

SUPPLEMENTARY INFORMATION

Addressing the energy sustainability of biowaste-derived hard carbon materials for battery electrodes

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SI 1 – Electrode cycling results

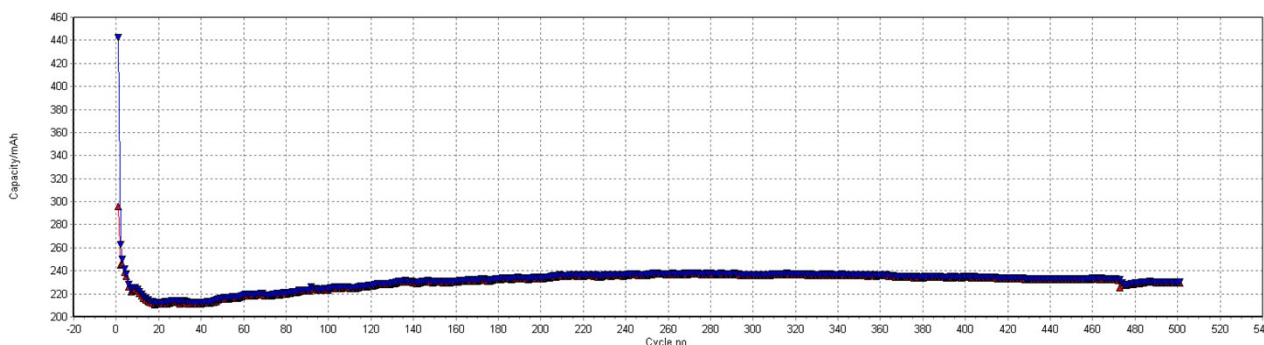


Figure SI-1. Long-term cycling performance of an apple pomace-derived hard carbon electrode in the Na//Hard carbon cell. Electrolyte: 1M NaClO₄ in EC: PC (1:1 wt.%). Charge and discharge current density: 200 mA h g⁻¹. For further details, please, see reference X. Dou, D. Buchholz, S. Passerini in *APL Materials* **6** (2018) 047501.

SI 2 - Extended model

Hereinafter detailed modelling equations are shown. In the main manuscript a condensed version of the equations is provided, in order to introduce theoretical elements which are accounted for. Yet, the main manuscript do not go to computational details. Moreover, a complete list of symbols adopted in the equations is attached to the Supplementary Information for the sake of clarity.

$$\sum_{j=1}^3 x_j = x_{cell} + x_{hemi} + x_{lign} = 1 \quad \text{Eq. SI - 1}$$

$$w_{subs}(T) = \sum_{j=1}^3 x_j w_j(T) = x_{cell} w_{cell}(T) + x_{hemi} w_{hemi}(T) + x_{lign} w_{lign}(T) \quad \text{Eq. SI - 2}$$

$$h_{subs}(T) = \sum_{j=1}^3 x_j w_j(T) h_j(T) = x_{cell} w_{cell}(T) h_{cell}(T) + x_{hemi} w_{hemi}(T) h_{hemi}(T) + x_{lign} w_{lign}(T) h_{lign}(T) \quad \text{Eq. SI - 3}$$

$$\dot{n}_{vi,subs}(T) = \sum_{j=1}^3 x_j \dot{n}_{vi,j}(T) = x_{cell} \dot{n}_{vi,cell}(T) + x_{hemi} \dot{n}_{vi,hemi}(T) + x_{lign} \dot{n}_{vi,lign}(T) \quad \text{Eq. SI - 4}$$

$$\dot{m}_{i,subs}(T) = \frac{\dot{n}_{vi,subs}(T) PM_i}{V_{m,std}} \quad \text{Eq. SI - 5}$$

$$h_{subs}|_{A-B} = \int_{T_a}^{T_h} h_{subs}(T) dt = \int_{T_a}^{T_h} h_{subs}(T) \frac{dT}{TR} \quad \text{Eq. SI - 6}$$

$$H_{subs}|_{A-B} = m_{subs} h_{subs}|_{A-B} \quad \text{Eq. SI - 7}$$

$$h_{comb}|_{K-J} = \int_{T_k}^{T_j} \sum_{i=1}^l \dot{m}_i(T) LHV_i dt = \int_{T_k}^{T_j} \sum_{i=1}^l \dot{m}_i(T) LHV_i \frac{dT}{TR'} \quad \text{Eq. SI - 8}$$

$$(m_{hc} + m_{lbs} + m_{tar}) + m_{pyrogas} = (m_{hc} + m_{lbs} + m_{tar}) + \sum_{i=1}^l \int_{RT}^{T_{fin}} \dot{m}_i(T) \frac{dT}{TR'} = 1 \quad \text{Eq. SI - 9}$$

$$h_{subs}|_{K-J} = \varphi \Delta H_{wat,(RT-T_{ev})} + (1 - \varphi) \Delta H_{subs,(RT-T_{pyr})} \quad \text{Eq. SI - 10}$$

$$\Delta H_{wat,(RT-T_{ev})} = \bar{c}_{wat}^-(T_{ev} - RT) + \Delta H_{ev,wat}(T_{ev}) \quad \text{Eq. SI - 10a}$$

$$\Delta H_{subs,(RT-T_{pyr})} = \bar{c}_{p,subs}^-(T_{pyr} - RT) + \int_{T_{pyr}}^{T_{fin}} h_{subs}(T) \frac{dT}{TR} \quad \text{Eq. SI - 10b}$$

η_{pyr}

$$\eta_{pyr} = \frac{m_{pyrogas} LHV_{pyrogas} + m_{tars}^{\sim} LHV_{tars} + m_{hc} LH}{m_{subs} \left[\left(\sum_{j=1}^3 x_j LHV_j \right) + |DH_{pyr}| \right]} \quad \text{Eq. SI - 11}$$

List of Symbols and abbreviations

Symbols

\dot{h}	[W g ⁻¹]	reaction specific heat release/uptake rate
h	[J g ⁻¹]	reaction specific heat release/uptake
i	-	fibres components index (from 1 to 3)
j	-	volatile gases components index (from 1 to 6)
H	[J]	reaction total heat release/uptake
LHV	[MJ kg ⁻¹]	Low Heating Value
\dot{m}	[g min ⁻¹]	mass flow rate (absolute or specific to substrate initial mass)
m	[g],[kg]	mass (absolute or specific to substrate initial mass)
\dot{n}_v	[ml min ⁻¹]	volume flowrate (absolute or specific to substrate initial mass)
PM	[g mol ⁻¹]	molecular weight
t	[s], [min]	time
T	[°C]	temperature
TR	[°C min ⁻¹]	temperature rate
V_m	[l mol ⁻¹]	molar volume
w	[%]	residual mass (as a fraction of the original specimen mass)
x	[%]	mass fraction

Subscripts

as	as received
cell	cellulose
comb	combustion
hemi	hemicellulose
lbs	low boiling substances (including water)
lign	lignin
std	standard
subs	substrate
wat	water