

**Supplementary information for**

**Organosulfates in aerosols downwind of an urban region in central  
Amazon**

Marianne Glasius<sup>1</sup>, Mads S. Bering<sup>1</sup>, Lindsay D. Yee<sup>2</sup>, Suzane S. de Sá<sup>3</sup>, Gabriel Isaacman-VanWertz<sup>2\*</sup>, Rebecca A. Wernis<sup>4</sup>, Henrique M.J. Barbosa<sup>5</sup>, M. Lizabeth Alexander<sup>6</sup>, Brett B. Palm<sup>7\*\*</sup>, Weiwei Hu<sup>7</sup>, Pedro Campuzano-Jost<sup>7</sup>, Douglas A. Day<sup>7</sup>, Jose L. Jimenez<sup>7</sup>, M. Shrivastava<sup>6</sup>, Scot T. Martin<sup>3</sup>, and Allen H. Goldstein<sup>2</sup>

<sup>1</sup>Aarhus University, Dept. of Chemistry, 8000 Aarhus C, Denmark

<sup>2</sup>University of California, Berkeley, Dept. of Environmental Science, Policy, and Management, Berkeley, 94720, USA

<sup>3</sup> Harvard University, School of Engineering and Applied Sciences, Cambridge, 02138, USA

<sup>4</sup>University of California, Berkeley, Dept. of Civil and Environmental Engineering, Berkeley, 94720, USA

<sup>5</sup>University of Sao Paulo, Institute of Physics, 05508-090, Sao Paulo, Brazil

<sup>6</sup>Pacific Northwest National Laboratory, Environmental Molecular Sciences Laboratory, Richland, 99352, USA

<sup>7</sup>University of Colorado, Boulder, Dept. of Chemistry & Biochemistry and Cooperative Institute for Research in Environmental Sciences (CIRES), Boulder, 80309, USA

\*Now at Virginia Tech, Dept. of Civil and Environmental Engineering, Blacksburg, 24061, USA

\*\* Now at: University of Washington, Dept. of Atmospheric Science, Seattle, WA, 98155, USA

Table S1. Overview of suggested molecular structures and precursors of organosulfates detected in this study.

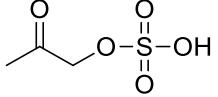
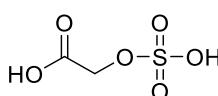
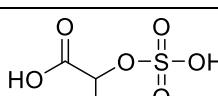
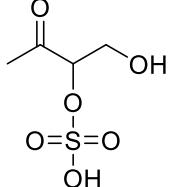
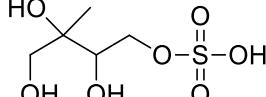
Compound	Molecular formula	Suggested structure	Suggested precursor
Acetone sulfate (OS 154)	C <sub>3</sub> H <sub>6</sub> O <sub>5</sub> S		hydroxyacetone <sup>1</sup>
Acetic acid sulfate (OS 156)	C <sub>2</sub> H <sub>4</sub> O <sub>6</sub> S		hydroxyacetic acid (glycolic acid) <sup>2,3</sup>
Propanoic acid 2-sulfate (OS 170)	C <sub>3</sub> H <sub>6</sub> O <sub>6</sub> S		2-hydroxy propanoic acid <sup>2,3</sup>
OS 184 (hydroxybutanone-sulfate)	C <sub>4</sub> H <sub>8</sub> O <sub>6</sub> S		methyl vinyl ketone <sup>2,4</sup>
OS 200	C <sub>4</sub> H <sub>8</sub> O <sub>7</sub> S	Multiple isomers possible	2-methylglyceric acid <sup>1</sup> , MBO <sup>5</sup>
OS 212	C <sub>5</sub> H <sub>8</sub> O <sub>7</sub> S	Multiple isomers possible	isoprene <sup>6</sup>
OS 214	C <sub>5</sub> H <sub>10</sub> O <sub>7</sub> S	Multiple isomers possible	isoprene <sup>6,7</sup>
IEPOX OS (1,2,4-hydroxy-2-methylbutyl-3-sulfate, OS 216)	C <sub>5</sub> H <sub>12</sub> O <sub>7</sub> S		isoprene <sup>1,6</sup>

Table S2. Correlation coefficients ( $R^2$ ) of individual organosulfates, sulfate and organic PM<sub>1</sub> (OA) (measured by AMS) during wet season (a) and dry season (b). Correlations  $> 0.75$  are highlighted in bold font.

<b>a. Wet season (n=28)</b>	<b>OS 154</b>	<b>OS 184</b>	<b>OS 200</b>	<b>OS 212</b>	<b>OS 214</b>	<b>OS 216</b>
<b>OS 154</b>	1	<b>0.92</b>	0.14	<b>0.85</b>	<b>0.91</b>	<b>0.84</b>
<b>OS 184</b>	<b>0.92</b>	1	0.24	<b>0.85</b>	<b>0.84</b>	<b>0.80</b>
<b>OS 200</b>	0.14	0.24	1	0.20	0.09	0.16
<b>OS 212</b>	<b>0.85</b>	<b>0.85</b>	0.20	1	0.73	0.65
<b>OS 214</b>	<b>0.91</b>	<b>0.84</b>	0.09	0.73	1	<b>0.87</b>
<b>OS 216</b>	<b>0.84</b>	<b>0.80</b>	0.16	0.65	<b>0.87</b>	1
<b>Sulfate</b>	0.70	0.71	0.07	0.64	0.71	0.50
<b>Organic</b>	0.50	0.41	0.01	0.48	0.37	0.29

<b>b. Dry season (n=70)</b>	<b>OS 154</b>	<b>OS 156</b>	<b>OS 170</b>	<b>OS 184</b>	<b>OS 200</b>	<b>OS 214</b>	<b>OS 216</b>
<b>OS 154</b>	1	<b>0.78</b>	0.56	0.71	<b>0.87</b>	<b>0.83</b>	0.45
<b>OS 156</b>	<b>0.78</b>	1	0.64	0.61	<b>0.81</b>	<b>0.76</b>	0.39
<b>OS 170</b>	0.56	0.64	1	0.61	0.58	0.41	0.11
<b>OS 184</b>	0.71	0.61	0.61	1	0.63	0.61	0.29
<b>OS 200</b>	<b>0.87</b>	<b>0.81</b>	0.58	0.63	1	<b>0.77</b>	0.36
<b>OS 214</b>	<b>0.83</b>	<b>0.76</b>	0.41	0.61	<b>0.77</b>	1	0.67
<b>OS 216</b>	0.45	0.39	0.11	0.29	0.36	0.67	1
<b>Sulfate</b>	0.25	0.24	0.08	0.21	0.16	0.43	0.69
<b>Organic</b>	0.17	0.23	0.25	0.11	0.15	0.12	0.06

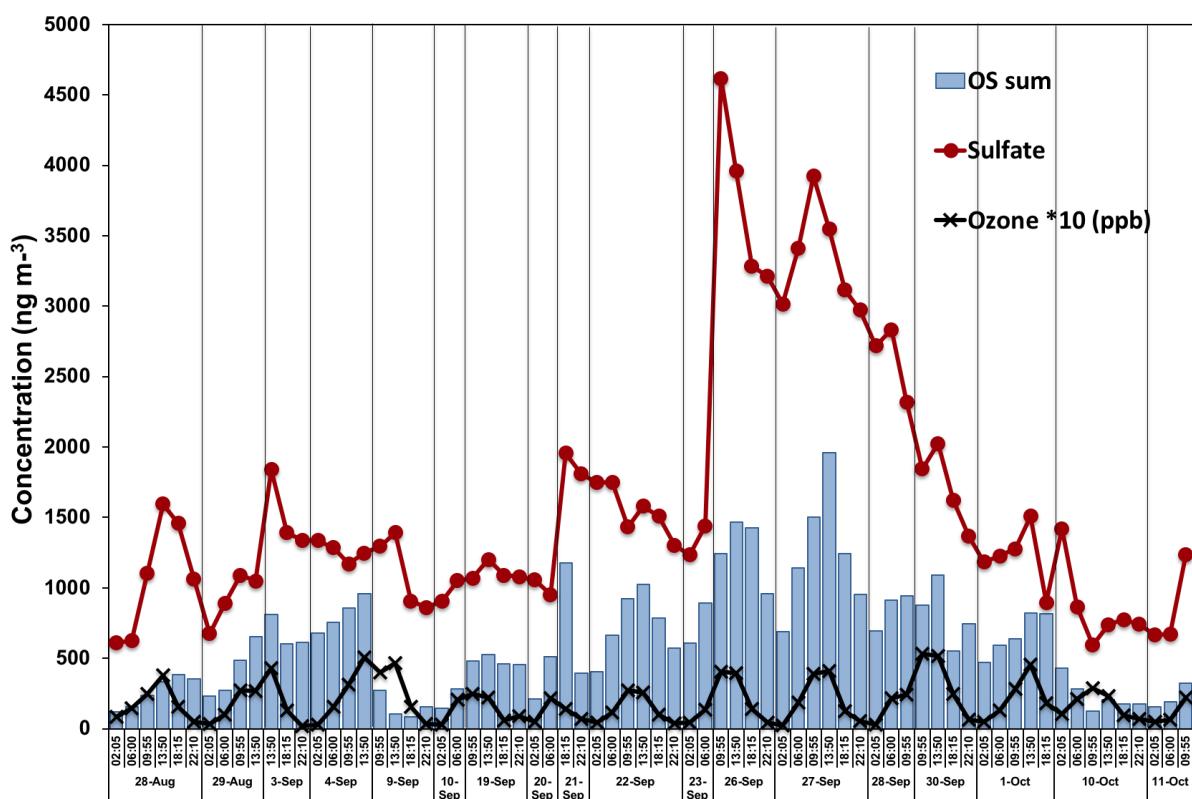
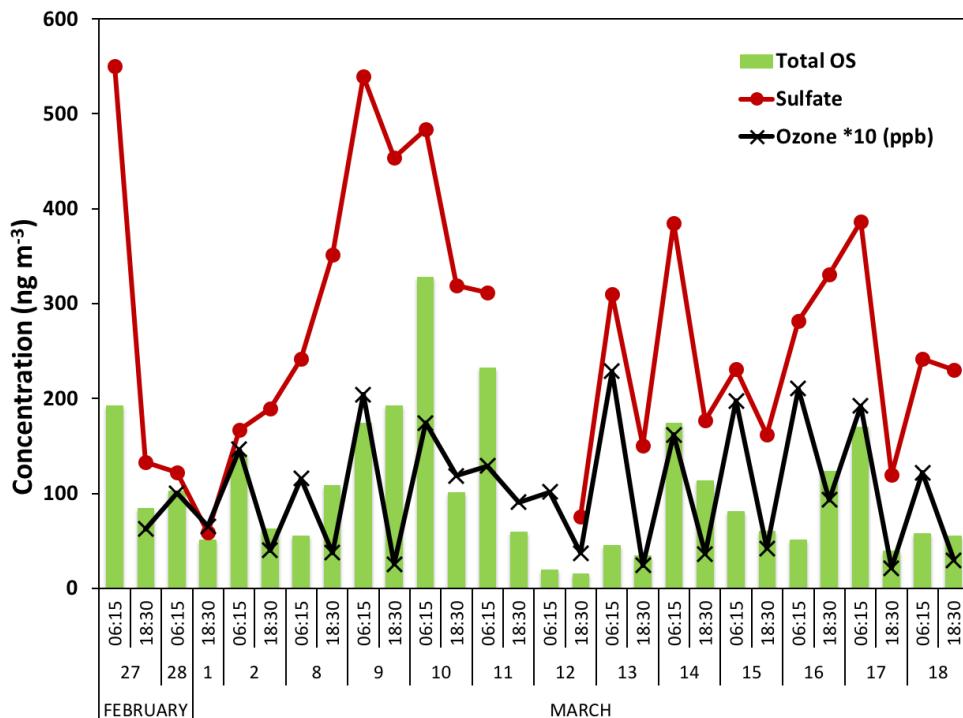


Figure S1. Concentrations of OS in aerosol samples collected during the wet season (upper panel) and dry season (lower panel), as well as sulfate and ozone. Note that the timeline is not continuous, and lines have been added to guide the eye.

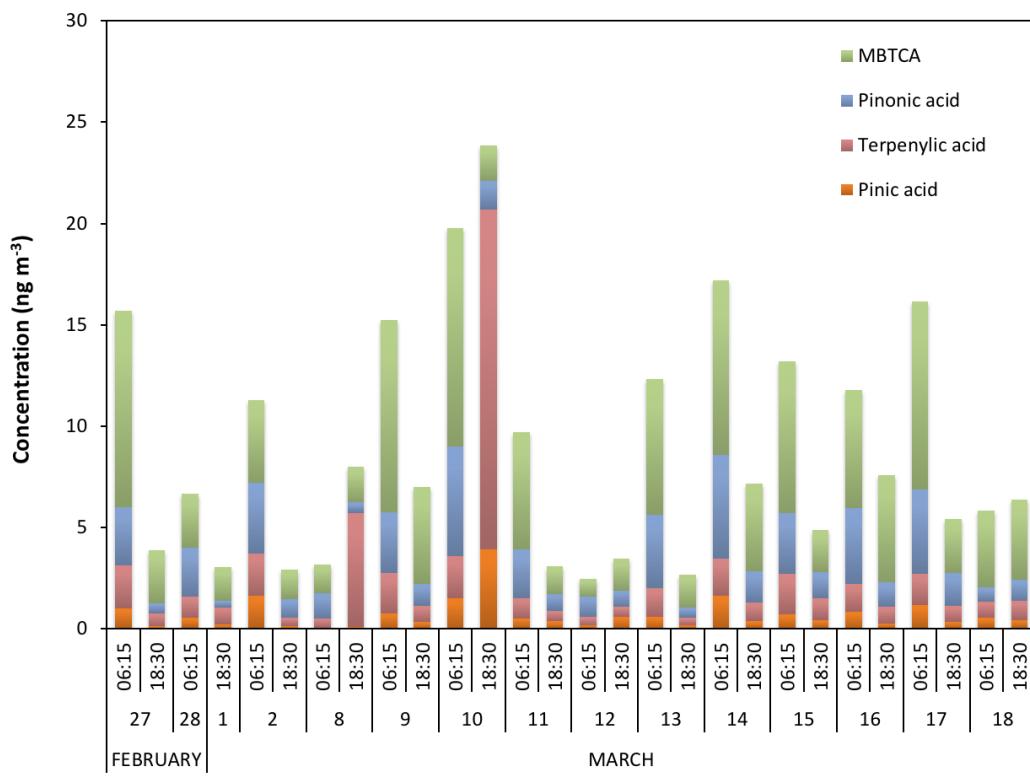
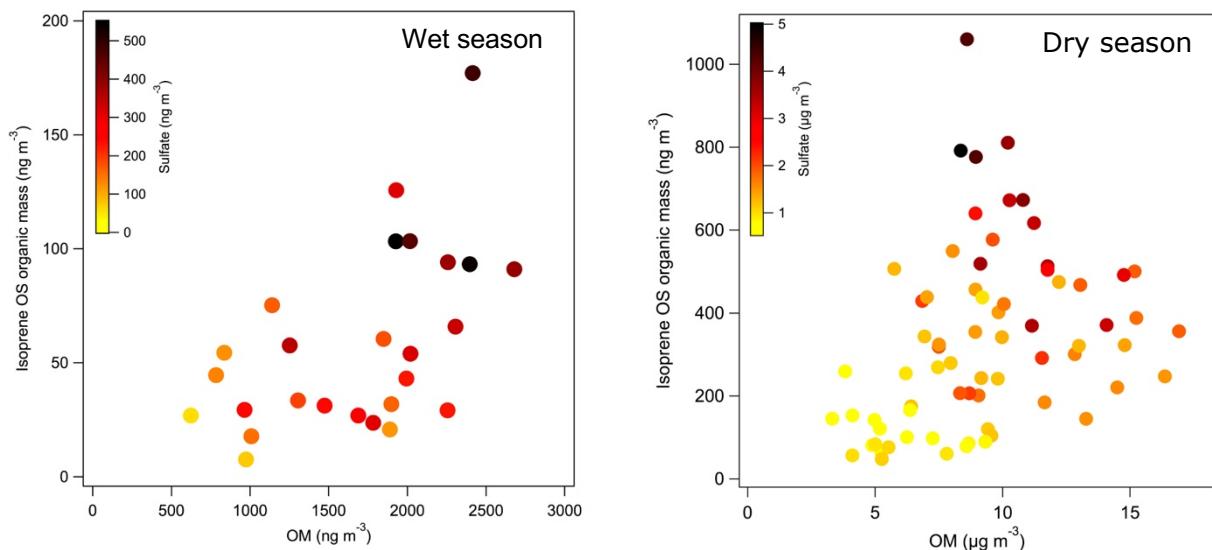


Figure S2. Concentrations of carboxylic acids derived from  $\alpha$ -pinene and  $\beta$ -pinene in aerosol samples collected during the wet season. MBTCA is 3-methylbutane tricarboxylic acid.



## References for supplementary information

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