

**Feasibility evaluation of the treatment and recycling of shale gas  
produced water: A case study of the first shale gas field in the Eastern  
Sichuan Basin, China**

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Supporting data: 12 pages, 5 tables, 6 figures.

**Table S1** Operational strategy and parameters for the laboratory-scale PW treatment

Step	Treatment process		Parameters
1	Pretreatment and combined advanced chemical oxidation	Coagulation	Dosing $\text{Al}_2(\text{SO}_4)_3$ with pH = 6.0
2		Modified Fenton oxidation <sup>a</sup>	$\text{H}_2\text{O}_2 = 10 \text{ mL/L}$ ; $\text{FeSO}_4 \cdot 7\text{H}_2\text{O} = 0.5 \text{ and } 2.5 \text{ g/L}$ ; pH = 3.2
3		NaClO oxidation	NaClO = 1 mL/L; pH = 7.0
4		Residual chlorine removal	$\text{Na}_2\text{SO}_3$ dosing
5	Multi-media filtration	Sand-artificial zeolite filtration	-
6		Activated carbon filtration	-
7	Desalination	RO	RO membrane material:GE1812; Pressure = 3.0–4.0 MPa; Pure water conductivity = 450–650 $\mu\text{S/cm}$

a. Modified Fenton oxidation was designed by dosing  $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$  twice in an entire Fenton oxidation experiment.

**Table S2** Operational strategy and parameters for recycling RO concentrate

Step	Treatment process		Parameters
1	Pretreatment	Coagulation-softening	-
2		NaClO oxidation	NaClO = 21 mL/L; pH = 10.0–11.0
3	Advanced oxidation <sup>a</sup>	O <sub>3</sub> -H <sub>2</sub> O <sub>2</sub> oxidation	O <sub>3</sub> gas flow rate = 30 L/h; pH = 11.0
4		Electrochemical oxidation	DSA-type anode; Stainless steel cathode; Current = 0.77 A; pH = 7.5
5	Advanced filtration	Ion exchange	D-301 anion exchange resin; Filtration flow rate = 1.0 L/h
6		Activated carbon filtration	Granular activated carbon; Filtration flow rate = 1.0 L/h
7	Concentration	Evaporation concentration	-

a. Removal efficiencies of residual refractory organic matters in the pretreated RO concentrate by O<sub>3</sub>-H<sub>2</sub>O<sub>2</sub> oxidation and DSA-type electrochemical oxidation techniques were compared.

**Table S3** Operational strategy and parameters of MBBR system for the co-treatment of PW with domestic wastewater

Reactor volume (L)	Sequencing batch operational mode	HRT (h)	DO in Oxidic stage (mg/L)	MLSS concentration	Biofilm carriers
8.0	Feeding (0.25 h)/Anoxic (2 h)/Oxidic (2 h)/Anoxic (2 h)/Oxidic (1 h)/Setting (0.5 h)/Withdrawing (0.25 h)	8.0	4.5–6.5	3000–4000 mg/L	AnoxKaldnes K5 type; Fill ratio = 30%

**Table S4** Representative of identified molecules in the PW from the Eastern Sichuan Basin by TD-GC-MS analysis

Formula	Compound name	CAS
C <sub>7</sub> H <sub>5</sub> N <sub>5</sub> O <sub>3</sub>	Pterin-6-carboxylic acid	948-60-7
C <sub>9</sub> H <sub>11</sub> F <sub>2</sub> NO <sub>3</sub>	Benzeneethanamine	152434-78-1
C <sub>2</sub> H <sub>6</sub> N <sub>2</sub> O <sub>2</sub>	Hydroxyacetic acid	3530-14-1
C <sub>10</sub> H <sub>18</sub> O <sub>2</sub>	8-Methyl-6-nonenic acid	21382-25-2
C <sub>21</sub> H <sub>34</sub> S <sub>2</sub>	cyclic ethylene mercaptole	2759-86-6
C <sub>14</sub> H <sub>25</sub> F <sub>3</sub> O <sub>2</sub>	2-Trifluoroacetoxydodecane	1894-68-4
C <sub>8</sub> H <sub>24</sub> O <sub>4</sub> Si <sub>4</sub>	Cyclotetrasiloxane	556-67-2
C <sub>22</sub> H <sub>45</sub> Cl <sub>3</sub> Si	Docosyltrichlorosilane	7325-84-0
C <sub>13</sub> H <sub>28</sub>	5-ethyl-5-methyldecane	17312-74-2
C <sub>31</sub> H <sub>56</sub>	Pentacosan-13-ylbenzene	6006-90-2
C <sub>11</sub> H <sub>21</sub> ClO	4-Chloro-3-n-hexyltetrahydropyran	66555-66-6
C <sub>12</sub> H <sub>24</sub> O	Z-2-Dodecenol	69064-36-4
C <sub>10</sub> H <sub>9</sub> NO <sub>2</sub>	4-Methoxy-2(1H)-quinolone	27667-34-1
C <sub>17</sub> H <sub>36</sub>	2,6,10-trimethyltetradecane	14905-56-7
C <sub>9</sub> H <sub>16</sub> BrNO	2-Piperidinone	195194-80-0
C <sub>20</sub> H <sub>34</sub> O	8,11,14-Eicosatrienoic acid	1783-84-2
C <sub>37</sub> H <sub>76</sub> O	1-Heptatriacotanol	105794-58-9
C <sub>16</sub> H <sub>26</sub> O <sub>3</sub>	2-Dodecen-1-yl(-)succinic anhydride	19780-11-1
C <sub>16</sub> H <sub>30</sub> O	11-Hexadecyn-1-ol	65686-49-9
C <sub>14</sub> H <sub>24</sub> O <sub>2</sub>	3-Tetradecanynoic acid	55182-76-8
C <sub>40</sub> H <sub>56</sub>	$\pi$ Carotene	7235-40-7
C <sub>15</sub> H <sub>28</sub> O	2-Pentadecyn-1-ol	2834-00-6
C <sub>15</sub> H <sub>28</sub> O	Z,Z-2,5-Pentadecadien-1-ol	139185-79-8
C <sub>15</sub> H <sub>24</sub> O	Butylated Hydroxytoluene	128-37-0
C <sub>22</sub> H <sub>38</sub> O <sub>2</sub>	Cyclopropaneoctanoic acid	10152-71-3
C <sub>12</sub> H <sub>20</sub> O <sub>4</sub>	trans-Traumatic acid	6402-36-4
C <sub>15</sub> H <sub>26</sub> O <sub>2</sub>	Geranyl isovalerate	109-20-6
C <sub>16</sub> H <sub>34</sub> N <sub>2</sub> O	16-Hexadecanoyl hydrazide	2619-88-7
C <sub>16</sub> H <sub>22</sub> O <sub>2</sub>	1,1'-Butadiynylenedicyclohexanol	5768-10-5
C <sub>19</sub> H <sub>32</sub> O <sub>2</sub>	9-Octadecen-12-ynoic acid, methyl ester	56847-05-3
C <sub>18</sub> H <sub>38</sub> N <sub>2</sub> O	Stearic acid hydrazide	4130-54-5
C <sub>16</sub> H <sub>30</sub> O	7-Hexadecenal, (Z)-	56797-40-1
C <sub>10</sub> H <sub>16</sub> O <sub>2</sub>	2-Decanynoic acid	1851-90-7
C <sub>14</sub> H <sub>24</sub> O <sub>2</sub>	2-Myristynoic acid	67587-19-3

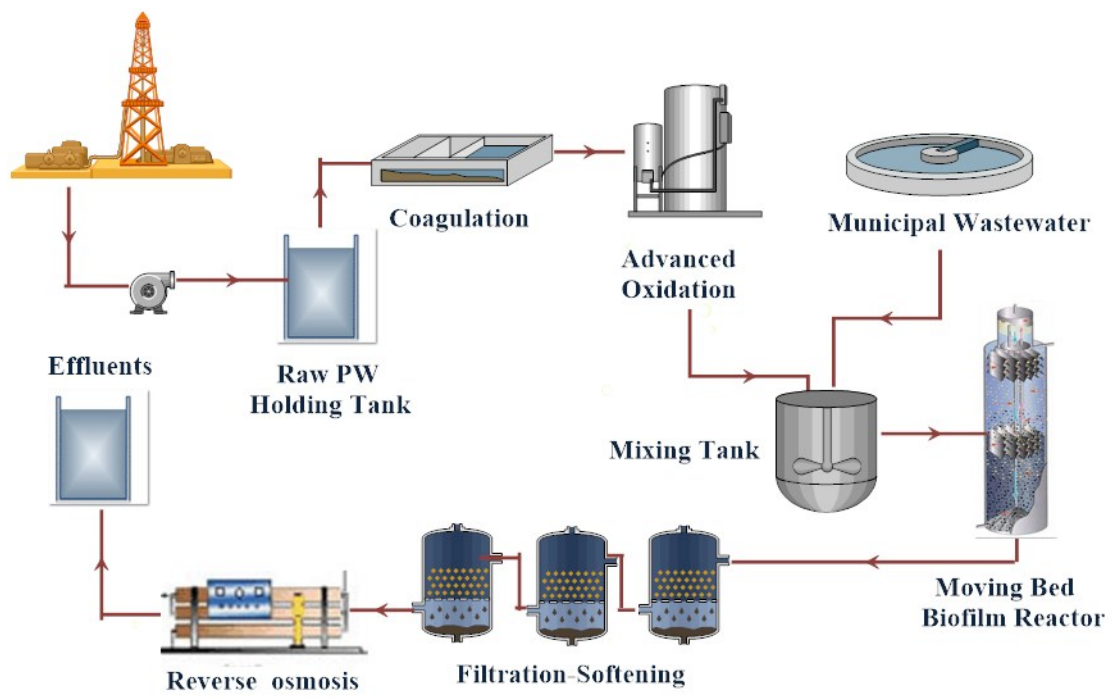
**Table S5** Representative wastewater discharge standards of oil & gas industry in  
China

Items (mg·L <sup>-1</sup> , except for dimensionless pH)	Emission standard of pollutants for petroleum chemistry industry (GB31571-2015)		Emission standard of pollutants for petroleum refining industry (GB31571-2015)	
	Direct discharge <sup>a</sup>	Indirect discharge <sup>b</sup>	Direct discharge	Indirect discharge
pH	6.0-9.0	-	6-9	-
TSS	70	-	70	-
COD <sub>cr</sub>	60 100 <sup>c</sup>	-	60	-
BOD <sub>5</sub>	20	-	20	-
NH <sub>3</sub> -H	8.0	-	8.0	-
TN	40	-	40	-
TP	1.0	-	1.0	-
TOC	20 30 <sup>c</sup>	-	20	-
Petroleum-like	5.0	20	5.0	20
Sulfide	1.0	1.0	1.0	1.0
Fluoride	10	20	-	
Volatile phenol	0.5	0.5	0.3	0.5
Cu	0.5	0.5	-	-
Total cyanide	0.5	0.5	0.3	0.5

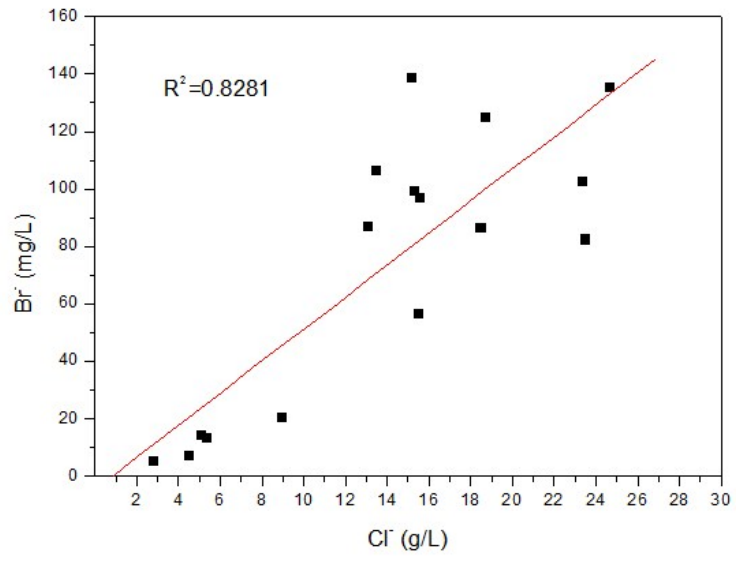
a. Direct discharge : wastewater discharged to municipal WWTPs or sewage pipe network.

b. Indirect discharge : wastewater discharged to private industrial WWTPs.

c. Wastewater from acrylonitrile- acrylic fibres, caprolactam, epichlorohydrin,  
2,6-di-tert-butyl-4-methylphenol, pure terephthalic acid, cresol, epoxide propane, and  
naphthalene production factories.



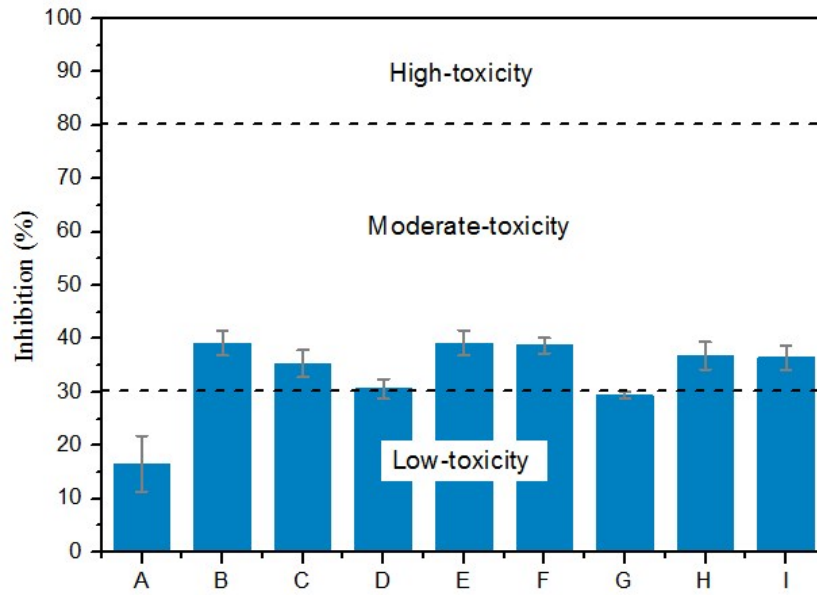
**Fig. S1** Flow diagram of the laboratory-scale co-treatment of PW and domestic wastewater.



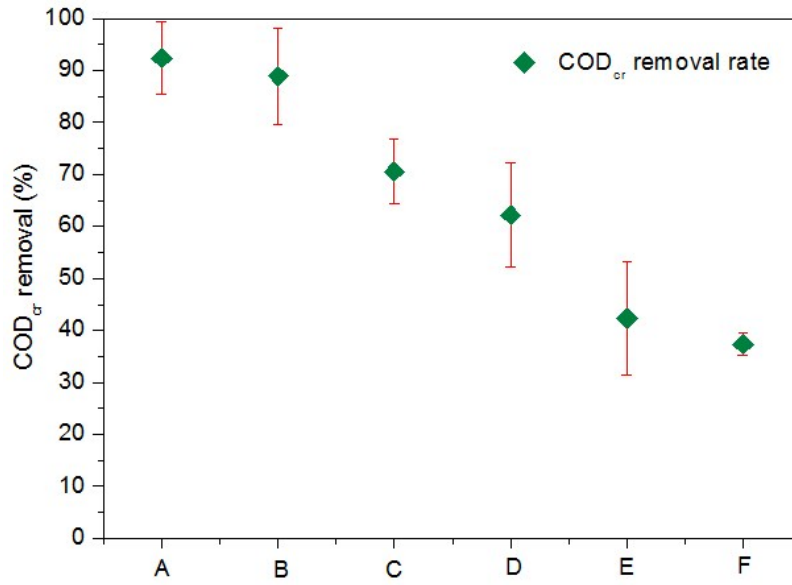
**Fig. S2** Relationship of Cl<sup>-</sup> and Br<sup>-</sup> among PW wells in the shale gas field in the Eastern Sichuan

Basin.

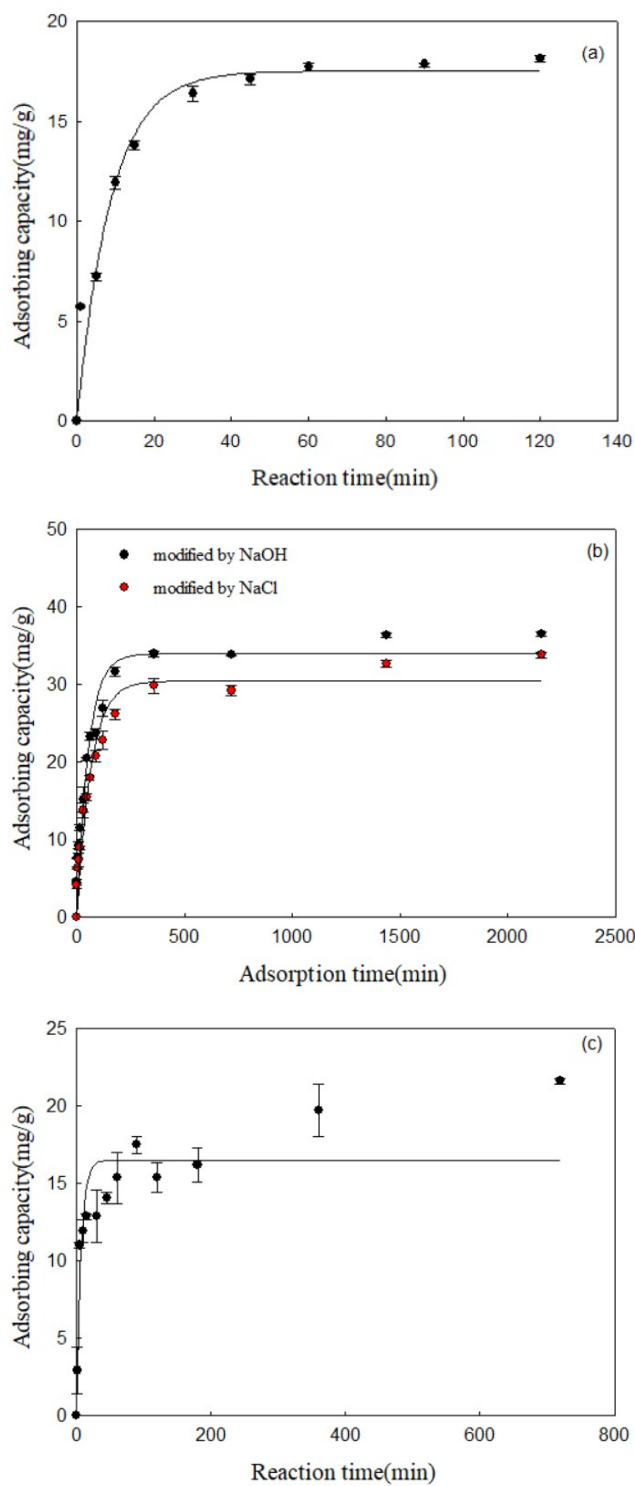




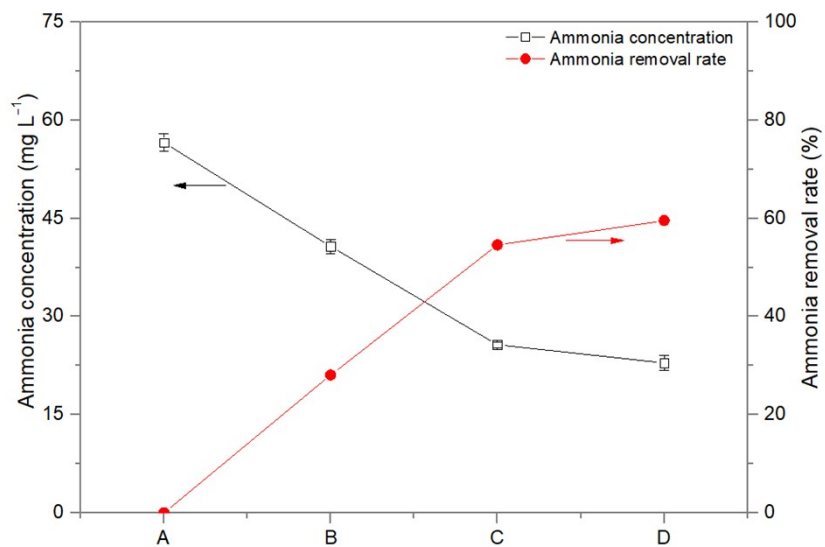
**Fig. S3** Acute toxicity tests of the raw PW samples among nine wells represented as the luminescence inhibition rate (low-toxicity level: inhibition ratio <30%; mid-toxicity level:  $\leq$ 30% inhibition ratio  $\leq$ 80%; high-toxicity level: inhibition ratio >80%).



**Fig. S4** Comparison among different advanced and chemical oxidation processes on the removal of COD<sub>cr</sub> (A. Modified Fenton–NaClO oxidation; B. Activated potassium persulfate– NaClO oxidation; C. NaClO oxidation; D. Modified Fenton; E. Activated potassium persulfate; F. Fenton).



**Fig. S5** First-order kinetic fitting curve of a modified artificial zeolite for ammonia adsorption in deionized water (a. Raw artificial zeolite; b. NaOH- and NaCl-modified artificial zeolite; c. HCl-modified artificial zeolite).



**Fig. S6** Ammonia removal rate of pretreated PW by a modified artificial zeolite (A: Sequential advanced oxidation pretreated PW; B. Filtration with a raw artificial zeolite; C. Filtration with a NaCl-modified artificial zeolite; D. Filtration with a NaOH-modified artificial zeolite).