

Synthesis and characterisation of homoleptic 2,9-diaryl-1,10-phenanthroline copper(II) complexes: influencing selectivity in photoredox-catalysed atom-transfer radical addition reactions

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Supplementary Information

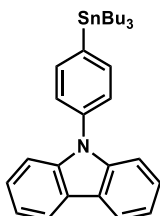
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I. Preparation of Materials

Experimental procedures for the preparation of previously unreported compounds are described below. These procedures were not optimised.

9-(4-(Tributylstannyl)phenyl)-9*H*-carbazole



n-BuLi (1.46 mL, 1.6 M in hexane, 2.33 mmol) was added drop wise to a magnetically stirred solution of 9-(4-bromophenyl)-9*H*-carbazole¹ (500 mg, 1.55 mmol) in THF (5 mL) maintained at –78 °C under N₂. After 1 h, Bu₃SnCl (505 mg, 1.55 mmol) was added dropwise. The ensuing mixture was slowly warmed to room temperature over 4 h and then concentrated under reduced pressure. The resulting pale-yellow residue was redissolved in CH₂Cl₂ (10 mL) and filtered through a Celite® plug then concentrated under reduced pressure. The ensuing residue was subjected to flash column chromatography (hexanes; silica gel) to provide the 9-(4-(tributylstannyl)phenyl)-9*H*-carbazole (610 mg, 74% yield) as a colourless oil.

¹H NMR (400 MHz, CDCl₃): δ 8.17 (d, *J* = 7.8 Hz, 2H), 7.77–7.65 (m, 2H), 7.54 (d, *J* = 7.9 Hz, 2H), 7.50 (complex m, 4H), 7.31 (t, *J* = 7.2 Hz, 2H), 1.70–1.60 (m, 6H), 1.21–1.14 (m, 6H), 0.97–0.94 (m, 9H) ppm.

¹³C NMR (100 MHz, CDCl₃): δ 141.8, 141.0, 137.9, 137.6, 126.5, 126.0, 123.5, 120.4, 120.0, 110.0, 29.3, 27.6, 13.8, 9.9 ppm.

IR (NaCl): 1694, 1653, 1586, 1559, 1501, 1452, 1231, 748, 725 cm⁻¹.

1. G. K.-M. So, G. Cheng, J. Wang, X. Chang, C.-C. Kwok, H. Zhang, C.-M. and W. Sun, *Chem. – Asian J.* 2013, 12, 2017, 1490.

II. Cyclic Voltammetry

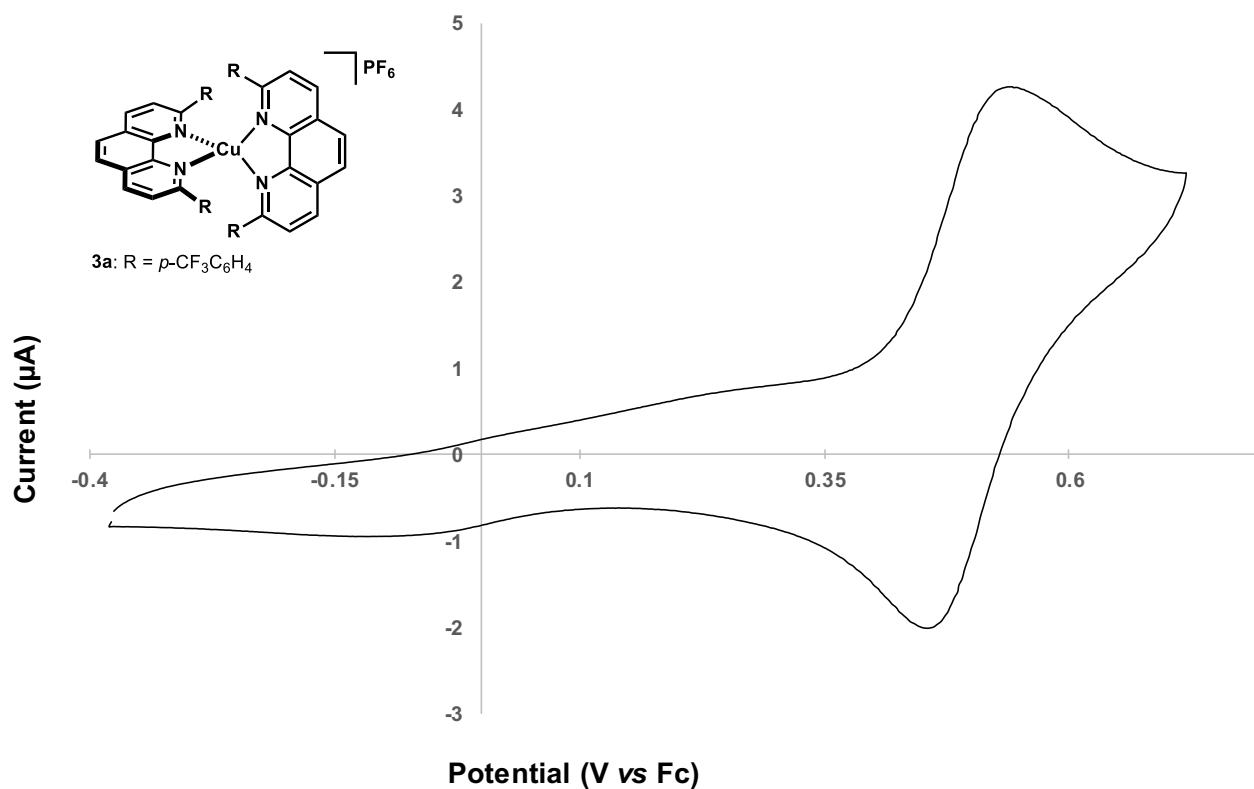


Figure S1a: Cyclic voltammogram of complex **3a** in MeCN (0.25 mM) using NBu₄PF₆ (0.1 M) as supporting electrolyte at scan rate of 100 mV s⁻¹.

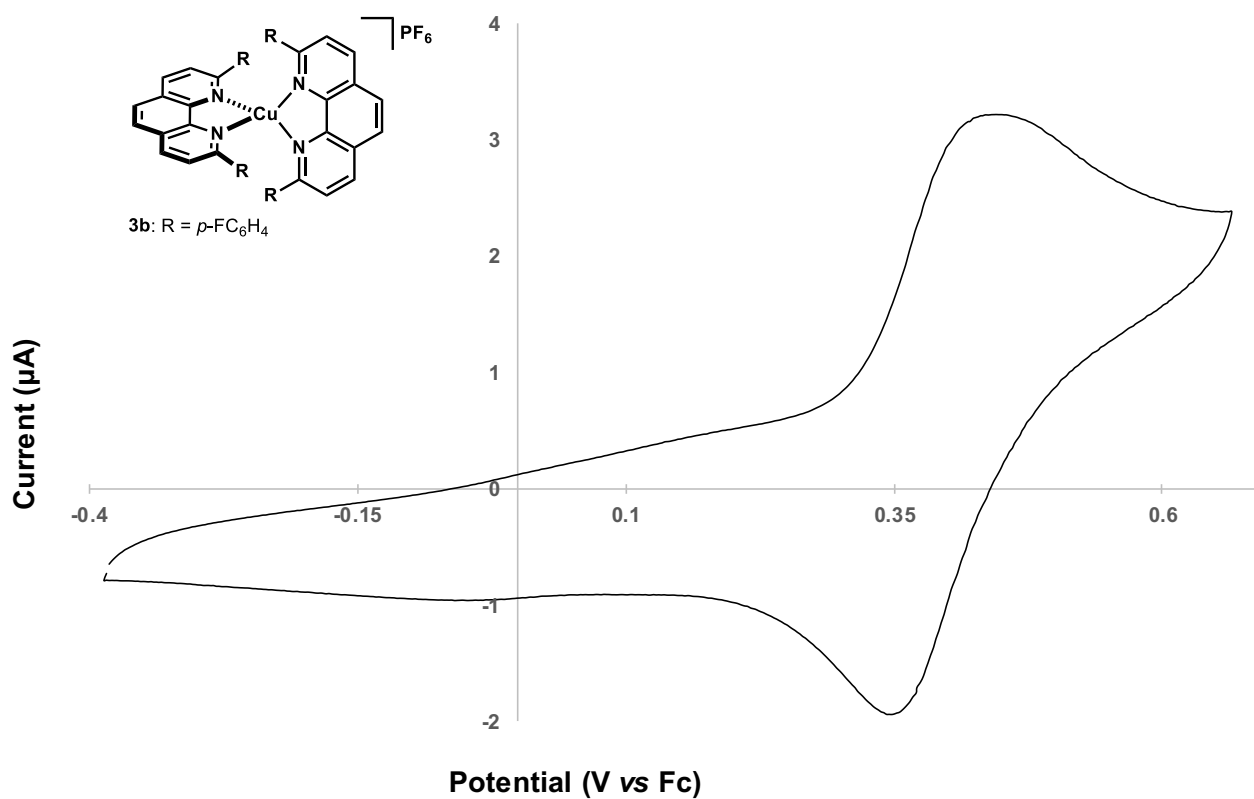


Figure S1b: Cyclic voltammogram of complex **3b** in MeCN (0.25 mM) using NBu₄PF₆ (0.1 M) as supporting electrolyte at scan rate of 100 mV s⁻¹.

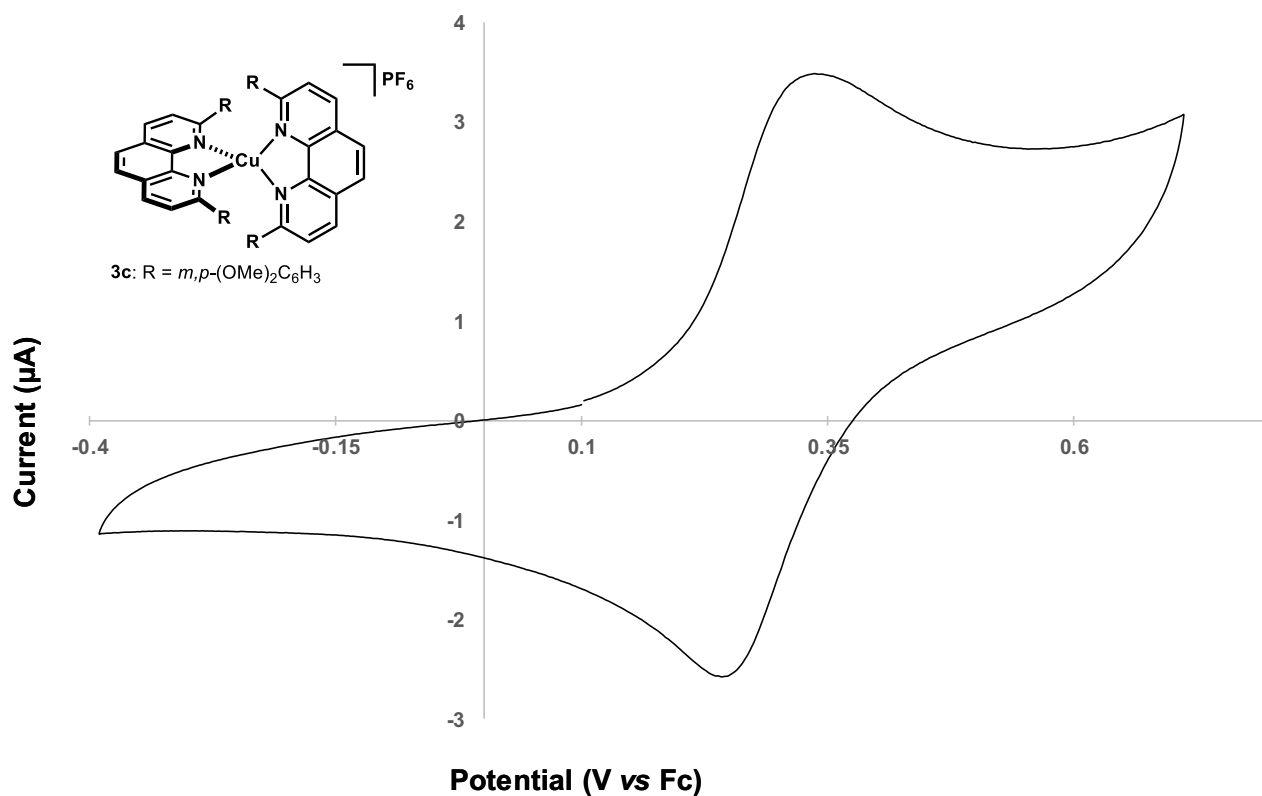


Figure S1c: Cyclic voltammogram of complex **3c** in MeCN (0.25 mM) using NBu₄PF₆ (0.1 M) as supporting electrolyte at scan rate of 100 mV s⁻¹.

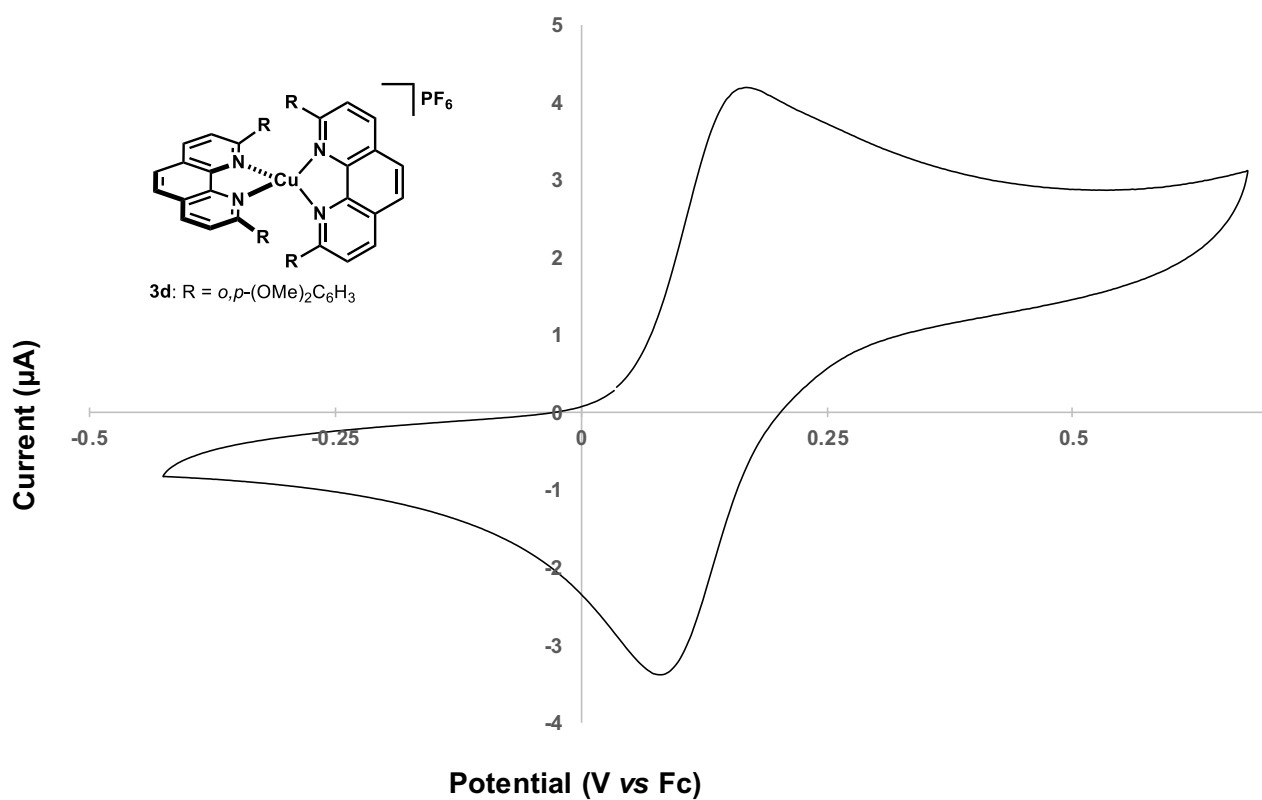


Figure S1d: Cyclic voltammogram of complex **3d** in MeCN (0.25 mM) using NBu₄PF₆ (0.1 M) as supporting electrolyte at scan rate of 100 mV s⁻¹.

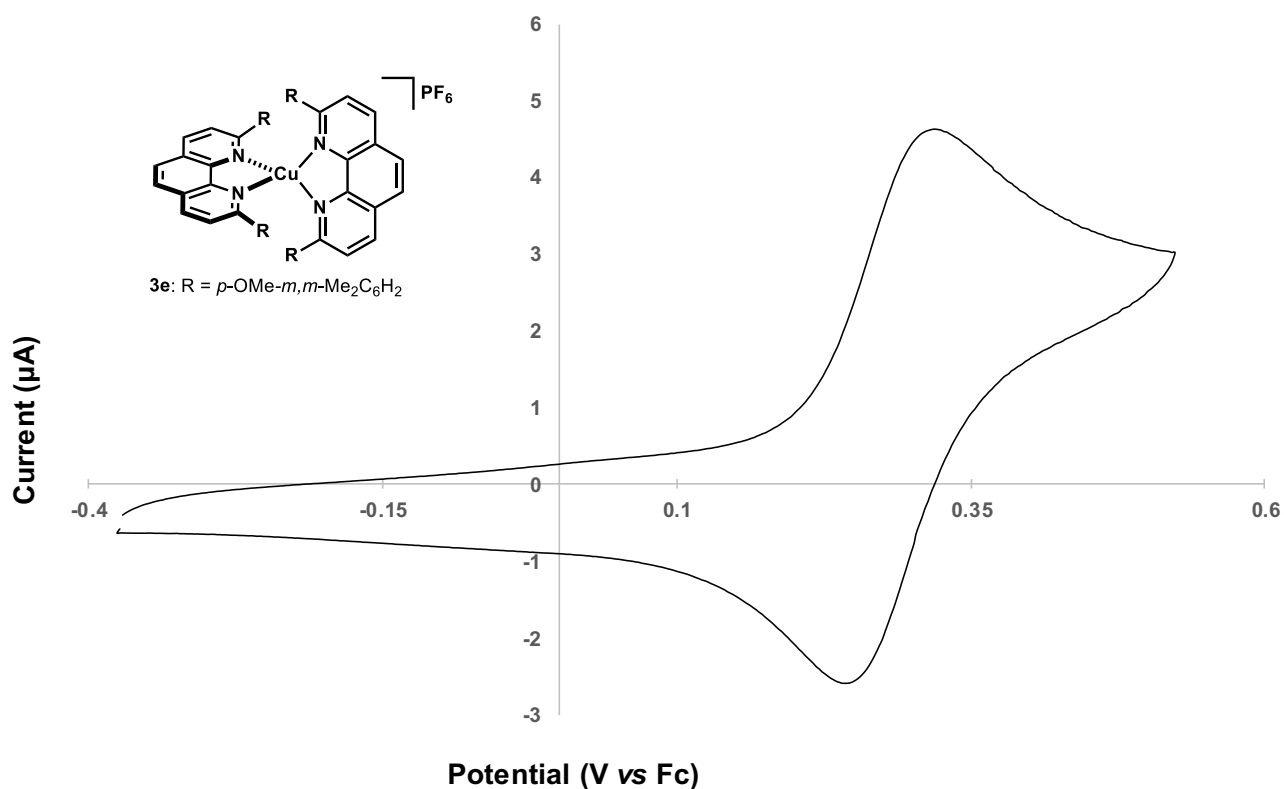


Figure S1e: Cyclic voltammogram of complex **3e** in MeCN (0.25 mM) using NBu₄PF₆ (0.1 M) as supporting electrolyte at scan rate of 100 mV s⁻¹.

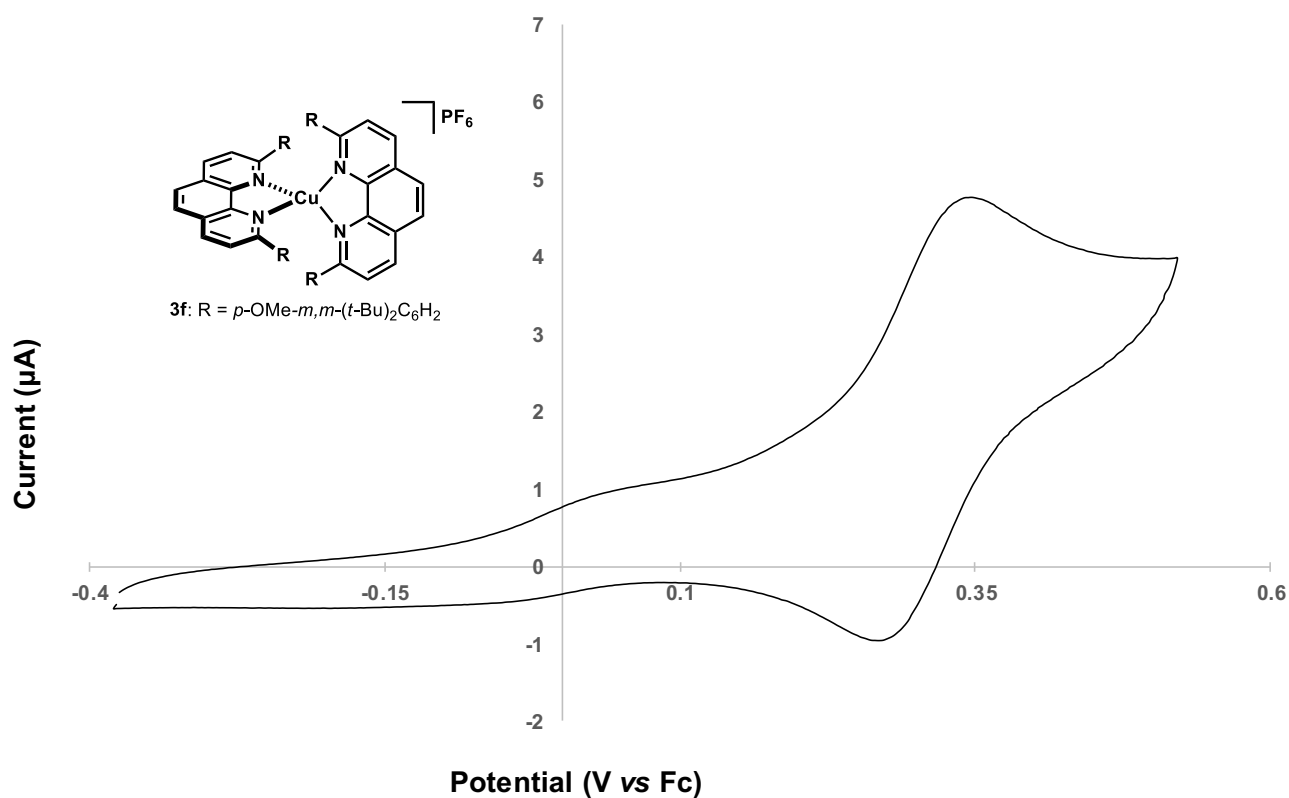


Figure S1f: Cyclic voltammogram of complex **3f** in MeCN (0.25 mM) using NBu₄PF₆ (0.1 M) as supporting electrolyte at scan rate of 100 mV s⁻¹.

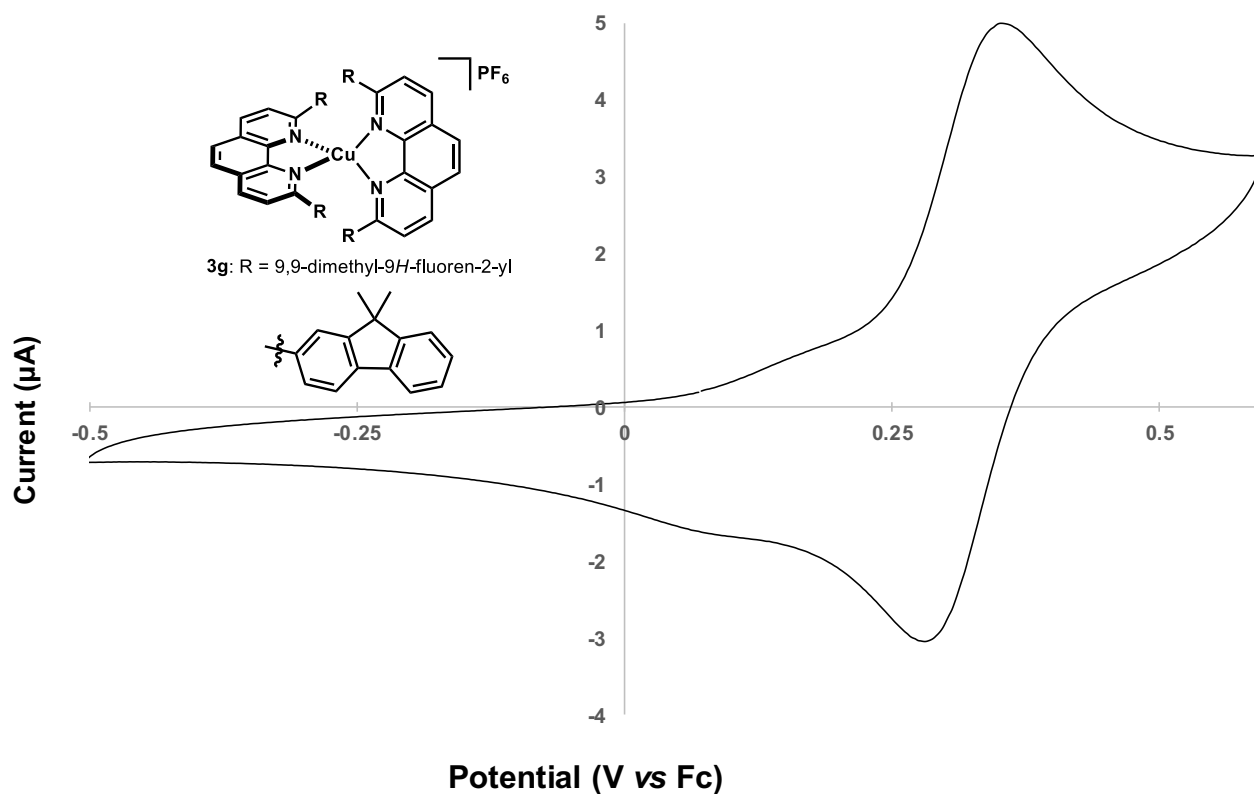


Figure S1g: Cyclic voltammogram of complex **3g** in MeCN (0.25 mM) using NBu₄PF₆ (0.1 M) as supporting electrolyte at scan rate of 100 mV s⁻¹.

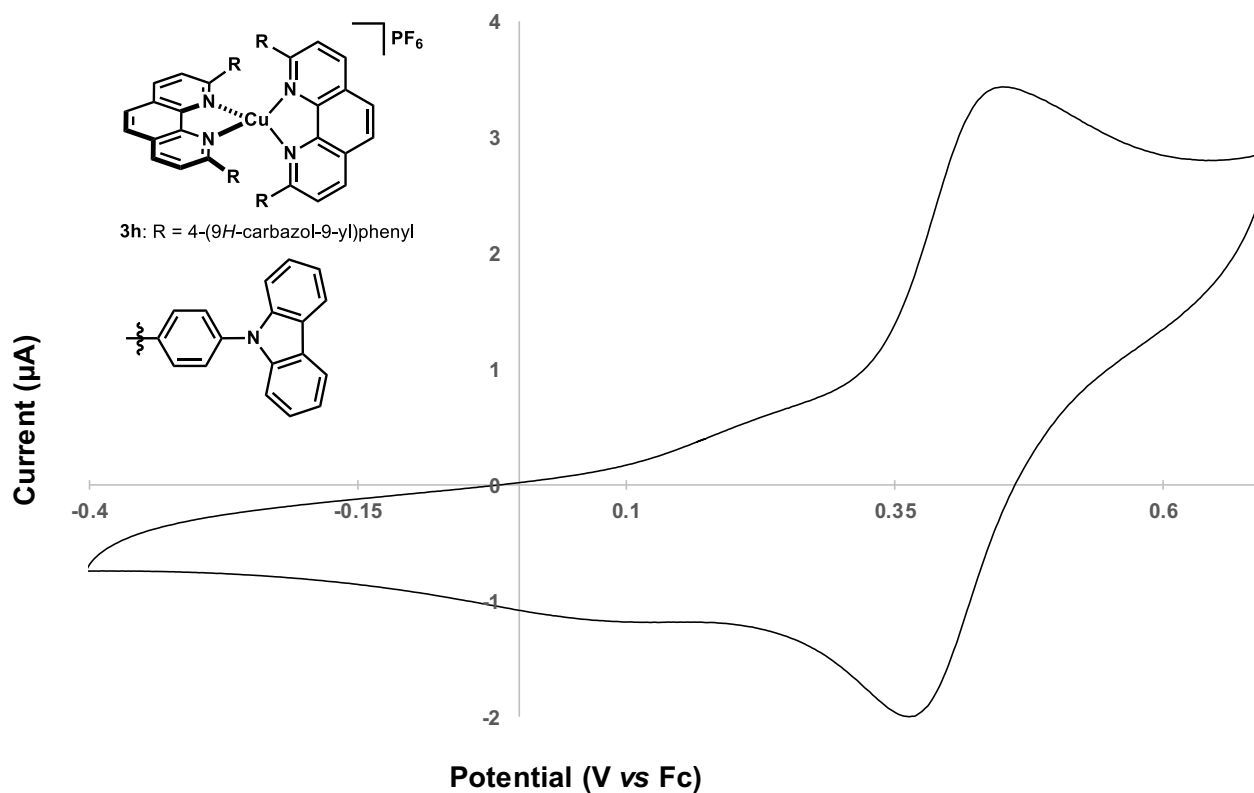


Figure S1h: Cyclic voltammogram of complex **3h** in MeCN (0.25 mM) using NBu₄PF₆ (0.1 M) as supporting electrolyte at scan rate of 100 mV s⁻¹.

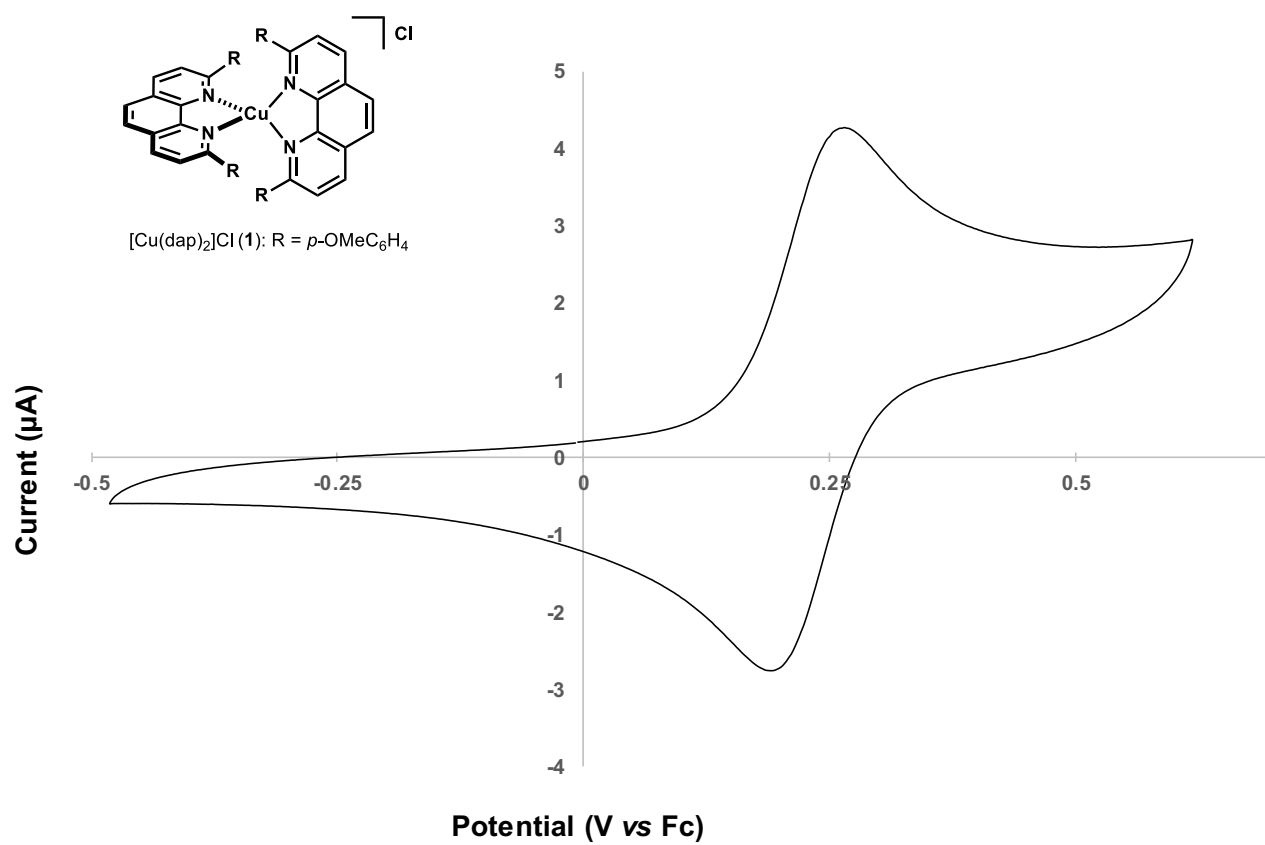


Figure S1i: Cyclic voltammogram of complex **1** in MeCN (0.25 mM) using NBu_4PF_6 (0.1 M) as supporting electrolyte at scan rate of 100 mV s^{-1} .

III. Absorption Spectra

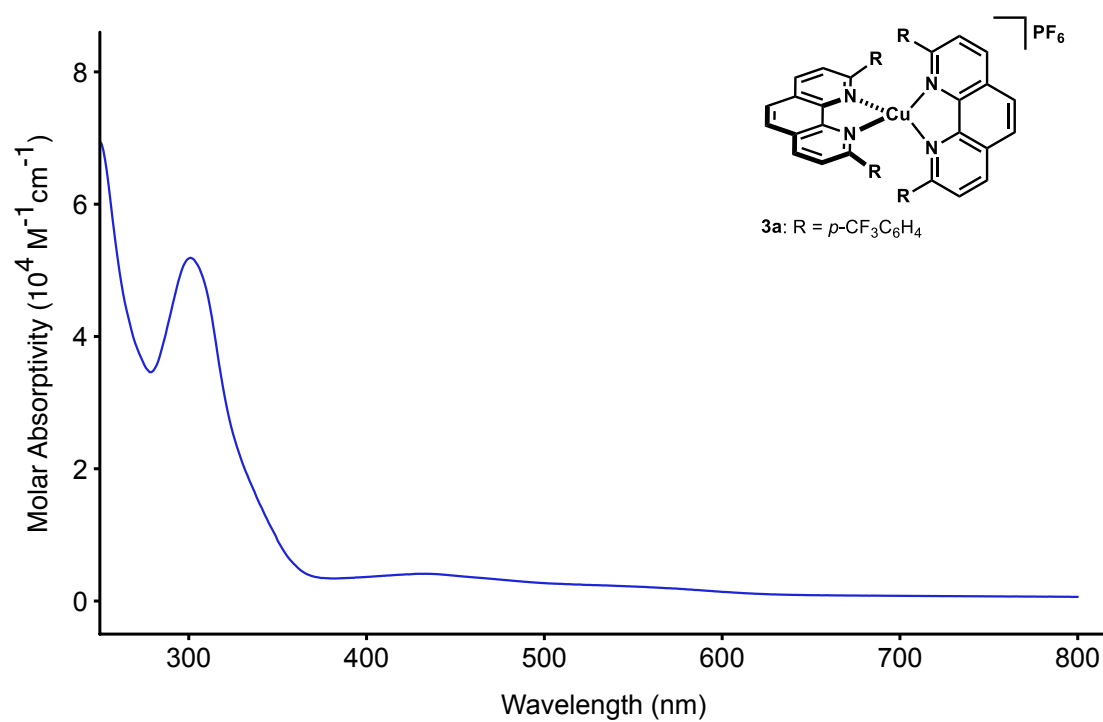


Figure S2a: Absorption spectrum of complex **3a** in CH_2Cl_2 (10^{-5} M).

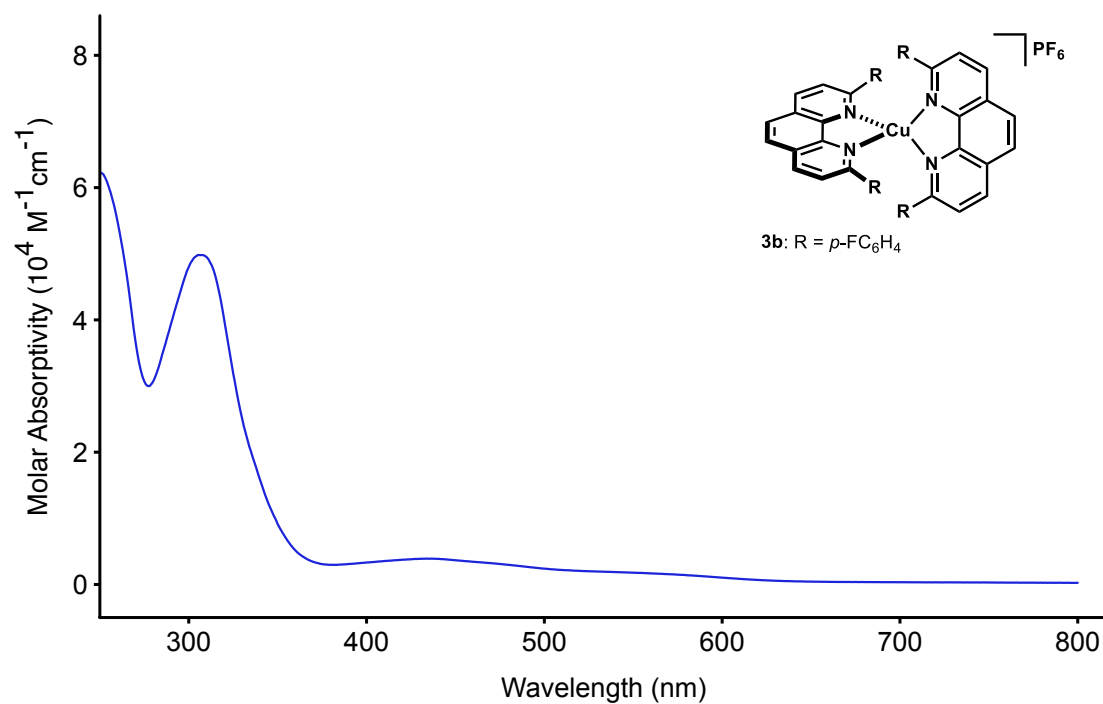


Figure S2b: Absorption spectrum of complex **3b** in CH_2Cl_2 (10^{-5} M).

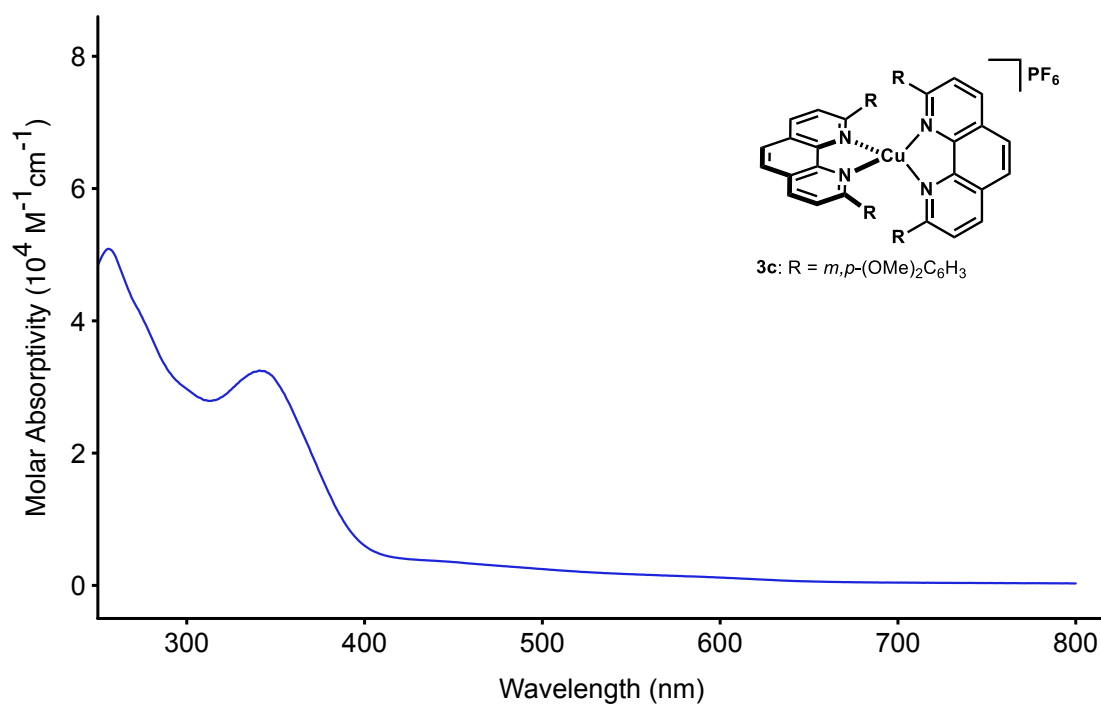


Figure S2c: Absorption spectrum of complex **3c** in CH₂Cl₂ (10⁻⁵ M).

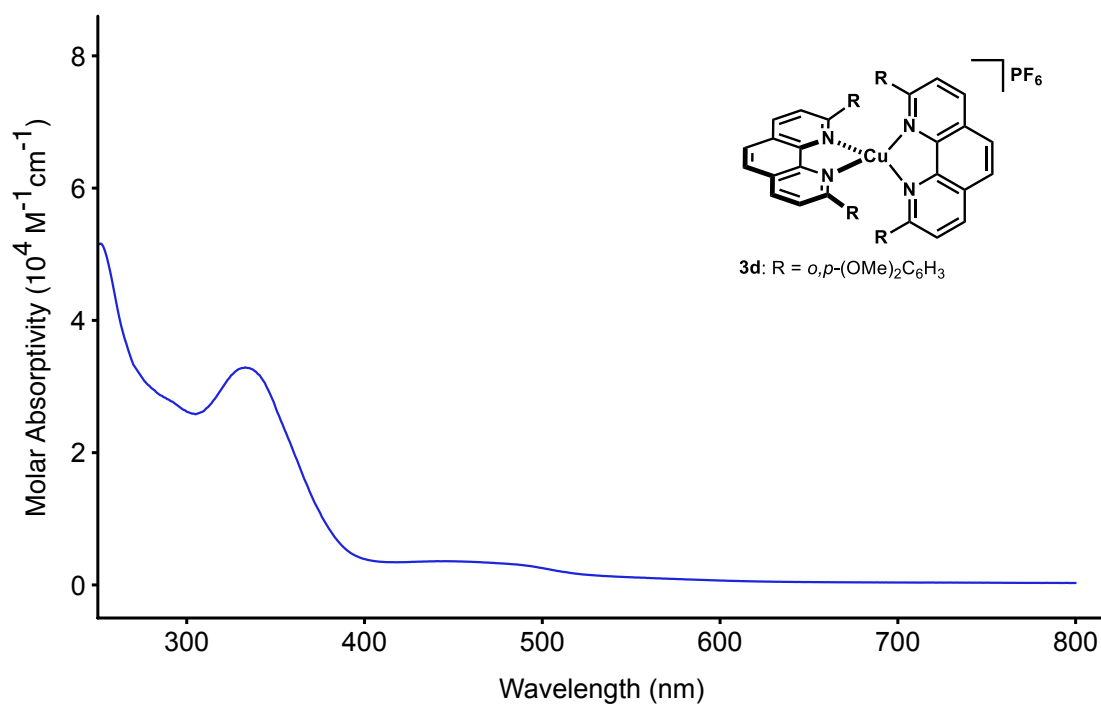


Figure S2d: Absorption spectrum of complex **3d** in CH₂Cl₂ (10⁻⁵ M).

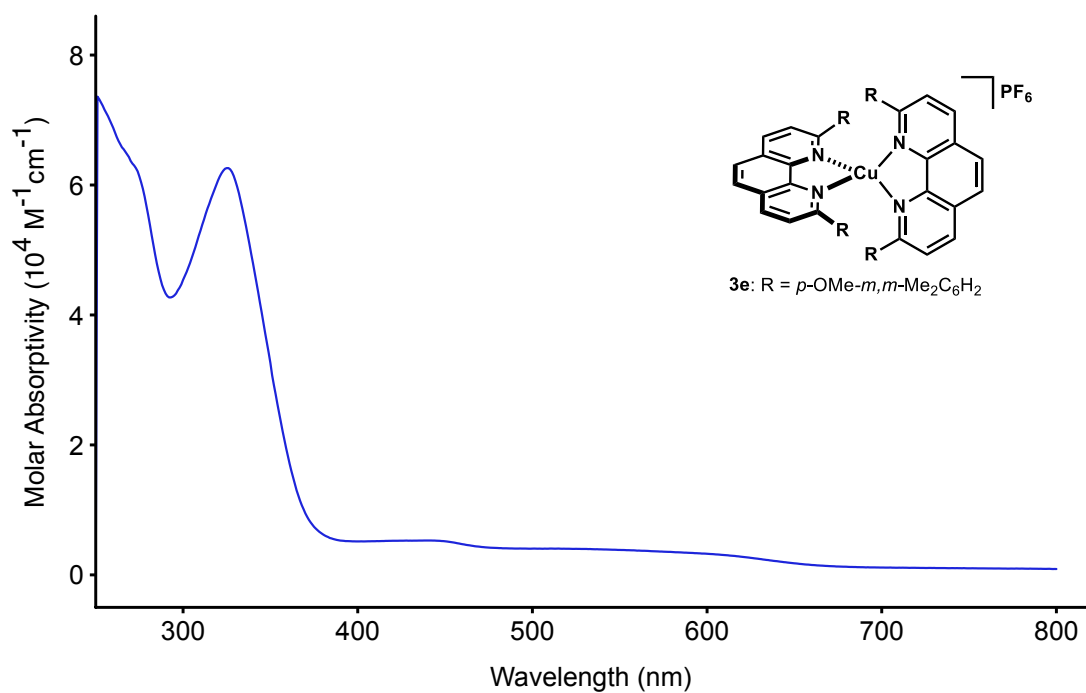


Figure S2e: Absorption spectrum of complex **3e** in CH₂Cl₂ (10⁻⁵ M).

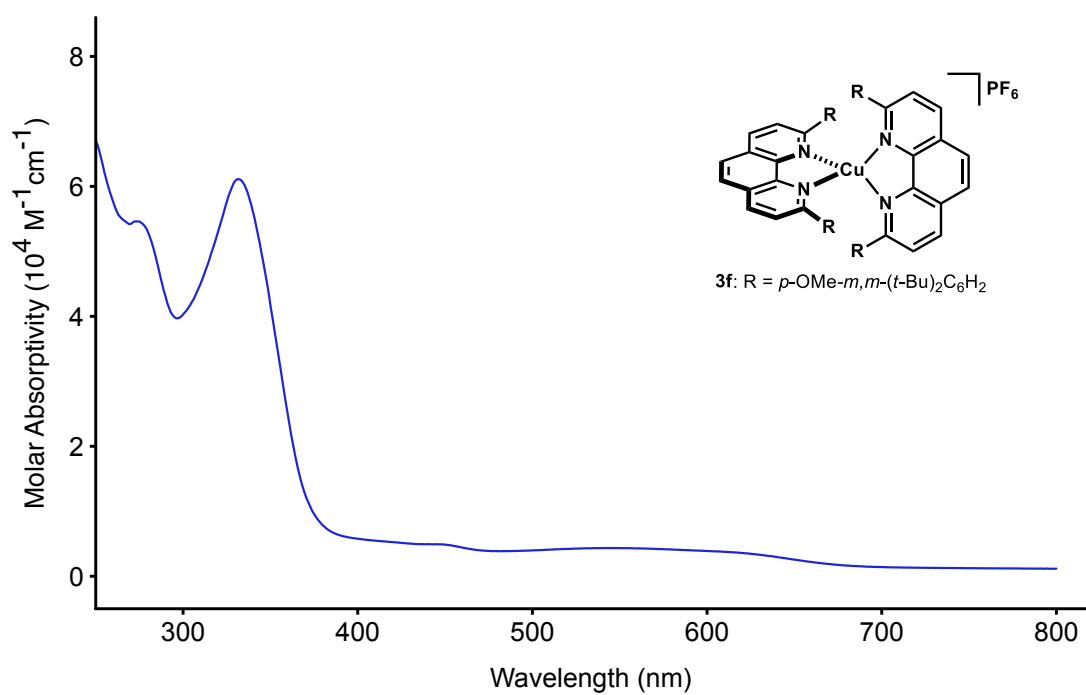


Figure S2f: Absorption spectrum of complex **3f** in CH₂Cl₂ (10⁻⁵ M).

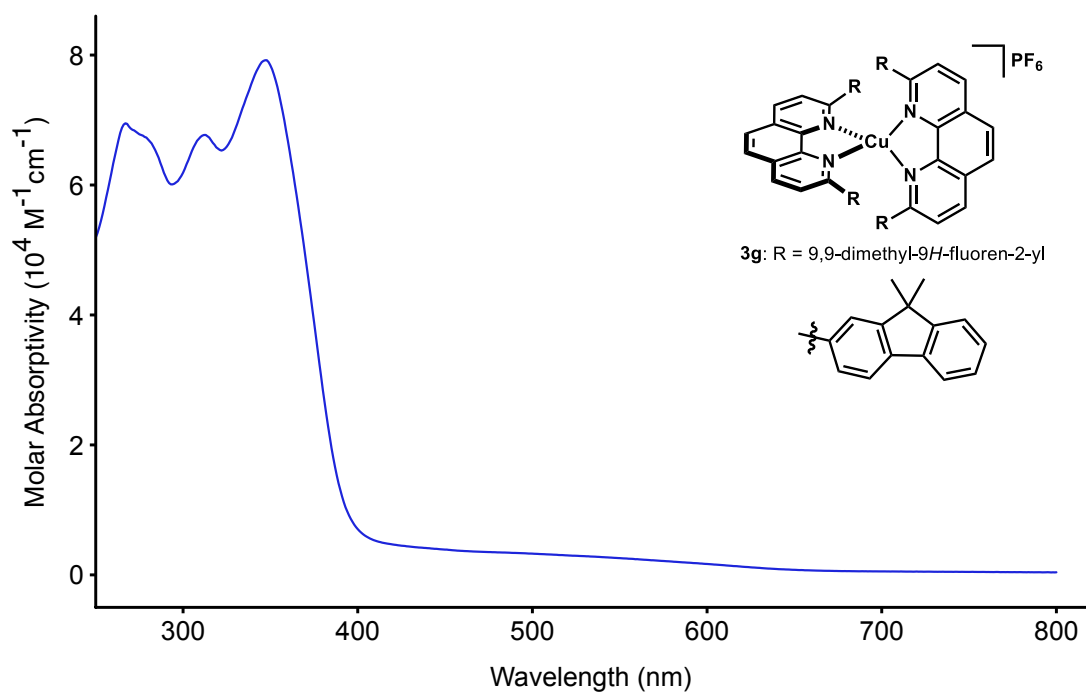


Figure S2g: Absorption spectrum of complex **3g** in CH_2Cl_2 (10^{-5} M).

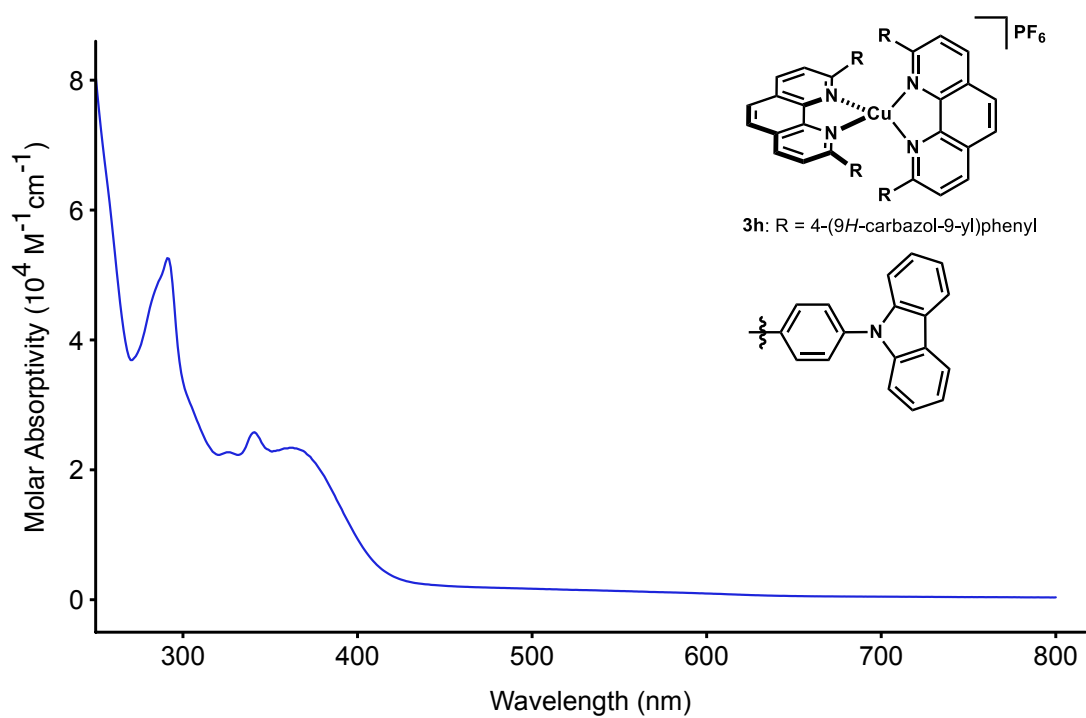


Figure S2h: Absorption spectrum of complex **3h** in CH_2Cl_2 (10^{-5} M).

IV. Excitation Spectra

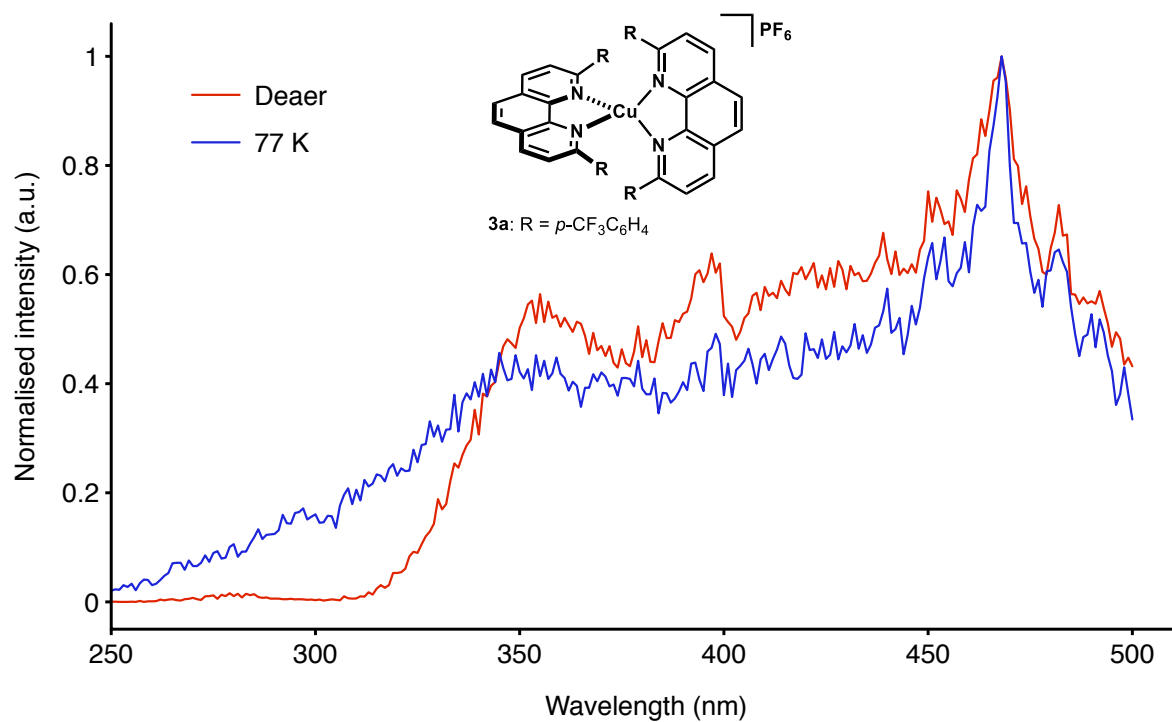


Figure S3a: Excitation spectrum of complex **3a** in CH_2Cl_2 (10^{-5} M).

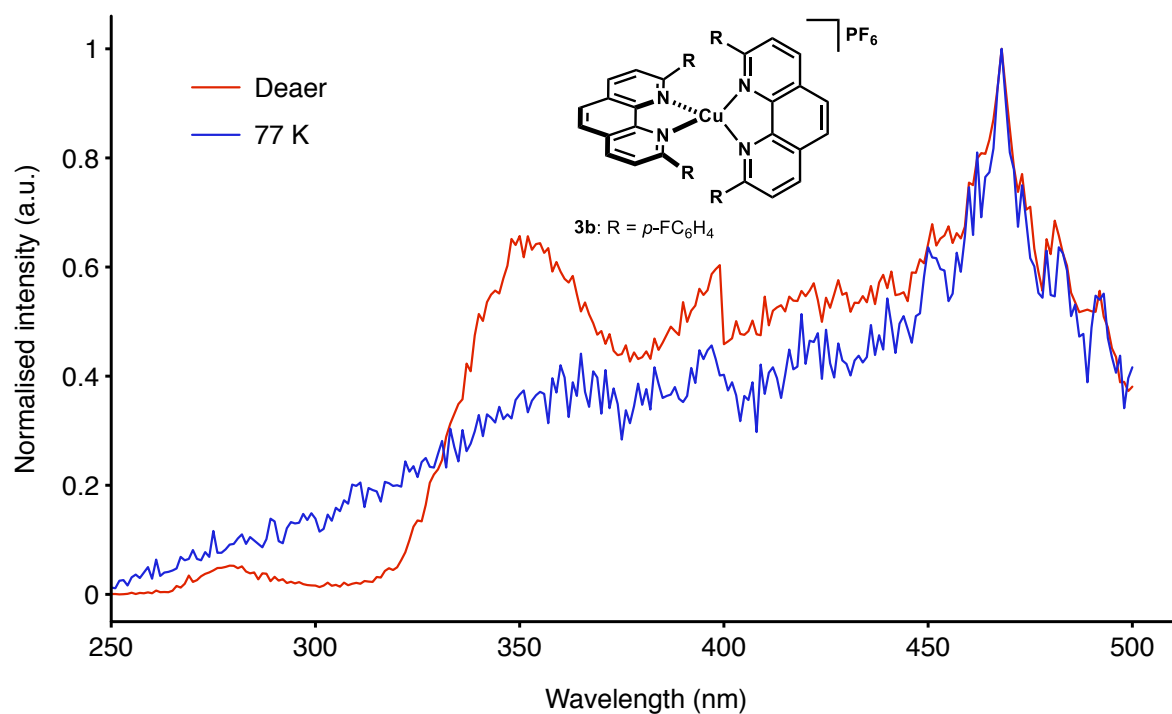


Figure S3b: Excitation spectrum of complex **3b** in CH_2Cl_2 (10^{-5} M).

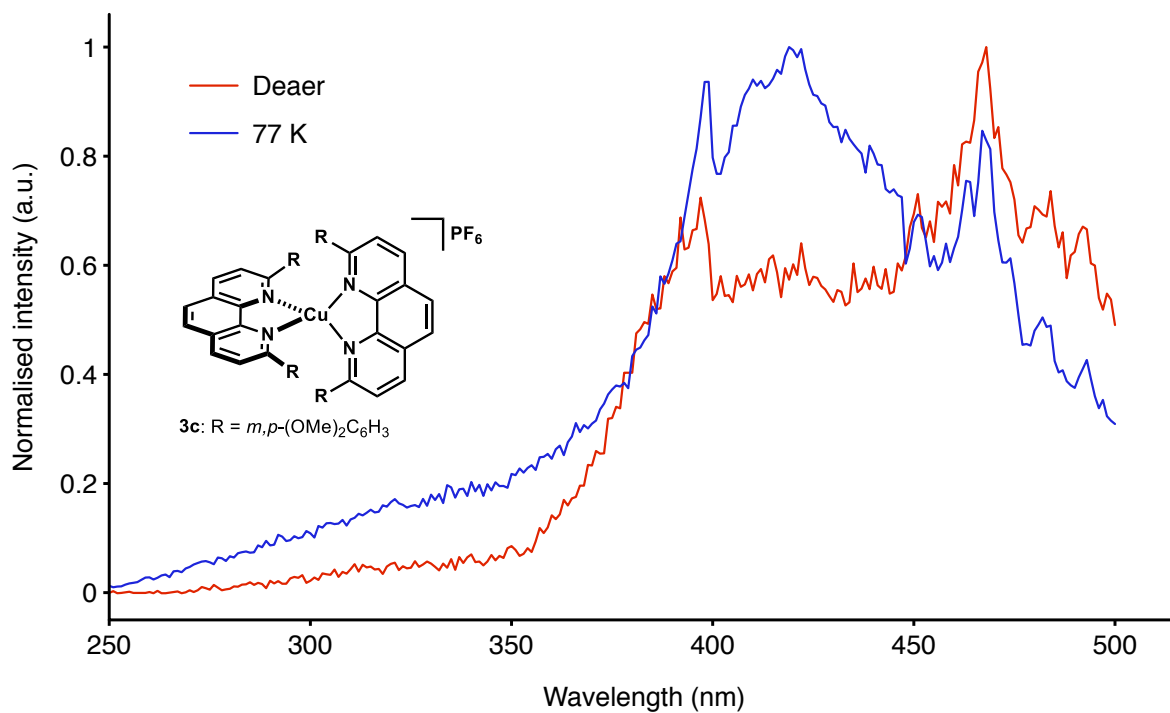


Figure S3c: Excitation spectrum of complex **3c** in CH₂Cl₂ (10⁻⁵ M).

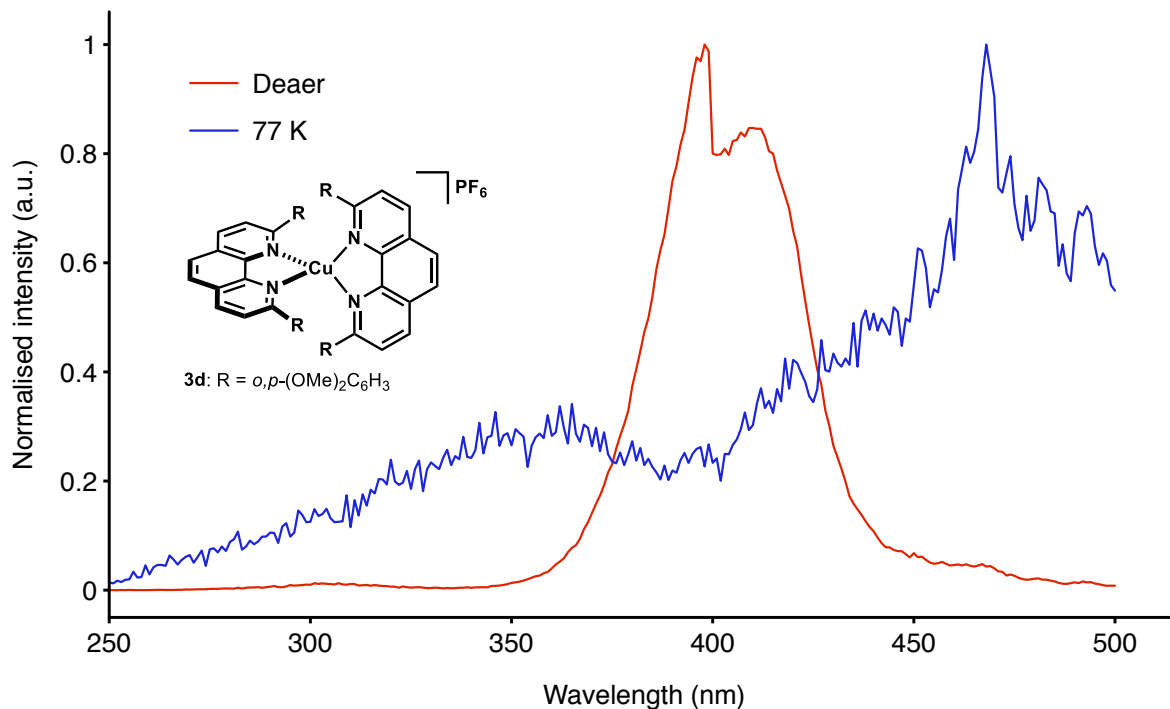


Figure S3d: Excitation spectrum of complex **3d** in CH₂Cl₂ (10⁻⁵ M).

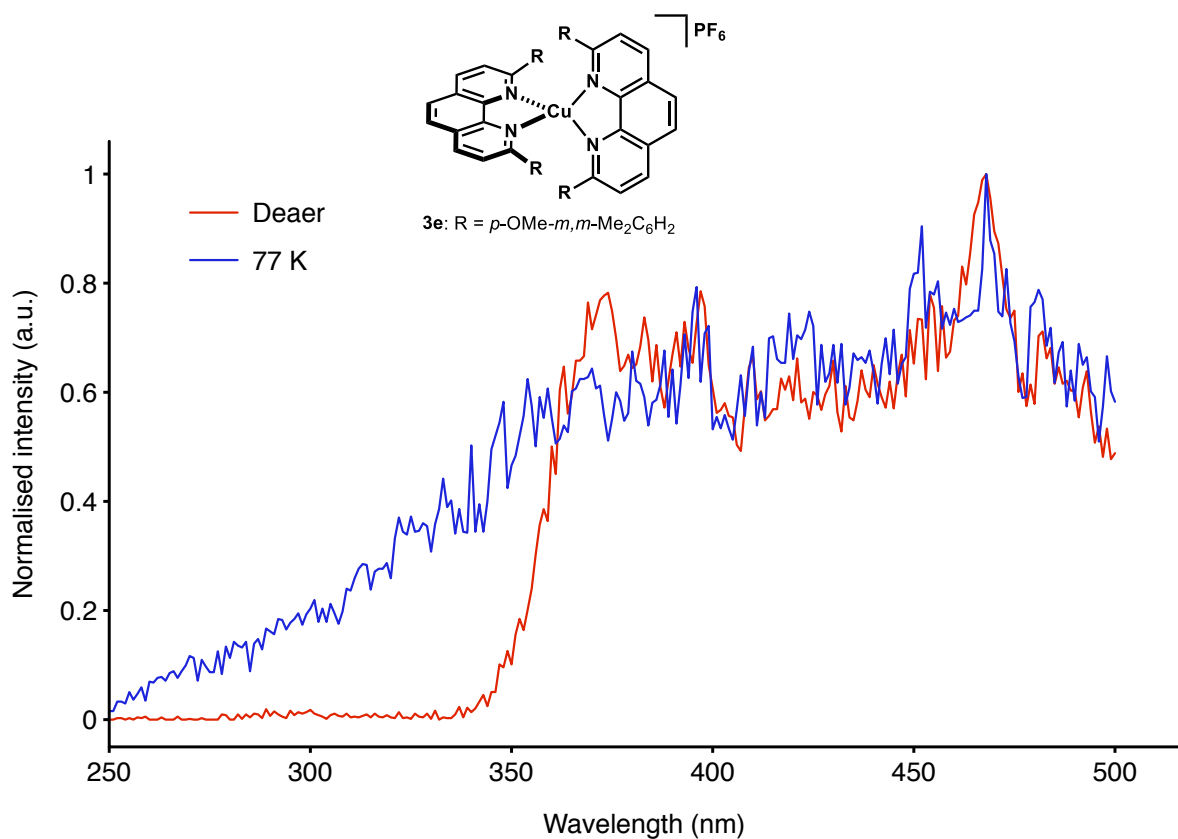


Figure S3e: Excitation spectrum of complex **3e** in CH_2Cl_2 (10^{-5} M).

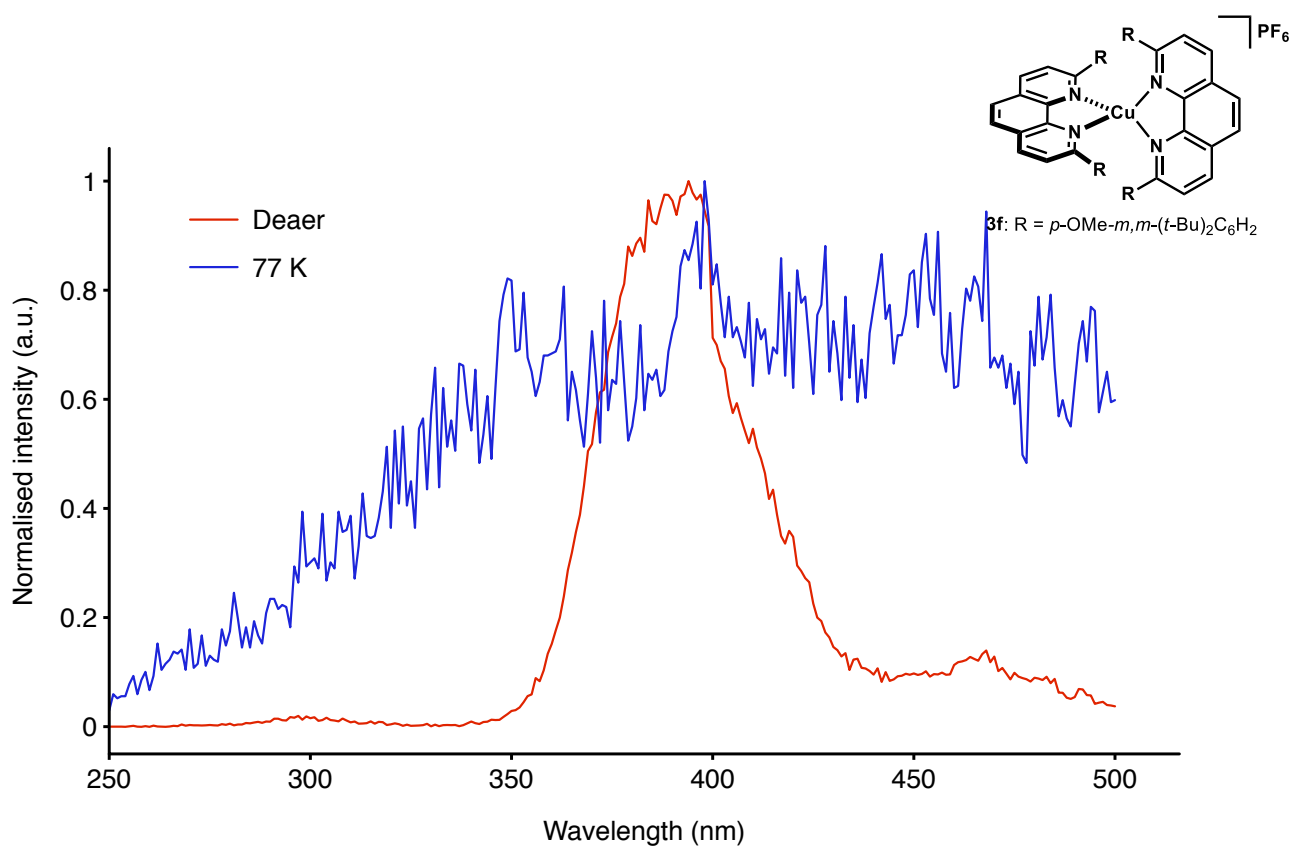


Figure S3f: Excitation spectrum of complex **3f** in CH_2Cl_2 (10^{-5} M).

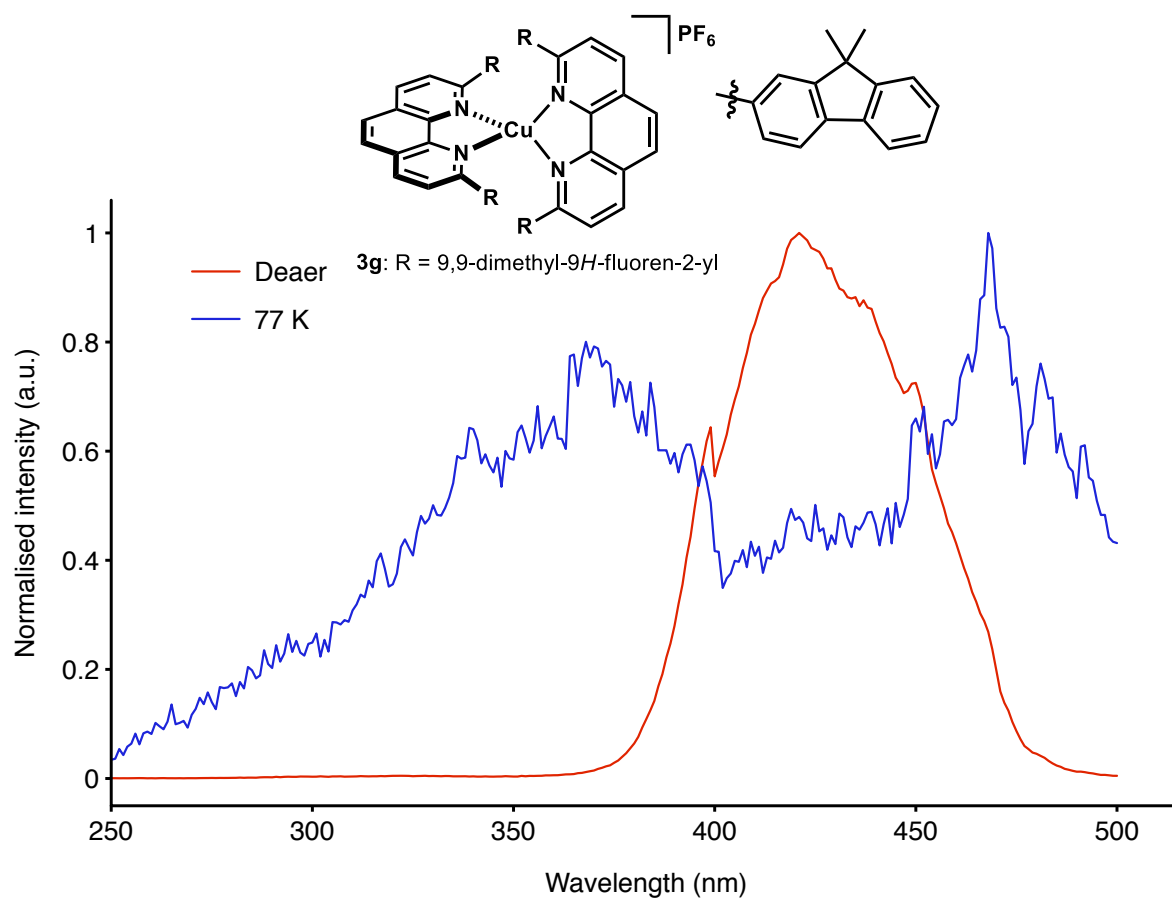


Figure S3g: Excitation spectrum of complex **3g** in CH_2Cl_2 (10^{-5} M).

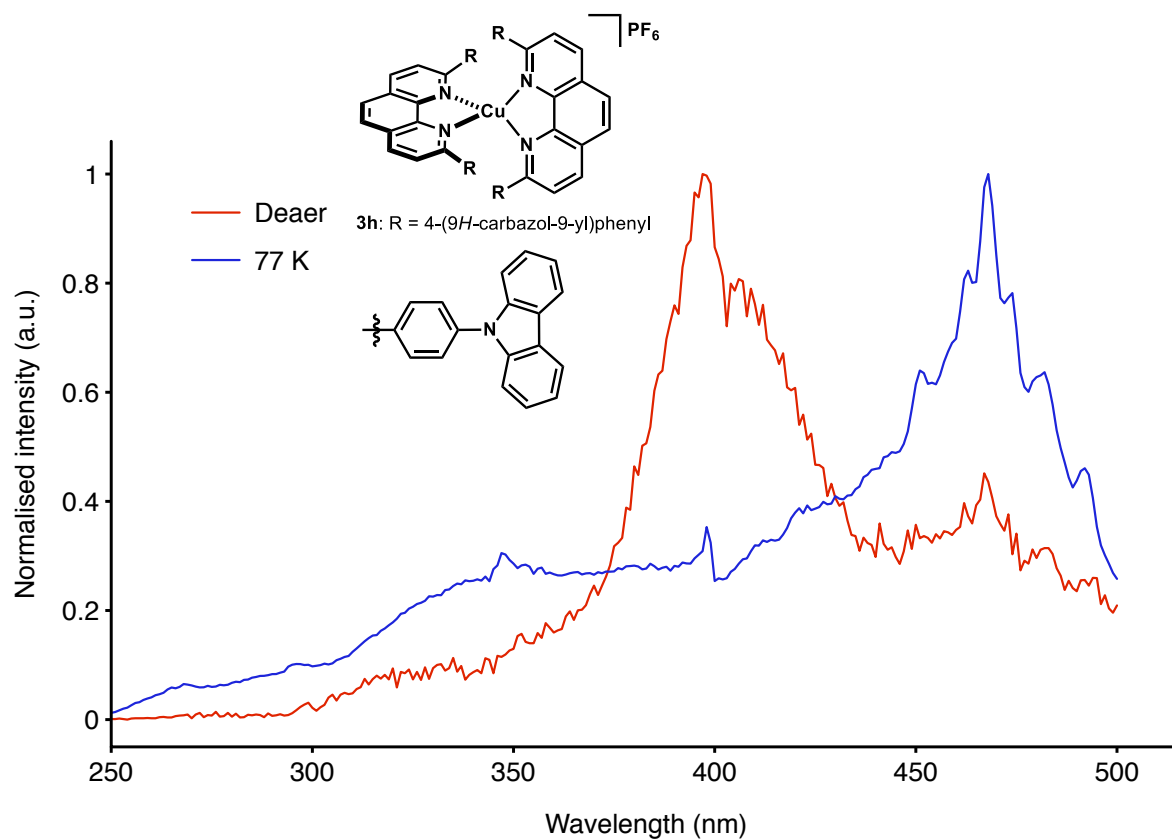


Figure S3h: Excitation spectrum of complex **3h** in CH_2Cl_2 (10^{-5} M).

V. Emission Spectra

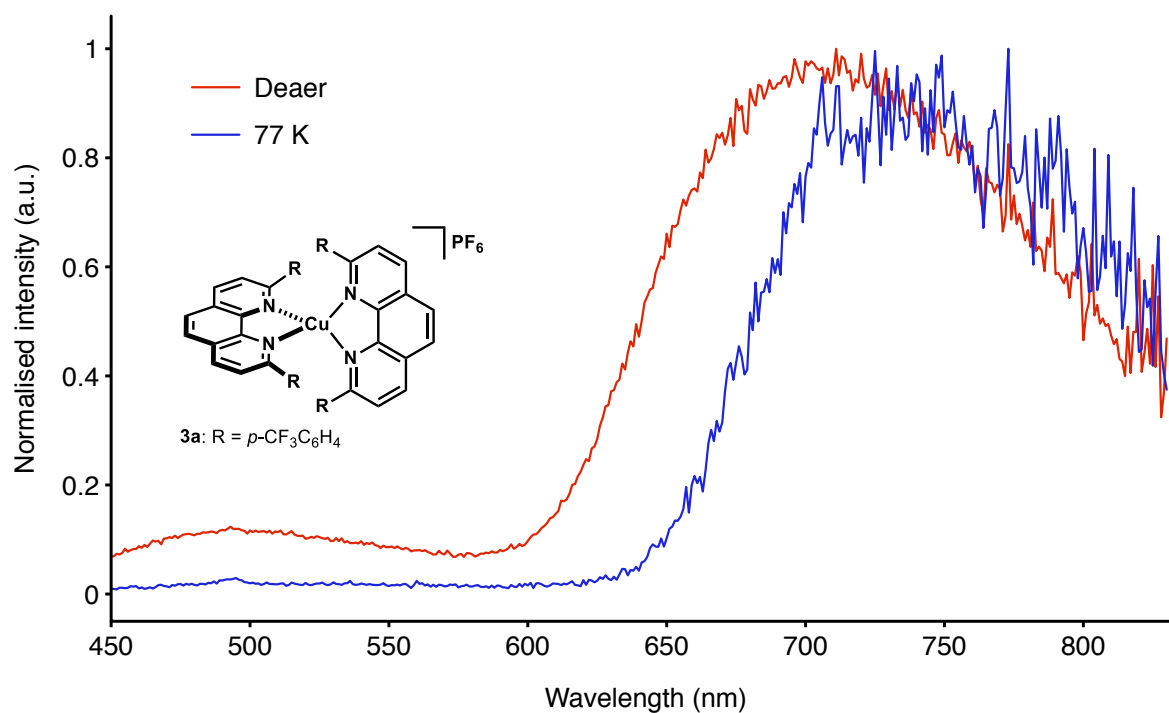


Figure S4a: Emission spectrum of complex **3a** in CH_2Cl_2 (10^{-5} M).

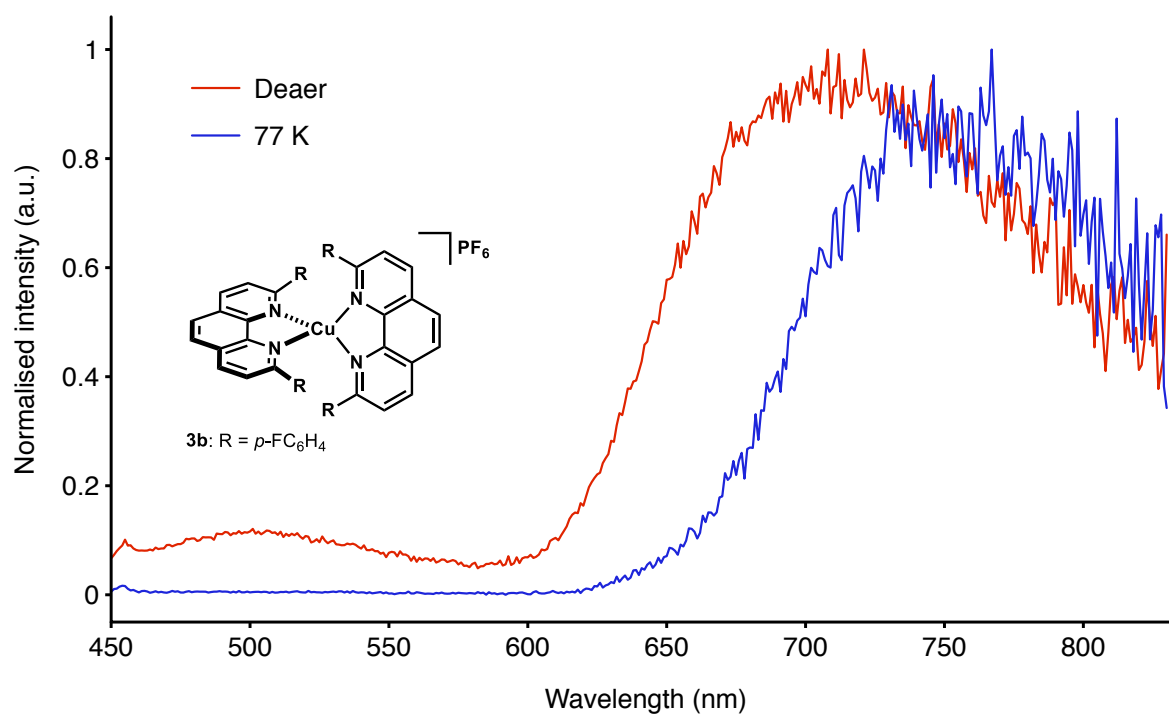


Figure S4b: Emission spectrum of complex **3b** in CH_2Cl_2 (10^{-5} M).

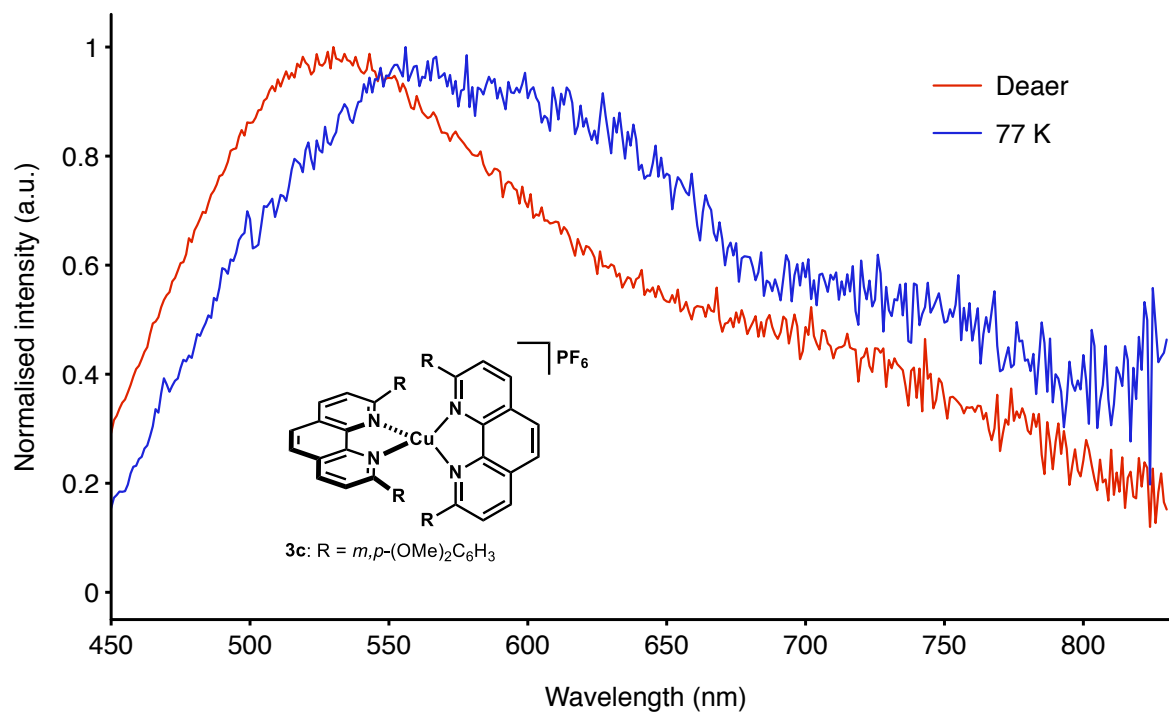


Figure S4c: Emission spectrum of complex **3c** in CH₂Cl₂ (10⁻⁵ M).

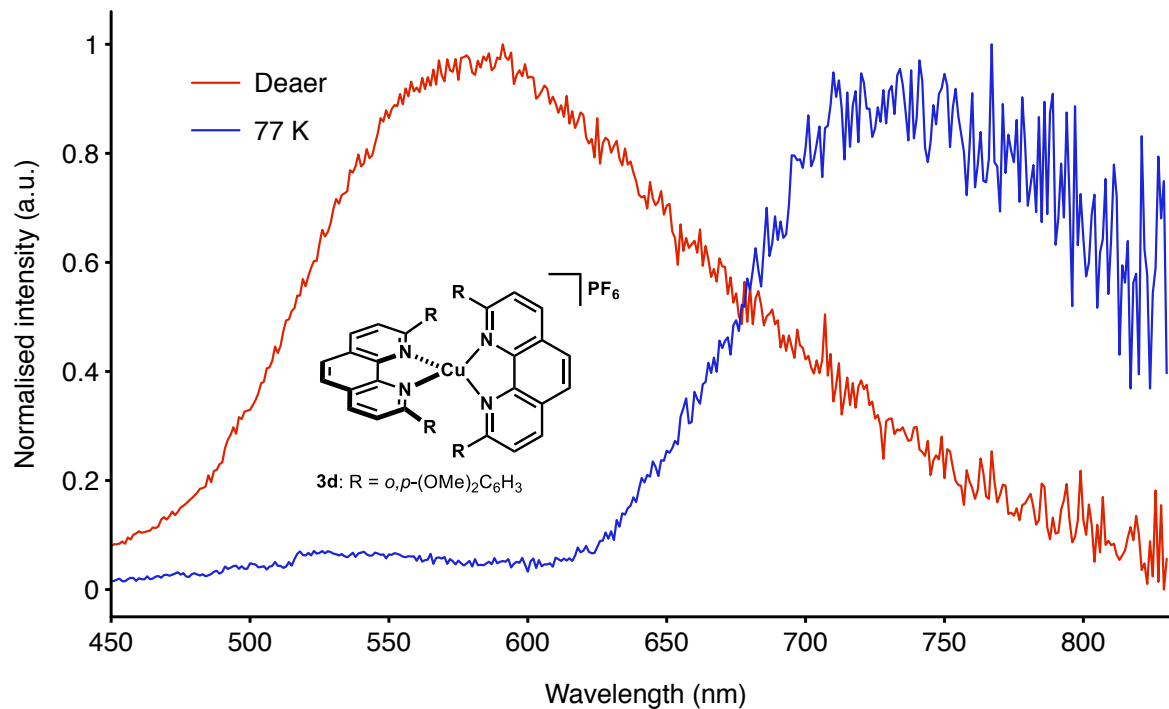


Figure S4d: Emission spectrum of complex **3d** in CH₂Cl₂ (10⁻⁵ M).

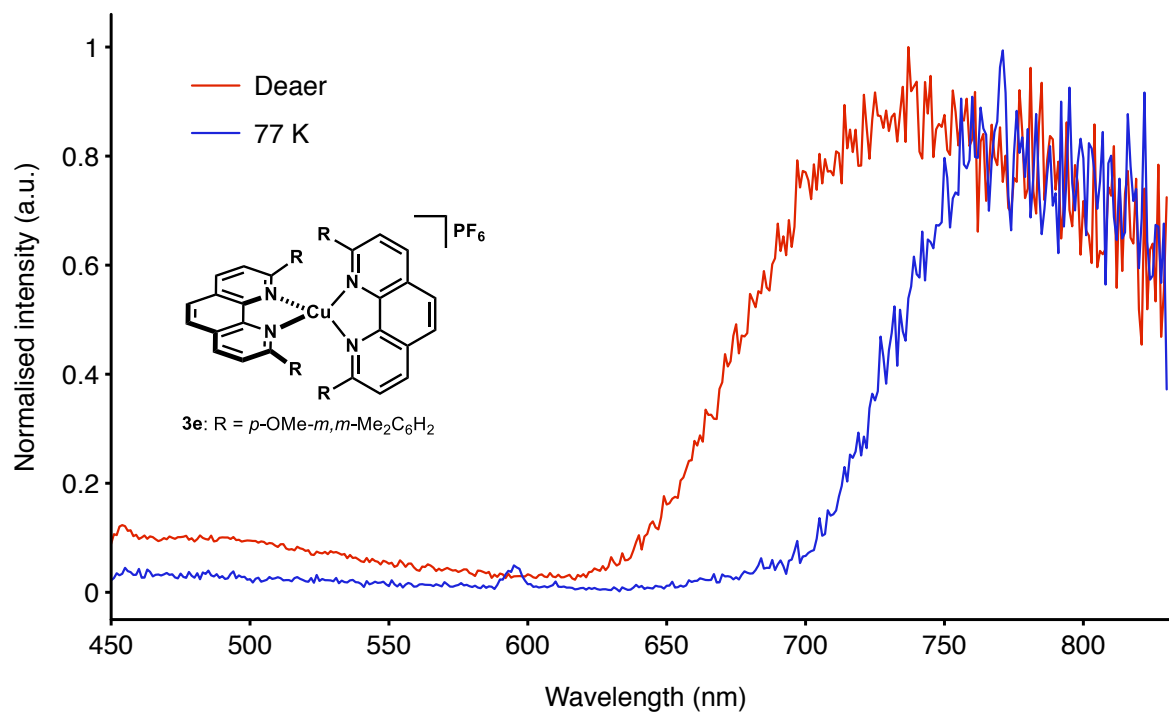


Figure S4e: Emission spectrum of complex **3e** in CH₂Cl₂ (10⁻⁵ M).

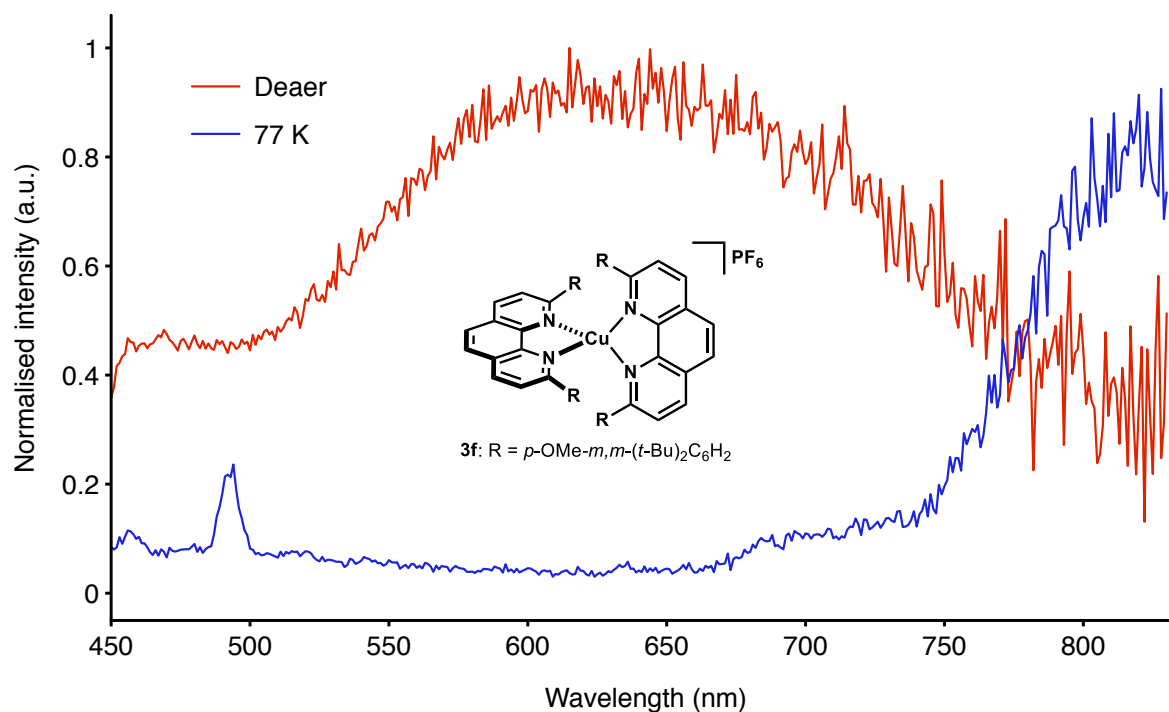


Figure S4f: Emission spectrum of complex **3f** in CH₂Cl₂ (10⁻⁵ M).

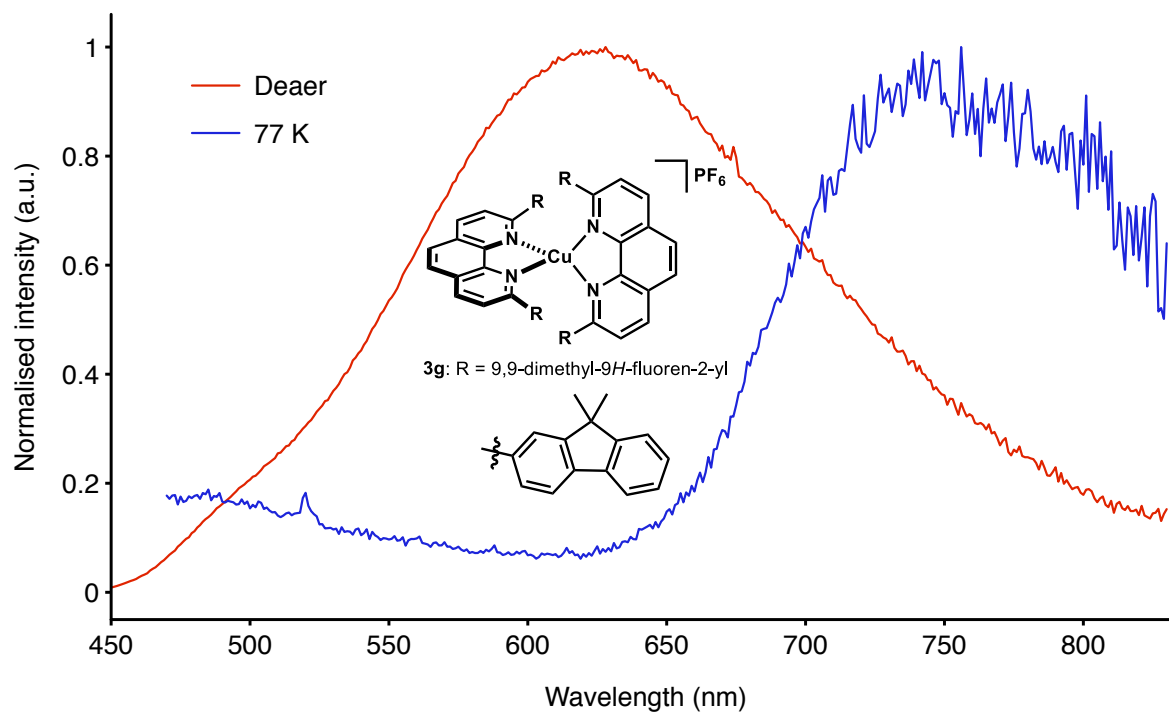


Figure S4g: Emission spectrum of complex **3g** in CH_2Cl_2 (10^{-5} M).

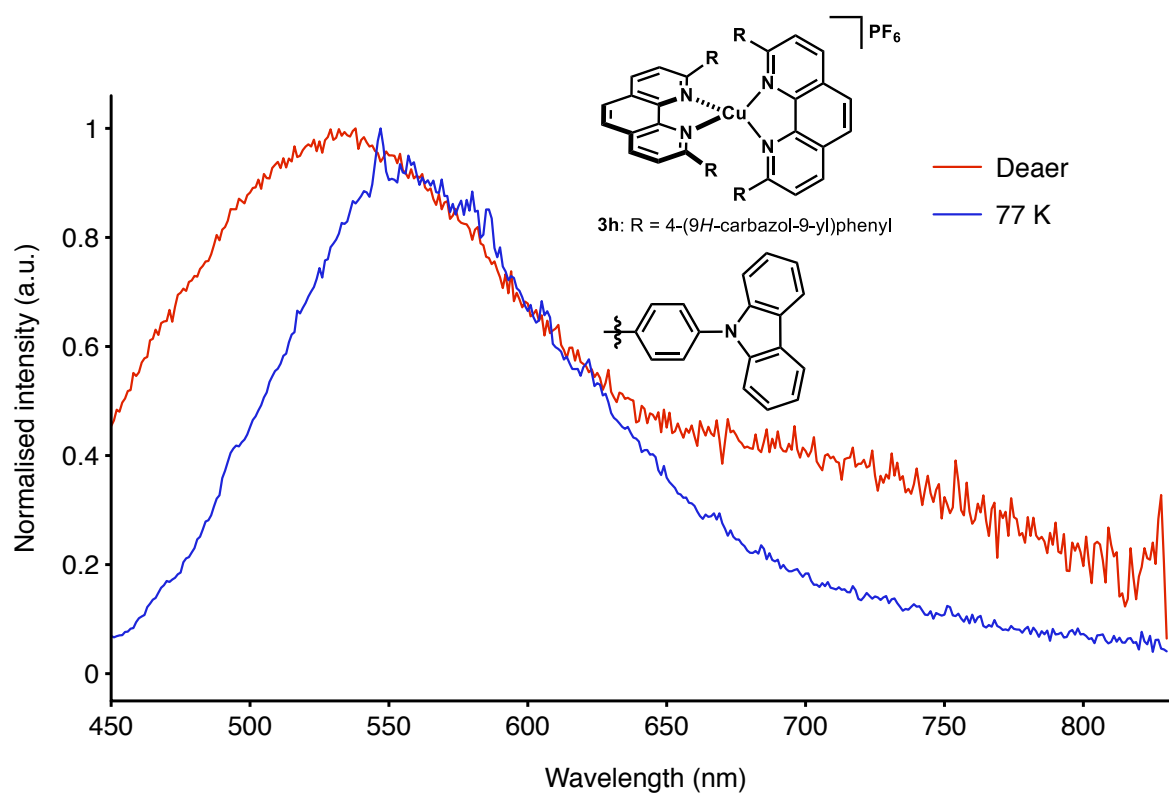


Figure S4h: Emission spectrum of complex **3h** in CH_2Cl_2 (10^{-5} M).

VI. X-Ray Crystallographic Data

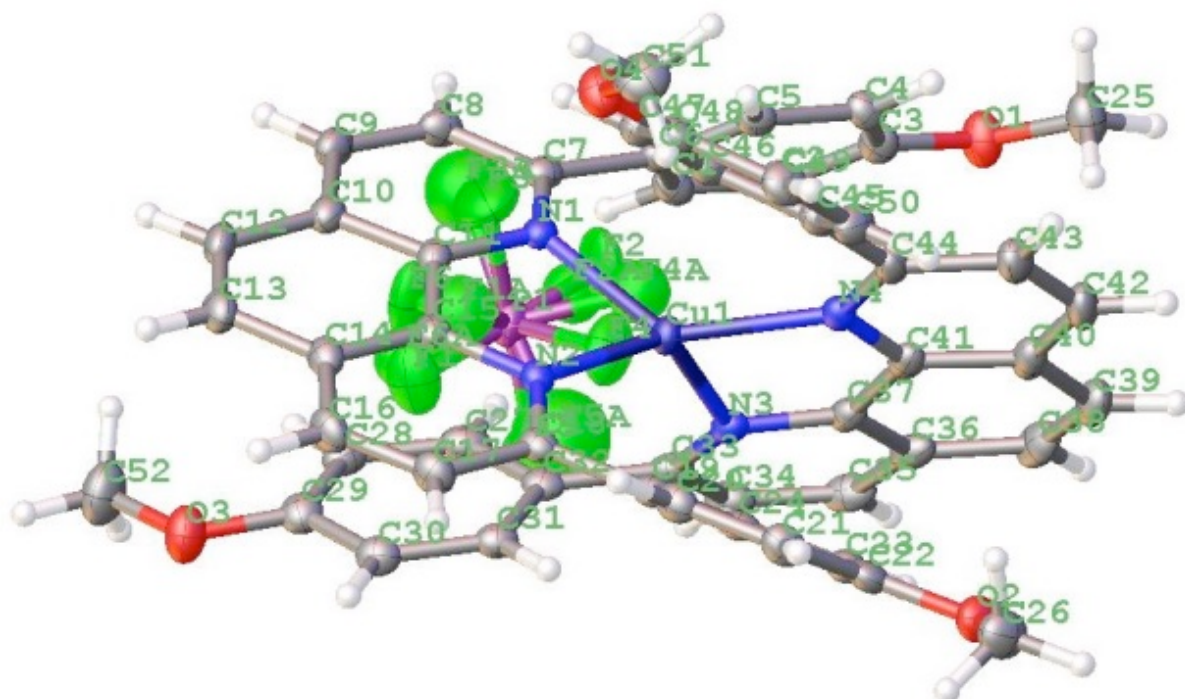


Figure S5a: ORTEP representation of copper(I) complex **1**.PF₆. Thermal ellipsoids are drawn at the 50% probability level.

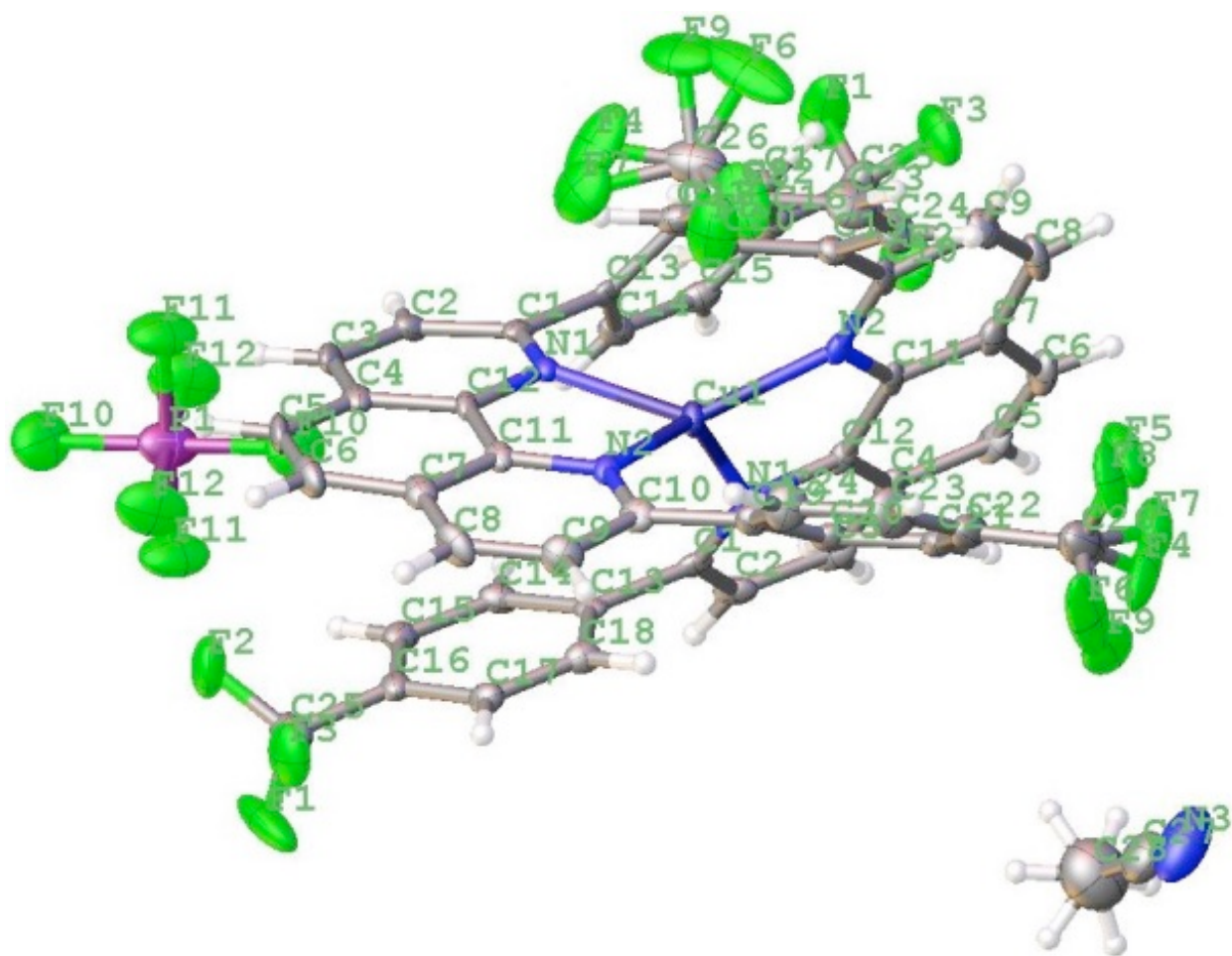


Figure S5b: ORTEP representation of copper(I) complex **3a**. Thermal ellipsoids are drawn at the 50% probability level.

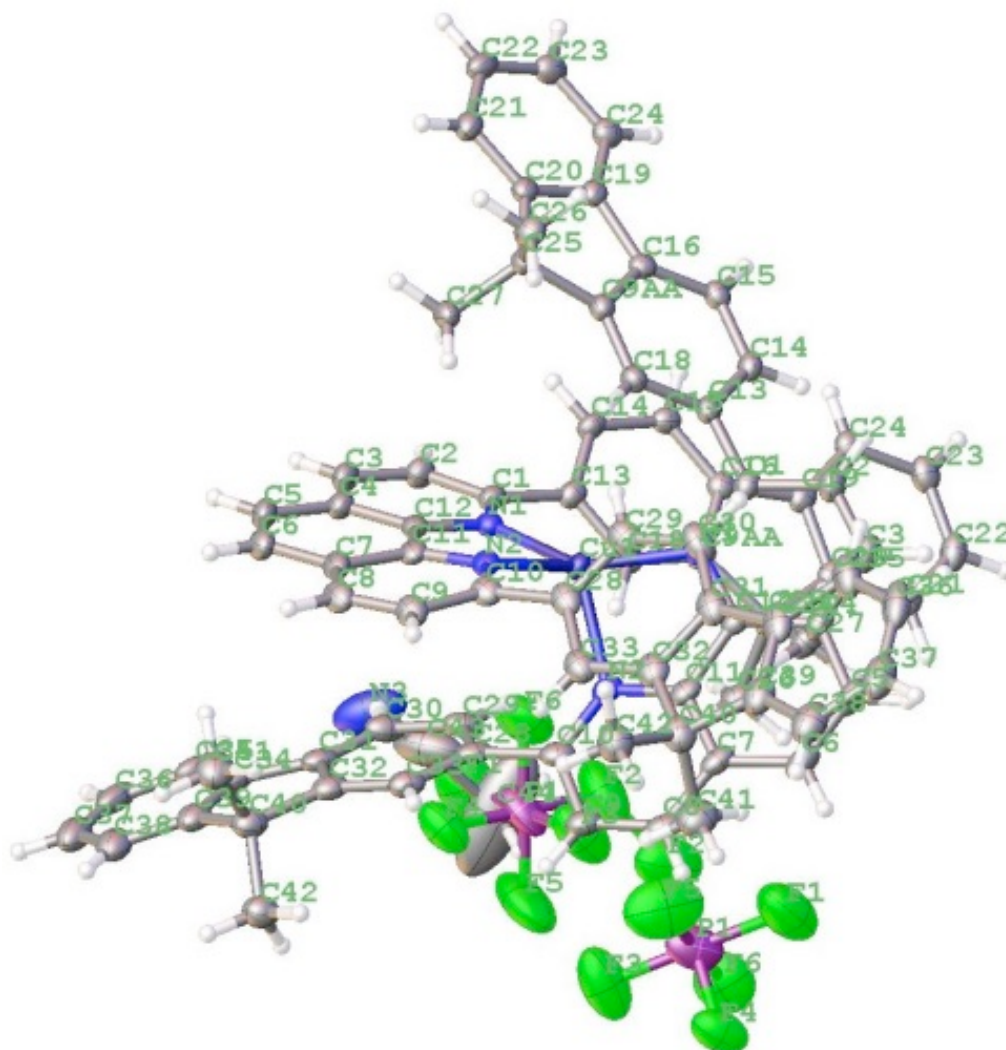


Figure S5d: ORTEP representation of copper(I) complex **3g** Thermal ellipsoids are drawn at the 50% probability level.

VII. Reported Electrochemical Data

Table S1. Reported electrochemical data for [Cu(dmp)₂]PF₆ and complexes **8a–d** in degassed CH₃CN (unless otherwise noted) (V vs. SCE).^{2–6} ^aMeasurements were performed in CH₂Cl₂.

[M]	E _{1/2} (M ⁺ /M [*])	E _{1/2} (M [*] /M [−])	E _{1/2} (M ⁺ /M)	E _{1/2} (M/M [−])	ref.
[Cu(dmp) ₂]PF ₆	−1.54	—	+0.63	—	2
8a	−1.44	+1.00	+1.20	−1.73	3
8b	−1.34	+1.04	+1.34	−1.64	4
8c	−1.35	+1.01	+1.26 ^a	−1.60 ^a	5
8d	−1.02	+0.63	+1.25	−1.64	6

2. D. R. McMillin, M. T. Buckner and B. T. Ahn, *Inorg. Chem.*, 1977, **16**, 943.

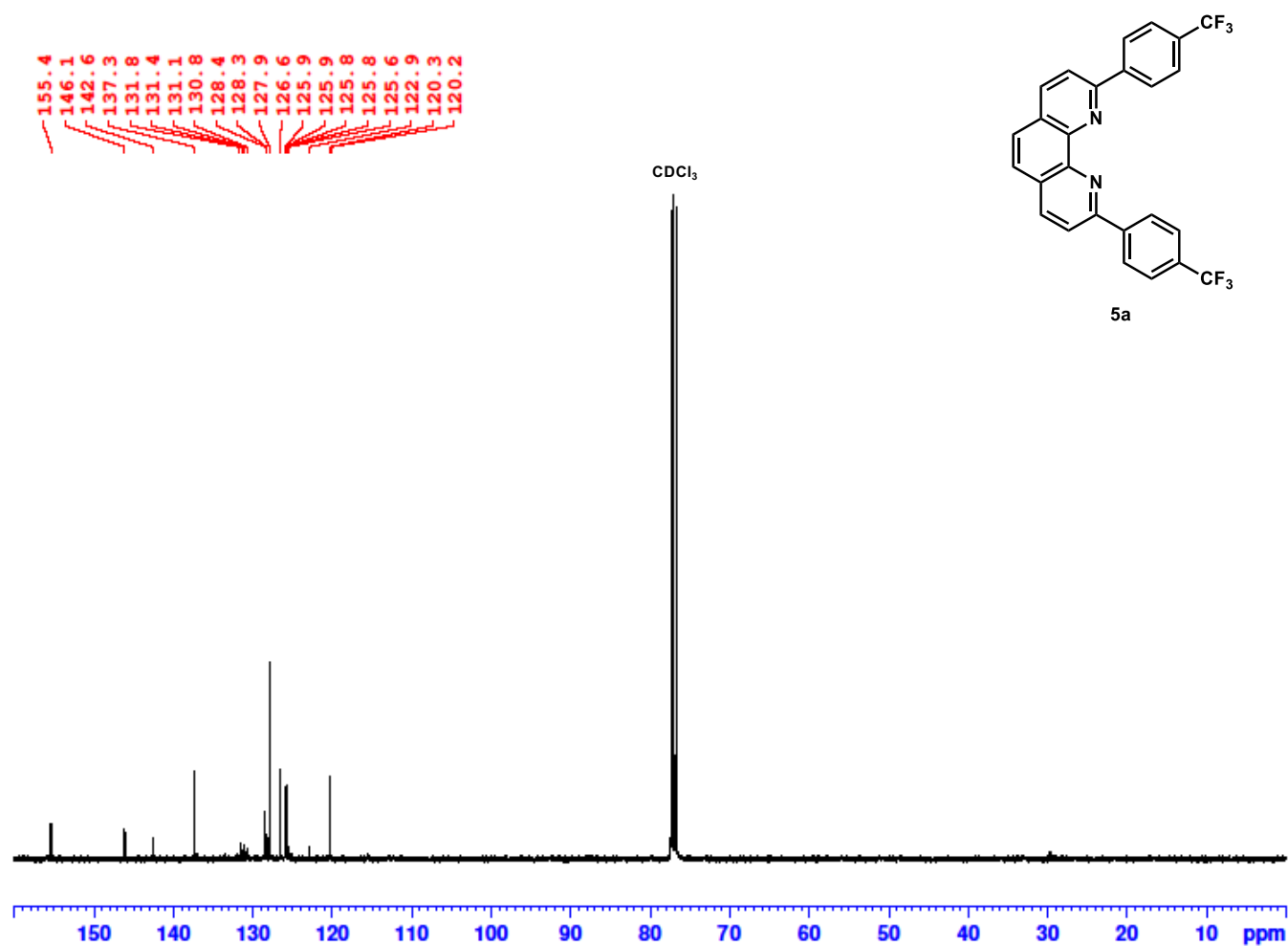
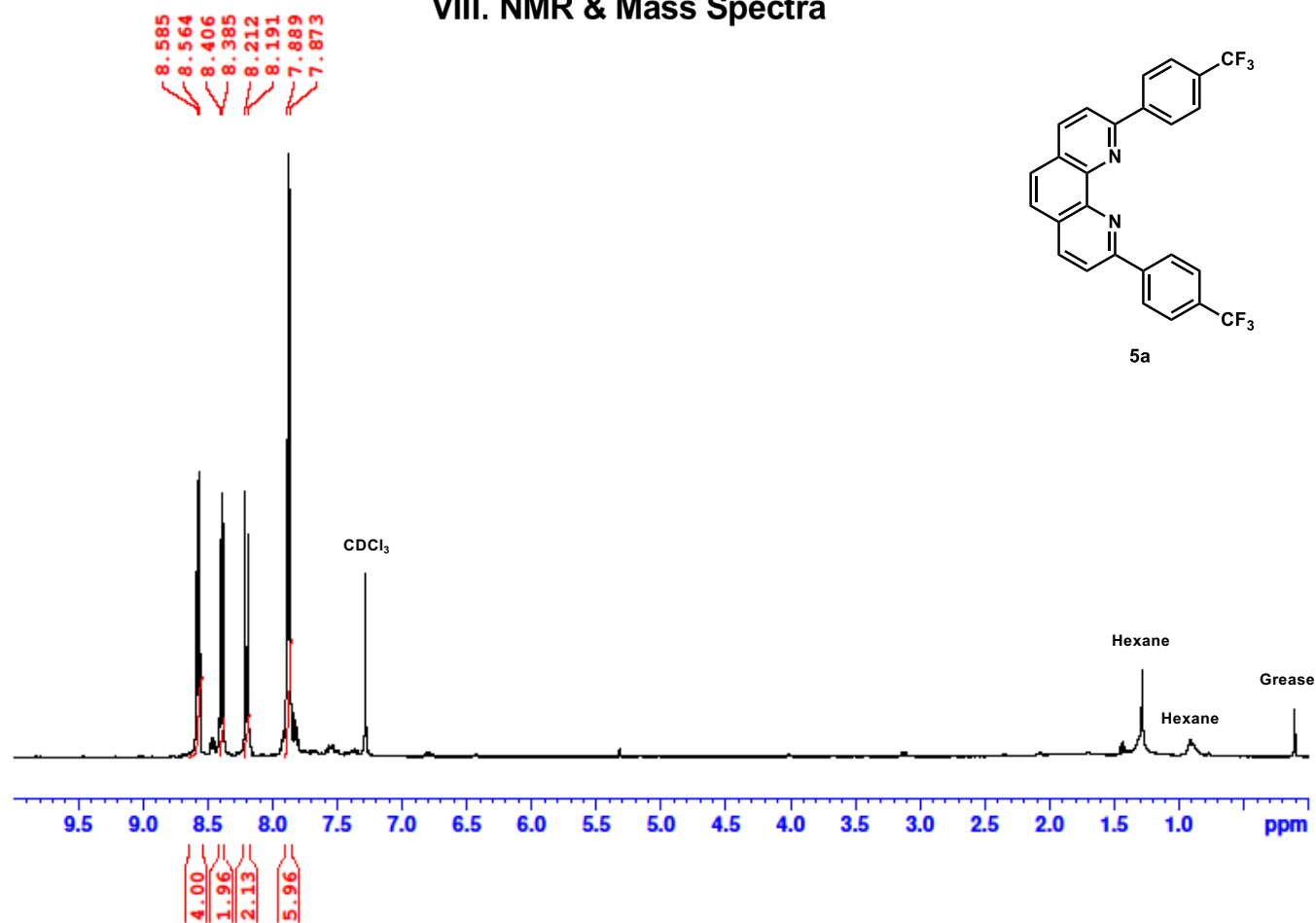
3. Y. Zhang, M. Heberle, M. Wächler, M. Karnahl and B. Dietzek, *RSC adv.*, 2016, **6**, 105801.

4. E. Mejía, S.-P. Luo, M. Karnahl, A. Friedrich, S. Tschierlei, A.-E. Surkus, H. Junge, S. Gladiali, S. Lochbrunner and M. Beller, *Chem. Eur. J.*, 2013, **19**, 15972.

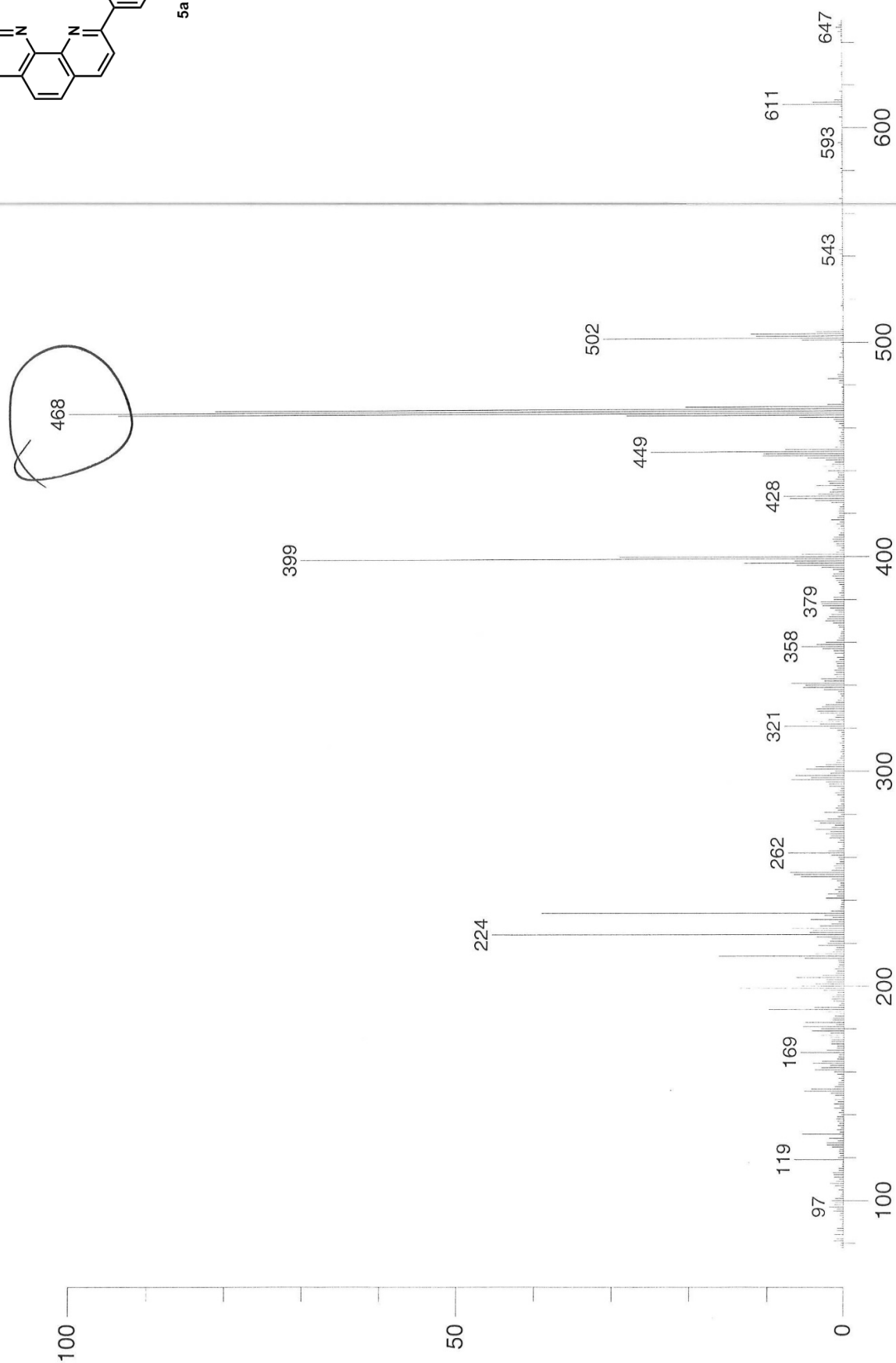
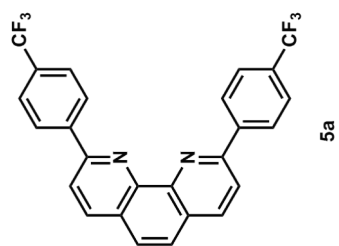
5. (a) S. Kuang, D. G. Cuttall, D. R. McMillin, P. E. Fanwick, R. A. Walton and W. Lafayette, *Inorg. Chem.*, 2002, **41**, 3313; (b) D. G. Cuttall, S.-M. Kuang, P. E. Fanwick, D. R. McMillan and R. A. Walton, *J. Am. Chem. Soc.*, 2002, **124**, 6.

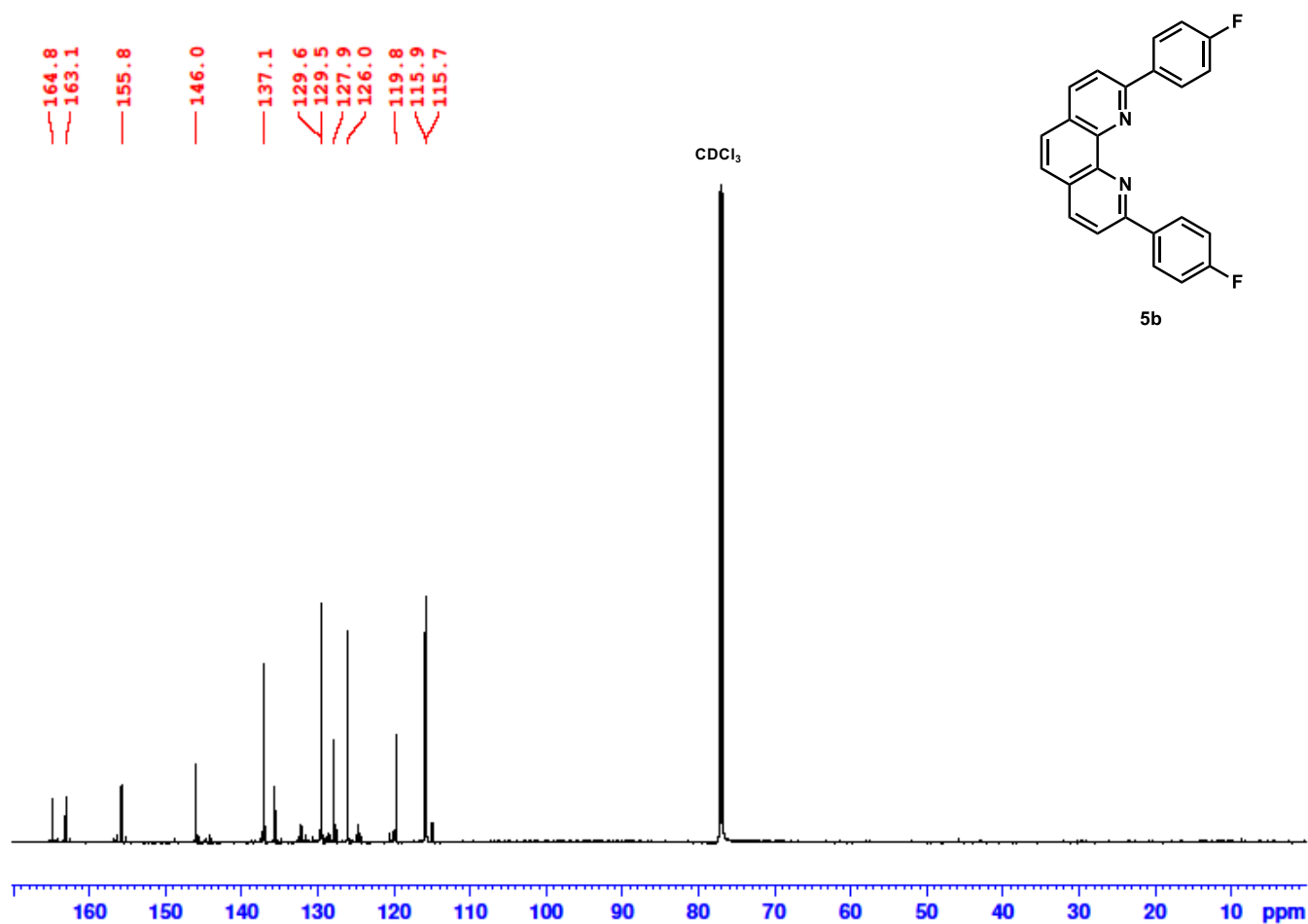
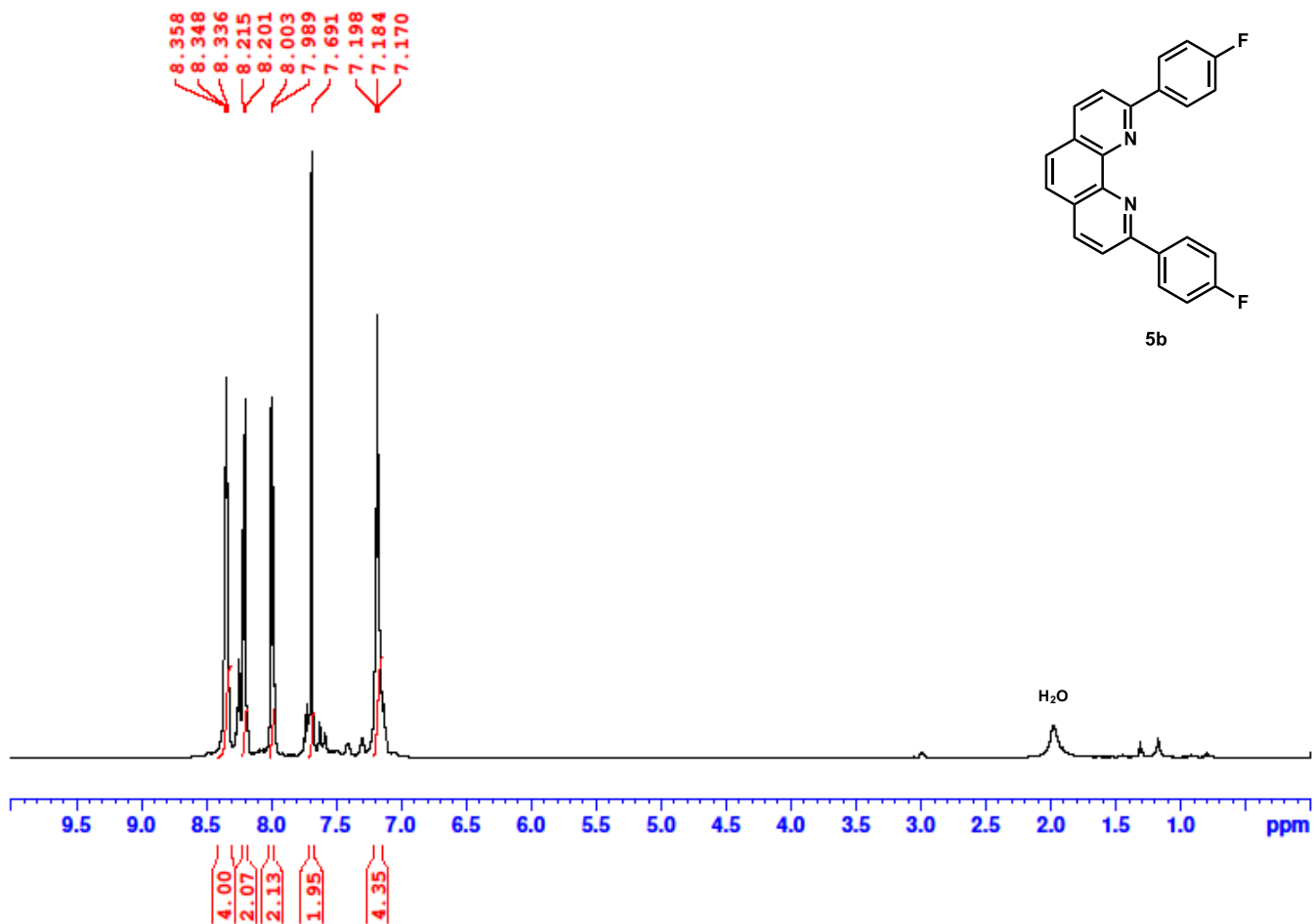
6. A. Call, C. Casadevall, F. Acuña-Parés, A. Casitas and J. Lloret-Fillol, *Chem. Sci.*, 2017, **8**, 4739.

VIII. NMR & Mass Spectra



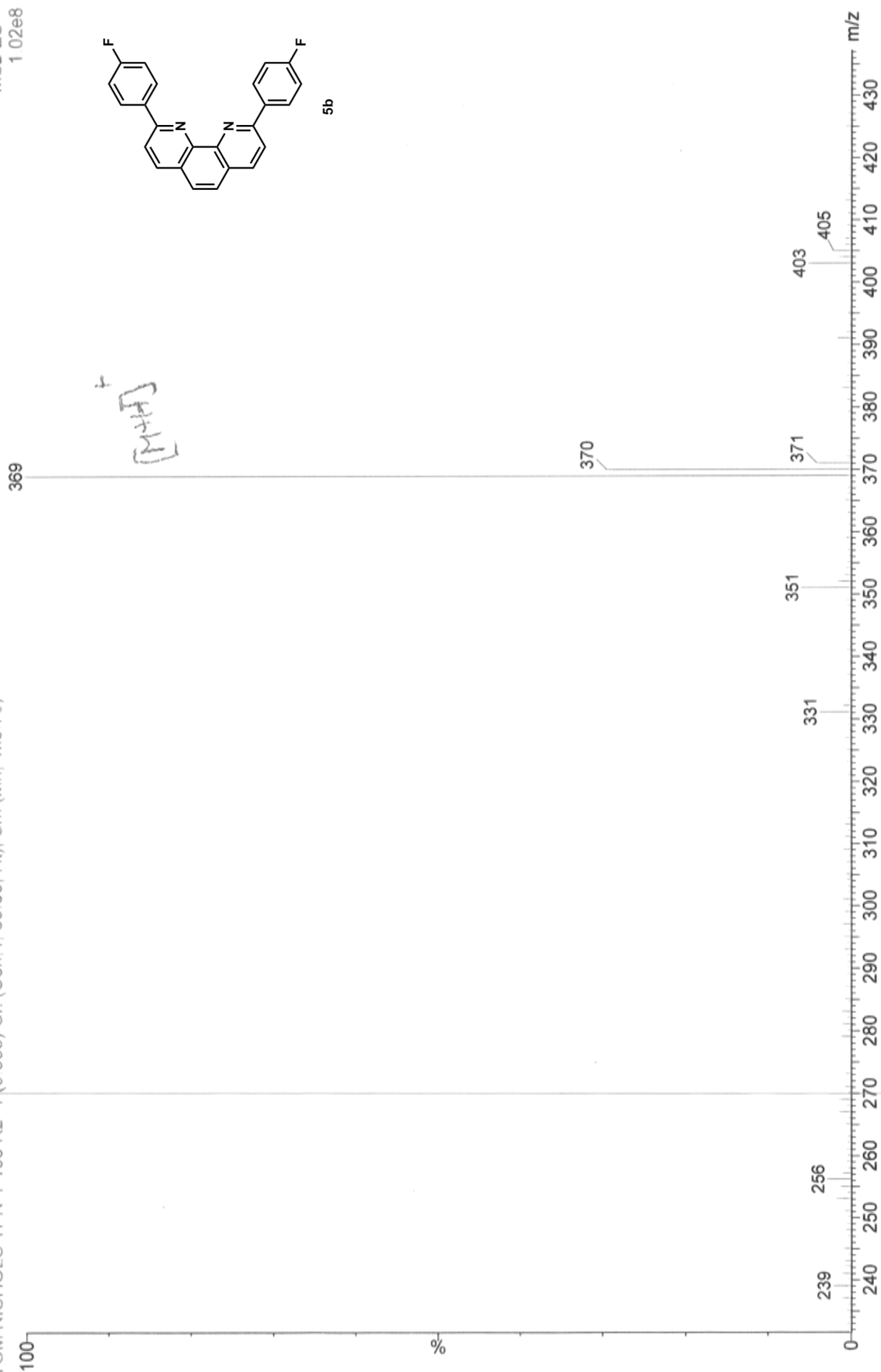
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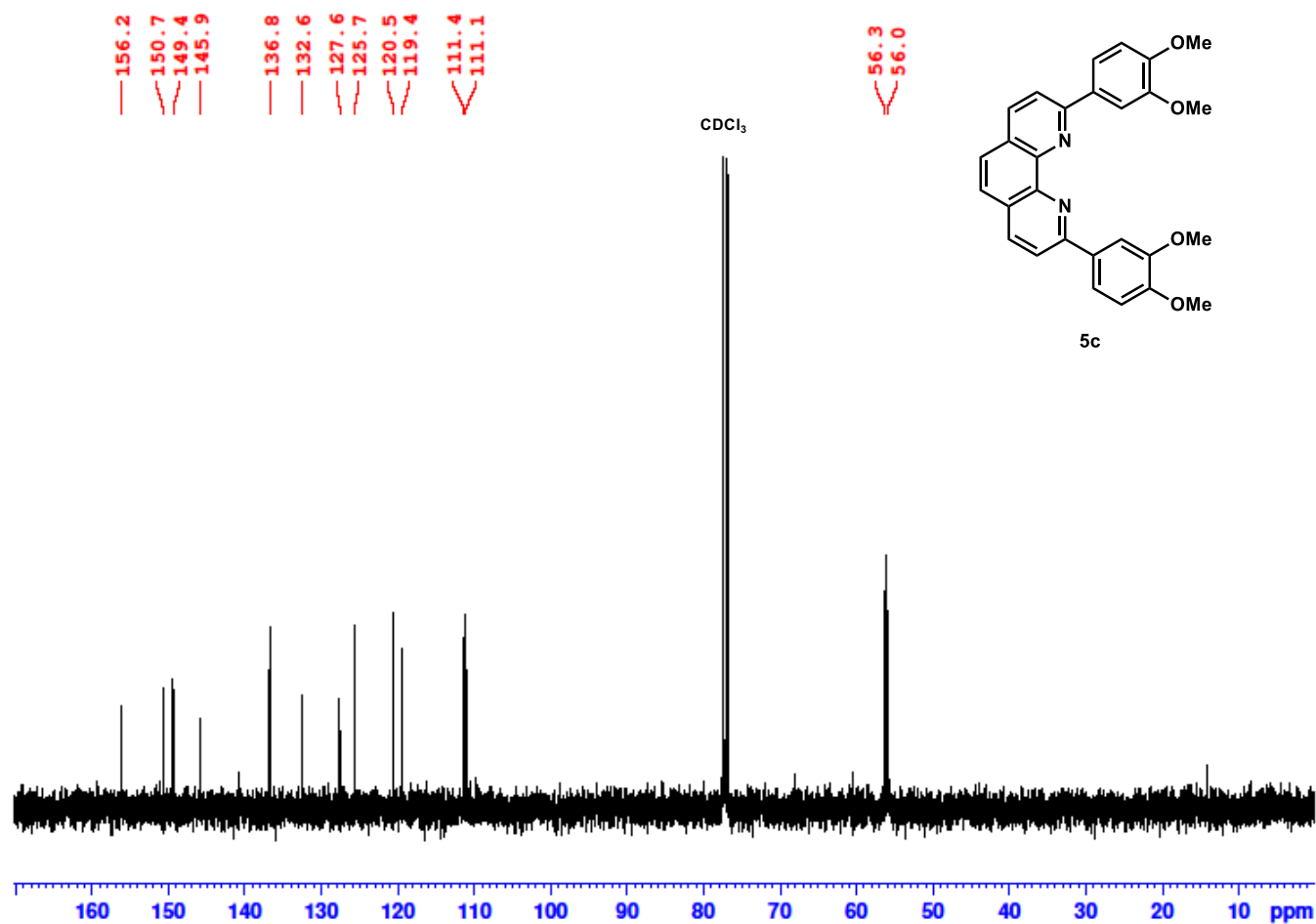
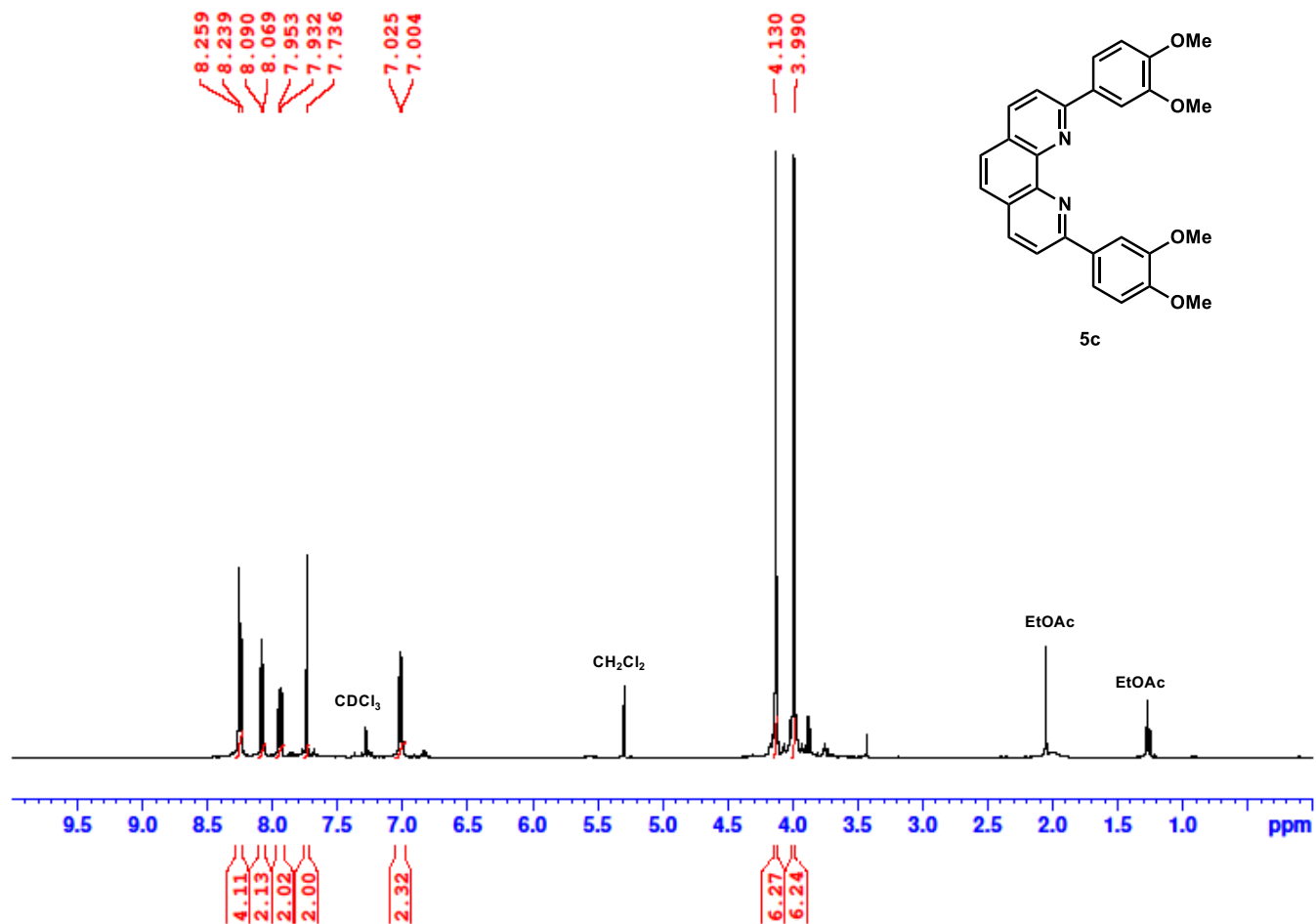




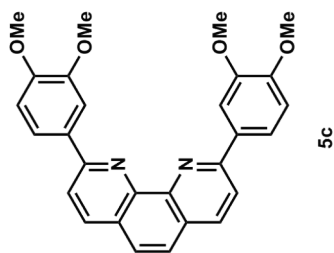
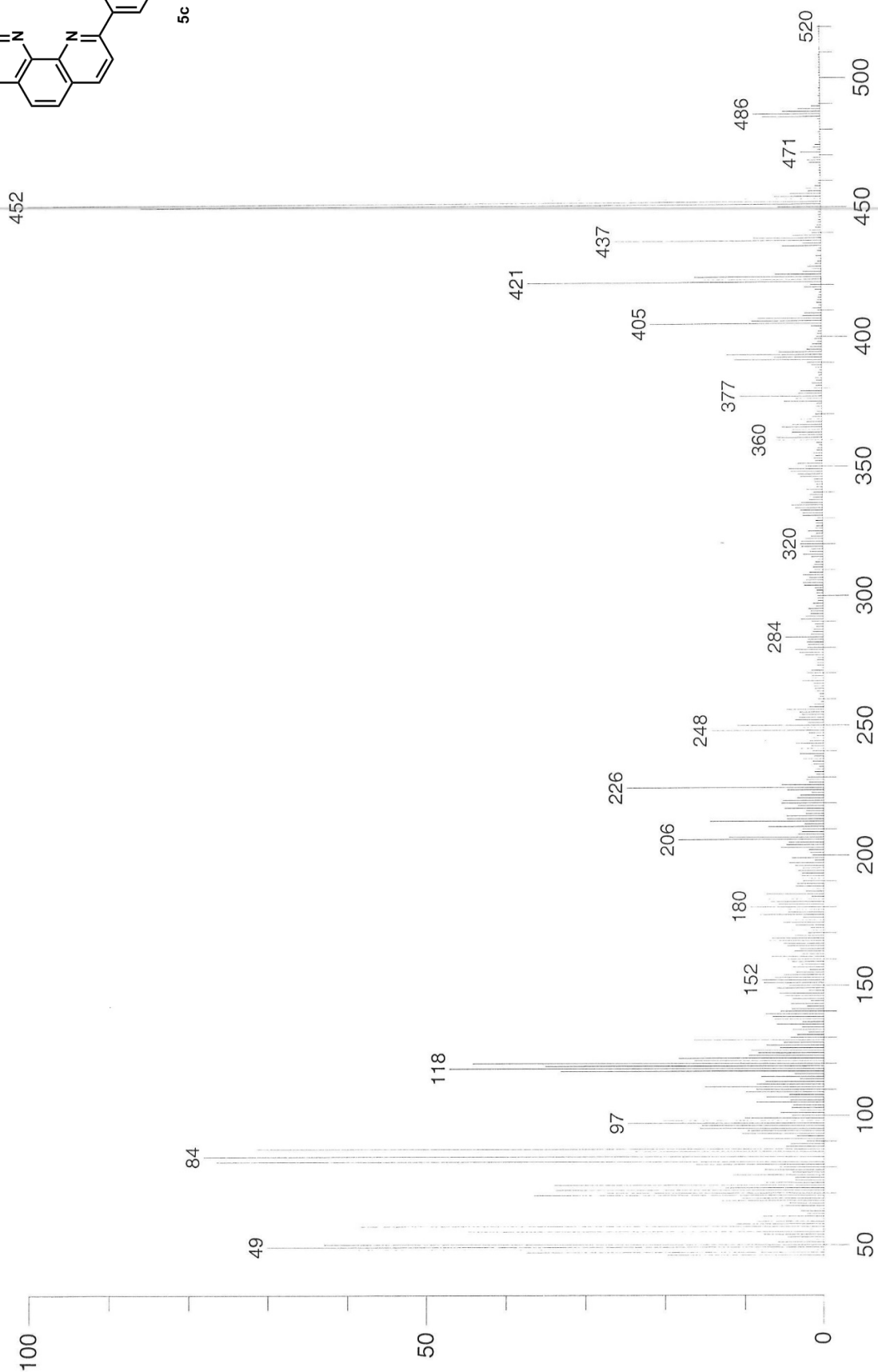
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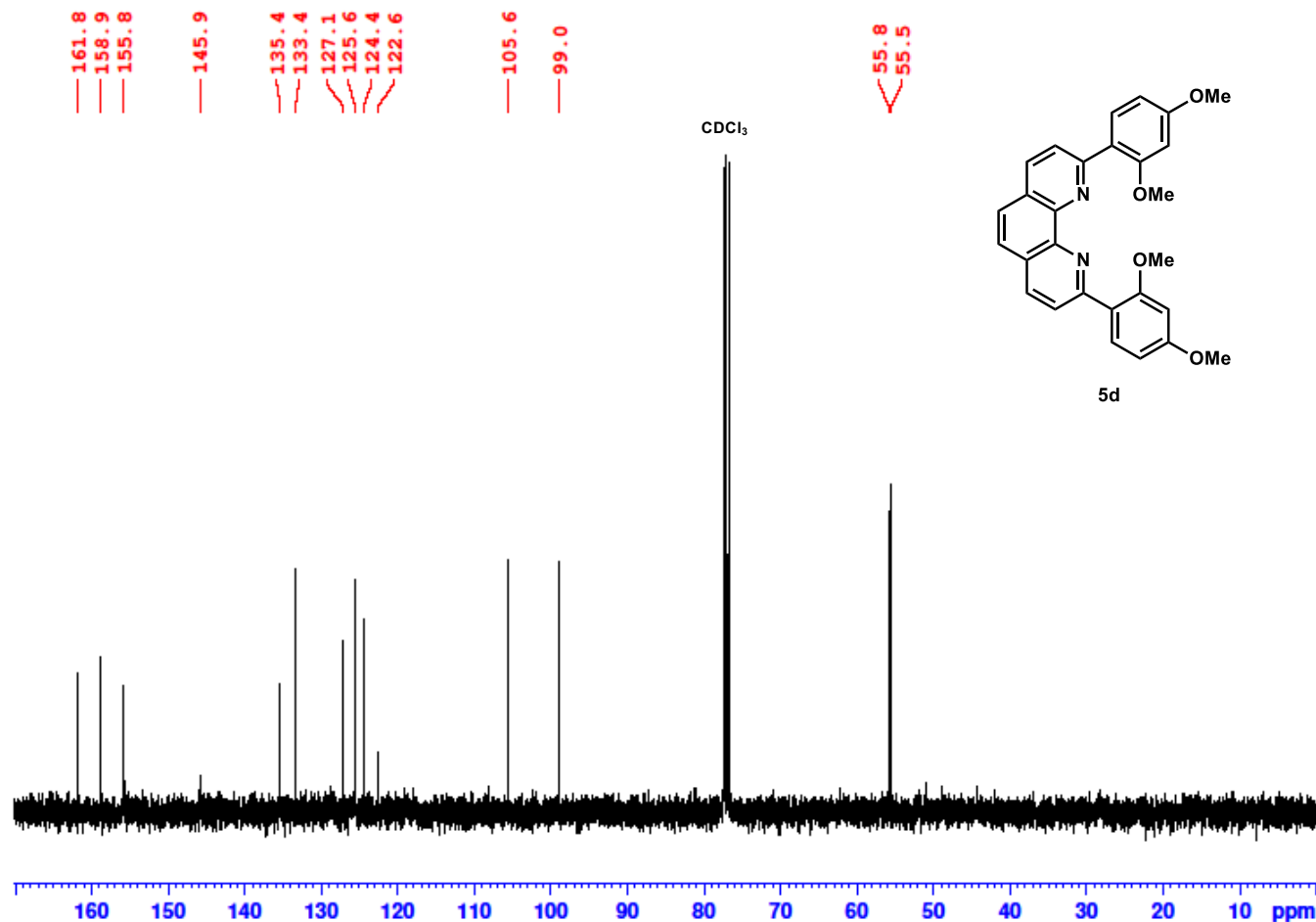
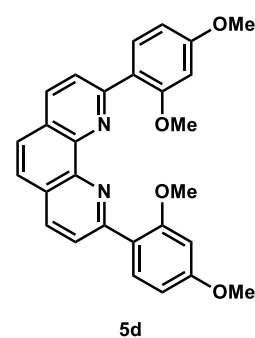
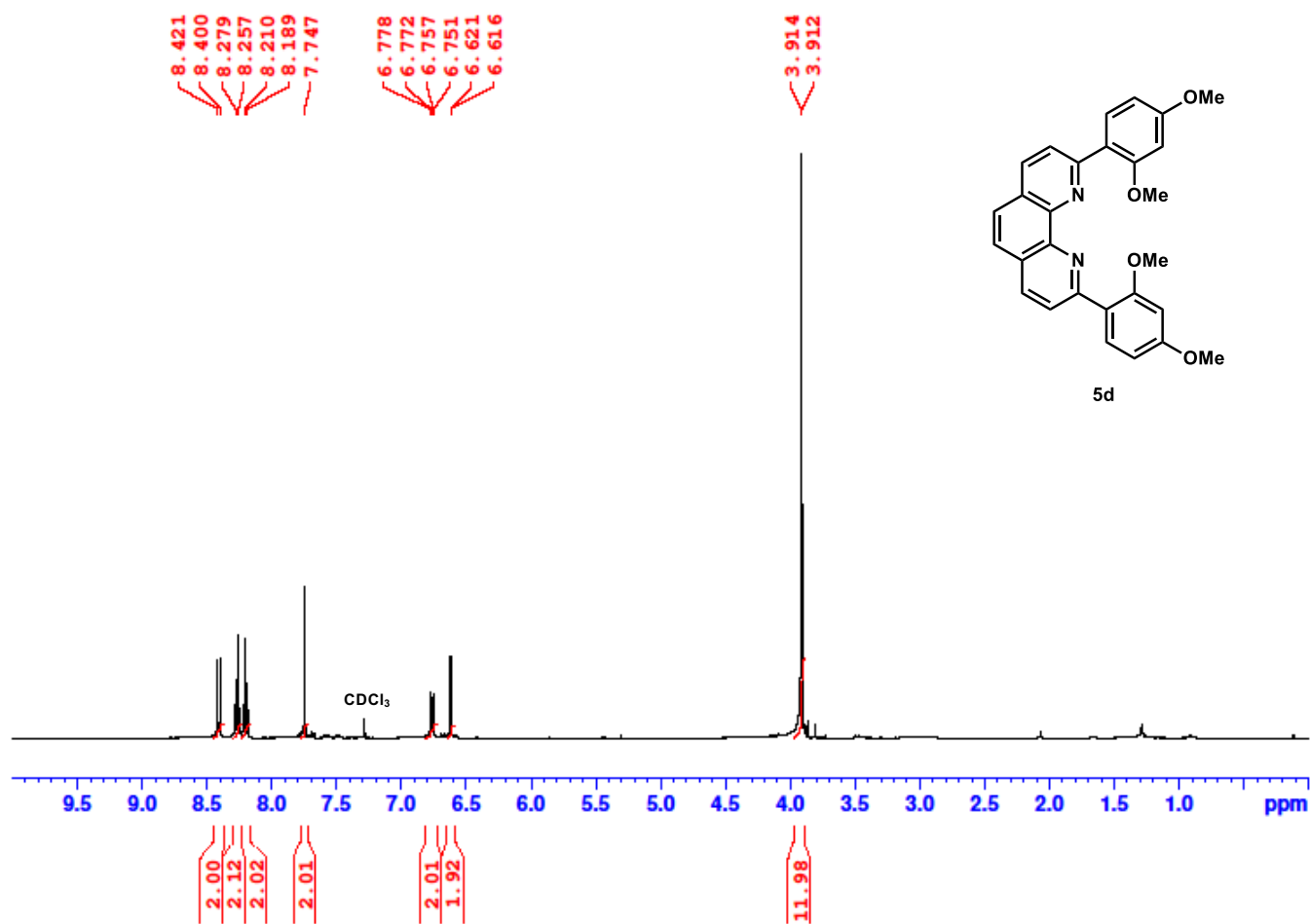
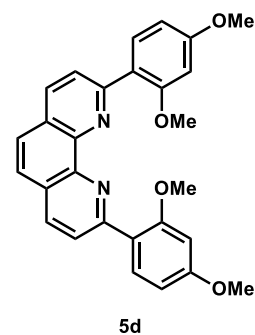
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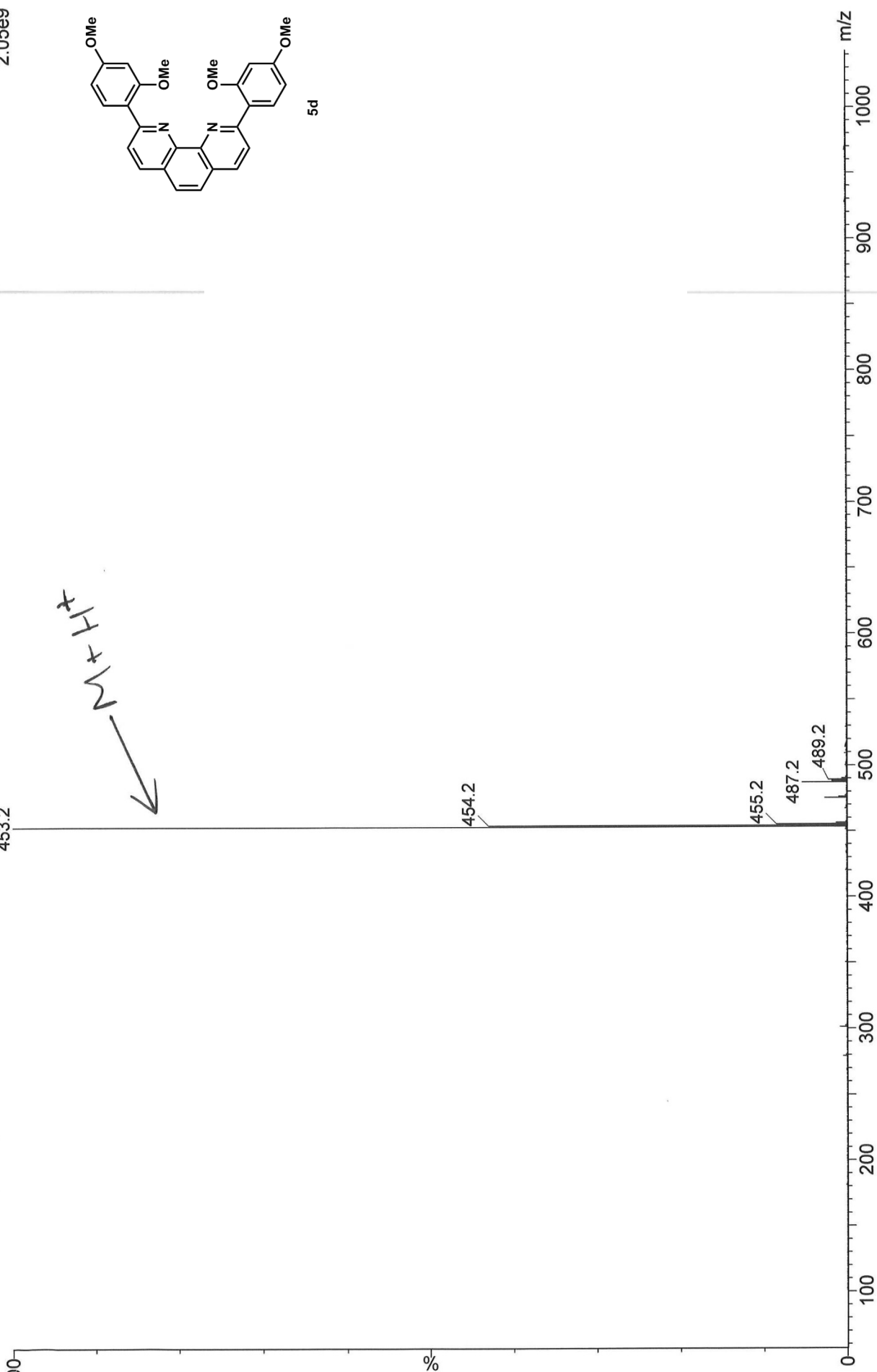
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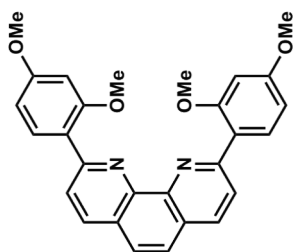


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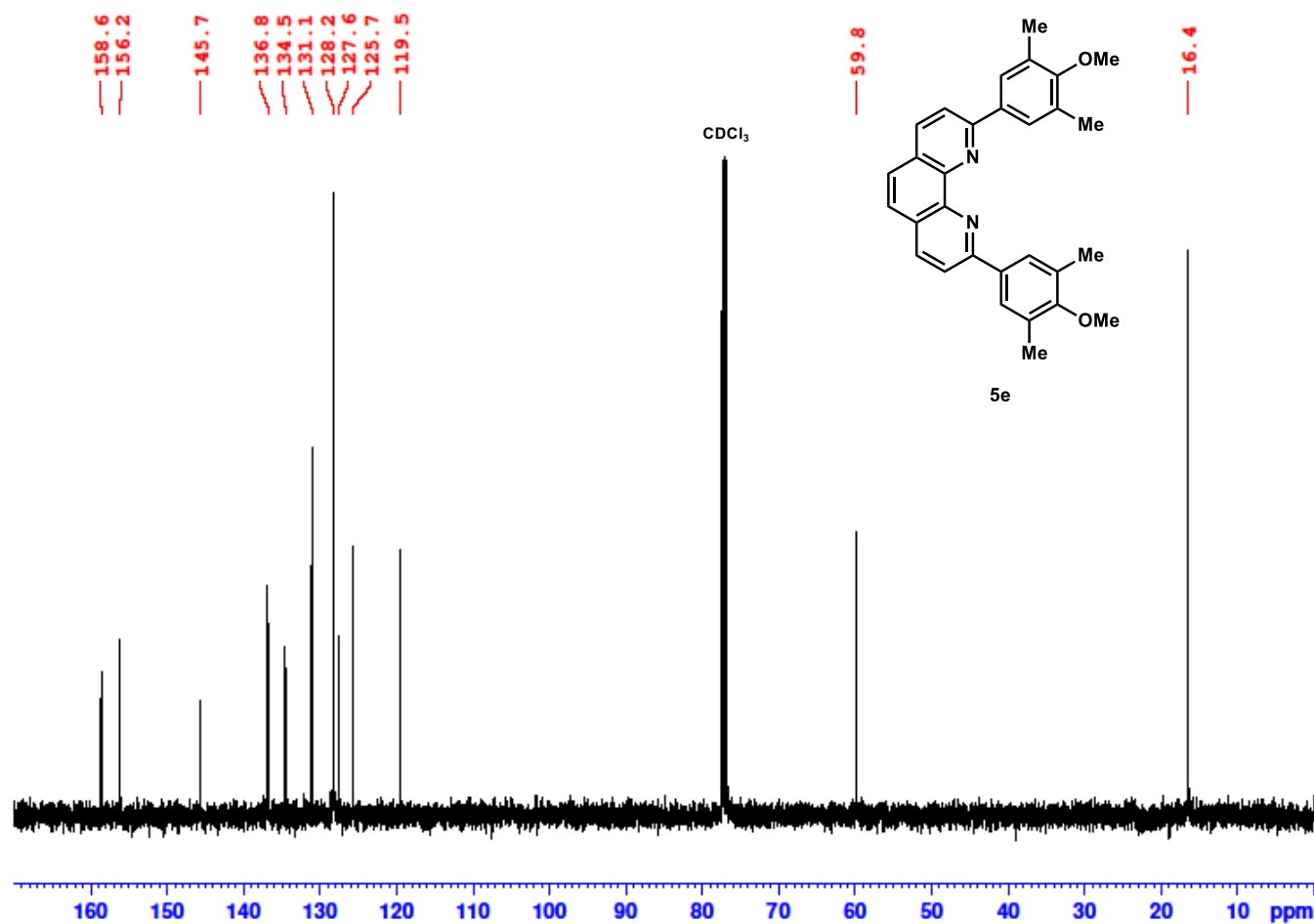
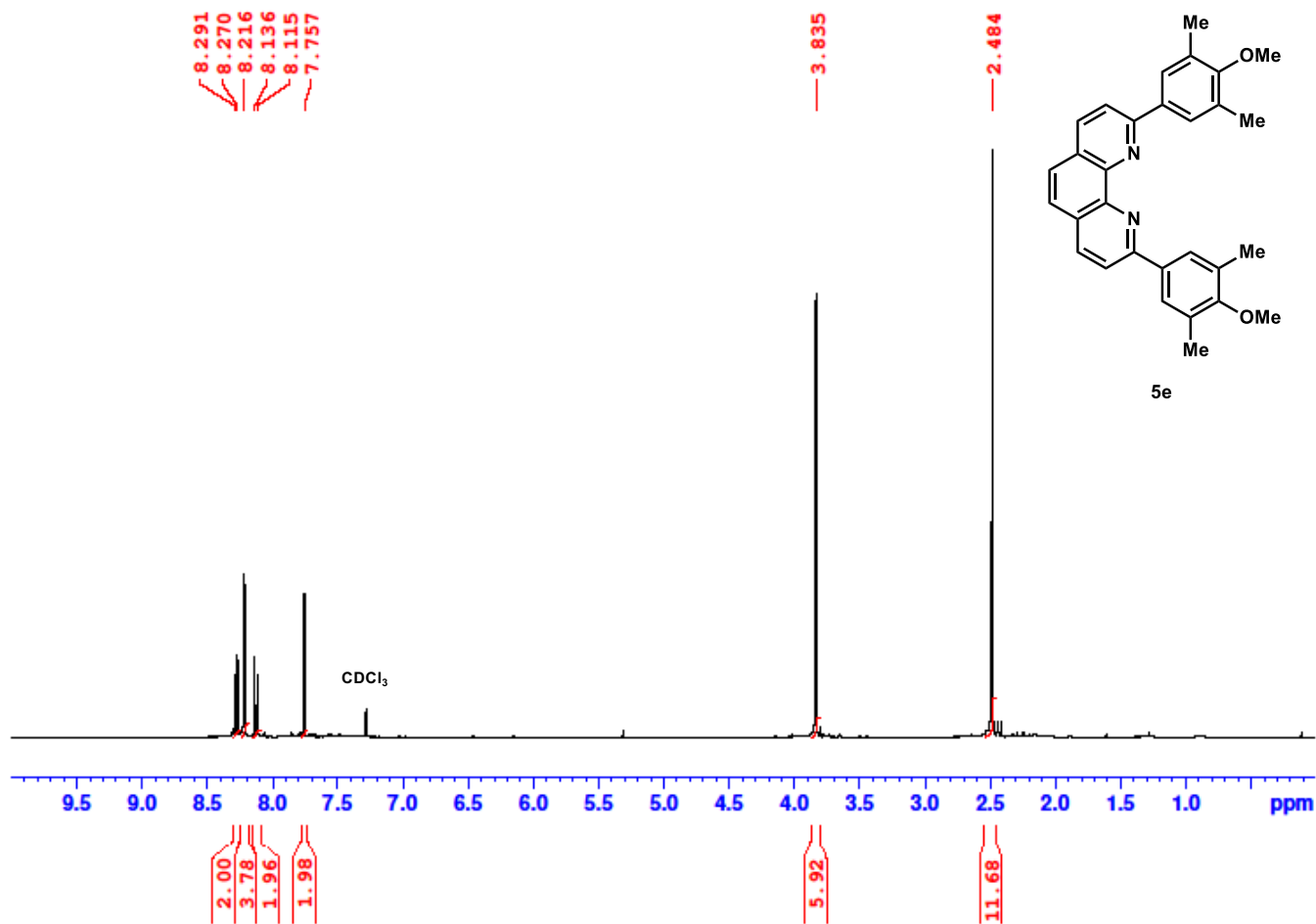
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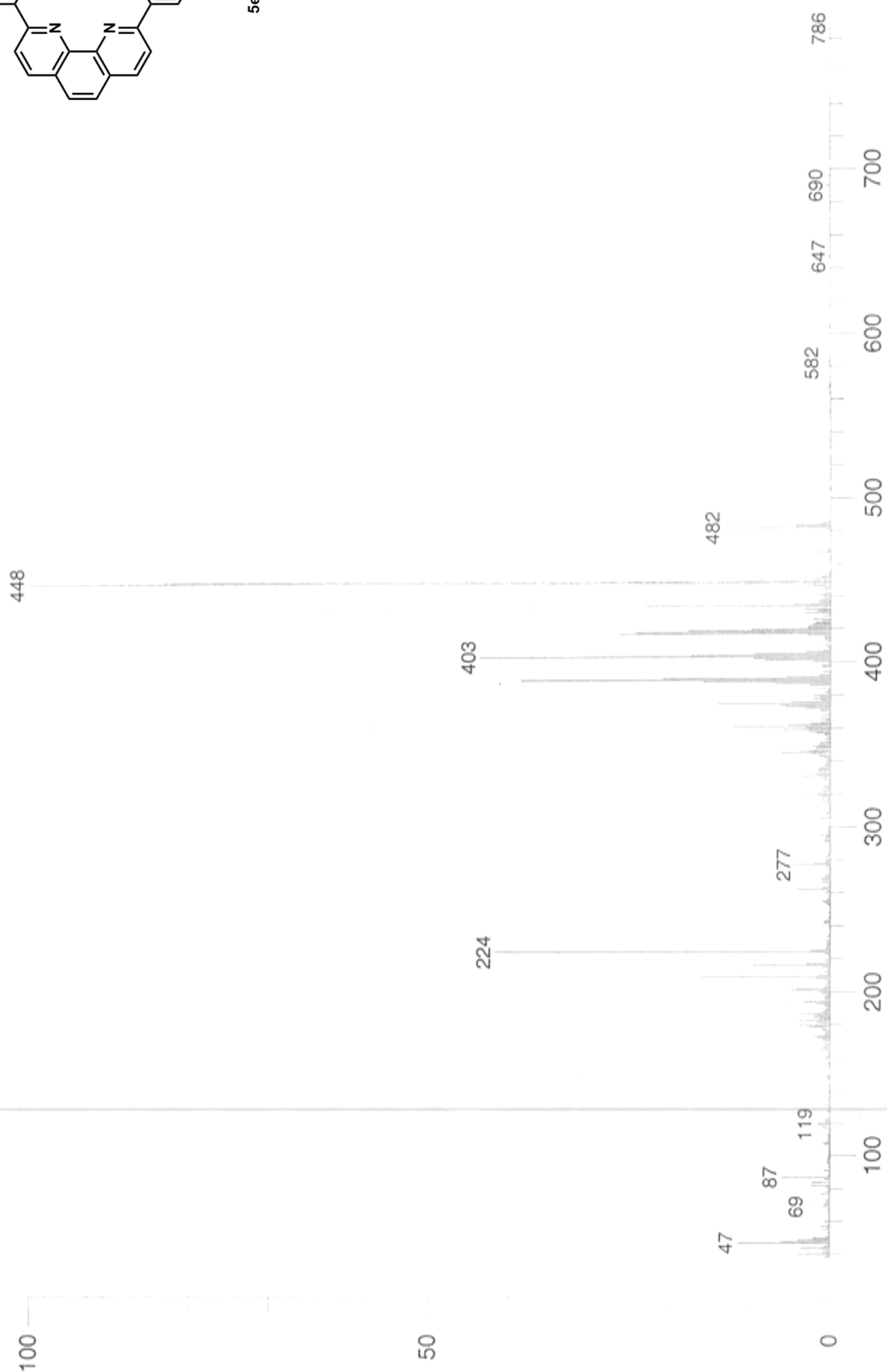
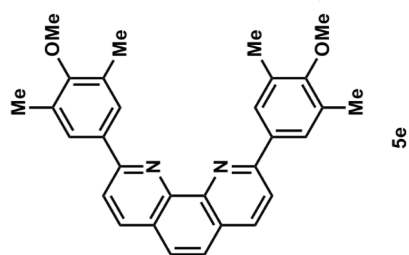
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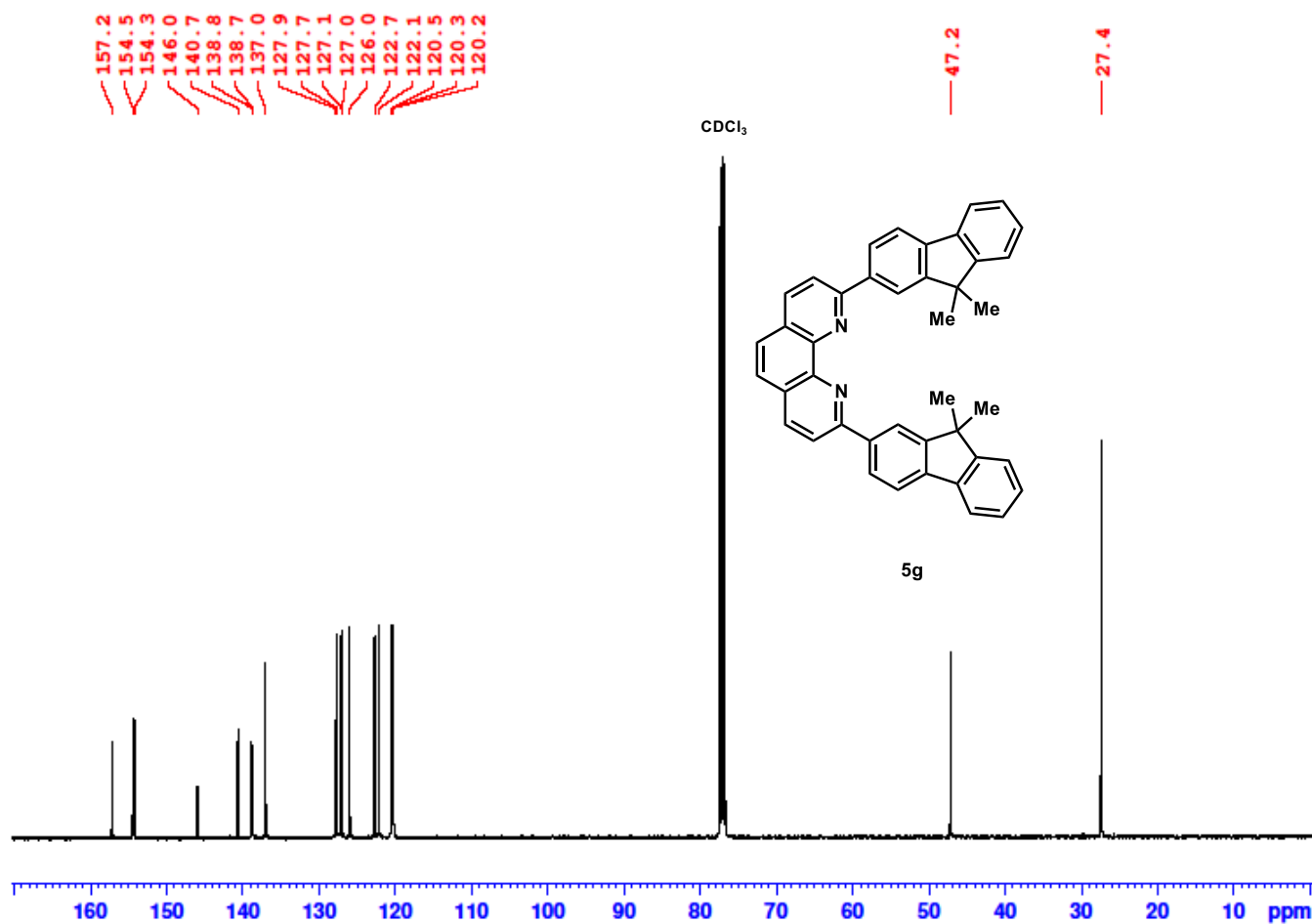
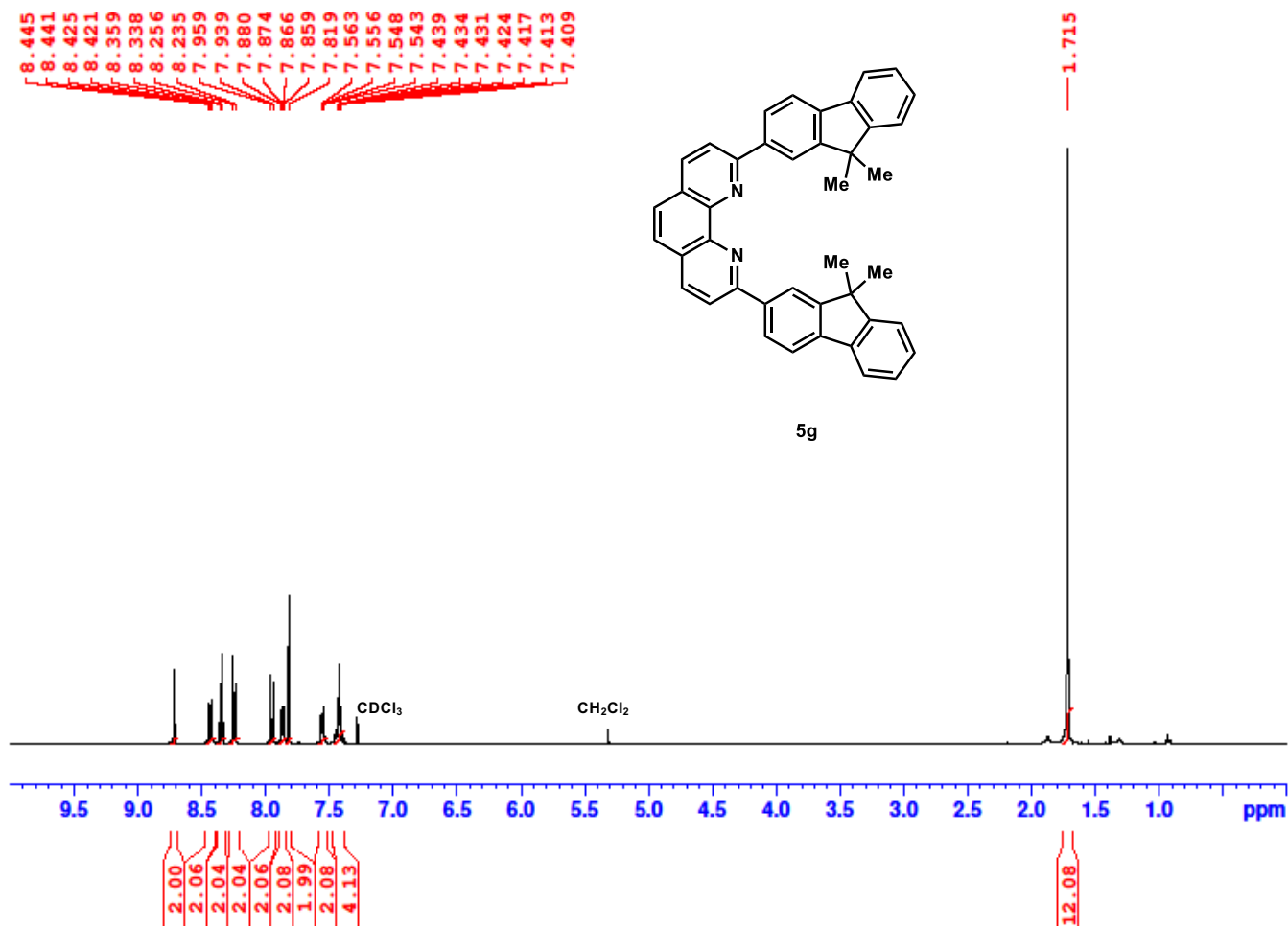


5d



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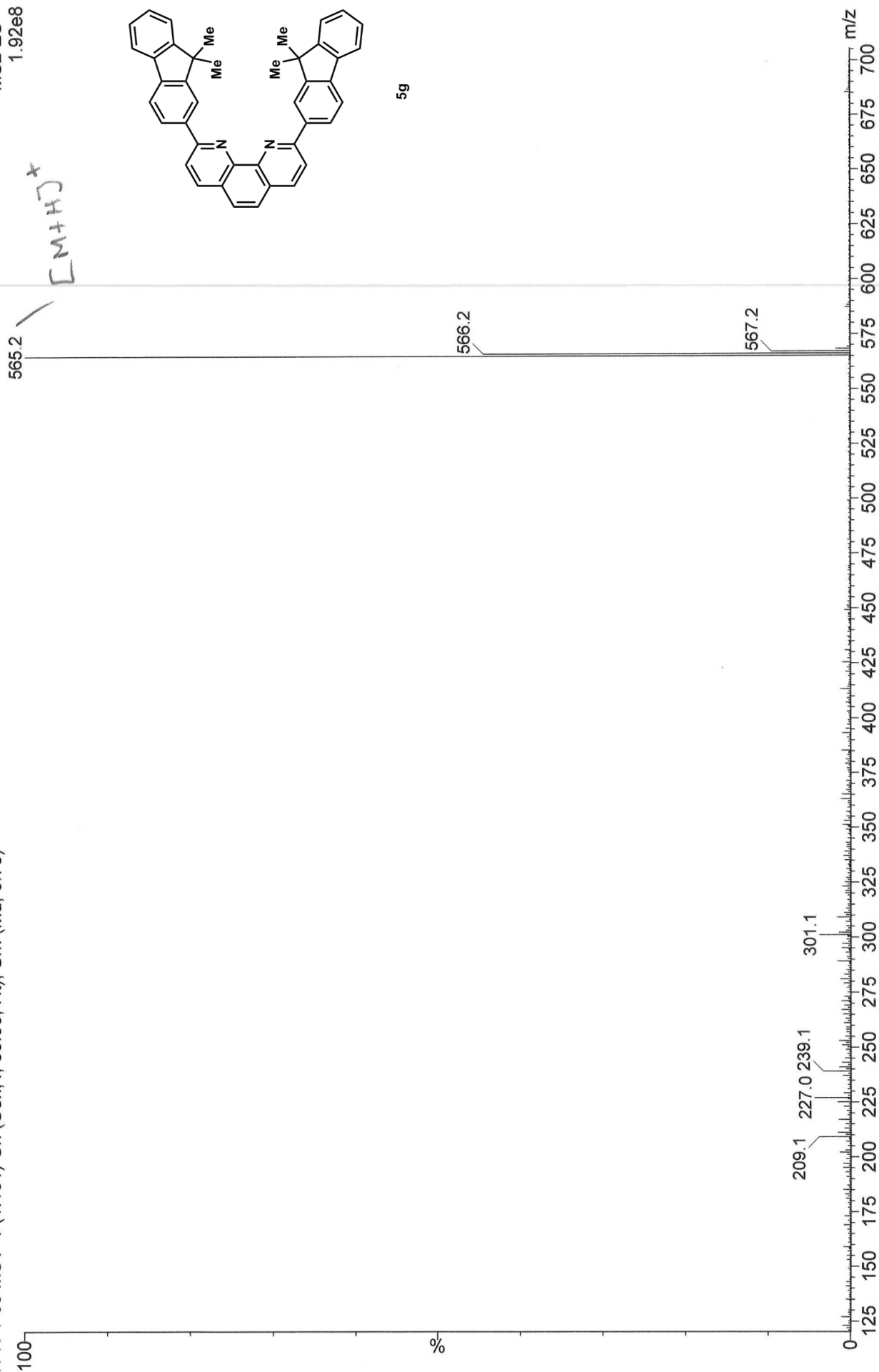


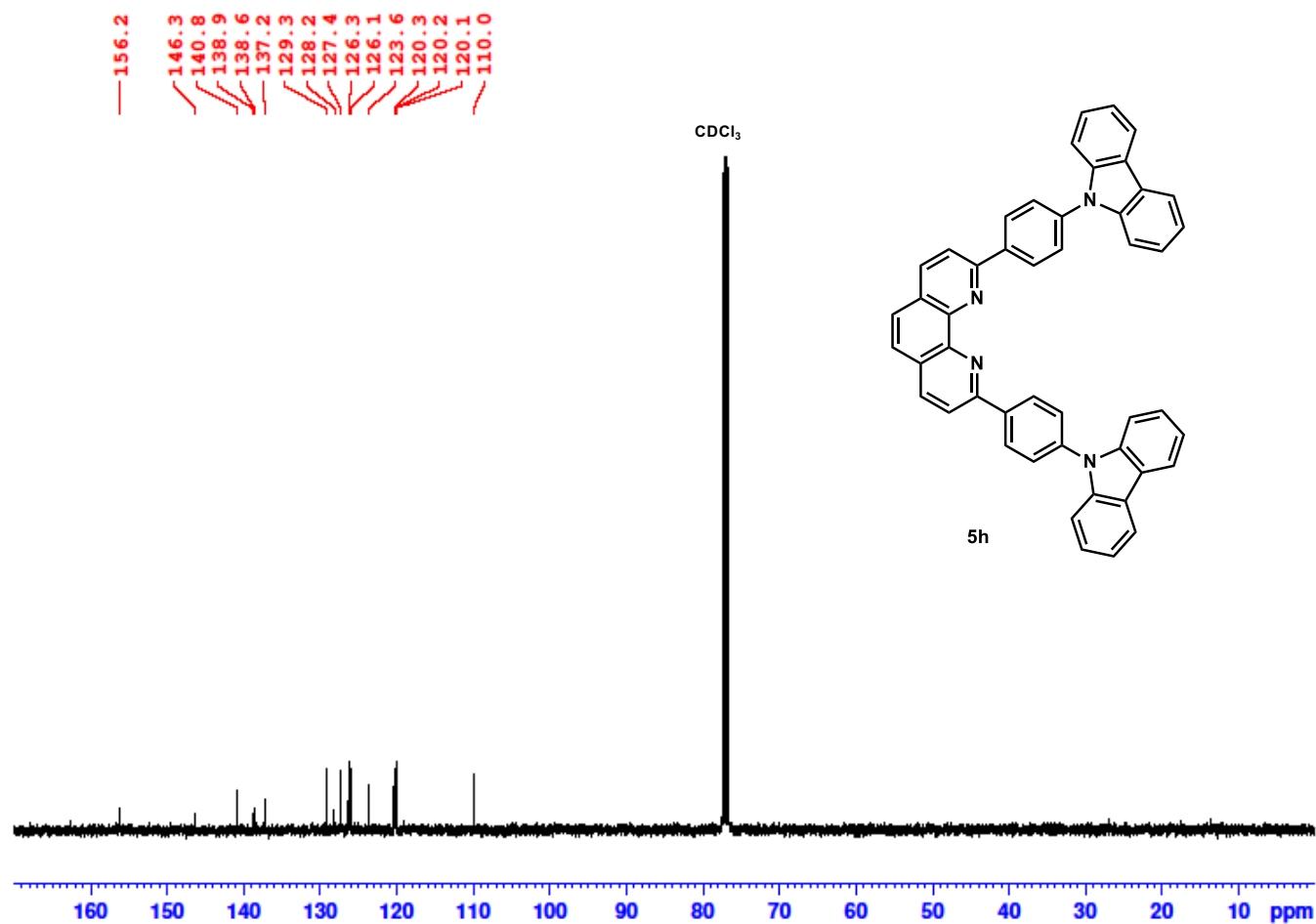
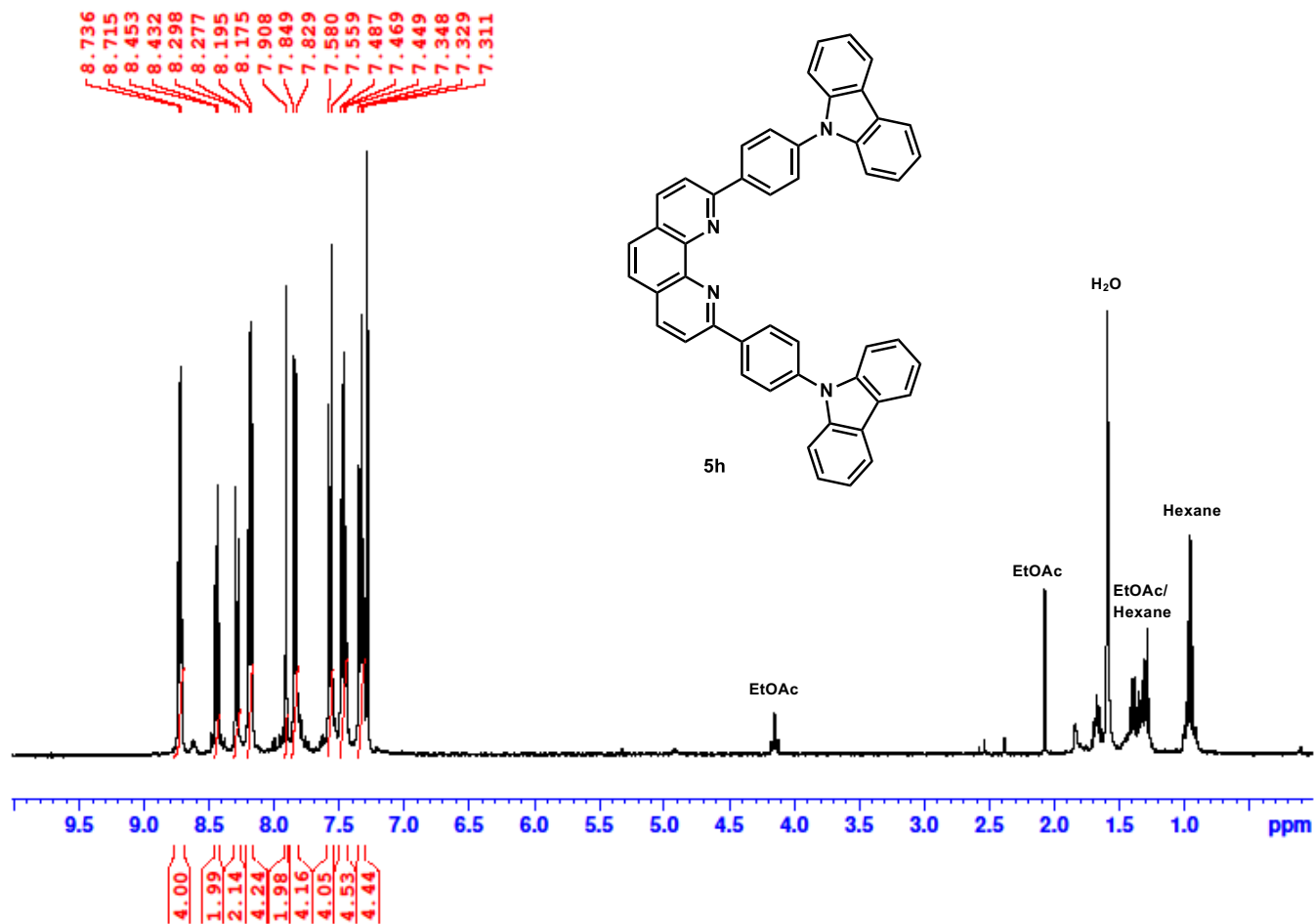


cap 2.8 kV cone 30V PESI Infusion

TPN-1-65 MS1 1 (1.101) Cn (Cen, 4, 80.00, Ht); Sm (Md, 0.70)

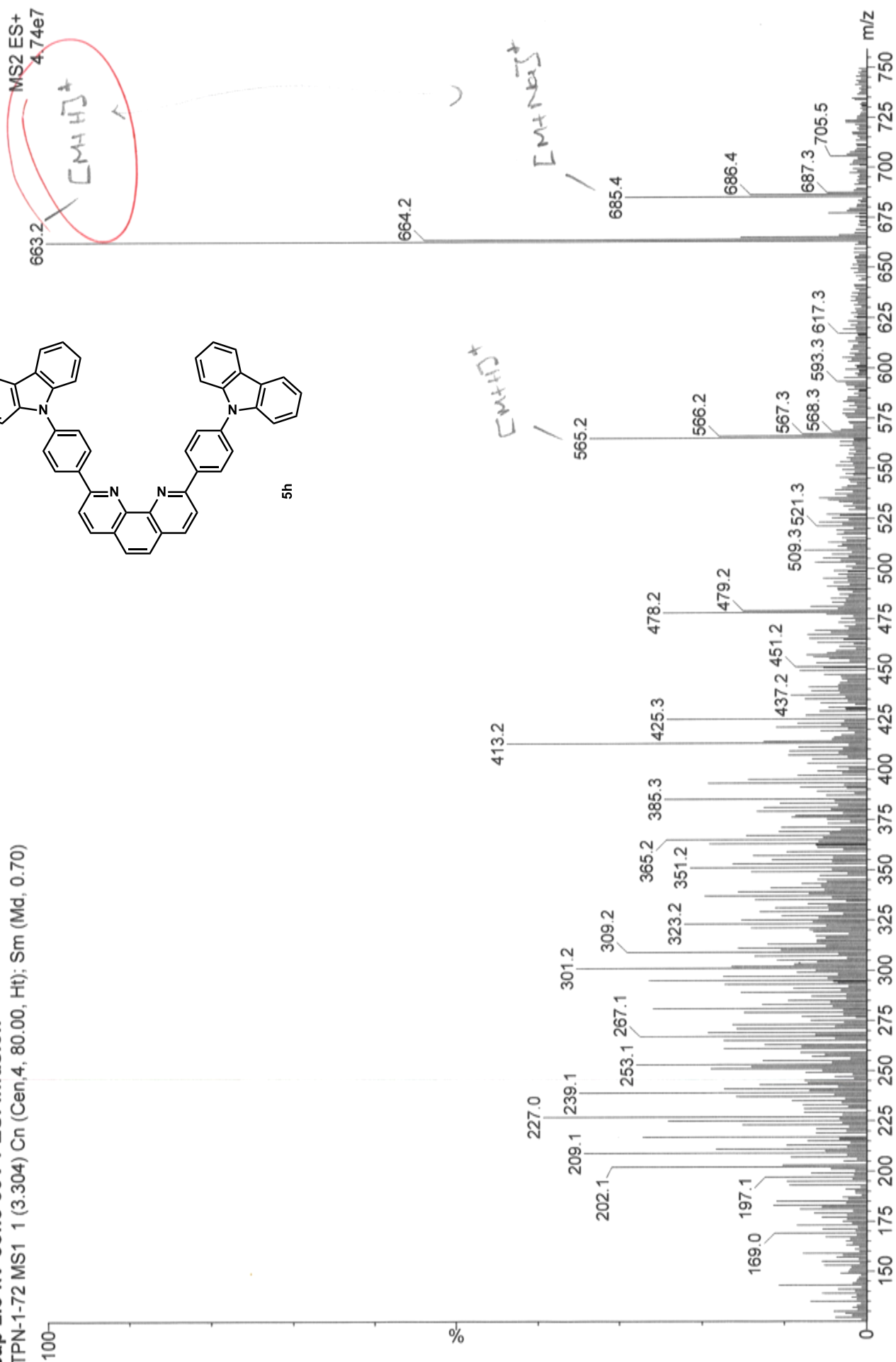
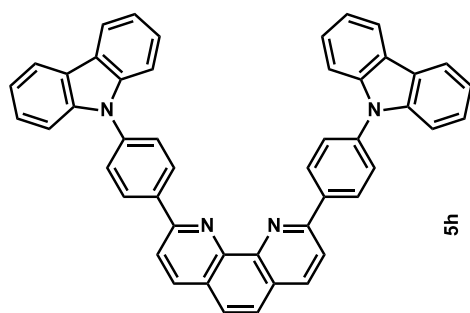
MS2 ES+
1.92e8

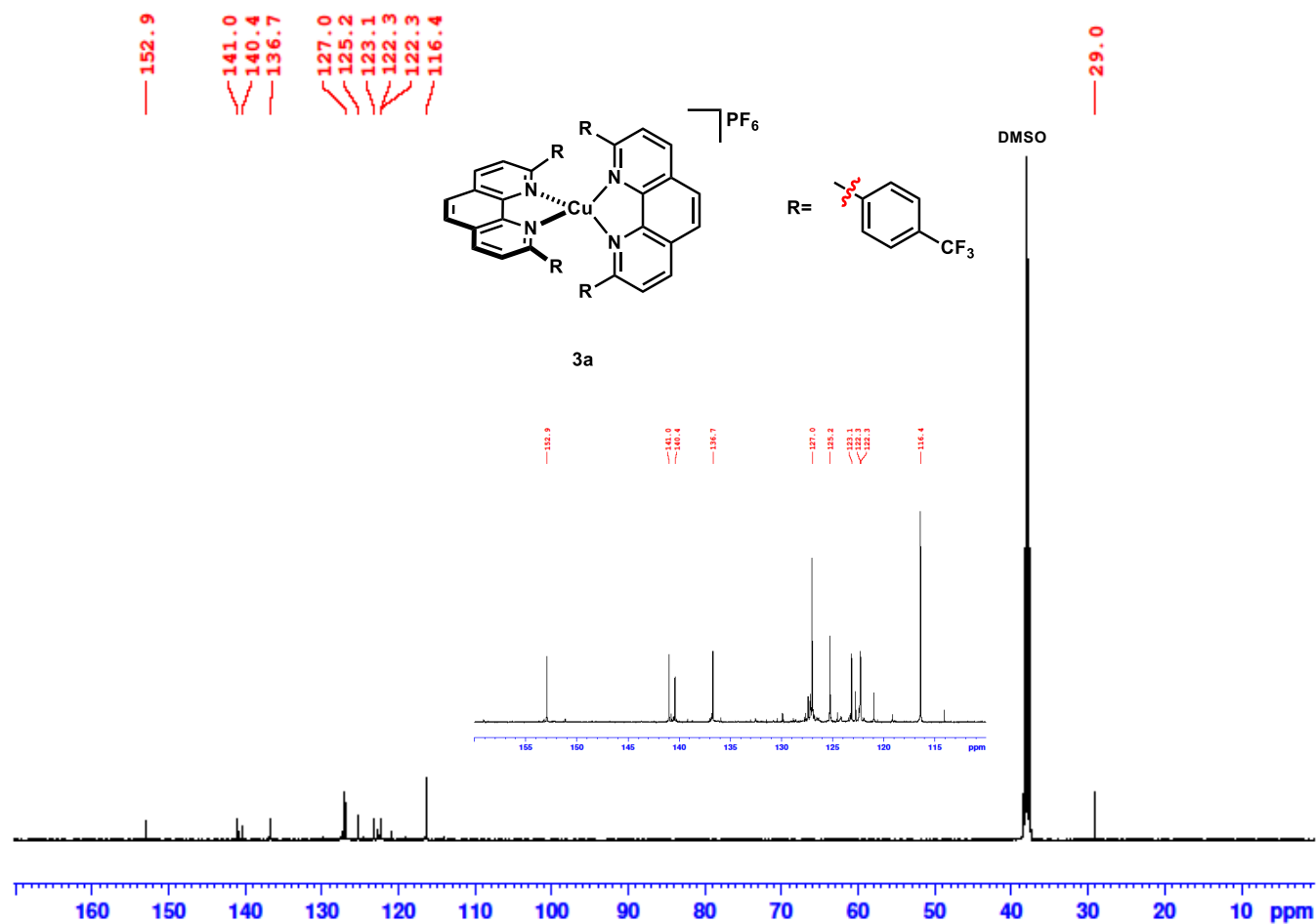
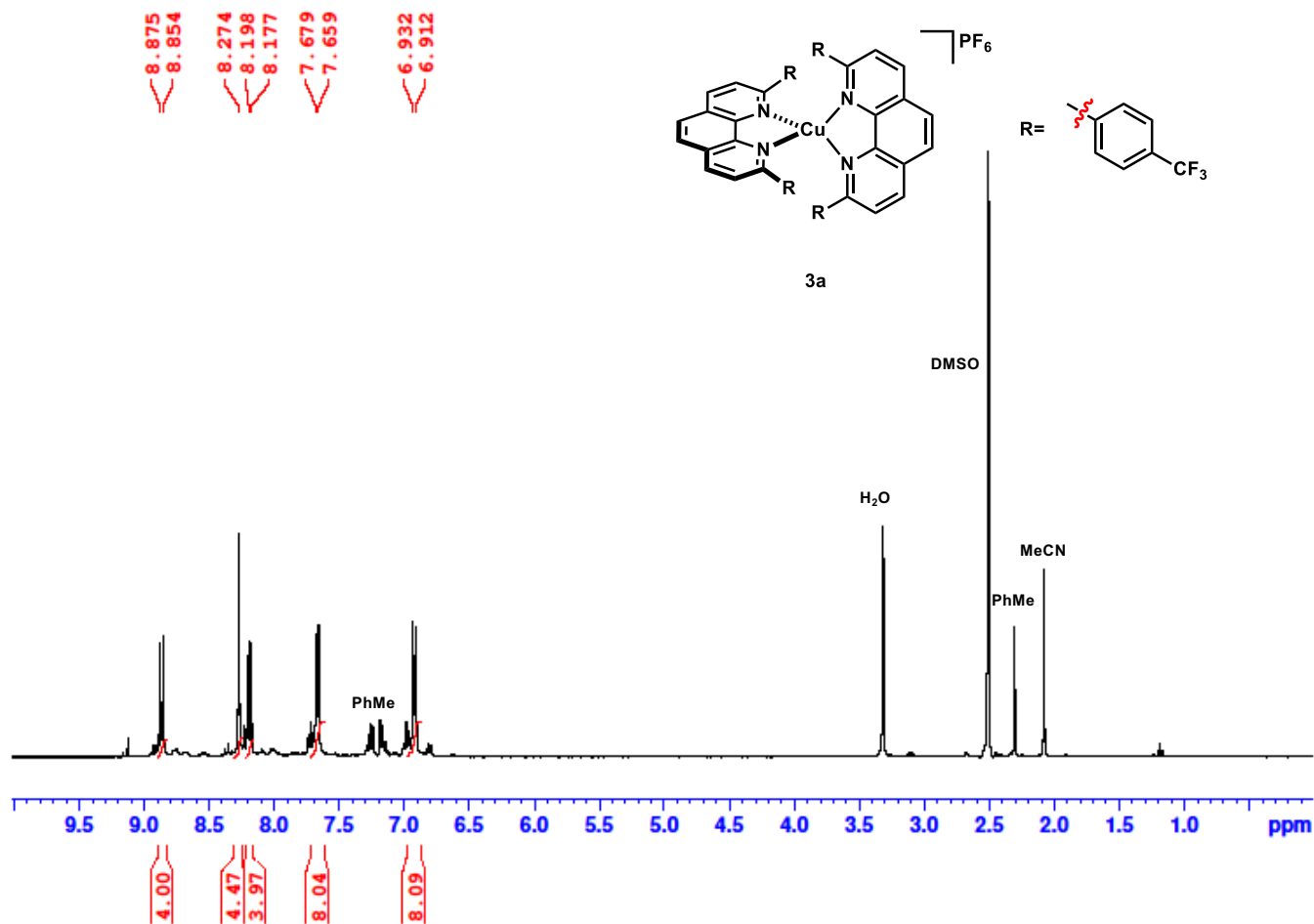




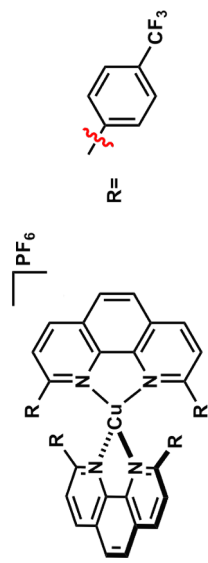
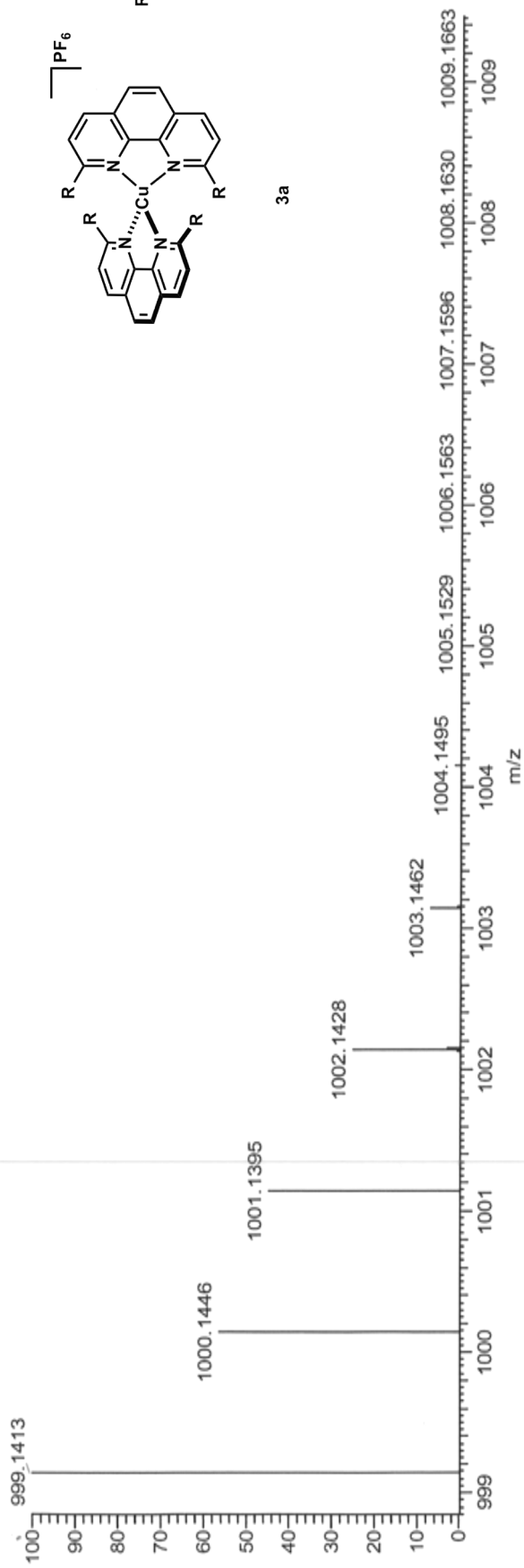
cap 2.8 kV cone 30V PESI Infusion

TPN-1-72 MS1 1 (3.304) Cn (Cen, 4, 80.00, Ht); Sm (Md, 0.70)



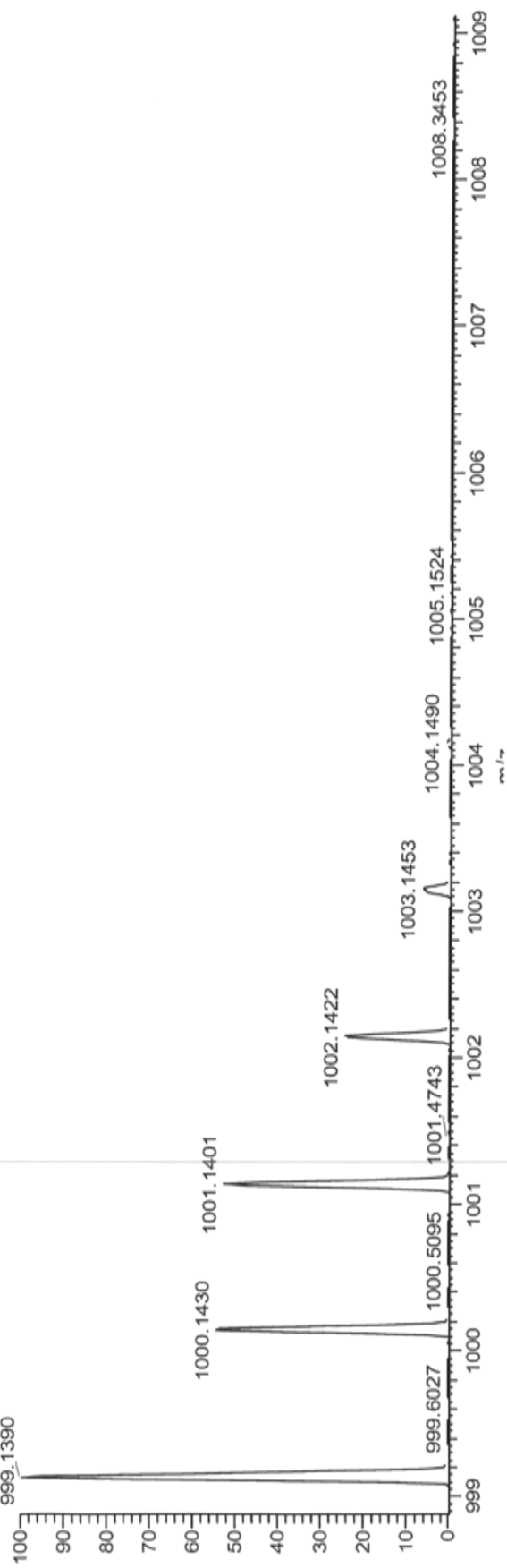


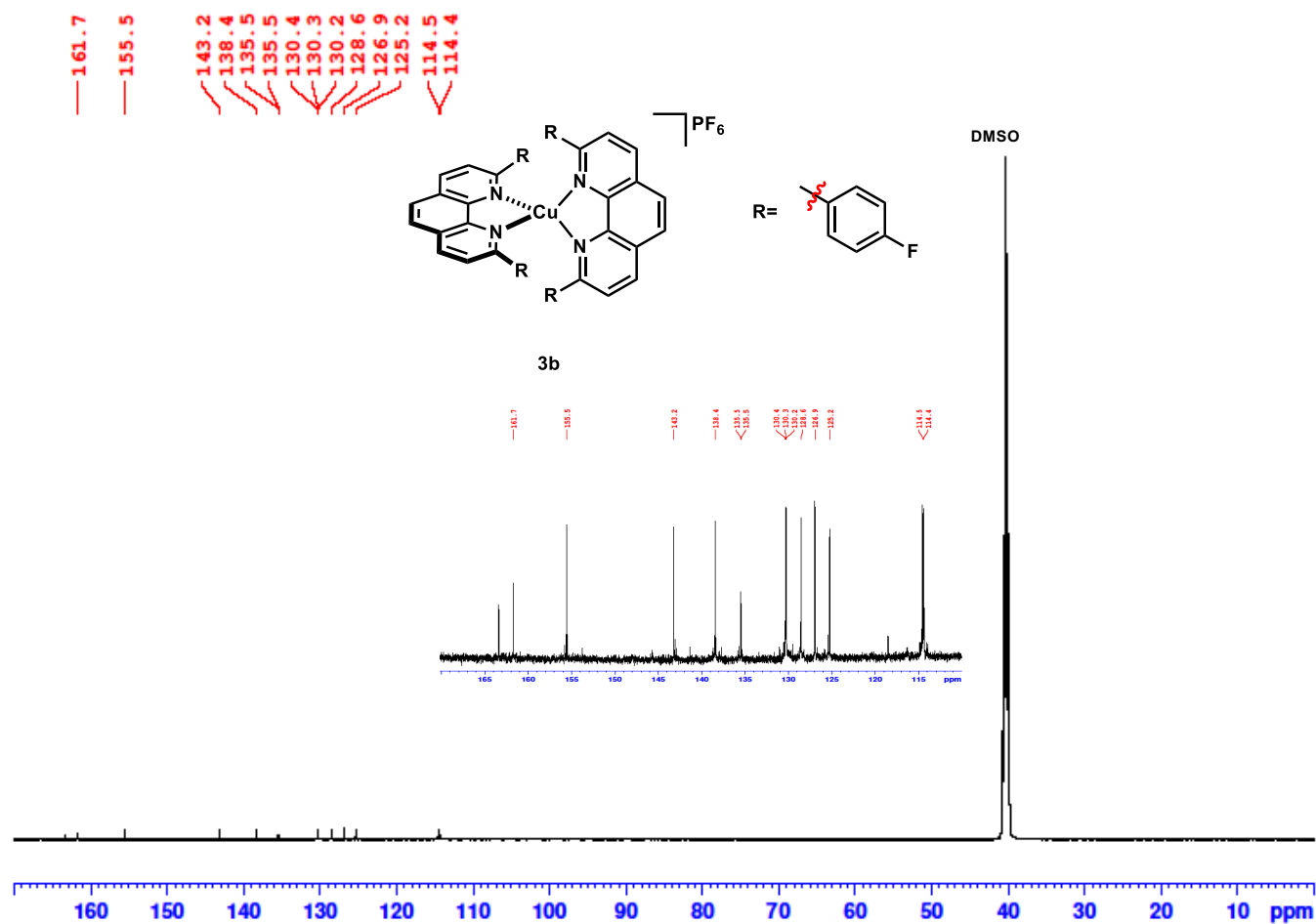
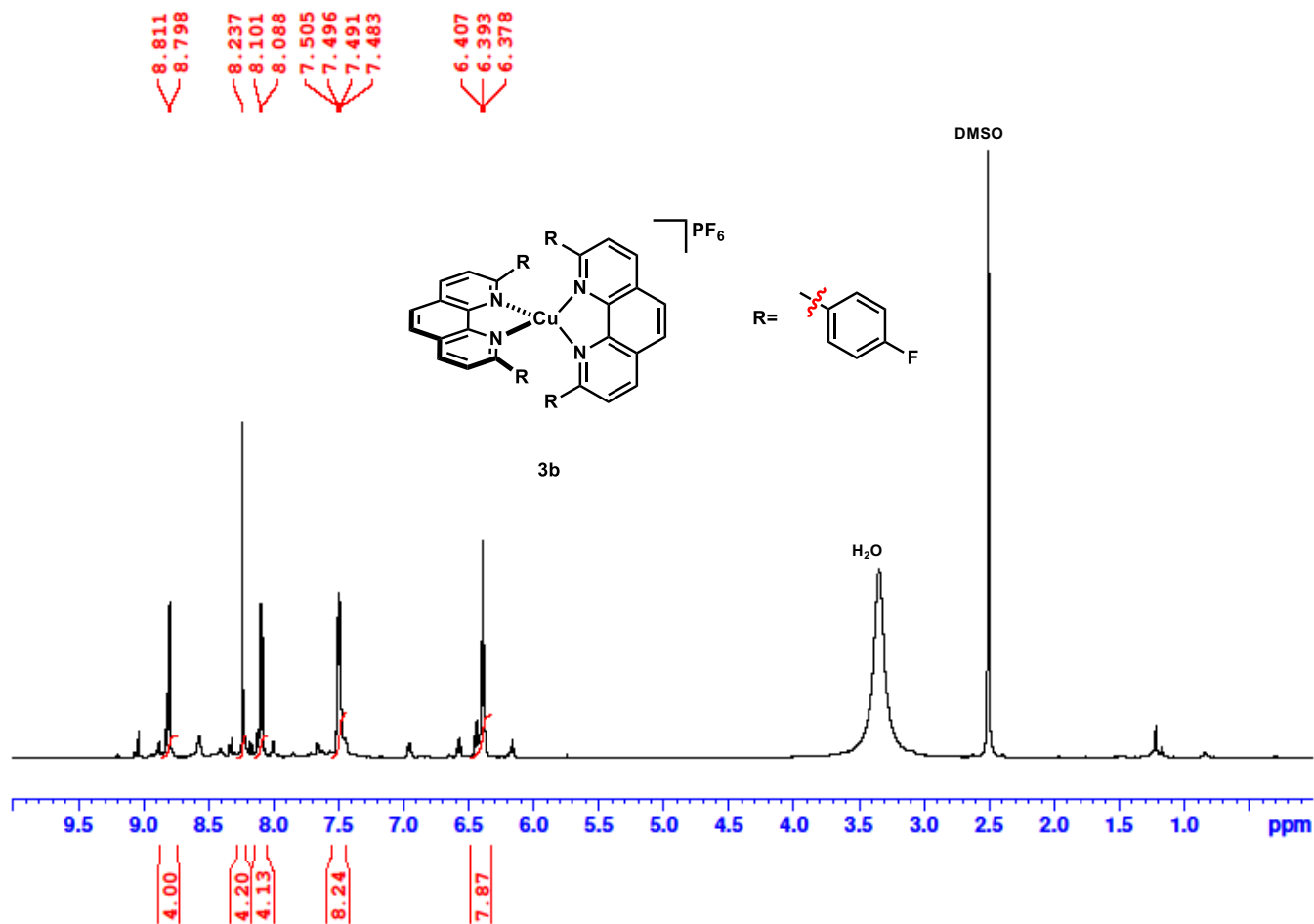
52H28N4F12Cu: C52 H28 N4 F12 Cu1 pa Chrg

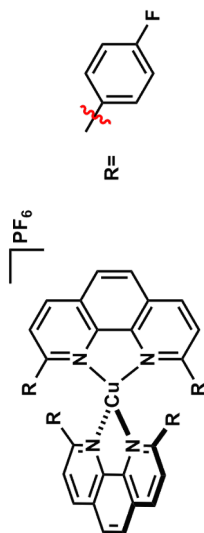
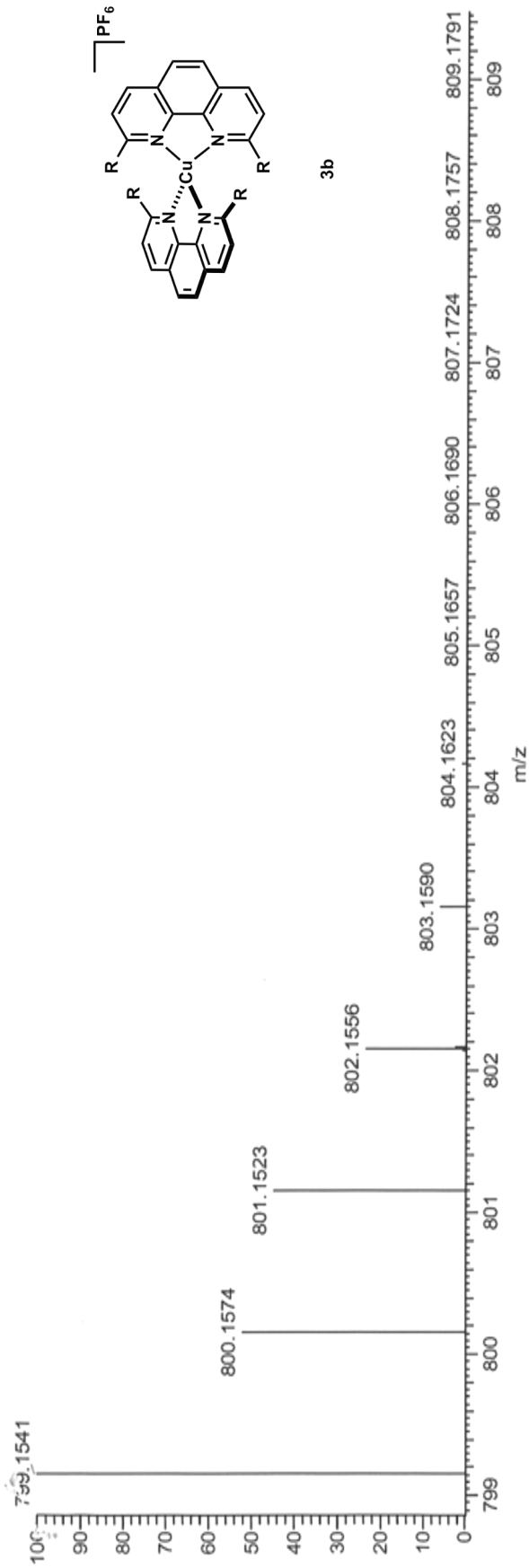


3a

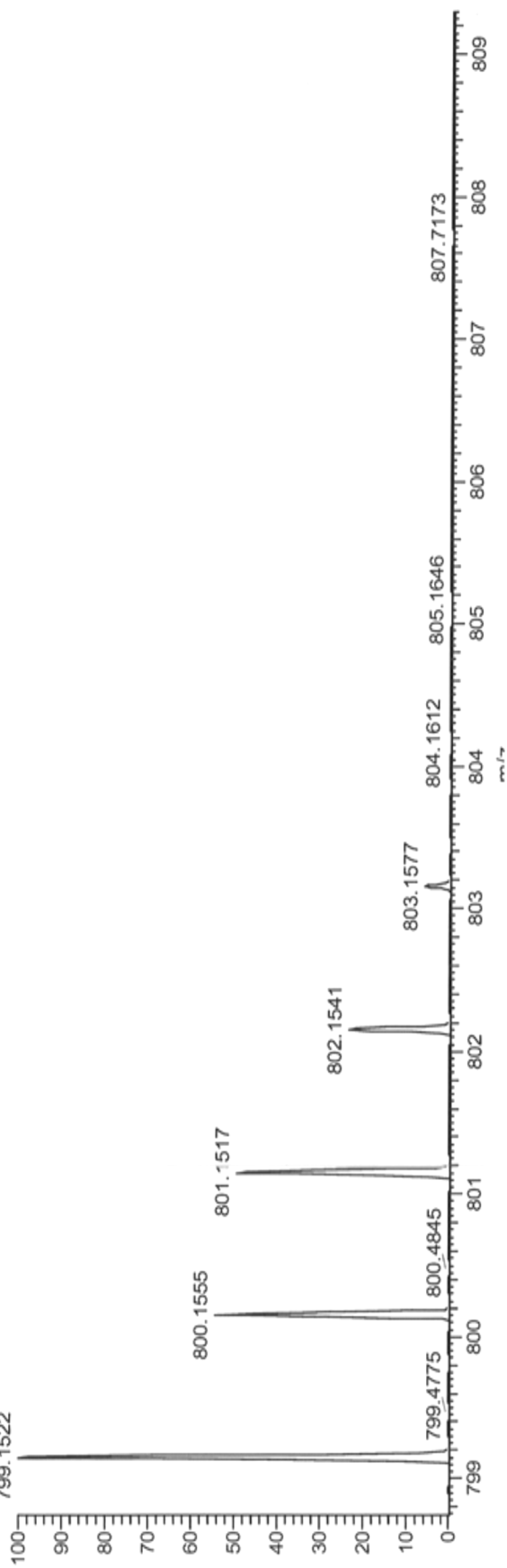
PN_6_2 #1-20 RT: 0.00-0.15 AV: 20 NL: 2.79E7
; FTMS + p ESI Full ms [50.00-2000.00]

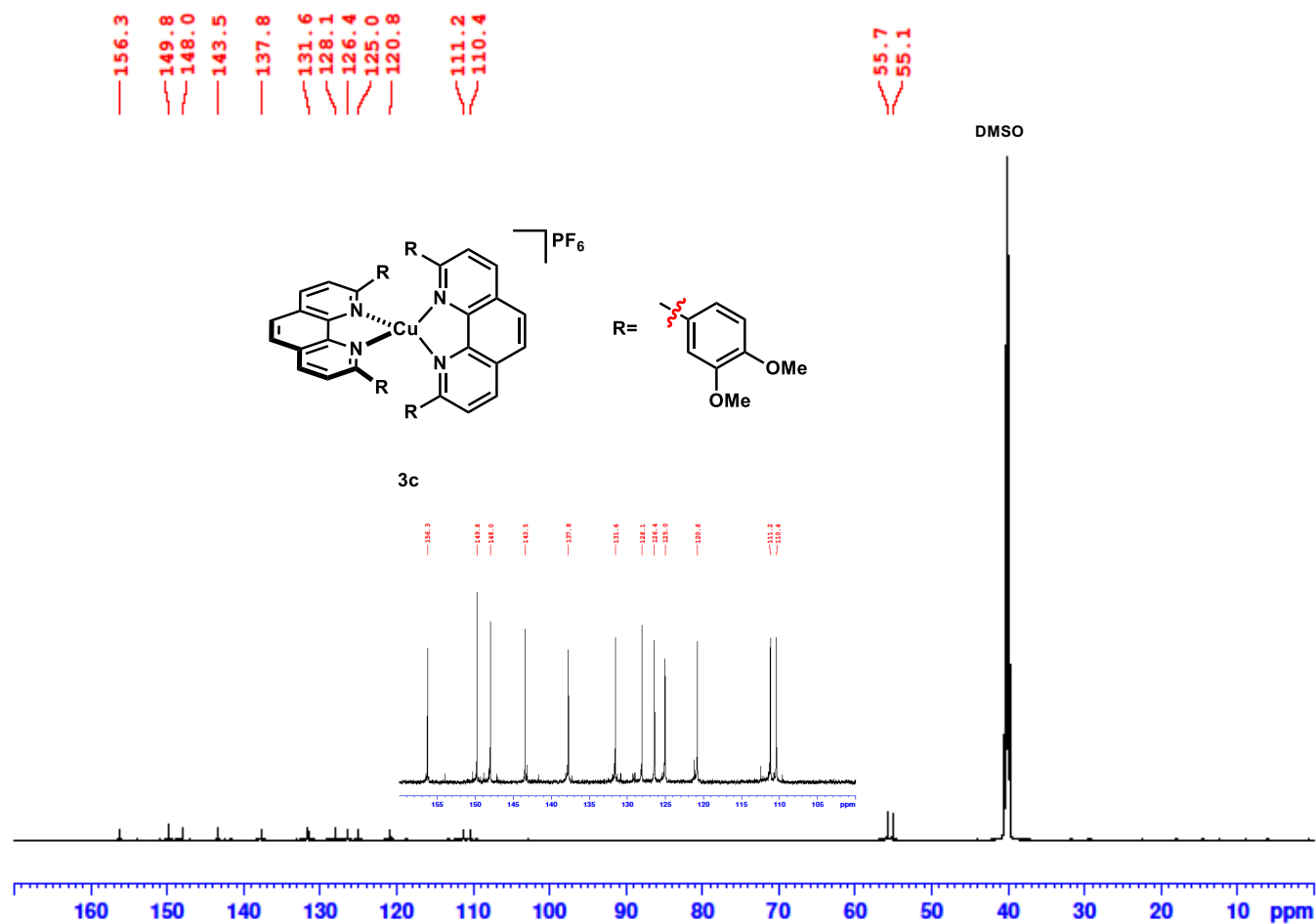
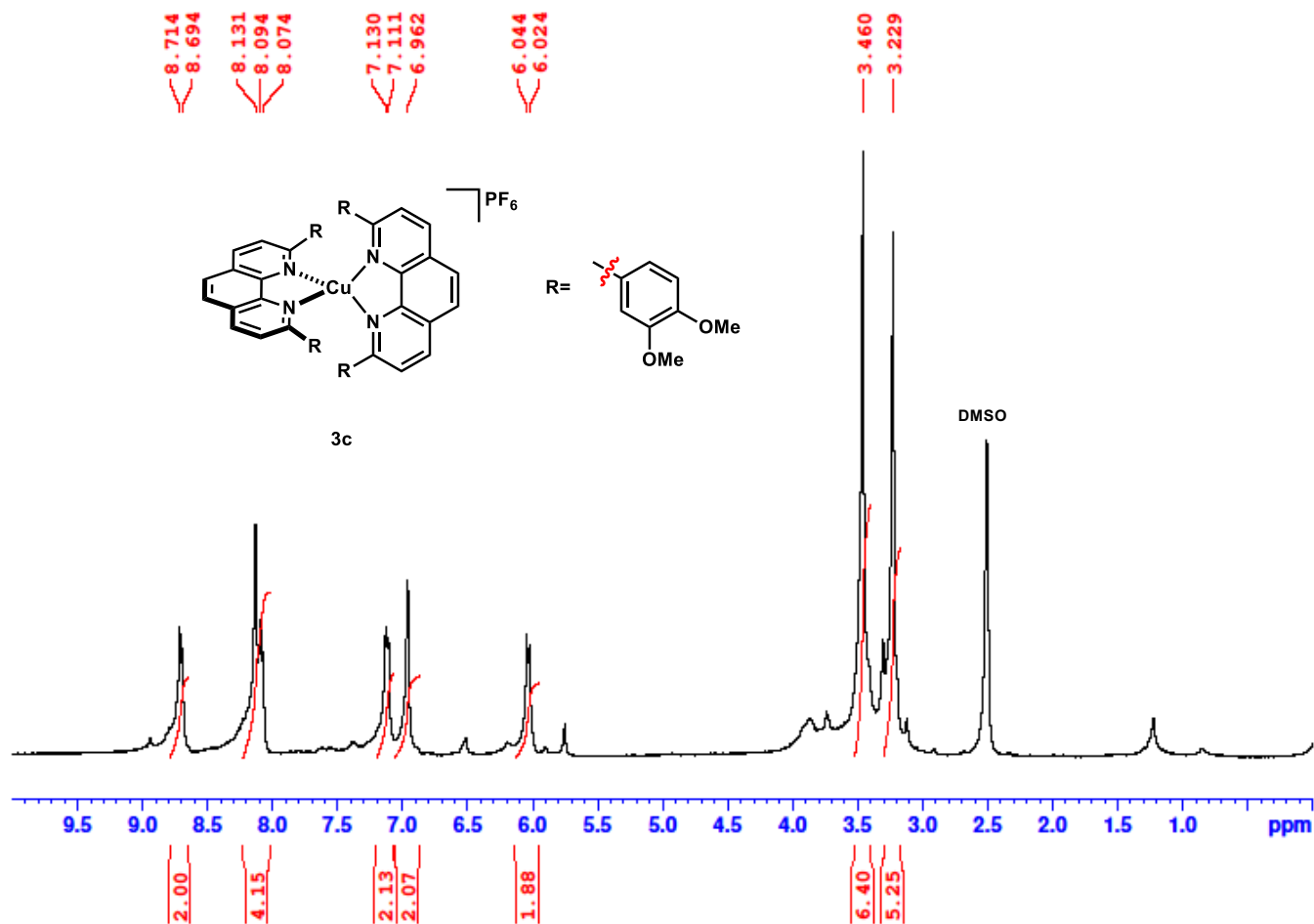




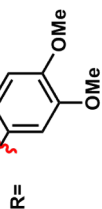
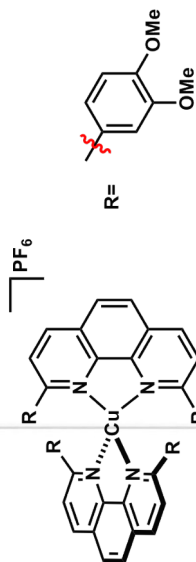
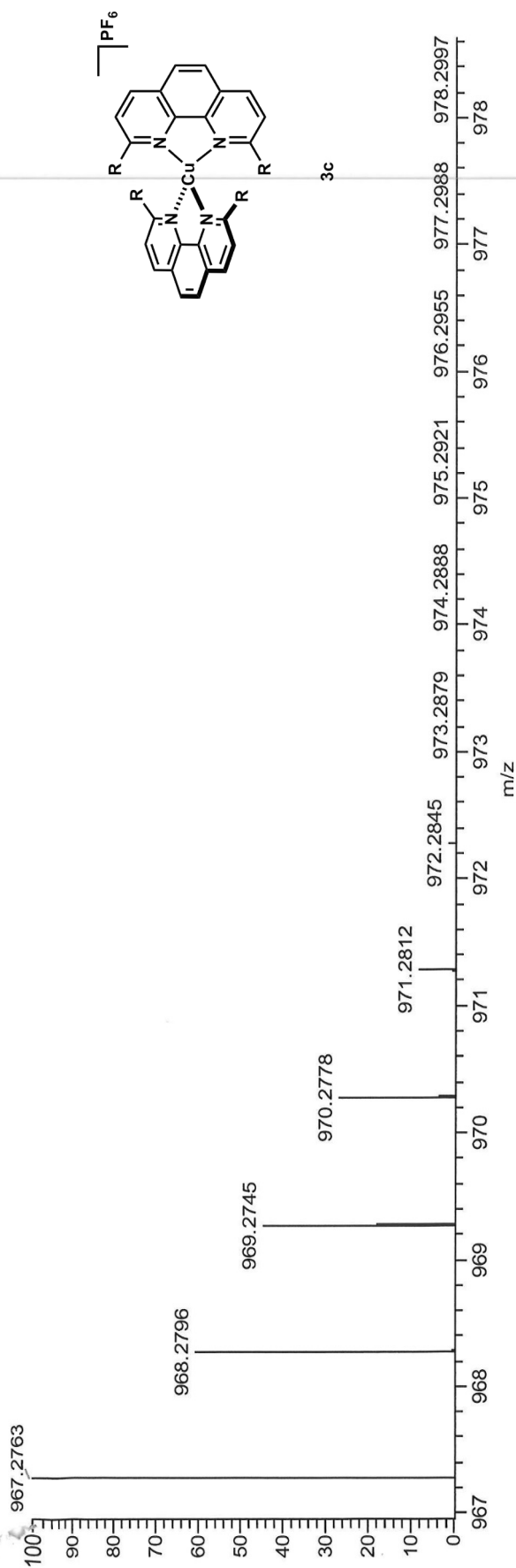


3b



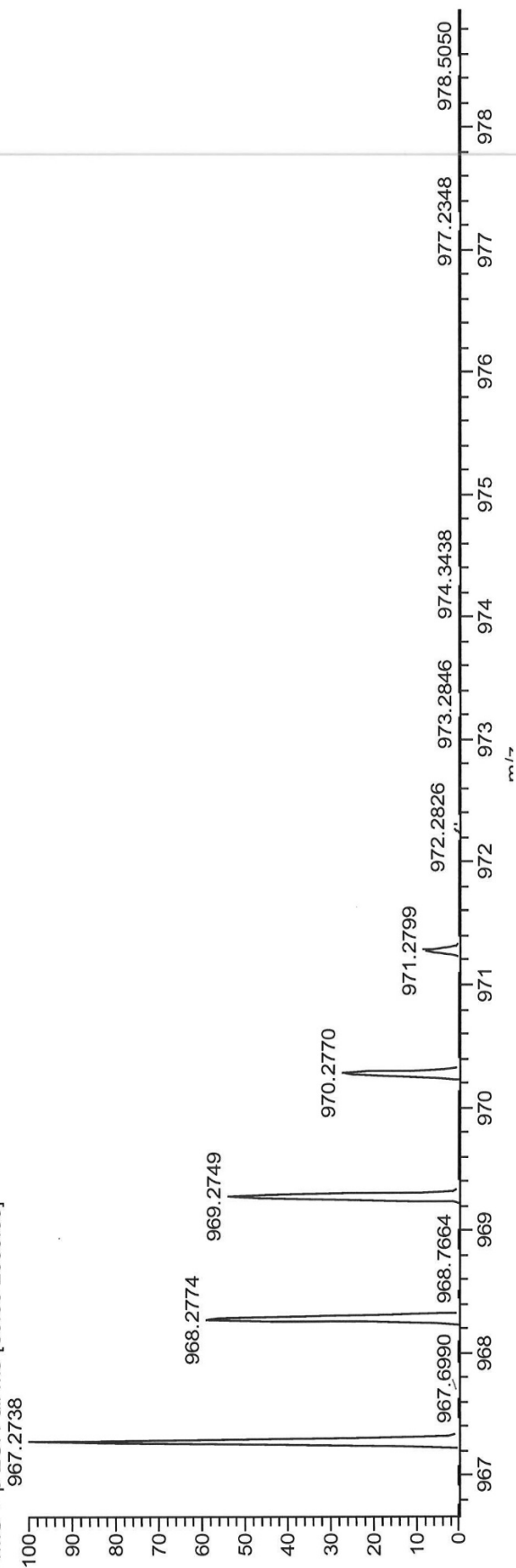


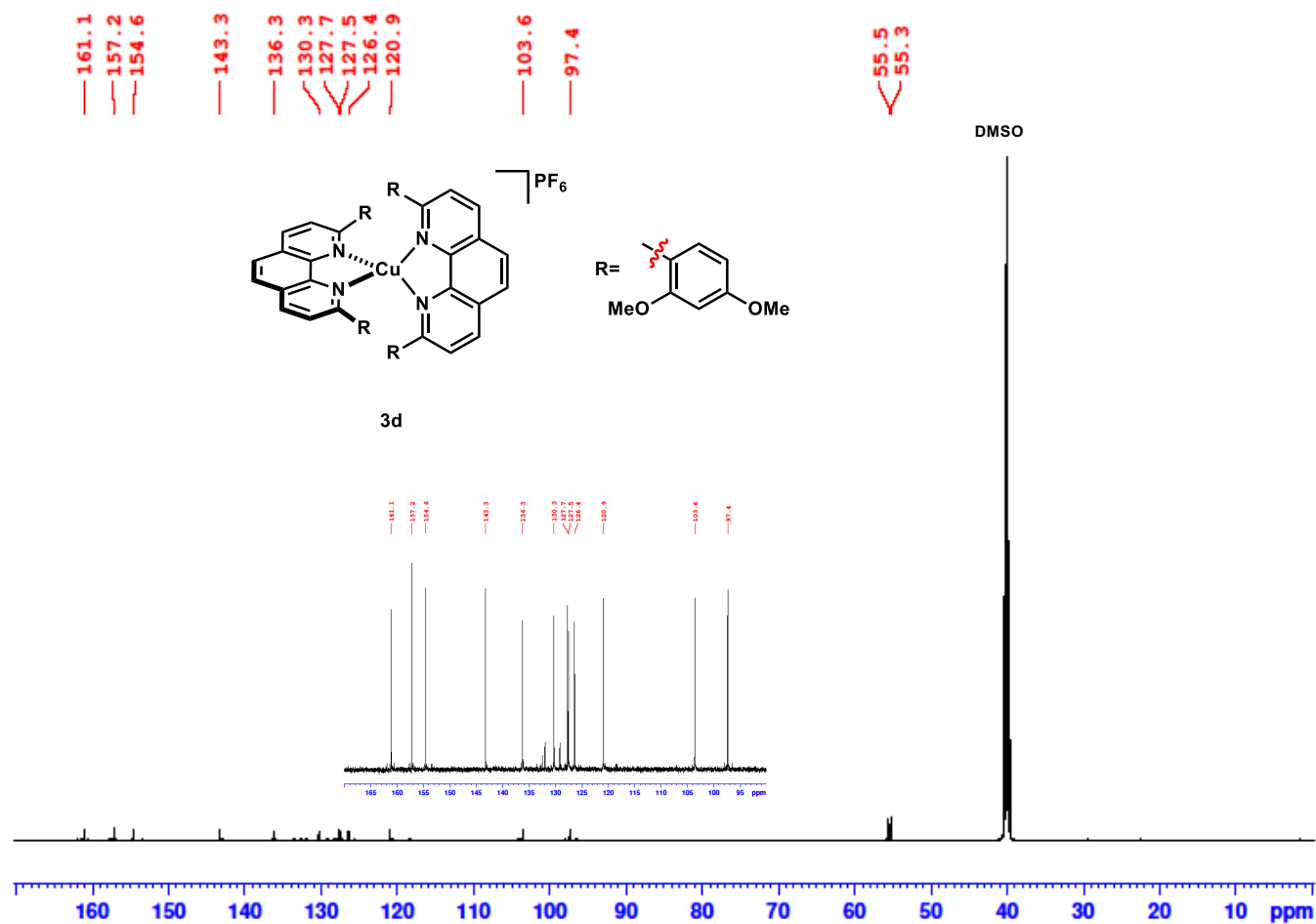
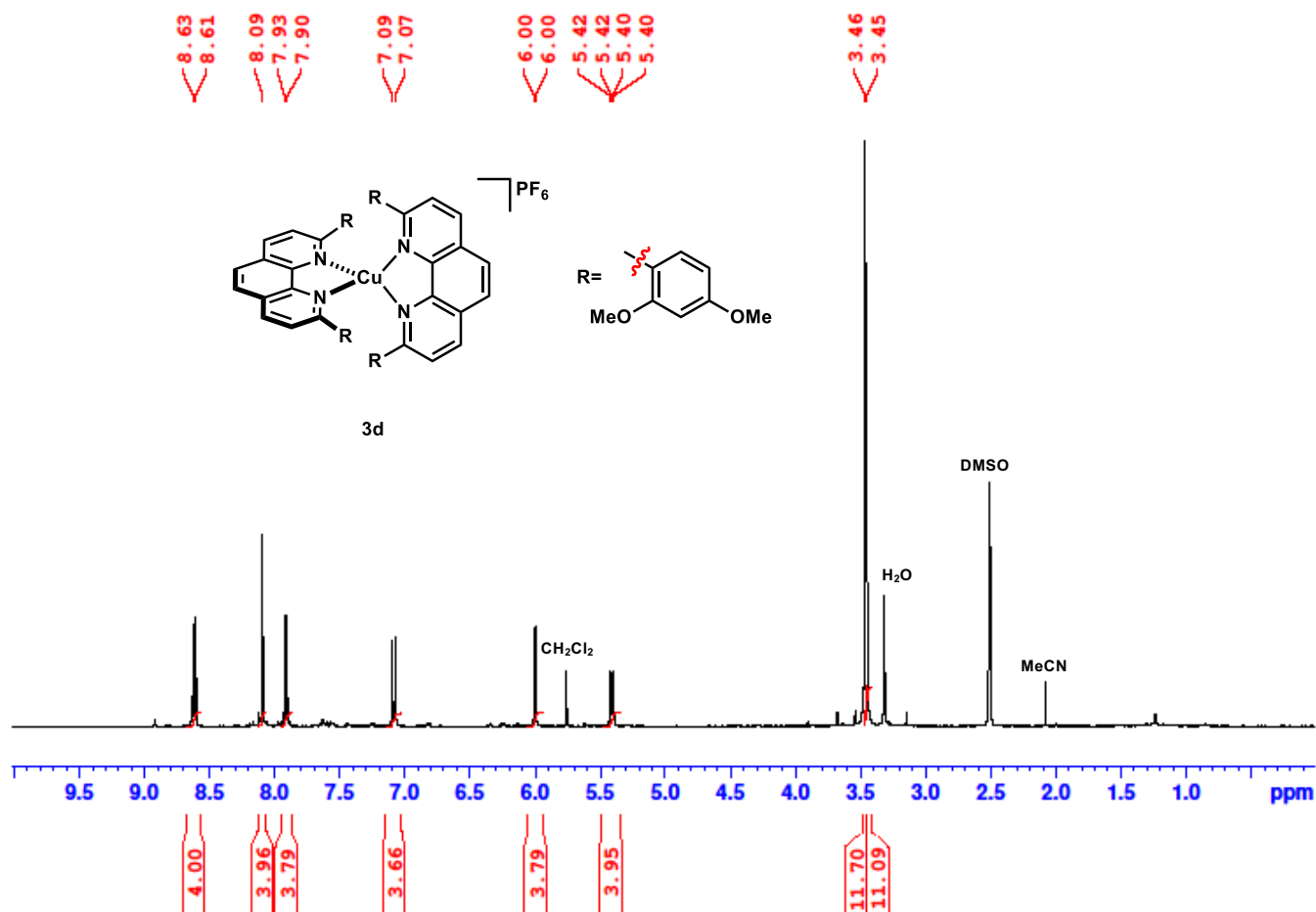
5. 148N4O8Cu: C56 H48 N4 O8 Cu1 pa Chrg 1



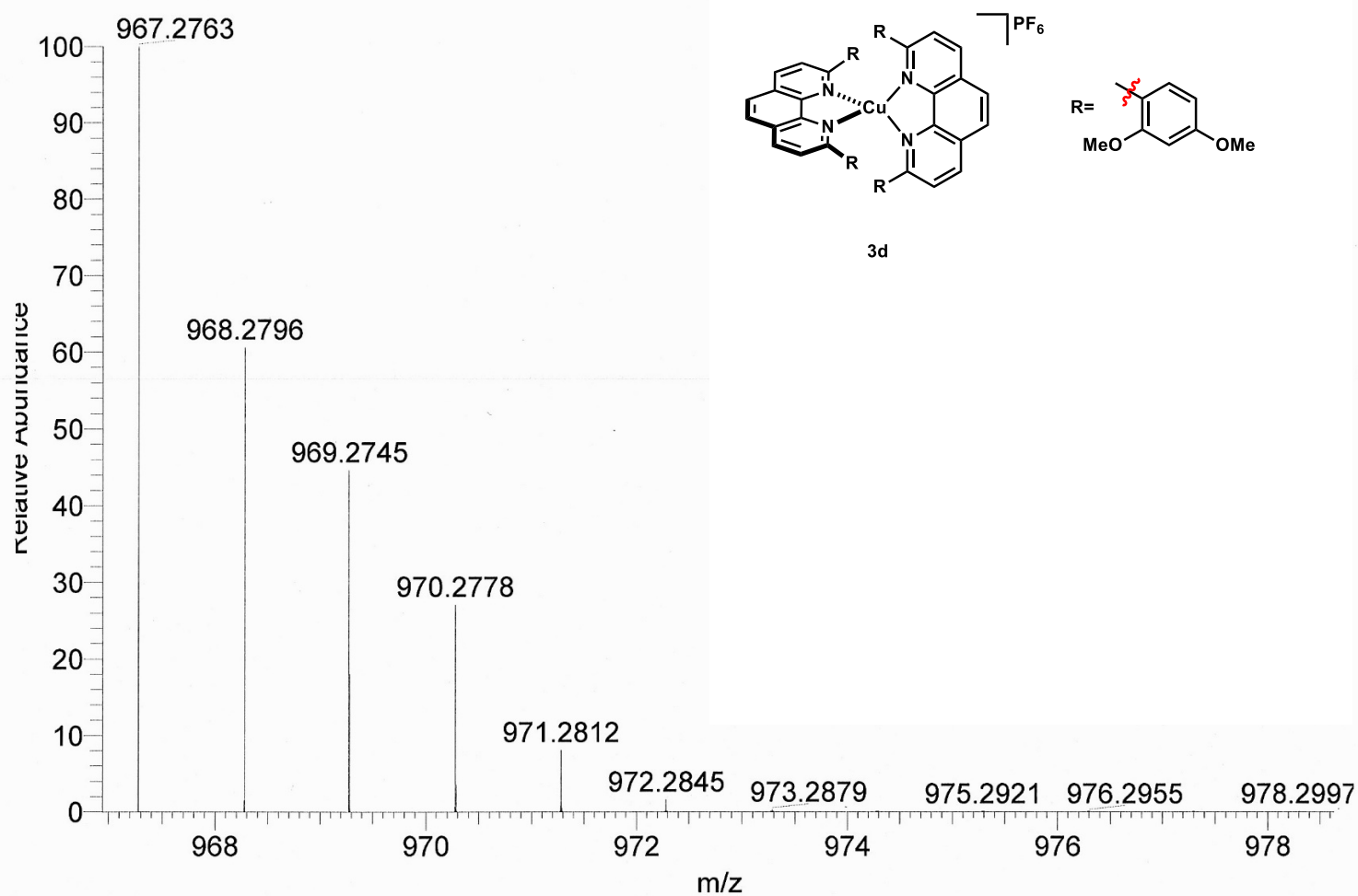
S 42

PN_5_2 #2-20 RT: 0.01-0.15 AV: 19 NL: 3.06E7
: FTMS + p ESI Full ms [50.00-2000.00]



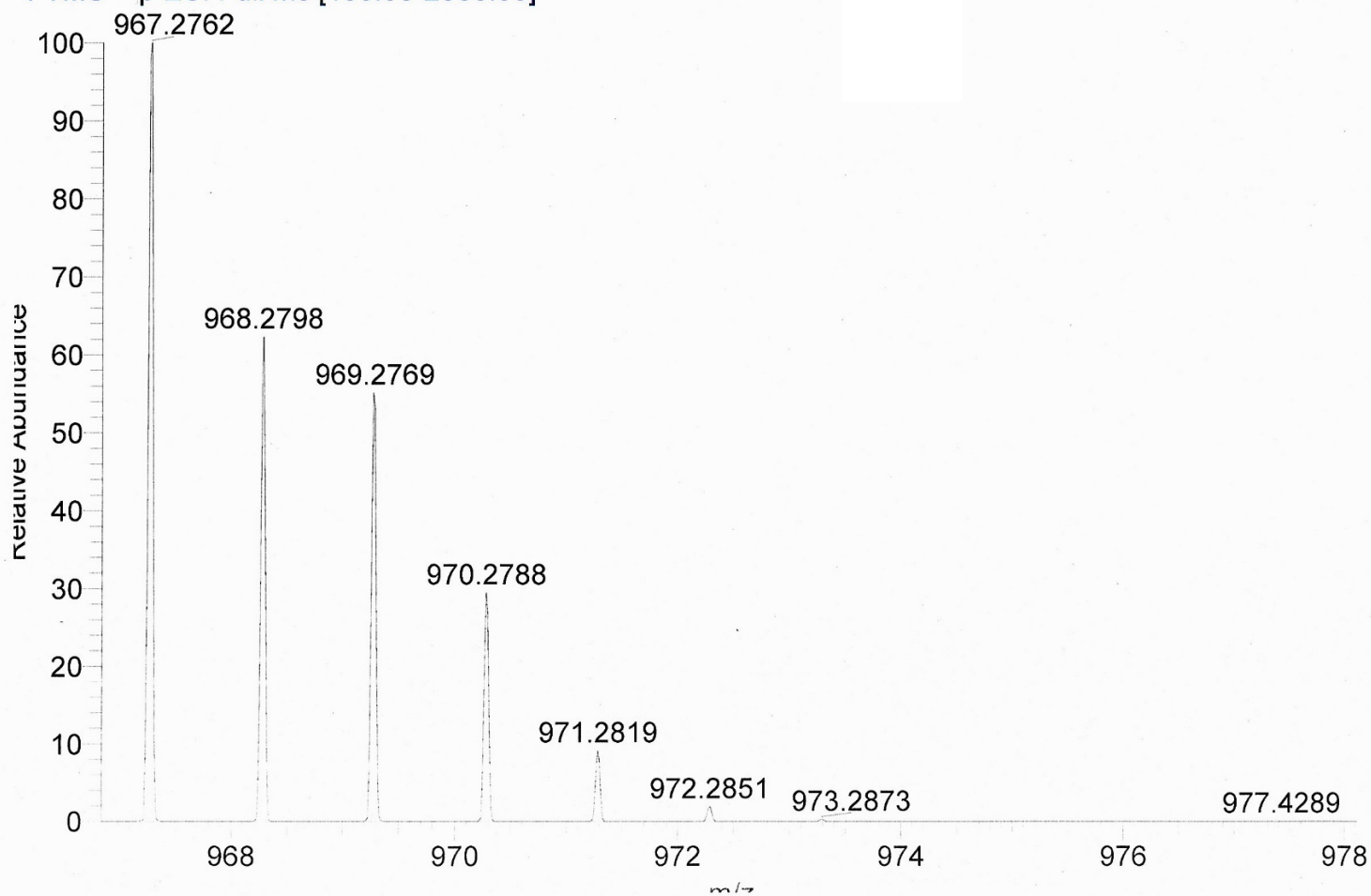


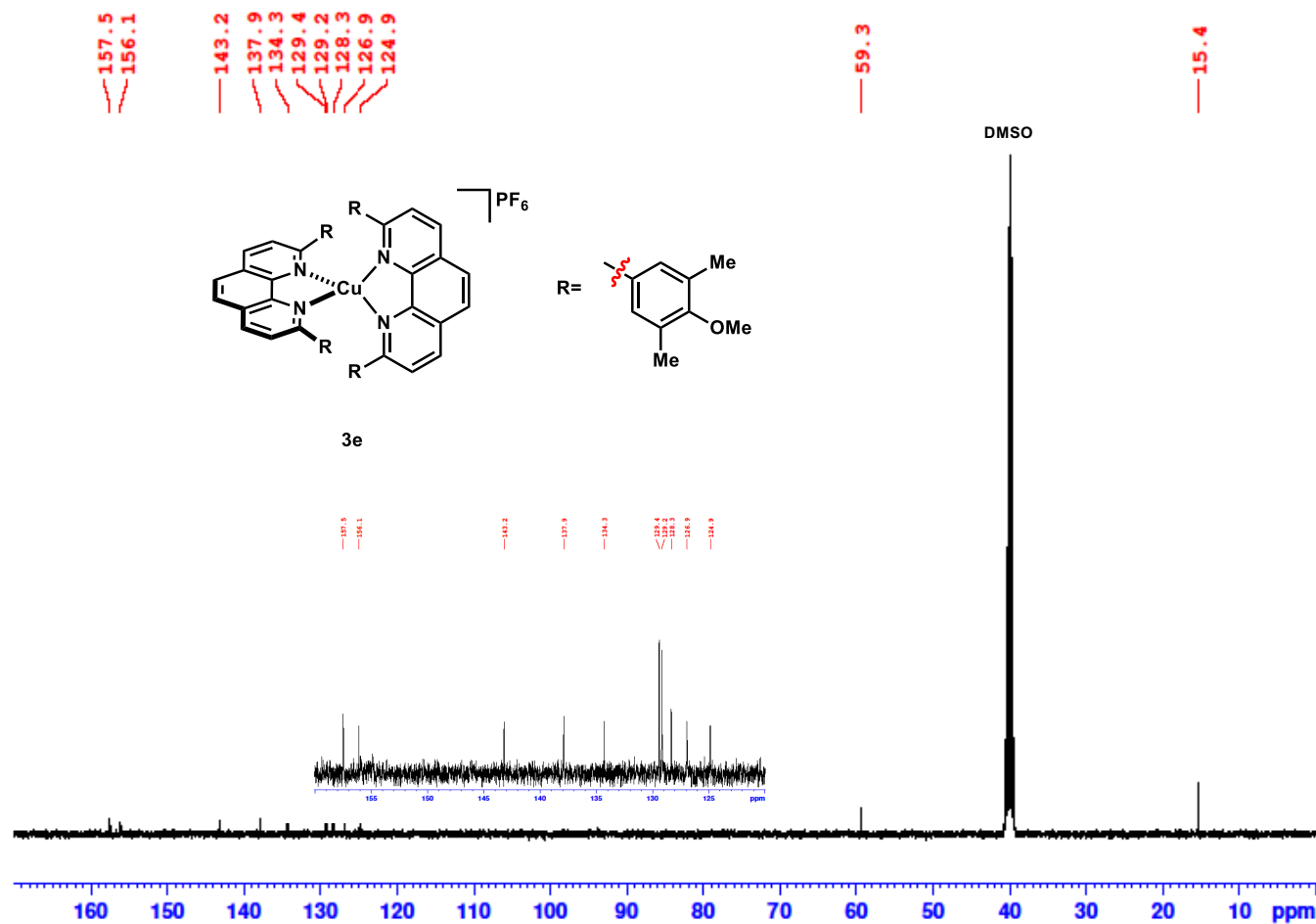
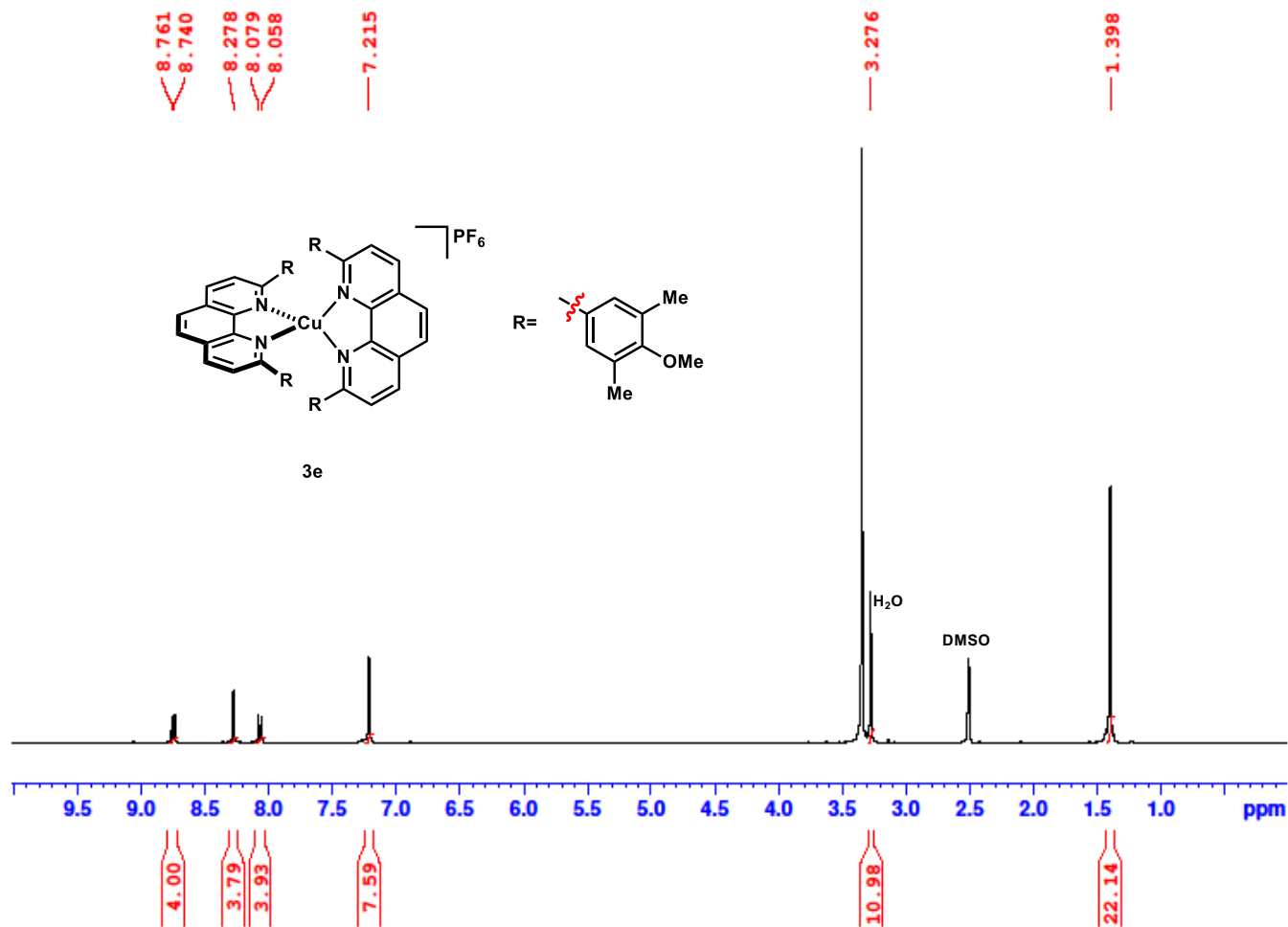
uC56H47N4O8 +H: Cu1 C56 H48 N4 O8 pa Chrg 1



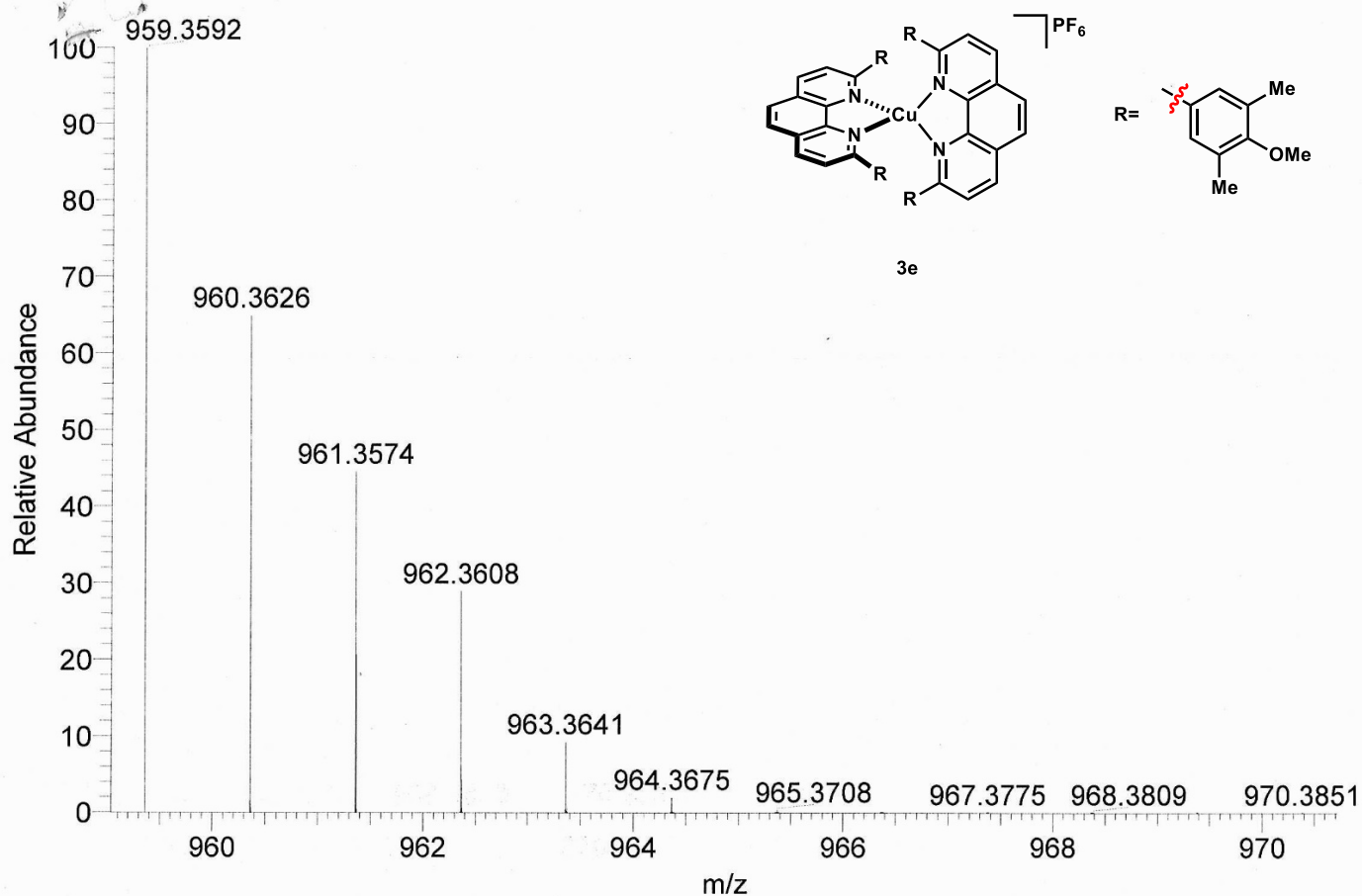
V-1-93 #3-27 RT: 0.05-0.62 AV: 25 NL: 1.14E7

FTMS + p ESI Full ms [150.00-2000.00]

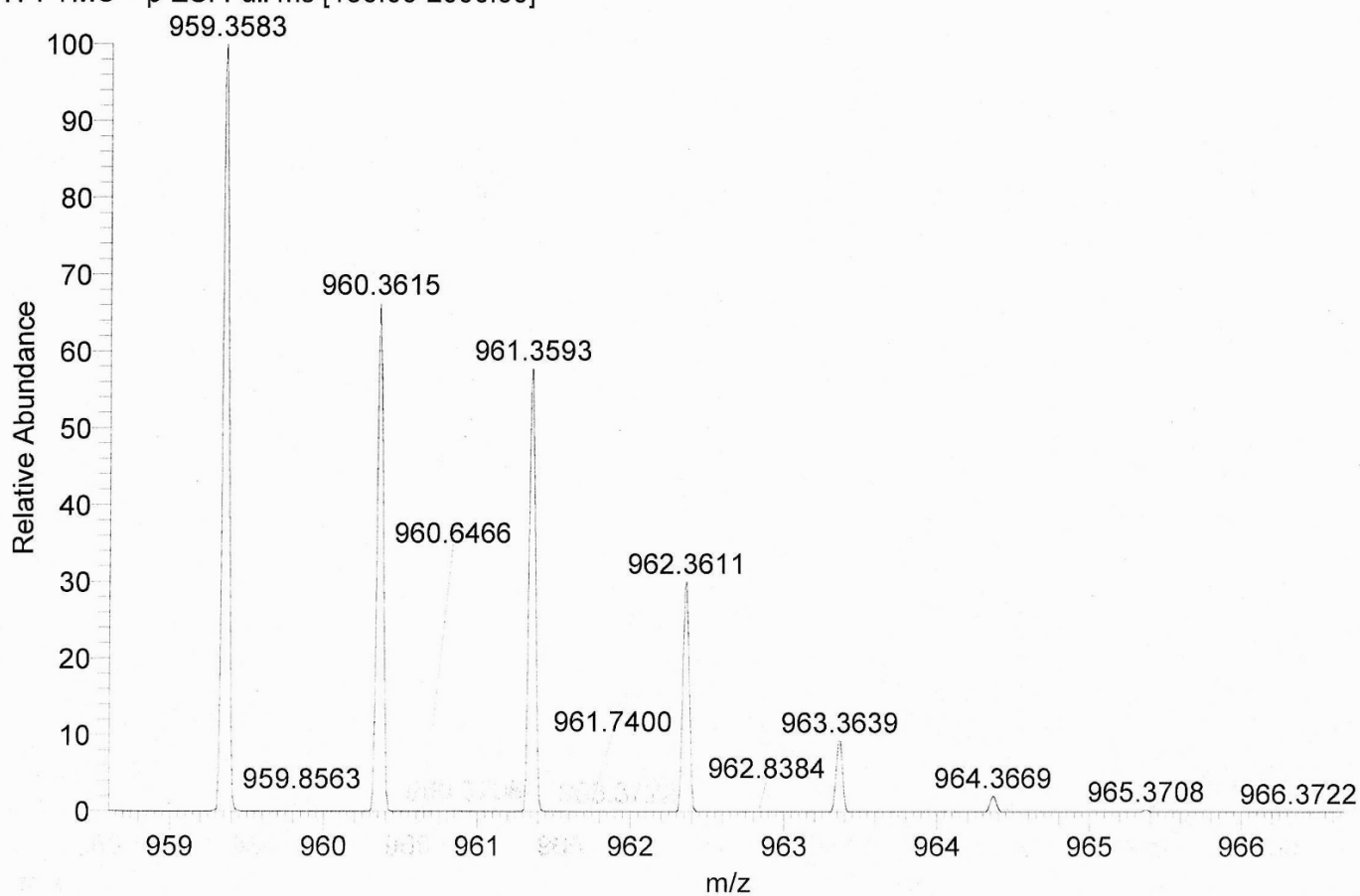


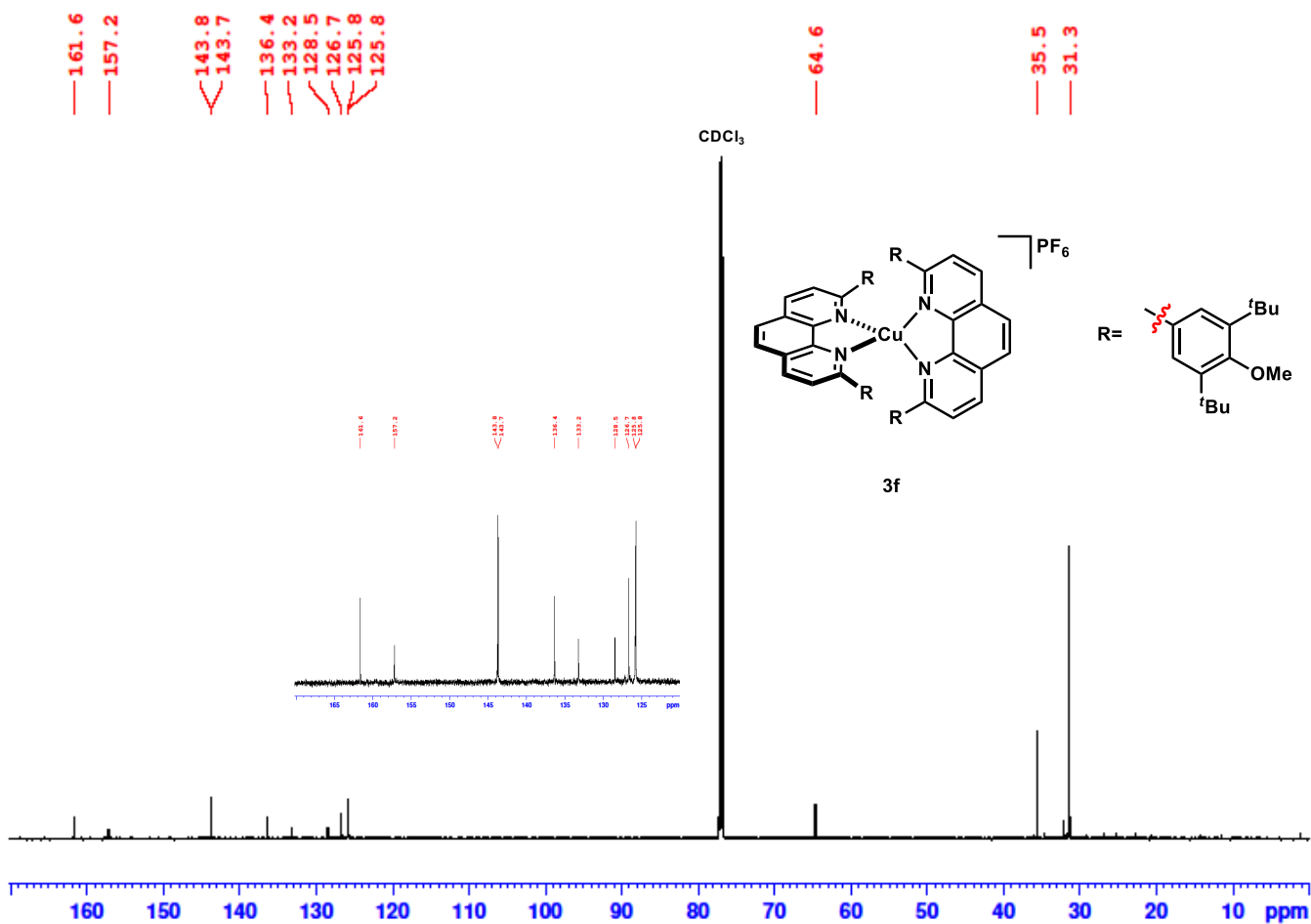
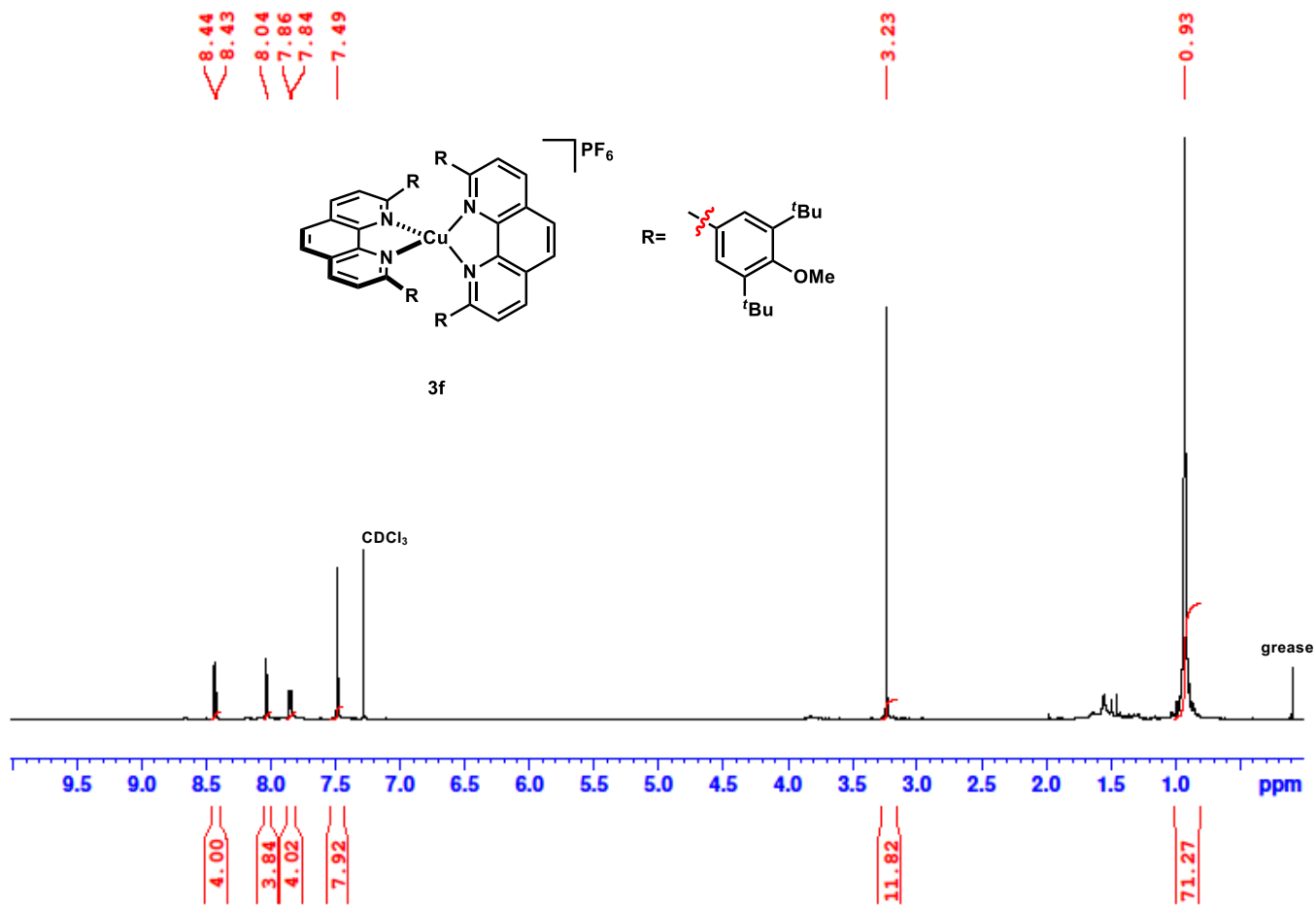


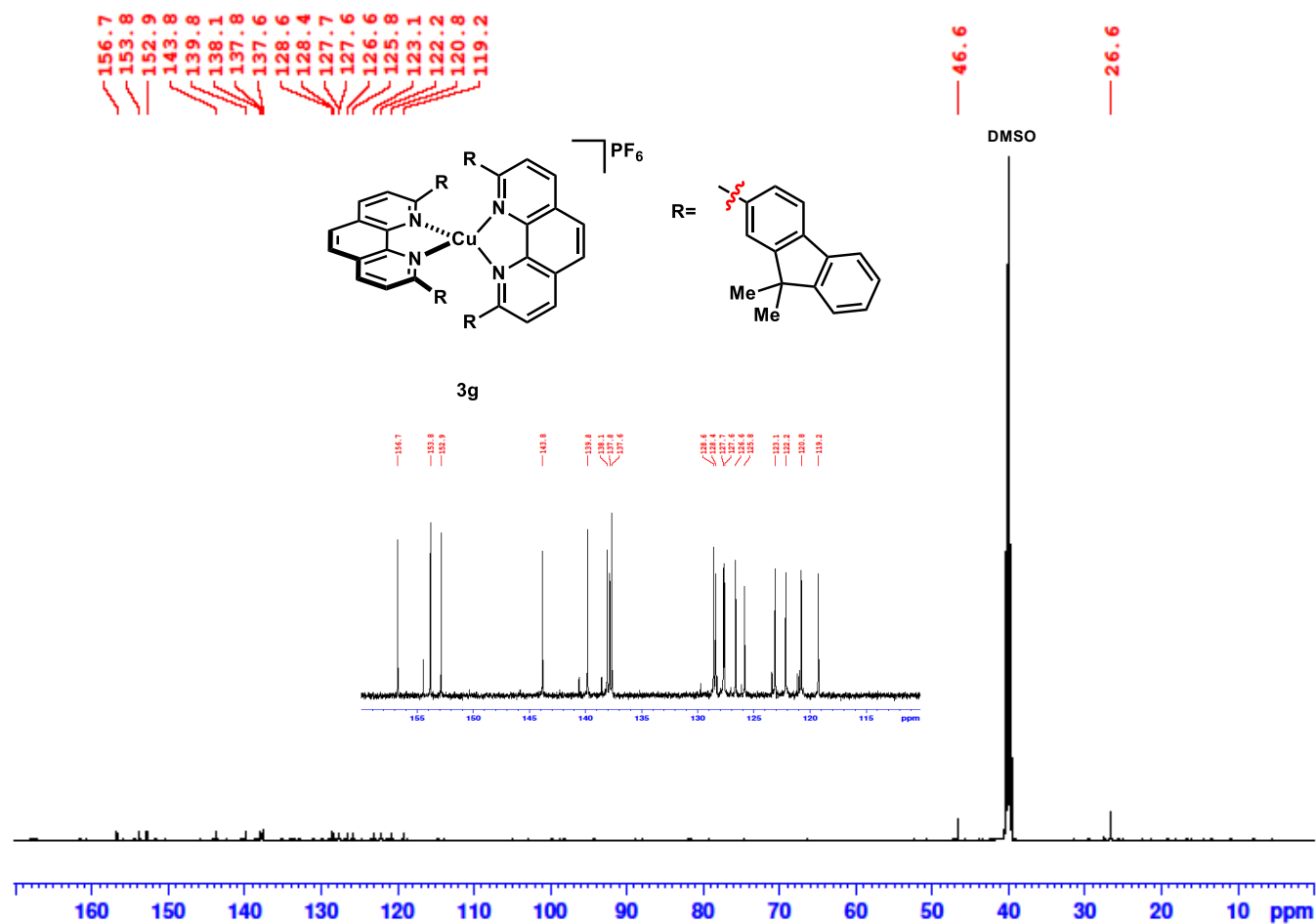
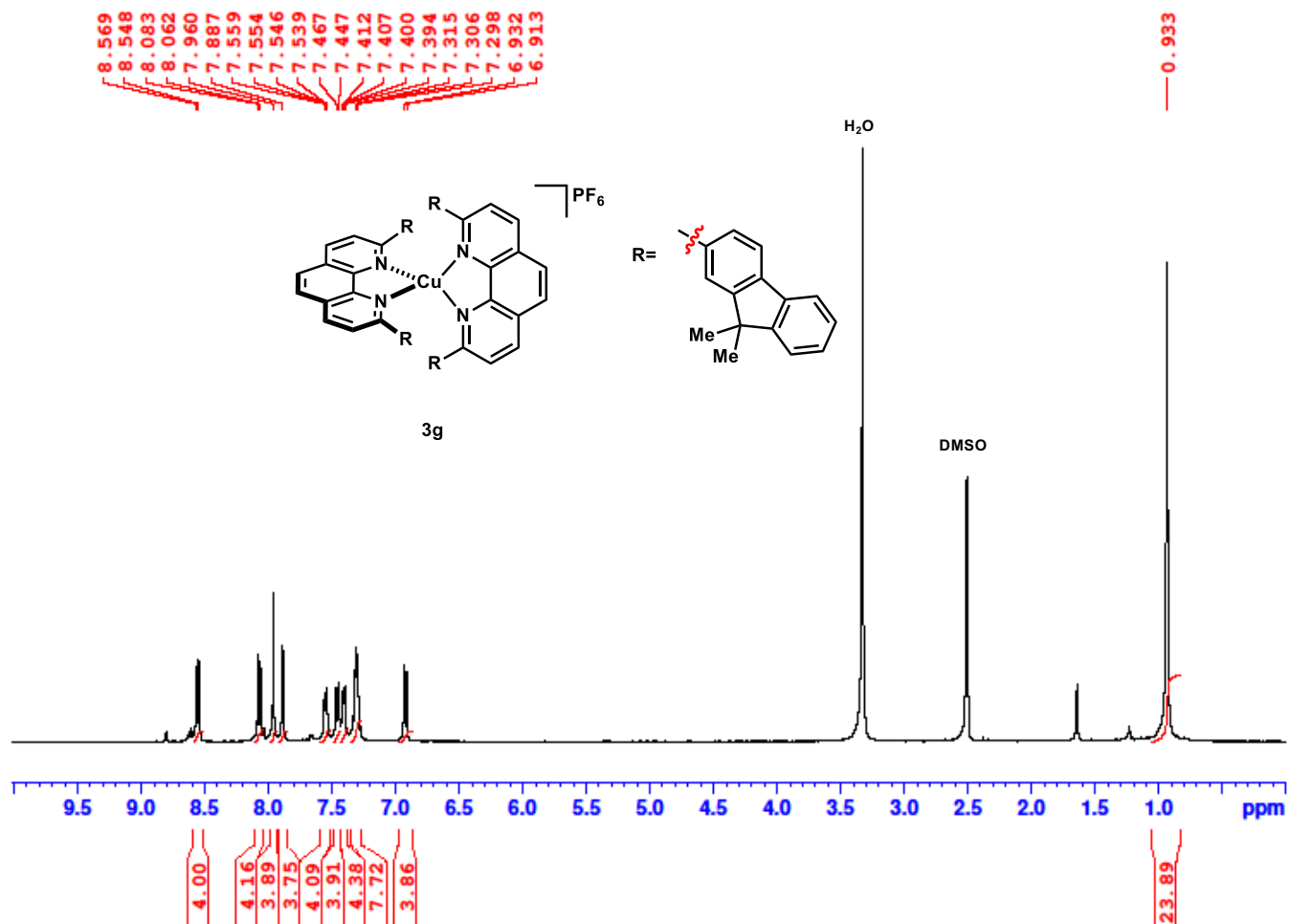
CuC60H56N4O4: Cu1 C60 H56 N4 O4 pa Chrg 1

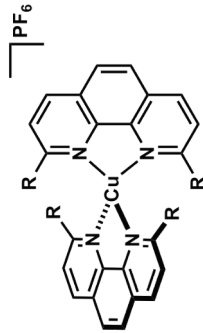
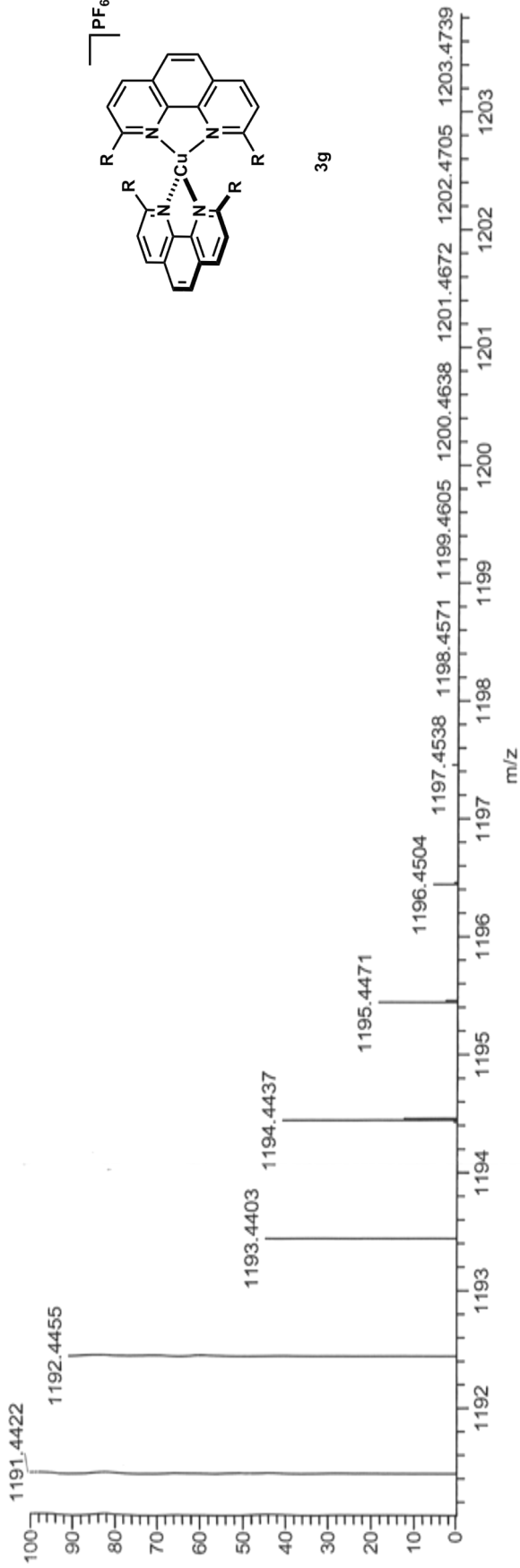


tpn-99 #1-37 RT: 0.01-0.29 AV: 37 NL: 2.48E7
T: FTMS + p ESI Full ms [150.00-2000.00]

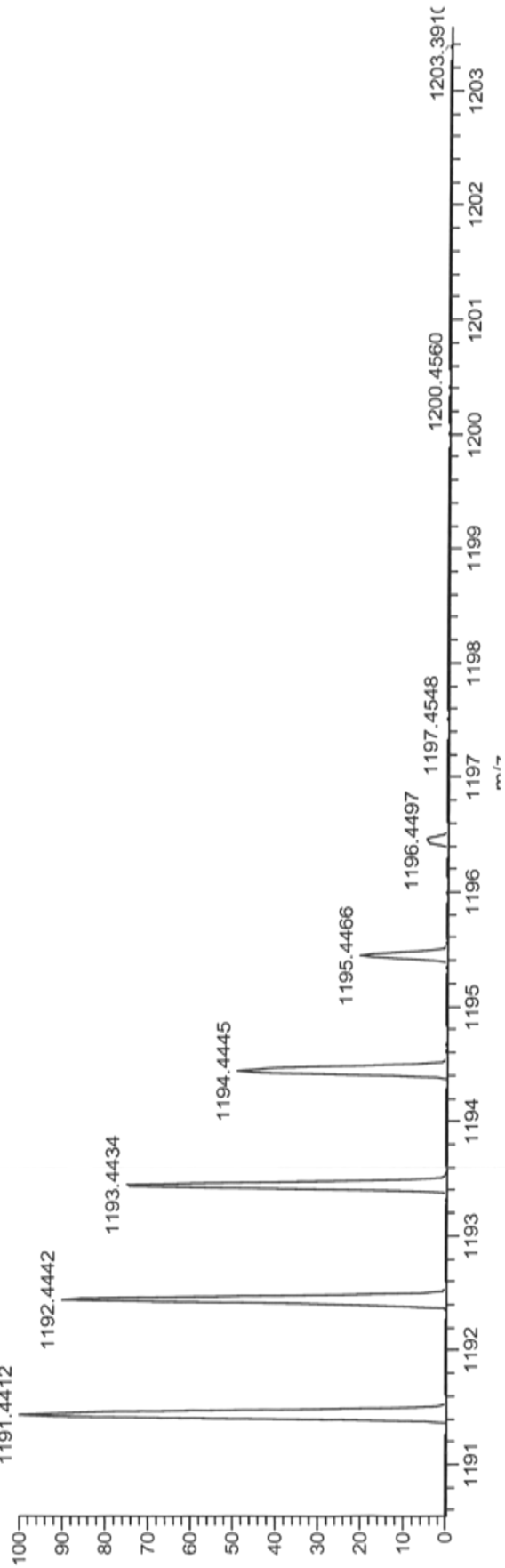


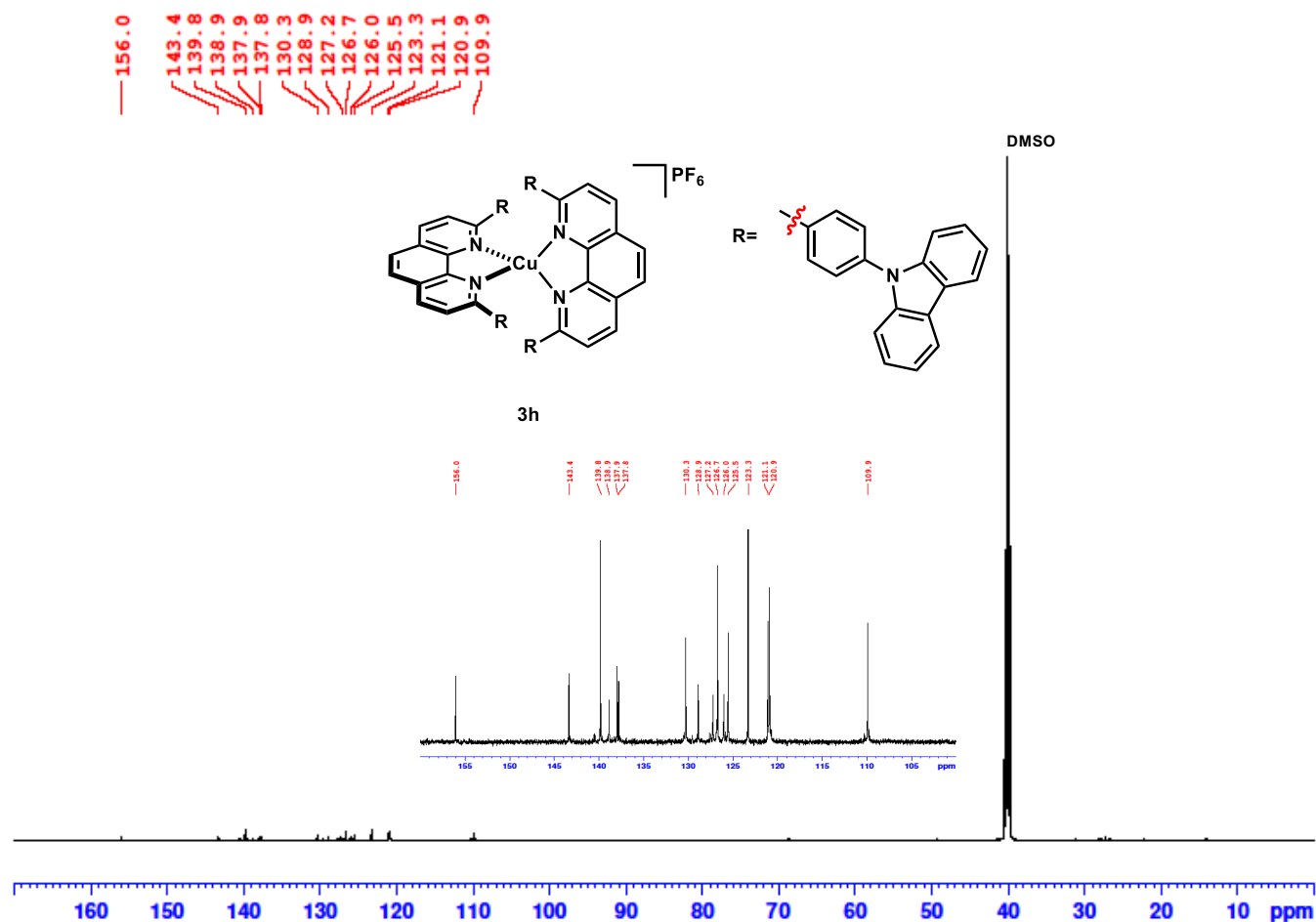
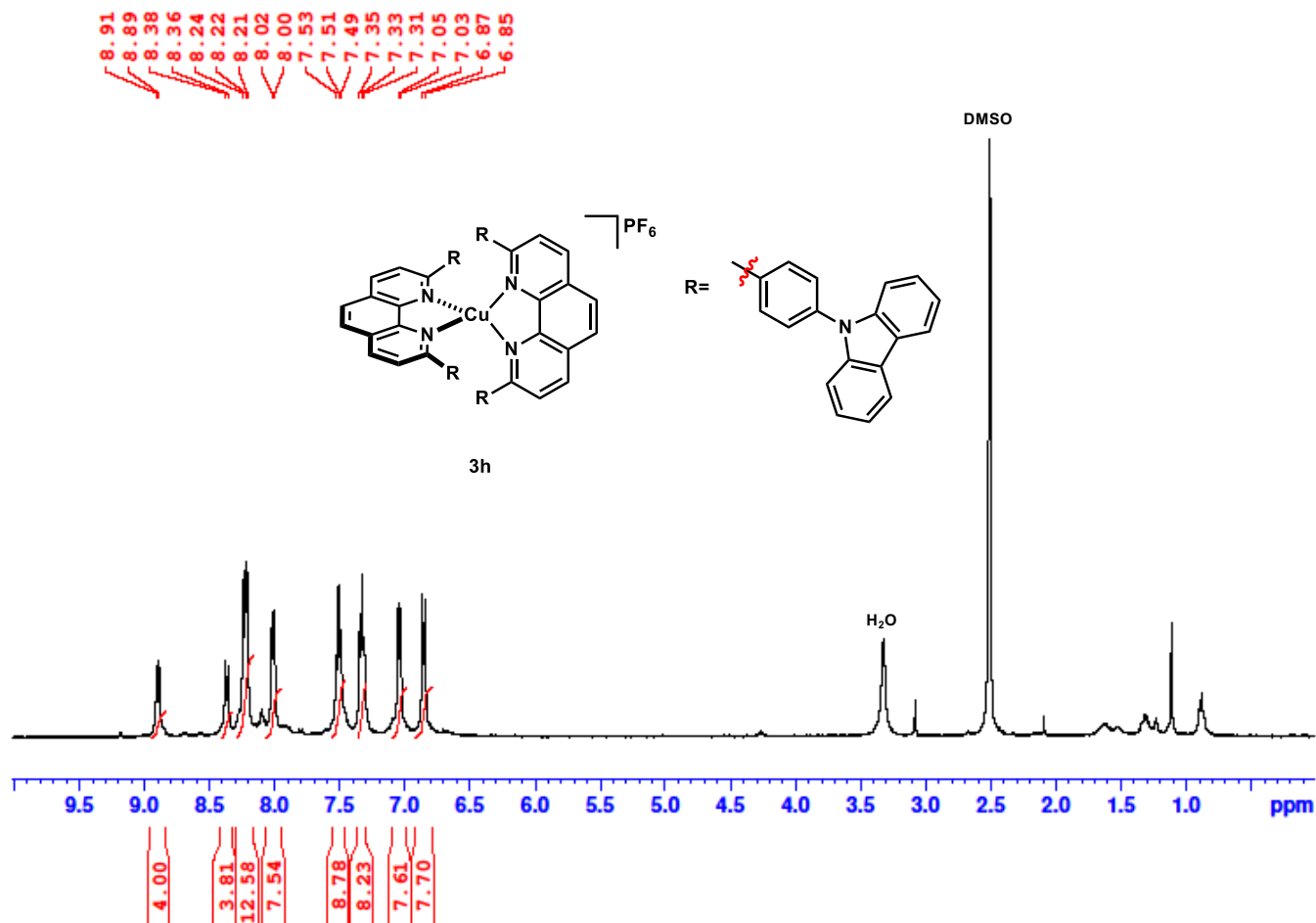




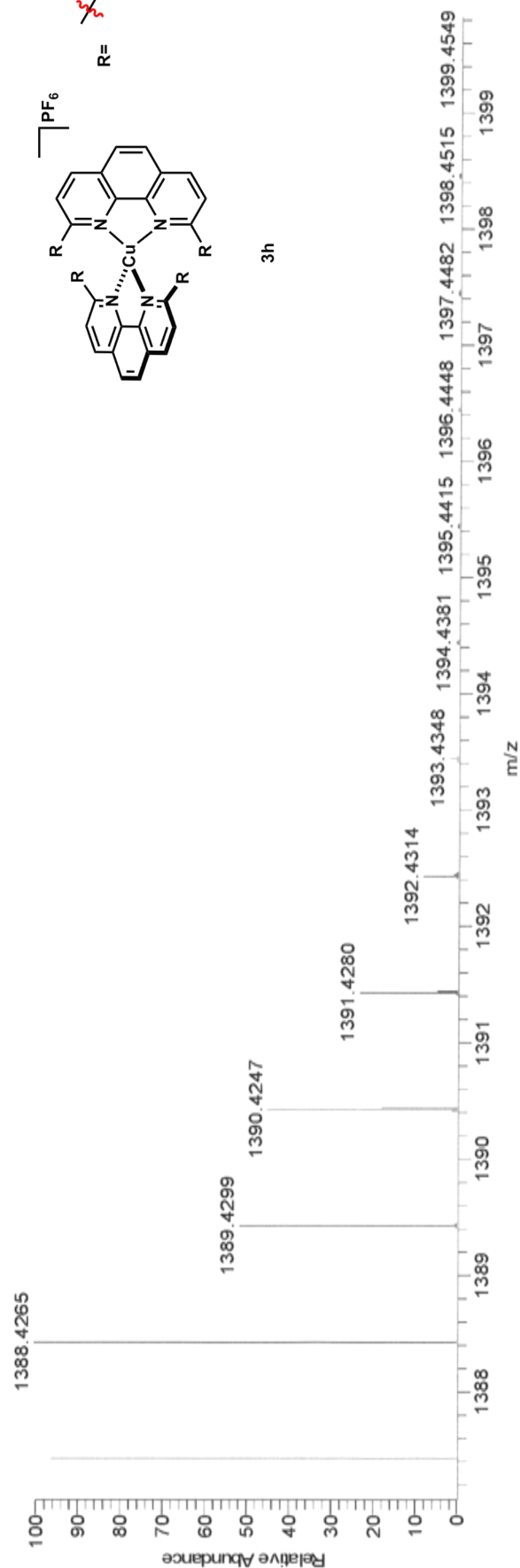


3g





C96H60N8Cu: C96 H60 N8 Cu1 pa Chrg 1

TN_04 #2-28 RT: 0.02-0.40 AV: 27 NL: 9.06E4
T: FTMS + p ESI Full ms [150.00-2000.00]