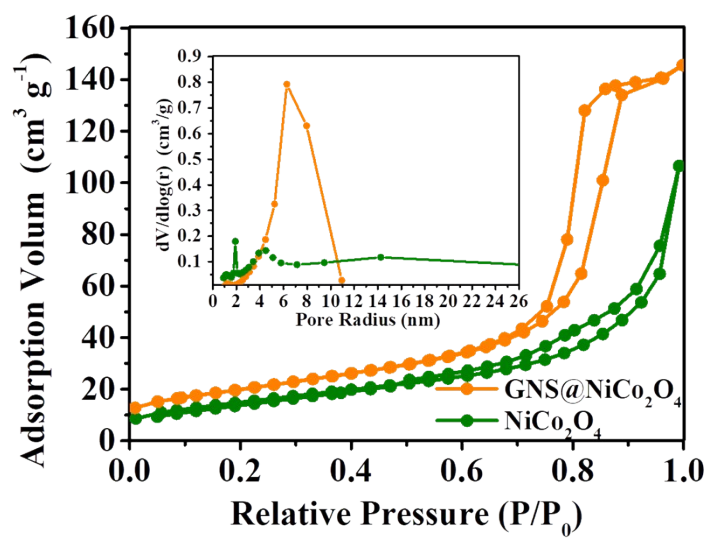


Figure S5. Nitrogen adsorption-desorption isotherms and BJH pore size distributions of NiCo_2O_4 spheres and $\text{GNS@NiCo}_2\text{O}_4$ composites.

Figure S6.

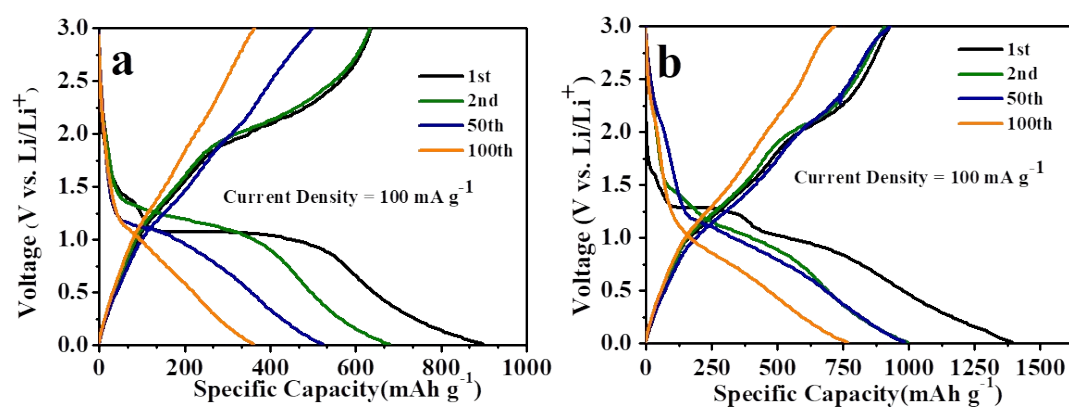


Figure S6. Galvanostatic discharge-charge voltage profiles of (a) NiCo₂O₄ spheres and (b) GNS@NiCo₂O₄ composites at a current density of 100 mA g⁻¹.

Figure S7.

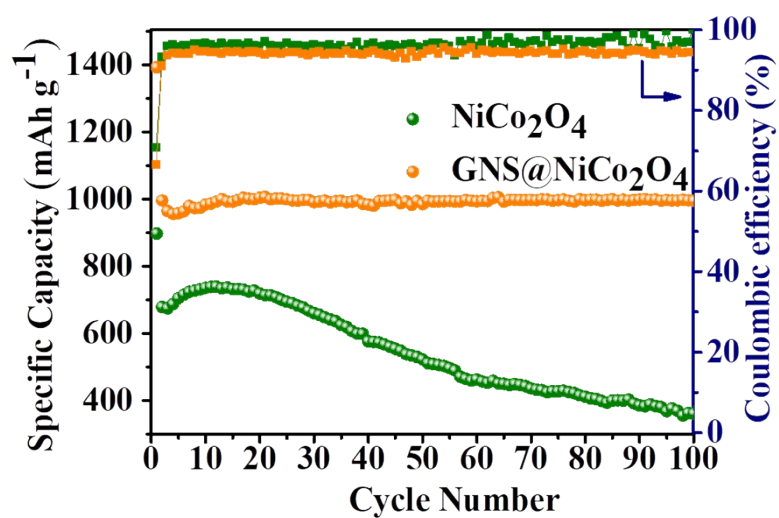


Figure S7. The comparative cycling performance of the $\text{GNS@NiCo}_2\text{O}_4$ composites and NiCo_2O_4 spheres at a current density of 100 mA g^{-1} for 100 cycles.

Figure S8.

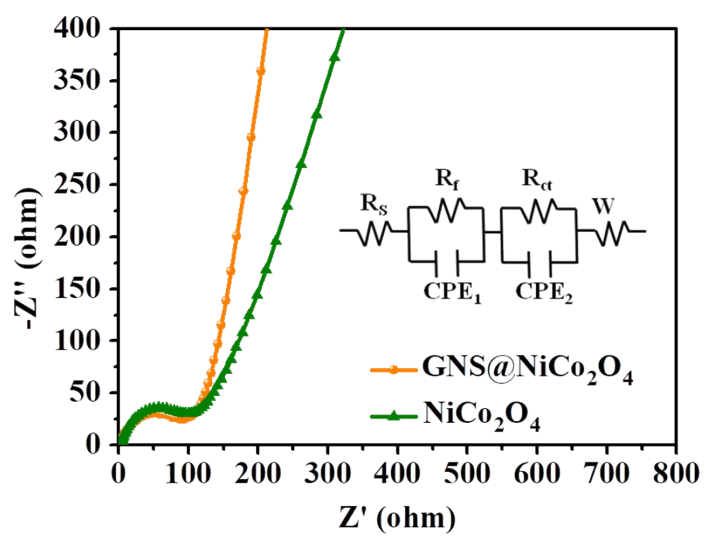


Figure S8. The EIS spectras of GNS@NiCo₂O₄ composites and NiCo₂O₄ spheres before cycles at a frequency range of 10^5 Hz to 0.1 Hz.

Figure S9

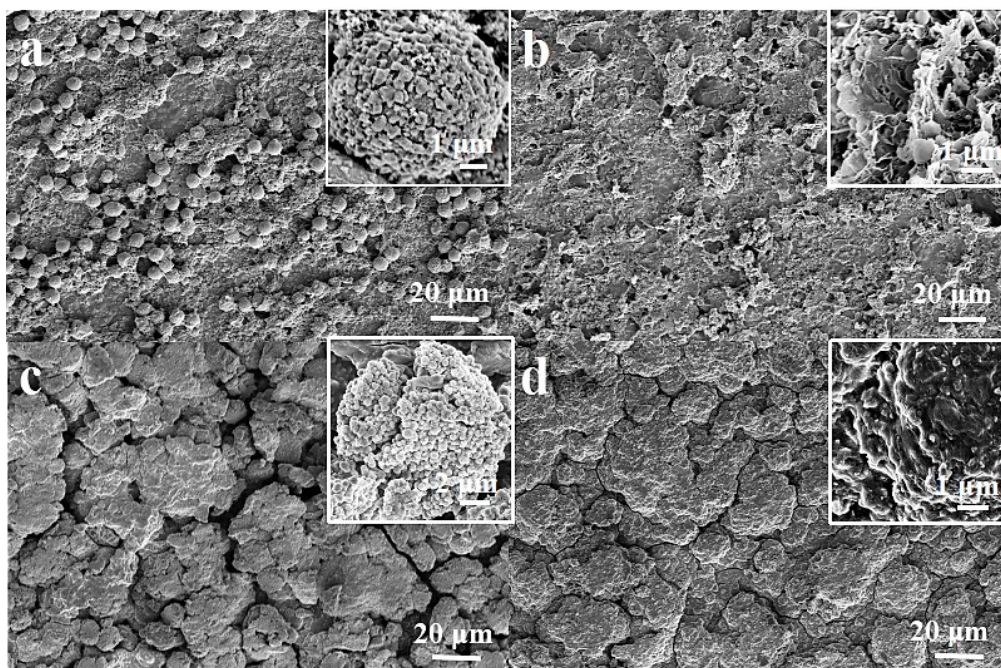


Figure S9. SEM images of the electrodes on copper foil (a,b) before and (c,d) after discharged/charged at a current density of 500 mA g⁻¹ for 200 cycles. (a,c) bare NiCo₂O₄ spheres; (b,d) GNS@NiCo₂O₄ composites.

Table S1. Comparing with NiCo₂O₄ materials or its composite with conductive matrix as anode for LIB reported in previous literature

Materials	Current Density	Reversible capacity	Ref.
	(mA g ⁻¹)	(mA h ⁻¹) / cycle number	
NiCo ₂ O ₄ -C nanorods	500	863/200th	1
NiCo ₂ O ₄ @C/carbon cloth composite	500	807/100th	2
NiCo ₂ O ₄ flower-like structure	100	939/60th	3
NiCo ₂ O ₄ dried plum-like spheres	100	801/50th	4
NiCo ₂ O ₄ nanoplates on RGO	100	816/70th	5
NiCo ₂ O ₄ nanorods	100	856/100th	6
This work	500	1024/200th	
	100	998/100th	

References

- 1 L. Peng, H. Zhang, Y. Bai, J. Yang and Y. Wang, *J. Mater. Chem. A*, 2015, **3**, 22094-22101.
- 2 K. Wang, Y. Huang, M. Wang, M. Yu, Y. Zhu and J. Wu, *Carbon*, 2017, **125**, 375-383.
- 3 L. Li, Y. Cheah, Y. Ko, P. Teh, G. Wee, C. Wong, S. Peng and M. Srinivasan, *J. Mater. Chem. A*, 2013, **1**, 10935-10941.
- 4 T. Li, X. Li, Z. Wang, H. Guo and Y. Li, *J. Mater. Chem. A*, 2015, **3**, 11970-11975.
- 5 Y. Chen, M. Zhuo, J. Deng, Z. Xu, Q. Li and T. Wang, *J. Mater. Chem. A*, 2014, **2**, 4449-4456.
- 6 F. Fu, J. Li, Y. Yao, X. Qin, Y. Dou, H. Wang, J. Tsui, K. Y. Chan and M. Shao, *ACS Appl. Mater. Interfaces*, 2017, **9**, 16194-16201.