Electronic Supplementary Information for

Carbon Dioxide in the Cage: Manganese Metal-Organic Frameworks for High Performance CO₂ Electrodes in Li-CO₂ Batteries

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Fig. S1 Discharge-charge voltage curves of CNT electrodes cycled at (a) 50 mA g^{-1} , (b) 100 mA g^{-1} , and (c) 200 mA g^{-1} with a capacity limit of 1000 mA h g^{-1} .



Fig. S2 Crystal structures of (a) $Mn(C_2H_2N_3)_2$, (b) $Mn(HCOO)_2$, (c) $MnCO_3$ and (d) MnO (left) and their corresponding discharge-charge voltage curves at 50 mA g⁻¹ with a capacity limit of 1000 mA h g⁻¹ (right).



Fig. S3 Discharge-charge voltage curves of $Mn_2(dobdc)$ electrodes cycled at (a) 100 mA g⁻¹ and (b) 200 mA g⁻¹ with a capacity limit of 1000 mA h g⁻¹.



Fig. S4 Discharge-charge voltage curves of $Mn(HCOO)_2$ electrodes cycled at (a) 100 mA g⁻¹ and (b) 200 mA g⁻¹ with a capacity limit of 1000 mA h g⁻¹.



Fig. S5 Discharge-charge voltage curves of $MnCO_3$ electrodes cycled at (a) 100 mA g⁻¹ and

(b) 200 mA g^{-1} with a capacity limit of 1000 mA h g^{-1} .



Fig. S6 CO_2 adsorption isotherms of $Mn_2(dobdc)$, $Mn(HCOO)_2$, $MnCO_3$ and CNT at (a) 273 K and (b) 283 K. (c) N_2 adsorption isotherms of $Mn_2(dobdc)$ and CNT at 77 K.



Fig. S7 PXRD patterns of a fresh, a discharged and a recharged $Mn(HCOO)_2$ electrode. The cells were operated at 50 mA g⁻¹ with a capacity limit of 1000 mA h g⁻¹. The enlarged figure highlights the formation and decomposition of Li₂CO₃.



Fig. S8 PXRD patterns of a fresh, a discharged and a recharged $Ni_2(dobdc)$ electrode. The cells were operated at 50 mA g⁻¹ with a capacity limit of 1000 mA h g⁻¹.



Fig. S9 PXRD patterns of a fresh, a discharged and a recharged $Mn_2(dobdc)$ electrode. The cells were operated at 50 mA g⁻¹ with a capacity limit of 1000 mA h g⁻¹.



Fig. S10 PXRD patterns of a fresh, a discharged and a recharged $Co_2(dobdc)$ electrode. The cells were operated at 50 mA g^{-1} with a capacity limit of 1000 mA h g^{-1} .



Fig. S11 PXRD patterns of a fresh, a discharged and a recharged $MnCO_3$ electrode. The cells were operated at 50 mA g⁻¹ with a capacity limit of 1000 mA h g⁻¹.



Fig. S12 (a) Typical discharge-charge voltage curves with plots showing the capacity at which an EIS experiment was conducted. (b) Equivalent circuit model for fitting the Nyquist plots. Nyquist plots of (c) CNT, (d) $Mn_2(dobdc)$, (e) $Mn(HCOO)_2$, and (f) $MnCO_3$ electrodes at different discharge and charge states (F: fresh electrode; D1-D3: discharged to 200, 500 and 1000 mA h g⁻¹, respectively; C1-C3: recharged to 200, 500 and 1000 mA h g⁻¹, respectively).



Fig. S13 (a) Discharge-charge voltage curves of a CNT electrode. (b-f) SEM images of CNT electrodes at different states during operation at 50 mA g^{-1} (F: fresh electrode; D1-D3: discharged to 200, 500 and 1000 mA h g^{-1} , respectively; R: recharged to 1000 mA h g^{-1}).



Fig. S14 (a) Discharge-charge voltage curves of a $Mn_2(dobdc)$ electrode. (b-f) SEM images of $Mn_2(dobdc)$ electrodes at different states during operation at 50 mA g⁻¹ (F: fresh electrode; D1-D3: discharged to 200, 500 and 1000 mA h g⁻¹, respectively; R: recharged to 1000 mA h g⁻¹).



Fig. S15 (a) Discharge-charge voltage curves of a $Mn(HCOO)_2$ electrode. (b-f) SEM images of $Mn(HCOO)_2$ electrodes at different states during operation at 50 mA g⁻¹ (F: fresh electrode; D1-D3: discharged to 200, 500 and 1000 mA h g⁻¹, respectively; R: recharged to 1000 mA h g⁻¹).



Fig. S16 (a) Discharge-charge voltage curves of a MnCO₃ electrode. (b-f) SEM images of MnCO₃ electrodes at different states during operation at 50 mA g^{-1} (F: fresh electrode; D1-D3: discharged to 200, 500 and 1000 mA h g^{-1} , respectively; R: recharged to 1000 mA h g^{-1}).



Fig. S17 (a) PXRD patterns, (b) and (c) SEM images of $Mn_2(dobdc)@Ni$ foam.



Fig. S18 Schematics of the DEMS system.



Fig. S19 Discharge-charge voltage curves of a $Mn_2(dobdc)$ electrode tested in Ar atmosphere at 50 mA g⁻¹.



Fig. S20 Photographs of a customized Swagelok cell for *in situ* DEMS.

CO ₂ electrode	Full discharge capacity (mA h g ⁻¹) ^[a]	Average charge potential (V) ^[b]	Cycle life	Ref.	
Mn ₂ (dobdc)	18022 (50 mA g ⁻¹)	nA g^{-1}) 3.96 (50 mA g^{-1}) 50 (200 mA g^{-1})			
Mn(HCOO) ₂	15510 (50 mA g ⁻¹)	4.00 (100 mA g ⁻¹)	50 (200 mA g ⁻¹)	I IIS WORK	
Ketjen-black	<100 (0.1 mA cm ⁻²)	N/A	N/A	1	
	1808 (30 mA g ⁻¹)	4.26 (30 mA g ⁻¹)	9 (30 mA g ⁻¹)	2	
High surface area carbon	~800 (0.05 mA cm ⁻²)	N/A	N/A	3	
	~3000 (0.05 mA cm ⁻²) ^[c]	N/A	N/A		
CNTs	8379 (50 mA g ⁻¹)	4.3 (50 mA g^{-1})	29 (50 mA g^{-1})	4	
Graphene	14774 (50 mA g ⁻¹)	4.07 (50 mA g ⁻¹)	$0 \text{ mA } g^{-1}$) 20 (50 mA g^{-1})		
B,N-codoped holey graphene	16033 (300 mA g ⁻¹)	3.97 (100 mA g ⁻¹)	$mA g^{-1}$) 200 (1000 mA g^{-1})		
Ru@Super P	8229 (100 mA g ⁻¹)	$3.93 (100 \text{ mA g}^{-1})$	70 (100 mA g^{-1})	7	
Mo ₂ C@CNT ^[d]	$\sim 360 (\sim 5 \text{ mA g}^{-1})$	$\sim 3.5 (\sim 5 \text{ mA g}^{-1})$	40 (~5 mA g ⁻¹)	8	
Ni-NG	17625 (100 mA g^{-1})	4.2 (100 mA g^{-1})	101 (100 mA g ⁻¹)	9	

Table S1. Summary of performance of existing CO₂ electrodes working in pure CO₂.

[a] Full discharge capacity is the discharge capacity toward a cutoff potential of 2.0 V under the given current density. All the capacity and current density values, unless specified otherwise, are normalized by the total mass of catalyst and conductive agent.

[b] Average charge potential is the average of potential during a charge process with a capacity limit of 1000 mA h g^{-1} .

[c] This value was obtained at 100 °C.

[d] The capacity and current density values of $Mo_2C@CNT$ are calculated based on the loading mass of 4 mg on the electrode. The capacity limit for deriving the average charge potential is ~30 mA h g⁻¹.

CO_2 electrode –				$R_{\rm ct}\left(\Omega\right)$			
	F	D1	D2	D3	C1	C2	C3
CNT	276	295	1474	3397	1667	683	623
Mn ₂ (dobdc)	279	291	712	1115	650	458	434
Mn(HCOO) ₂	281	383	672	1326	881	535	507
MnCO ₃	308	329	745	1241	1072	542	414

Table S2. Charge transfer resistance (R_{ct}) of different CO₂ electrodes at different discharge and charge states (Fitting results of **Fig. S12**).

F: fresh electrode; D1-D3: discharged to 200, 500 and 1000 mA h g^{-1} , respectively; C1-C3: recharged to 200, 500 and 1000 mA h g^{-1} , respectively.

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