

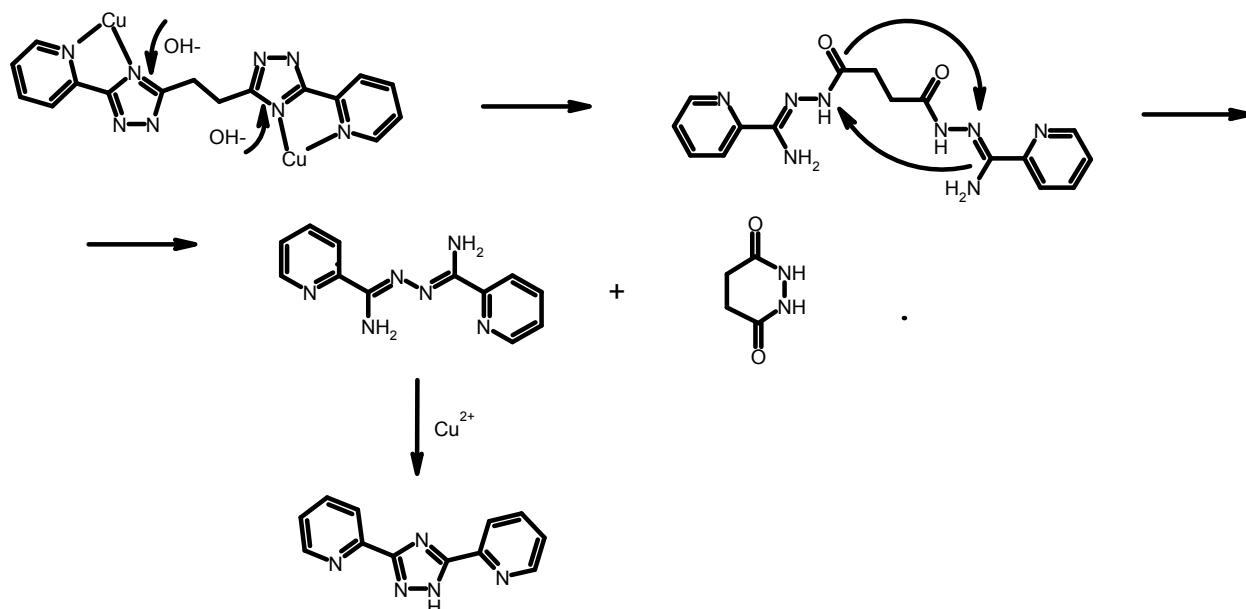
Supplementary Information

Copper(II) self-assembled clusters of bis((pyridin-2-yl)-1,2,4-triazol-3-yl)alkanes. Unusual rearrangement of ligand upon reaction condition.

Table S1. Crystal data and structure refinements for **1-4**.

Parameter/Complex	1	2	3	4*
Molecular formula	[Cu ₂ (H ₂ L1)Cl ₂]Cl ₂ ·6.5 MeOH	[Cu ₄ L1 ₄]·6H ₂ O·2.8 MeOH	[Cu(H ₂ L2)(ClO ₄) ₂]	[Cu ₃ (OH)Na ₂ (L') ₆]·(ClO ₄) ₃ ·11H ₂ O
Formula	C _{36.5} H ₅₀ Cl ₄ Cu ₂ N ₁₆ O _{6.5}	C _{62.8} H _{57.6} Cu ₄ N ₃₂ O _{8.8}	C ₁₆ H ₁₄ Cl ₂ CuN ₄ O ₈	C ₇₂ H ₇₀ ClCu ₃ N ₃₀ Na ₂ O ₁₆
Crystal system	Monoclinic	Tetragonal	Monoclinic	Triclinic
Space group	<i>C2/c</i>	<i>I</i> $\bar{4}$	<i>C2/c</i>	<i>P</i> $\bar{1}$
<i>a</i> , Å	19.725(3)	13.1705(9)	13.0990(11)	15.506(3)
<i>b</i> , Å	18.998(3)	13.1705(9)	11.6970(7)	16.160(3)
<i>c</i> , Å	12.833(2)	19.6437(14)	14.5274(12)	20.311(4)
α°	90	90	90.0	86.840(3)
β°	95.971(5)	90	112.745(10)	89.884(3)
γ°	90	90	90.0	61.689(3)
<i>V</i> , Å ³	4783.0(14)	3407.4(5)	2052.8(3)	4472.5(16)
<i>Z</i>	4	2	4	2
μ_{Mo} , mm ⁻¹	1.175	1.314	1.392	0.824
Parameters	328	254	159	1124
No. unique	5761	4536	1819	14633
No. <i>I</i> > 2σ(<i>I</i>)	4107	3931	1595	4410
<i>GOF</i>	1.017	1.058	1.084	0.713
<i>R</i> (<i>I</i> > 2σ(<i>I</i>))	0.0531	0.0371	0.0250	0.0658
w <i>R</i> ₂	0.1544	0.0939	0.0695	0.1459

* the formula does not include masked solvent



Scheme S1. Proposed mechanism for the rearrangement of the ligand H₂L2 to afford the 3,5-bis-(pyridin-2-yl)-1,2,4-triazole

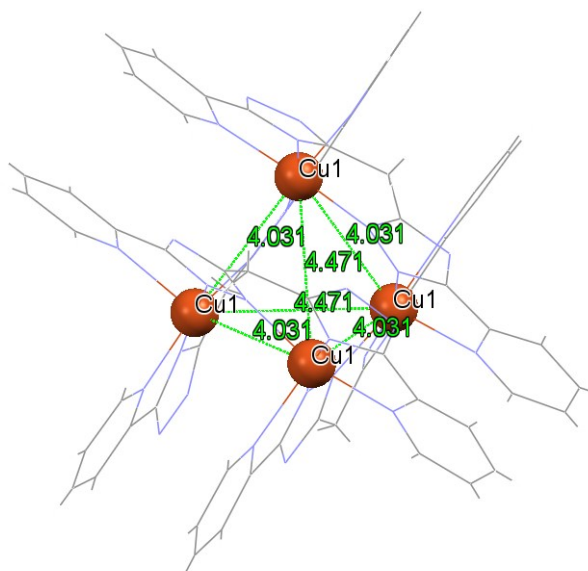


Figure.S1. Copper-copper distances in tetranuclear core of complex 2.

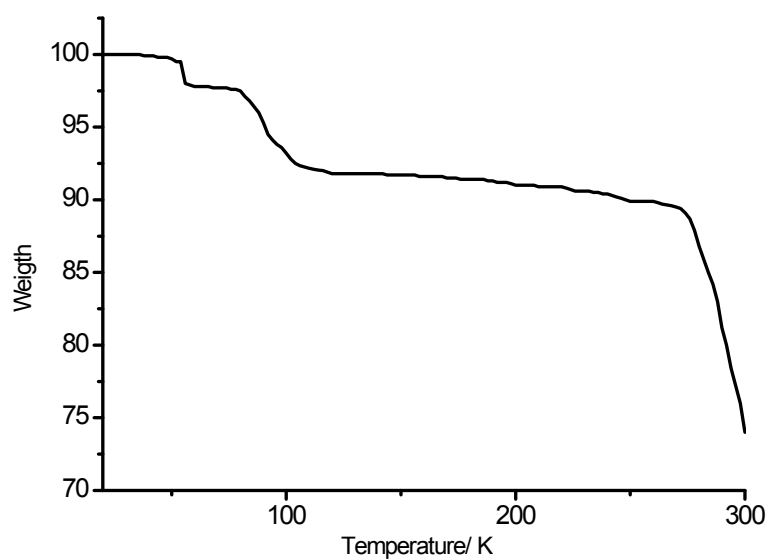


Figure S2. TG curve of 4.

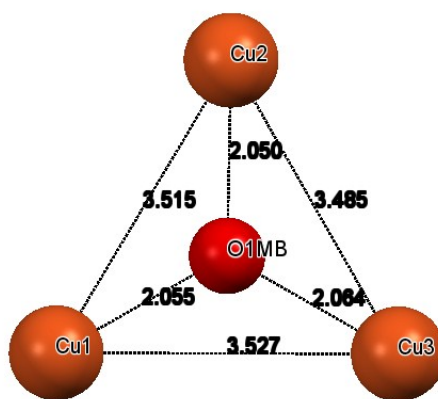


Figure S3. Structure of trinuclear Cu_3OH core in complex 4.

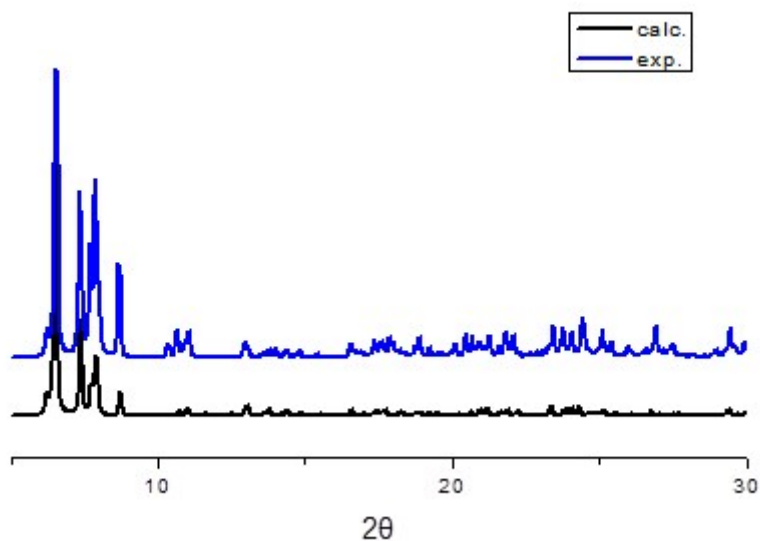


Figure S4. XRPD curve of **4**.

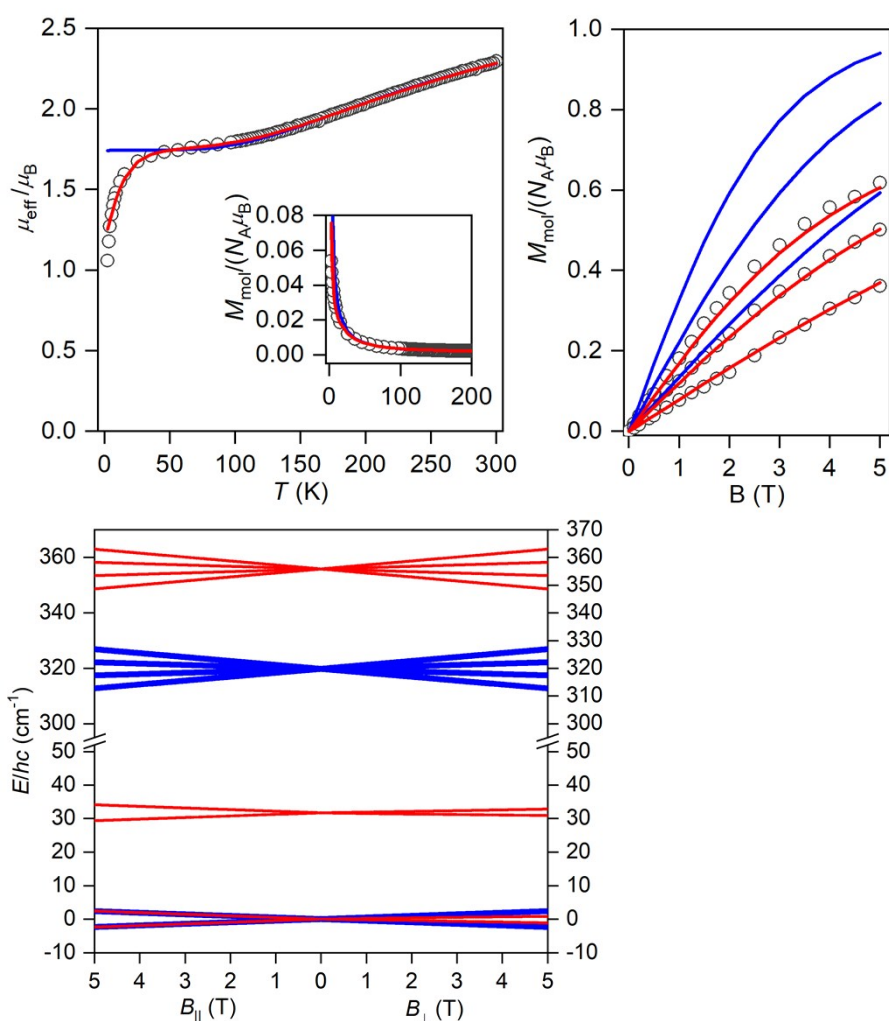


Figure S5. *Top*: Temperature dependence of the effective magnetic moment (calculated from magnetization at $B = 0.5$ T) of **4** with the low-temperature region expanded in the inset and the isothermal magnetization data measured at $T = 2, 3$ and 5 K. Circles are experimental points, blue lines are calculated using the best-fit parameters: $J = -213$ cm⁻¹ and $g = 2.02$, red lines = calculated using the best-fit parameters: $J_{12} = -218$ cm⁻¹, $J_{13} = J_{23} = -231$ cm⁻¹, $g = 2.05$

and $|d_z| = 16.7 \text{ cm}^{-1}$. *Bottom*: the calculated energy levels in magnetic field using $J = -213 \text{ cm}^{-1}$ and $g = 2.02$ (blue color) and $J_{12} = -218 \text{ cm}^{-1}$, $J_{13} = J_{23} = -231 \text{ cm}^{-1}$, $g = 2.05$ and $|d_z| = 16.7 \text{ cm}^{-1}$ (red color).

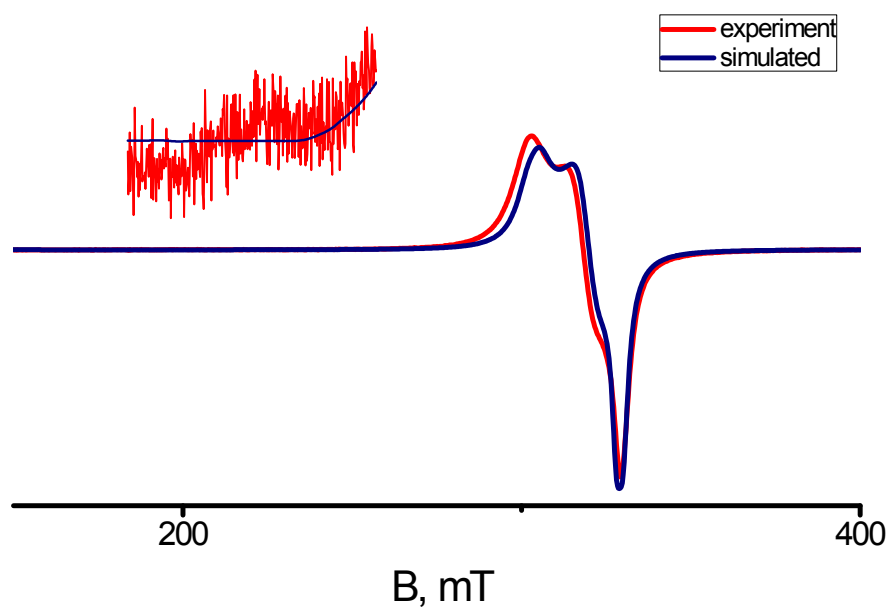


Figure S6. Experimental and simulated X-Band EPR spectra of a polycrystalline sample of complex **1** at 50 K

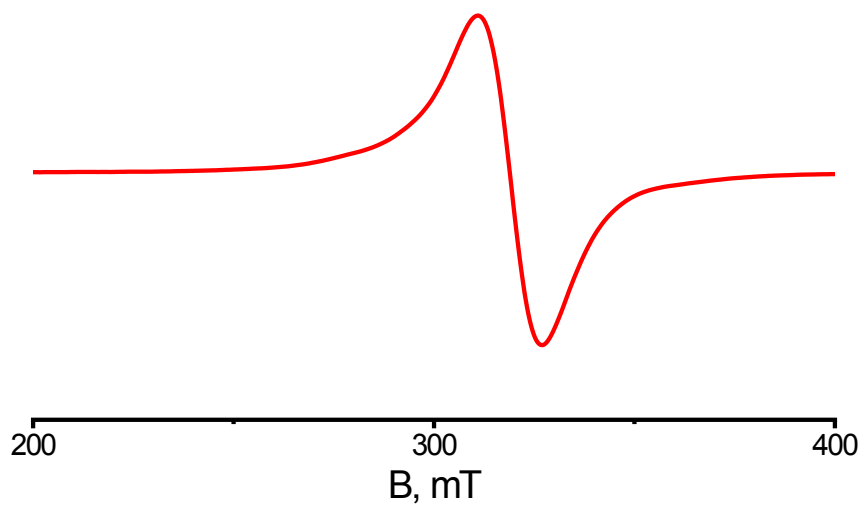


Figure S7. Experimental X-Band EPR spectra of a polycrystalline sample of complex **4** at 294 K