Electronic Supplementary Information

Synthesis of Energetic Polynorbornene with Pendant Bis-Azidoacetyloxymethyl Groups (PNBAA)

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Entry	n _{CPD} :n _{diol}	Temperature(°C)	Time(h)	Yield (%)
1	1:1	100	5	<10
2	1:1	150	5	<10
3	1:1	170	5	<20
4	1:1	190	5	45
5	1:1	200	5	60
6	1:1	250	5	55
7	1.05:1	200	5	55
8	1.2: 1	200	5	40
9	2.5: 1	200	5	41
10	1: 1.2	200	5	50
11	1:1	200	1	<10
12	1:1	200	2	<10
13	1:1	200	3	30
14	1:1	200	4	50

Table S1 : The optimization on solvent-free Diels-Alder reaction

Table S2 The comparison of 1H-NMR between 3 and reported data

norbornene diolreported by Paul R. Hanson (Norbornene-dimethanol (3)(in	Δδ
in CDCl ₃)	CDCl ₃)	
6.02(H-1)	6.00(H-1)	0.02
4.07 (H-6)	4.33 (H-6)	0.26
3.61 (H-4)	3.59-3.56 (H-4)	0.02
3.39-3.30(H-5)	3.33 (H-5)	0.00
2.78 (H-2)	2.76 (H-2)	0.02
2.55-2.47(H-3)	2.50(H-3)	0.00
1.42-1.35(H-7)	1.34-1.40(H-7)	0.02
Ref: J. Org. Chem. 2007, 72, 3194-3198.		

norbornene diolreported by Paul R.	Diels-Alder-norbornene diol(in	$ riangle \delta$
Hanson (in CDCl3)	CDCl3)	
45.0 (C-c)	45.2 (C-c)	0.2
46.5 (C-b)	46.6 (C-b)	0.1
49.9 (C-e)	49.9 (C-e)	0.0
63.4 (C-d)	63.6 (C-d)	0.1
134.7 (C-a)	134.8 (C-a)	0.1
Ref: J. Org. Chem. 2007, 72, 3194-3198.		

Table S3 The comparison of ¹³C-NMR between 3 and reported data

Table S4 Elemental analysis on NBCA, PNBCA and PNBAA

	Mass Percentage (%)					
Sample	mple		Found		Theoretical ^a	
	Ν	С	Η	Ν	С	Η
NBCA	0	51.04	5.49	0	50.83	5.25
PNBCA	0	50.14	5.30	0	50.83	5.25
PNBAA	24.10	48.90	5.27	26.24	48.75	5.04

[a] Theoretical values were calculated assuming an ideal structure of PNBCA for the elemental analysis.

Table S5 ICP-Ms analysis on PNBCA for Ru content

PNBCA ^[a]	Ru Content (ppm)
First -precipitating ^[b]	966
Second -precipitating ^[b]	823
Third -precipitating ^[b]	790

[a] PNBCA was synthesis through solvent free way with 1 mol% of Grubbs' I catalyst. [b] 500 mg PNBCA was dissolve in 5 ml 3-pentone and added 5 ml methanol to precipitate.



Figure S1 : The ¹H-NMR spectrum of norbornene-dimethanol (3) in $CDCl_3$



Figure S2 : The ¹³C-NMR spectrum of norbornene-dimethanol (3) in CDCl₃



Figure S3 : The gCOSY-NMR spectrum of norbornene-dimethanol (3) in CDCl₃



Figure S4 : The gHSQC-NMR spectrum of norbornene-dimethanol (3) in CDCl₃



Figure S5 : The FT-IR spectrum of norbornene-dimethanol (3)

Entry-12		99%
Entry-11		85%
Entry-10	l l.	500
Entry-9		67
Entry-8		679
Entry-7		50%
Entry-6		8%
Entry-5		27%
Entry-4	l	57%
Entry-3		54%

Figure S6 Array NMR study on the optimization on chloroacetylation on diol 3



Figure S7 The ¹H-NMR spectrum of NBCA (4) in CDCl₃



Figure S8 : The¹³C-NMR spectrum of NBCA (4)in CDCl₃



Figure S9 :The gCOSY-NMR spectrum of NBCA (4)in CDCl₃



Figure S10 : The gHSQC-NMR spectrum of NBCA (4)in CDCl₃



Figure S11 : TheFT-IR spectrum of NBCA (4)



Figure S12 : The HR-MS spectrum of NBCA(4)



Figure S13:The DSCgraph of NBCA (4)



Figure S14: The TGAgraph of NBCA (4)



Figure S15 : The ¹H-NMR spectrum of PNBCA (5) in CDCl₃



Figure S16:The¹³C-NMR spectrum of PNBCA (5) in CDCl₃



Figure S17: The array ¹H-NMR on ROMP reaction in CDCl₃ (Entry 1, Table 2)



Figure S18: The array ¹H-NMR on ROMP reaction in CDCl₃ (Entry 2, Table 2)



Figure S19: The array ¹H-NMR on ROMP reaction in CDCl₃ (Entry 4, Table 2)



Figure S20 Photograph on solvent-free ROMP reaction on NBCA (4)



Figure S21: Photograph of ROMP adduct before (top) and after (bottom) removal of Ru comtaminant



Figure S22 : The FT-IR spectrum f PNBCA (5)



Figure S23 : The ¹H-NMR spectrum of PNBAA (6) in $CDCl_3$



Figure S24 : The FT-IR spectrum f PNBAA (6)



Figure S25 The comparison among the FT-IR spectra of monomers and polymers



Figure S26 The GPC graph of PNBCA with different Mn



Figure S27 The comparison of FT-IR spectra between NBCA and PNBCA



Figure S28: The TG and DTG graph of PNBAA (6)



Figure S29 :The DTG graph of PNBCA(5) and PNBAA (6)



Figure S30 : The DSC graph of PNBCA(5) and PNBAA (6) from 80 $^\circ C$ to 600 $^\circ C$