Supporting Information

Selective Hydrosiloxane Synthesis via Dehydrogenative Coupling of

Silanols with Hydrosilanes Catalyzed by Fe Complexes Bearing a

Tetradentate PNNP Ligand

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Compound characterization data

[FeCl₂(PNNP-Ph)] (1a)

¹H NMR:



$[Fe(PNNP-Ph)]_2(\mu-N_2)$ (2a)



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{}^{31}P{}^{1}H} NMR:
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 $[Fe(PNNP-Cy)]_2(\mu-N_2)$ (2b)

¹H NMR:



[Fe(H)₂(PNNP-Cy)] (3b)



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^{13}C{^{1}H} NMR:
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[Fe(PNNP-Ph)(CO)] (4a)





[Fe(η^2 -HSiH₂Ph)(PNNP-Ph)] (5a)





Compounds characterization data of hydrosiloxanes

The final product was characterized by ¹H, ¹³C{¹H} and ²⁹Si{¹H} NMR as well as HRMS. (Me₃SiO)(SiH₂Ph),¹ (Me₃SiO)₂(SiHPh),^{1,2} ((*t*BuO)₃SiO)(SiH₂Ph),¹ (Me₃SiO)(SiHPh₂)³ and (Me₃SiO)(SiHPhMe)² were identified by comparing their ¹H, ¹³C{¹H} and ²⁹Si{¹H} NMR data with those previously reported. 4-Methylphenylsilane⁴ was synthesized by following the reported procedures.

(*i***Pr**₃**SiO**)₂(**SiHPh**) Using PhSiH₃ (32.4 mg, 0.30 mmol) and *i***Pr**₃SiOH (108.1 mg, 0.62 mmol). ¹H NMR (600 MHz, C₆D₆, 25 °C): δ 7.72 (d, 2H, *J* = 1.50 Hz, Ar-*H*), 7.21-7.18 (m, 3H, Ar-*H*), 5.39 (s, 1H, ¹*J*_{SiH}= 246.6 Hz), 0.98-0.89 (m, 42H) ppm. ¹³C{¹H} NMR (150 MHz, C₆D₆, 25 °C): δ 137.6, 132.8, 130.1, 128.1, 17.7, 12.8 ppm. ²⁹Si{¹H} NMR (119 MHz, C₆D₆, 25 °C): δ 9.5, -49.2 ppm.

HRMS (ESI): m/z calcd for $[C_{23}H_{48}O_2Si_3H]^+$ (M+H): 453.3040 ; found 453.3029.





(*t*BuMe₂SiO)₂(SiHPh) Using PhSiH₃ (32.4 mg, 0.30 mmol) and *t*BuMe₂SiOH (108.1 mg, 0.62 mmol). ¹H NMR (600 MHz, C₆D₆, 25 °C): δ 7.73 (dd, 2H, *J* = 7.74, 1.62 Hz, Ar-*H*), 7.23-7.19 (m, 3H, Ar-*H*), 5.39 (s, 1H, ¹*J*_{Si, H} = 243.8 Hz, Si*H*), 0.96 (s, 18H), 0.14 (s, 6H), 0.12 (s, 6H) ppm. ¹³C{¹H} NMR (150 MHz, C₆D₆, 25 °C): δ 137.4, 133.4, 130.5, 128.2, 25.9, 18.4, -2.8 ppm. ²⁹Si{¹H} NMR (119 MHz, C₆D₆, 25 °C): δ 13.2, -48.5 ppm.

HRMS (ESI): m/z calcd for $[C_{18}H_{36}O_2Si_3H]^+$ (M+H): 369.2101 ; found 369.2111.



²⁹Si{¹H} NMR:



(Me₃SiO)₂(SiHTol) Using TolSiH₃ (36.6 mg, 0.30 mmol) and Me₃SiOH (56.1 mg, 0.62 mmol). ¹H NMR (600 MHz, C₆D₆, 25 °C): δ 7.67 (d, 2H, *J* = 7.87, Ar-*H*), 7.07 (d, 2H, *J* = 7.07, Ar-*H*), 5.41(s, 1H, ¹*J*_{Si, H} = 243.3 Hz, Si*H*), 2.08 (s, 3H, PhC*H*₃), 0.17 (s, 18H, Si(C*H*₃)₃) ppm. ¹³C{¹H} NMR (150 MHz, C₆D₆, 25 °C): δ 139.9, 133.6, 133.2, 128.7, 25.4, 1.5 ppm. ²⁹Si{¹H} NMR (119 MHz, C₆D₆, 25 °C): δ 10.2, –48.1 ppm.

HRMS (ESI): m/z calcd for $[C_{18}H_{36}O_2Si_3H]^+$ (M+H): 369.2101 ; found 369.2111.







(Me₃SiO)₂(SiH*n*Hex) Using *n*HexSiH₃ (34.8 mg, 0.30 mmol) and Me₃SiOH (56.0 mg, 0.62 mmol). ¹H NMR (600 MHz, C₆D₆, 25 °C): δ 5.0 (s, 1H, ¹J_{Si, H} = 232.4 Hz, Si*H*), 1.50 (quin, 2H, *J* = 7.80, SiCH₂CH₂), 1.35 (quin, 2H, *J* = 7.35, SiCH₂CH₂CH₂), 1.32-1.23 (m, 4H, CH₂CH₂CH3), 0.89 (t, 3H, *J* = 7.02, CH₃), 0.66 (t, 2H, *J* = 8.01, SiCH₂), 0.19 (s, 18H, Si(CH₃)₃) ppm. ¹³C{¹H} NMR (150 MHz, C₆D₆, 25 °C): δ 32.6, 31.7, 22.6, 22.2, 17.5, 14.0, 1.5 ppm. ²⁹Si{¹H} NMR (119 MHz, C₆D₆, 25 °C): δ 9.00, -36.0 ppm. HRMS (ESI): m/z calcd for [C₁₂H₃₂O₂Si₃H]⁺ (M+H): 293.1788 ; found 293.1787.





(thiophen-2-ylMe₂SiO)₂(SiHPh) Using PhSiH₃ (32.4 mg, 0.30 mmol) and thiophen-2ylMe₂SiOH (98.1 mg, 0.62 mmol). ¹H NMR (600 MHz, CD₃CN): δ 7.72 (dd, 2H, *J* = 4.59, 0.87 Hz, Ar-*H*), 7.55 (dd, 2H, *J* = 7.21, 1.38 Hz, Ar-*H*), 7.42 (dt, 2H, *J* = 7.47, 1.41 Hz, Ar-*H*), 7.34 (t, 2H, *J* = 7.32 Hz, Ar-*H*), 7.31 (dd, 2H, *J* = 3.36, 0.84 Hz, Ar-*H*), 7.23-7.21 (m, 2H, Ar-*H*), 5.45 (s, 1H, ¹*J*_{SiH} = 248.9 Hz, Si*H*), 0.39 (s, 6H, SiC*H*₃), 0.38 (s, 6H, SiC*H*₃) ppm. ¹³C{¹H} NMR (150 MHz, C₆D₆): δ 138.2, 136.1, 134.7, 133.2, 130.8, 130.3. 128.0, 1.27, 1.26 ppm. ²⁹Si{¹H} NMR (119 MHz, C₆D₆): δ -3.5, -47.7 ppm.

HRMS (ESI): m/z calcd for $[C_{18}H_{24}O_2S_2Si_3H]^+$ (M+H): 421.0604 ; found 421.0604. ^1H NMR:



Single crystal X-ray diffraction studies.

The single crystal X-ray diffraction measurements of **1a**, **1b**, **2a**, **2b**, **3b**, **4a** and **5a** was performed under a cold nitrogen stream on a Rigaku XtaLAB P200 diffractometer with a Pilatus 200K detector using multi-layer mirrore monochromated Mo K α radiation. The determination of crystal systems and unit cell parameters and data processing were performed with the *CrystalClear* program package. The data sets were corrected for Lorentz and polarzation effects and absorption. The structure was solved by direct methods using SIR2014 program,⁵ and refined by full-matrix least squares calculations on F² for all reflections (SHELXL-2014/7),⁶ using Yadokari-XG 2009.⁷

	1 a	1b	2a	2b	3b	4a	5a
Emprical formula	$C_{38}H_{30}Cl_2FeN_2P_2$	$C_{38}H_{54}Br_2FeN_2P_2$	$C_{40}H_{34}FeN_3OP_2$	$C_{82}H_{122}Fe_2N_6P_4$	$C_{40}H_{34}FeN_3OP_2$	$C_{39}H_{30}FeN_2OP_2$	C44H38FeN2P2Si
Formula weifht	703.33	816.42	690.49	1427.44	658.63	660.44	740.64
Temperature/K	93 (2)	93 (2)	93 (2)	93 (2)	93 (2)	93 (2)	93 (2)
Crystal system	monoclinic	monoclinic	triclinic	monoclinic	Ortho	triclinic	triclinic
Space group	P 21/c	P 21/c	P-1	P 2/n	P na21	P-1	P-1
<i>a /</i> Å	12.0414 (19)	18.299 (5)	9.785 (2)	26.075 (4)	15.1645 (19)	9.8699 (9)	96.735 (9)
<i>b</i> / Å	20.358 (3)	14.140 (4)	12.336 (2)	10.1672 (15)	13.1906 (19)	10.7795 (13)	12.1191 (8)
<i>c</i> / Å	14.058 (2)	15.178 (5)	14.142 (3)	28.018 (4)	16.823 (2)	16.578 (2)	17.8980 (18)
lpha / °	90	90	92.154 (4)	90	90	82.081 (11)	70.273 (16)
eta / °	107.348 (3)	107.351 (7)	103.057 (5)	93.366 (3)	90	75.1840 (10)	74.867 (16)
γ/°	90	90	101.480 (5)	90	90	65.260 (8)	67.391 (14)
Volume / Å ³	3289.4 (9)	374.6 (19)	1623.5 (6)	7415.0 (19)	3365.1 (8)	1547.7 (3)	1802.4 (4)
Ζ	4	4	2	4	4	2	2
<i>R</i> 1 / %	0.029	0.0744	0.0507	0.0771	0.0338	0.0519	0.0576
wR2 / %	0.0644	0.189	0.104	0.1849	0.0802	0.1322	0.1111
GOF	1.028	1.107	1.063	1.128	1.022	1.07	1.672

Table S1. Crystallographic data of 1a, 1b, 2a, 2b, 3b, 4a, 5a



Figure S-1. Molecular structure of **1b** with 50% probability ellipsoids. Hydrogen atoms are omitted for clarity.



Figure S-2. Molecular structure of **1b** with 50% probability ellipsoids. Hydrogen atoms are omitted for clarity.



Figure S-3. Molecular structure of **2b** with 50% probability ellipsoids. Hydrogen atoms are omitted for clarity.



Figure S-4. Molecular structure of **3b** with 50% probability ellipsoids. Hydrogen atoms are omitted for clarity.

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