## **Electronic Supplementary Information**

## Fully Solution-Processable Cu<sub>2</sub>O-BiVO<sub>4</sub> Photoelectrochemical Cells for Bias-Free Solar Water Splitting

Hyunwoo Kim, <sup>†</sup> Sanghyun Bae, <sup>†</sup> Dasom Jeon, and Jungki Ryu<sup>\*</sup>

Department of Energy Engineering, School of Energy and Chemical Engineering, Ulsan National Institute of Science and Technology (UNIST), Ulsan 44919, Republic of Korea

<sup>†</sup>These authors contributed equally to this work.

\*To whom correspondence should be addressed: jryu@unist.ac.kr



**Fig. S1** Cyclic voltammograms showing the HER and OER catalytic activity of NiPOM and CoPOM, respectively.



**Fig. S2** LSV curves showing the effect of the number of HER BLs on the performance and stability of the Cu<sub>2</sub>O photocathode. LSV curves were measured under periodic and continuous illumination.



**Fig. S3** *Ex situ* XPS spectra of the bare Cu<sub>2</sub>O and the Cu<sub>2</sub>O with 10 BL of the HER CMs before and after the PEC hydrogen evolution test at an applied bias of 0.35V vs. RHE for 1h. (Black dotted line for Cu<sub>2</sub>O, Blue dotted line for Cu<sub>2</sub>O-Ni10BL)



**Fig. S4** Performance of Cu<sub>2</sub>O photocathodes modified with Pt nanoparticles (Cu<sub>2</sub>O-Pt). (a) Chronoamperograms of various Cu<sub>2</sub>O photocathodes—the bare Cu<sub>2</sub>O, Cu<sub>2</sub>O with Pt nanoparticles, and Cu<sub>2</sub>O with the HER CMs—are shown for comparison. (b, c) SEM images of the Cu<sub>2</sub>O-Pt photocathode before and after the PEC test at an applied bias of 0.35 V vs. RHE for 1h.

R (Ω)	Rs	$R_{I}$	$R_2$
Cu <sub>2</sub> O	30.26	79.5	992.3
Cu <sub>2</sub> O-15BL	32.73	38.25	589.3

**Table S1.** The fitting results for the EIS spectra (Fig. 4a) of the  $Cu_2O$  photocathode with and without 15 BL of the HER CMs.



**Fig. S5** (a) Comparison between the charge separation efficiency for the bare  $Cu_2O$  and the  $Cu_2O$  with 15 BL of the HER CMs. (b) Chronoamperograms measured for the calculation of the charge separation efficiency. 0.1 M sodium persulfate (Na<sub>2</sub>S<sub>2</sub>O<sub>8</sub>) was used as an electron scavenger.



**Fig. S6** TEM, STEM and elemental mapping analyses of BiVO<sub>4</sub> photoanodes before and after the modification with 10 BLs of the OER CMs.



**Fig. S7** Formation of the OER CMs was investigated with (a) UV-vis absorption spectroscopy, (b) QCM analysis, (c) ellipsometry, and (d) XPS.



**Fig. S8** LSV curves showing the effect of (a) the number of OER BL and (b) illuminationdirection on the performance of the BiVO<sub>4</sub> photoanode.

	$R_{S}(\Omega)$	$R_{I}(\Omega)$	CPE1 (F)	$R_2$ ( $\Omega$ )	<i>CPE2</i> (F)
BiVO4	26.51	360.3	7.66 x 10 <sup>-5</sup>	2.00 x 10 <sup>5</sup>	7.95 x 10 <sup>-5</sup>
BiVO <sub>4</sub> -5BL	31.95	173.3	4.58 x 10 <sup>-5</sup>	988.4	1.75 x 10 <sup>-4</sup>
BiVO4-10BL	21.58	123.2	4.00 x 10 <sup>-5</sup>	727.6	4.39 x 10 <sup>-4</sup>
BiVO4-15BL	24.42	134.2	6.90 x 10 <sup>-5</sup>	832.4	3.25 x 10 <sup>-4</sup>

**Table S2.** The fitting results for the EIS spectra (Fig. 5e) of the BiVO<sub>4</sub> photoanode with and without OER CMs.



Figure S9. (a) Comparison between the charge separation efficiency for the bare  $BiVO_4$  and the  $BiVO_4$  with 10 BL of the OER CMs. (b, c) Chronoamperograms measured for the calculation of the charge separation efficiency. 0.5 M sodium sulfite (Na<sub>2</sub>SO<sub>3</sub>) was used as a hole scavenger.