

Supplementary Information for:

## **Tuning the Electrochemiluminescent properties of Iridium Complexes of N-Heterocyclic Carbene Ligands**

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### **X ray crystallography**

X-ray Crystallography. Single crystals of the Ir(III) complexes **1**, **3**, **5**, and **6** suitable for X-ray diffraction studies were grown by slow diffusion of ethyl acetate into acetonitrile solutions of each compound at ambient temperature. Crystallographic data for all structures determined are given in Tables 1 and S1. For all samples, crystals were removed from the crystallisation vial and immediately coated with paratone oil on a glass slide. A suitable crystal was mounted in Paratone oil on a glass fibre and cooled rapidly to 173 K in a stream of cold N<sub>2</sub> using an Oxford low temperature device. Diffraction data were measured using an Oxford Gemini diffractometer mounted with Mo-K $\alpha$   $\lambda$  = 0.71073 Å and Cu-K $\alpha$   $\lambda$  = 1.54184. Data were reduced and corrected for absorption using the CrysAlis Pro program. The SHELXL2013-2 program was used to solve the structures with Direct Methods, with refinement by the Full-Matrix Least-Squares refinement techniques on F<sup>2</sup>. The non-hydrogen atoms were refined anisotropically and hydrogen atoms were placed geometrically and refined using the riding model. Coordinates and anisotropic thermal parameters of all non-hydrogen atoms were refined. All calculations were carried out using the program Olex2. Images were generated by using ORTEP-3. Further XRD details are provided in the Supporting Information. CCDC 1869253-1869256 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via [www.ccdc.cam.ac.uk/data\\_request/cif](http://www.ccdc.cam.ac.uk/data_request/cif).

Table S1. Crystallographic data for all structures **1**, **2**, **5**, and **6**.

Compound	<b>1</b>	<b>3</b>	<b>5</b>	<b>6</b>
Empirical formula	C <sub>34</sub> H <sub>24</sub> N <sub>5</sub> F <sub>4</sub> Ir	C <sub>34</sub> H <sub>25</sub> F <sub>4</sub> IrN <sub>4</sub> O <sub>2</sub>	C <sub>32</sub> H <sub>19</sub> N <sub>4</sub> F <sub>6</sub> Ir	C <sub>36</sub> H <sub>29</sub> N <sub>5</sub> F <sub>4</sub> IrCl
Formula weight	770.78	789.78	765.71	835.29
Temperature/K	173	173	173	173
Crystal system	triclinic	triclinic	monoclinic	triclinic
Space group	<i>P</i> -1	<i>P</i> -1	<i>P</i> 2 <sub>1</sub> / <i>n</i>	<i>P</i> -1
<i>a</i> /Å	9.8986(3)	10.3830(3)	10.84307(11)	9.0031(3)
<i>b</i> /Å	11.6908(4)	10.6683(3)	15.66542(16)	12.7459(4)
<i>c</i> /Å	13.1739(4)	13.4176(4)	15.57142(16)	15.4060(5)
$\alpha$ /°	71.971(3)	76.274(2)	90	104.264(3)
$\beta$ /°	89.258(2)	82.985(2)	92.5122(9)	105.973(3)
$\gamma$ /°	88.706(3)	84.990(2)	90	98.494(2)
Volume/Å <sup>3</sup>	1449.28(8)	1430.37(7)	2642.44(5)	1602.56(9)
<i>Z</i>	2	2	4	2
$\rho$ calc/mg/mm <sup>3</sup>	1.766	1.834	1.925	1.731
$\mu$ /mm <sup>-1</sup>	9.429	4.734	10.44	4.307
F(000)	752	772	1480	820
Crystal size/mm <sup>3</sup>	0.15 × 0.15 × 0.04	0.08 × 0.04 × 0.04	0.1 × 0.08 × 0.05	0.1 × 0.05 × 0.05
Radiation	CuK $\alpha$ ( $\lambda$ = 1.54184)	MoK $\alpha$ ( $\lambda$ = 0.71073)	CuK $\alpha$ ( $\lambda$ = 1.54184)	MoK $\alpha$ ( $\lambda$ = 0.71073)
2 $\theta$ range for data collection	7.056 to 147.464°	6.286 to 52.74 -12 ≤ <i>h</i> ≤ 11, -13	8.01 to 147.702°	5.77 to 52.738°
Index ranges	-12 ≤ <i>h</i> ≤ 12, -14 ≤ <i>k</i> ≤ 13, -16 ≤ <i>l</i> ≤ 11	≤ <i>k</i> ≤ 13, -16 ≤ <i>l</i> ≤ 16	-8 ≤ <i>h</i> ≤ 13, -19 ≤ <i>k</i> ≤ 18, -19 ≤ <i>l</i> ≤ 19	-11 ≤ <i>h</i> ≤ 11, -15 ≤ <i>k</i> ≤ 15, -19 ≤ <i>l</i> ≤ 19
Reflections collected	9129	11588 5844 [R <sub>int</sub> =	17700	14873
Independent reflections	5672 [R <sub>int</sub> = 0.0346, R <sub>sigma</sub> = 0.0290]	0.0181, R <sub>sigma</sub> = 0.0280]	5281 [R <sub>int</sub> = 0.0209, R <sub>sigma</sub> = 0.0172]	6542 [R <sub>int</sub> = 0.0255, R <sub>sigma</sub> = 0.0347]
Data/restraints/parameters	5672/0/399	5844/0/409	5281/0/389	6542/0/428
Goodness-of-fit on F <sup>2</sup>	1.054	1.075	1.16	1.043
Final R indexes [ <i>I</i> ≥ 2 $\sigma$ ( <i>I</i> )]	R <sub>1</sub> = 0.0473, wR <sub>2</sub> = 0.1270	R <sub>1</sub> = 0.0204, wR <sub>2</sub> = = 0.0498	R <sub>1</sub> = 0.0205, wR <sub>2</sub> = 0.0506	R <sub>1</sub> = 0.0209, wR <sub>2</sub> = 0.0473
Final R indexes [all data]	R <sub>1</sub> = 0.0480, wR <sub>2</sub> = 0.1279	R <sub>1</sub> = 0.0233, wR <sub>2</sub> = = 0.0515	R <sub>1</sub> = 0.0212, wR <sub>2</sub> = 0.0509	R <sub>1</sub> = 0.0244, wR <sub>2</sub> = 0.0492
Largest diff. peak/hole / e Å <sup>-3</sup>	3.29/-2.19	0.89/-0.82	0.74/-1.13	0.85/-0.71

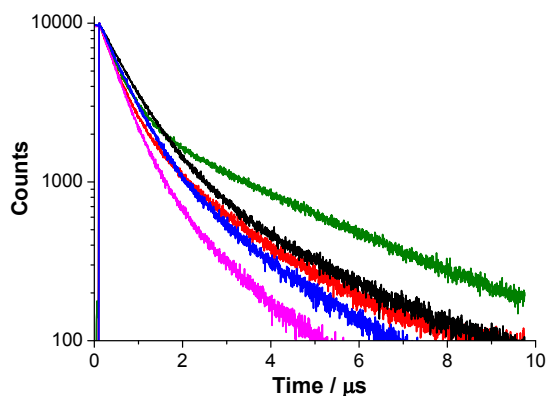
**Details:**

**1:** Solved using direct methods in the triclinic space group *P*-1. The asymmetric unit contains one molecule of the title compound.

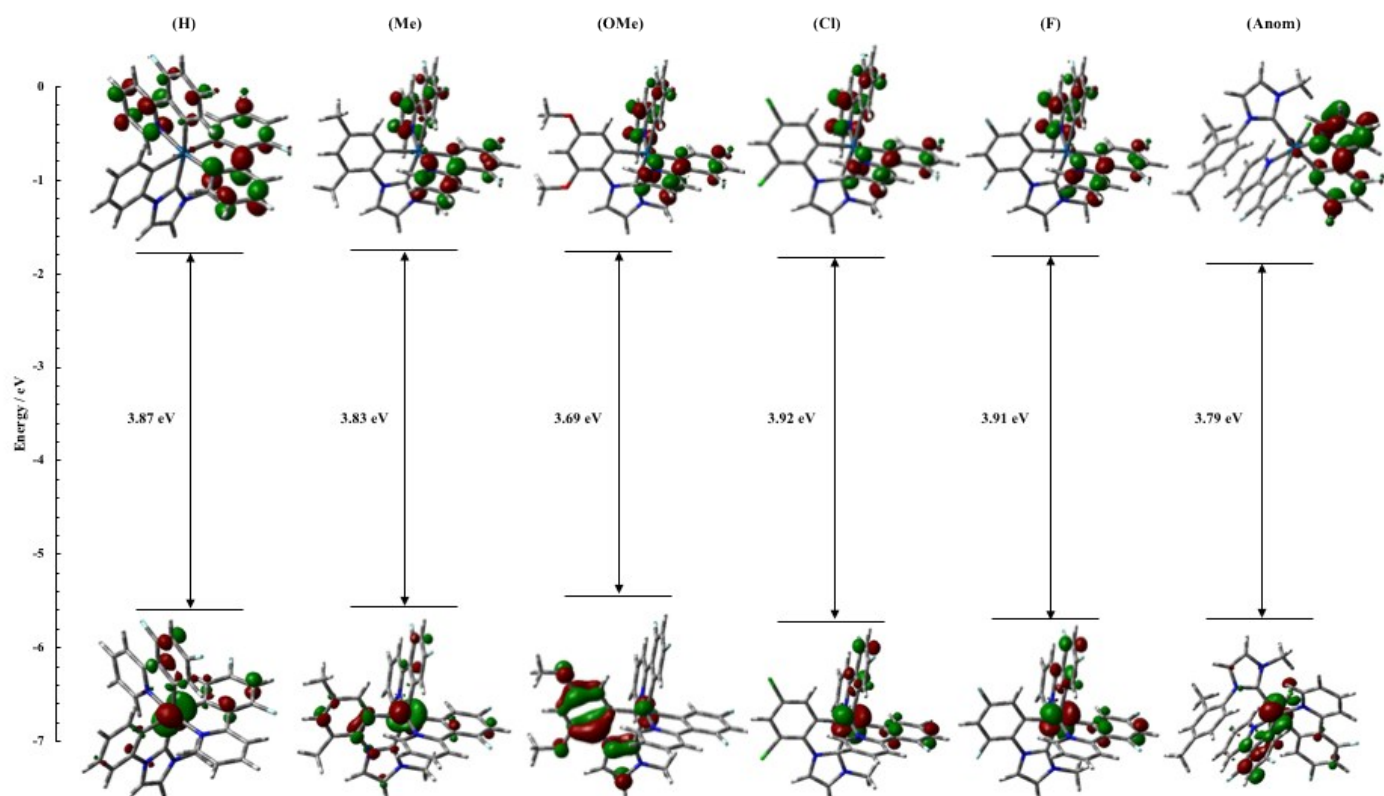
**3:** Solved using direct methods in the triclinic space group *P*-1. The asymmetric unit contains one molecule of the title compound. Disorder was identified in the positions of fluorine atoms: F1 and F3. Attempts to model this disorder using two different closely spaced positions for each of these F atoms were unsatisfactory with the site occupancy for one component refining to unreasonably low levels. As a result, disorder in the positions of these atoms was not modelled. Three level B Checkcif alerts result from the disorder associated with these F atoms: Hirshfeld Test Diff for F1 - C12 ..10.7 su, Hirshfeld Test Diff for F3 - C23 .. 9.0 su and Short Inter HL..HL Contact F1 .. F3 . 2.42 Ang.

**5:** Solved using direct methods in the monoclinic space group  $P2_1/n$ . The asymmetric unit contains one molecule of the title compound.

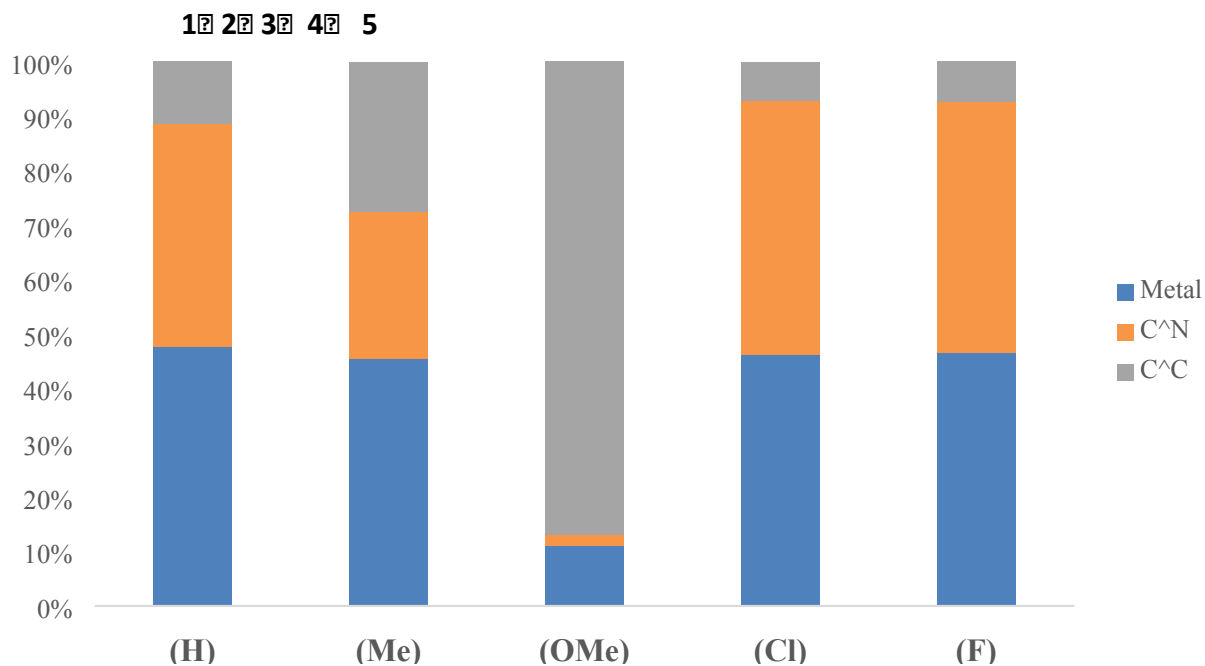
**6:** Solved using direct methods in the triclinic space group  $P-1$ . The asymmetric unit contains one molecule of the title compound and one molecule of acetonitrile as a solvent of crystallisation.



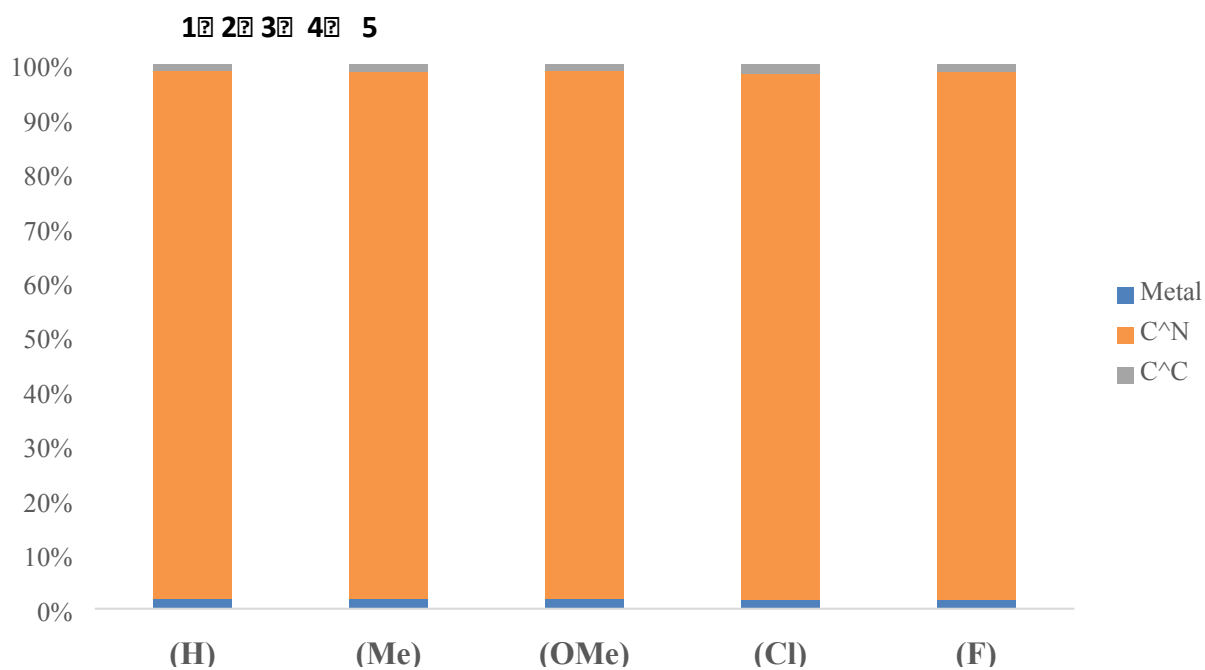
**Figure S1.** Photoluminescence decay profiles for dilute solutions of complexes 1 (red) 2 (green) 3 (black) 4 (pink) and 5 (blue) in de-aerated acetonitrile



**Figure S2.** B3LYP/def2-TZVP//B3LYP/def2-SVP calculated molecular orbital energy HOMO-LUMO gaps and surface plots of the HOMO and LUMO for compounds 1-5. Acetonitrile solvent correction included with single-point SCRF. Units of eV.



**Figure S3.** Percentage contribution to the HOMO from the metal centre and ligands (C<sup>C</sup> and C<sup>N</sup> ligands) of complexes 1-5 (B3LYP/def2-TZVP calculations).



**Figure S4.** Percentage contribution to the LUMO from the metal centre and ligands (C<sup>C</sup> and C<sup>N</sup> ligands) of complexes 1-5 (B3LYP/def2-TZVP calculations).