Supplementary Information for

Enhanced Debromination of Tetrabromobisphoenol A by Zero-valent Copper Nanoparticles Modified Green Rusts

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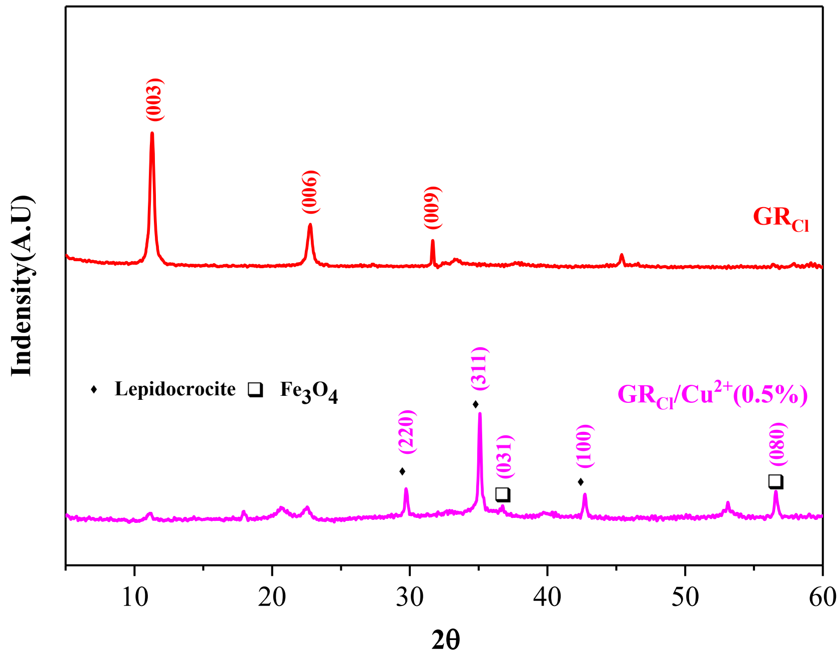


Figure S1. XRD of the GR(Cl)-Cu^{2+} and GR(Cl).

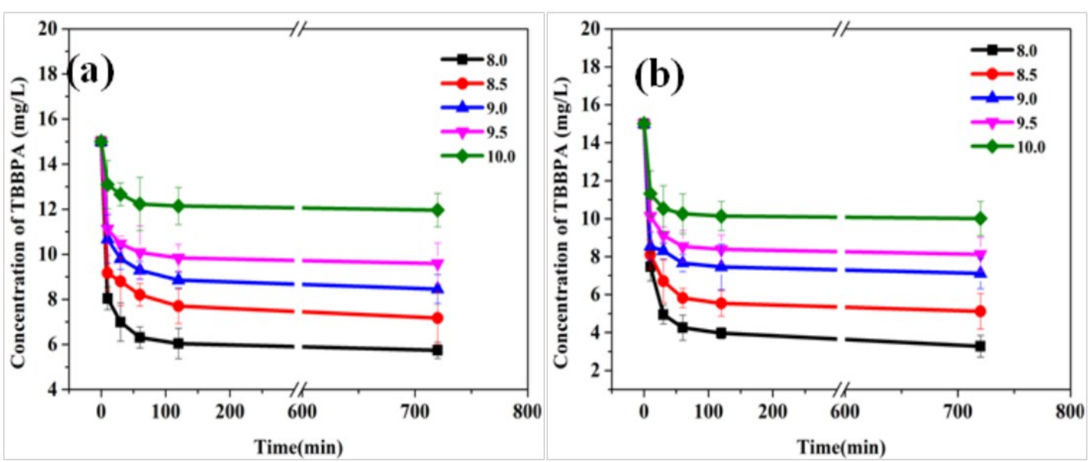


Figure S2. Effect of pH on TBBPA reduction by GR(Cl) and GR(Cl)-Cu NPs.

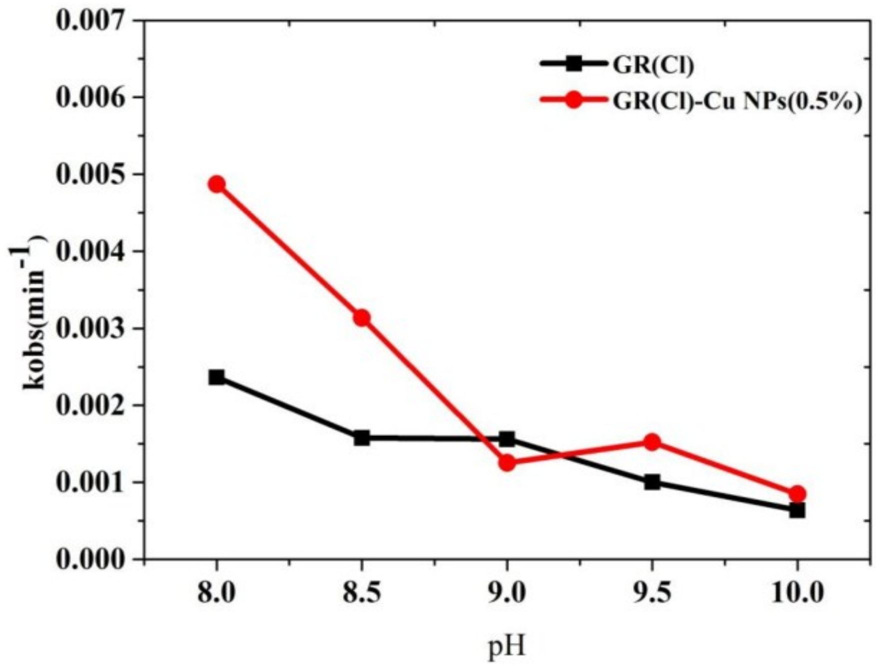


Figure S3. Effect of pH on pseudo first order reaction rate of TBBPA reduction by GR(Cl) and GR(Cl)-Cu NPs.

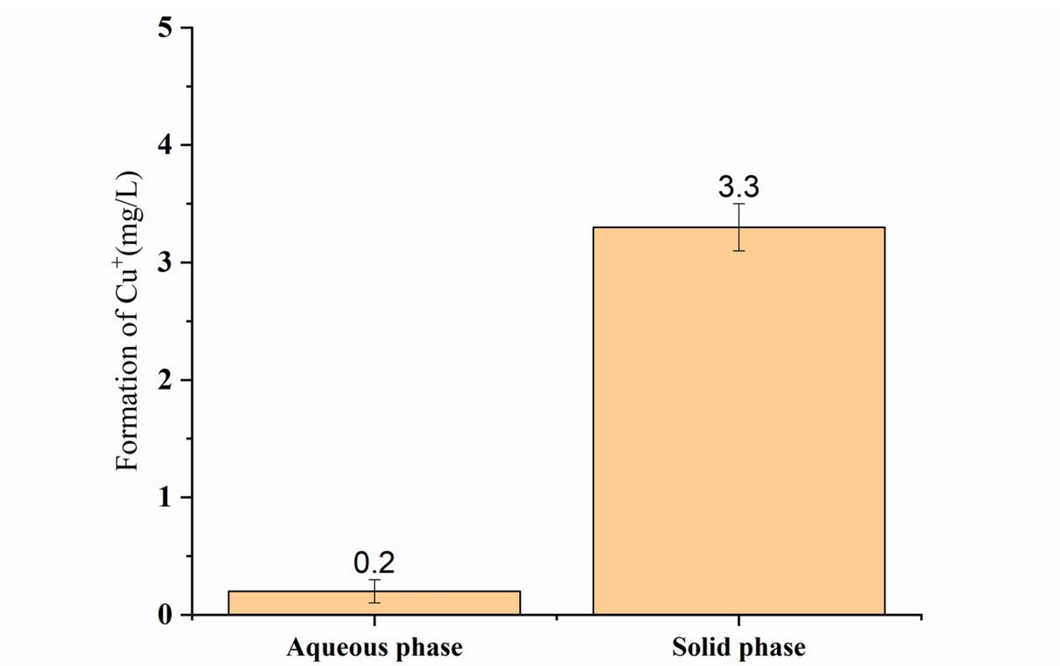


Figure S4. Cu^{+} content in GR(Cl)-Cu^{2+} system (0.5%). The concentration is normalized to the reaction aqueous volume.

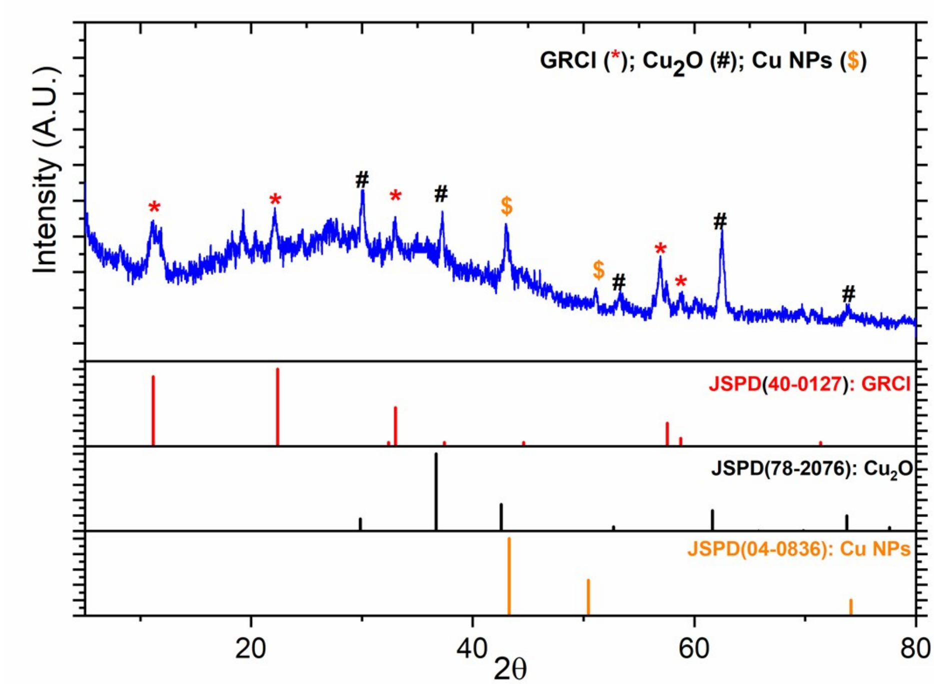
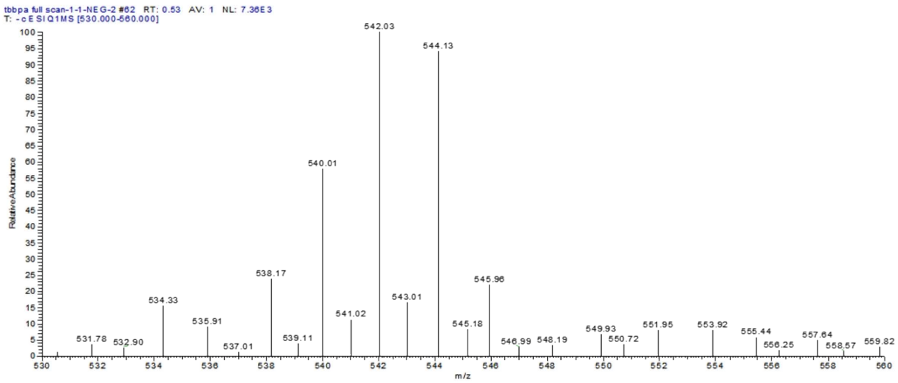
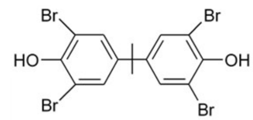
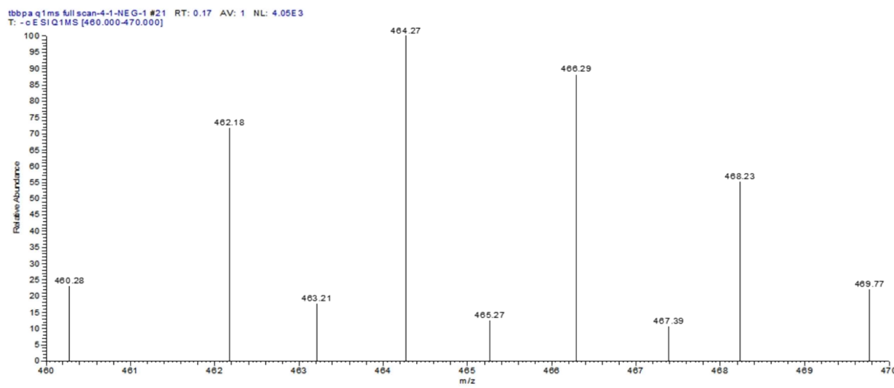


Figure S5. Cu_2O generation after GR(Cl)-Cu NPs reaction with TBBPA. High Cu NPs dosage (10%) in GR(Cl)-Cu NPs was used for Cu_2O measurement.

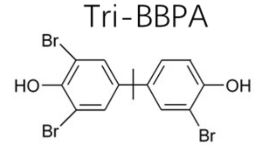


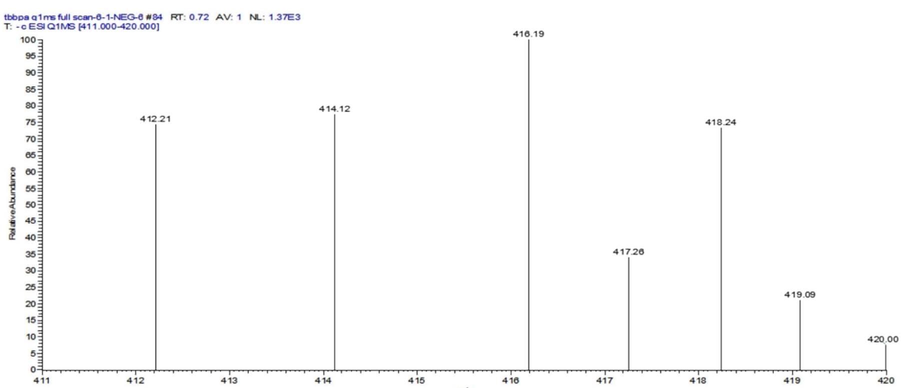
TBBPA



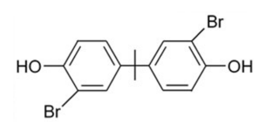


Tri-BBPA





DBBPA



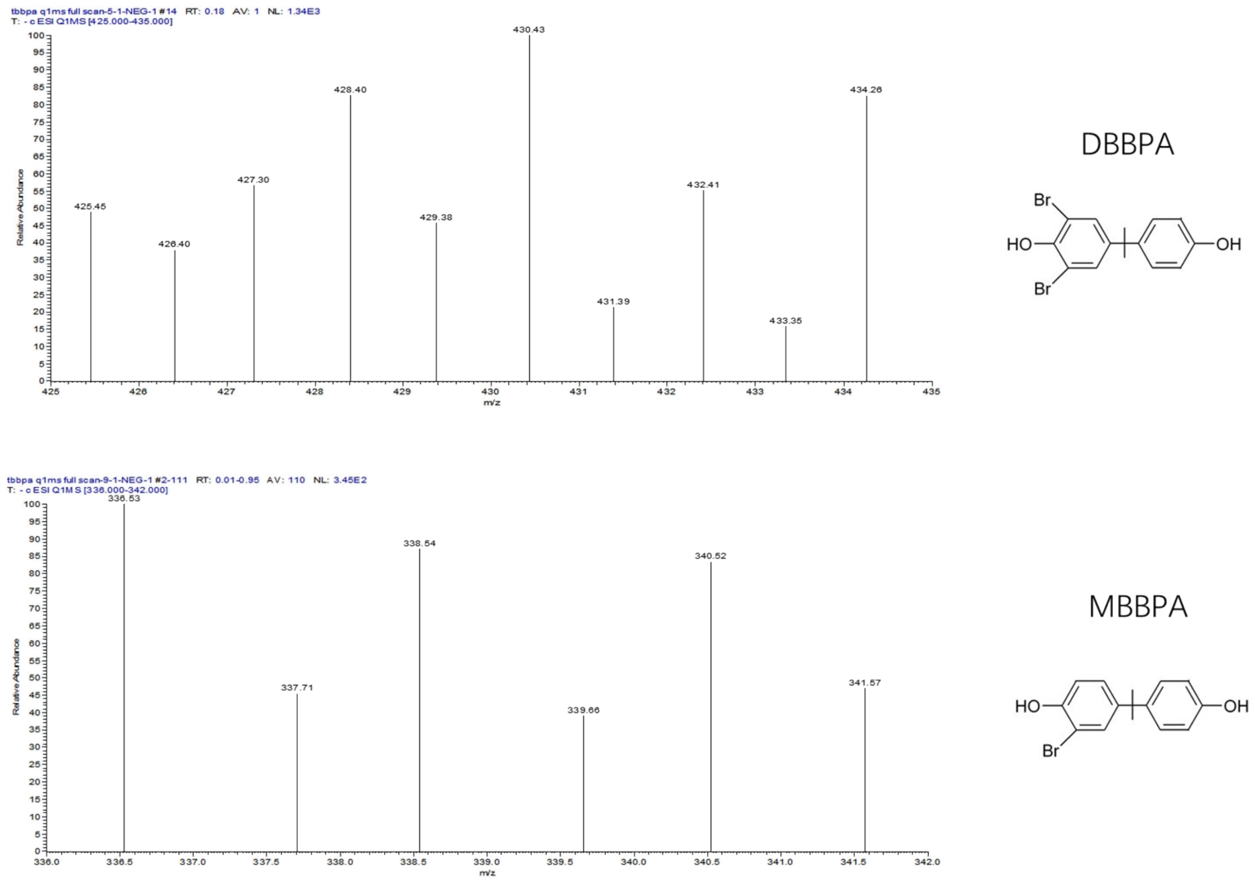


Figure S6. Mass spectra of potential intermediates from the degradation of TBBPA by GR(Cl)-Cu NPs.