Supplementary information to

## Model Potential Study of Non-Valence Correlation-Bound Anions of

## (C<sub>60</sub>)<sub>n</sub> Clusters: the Role of Electric Field-Induced Charge Transfer

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## 1. The extended Mayer-Åstrand model allowing for intermolecular charge transfer

For a molecular dimer such as  $(C_{60})_2$ , the constrained charge-flow equation in the Mayer-Åstrand model, which does not permit intermolecular charge transfer, is

$$f = E_{tot} + \lambda_1 \left( \sum_{i=1}^{N_1} q_{1,i} - Q_{1,tot} \right) + \lambda_2 \left( \sum_{i=1}^{N_2} q_{2,i} - Q_{2,tot} \right)$$
(S1)

where  $\lambda_1$  and  $\lambda_2$  are Lagrange multipliers and  $Q_{1,tot}$  and  $Q_{2,tot}$  are the total charges of the isolated molecules 1 and 2, respectively. The total polarization energy,  $E_{tot}$  is given by eqn (2) in the manuscript. In extending the model to allow for intermolecular charge transfer we use the expression:

$$f = E_{tot} + \lambda_1 \left( \sum_{i=1}^{N_1} q_{1,i} - Q_{1,tot} \right) + \lambda_2 \left( \sum_{i=1}^{N_2} q_{2,i} - Q_{2,tot} \right) + \eta (\lambda_1 - \lambda_2)^2$$
(S2)

where  $\eta$  is represented by exponential depending on the internuclear separation as described below. By applying  $df/q_1 = 0$ ,  $df/q_2 = 0$ ,  $df/p_1 = 0$ ,  $df/p_2 = 0$ ,  $df/d\lambda_1 = 0$ , and  $df/d\lambda_2 = 0$ , the following set of

equations is obtained:

$$\begin{pmatrix} \mathbf{T}_{11}^{\mathbf{q}\mathbf{q}} & \mathbf{T}_{12}^{\mathbf{q}\mathbf{q}} & -\mathbf{T}_{11}^{\mathbf{p}\mathbf{q}} & -\mathbf{T}_{21}^{\mathbf{p}\mathbf{q}} & 1 & 0 \\ \mathbf{T}_{21}^{\mathbf{q}\mathbf{q}} & \mathbf{T}_{22}^{\mathbf{q}\mathbf{q}} & -\mathbf{T}_{12}^{\mathbf{p}\mathbf{q}} & -\mathbf{T}_{22}^{\mathbf{p}\mathbf{q}} & 0 & 1 \\ -\mathbf{T}_{11}^{\mathbf{p}\mathbf{q}} & -\mathbf{T}_{12}^{\mathbf{p}\mathbf{q}} & -\mathbf{T}_{12}^{\mathbf{p}\mathbf{p}} & -\mathbf{T}_{12}^{\mathbf{p}\mathbf{p}} & 0 & 0 \\ -\mathbf{T}_{21}^{\mathbf{p}\mathbf{q}} & -\mathbf{T}_{22}^{\mathbf{p}\mathbf{q}} & -\mathbf{T}_{21}^{\mathbf{p}\mathbf{p}} & -\mathbf{T}_{22}^{\mathbf{p}\mathbf{p}} & 0 & 0 \\ 1 & 0 & 0 & 0 & 2\eta & -2\eta \\ 0 & 1 & 0 & 0 & -2\eta & 2\eta \end{pmatrix} \begin{pmatrix} \mathbf{q}_1 \\ \mathbf{q}_2 \\ \mathbf{p}_1 \\ \mathbf{p}_2 \\ \lambda_1 \\ \lambda_2 \end{pmatrix} = \begin{pmatrix} -\mathbf{V}_1 \\ -\mathbf{V}_2 \\ \mathbf{E}_1 \\ \mathbf{E}_2 \\ Q_{1,tot} \\ Q_{2,tot} \end{pmatrix}$$
(S3)

where  $\mathbf{T}^{\mathbf{qq}}$ ,  $\mathbf{T}^{\mathbf{pq}}$ , and  $\mathbf{T}^{\mathbf{pp}}$  are charge-charge, charge-dipole and dipole-dipole interaction matrices, respectively. The matrix on the left-hand side is of Eq. S3 referred to as the molecular tensor matrix.  $\mathbf{q}_j$  and  $\mathbf{p}_j$  are vectors that contain, respectively, the  $q_{j,i}$  and the components of  $p_{j,i}$ , for molecule *j*.  $\mathbf{V}_j$  is a vector of length  $N_1$  that contains the  $V_{j,i}$ , and  $\mathbf{E}_j$  is a vector of length  $3N_j$  that contains the components of the external field  $\mathbf{E}_{ext}$ . The inverse of the molecular tensor matrix become the molecular polarizability matrix,  $\boldsymbol{\alpha}^{mol}$ , which has charge-charge, charge-dipole, dipole-dipole components:

$$\boldsymbol{\alpha}^{mol} = \begin{pmatrix} \boldsymbol{\alpha}_{qq}^{mol} & \boldsymbol{\alpha}_{qp}^{mol} & \cdots \\ \boldsymbol{\alpha}_{pq}^{mol} & \boldsymbol{\alpha}_{pp}^{mol} & \cdots \\ \vdots & \vdots & \ddots \end{pmatrix}$$
(S4)

The molecular polarizability tensor,  $\alpha$  is computed as

$$\boldsymbol{\alpha} = \sum_{j=1}^{N} \sum_{i=1}^{N} \left( \alpha_{q_i q_j}^{mol} \mathbf{R}_j \otimes \mathbf{R}_i - 2 \boldsymbol{\alpha}_{q_i \mathbf{p}_j}^{mol} \otimes \mathbf{R}_i + \boldsymbol{\alpha}_{\mathbf{p}_i \mathbf{p}_j}^{mol} \right)$$
(S5)

where *N* is the total number of atoms and  $\mathbf{R}_i$  is the vector defining atomic site *i*. The first, second, and third terms on the right-hand of eqn (S5) indicate charge flow, charge-dipole coupling, and induced dipole contributions to the molecular polarizability, respectively.

## 2. Charge transfer parameter and applying the extended Mayer-Åstrand model to the $(C_{60})_n$ cluster

The quantity  $2\eta$  is represented by exponential  $Ae^{-B\cdot R}$ , where A and B are determined by fitting the distance dependence of the intermolecular charge transfer of the C<sub>60</sub> dimer from B3LYP calculations resulting from an external field of 0.001 a.u. in the *z* direction. Fig. 2 in the manuscript shows that the extended Mayer-Åstrand model closely reproduces the charge transfer from the B3LYP calculations over a wide range of *R* values. The charge transfer model described above is readily extended to clusters with more than two fullerene molecules.

Fig. S1 shows the molecular tensor matrices for  $(C_{60})_3$ .



(b) Triangular C<sub>60</sub> trimer



Fig. S1 Molecular tensor matrices with  $\eta$  terms that damp the intermolecular charge transfer for linear and triangular C<sub>60</sub> trimers.