

Supplementary Information for: Effects of symmetry breaking on the translation-rotation eigenstates of H₂, HF, and H₂O inside the fullerene C₆₀[†]

Zlatko Bačić,^{*,†,‡} Vojtěch Vlček,[¶] Daniel Neuhauser,[¶] and Peter M. Felker^{*,¶}

[†]*Department of Chemistry, New York University, New York, New York 10003, USA*

[‡]*NYU-ECNU Center for Computational Chemistry at NYU Shanghai, 3663 Zhongshan
Road North, Shanghai 200062, China*

[¶]*Department of Chemistry and Biochemistry, University of California, Los Angeles,
California 90095-1569, USA*

E-mail: zlatko.bacic@nyu.edu; felker@chem.ucla.edu

1 Hamiltonian Parameters

1.1 H₂@C₆₀

The kinetic energy operator for H₂@C₆₀ was taken to be

$$\hat{T} = -\frac{\nabla^2}{2M} + B\hat{j}^2 \quad (1)$$

where ∇^2 is the Laplacian associated with \mathbf{R} , \hat{j}^2 is the operator corresponding to the square of the rotational angular momentum of the H₂, M is the mass of the H₂, and B is the

rotational constant of the H₂. We used $M = 2.0104$ amu, $B = 58.378$ cm⁻¹ for the $v = 0$ manifold, and $B = 54.83$ cm⁻¹ for the $v = 1$ manifold.

The $V_{M-C_{60}}$ function for both the $v = 0$ and the $v = 1$ manifolds was taken to be a pairwise-additive Lennard-Jones one of the form

$$V_{H_2-C_{60}} = \sum_{i=1}^3 \sum_{k=1}^{60} 4w_i \epsilon \left[\left(\frac{\sigma}{r_{ik}} \right)^{12} - \left(\frac{\sigma}{r_{ik}} \right)^6 \right], \quad (2)$$

where i runs over three H₂ sites, k runs over the 60 nuclear positions of the C atoms in the central cage, and r_{ik} is the distance between site i and site k . For both manifolds the H₂ site 1 was located at the center of the HH bond, and sites 2 and 3 were located at the H nuclei. For $v = 0$ the HH bond distance was taken to be 0.74 Å, $w_1 = 6.7$, $w_2 = w_3 = 1$, $\sigma = 2.95$ Å, and $\epsilon = 3.07$ cm⁻¹.¹ For $v = 1$ the HH bond distance was taken to be 0.78132 Å, $w_1 = 7.5$, $w_2 = w_3 = 1$, $\sigma = 2.95$ Å, and $\epsilon = 2.9886668$ cm⁻¹.² The C₆₀ geometry was taken to be that used in Felker, et al.³

As to V_{quad} , the BF \hat{z} axis was taken to be the internuclear axis, and the one nonzero BF quadrupole component for H₂ was taken to be $Q_0^{\text{BF}} = 0.499$ au for both the $v = 0$ and $v = 1$ manifolds. This is the same value that was used in Felker, et al.³

1.2 HF@C₆₀

The kinetic energy operator for HF@C₆₀ was taken to have the same form as eqn (1) but with $M = 20.006225$ amu and $B = 18.523$ cm⁻¹. This value for B is the cage-modified one determined by Kalugina and Roy.⁴

The $V_{HF-C_{60}}$ function was taken directly from Kalugina and Roy.⁴ It is an expansion over bipolar spherical tensors dependent on the four angles (Θ, Φ, ω) with R -dependent expansion coefficients. It does not require any input as to the C₆₀ geometry.

The BF \hat{z} axis was taken by Kalugina and Roy⁵ to be the internuclear axis pointing from the H nucleus to the F nucleus. As such $\vec{\mu} = \mu \hat{z}$ is antiparallel to \hat{z} and μ is negative. We

take the magnitude of μ to be the screened value of -0.177 au from Krachmalnicoff, et al.⁶

1.3 H₂O@C₆₀

The kinetic energy operator for H₂O@C₆₀ was taken to be

$$\hat{T} = -\frac{\nabla^2}{2M} + B_x \hat{j}_x^2 + B_y \hat{j}_y^2 + B_z \hat{j}_z^2 \quad (3)$$

where ∇^2 is the Laplacian associated with \mathbf{R} , \hat{j}_x , \hat{j}_y , and \hat{j}_z are the operators associated with the components of the rotational angular momentum of the H₂O along the BF axes, which are take to be its principal inertial axes. We used $M = 18.0105$ amu, $B_x = 27.877$ cm⁻¹, $B_y = 9.285$ cm⁻¹, and $B_z = 14.512$ cm⁻¹. This choice of the BF axes locates the bisector of the HOH bond angle to be along the BF \hat{z} axis.

The $V_{M-C_{60}}$ for H₂O@C₆₀ was taken from Felker and Bačić⁷ and is given by

$$V_{H_2O-C_{60}} = \sum_{i=1}^3 \sum_{k=1}^{60} 4\epsilon_i \left[\left(\frac{\sigma_i}{r_{ik}} \right)^{12} - \left(\frac{\sigma_i}{r_{ik}} \right)^6 \right], \quad (4)$$

where i runs over three H₂O sites, k runs over the 60 nuclear positions of the C atoms in the central cage, r_{ik} is the distance between site i and site k , $\sigma_1 = 3.372$ Å, $\sigma_2 = \sigma_3 = 2.640$ Å, $\epsilon_1 = 36.34$ cm⁻¹, and $\epsilon_2 = \epsilon_3 = 8.95384$ cm⁻¹. The three H₂O sites are given in Table 2 of the ESI of Felker, et al.³ The C₆₀ geometry was taken to be the same as that used for H₂@C₆₀.

As to V_{quad} , since we take the BF \hat{z} axis to point from the c.m. of the water moiety toward the O nucleus along the HOH bond-angle bisector, then $\vec{\mu} = \mu\hat{z}$ is antiparallel to \hat{z} , and μ is negative. We used the screened dipole value, $\mu = -0.200$ au, from Goh, et al.⁸ The BF quadrupole components of the H₂O were taken to be the same as in Felker, et al.:³ $Q_0^{(\text{BF})} = -0.09973$ au and $Q_{\pm 2}^{(\text{BF})} = 1.53843$ au.

2 Grid Parameters

As mentioned in Subsection 2.2 of the main body of the paper the TR state function, $|\psi\rangle$, employed in the Chebyshev filter diagonalization procedure was transformed to a grid representation to effect its multiplication by the potential-energy portion of \hat{H} . The general nature of the five-dimensional (5D) grid points used for $\text{H}_2@\text{C}_{60}$ and for $\text{HF}@\text{C}_{60}$, and the six-dimensional (6D) grid points used for $\text{H}_2\text{O}@\text{C}_{60}$ are described in Section 2.5 of Felker, et al.³ Further specifics as to the grids used in this work follow.

For $\text{M}=\text{H}_2$ we used (i) 12 Gauss-associated-Laguerre quadrature points generated as per Felker and Bačić⁹ with $\beta = 2.9888989$ au for the R coordinate, (ii) 10 Gauss-Legendre quadrature points for the $\cos \Theta$ coordinate, (iii) 18 Fourier grid points for the Φ coordinate, (iv) 10 Gauss-Legendre quadrature points for the $\cos \theta$ coordinate, and (v) 18 Fourier grid points for the ϕ coordinate. Here, the relevant Euler angles are $\omega = (\theta, \phi)$, where θ is the polar angle, and ϕ the azimuthal angle describing the orientation of the BF \hat{z} axis with respect to the SF axis system.

For $\text{M}=\text{HF}$ we used (i) 14 Gauss-associated-Laguerre quadrature points generated with $\beta = 12.0$ au for the R coordinate, (ii) 12 Gauss-Legendre quadrature points for the $\cos \Theta$ coordinate, (iii) 24 Fourier grid points for the Φ coordinate, (iv) 10 Gauss-Legendre quadrature points for the $\cos \theta$ coordinate, and (v) 18 Fourier grid points for the ϕ coordinate.

For $\text{M}=\text{H}_2\text{O}$ we used (i) 12 Gauss-associated-Laguerre quadrature points generated with $\beta = 24.38$ au for the R coordinate, (ii) 10 Gauss-Legendre quadrature points for the $\cos \Theta$ coordinate, (iii) 18 Fourier grid points for the Φ coordinate, (iv) 10 Gauss-Legendre quadrature points for the $\cos \theta$ coordinate, (v) 18 Fourier grid points for the ϕ coordinate, and (vi) 18 Fourier grid points for the χ coordinate. Here, $\omega = (\phi, \theta, \chi)$ are the Euler angles, defined with the convention used in Zare,¹⁰ that specify the orientation of the BF axes of the H_2O with respect to the SF axes.

References

- (1) M. Xu, S. Ye, A. Powers, R. Lawler, N. J. Turro, and Z. Bačić, *J. Chem. Phys.*, 2013, **139**, 064309.
- (2) M. Xu, F. Sebastianelli, B. R. Gibbons, Z. Bačić, R. Lawler and N. J. Turro, *J. Chem. Phys.*, 2009, **130**, 224306.
- (3) P. M. Felker, V. Vlček, I. Hietanen, S. FitzGerald, D. Neuhauser and Z. Bačić, *Phys. Chem. Chem. Phys.*, 2017, **19**, 31274.
- (4) Y. N. Kalugina and P. N. Roy, *J. Chem. Phys.*, 2017, **147**, 244303.
- (5) P. N. Roy, private communication.
- (6) A. Krachmalnicoff, R. Bounds, S. Mamone, S. Alom, M. Concistrè, B. Meier, K. Kouřil, M. E. Light, M. R. Johnson, S. Rols, A. J. Horsewill, A. Shugai, U. Nagel, T. Rõõm, M. Carravetta, M. Levitt and R. J. Whitby, *Nature Chem.*, 2016, **8**, 953.
- (7) P. M. Felker and Z. Bačić, *J. Chem. Phys.*, 2016, **144**, 201101.
- (8) K. S. K. Goh, M. Jiménez-Ruiz, M. R. Johnson, S. Rols, J. Ollivier, M. S. Denning, S. Mamone, M. H. Levitt, X. Lei, Y. Li, N. J. Turro, Y. Murata and A. J. Horsewill, *Phys. Chem. Chem. Phys.*, 2014, **16**, 21330.
- (9) See P. M. Felker and Z. Bačić, *J. Chem. Phys.*, 2016, **145**, 084310, Supplementary Material, Section II.
- (10) R. N. Zare, *Angular Momentum* (Wiley, New York, 1988).