

**SUPPORTING INFORMATION**

**Nature-Based Catalyst for Visible-Light-Driven Photocatalytic  
CO<sub>2</sub> Reduction**

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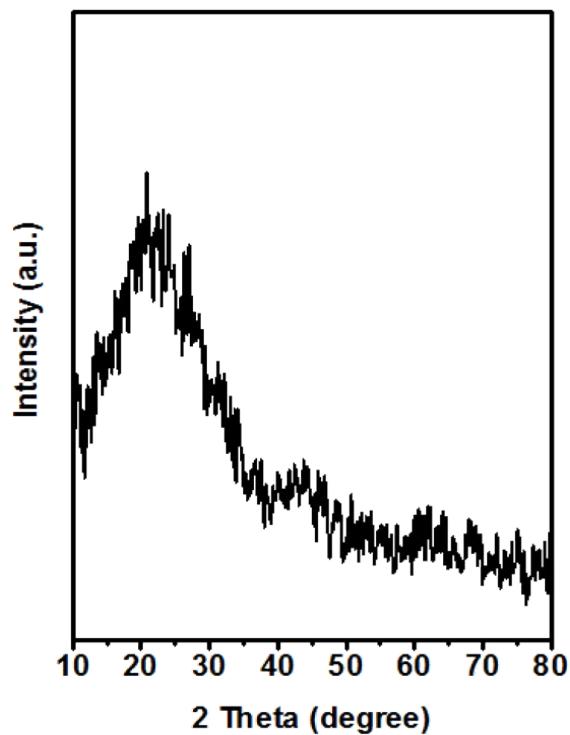
**Table S1** The photocatalytic CO<sub>2</sub> reduction efficiency of various catalysts.

Photocatalyst	Light source	CO evolution rate (μmol h <sup>-1</sup> g <sup>-1</sup> )	CH <sub>4</sub> evolution rate (μmol h <sup>-1</sup> g <sup>-1</sup> )	Reference
Bi <sub>4</sub> O <sub>5</sub> I <sub>2</sub> /g-C <sub>3</sub> N <sub>4</sub>	300 W Xenon lamp (with λ > 400 nm filter)	45.6	6	1
LaPO <sub>4</sub> /g-C <sub>3</sub> N <sub>4</sub>	300W Xenon lamp	14.4	N/A	2
SnO <sub>2</sub> /g-C <sub>3</sub> N <sub>4</sub>	500W Xenon lamp	18	2	3
Pt/B <sub>4</sub> C/g-C <sub>3</sub> N <sub>4</sub>	Visible light (405–723 nm)	N/A	0.84	4
ZnO/g-C <sub>3</sub> N <sub>4</sub>	500 W Xenon lamp (with a 420 nm cut filter)	29	4	5

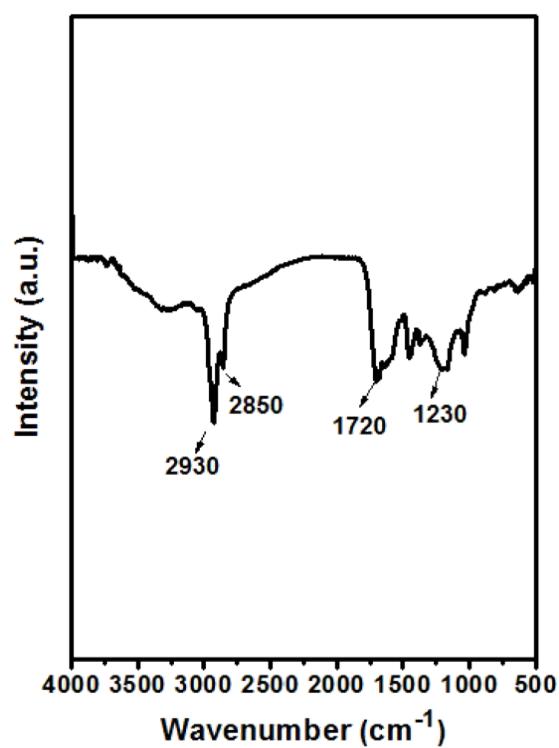
NA= not available

**Table S2** The photocatalytic CO<sub>2</sub> reduction efficiency and the corresponding QE of g-C<sub>3</sub>N<sub>4</sub> based photocatalysts

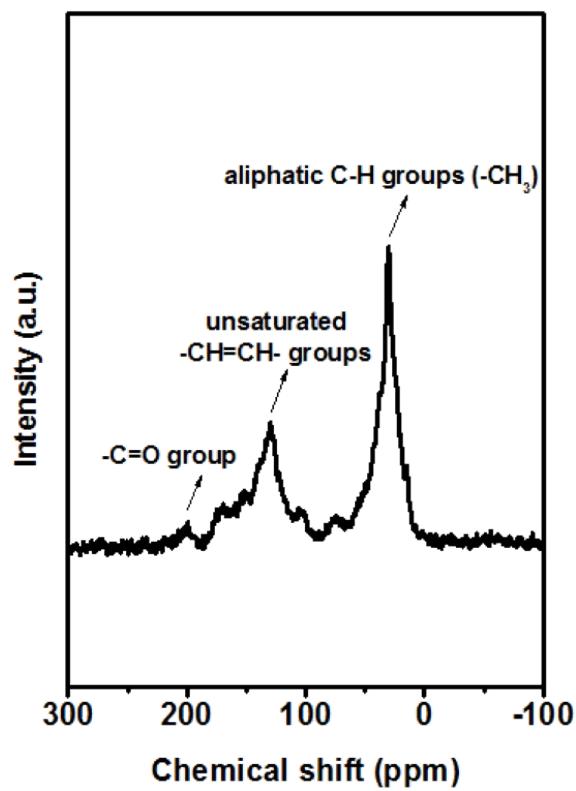
Photocatalyst	Light source	CO evolution rate (μmol g <sup>-1</sup> h <sup>-1</sup> )	QE (%)	Reference
KOH/g-C <sub>3</sub> N <sub>4</sub>	300 W Xeon lamp (PLS-SXE300UV)	2.0	N/A	6
Co-porphyrin/g-C <sub>3</sub> N <sub>4</sub>	300W Xenon lamp (with a UV-cut filter)	17.0	0.8 (420 nm)	7
N-TiO <sub>2</sub> /g-C <sub>3</sub> N <sub>4</sub>	300 W Xenon lamp (Perfect Light)	12.2	N/A	8
LaPO <sub>4</sub> /g-C <sub>3</sub> N <sub>4</sub>	300W Xenon lamp (Aulight CEL-HX)	14.4	N/A	9
BiOI/g-C <sub>3</sub> N <sub>4</sub>	300 W Xenon lamp (with a UV cutoff filter λ > 400 nm)	3.45	N/A	10
rGO/g-C <sub>3</sub> N <sub>4</sub>	15 W energy-saving daylight lamp (Philips)	N/A	0.56 (420 nm)	11
Co-ZIF-9/g-C <sub>3</sub> N <sub>4</sub>	Xenon lamp (with a 420 nm cutoff filter)	10.4	0.9 (420 nm)	12
Bi <sub>2</sub> WO <sub>6</sub> /g-C <sub>3</sub> N <sub>4</sub>	300 W Xenon lamp (Aulight CEL-HX)	5.19	N/A	13
N-TiO <sub>2</sub> /g-C <sub>3</sub> N <sub>4</sub>	400 W Xenon lamp	12.28	N/A	14



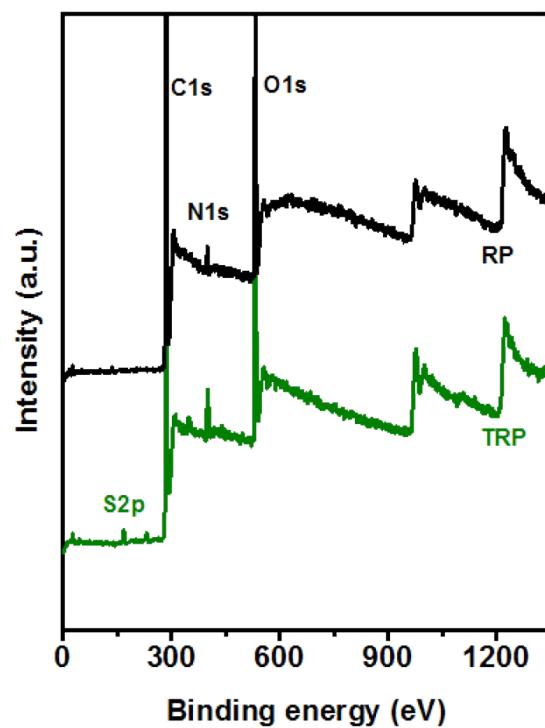
**Figure S1.** XRD pattern of TRP catalyst.



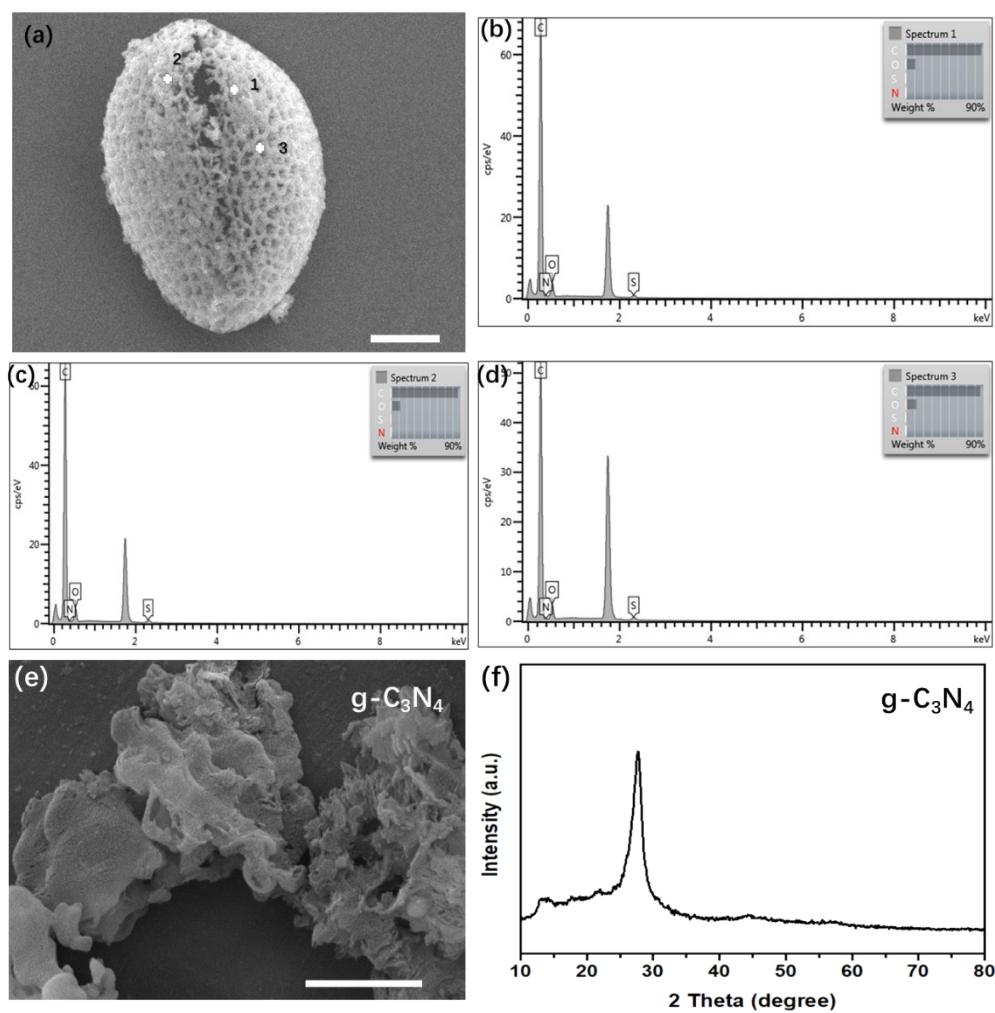
**Figure S2.** FT-IR spectrum of TRP catalyst.



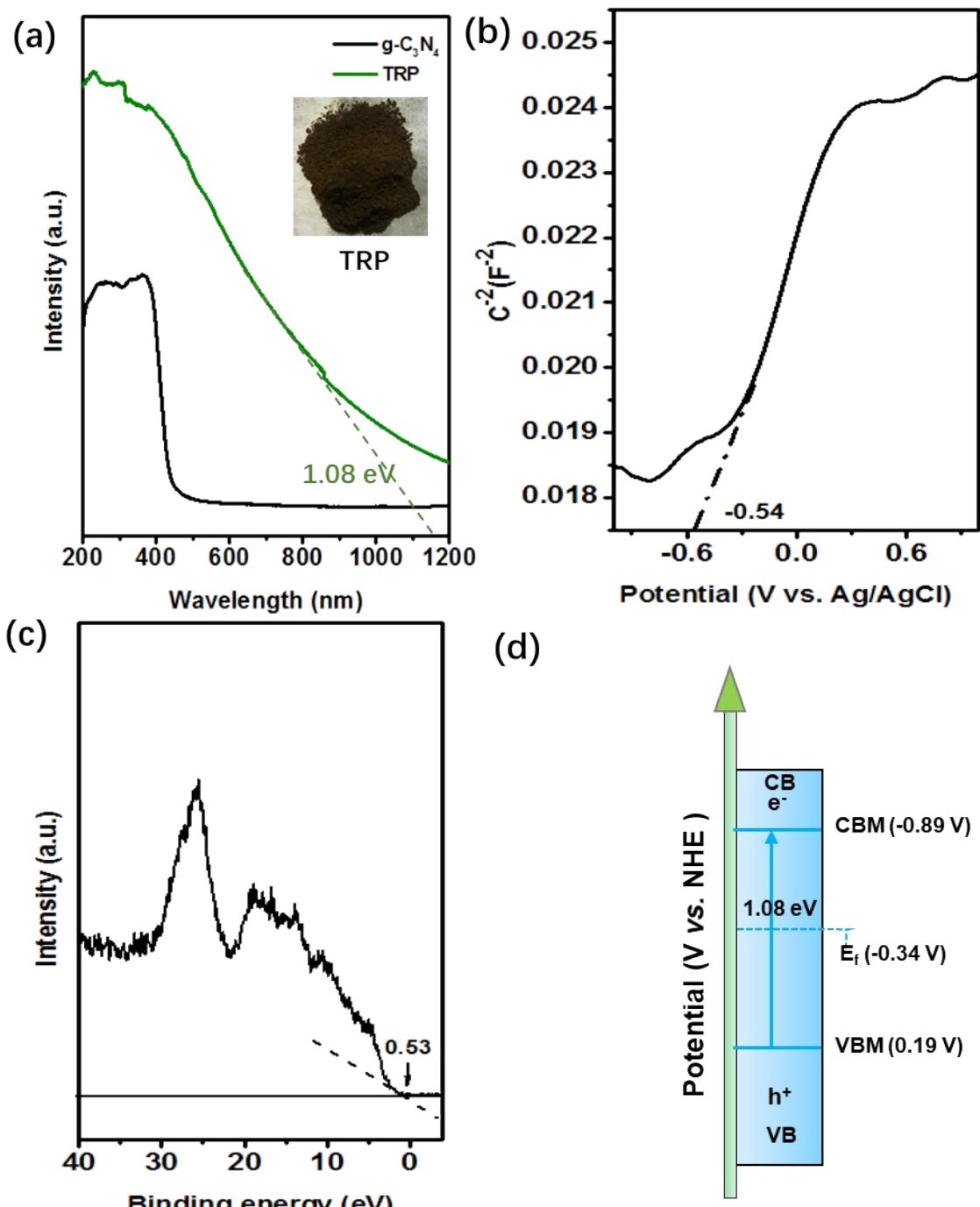
**Figure S3.** <sup>13</sup>C Solid-state NMR spectrum of TRP catalyst.



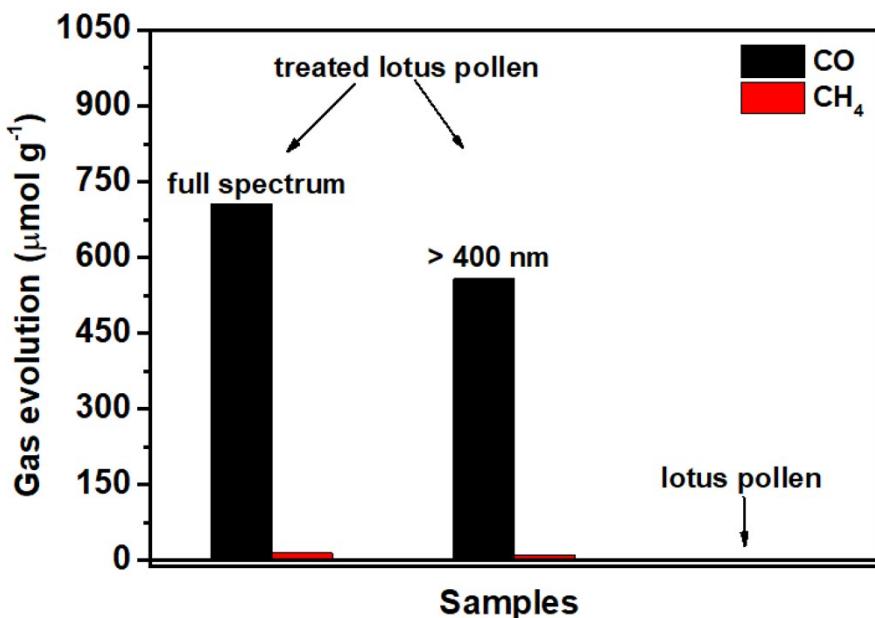
**Figure S4.** XPS spectra of RP and TRP catalyst.



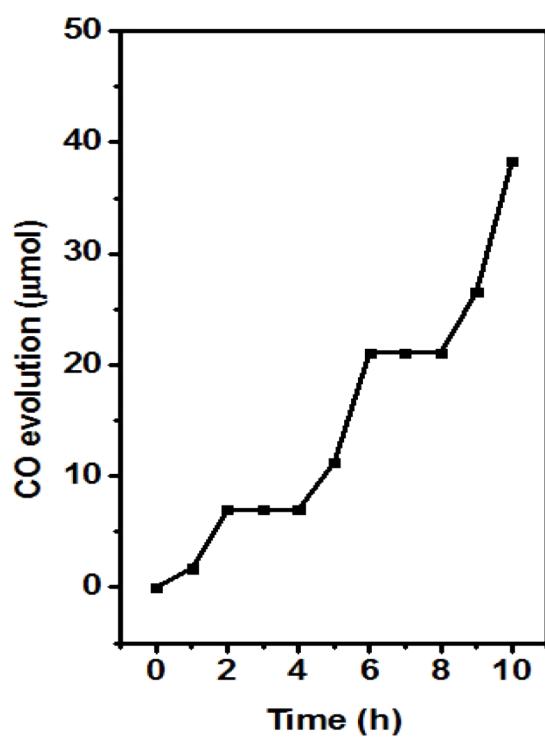
**Figure S5.** SEM image (a) and corresponding EDX spectra (b-d) of TRP catalyst, SEM image (e) and XRD pattern (f) of  $\text{g-C}_3\text{N}_4$ . Scale bar: 5  $\mu\text{m}$  (a), 1  $\mu\text{m}$  (e)



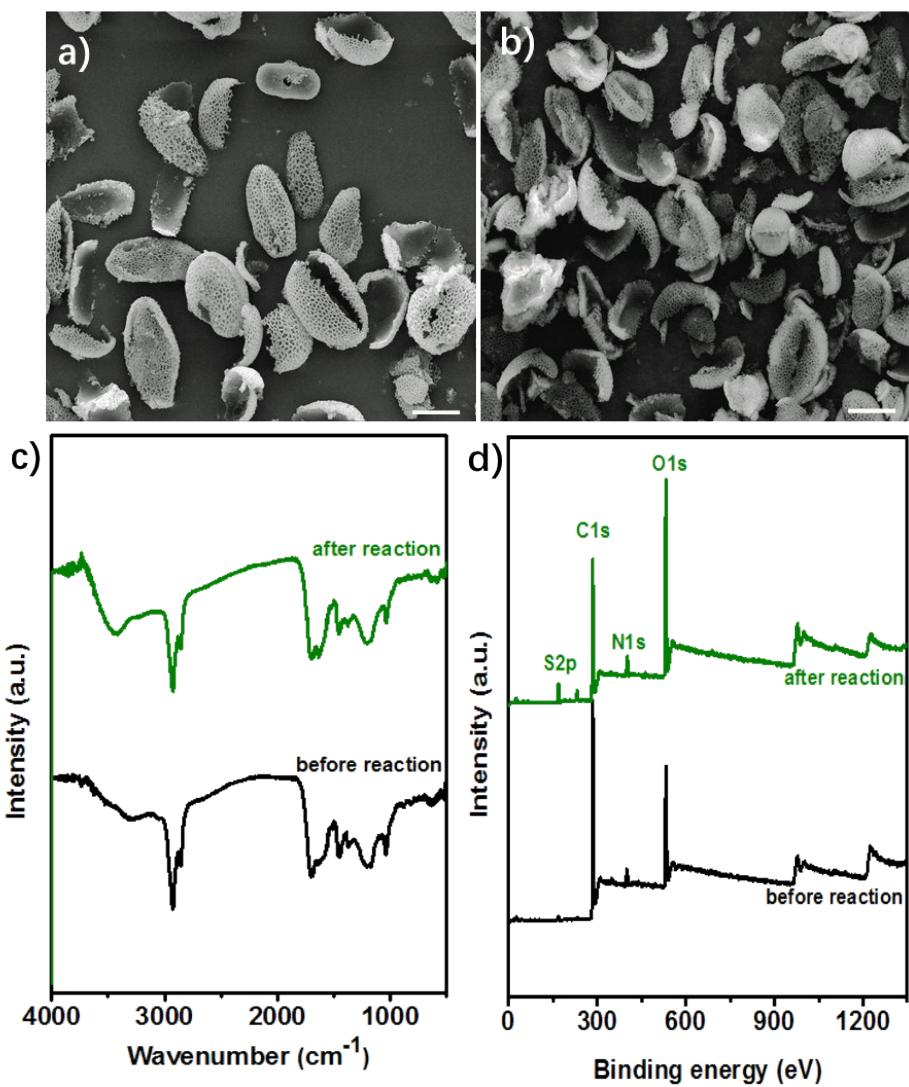
**Figure S6.** UV-vis diffuse reflectance spectra of  $\text{g-C}_3\text{N}_4$  and TRP (inset: Photograph of TRP sample) (a), Mott-Schottky plot (b), valance band XPS spectrum (c) and energy band diagram (d) of TRP catalyst.



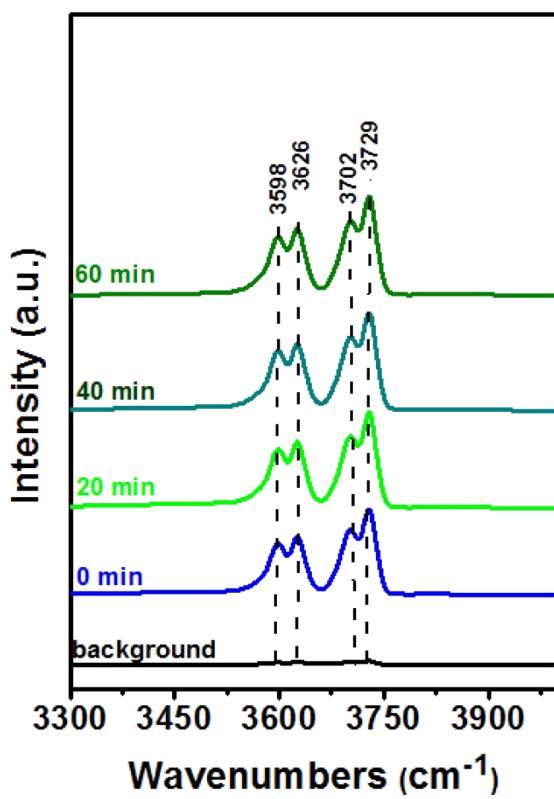
**Figure S7.** Photocatalytic CO<sub>2</sub> reduction activity of the treated lotus pollen and lotus pollen.



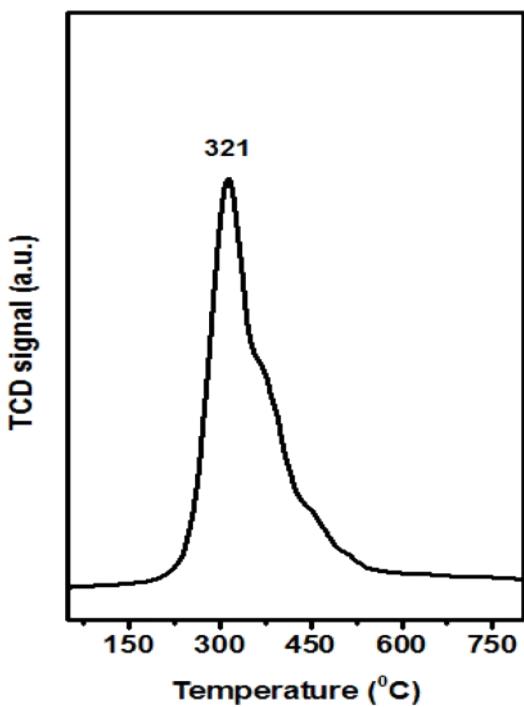
**Figure S8.** The possibility of recycling ability of TRP catalyst.  
(Reaction condition: 15 mg TRP, 20 mL water)



**Figure S9.** SEM images of TRP before (a) and after (b) reaction, FT-IR (c) and XPS (d) spectra of TRP catalyst before and after reaction. Scale bar (a and b: 10  $\mu\text{m}$ )



**Figure S10.** *In situ* FT-IR spectra for co-adsorption of a mixture of  $\text{CO}_2$  and  $\text{H}_2\text{O}$  vapor on the TRP catalyst in the dark and under visible light (LED lamp,  $\lambda = 420 \text{ nm}$ ) irradiation for 20, 40, 60 min, respectively.



**Figure S11.**  $\text{CO}_2$ -TPD spectrum of TRP catalyst.

## References

1. Y. Bai et al. *Appl. Catal. B: Environ.* **2016**, *194*, 98-104.
2. M. Li et al. *Appl. Catal. B: Environ.* **2017**, *201*, 629-635.
3. Y. He et al. *Solar Energy Mater. Solar Cells*, **2015**, *137*, 175-184.
4. X. Zhang et al. *J. Colloid Interf. Sci.*, **2016**, *464*, 89-95.
5. W. Yu et al. *J. Mater. Chem. A*, **2015**, *3*, 19936-19947.
6. Z. Sun et al. *Appl. Catal. B: Environ.* **2017**, *216*, 146.
7. G. Zhao et al. *Appl. Catal. B: Environ.* **2017**, *200*, 141.
8. S. Zhou et al. *Appl. Catal. B: Environ.* **2014**, *20*, 158.
9. M. Li et al. *Appl. Catal. B: Environ.* **2017**, *201*, 629.
10. J. Wang et al. *ACS Appl. Mater. Interf.* **2016**, *8*, 3765.
11. W. J. Ong et al. *Nano Energy* **2015**, *13*, 757.
12. S. Wang et al. *Phys. Chem. Chem. Phys.* **2014**, *16*, 14656.
13. M. Li et al. *J. Mater. Chem. A* **2015**, *3*, 5189.
14. W. Wang et al. *J. Am. Chem. Soc.* **2012**, *134*, 11276.