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- 1 Supporting Information for:
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3 Effects of Vertical Hydrodynamic Mixing on Photomineralization of Dissolved Organic
4 Carbon in Arctic Surface Waters

- 5
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- 68 River discharge and channel geometry data were collected from 8 gauging stations along the
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- 71



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- 83





87 CDOM ( $\Phi_{\lambda}$ ) for arctic streams, rivers, and lakes<sup>1, 7</sup>. Parameters *aCDOM*<sub> $\lambda$ </sub> and  $\Phi_{\lambda}$  exhibit an

88 exponential relationship with wavelength, with decadic slopes of -0.005 and -0.009, respectively.

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91



**Figure S4.** Water column depth (*H*) and water surface slope (*S*) as a function of flow discharge

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101 Figure S5. Reaction efficiency heatmap of (A) beaded streams under slow flow, (B) shallow

- 102 lakes, and (C) deep lakes. Natural variations of surface Damkohler number ( $d^*$ ) and
- 103 dimensionless light attenuation ( $p^*$ ) values for each type of systems are plotted in white lines. 104

### 106 SI-1. No mixing limitation when CDOM is the only light attenuating constituent.

107 When CDOM is the only light attenuating constituent, light does not necessarily decay exponentially with depth, especially when CDOM is nonhomogeneously distributed. At each 108 vertical position (y) in the water column, photon flux  $(Q_{\lambda}(y))$  depends on the total amount of

109 CDOM above that position<sup>8</sup>. 110

111 
$$Q_{\lambda}(y) = Q_{dso-\lambda} e^{-\int_{0}^{y} aCDOM_{\lambda}(y)dy}$$
(S1)

112 where  $Q_{dso-\lambda}$  is the photon flux at the water surface,  $aCDOM_{\lambda}(y)$  is the light attenuation 113 coefficient of CDOM, and  $\lambda$  is wavelength. The photomineralization rate profile (PM(y)) is the

114 product of the photo-lability of CDOM to photomineralization ( $\Phi_{\lambda}$ ), the light attenuation 115 coefficient of CDOM, and the photon flux<sup>1</sup>.

116 
$$PM(y) = \sum_{\lambda} \phi_{\lambda} aCDOM_{\lambda}(y) Q_{dso-\lambda} e^{-\int_{0}^{y} aCDOM_{\lambda}(y) dy}$$
(S2)

Using integration by parts and calculus theorem, the total attenuation of light across the 117 118 water column ( ${}^{\Delta Q_{\lambda}}$ ) can be derived as

119 
$$\Delta Q_{\lambda} = Q_{dso-\lambda} - Q_{dso-\lambda} e^{-\int_{0}^{H} aCDOM_{\lambda}(y)dy}$$
(S3)

120 where the exponential term on the right-hand side of (S3) is a function of the total CDOM

amount in the water column, so the total attenuation of light across the water column is the same 121 regardless of the distribution of CDOM. Similarly, the upscaled photomineralization rate over 122

123 the water column can be derived as

124 
$$\int_{0}^{H} PM(y) dy = \phi_{\lambda} Q_{dso-\lambda} \left( 1 - e^{-\int_{0}^{H} aCDOM_{\lambda}(y) dy} \right)$$
(S4)

125 which is independent of the distribution of CDOM. Therefore, in arctic waters where CDOM is

the only constituent that attenuates light, upscaled phtomineralization rate over the water column 126 is not limited by vertical hydrodynamic mixing. 127

128

#### 129 SI-2. Compiling photochemical and hydrological data of arctic waters.

Spectra of light attenuation coefficient of CDOM (aCDOM) and photo-lability of 130 CDOM ( $\Phi_{\lambda}$ ) are available from over 1000 samples for arctic streams and rivers and over 2000 131 samples for arctic lakes<sup>1,7</sup>. These spectra show exponential relationships with wavelength 132 (Figure S3), although *aCDOM*<sub> $\lambda$ </sub> and  $\Phi_{\lambda}$  were usually reported as single values at specific 133 wavelengths in many other studies. In latter cases, full spectra were extrapolated by assuming the 134 same decadic slopes (-0.005 for *aCDOM*<sub> $\lambda$ </sub> and -0.009 for  $\Phi_{\lambda}$ ) with wavelength as the available 135 spectra (Figure S3). Note that these slopes are based on log10 of  $aCDOM_{\lambda}$  and  $\Phi_{\lambda}$ , which is the 136 same as exponential slopes -0.012 and -0.021 for commonly reported natural log of  $aCDOM_{\lambda}$ 137 and  $\Phi_{2}^{9}$ . We also assumed that, within each type of system, high DOC concentration 138 corresponds to high  $aCDOM_{\lambda}$  and low DOC concentration corresponds to low  $aCDOM_{\lambda}$ , due to 139 the fact that DOC concentration and  $aCDOM_{i}$  were usually not reported together. 140 In streams and rivers, water surface slope (S) and water column depth (H) are expected to 141 142 co-vary, and both are a function of flow discharge (Q), although from the literature S, H, and Q 143 were usually not reported together. In order to predict S and H from reported values while

capturing their covariation, we plotted S-O and H-O relationships based on data when S-O or H-144

Q were reported together (Figure S4). For streams and rivers, both S and H exhibit power law 145

- 146 relationships with discharge,  $\log_{10} H = 0.25 \log_{10} Q 0.46$  and  $\log_{10} S = -0.12 \log_{10} Q 2.55$ . For
- 147 streams and rivers, S and H can therefore be predicted by the power laws and corresponding
- 148 prediction intervals, when at least one of S, H, or Q were reported. In beaded streams, existing S-
- 149 *Q* data do not show a clear relationship (Figure S4), and pool structure is expected to behave
- 150 differently than common streams, so we assumed that ranges of H and S for beaded streams do
- 151 not co-vary with Q.
- 152

## 153 SI-3. Non-dimensionalization of the dispersion-reaction equation.

154 The original dispersion-reaction equation is

155 
$$\frac{\partial C(y,t)}{\partial t} = \frac{\partial}{\partial y} \left( D(y) \frac{\partial C(y,t)}{\partial y} \right) - \sum_{\lambda} \phi_{\lambda} a CDOM_{\lambda}(y,t) Q_{dso-\lambda} e^{-K_{d\lambda}(y)y}$$
(S5)

156 with non-flux boundary conditions

157 
$$\frac{\partial C(y,t)}{\partial y}\Big|_{y=0} = 0$$
(S6a)

158 
$$\frac{\partial C(y,t)}{\partial y}\Big|_{y=H} = 0$$
(S6b)

159 where C(y,t) is DOC concentration, t is time, y is the vertical position that is 0 at water surface

- 160 and positive in the downward direction, and D(y) is the y-dependent vertical dispersion
- 161 coefficient that dictates vertical mixing and sets the rate of resupply of CDOM from bottom
- waters to the surface,  $\Phi_{\lambda}$  is the photo-lability of CDOM,  $Q_{dso-\lambda}$  is the photon flux at the water surface,  $aCDOM_{\lambda}(y,t)$  is the light attenuation coefficient of CDOM, and  $K_{d\lambda}(y)$  is the total light attenuation coefficient. Notations in this study are summarized in Table S3.
- 165 The dimensional variables contained in (S5) are *C*, *t*, and *y*. Thus, the dimensions to be 166 normalized are concentration, time, and length. We define the following scales to normalize 167 dimensional variables into dimensionless ones.

$$168 y = Hy^* (S7a)$$

169 
$$C(y,t) = \frac{1}{\varepsilon_{280}H} C^*(y^*,t^*)$$
 (S7b)

170 
$$t = \frac{H^2}{\langle D(y) \rangle} t^*$$
(S7c)

171 where superscript \* indicates dimensionless quantities,  $\varepsilon_{280}$  denotes empirical parameter  $\varepsilon_{\lambda}$  at 172 wavelength 280 nm, and  $\langle D \rangle$  is the depth-averaged vertical dispersion coefficient. Wavelength 173 280 nm was picked because it is the high end of the spectra such that  $\varepsilon_{280}$  is always non-zero.

174 Plug (S7) into (S5)-(S6), (S5)-(S6) become

175
$$\frac{\partial C^{*}(y^{*},t^{*})}{\partial t^{*}} = \frac{\partial}{\partial y^{*}} \left( \frac{D(y^{*})}{\langle D \rangle} \frac{\partial C^{*}(y^{*},t^{*})}{\partial y^{*}} \right) \\
- \frac{H^{2}}{\langle D \rangle} \sum_{\lambda} \phi_{\lambda} \frac{aCDOM_{\lambda}(y^{*},t^{*})}{C(y^{*},t^{*})} Q_{dso-\lambda} e^{-K_{d\lambda}(y^{*})Hy^{*}} C^{*}(y^{*},t^{*})$$
(S8)

176 with non-flux boundary conditions

177 
$$\frac{\partial C^*\left(y^*,t^*\right)}{\partial y^*}\Big|_{y^*=0} = 0$$
(S9a)

 $\frac{\partial C^*\left(y^*,t^*\right)}{\partial v^*}\Big|_{v^*=1} = 0$ (S9b)

(S8) can be rewritten as 179

180 
$$\frac{\partial C^*\left(y^*,t^*\right)}{\partial t^*} = \frac{\partial}{\partial y^*} \left( \frac{D\left(y^*\right)}{\langle D \rangle} \frac{\partial C^*\left(y^*,t^*\right)}{\partial y^*} \right) - \sum_{\lambda} d_{\lambda}^* e^{-p_{\lambda}^* y^*} C^*\left(y^*,t^*\right)$$
(S10)

181 where  $d_{\lambda}^{*}$  and  $p_{\lambda}^{*}$  are wavelength-specific dimensionless parameters defined by

182 
$$d_{\lambda}^{*} = \frac{H^{2}}{\langle D \rangle} \phi_{\lambda} \frac{aCDOM_{\lambda} (y^{*}, t^{*})}{C(y^{*}, t^{*})} Q_{dso-\lambda}$$
(S11a)

183 
$$p_{\lambda}^{*} = K_{d\lambda} \left( y^{*} \right) H \tag{S11b}$$

184

193

178

#### SI-4. Solving for the depth-integrated photomineralization rate. 185

The generalized dispersion-reaction equation is: 186

187 
$$\frac{\partial C^*\left(y^*,t^*\right)}{\partial t^*} = \frac{\partial}{\partial y^*} \left( \frac{D\left(y^*\right)}{\langle D \rangle} \frac{\partial C^*\left(y^*,t^*\right)}{\partial y^*} \right) - d^* e^{-p^* y^*} C^*\left(y^*,t^*\right)$$
(S12)

- 188 with non-flux boundary condition (S9), where  $d^*$  is surface Damkohler number, and  $p^*$  is
- dimensionless light attenuation. The moment method<sup>10</sup> solves for the dimensionless depth-189
- integrated photomineralization rate,  $r^*$ , as well as the vertical distribution of DOC concentration, 190
- at asymptotic regime. At the asymptotic regime, the dimensionless depth-averaged DOC 191
- concentration decays at a first-order rate constant  $r^*$  over time<sup>10</sup> 192
  - $\frac{d < C^* >}{dt^*} = r^* < C^* >$ (S13)
- The moment method defines a representative unit cell, where the vertical water column is 194

mirrored, such that reaction and dispersion profiles are symmetric about  $y^* = 0$ . The governing 195 196 equation within a unit cell becomes:

197 
$$\frac{\partial C^*\left(y^*,t^*\right)}{\partial t^*} = \frac{\partial}{\partial y^*} \left( D^*\left(y^*\right) \frac{\partial C^*\left(y^*,t^*\right)}{\partial y^*} \right) - d^* e^{-p^*\left|y^*\right|} C^*\left(y^*,t^*\right)$$
(S14)

198 with periodic boundary conditions:

199 
$$C^{*}(y^{*} = -1, t^{*}) = C^{*}(y^{*} = 1, t^{*})$$
 (S15a)  
200  $\frac{\partial C^{*}(y^{*}, t^{*})}{\partial y^{*}}\Big|_{y^{*} = -1} = \frac{\partial C^{*}(y^{*}, t^{*})}{\partial y^{*}}\Big|_{y^{*} = 1}$  (S15b)

(S15b)

201 At asymptotic regime, DOC concentration has been fully developed in the unit cell, such that

solute concentration is symmetric about  $y^* = 0$ . Therefore, non-flux boundary condition (S9) 202

203 applies in vertical water column.

The asymptotic regime solution  $r^*$  is given by the smallest eigenvalue of the following 204 205 eigenvalue problem<sup>10, 11</sup>

206 
$$\frac{d}{dy^*} \left( D^*\left( \left| y^* \right| \right) \frac{dE\left( y^* \right)}{dy^*} \right) - d^* e^{-p^* \left| y^* \right|} E\left( y^* \right) = -\Lambda E\left( y^* \right)$$
(S16)

208 
$$E(y^* = -1) = E(y^* = 1)$$
 (S17a)

(S17b)

209 
$$\frac{dE\left(y^{*}\right)}{dy^{*}}\Big|_{y^{*}=-1} = \frac{dE\left(y^{*}\right)}{dy^{*}}\Big|_{y^{*}=1}$$

where  $E(y^*)$  is the eigenfunction and  $\Lambda$  is the eigenvector. The eigenvector that corresponds to the smallest eigenvalue is the vertical distribution of DOC concentration at asymptotic regime. 210 211

- We used finite difference method to solve the eigenvalue problem (S16)-(S17). We discretized the vertical water column into n + 1 layers (n = 1024). The unit cell is therefore 213
- discretized into 2n + 2 layers, each with thickness  $h = \frac{1}{n+1}$ . For each layer i ( i = 0, 1, ..., 2n + 1) within the unit cell, denote  $y_i^* \equiv -1 + hi$  such that  $y_0^* = -1$  and  $y_{n+1}^* = 0$ . 214 215 216 Denote  $D_i^* \equiv D^* (y_i^*)$  and  $E_i \equiv E(y_i^*)$ , (S12) becomes  $D_i^* = E_i = D_i^* (y_i^*) = D_i^* = E_i^* = D_i^* =$

217 
$$\frac{D_{i+\frac{1}{2}}E_{i+1}}{h^2} + \frac{D_{i-\frac{1}{2}}E_{i-1}}{h^2} - \frac{(D_{i+\frac{1}{2}} + D_{i-\frac{1}{2}})E_i}{h^2}}{h^2} - d^*e^{-p^*y_i^*}E_i = -\Lambda E_i$$
(S18)

218 Define

$$\eta_i \equiv -\frac{(D_{i+\frac{1}{2}}^* + D_{i-\frac{1}{2}}^*)}{h^2} - d^* e^{-p^* y_i^*}$$
(S19a)

220 
$$\theta_i = \frac{D_{i+\frac{1}{2}}^*}{h^2}$$
 (S19b)

221 
$$\xi_i = \frac{D_{i-\frac{1}{2}}}{h^2}$$
 (S19c)

222

219

223 (S18) becomes

ה\*

224 
$$\xi_i E_{i-1} + \eta_i E_i + \theta_i E_{i+1} = -\Lambda E_i$$
(S20)

225

With periodic boundary conditions (S17) applied, (S20) can be written as 226

$$-\begin{pmatrix} \eta_{0} & \theta_{0} & 0 & \dots & 0 & 0 & \xi_{0} \\ \xi_{1} & \eta_{1} & \theta_{1} & \dots & 0 & 0 & 0 \\ 0 & \xi_{2} & \eta_{2} & \dots & 0 & 0 & 0 \\ & & & \ddots & & & \\ 0 & 0 & 0 & \dots & \eta_{2n-1} & \theta_{2n-1} & 0 \\ \theta_{2n+1} & 0 & 0 & \dots & 0 & \xi_{2n+1} & \eta_{2n+1} \end{pmatrix} \begin{pmatrix} E_{0} \\ E_{1} \\ E_{2} \\ \vdots \\ E_{2n-1} \\ E_{2n} \\ E_{2n+1} \end{pmatrix} = \Lambda \begin{pmatrix} E_{0} \\ E_{1} \\ E_{2} \\ \vdots \\ E_{2n-1} \\ E_{2n} \\ E_{2n+1} \end{pmatrix}$$
(S21)

230 To this point, the smallest eigenvalue and the corresponding eigenvector can be solved.

231

### 232 SI-5. Estimating vertical dispersion profiles of arctic waters.

In the Kuparuk River, the vertical dispersion coefficient D(y) follows a standard model<sup>12</sup>, <sup>13</sup>:

235 
$$D(y) = \kappa u_* y (1 - \frac{y}{H})$$
 (S22)

where  $\kappa$  is the von Karman constant,  $u_*$  is the shear velocity [m s<sup>-1</sup>], and *H* is the water column depth [m]. Because the observed river channel width is always at least 10 times larger than the observed river depth, turbulent properties are independent of river width<sup>12</sup>. Therefore, we approximate the shear velocity by  $u_* = \sqrt{gHS}$ , where g is gravitational acceleration [m s<sup>-2</sup>] and S

240 is the water surface slope. The depth-averaged vertical dispersion coefficient therefore is

241 
$$\left\langle D\left(y^*\right)\right\rangle = \frac{\kappa}{6}u_*H$$
 (S23)

We assume arctic streams and rivers to follow the same vertical dispersion profile as equation (S22). Arctic lakes and beaded stream pools usually have more complicated dispersion profiles. We assume two scenarios to capture the natural ranges of vertical mixing in beaded stream pools: the vertical dispersion profile in equation (S22) at fast flow, and a layered dispersion profile at slow flow. In the latter case, we assume that the top 10% of the pool depth mixes rapidly at the mean dispersion value of the fast flow scenario, while the deep layer of the pool mixes 2 orders of magnitude slower, a typical ratio for lakes worldwide<sup>14-16</sup>.

249 
$$D(y) = \begin{cases} \frac{\kappa}{6} u_* H, \quad y \le 10\% H\\ 1\% \frac{\kappa}{6} u_* H, \quad y > 10\% H \end{cases}$$
(S24)

250 In arctic lakes, because the epilimnion (or the surface layer) usually defines a turbulent mixing 251 layer that has much higher vertical dispersion than the deep layer, we define two scenarios of 252 vertical dispersion: deep lakes that are deeper than the mean epilimnion depth in arctic lakes, and 253 shallow lakes that are shallower than the mean epilimnion depth. Water transparency and mixing depth often relates to each other, because rapid absorption of light at the surface often creates a 254 "trapping depth" defined by a high density gradient<sup>17</sup>. A strong relationship between 255 256 transparency and mixing depth was found in a variety of lakes, including those larger than Toolik Lake (1.5 km<sup>2</sup> fetch) and those that are sheltered from the wind (crater lakes)<sup>18</sup>. Even in sheltered 257 258 lakes where diurnal thermoclines are more common during solar heating and calm conditions, the longer-term, average mixing depth is still related strongly to the optical properties of the 259 water<sup>18-20</sup>. Secchi depth, which characterizes the optical property of the water column, strongly 260

261 affects epilimnion depth in relatively small temperate lakes (defined as fetch < 500 ha in <sup>20</sup>), and 262 becomes less effective in predicting epilimnion depth of large temperate lakes (defined as fetch > 500 ha in <sup>20</sup>). However, very large lakes in the Arctic are rare compared to temperate lakes, and 263 studies typically report a much smaller range of fetch (0.003-1.5 km2, which is 0.3-150 ha, based 264 on 7 studies of 51 arctic lake)<sup>21-27</sup>. Further, given that arctic waters are typically light limited and 265 have high CDOM, even if mixing is confined to a near-surface layer, the amount CDOM is 266 267 difficult to consume in short periods of time (days to weeks), and CDOM would be resupplied 268 from greater depths as soon as a diurnal thermocline is erased, for example by wind mixing or convective mixing at night<sup>28</sup>. Therefore, Secchi depth was chosen to estimate epilimnion depth. 269 270 We estimate the mean epilimnion depth in arctic lakes using an empirical relationship with Secchi depth reported for temperate-zone lakes<sup>19</sup>, assuming that arctic lakes behave similarly to 271 temperate-zone lakes 272

$$\overline{E}_d = 3.24 + 0.35\overline{S}_d \tag{S25}$$

274 where  $E_d$  is the mean epilimnion depth (4.3 m) and  $S_d$  is the mean Secchi depth reported in arctic lakes. In shallow lakes, we treat the entire water column as the surface layer where vertical 275 276 dispersion is uniform over depth and spans the range 10<sup>-5</sup>–10<sup>-2</sup> m<sup>2</sup> s<sup>-1</sup>. In deep lakes, we assumed that deep-layer dispersion is 2 orders of magnitude lower than surface layer dispersion, which is 277

representative of reported ranges for lakes and oceans<sup>14-16</sup>. For each deep lake, given that Secchi 278 depth and lake depth are usually not reported together (Table S1), we assumed that the

279

epilimnion depth (or the surface layer thickness)  $E_d$  can be estimated by 280

281 
$$E_d = \frac{E_d}{\overline{H}_{DL}} H$$
(S26)

282 where  $H_{DL}$  is the mean depth of deep arctic lakes (12.5 m). (S26) only serves as a rough estimate 283 of  $E_d$  for each deep lake system.

286 Table S1. Data sources of photochemical and hydrological parameters in arctic stream, rivers,

287 beaded streams, and lakes.

288

	DOC	$aCDOM_{\lambda}$	$arPsi_\lambda$	Q	Н	S	SD
rivers	1, 7, 29-32	1, 7, 32	1,7	2, 32-35	2, 35	2, 33, 34	-
streams	1, 7, 29-31, 36, 37	1, 7, 38	1,7	2, 33, 35, 36, 39, 40	2, 35	2, 33, 34, 39, 40	-
beaded streams	1, 29-31, 41-44	38, 41, 43	1, 41, 42	33, 38-40, 43-47	38, 44-47	33, 39, 40, 45, 47	-
lakes	7, 21, 26, 31, 37, 48- 52	1,7	1, 7	-	21-27, 48, 49, 53	-	31, 54-58

289

290 \* From left to right, *DOC*, *aCDOM*<sub> $\lambda$ </sub>,  $\Phi_{\lambda}$ , *Q*, *H*, *S*, and *SD* denote DOC concentration, light attenuation

291 coefficient of CDOM, photo-lability of CDOM, flow discharge, water column depth, water surface slope,

and Secchi depth respectively. The reference numbers correspond to the references in SI.

293

**Table S2.** Reported ranges (min, max) of photochemical and hydrological parameters in arctic streams and rivers (rocky bottom), beaded streams (peat bottom), and lakes.

	DOC (mol m <sup>-3</sup> )	<i>aCDOM</i> <sub>300</sub> (m <sup>-1</sup> )	$\Phi_{300} \text{ (mol CO}_2 \text{ mol}^{-1} \text{ photon)}$	<i>H</i> (m)	S (-)	SD (m)
rivers	0.028, 18	3.6, 103	4×10 <sup>-4</sup> , 0.08	0.14, 15.5	5×10 <sup>-5</sup> , 0.03	-
streams	0.028, 18	4.6, 334	5×10 <sup>-4</sup> , 0.04	0.015, 2.1	1.7×10 <sup>-4</sup> , 0.1	-
beaded streams	0.12, 3.1	33, 359	0.003, 0.04	0.12, 3.0	3×10 <sup>-4</sup> , 0.009	-
lakes	0.008, 2.7	4.3, 79	4×10 <sup>-4</sup> , 0.08	0.08, 43	-	0.3, 11.2

298

299 \* Nomenclatures same as Table S1. Light attenuation coefficient of CDOM ( $aCDOM_{\lambda}$ ) and photo-lability

300 of CDOM ( $\Phi_{\lambda}$ ) are adjusted to wavelength = 300 nm.

301

303 Table S3. (A) Percentage of systems partially or substantially limited by mixing for each type of

304 arctic systems that exhibit exponential light attenuation over depth. (B) Depth-integrated

305 attenuation ratio (mean  $\pm$  standard error) for each type of arctic systems that are partially or 306 substantially limited by mixing.

307

308	(A)Percer	ntage within each type	of arctic systems		
		partially limited	substantially limited		
	streams and rivers	0%	0%		
	beaded streams	0.1%	0%		
	lakes	12%	30%		
309					
310	(B) Depth-integrated attenuation ratio				
		partially limited	substantially limited		
	streams and rivers				
	beaded streams	$0.92 \pm 0.001$			
	lakes	$0.95 \pm 0.001$	$0.58 \pm 0.002$		
811					
312					

## 314 **Table S4.** List of notations.

315

316 English letters

$aCDOM_{\lambda} [m^{-1}]$	light attenuation coefficient of CDOM
$aSS_{\lambda} [m^{-1}]$	light attenuation coefficient of suspended sediment (SS)
$C [\text{mol } \text{m}^{-3}]$	DOC concentration
$D [m^2 s^{-1}]$	vertical dispersion coefficient
$d^*$	surface Damkohler number
E	eigenfunction
$E_{d}$ [m]	epilimnion depth
g [m s <sup>-2</sup> ]	gravitational acceleration
<i>H</i> [m]	water column depth
$H_{DL}[m]$	mean depth of deep arctic lakes
$K_{d\lambda} [\mathrm{m}^{-1}]$	total light attenuation coefficient
$p^*$	dimensionless light attenuation
P <sub>e</sub>	statistical equilibrium of the vertical distribution of DOC
<i>PM</i> [mol m <sup>-3</sup> s <sup>-1</sup> ]	DOC photo-mineralization rate
$Q [m^{-3} s^{-1}]$	discharge
$Q_{dso-\lambda} [\text{mol } \text{m}^{-2} \text{s}^{-1}]$	photon flux at the water surface
$Q_{\lambda \text{ [mol m}^{-2} \text{ s}^{-1}]}$	photon flux
$r[s^{-1}]$	depth-integrated photomineralization rate
$r_{wm [s^{-1}]}$	depth-integrated photomineralization rate under well-mixed assumption
S	water surface slope
<i>S</i> <sub>d</sub> [m]	Secchi depth
<i>t</i> [s]	time
$u_{*} [m s^{-1}]$	shear velocity
W [m]	river width
y [m]	vertical position in the water column

# 317

## 318 Greek letters

$\alpha_{\lambda [m^{-1}]}$	$= \int \alpha_{\lambda} + \varepsilon_{\lambda} C  if  \alpha_{\lambda} + \varepsilon_{\lambda} C > 0$		
$\varepsilon_{\lambda}  [\mathrm{m}^2  \mathrm{mol}^{-1}]$	$\begin{bmatrix} a \in DOM_{\lambda} - \\ 0 & if  \alpha_{\lambda} + \varepsilon_{\lambda} C \le 0 \end{bmatrix}$		
$\eta_{react}$	reaction efficiency		
κ	von Karman constant		
Λ	eigenvector		
$\lambda$ [m]	wavelength		
$\phi_{1}$ [mal CO malti shotas]	photo-lability of CDOM to photomineralization, also known as		
$r_{\lambda}$ [mol CO <sub>2</sub> mol <sup>-1</sup> photon]	the apparent quantum yield for photomineralization		

319

320 Others

superscript *	dimensionless variable or	parameter
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	$\langle \rangle$	depth-averaged value	
	-	mean value	
321			

## 322 **References**

- 323
- 1. R. M. Cory, C. P. Ward, B. C. Crump and G. W. Kling, Sunlight controls water column processing of carbon in arctic fresh waters, *Science*, 2014, **345**, 925-928.
- T. V. King, B. T. Neilson, L. D. Overbeck and D. L. Kane, Water temperature controls in low arctic rivers, *Water Resources Research*, 2016, **52**, 4358-4376.
- G. W. Kling and R. M. Cory, Biogeochemistry data set for NSF Arctic Photochemistry
   project on the North Slope of Alaska, *Environmental Data Initiative*, 2015,
- 330 <u>http://dx.doi.org/10.6073/pasta/22a3a3fc2dc74b7aabe8a10ab9061cf0</u>.
- G. W. Kling and R. M. Cory, Photochemistry data set for NSF Photochemistry project on
  the North Slope of Alaska, 2015,
- 333 <u>http://dx.doi.org/10.6073/pasta/2f9433d6a608e82e1dd4fa23175c1f59</u>.
- G. W. Kling and R. M. Cory, Light profile data set for NSF Photochemistry project on
  the North Slope of Alaska, *Environmental Data Initiative*, 2015,
- 336 <u>http://dx.doi.org/10.6073/pasta/8e8cb22fd7ee278168f8eb6ad7e1a48c</u>.
- G. W. C. Kling, R. M., Apparent quantum yield data set for NSF Photochemistry project on the North Slope of Alaska, *Environmental Data Initiative*, 2015,
- 339 <u>http://dx.doi.org/10.6073/pasta/aa2d0ed4ddef6e76c3ef8d6c12460607</u>.
- R. M. Cory, B. C. Crump, J. A. Dobkowski and G. W. Kling, Surface exposure to
  sunlight stimulates CO2 release from permafrost soil carbon in the Arctic, *Proceedings of the National Academy of Sciences of the United States of America*, 2013, **110**, 3429-3434.
- 343 8. J. D. J. Ingle and S. R. Crouch, Spectrochemical analysis, 1988.
- 344 9. C. L. Osburn, N. J. Anderson, C. A. Stedmon, M. E. Giles, E. J. Whiteford, T. J.
- McGenity, A. J. Dumbrell and G. J. Underwood, Shifts in the Source and Composition of
   Dissolved Organic Matter in Southwest Greenland Lakes Along a Regional
- Hydro-climatic Gradient, *Journal of Geophysical Research: Biogeosciences*, 2017, 122,
  3431-3445.
- 349 10. M. Shapiro and H. Brenner, Dispersion of a chemically reactive solute in a spatially
- periodic model of a porous medium, *Chemical Engineering Science*, 1988, **43**, 551-571.
- B. B. Dykaar and P. K. Kitanidis, Macrotransport of a biologically reacting solute
  through porous media, *Water Resources Research*, 1996, **32**, 307-320.
- H. B. Fischer, J. E. List, C. R. Koh, J. Imberger and N. H. Brooks, *Mixing in inland and coastal waters*, Academic Press, San Diego, CA, 1979.
- M. R. Leeder, Sedimentology and sedimentary basins: from turbulence to tectonics,
  Wiley-Blackwell, Oxford, U.K., 2nd edn., 2011.
- 357 14. S. W. Zison, *Rates, constants, and kinetics formulations in surface water quality* 358 *modeling*, Environmental Protection Agency, Environmental Research Laboratory,
- 359 Athens, GA, 1978.
- 360 15. G. E. Likens, *Biogeochemistry of inland waters*, Academic Press, San Diego, CA, 2010.
- 361 16. S. C. Chapra, *Surface water-quality modeling*, Waveland press, Long Grove, IL, 2008.
- J. F. Price, R. A. Weller and R. Pinkel, Diurnal cycling: Observations and models of the
  upper ocean response to diurnal heating, cooling, and wind mixing, *Journal of Geophysical Research: Oceans*, 1986, **91**, 8411-8427.
- 365 18. G. W. Kling, Comparative transparency, depth of mixing, and stability of stratification in
- 366 lakes of Cameroon, West Africa1, *Limnology Oceanography*, 1988, **33**, 27-40.

- A. Mazumder and W. D. Taylor, Thermal structure of lakes varying in size and water
   clarity, *Limnology and Oceanography*, 1994, **39**, 968-976.
- E. Fee, R. Hecky, S. Kasian and D. Cruikshank, Effects of lake size, water clarity, and
  climatic variability on mixing depths in Canadian Shield lakes, *Limnology Oceanography*,
  1996, 41, 912-920.
- J. E. Hobbie and G. W. Kling, *Alaska's changing Arctic: Ecological consequences for tundra, streams, and lakes*, Oxford University Press, 2014.
- M. H. Hermanson, Anthropogenic mercury deposition to Arctic lake sediments, *Water, Air, and Soil Pollution*, 1998, **101**, 309-321.
- P. T. Doran, C. P. McKay, W. P. Adams, M. C. English, R. A. Wharton and M. A. Meyer,
  Climate forcing and thermal feedback of residual lake-ice covers in the high Arctic, *Limnology and Oceanography*, 1996, 41, 839-848.
- A. Evenset, G. N. Christensen, T. Skotvold, E. Fjeld, M. Schlabach, E. Wartena and D.
  Gregor, A comparison of organic contaminants in two high Arctic lake ecosystems,
- Bjørnøya (Bear Island), Norway, *Science of the Total Environment*, 2004, **318**, 125-141.
- B. M. Jones, C. D. Arp, K. M. Hinkel, R. A. Beck, J. A. Schmutz and B. Winston, Arctic
  lake physical processes and regimes with implications for winter water availability and
  management in the National Petroleum Reserve Alaska, *Environmental Management*,
  2009, 43, 1071-1084.
- 386 26. S. Kokelj, R. Jenkins, D. Milburn, C. Burn and N. Snow, The influence of thermokarst
  disturbance on the water quality of small upland lakes, Mackenzie Delta region,
- 388 Northwest Territories, Canada, *Permafrost and Periglacial Processes*, 2005, **16**, 343-353.
- D. Schindler, J. Kalff, H. Welch, G. Brunskill, H. Kling and N. Kritsch, Eutrophication in
  the high arctic-meretta lake, cornwallis island (75 N lat.), *Journal of the Fisheries Board*of Canada, 1974, **31**, 647-662.
- 392 28. S. MacIntyre, J. P. Fram, P. J. Kushner, N. D. Bettez, W. O'brien, J. Hobbie and G. W.
  393 Kling, Climate-related variations in mixing dynamics in an Alaskan arctic lake,
- *Limnology Oceanography*, 2009, **54**, 2401-2417.
- W. B. Bowden, ArcLTER streams chemistry data 1983 to present, *Environmental Data Initiative*, 2016, <u>http://dx.doi.org/10.6073/pasta/8b81b5b0cf936c8b2bb173adab506fd7</u>.
- J. R. Larouche, B. W. Abbott, W. B. Bowden and J. B. Jones, The role of watershed
  characteristics, permafrost thaw, and wildfire on dissolved organic carbon
  biodegradability and water chemistry in Arctic headwater streams, *Biogeosciences*, 2015,
  400 **12**, 4221-4233.
- 401 31. G. W. Kling, Chemistry from thermokarst impacted soils, lakes, and streams near Toolik
  402 Lake Alaska, 2008-2011, *Environmental Data Initiative*, 2012,
- 403 http://dx.doi.org/10.6073/pasta/2e55d1587290e642938ac1a6caed6ec6.
- 404 32. P. Mann, R. Spencer, P. Hernes, J. Six, G. Aiken, S. Tank, J. McClelland, K. Butler, R.
- 405Dyda and R. Holmes, Pan-arctic trends in terrestrial dissolved organic matter from optical406measurements, *Earth Sci*, 2016, 4, 25.
- 407 33. K. J. Edwardson, W. B. Bowden, C. Dahm and J. Morrice, The hydraulic characteristics
  408 and geochemistry of hyporheic and parafluvial zones in Arctic tundra streams, north
  409 slope, Alaska, *Advances in Water Resources*, 2003, 26, 907-923.
- 410 34. K. M. Scott, *Effects of permafrost on stream channel behavior in Arctic Alaska*, US Govt.
  411 Print. Off., 1978.

- W. M. Wollheim, B. J. Peterson, L. A. Deegan, J. E. Hobbie, B. Hooker, W. B. Bowden,
  K. J. Edwardson, D. B. Arscott, A. E. Hershey and J. Finlay, Influence of stream size on
  ammonium and suspended particulate nitrogen processing, *Limnology and Oceanography*,
  2001, 46, 1-13.
- 416 36. B. C. Crump, G. W. Kling, M. Bahr and J. E. Hobbie, Bacterioplankton community shifts
  417 in an arctic lake correlate with seasonal changes in organic matter source, *Applied and*418 *Environmental Microbiology*, 2003, **69**, 2253-2268.
- G. W. Kling, G. W. Kipphut, M. M. Miller and W. J. O'Brien, Integration of lakes and
  streams in a landscape perspective: the importance of material processing on spatial
  patterns and temporal coherence, *Freshwater Biology*, 2000, 43, 477-497.
- 422 38. M. Merck, B. Neilson, R. Cory and G. Kling, Variability of in-stream and riparian 423 storage in a beaded arctic stream, *Hydrological Processes*, 2012, **26**, 2938-2950.
- M. J. Greenwald, W. B. Bowden, M. N. Gooseff, J. P. Zarnetske, J. P. McNamara, J. H.
  Bradford and T. R. Brosten, Hyporheic exchange and water chemistry of two arctic
  tundra streams of contrasting geomorphology, *Journal of Geophysical Research: Biogeosciences*, 2008, 113.
- 428 40. J. P. Zarnetske, M. N. Gooseff, T. R. Brosten, J. H. Bradford, J. P. McNamara and W. B.
  429 Bowden, Transient storage as a function of geomorphology, discharge, and permafrost
  430 active laver conditions in Arctic tundra streams, *Water Resources Research*, 2007, 43.
- 431 41. R. M. Cory, K. H. Harrold, B. T. Neilson and G. W. Kling, Controls on dissolved organic
  432 matter (DOM) degradation in a headwater stream: the influence of photochemical and
  433 hydrological conditions in determining light-limitation or substrate-limitation of photo434 degradation, *Biogeosciences*, 2015, **12**, 6669-6685.
- 435 42. G. W. Kling, Biogeochemistry data set for Imnavait Creek Weir on the North Slope of
  436 Alaska, *Environmental Data Initiative*, 2015,

437 <u>http://dx.doi.org/10.6073/pasta/f540d09369a9065704df7df66927083b</u>.

- 438 43. B. L. Miller, Terrestrial-aquatic transfers of carbon dioxide, methane, and organic carbon
  439 from riparian wetlands to an arctic headwater stream, M.S., University of Michigan at
  440 Ann Arbor, 2014.
- 441 44. M. Oswood, J. Irons III and D. Schell, in *Landscape function and disturbance in arctic tundra*, Springer, 1996, pp. 275-289.
- 443 45. C. D. Arp, M. S. Whitman, B. M. Jones, G. Grosse, B. V. Gaglioti and K. C. Heim,
  444 Distribution and biophysical processes of beaded streams in Arctic permafrost landscapes,
  445 *Biogeosciences*, 2015, 12, 29-47.
- 446 46. M. Oswood, K. Everett and D. Schell, Some physical and chemical characteristics of an 447 arctic beaded stream, *Holarctic Ecology*, 1989, **12**, 290-295.
- 448 47. A. Tarbeeva and V. Surkov, Beaded channels of small rivers in permafrost zones, 449 *Geography and Natural Resources*, 2013, **34**, 216-220.
- 450 48. M. Amyot, D. Lean and G. Mierle, Photochemical formation of volatile mercury in high 451 Arctic lakes, *Environmental Toxicology and Chemistry*, 1997, **16**, 2054-2063.
- 452 49. P. B. Hamilton, K. Gajewski, D. E. Atkinson and D. R. Lean, Physical and chemical
  453 limnology of 204 lakes from the Canadian Arctic Archipelago, *Hydrobiologia*, 2001, 457,
  454 133-148.
- 455 50. D. P. Morris, H. Zagarese, C. E. Williamson, E. G. Balseiro, B. R. Hargreaves, B.
- 456 Modenutti, R. Moeller and C. Queimalinos, The attenuation of solar UV radiation in

457		lakes and the role of dissolved organic carbon, <i>Limnology and Oceanography</i> , 1995, 40,
458		1381-1391.
459	51.	S. Sobek, L. J. Tranvik, Y. T. Prairie, P. Kortelainen and J. J. Cole, Patterns and
460		regulation of dissolved organic carbon: An analysis of 7,500 widely distributed lakes,
461		<i>Limnology and Oceanography</i> , 2007, <b>52</b> , 1208-1219.
462	52.	S. E. Tank, L. F. Lesack, J. A. Gareis, C. L. Osburn and R. H. Hesslein, Multiple tracers
463		demonstrate distinct sources of dissolved organic matter to lakes of the Mackenzie Delta,
464		western Canadian Arctic, 2011.
465	53.	G. W. Kling, G. W. Kipphut and M. C. Miller, The flux of CO2 and CH4 from lakes and
466		rivers in arctic Alaska, Hydrobiologia, 1992, 240, 23-36.
467	54.	A. K. Giblin, George W, Physical and chemical data for various lakes near Toolik
468		Research Station, Arctic LTER. Summer 1983 to 1989, Environmental Data Initiative,
469		2015, http://dx.doi.org/10.6073/pasta/1f780cc1b9e31f58a87d72b8eb2693ea.
470	55.	A. K. Giblin, George W, Physical and chemical data for various lakes near Toolik
471		Research Station, Arctic LTER. Summer 1990 to 1999, Environmental Data Initiative,
472		2015, http://dx.doi.org/10.6073/pasta/1fd85582de93a281e5e5d3b80df97b52.
473	56.	A. K. Giblin, George W, Physical and chemical data for various lakes near Toolik
474		Research Station, Arctic LTER. Summer 2000 to 2009, Environmental Data Initiative,
475		2015, http://dx.doi.org/10.6073/pasta/791e3cb6288f75f602f23ef3e5532017.
476	57.	A. K. Giblin, George W, Physical and chemical data for various lakes near Toolik
477		Research Station, Arctic LTER. Summer 2010 to 2014, Environmental Data Initiative,
478		2015, http://dx.doi.org/10.6073/pasta/2d161db112e1855277c112d9b4ffc980.
479	58.	G. W. Kling, Biogeochemistry data set for soil waters, streams, and lakes near Toolik on
480		the North Slope of Alaska, Environmental Data Initiative, 2016,
481		http://dx.doi.org/10.6073/pasta/e240e6b082d5536e1c9dd2d2ddc2e88b.
100		
482		