

# Restricted Rotation of an Fe(CO)<sub>2</sub>(PL<sub>3</sub>)-subunit in [FeFe]-Hydrogenase Active Site Mimics by Intramolecular Ligation

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## Supporting Info

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### <sup>31</sup>P-NMR spectra

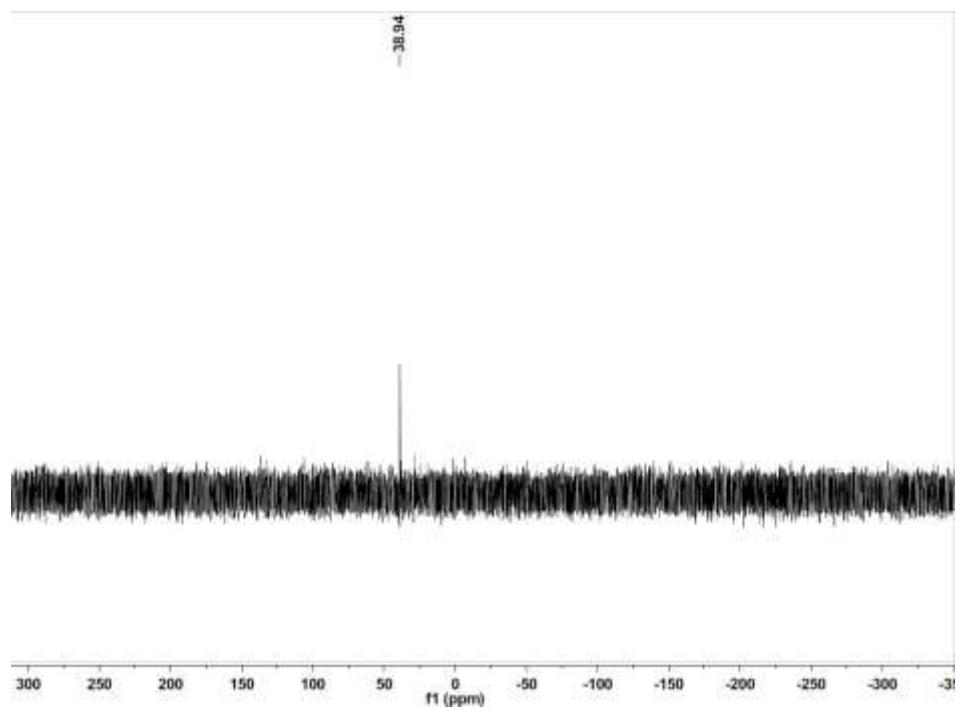


Figure S1:  $\text{Fe}_2(\text{propylamide-mabdt})(\text{CO})_5(\text{PPh}_2\text{Me})$  (**6**) ( $\text{CDCl}_3$ )

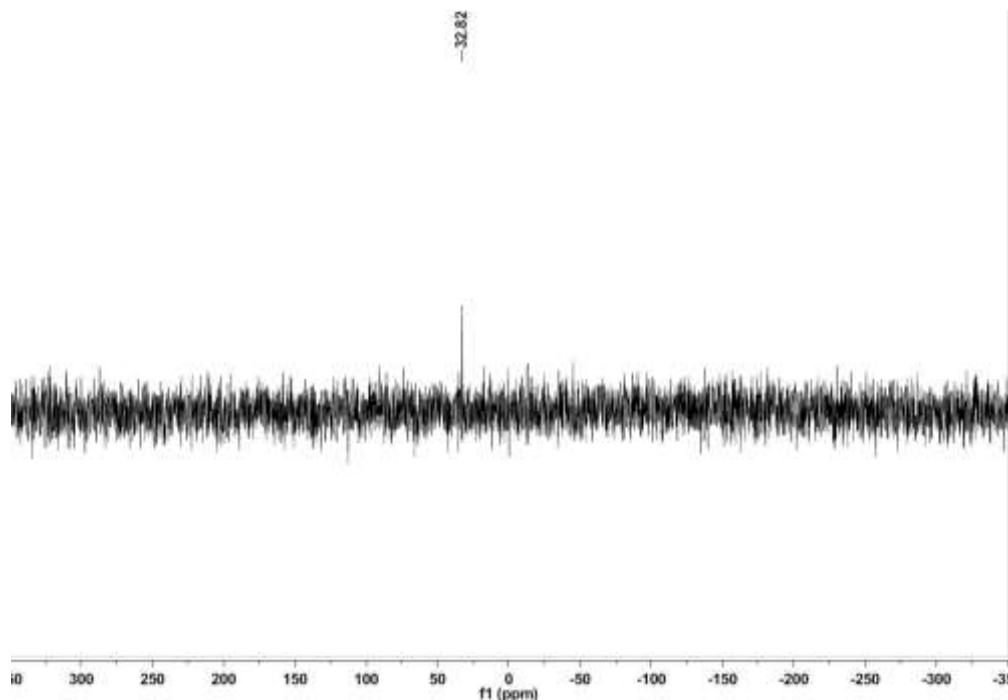
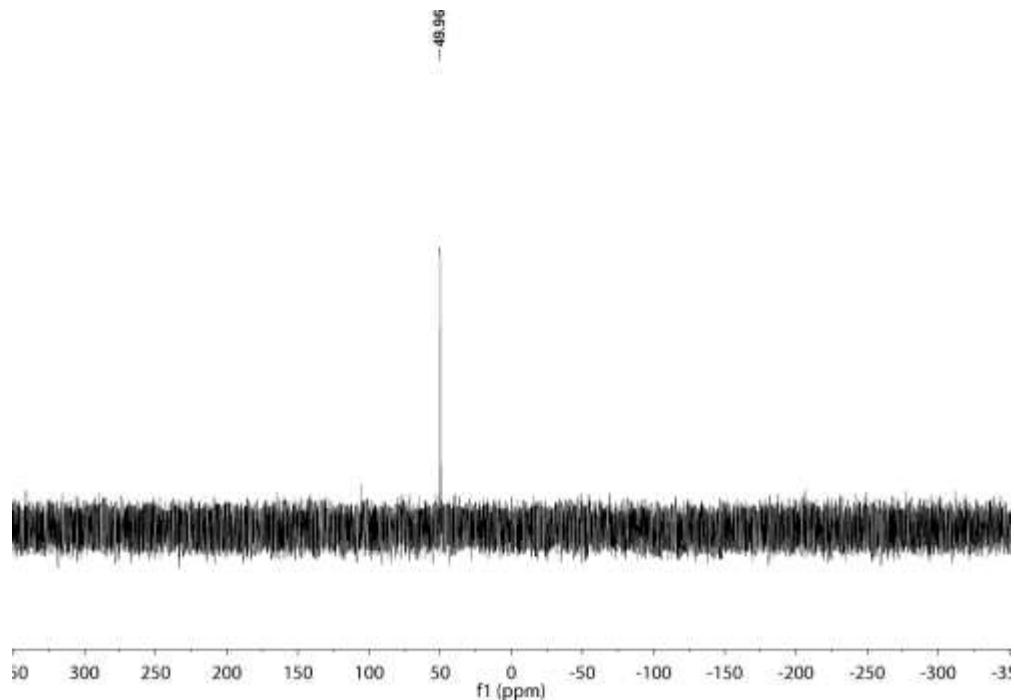
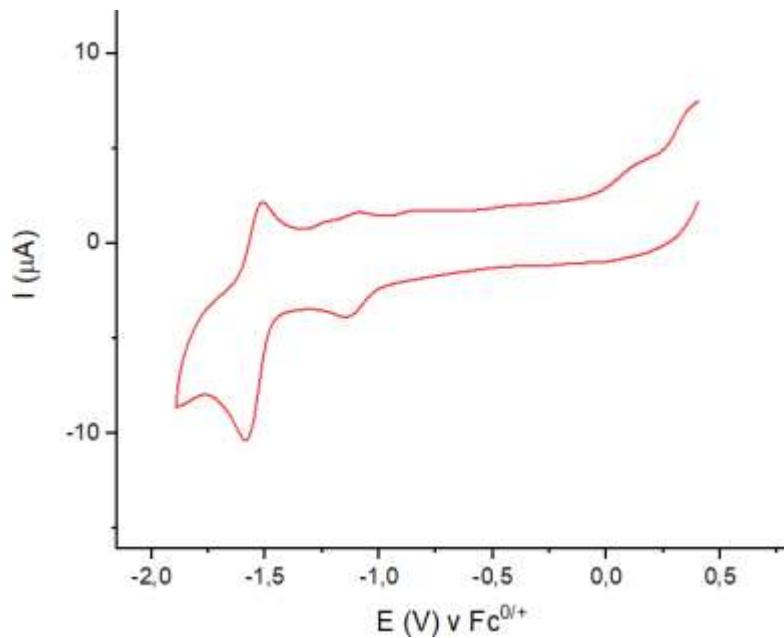


Figure S2:  $\text{Fe}_2(\text{propyl-tethered-mabdt-PPh}_2)(\text{CO})_5$  (**3**) ( $\text{CDCl}_3$ )



**Figure S3:**  $\text{Fe}_2(\text{butyl-tethered-mabdt-PPh}_2)(\text{CO})_5$  (**4**) ( $\text{CD}_2\text{Cl}_2$ )

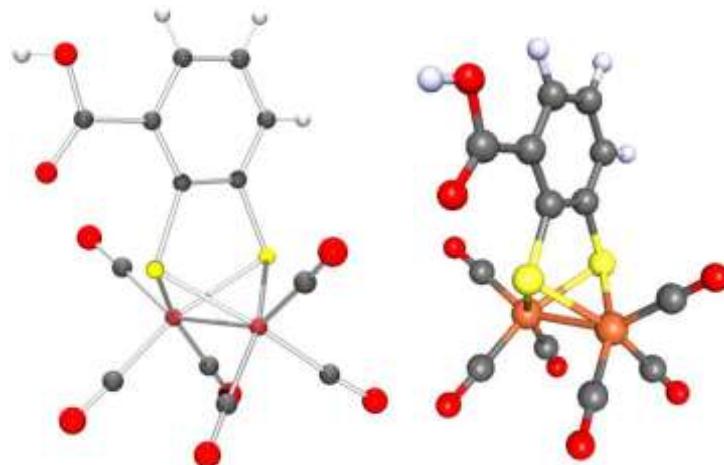
### Electrochemical data



**Figure S4:** Cyclic voltammogram of complex **3** in  $0.1 \text{ M TBAPF}_6$  in  $\text{MeCN}$ ,  $1 \text{ mM FeFe}$ ,  $100 \text{ mV/s}$  scan rate.

## DFT calculations

The geometry optimized structure of the ground state is well in accordance with X-ray structure (view table S1).

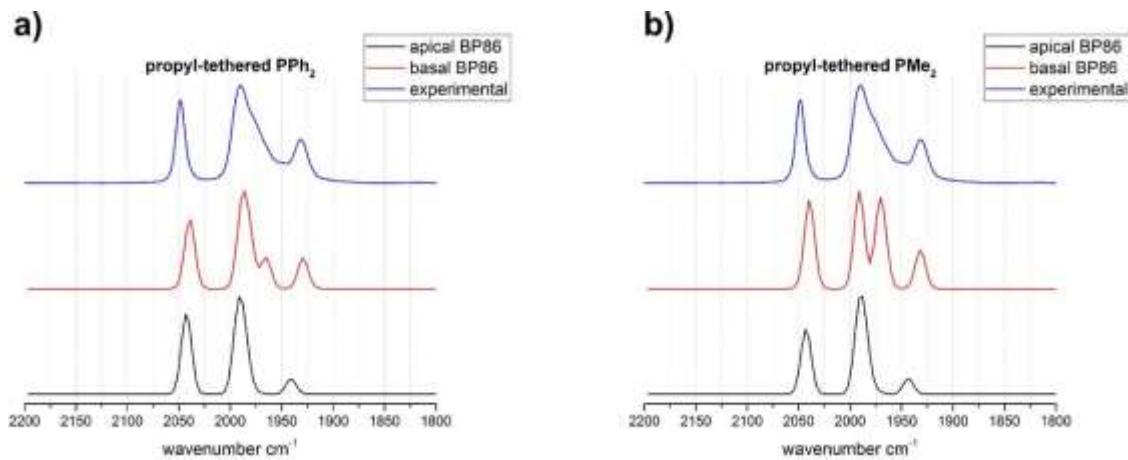


**Figure S5.** Left: Crystal structure of  $\text{Fe}_2(\text{mcbdt})(\text{CO})_6$  and Right: Geometry optimized structure.

**Table S1.** Comparison of computed ground state structure with X-ray crystallographic data<sup>8</sup>

bond/angle <sup>a</sup>	X-ray <sup>b</sup>	DFT (this study)
Fe-Fe	2.4986(8)	2.5200
Fe1-S1	2.2622(10)	2.2977
Fe1-S2	2.2636(10)	2.2844
S-C	1.781(4)	1.7942
Fe-C <sup>d</sup>	1.802(4)	1.7852
C-O <sup>d</sup>	1.135(5)	1.1548
S-Fe-S <sup>e</sup>	81.15(4)	80.29
Fe-S-Fe <sup>e</sup>	56.493(3)	60.74
S-Fe-C <sub>a</sub> <sup>e</sup>		102.12
C <sub>a</sub> -Fe-C <sub>b</sub> <sup>e</sup>		98.47
C <sub>b</sub> -Fe-C <sub>b</sub> <sup>e</sup>		90.82

<sup>a</sup>Bond lengths, Å; angles, degrees. <sup>b</sup>Experimental geometric parameters are averaged to  $\text{C}_{2v}$  symmetry for the molecule.  $\text{C}_a$  refers to the apical COs in the local square-based pyramid geometry of each Fe, and  $\text{C}_b$  refers to the basal COs. <sup>c</sup>See Experimental Section for details of the DFT method. <sup>d</sup>Averaged over all COs. <sup>e</sup>Averaged over all angles.



**Figure S6.** Calculated IR spectra for the propyl tethered complex with a)  $\text{PPh}_2$  and b)  $\text{PMe}_2$ . Black spectra represent the vibrational frequencies for the apical structure, red spectra the basal structure. In blue is the experimental spectrum given for complex 1.

**Table S2.** Lowest energies for all calculated complexes with  $\text{PMe}_2$  (def2-TZVP).

Chain length	functional	substitution	energy (Hartree)
ethyl	RP86	apical	-4790.895878096
		basal	n.a.
	R31 VP	apical	-4789.227840390
		basal	n.a.
	PWPB95	apical	-4789.428088305176
		basal	n.a.
propyl	DLPNO-CCSD(T)	apical	-4785.064883914102
		basal	n.a.
	RP86	apical	-4830.228996212
		basal	-4830.226841589
	R31 VP	apical	-4828.5334390690
		basal	-4828.5285008480
butyl	PWPB95	apical	-4828.732726636371
		basal	-4828.731521479146
	DLPNO-CCSD(T)	apical	-4824.314256716136
		basal	-4824.308893112706
	RP86	apical	-4869.549798348
		basal	-4869.554036785
pentyl	B3LYP	apical	-4867.8251102290
		basal	-4867.8282973880
	PWPB95	apical	-4868.023745984527
		basal	-4868.027199833423
	DLPNO-CCSD(T)	apical	-4863.548689363283
		basal	-4863.550183930706
	BP86	apical	-4908.890972635
		basal	-4908.887348598
	B3LYP	apical	-4907.1382768200
		basal	-4907.1325321590
	PWPB95	apical	-4907.336027496526
		basal	-4907.326278634325
	DLPNO-CCSD(T)	apical	-4902.804389701026
		basal	-4902.795260192346

reference	BP86	apical	-4792.122807997
		basal	-4792.121659707
	B3LYP	apical	-4790.4545319560
		basal	-4790.4515882420
	PWPB95	apical	-4790.640673854177
		basal	-4790.638805425159
	DLPNO-CCSD(T)	apical	-4786.284690443872
		basal	-4786.281398834074
propyl- $\mu$ -H	BP86	apical	-4830.593638823
		basal	-4830.592008635
	B3LYP	apical	-4828.912642687
		basal	-4828.910309877
propyl-H <sub>term.</sub>	BP86	apical	-4830.558741527
		basal	-4830.573430899
	B3LYP	apical	-4828.862578903
		basal	--4828.881269625

**Table2. Overview calculated structures (BP-86)**

