Electronic Supplementary Information

Fabrication of blue organic light-emitting diode from novel uranium complexes: Synthesis, characterization, and electroluminescence studies of uranium anthracene-9-carboxylate complexes

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Complex	1	2	3
Empirical formula	$C_{38}H_{42}O_{10}U \\$	$C_{36}H_{40}O_{14}U_2$	$C_{62}H_{48}O_{16}U_2$
Formula weight (g mol ⁻¹)	896.75	1172.74	1525.06
Т (К)	140	140	140
Wavelength λ (Å)	0.71073	0.71073	0.71073
Crystal system	monoclinic	triclinic	triclinic
Space group	$P2_{1}/c$	<i>P</i> -1	<i>P</i> -1
<i>a</i> (Å)	7.8016(8)	8.9252(7)	10.6301(6)
<i>b</i> (Å)	26.772(3)	9.2387(7)	11.9054(7)
<i>c</i> (Å)	17.7555(19)	12.1717(9)	11.9924(7)
α (°)	90	68.676(3)	110.639(3)
β (°)	93.142(5)	85.383(3)	100.789(3)
γ (°)	90	84.565(3)	98.333(3)
V (Å ³)	3702.9(7)	929.58(12)	1357.98(14)
Z	4	1	1
D _{calc} (mg m ⁻³)	1.609	2.095	1.865
F(000)	1768.0	552.0	732.0
μ (mm ⁻¹)	4.439	8.767	6.028
Crystal size (mm ³)	0.328×0.346×0.679	0.05×0.1×0.2	0.05×0.05×0.1
θ range (°)	1.52-26.37	1.80-27.81	1.88-27.91
Reflections collected	106147	19539	24875
Unique reflection (R _{int})	7573 (0.033)	4370 (0.037)	6208 (0.025)
$R_{1}, w R_{2}{}^{a}[I > 2\sigma (I)]$	0.020, 0.045	0.018, 0.048	0.020, 0.037
R_1 , wR_2^a (all data)	0.022, 0.046	0.019, 0.049	0.017, 0.038

Table S1 Crystal data and structure refinement for 1, 2 and 3.

 ${}^{\mathrm{a}}R_{1} = \Sigma ||F_{\mathrm{o}}| - |F_{\mathrm{c}}|| / \Sigma |F_{\mathrm{o}}|, wR_{2} = [\Sigma (w(F_{\mathrm{o}}{}^{2} - F_{\mathrm{c}}{}^{2})^{2}) / \Sigma w(F_{\mathrm{o}}{}^{2})^{2}]^{1/2}$

2.500(2) U(1)–O(1) 1.747(2) U(1)–O(5) U(1)–O(2) 1.750(2) U(1)–O(6) 2.4931(19) U(1)-O(3)2.481(2) U(1) - O(7)2.433(2)2.4246(19) U(1)-O(4)2.506(2) U(1)–O(8) 179.51(11) 179.27(9) O(1)-U(1)-O(2)O(3)-U(1)-O(6) 91.40(11) 92.34(10) O(1)-U(1)-O(8)O(1)-U(1)-O(5)88.83(12) 87.18(11) O(2)-U(1)-O(8)O(2)-U(1)-O(5)O(1)-U(1)-O(7)87.45(12) O(8)-U(1)-O(5)115.20(7) 65.09(7) O(2)-U(1)-O(7)92.32(12) O(7)-U(1)-O(5)O(8)-U(1)-O(7)178.83(9) O(3) - U(1) - O(5)128.82(7) O(1)-U(1)-O(3)92.13(10) O(6)-U(1)-O(5)50.90(6) O(2)-U(1)-O(3)88.15(10) O(1)-U(1)-O(4)91.01(10) O(8) - U(1) - O(3)115.61(7) O(2)-U(1)-O(4)89.48(11) O(7)-U(1)-O(3)64.23(7) O(8)-U(1)-O(4)64.94(7) 88.57(10) 114.84(7) O(1)-U(1)-O(6)O(7)-U(1)-O(4)91.15(10) O(3)-U(1)-O(4)50.74(7) O(2)-U(1)-O(6)64.58(6) 129.49(7) O(8)-U(1)-O(6)O(6)-U(1)-O(4)O(7)-U(1)-O(6)115.59(7) O(5)-U(1)-O(4)176.65(10)

Table S2 Selected band lengths (Å) and bond angles (°) for 1.

U(1)-O(3)	1.769(2)	U(1)–O(2)	2.453(2)
U(1)–O(4)	1.766(2)	U(1)–O(1)	2.489(2)
U(1)–O(6)	2.3264(19)	U(1)–O(5)	2.400(2)
O(3)-U(1)-O(4)	176.73(9)	O(4)–U(1)–O(1)	90.68(9)
O(3)-U(1)-O(6)	92.43(9)	O(6)–U(1)–O(1)	159.51(7)
O(4)-U(1)-O(6)	93.14(9)	O(5)–U(1)–O(1)	73.16(7)
O(3)-U(1)-O(6)	90.12(9)	O(2)–U(1)–O(1)	52.64(7)
O(4)-U(1)-O(6)	93.14(9)	O(3)–U(1)–O(1)	87.31(9)
O(6)-U(1)-O(6)	70.67(8)	O(5)–U(1)–O(2)	125.80(7)
O(3)-U(1)-O(5)	88.53(9)	O(6)–U(1)–O(2)	77.22(7)
O(4)-U(1)-O(5)	88.43(9)	O(4)–U(1)–O(2)	91.79(9)
O(6)-U(1)-O(5)	86.35(7)	O(3)–U(1)–O(2)	89.03(9)

 Table S3 Selected band lengths (Å) and bond angles (°) for 2.

U(1)–O(4)	1.7621(16)	U(1)–O(7)	2.3619(16)
U(1)–O(3)	1.7622(16)	U(1)–O(2)	2.5091(17)
U(1)–O(5)	2.3769(18)	U(1)–O(1)	2.4139(19)
U(1)–O(6)	2.3154(18)		
O(4)-U(1)-O(3)	177.51(9)	O(7)–U(1)–O(2)	122.44(6)
O(4)-U(1)-O(5)	85.33(8)	O(4)–U(1)–O(1)	90.13(8)
O(3)–U(1)–O(5)	92.31(8)	O(3)–U(1)–O(1)	91.91(8)
O(4)-U(1)-O(6)	90.87(7)	O(5)–U(1)–O(1)	126.73(6)
O(3)-U(1)-O(6)	87.98(7)	O(6)–U(1)–O(1)	152.13(6)
O(5)-U(1)-O(6)	81.09(6)	O(7)–U(1)–O(1)	70.83(6)
O(4)-U(1)-O(7)	91.30(7)	O(2)–U(1)–O(1)	52.31(6)
O(3)–U(1)–O(7)	90.71(7)	O(6)–U(1)–O(2)	154.73(6)
O(5)-U(1)-O(7)	162.02(7)	O(5)–U(1)–O(2)	75.52(6)
O(6)-U(1)-O(7)	81.30(6)	O(3)–U(1)–O(2)	83.57(7)
O(4)-U(1)-O(2)	96.57(7)		

 Table S4 Selected band lengths (Å) and bond angles (°) for 3.

Cg(I)····Cg(J)	Cg···Cg (Å)	α (°)	β (°)	γ(°)	Slippage (Å)	Symmetry operation on Cg(J)
2						
Cg1-Cg1	3.7136(19)	0.00(15)	21.5	21.5	1.359	2-x,1-y,-y
Cg1-Cg3	3.664(2)	1.11(17)	18.7	19.2	1.174	2-x,1-y,-z
Cg2-Cg3	3.747(2)	1.92(18)	22.2	20.3	1.414	2-x,1-y,-z
Cg1 (C2, C3, C8, C	C9, C10, C15), Cg	2 (C3>C8), 0	Cg3 (C1	0>C15)	1	
3						
Cg1- Cg1	3.8469 (13)	0.03(10)	17.6	17.6	1.160	-x, -y, -z
Cg1- Cg3	3.9136 (14)	0.61(11)	20.8	20.2	1.390	-x, -y, -z
Cg3- Cg2	3.8181 (14)	2.15(11)	17.0	18.4	1.114	-x, -y, -z
Cg1 (C2, C3, C8, C9, C10, C15), Cg2 (C3>C8), Cg3 (C10>C15)						
Cg is the centroid of the ring						
α=Dihedral angle between Planes Cg						
β =Angle between Cg(I) \rightarrow Cg(J) vector and normal to plane I						
γ =Angle between Cg \rightarrow Cg vector and normal to plane J						

Table S5 Summary of short π - π ring-interactions in the crystal structures of complexes 2 and 3.

Х-Н	Cg	H···Cg(Å)	X ··Cg(Å)	γ (°)	X-H···Cg (°)		
1							
С17-Н17	Cg (C4-C6, C11-C13)	2.83	3.3906(4)	12.76	119		
С19-Н19	Cg (C1-C4, C13-C14)	2.86	3.5220(4)	7.22	127		
С35-Н35В	Cg (C18-C20, C25-C27)	2.76	3.6936(4)	7.35	156		
С37-Н37В	Cg (C4-C6, C11-C13)	2.87	3.7345(4)	14.10	147		
2							
С17-Н17	Cg (C10>C15)	2.72	3.437(4)	8.59	131		
3							
O8-H8B	(C26>C31)	2.68(4)	3.324(3)	18.77	135(4)		
С25-Н25	(C10>C15)	2.92	3.746(2)	6.11	146		
С27-Н27	Cg (C2-C3, C8-C10, C15)	2.60	3.350(2)	15.58	136		
γ = Angle between Cg \rightarrow H vector and ring Cg normal							

Table S6 Summary of intermolecular X-H \cdots π interactions in the crystal structures of complexes 1, 2, and 3.

Table S7 Photophysical data of complexes, anthracene-9-carboxylic acid and uranyl acetate dihydrate.

abso	emission			
	λ_{max} (nm)	ε (L/mol cm)×10 ³	λ_{max} (nm)	FWHM (nm)
Complex 1	250, 368	2599, 363	440	75
Complex 2	256, 368	1113, 64	443	70
Complex 3	255, 367	2151, 700	443	75
Anthracene-9-carboxylic acid	252, 364	2.62, 1.64	441	75
Uranyl acetate dihydrate	239	0.80	615	72



Figure S1 Infrared spectra of compounds 1, 2 and 3.



Figure S2 TGA diagrams of compounds 1, 2 and 3.



Figure S3 X-ray molecular structure for $[UO_2(C_{15}H_9O_2)_2(CH_3CH_2OH)_2] \cdot 2CH_3CH_2OH$ (1) with the atom-labeling scheme. Displacement ellipsoids are drawn at the 50% probability level.



Figure S4 X-ray molecular structure for $[U_2O_4(C_{15}H_9O_2)_2(CH_3O)_2(CH_3OH)_2] \cdot 2CH_3OH$ (2) with the atom-labeling scheme. Displacement ellipsoids are drawn at the 50% probability level. Symmetry transformation i = 1-x, 1-y, 1-z.



Figure S5 X-ray molecular structure for $[U_2O_4(C_{15}H_9O_2)_4(CH_3OH)_2] \cdot 2H_2O$ (**3**) with the atomlabeling scheme. Displacement ellipsoids are drawn at the 50% probability level. Symmetry transformation i = 1-x, 1-y, -z.



Figure S6 UV spectra of anthracene-9-carboxylic acid (non-deprotonated ligand) and uranyl acetate dihydrate.



Figure S7 PL spectrum of anthracene-9-carboxylic acid at an excitation wavelength of 270 nm.



Figure S8 PL spectrum of uranyl acetate dihydrate at an excitation wavelength of 270 nm.





PVK



DCM









Scheme S1 Formulae of materials used to fabricate the BOLED: 2,9-dimethyl-4,7-diphenyl-1,10-phenanthroline (BCP), poly(9-vinylcarbazole) (PVK), 4-(dicyanomethylene)-2-methyl-6-(4-dimethylaminostyryl)-4*H*-pyran (DCM), *N*,*N*'-bis(3-methylphenyl)-*N*,*N*'-diphenylbenzidine (TPD), 2-phenyl-5-(4-biphenylyl)-1,3,4-oxadiazole (PBD), poly(3,4-ethylenedioxythiophene) poly(styrenesulfonate) (PEDOT:PSS.)