Supporting Information

Two Luminescent Coordination Polymers as Highly Selective and Sensitive Chemosensors for Cr^{VI}-Anions in Aqueous Medium

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	1	2	
Empirical formula	$C_{20}H_{20}Cd_2N_6O_{13}\\$	C ₂₀ H ₂₀ Zn ₂ N ₆ O ₁₃	
Formula weight	777.22	683.16	
Temperature/K	298	298	
Crystal system	monoclinic	monoclinic	
Space group	<i>I</i> 2/ <i>a</i>	<i>I</i> 2/ <i>a</i>	
a/Å	17.0062(5)	16.3819(2)	
b/Å	10.9658(3)	10.7728(2)	
c/Å	13.4337(4)	13.4974(2)	
$\alpha/^{\circ}$	90	90	
β/°	99.160(3)	98.2860(10)	
$\gamma/^{\circ}$	90	90	
Volume/Å ³	2473.26(13)	2357.14(6)	
Z	4	4	
Goodness-of-fit on F ²	1.062	1.041	
Final R indexes [I>=2 σ (I)]	$R_1 = 0.0260, wR_2 = 0.0682$ $R_1 = 0.0285, wR_2 $		
Final R indexes [all data]	$R_1 = 0.0281$, $wR_2 = 0.0694$	= 0.0694 $R_1 = 0.0301, wR_2 = 0.0770$	

Table S1 Crystal data and structure refinement results for compound 1 and 2.

Table S2 Selected bond lengths (\AA) of 1 and 2

1			2		
Cd1	O1 ^{#1}	2.448(2)	Zn1	O1#1	2.2998(18)
Cd1	O2 ^{#1}	2.287(2)	Zn1	O2 ^{#1}	2.1342(17)
Cd1	03	2.163(2)	Zn1	O4	1.9823(16)
Cd1	05	2.373(3)	Zn1	O5	2.163(2)
Cd1	O6	2.364(3)	Zn1	O6	2.1618(18)
Cd1	N3#2	2.286(3)	Zn1	N3 ^{#2}	2.0950(19)
^{#1} 1/2+X,2-Y,+Z; ^{#2} +X,1+Y,+Z		^{#1} -1/2+X,-Y,+Z; ^{#2} +X,-1+Y,+Z			



Fig. S1 ¹H-NMR spectrum of ligand H₂L (DMSO-d₆).



Fig. S2 PXRD patterns of 1 and 2.



Fig. S3 Thermogravimetric analyses (TGA) profiles of 1 and 2.



Fig. S4 PXRD patterns of 1 and 2 immersed in various solvents after 60 h.



Fig. S5 Excitation and emission spectra of H_2L in solid state.



Fig. S6 Excitation and emission spectra of 1 and 2 in solid-state.



Fig. S7 PXRD patterns of 1 toward $Cr_2O_7^{2-}$ and CrO_4^{2-} after 5 cyclic experiments.



Fig. S8 PXRD patterns of 2 toward $Cr_2O_7^{2-}$ and CrO_4^{2-} after 5 cyclic experiments.



Fig. S9 UV-Vis spectra of different salts in aqueous solution.



Fig. S10 Luminescence quenching effect of 1 and 2 towards different concentrations of $CrO_4^{2-}(a)/Cr_2O_7^{2-}(b)$.



Fig. S11 IR spectra of 1 and 2.