Electronic Supplementary Information

Wrinkle-Induced High Sorption Makes Fewlayered Black Phosphorus a Superior Adsorbent for Ionic Organic Compounds

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Fig. S1 Chemical structures of methylene blue (left) and Congo red (right).



Before Centrifugation After Centrifugation

Fig. S2 Tyndal effect observation of the supernatant before and after centrifugation.



Fig. S3 (a) UV-Vis spectra of MB and CR during sorption varied with time and initial concentration; (b) HPLC spectra of MB and CR after sorption. No new peaks were found in both spectra, suggesting no degradation occurred.

LM												
Compounds	Q^{o}	P of Q^0	b	Ро	P of b RMSEC		r_{adj}^2					
MB	856.81±28.61	< 0.0001	0.011±0.0	001 <0.0	0.0001 24.912		0.9870					
CR	218.59±4.90	< 0.0001	0.050±0.0	005 <0.0	0001	9.227	0.9800					
FM												
Compounds	K_{f}	P of K_f	n	Рc	ofn R	MSEC	r_{adj}^2					
MB	38.77±6.67	< 0.0001	0.505±0.0)33 <0.0	0001	38.81	0.9700					
CR	43.82±5.39	< 0.0001	0.283±0.0	024 <0.0	0001	16.10	0.9394					
DA												
Compounds	Q^{o}	P of Q^0	Ε	P of E	ł)	RMSEC	r _{adj} ²				
MB	1232±283	< 0.0001	20.3±0.9	< 0.0001	2.68±	⊧0.32	18.268	0.993				
CR	230±8.97	< 0.0001	26.6±0.1	< 0.0001	5.28±0.34		8.535	0.991				

Table S1. Results of LM, FM, and DA models fitted sorption data of two dyes by few-layered BP ^a

^a All estimated parameter values, their standard error, probability of assuming the null hypothesis (p) and the fitting adjusted square of correlation coefficient (r_{adj}^2) were determined by a commercial software program (Sigmaplot 11.1).



Fig. S4 Variation of separation factor R_L with dyes' initial concentration C_o .



Fig. S5 Comparison of amount sorption of two dyes at different Na⁺ concentration levels at initial concentration of 50 and 500 mg/L respectively.

PFOM													
Compounds	q_e	P of q_e		k_1	P of k_1	RMSEC	r_{adj}^{2}						
MB	174.32±16.81	< 0.0001	-1.27	25±0.6457	< 0.0001	36.923	0.5976						
CR	171.18±18.47	< 0.0001	-0.0234±0.0057		<0.0001	15.386	0.9048						
PSOM													
Compounds	q_e	P of q	le	k_2	RMSE	r_{adj}^2							
MB	251.52±14.99	< 0.000)1	0.00088	32.27	0.9622							
CR	174.89±16.61	< 0.0001		0.00032	17.20	5 0.8867							
IPDM													
Compounds	$k_{ m i}$	P of k_i		θ	P of θ	RMSEC	r_{adj}^2						
MB	27.24±1.85	< 0.0001	47.	10±7.41	< 0.0001	13.403	0.947						
CR	14.55±0.47	< 0.0001	6.4	42±2.48	< 0.0001	4.973	0.984						

Table S2. Results of PFOM, PSOM, and IPDM fitted with sorption kinetic data of two dyes by few-layered BP ^a

^a All estimated parameter values, their standard error, probability of assuming the null hypothesis (p) and the fitting adjusted square of correlation coefficient (r_{adj}^2) were determined by a commercial software program (Sigmaplot 11.1).



Fig. S6 TEM images (a-c), AFM images (d-f), and statistical analysis of the thickness (g-i) of few-layered BP before (a, d, and g) and after sorption at the initial CR concentration of 50 mg/L (b, e, and h) and 500 mg/L (c, f, and i).



Fig. S7 Intial (left panel) and energy minimized (right panel) configurations of adsorption complex, the relatively unfavorable cases: MB adsored on graphene (GP2-3) and phosphorene (BP4-6) (Dark gray balls: C atoms; Light gray balls: H atoms; blue balls: N atoms; yellow balls: S atoms; Pink balls: P atoms; Unit for distance is Å).