

Supplementary information for “Development of $\text{H}_3\text{PW}_{12}\text{O}_{40}/\text{CeO}_2$ catalyst for bulk ring-opening polymerization of cyclic carbonate”

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Table S1. Effect of the pre-calcination temperature of CeO₂

Pre-calcination temperature / °C	BET surface area / m ² g ⁻¹	Conv. ^{a)} / %	Ether linkage ^{a)} selectivity / %	$M_{n, theo}$ ^{b)} / g mol ⁻¹	$M_{n, GPC}$ ^{c)} / g mol ⁻¹	M_w/M_n ^{c)}
400	119	12	0	3,700	3,600	1.21
500	109	18	0	5,700	5,100	1.31
600	73	20	0	6,300	6,200	1.41
700	62	15	0	4,700	4,500	1.35
800	43	4	0	1,300	1,400	1.19
900	26	0	0	-	-	-

Reaction conditions: 1 wt% H₃PW₁₂O₄₀/CeO₂ 20 mg (pretreated by drying under reduced pressure, 180 °C, 0.5 h), TMC 3 mmol, methyl iodide 0.01 mmol, reaction temperature 60 °C, reaction time 2 h.

^{a)} calculated by ¹H NMR, ^{b)} calculated by $[TMC]_0/[CH_3I]_0 \times (\text{conv.}) \times (\text{molecular weight of TMC}) + (\text{molecular weight of } CH_3I)$, ^{c)} determined by GPC in chloroform relative to polystyrene standards.

Table S2. Effect of the pretreatment temperature

Pretreatment temperature / °C	Conversion ^{a)} / %	Ether linkage ^{a)} selectivity / %	$M_{n, theo}$ ^{b)} / g mol ⁻¹	$M_{n, GPC}$ ^{c)} / g mol ⁻¹	M_w/M_n ^{c)}
-	7	0	2,100	660	1.29
100	8	0	2,700	2,000	1.30
120	8	0	2,700	3,500	1.31
140	11	0	3,500	3,400	1.33
160	15	0	4,600	4,600	1.40
180	20	0	6,300	6,200	1.41
200	8	0	2,700	2,800	1.30

Reaction conditions: 1 wt% H₃PW₁₂O₄₀/CeO₂ 20 mg (pretreated by drying under reduced pressure, 0.5 h), TMC 3 mmol, methyl iodide 0.01 mmol, reaction temperature 60 °C, reaction time 2 h.

^{a)} calculated by ¹H NMR, ^{b)} calculated by $[TMC]_0/[CH_3I]_0 \times (\text{conv.}) \times (\text{molecular weight of TMC}) + (\text{molecular weight of } CH_3I)$, ^{c)} determined by GPC in chloroform relative to polystyrene standards.

Table S3. Effect of the loaded amount of H₃PW₁₂O₄₀ on CeO₂

Loaded amount of H ₃ PW ₁₂ O ₄₀ / wt%	BET surface area / m ² g ⁻¹	Conversion ^{a)} / %	Ether linkage ^{a)} selectivity / %	$M_{n, theo}$ ^{b)} / g mol ⁻¹	$M_{n, GPC}$ ^{c)} / g mol ⁻¹	M_w/M_n ^{c)}
0.1	66	19	0	5,900	19,000	1.26
0.5	70	20	0	6,300	14,000	1.29
1	73	20	0	6,300	6,200	1.41
2	70	3	0	1,000	470	1.39
5	68	2	0	750	250	1.36

Reaction conditions: H₃PW₁₂O₄₀/CeO₂ 20 mg (pretreated by drying under reduced pressure, 180 °C, 0.5 h), TMC 3 mmol, methyl iodide 0.01 mmol, reaction temperature 60 °C, reaction time 2 h.

^{a)} calculated by ¹H NMR, ^{b)} calculated by $[TMC]_0/[CH_3I]_0 \times (\text{conv.}) \times (\text{molecular weight of TMC}) + (\text{molecular weight of } CH_3I)$

weight of CH₃I), ^c) determined by GPC in chloroform relative to polystyrene standards.

Table S4. Catalyst recycle test

Usage time	Conversion ^{a)} / %	Ether linkage ^{a)} selectivity / %	$M_{n, theo}$ ^{b)} / g mol ⁻¹	$M_{n, GPC}$ ^{c)} / g mol ⁻¹	M_w/M_n ^{c)}
1	21	0	6,400	6,200	1.41
2	17	0	5,100	690	1.18
2 ^{d)}	17	0	5,100	1,300	1.23

Reaction conditions: H₃PW₁₂O₄₀/CeO₂ 20 mg (pretreated by drying under reduced pressure, 180 °C, 0.5 h), TMC 3 mmol, initiator 0.01 mmol, reaction temperature 60 °C, reaction time 2 h.

^{a)} calculated by ¹H NMR, ^{b)} calculated by [TMC]₀/[CH₃I]₀ × (conv.) × (molecular weight of TMC) + (molecular weight of CH₃I), ^{c)} determined by GPC in chloroform relative to polystyrene standards, ^{d)} calcined at 200 °C for 0.5 h.

Table S5. Effect of initiator

Entry	Initiator	Conversion ^{a)} / %	Ether linkage ^{a)} selectivity / %	$M_{n, theo}$ ^{b)} / g mol ⁻¹	$M_{n, GPC}$ ^{c)} / g mol ⁻¹	M_w/M_n ^{c)}
1		21	0	6,400	6,200	1.41
2		12	0	3,900	3,300	1.38
3		3	0	1,000	220	1.56
4		0	0	140	-	-
5		15	0	4,600	2,400	1.18
6		1	0	450	-	-
7		3	0	1,000	1,600	1.12
8		10	0	3,000	430	1.55
9		11	0	3,400	290	1.85
10		12	0	5,000	2,900	1.77
11	None	3	0	-	640	1.36

Reaction conditions: H₃PW₁₂O₄₀/CeO₂ 20 mg (pretreated by drying under reduced pressure, 180 °C, 0.5 h), TMC 3 mmol, initiator 0.01 mmol, reaction temperature 60 °C, reaction time 2 h.

^{a)} calculated by ¹H NMR, ^{b)} calculated by [TMC]₀/[initiator]₀ × (conv.) × (molecular weight of TMC) + (molecular weight of initiator), ^{c)} determined by GPC in chloroform relative to polystyrene standards.

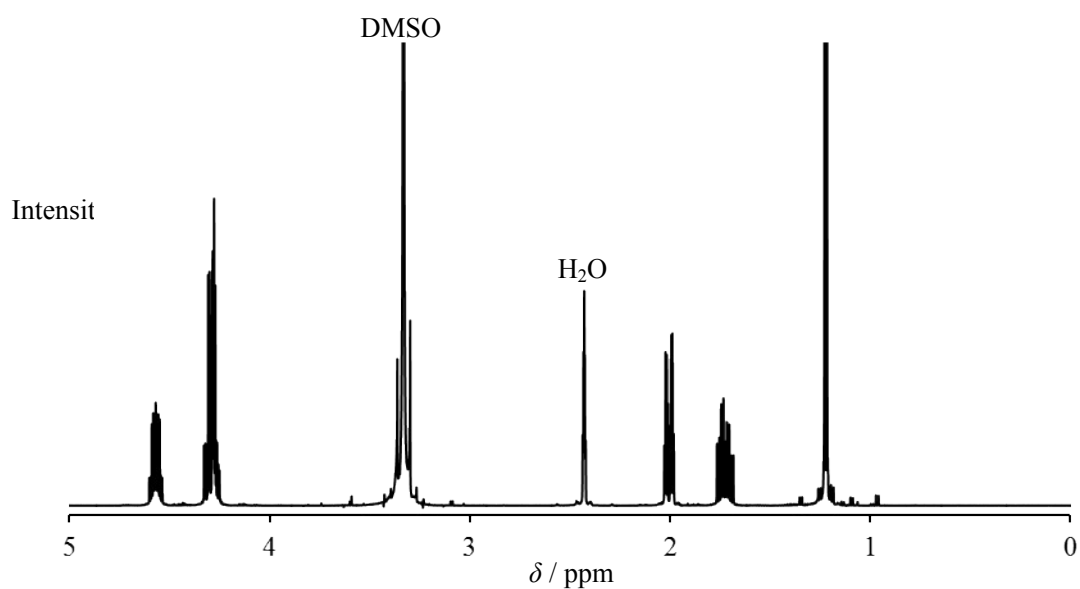


Figure S1. ¹H NMR spectra of 4-methyl-1,3-dioxan-2-one

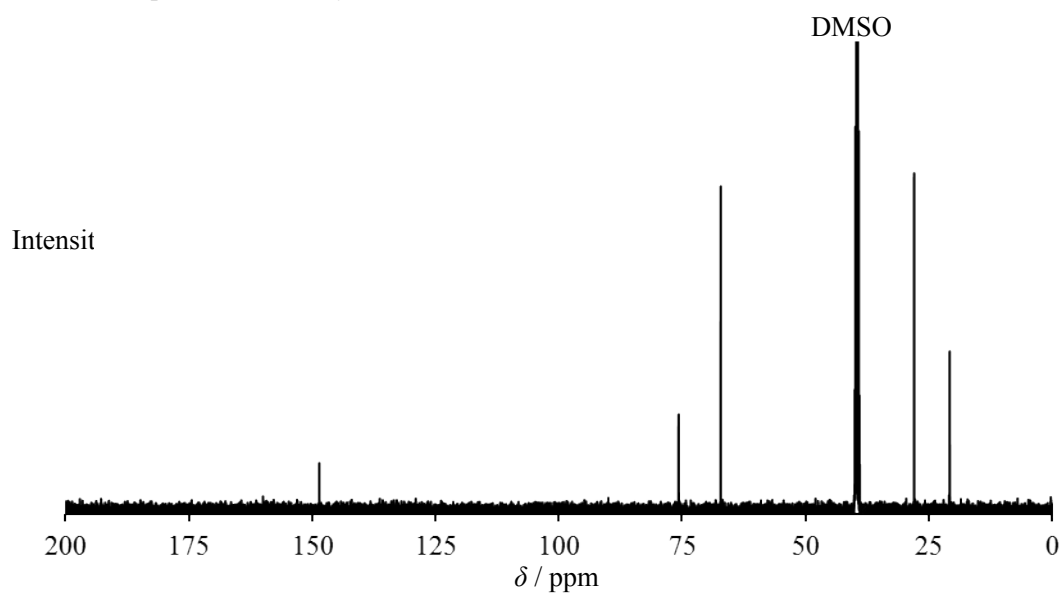


Figure S2. ¹³C NMR spectra of 4-methyl-1,3-dioxan-2-one

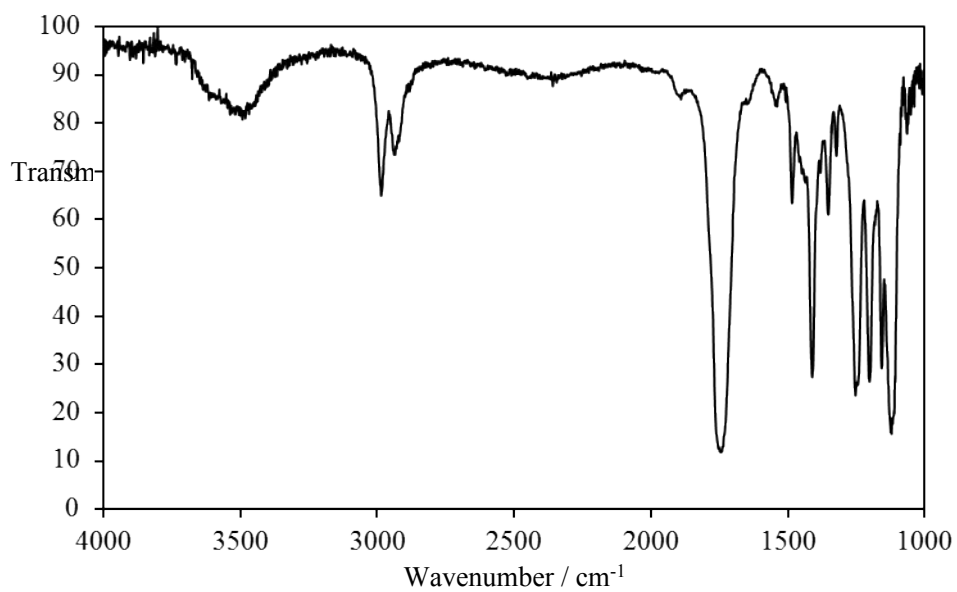


Figure S3. FTIR spectra of 4-methyl-1,3-dioxan-2-one

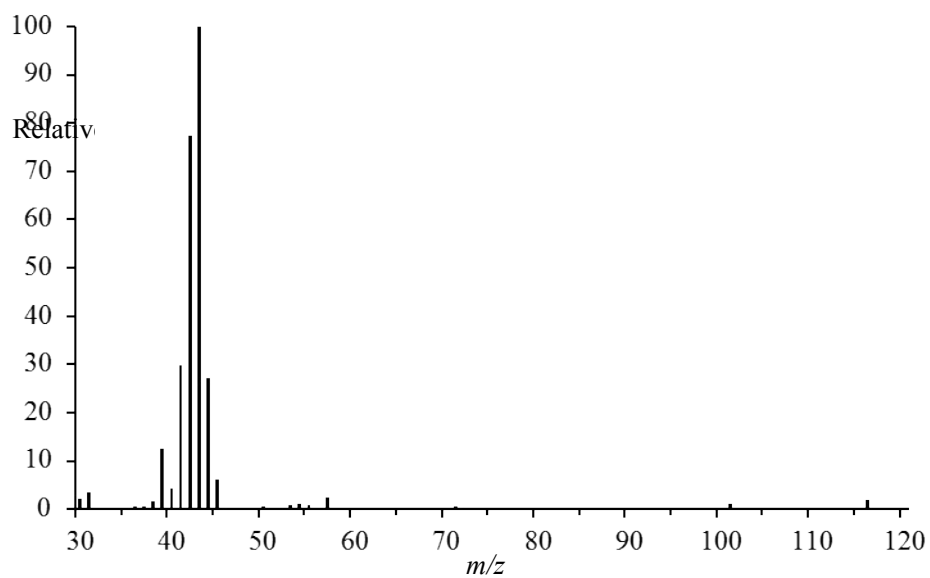


Figure S4. MS spectra of 4-methyl-1,3-dioxan-2-one

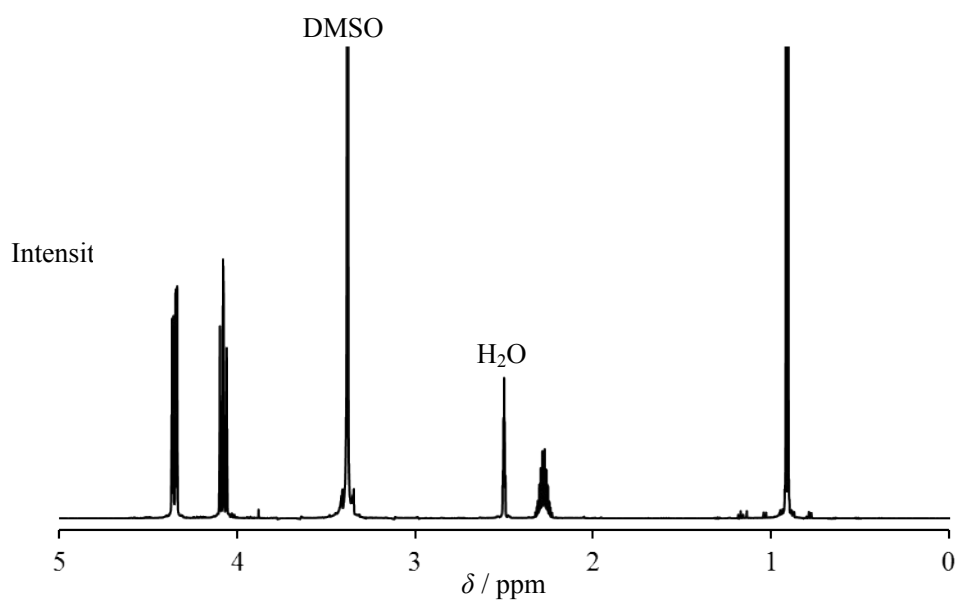


Figure S5. ^1H NMR spectra of 5-methyl-1,3-dioxan-2-one

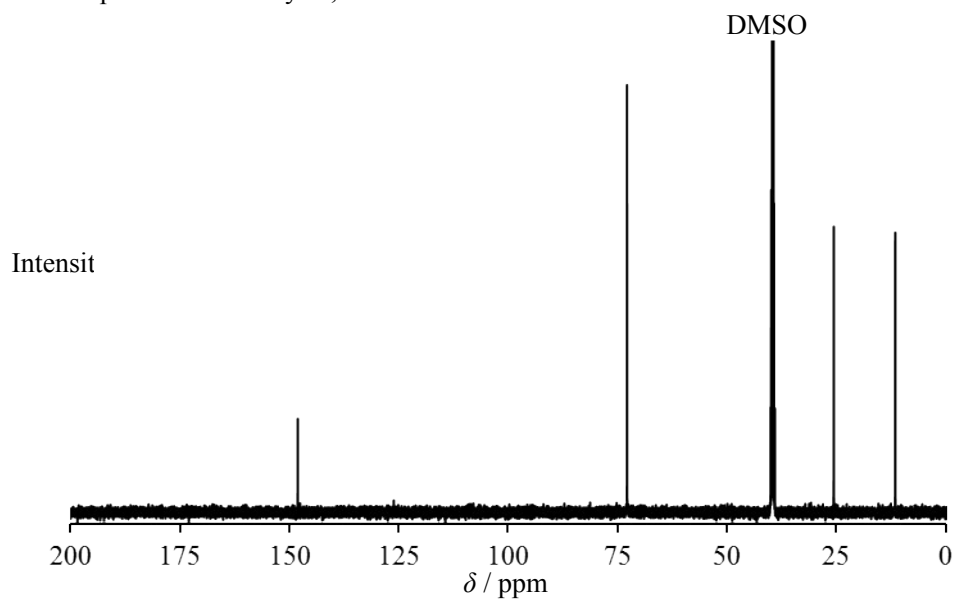


Figure S6. ^{13}C NMR spectra of 5-methyl-1,3-dioxan-2-one

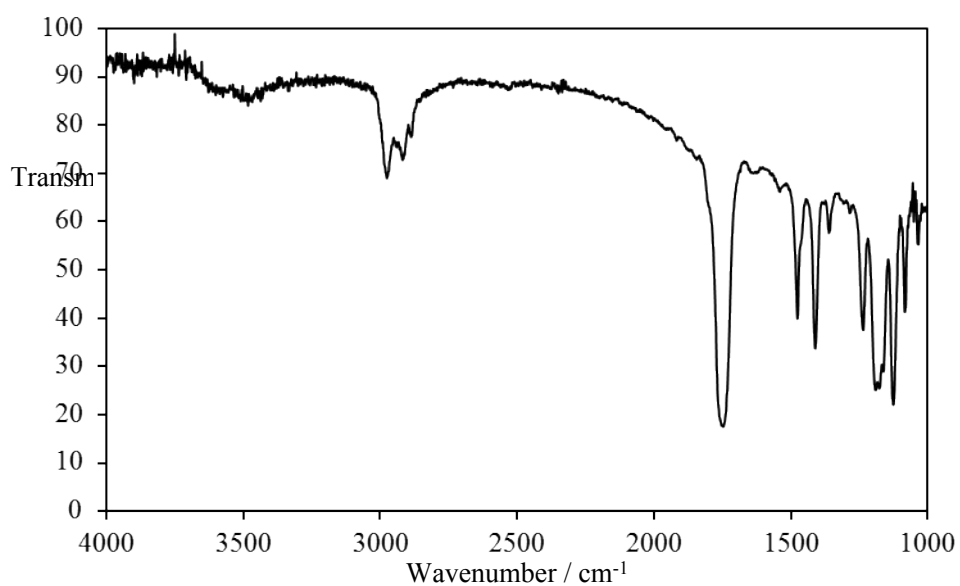


Figure S7. FTIR spectra of 5-methyl-1,3-dioxan-2-one

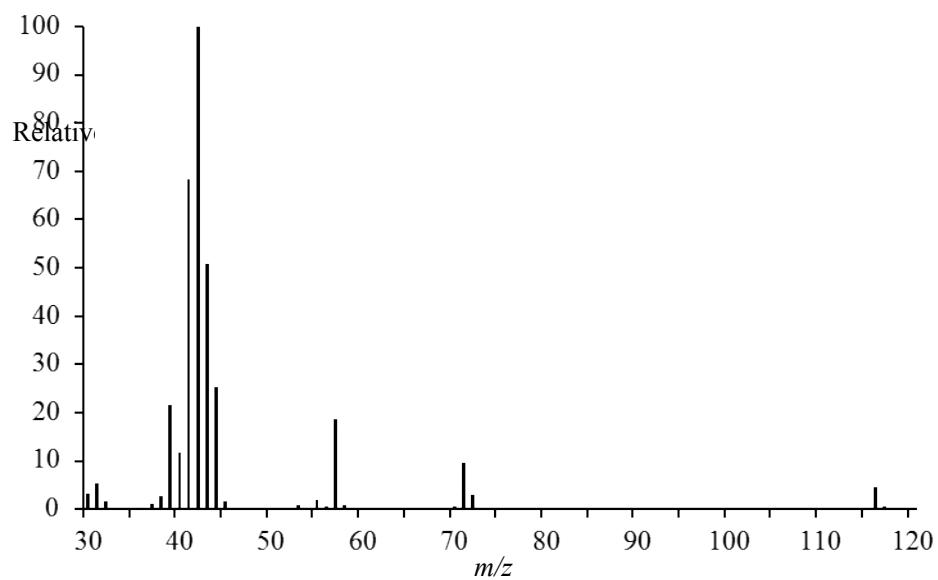


Figure S8. MS spectra of 5-methyl-1,3-dioxan-2-one

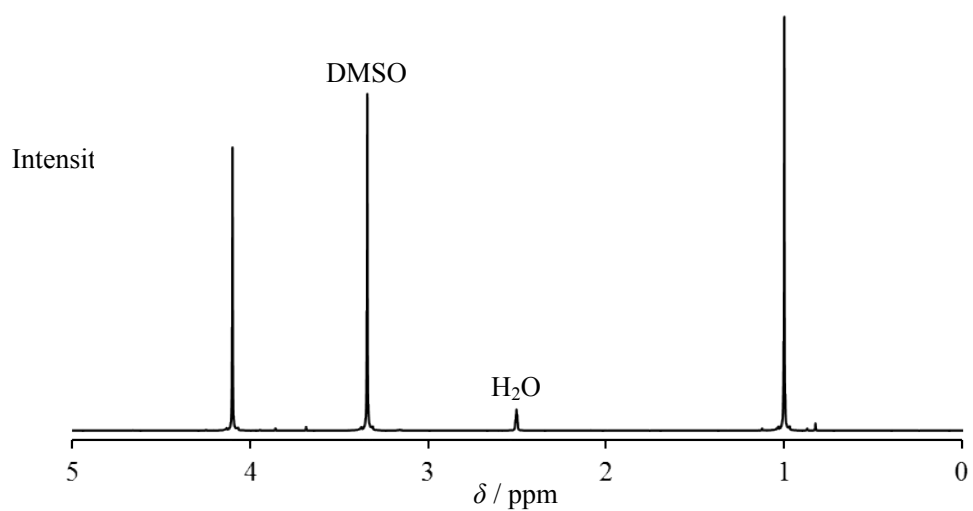


Figure S9. ^1H NMR spectra of 5,5-dimethyl-1,3-dioxan-2-one

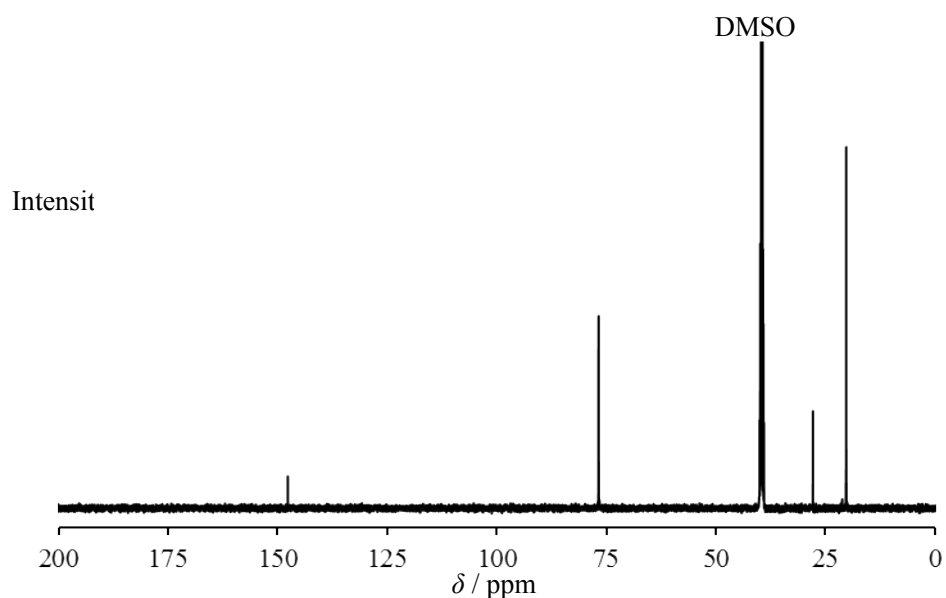


Figure S10. ^{13}C NMR spectra of 5,5-dimethyl-1,3-dioxan-2-one

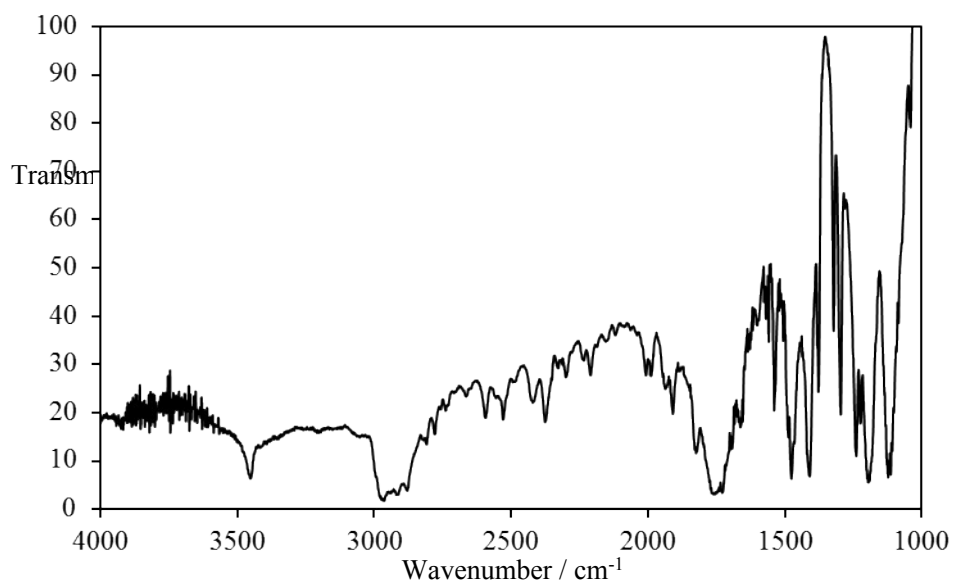


Figure S11. FTIR spectra of 5,5-dimethyl-1,3-dioxan-2-one

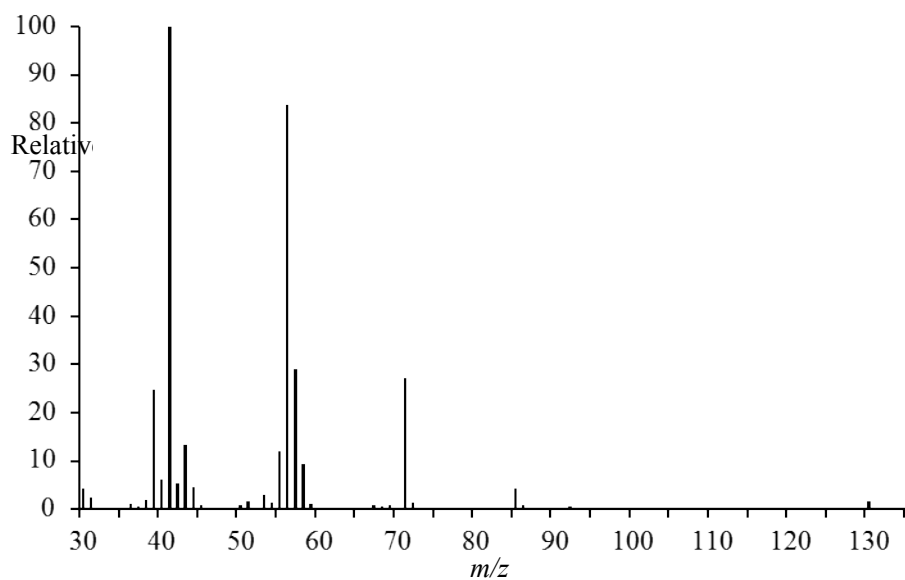


Figure S12. MS spectra of 5,5-dimethyl-1,3-dioxan-2-one

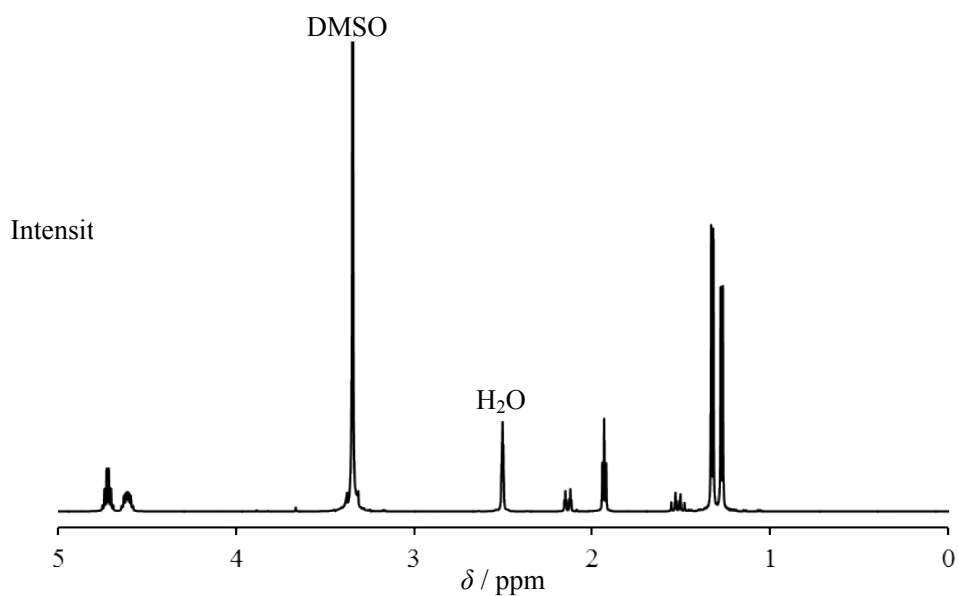


Figure S13. ¹H NMR spectra of 4,6-dimethyl-1,3-dioxan-2-one

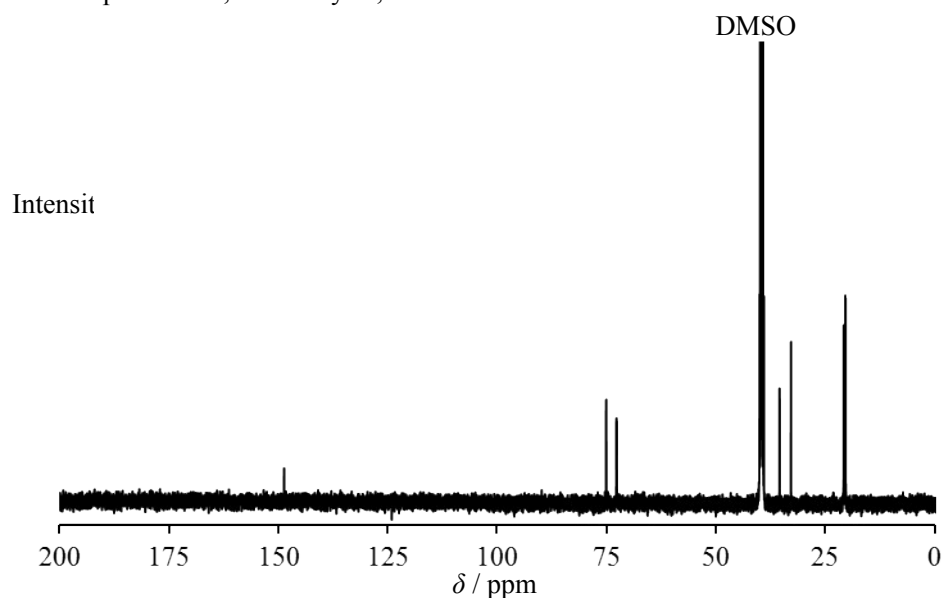


Figure S14. ¹³C NMR spectra of 4,6-dimethyl-1,3-dioxan-2-one

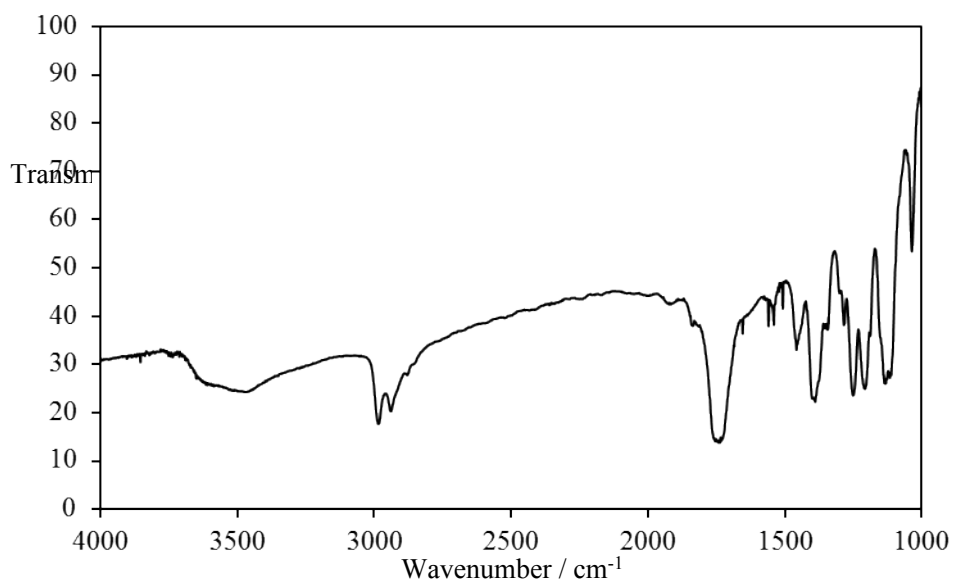


Figure S15. FTIR spectra of 4,6-dimethyl-1,3-dioxan-2-one

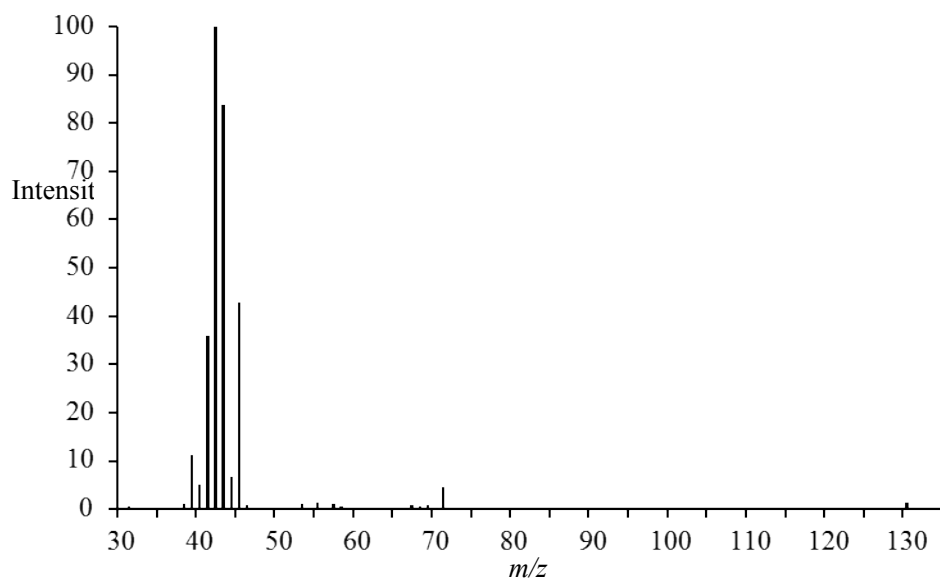


Figure S16. MS spectra of 4,6-dimethyl-1,3-dioxan-2-one

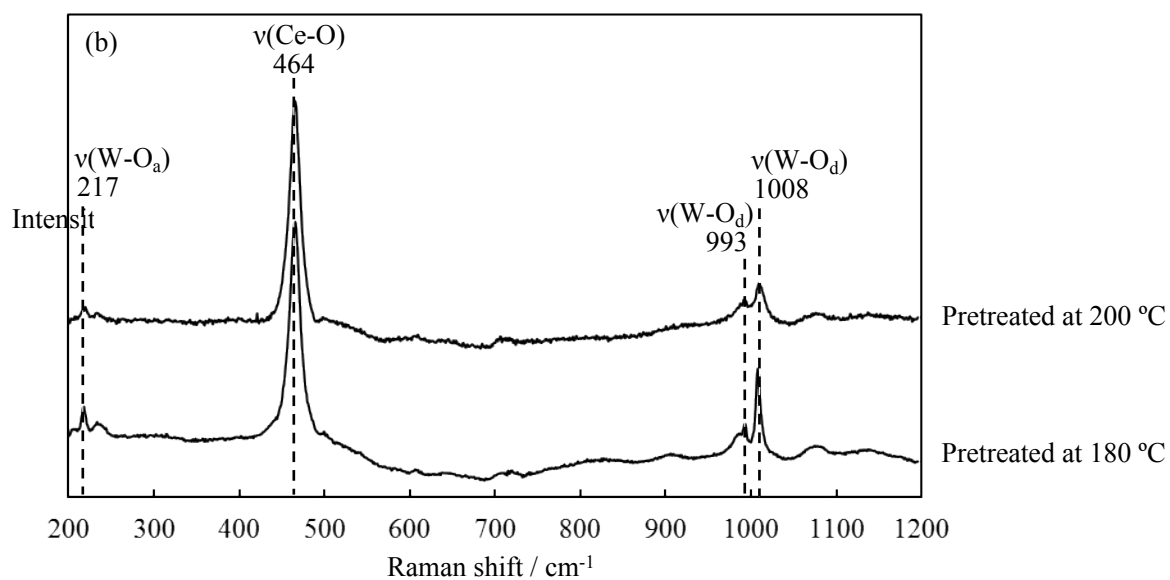


Figure S17. Raman spectra of 30 wt% $\text{H}_3\text{PW}_{12}\text{O}_{40}/\text{CeO}_2$ pretreated at different temperature

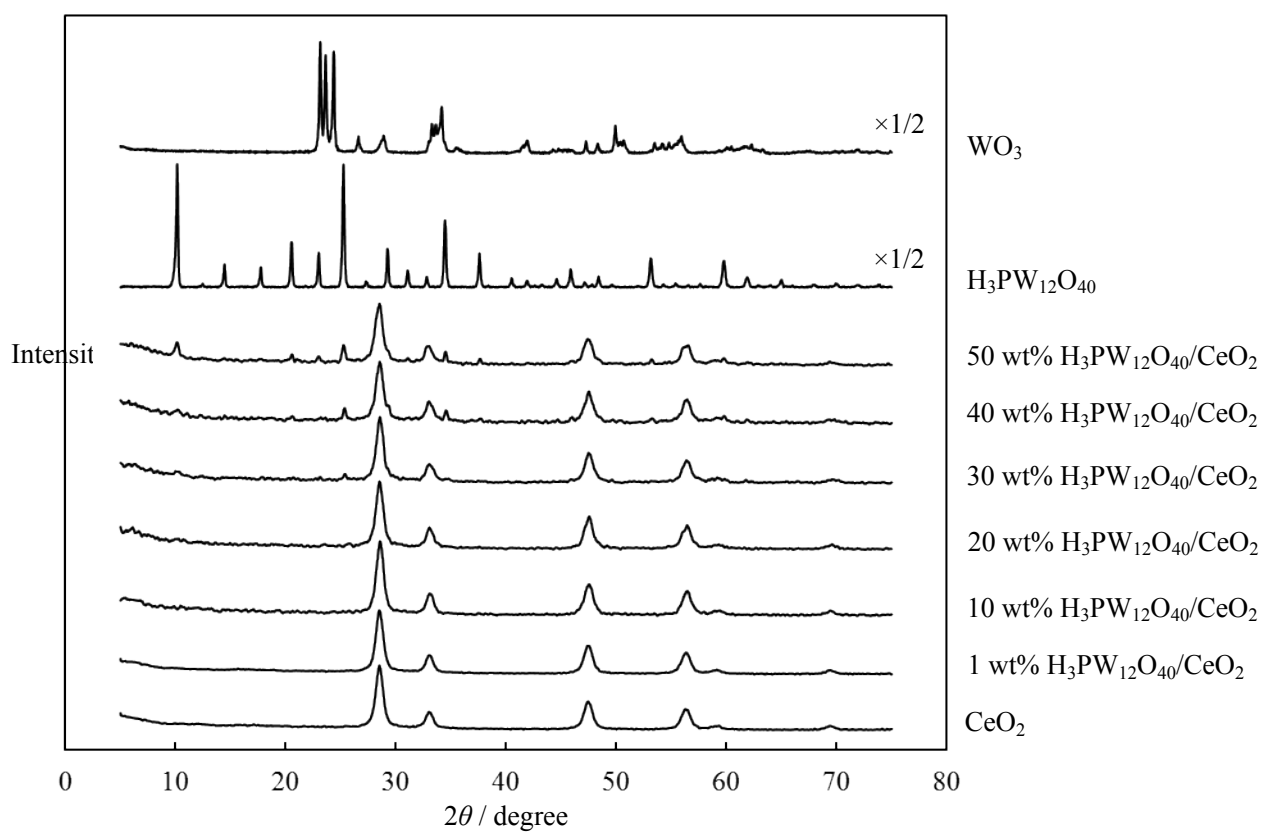


Figure S18. XRD profile of WO_3 , $\text{H}_3\text{PW}_{12}\text{O}_{40}$ and $\text{H}_3\text{PW}_{12}\text{O}_{40}/\text{CeO}_2$

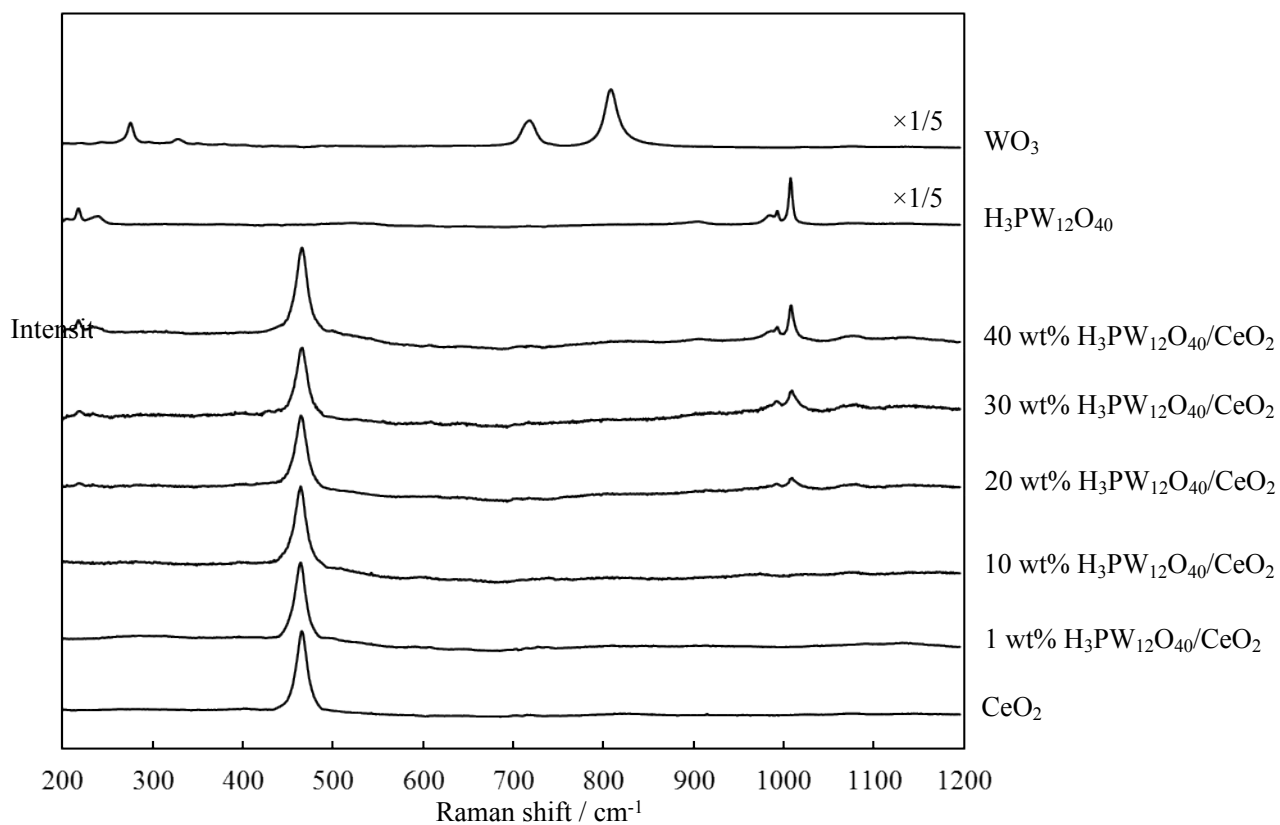


Figure S19. Raman spectra of WO_3 , $\text{H}_3\text{PW}_{12}\text{O}_{40}$ and $\text{H}_3\text{PW}_{12}\text{O}_{40}/\text{CeO}_2$

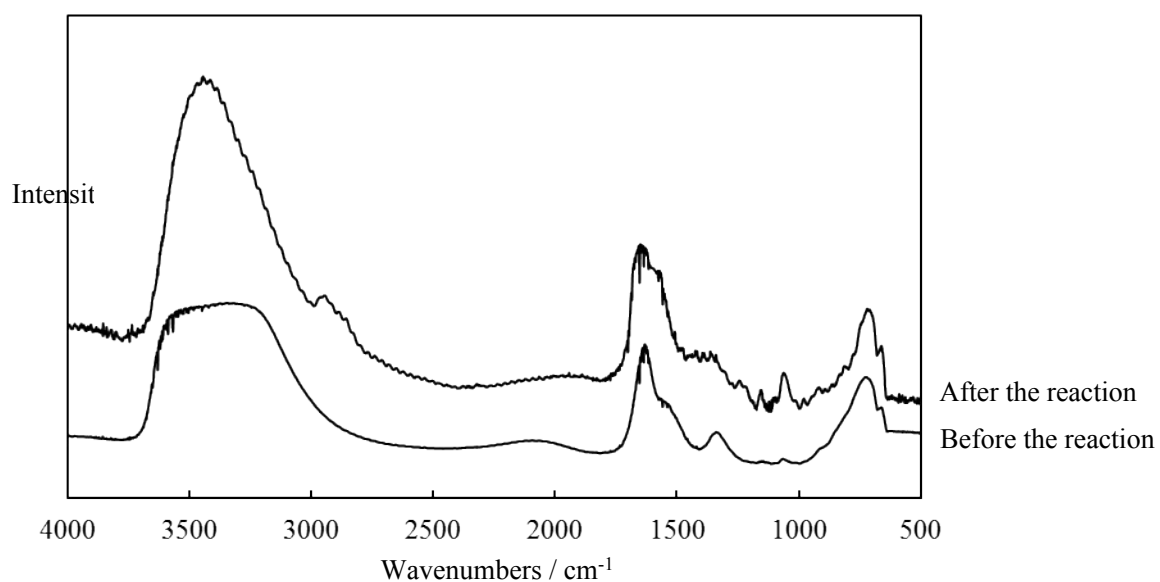


Figure S20. FTIR spectra of 1 wt% $\text{H}_3\text{PW}_{12}\text{O}_{40}/\text{CeO}_2$ catalyst before and after the reaction

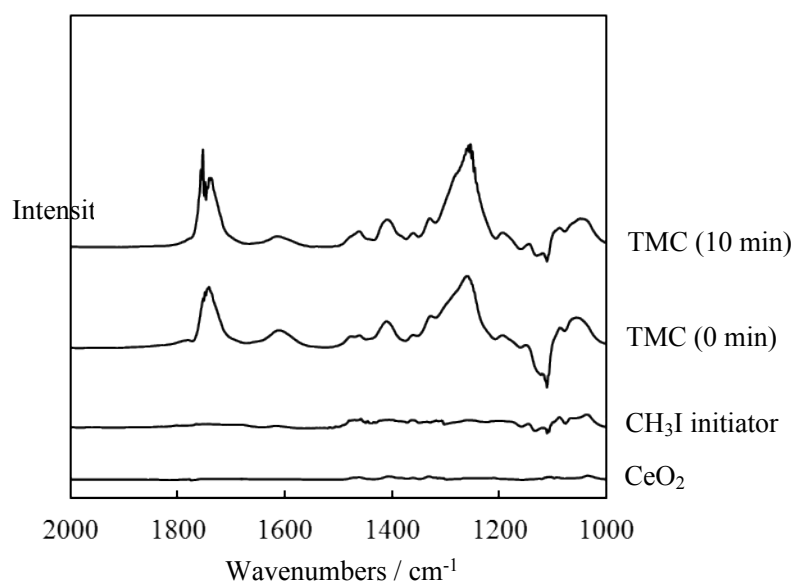


Figure S21. Differential FTIR spectra of CeO_2

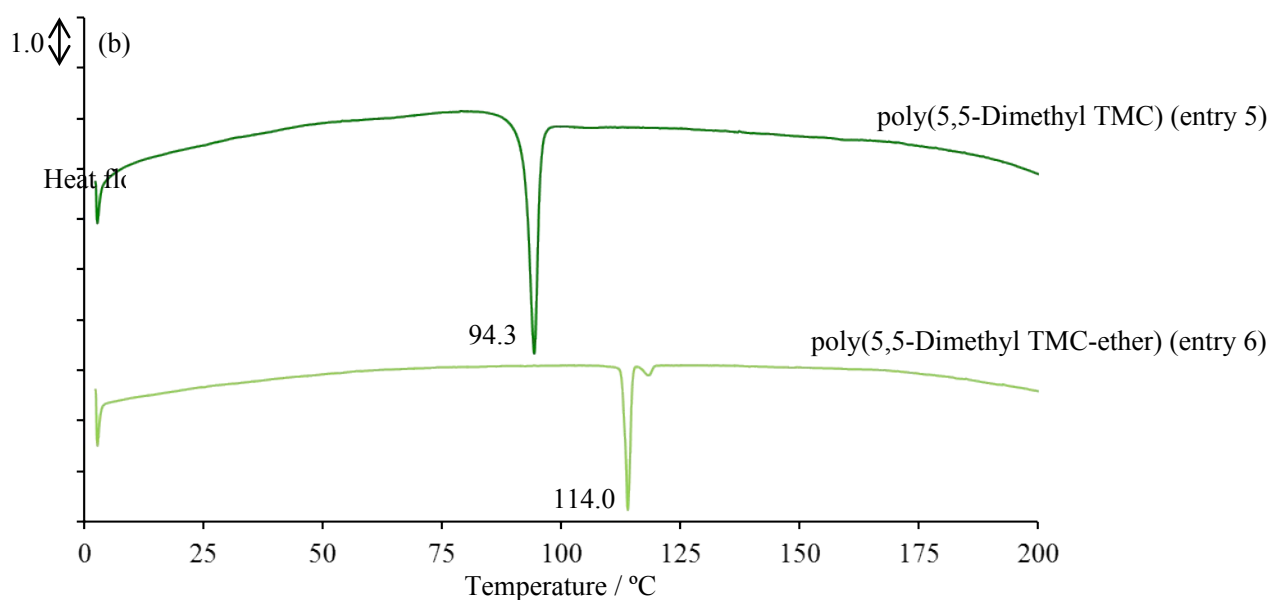
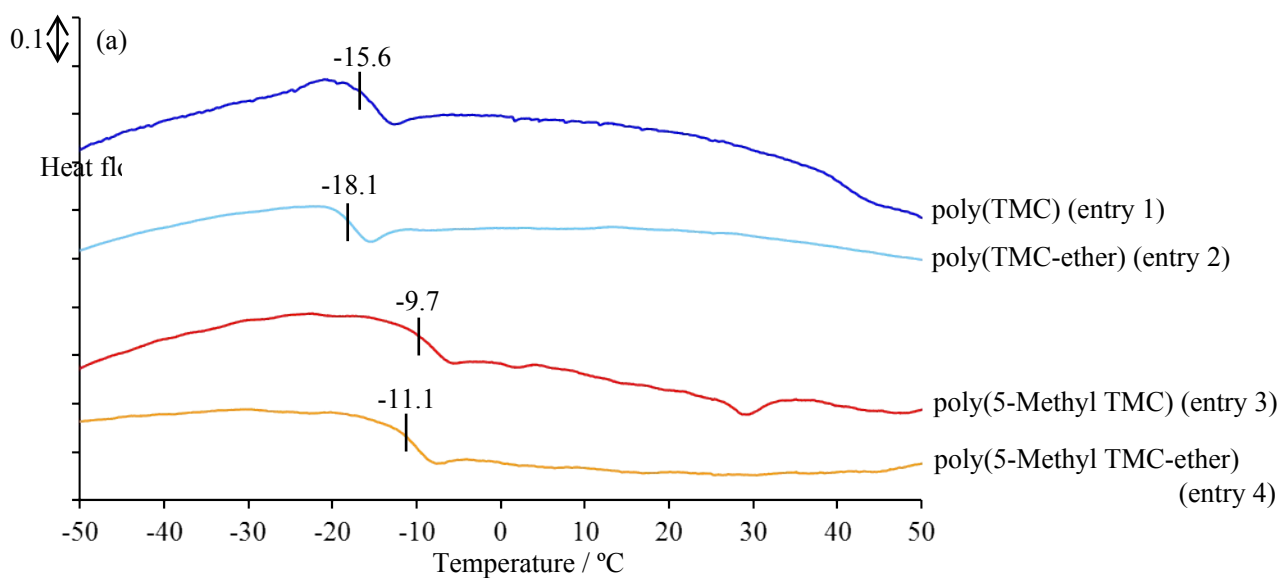


Figure S22. DSC profiles of (a) entries 1-4 (b) entries 5, 6 in Table 5

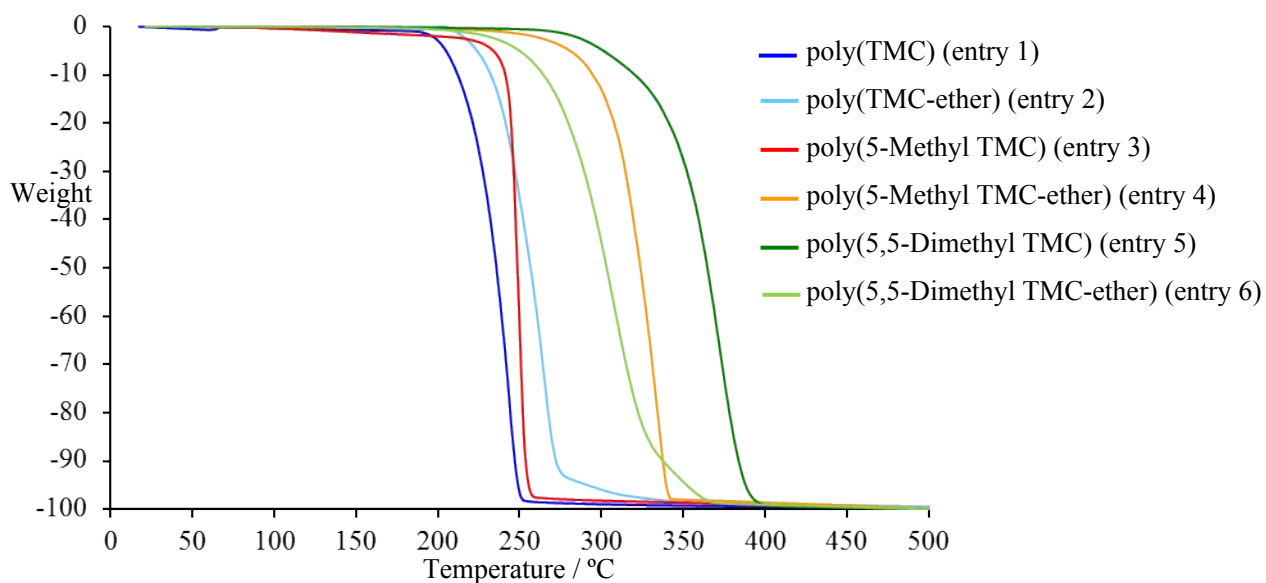


Figure S23. TG-DTA profiles of entries 1-6 in Table 5

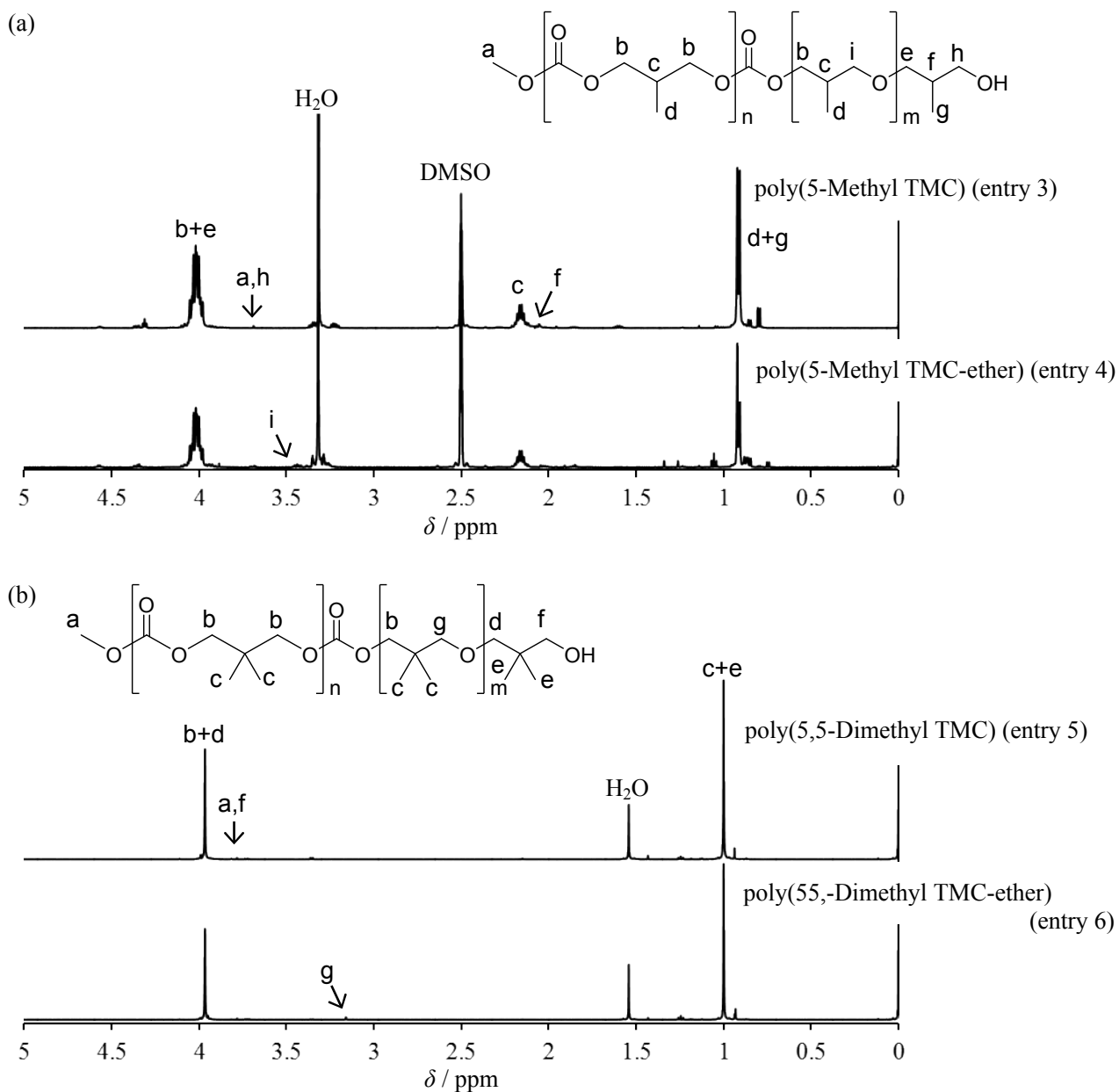


Figure S24. ^1H NMR spectra of (a) poly(5-Methyl TMC)s in $\text{DMSO}-d_6$ and (b) poly(5,5-dimethyl TMC)s in CDCl_3

(a) 220 $^\circ\text{C}$ (corresponds to 3% degradation)

(b) 330 $^\circ\text{C}$ (corresponds to 99% degradation)

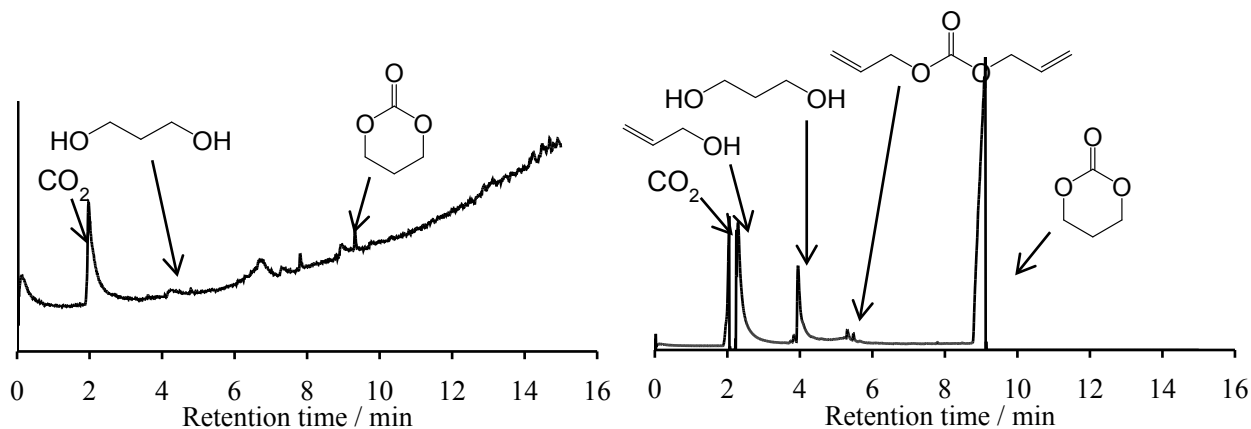


Figure S25. The chromatograms of Py-GCMS of poly(TMC-ether)