

**Supporting Information**

**High Performance Organic Sodium-ion Hybrid Capacitor Based on Nano  
Structured Disodium Rhodizinate Rivaling Inorganic Hybrid Capacitors**

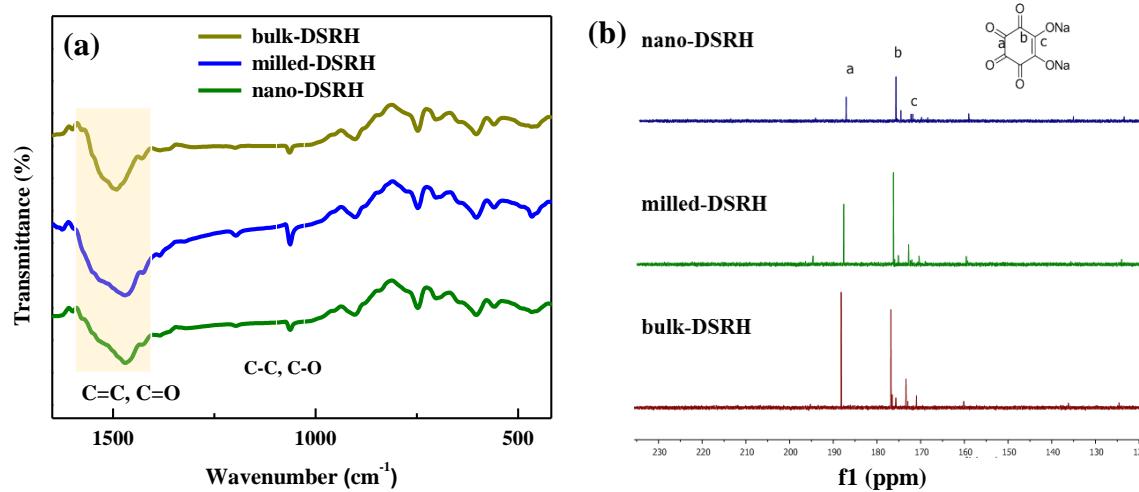
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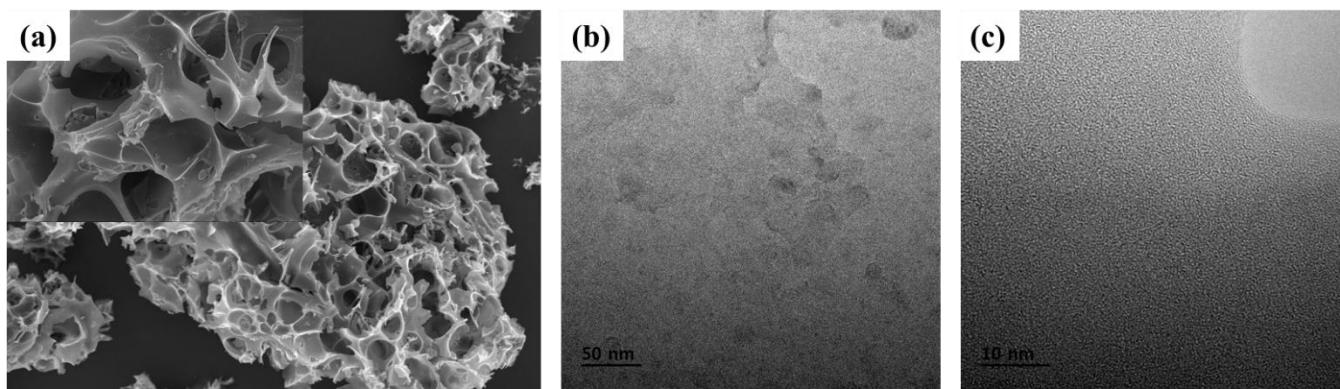
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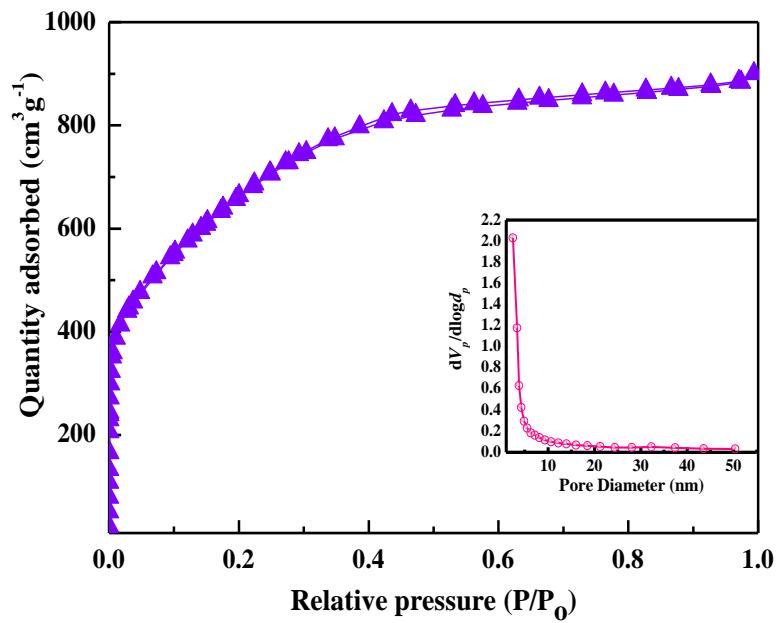
E-mail address: [leeys@chonnam.ac.kr](mailto:leeys@chonnam.ac.kr) (Y.S. Lee)



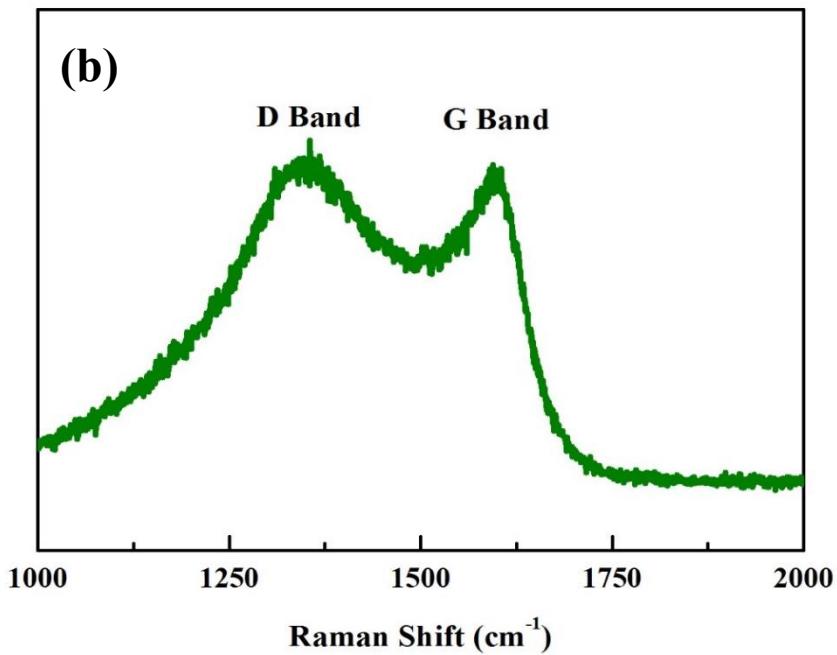
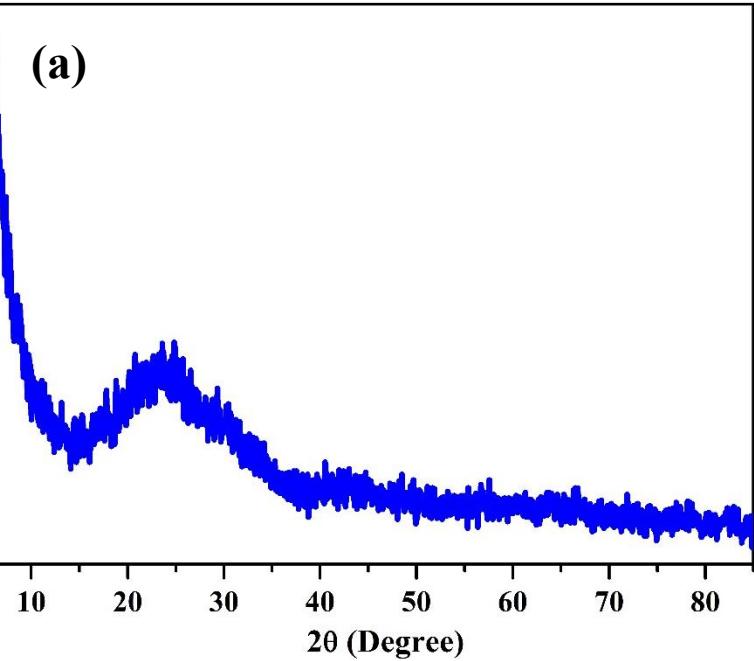
**Fig. S1** (a) FT-IR spectrum, and (b)  $^{13}\text{C}$ -NMR spectrum of various DSRH.



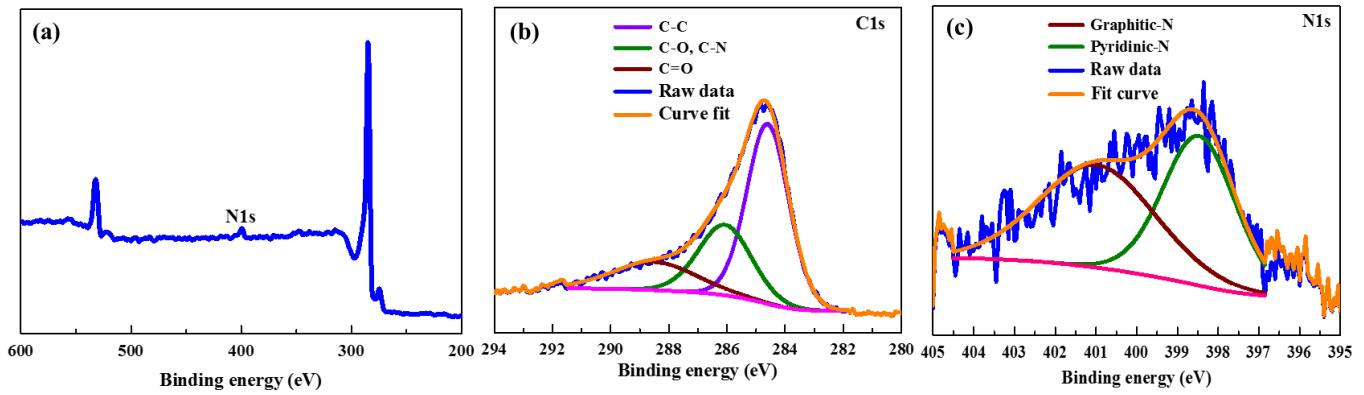
**Fig. S2** (a) SEM image of CDC, (b) – (c) TEM images of CDC.



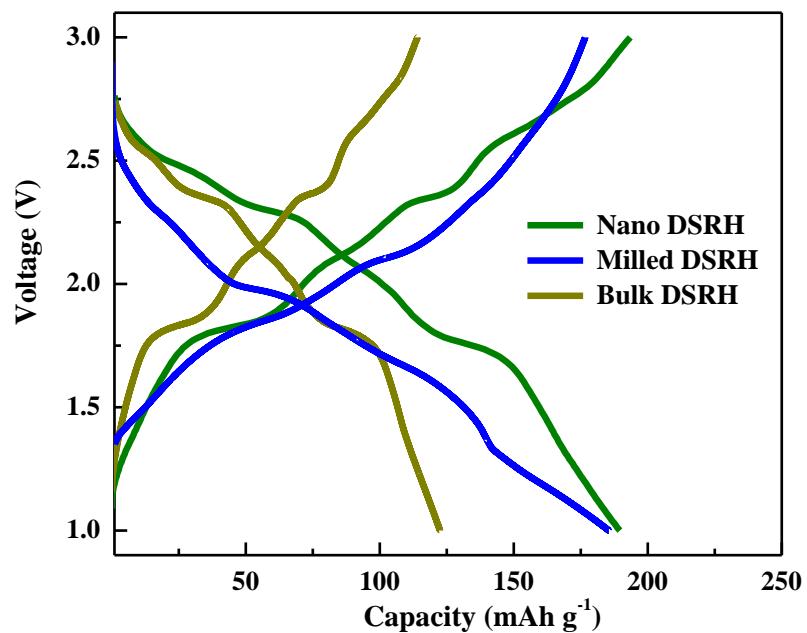
**Fig. S3** N<sub>2</sub> adsorption/desorption isotherm of CDC, Inset: BJH pore size distribution



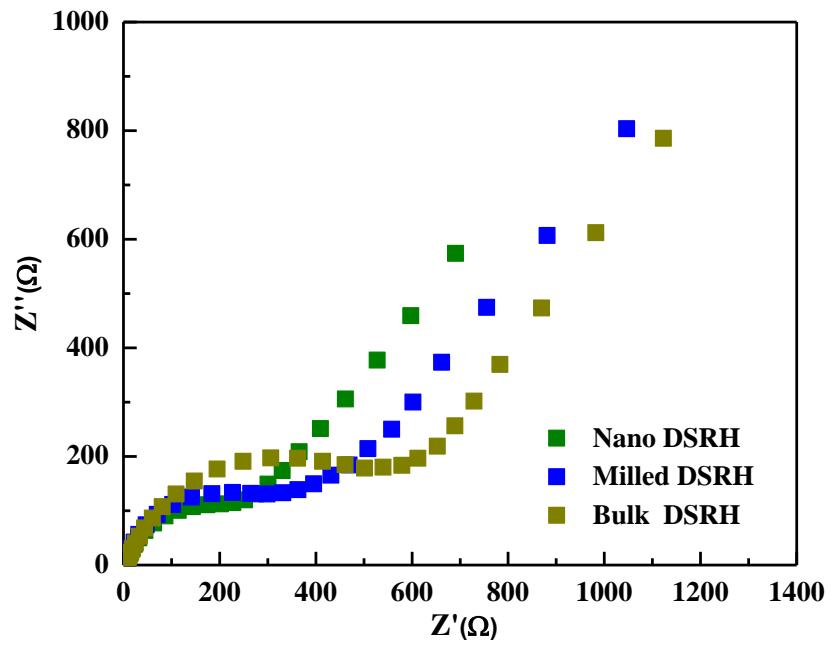
**Fig. S4** (a) XRD pattern of CDC, and (b) Raman spectrum of CDC



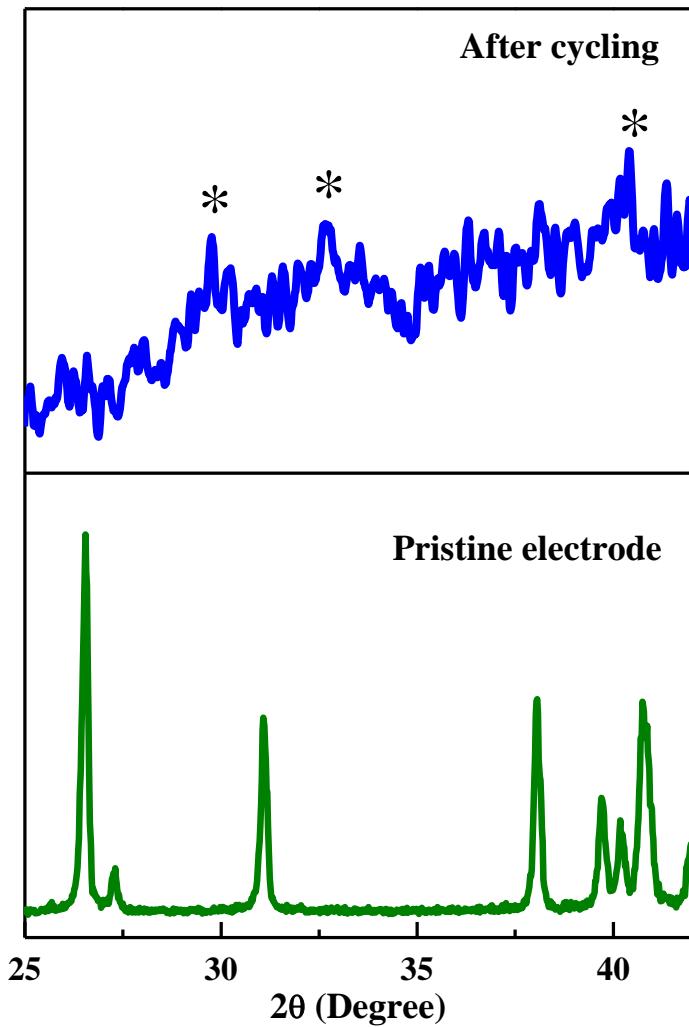
**Fig. S5** (a) XPS survey spectrum of CDC, (b) deconvoluted C1s spectrum, and (c) deconvoluted N1s spectrum



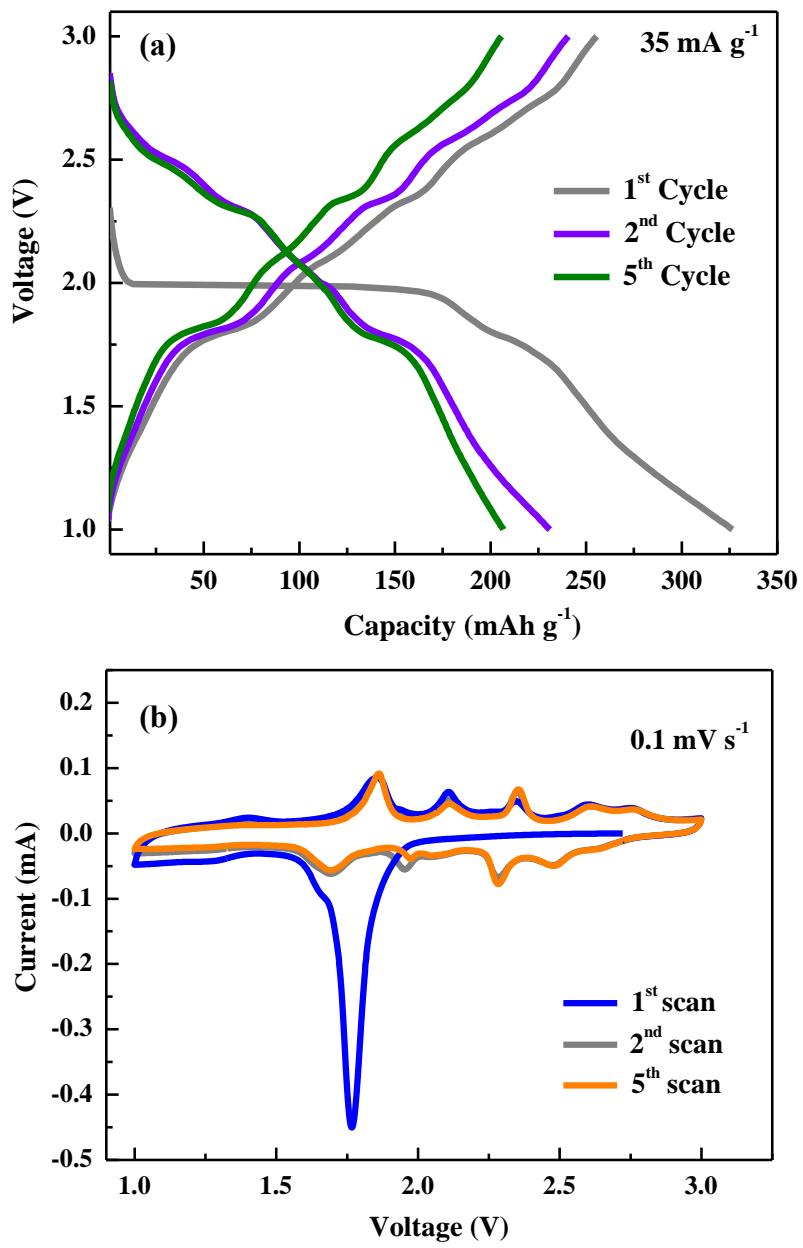
**Fig. S6** CD curves of various DSRH at 0.1 A g<sup>-1</sup>.



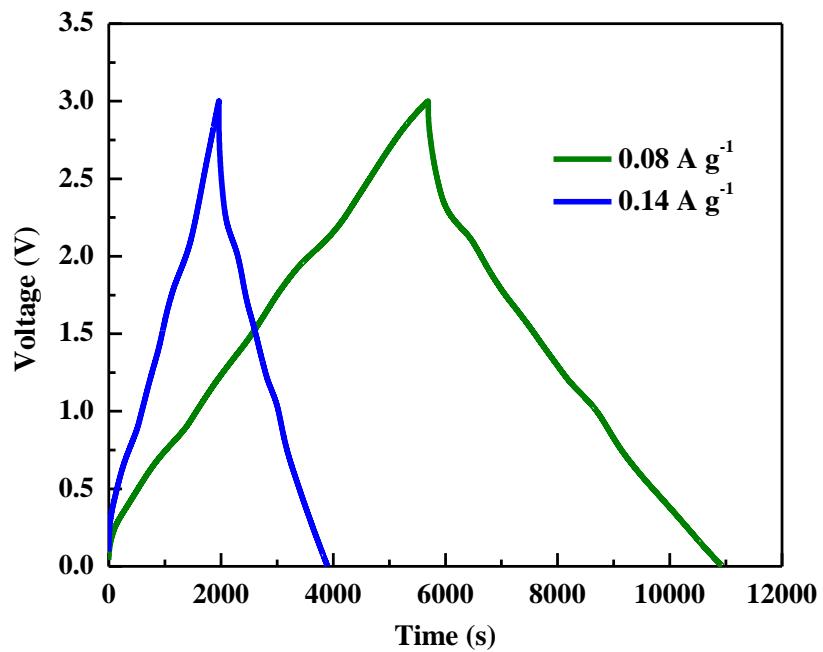
**Fig. S7** Nyquist plots of various nanostructured DSRH.



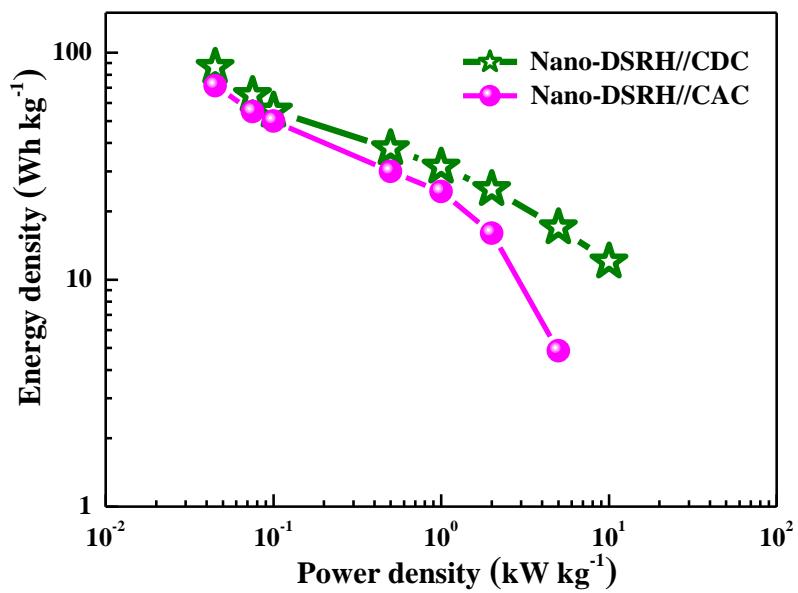
**Fig. S8** XRD pattern of nano-DSRH before and after cycling.



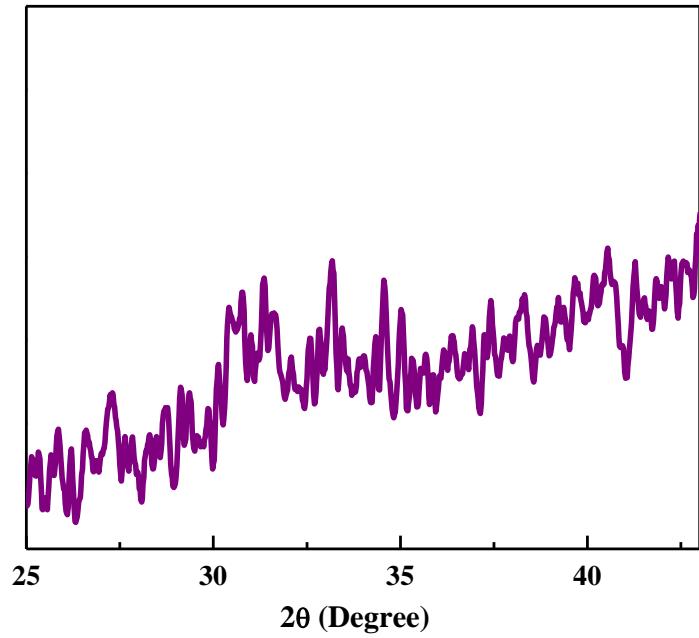
**Fig. S9** (a) CD curves of nano-DSRH Vs. Na, and (b) CV curves of nano-DSRH Vs. Na



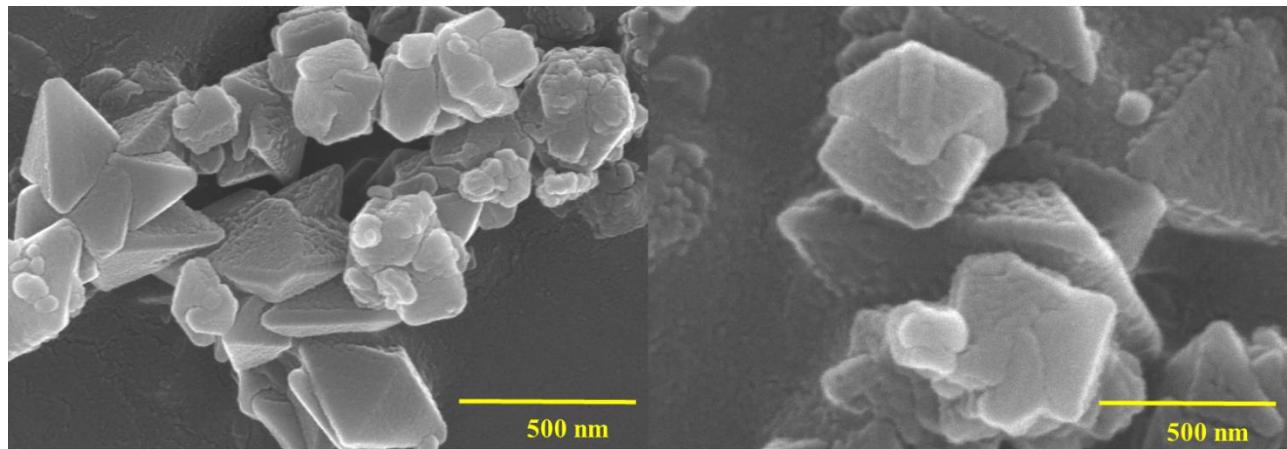
**Fig. S10** CD curves of OHNC



**Fig. S11** Ragone plot comparing the performance OHNC with CDC and CAC cathode



**Fig. S12** XRD patterns of DSRH in ONHC after cycling.



**Fig. S13** SEM images of DSRH in ONHC after cycling.

<i>System</i>	<i>Energy Density</i> (Wh kg <sup>-1</sup> )	<i>Power Density</i> (W kg <sup>-1</sup> )	<i>Stability</i>
<b>Organic sodium capacitor (This work)</b>	<b>87</b>	<b>10000</b>	<b>85% after 10000 cycles</b>
V <sub>2</sub> O <sub>5</sub> -CNT/AC <sub>[S1]</sub>	38	140	80% after 1000 cycles
rGO-TiO <sub>2</sub> /AC <sub>[S2]</sub>	8	9000	80% after 10000 cycles
rGO-Nb <sub>2</sub> O <sub>5</sub> /AC <sub>[S3]</sub>	76	8000	78% - 3000 cycles
Na-TNT/AC <sub>[S4]</sub>	34	889	98% after 1000 cycles
TiO <sub>2</sub> -B/AC <sub>[S5]</sub>	18	235	73% after 1200 cycles
Mesoporous TiO <sub>2</sub> //AC <sub>[S6]</sub>	64.2	1400	90% after 10000 cycles
NaMn <sub>1/3</sub> Ni <sub>1/3</sub> Co <sub>1/3</sub> PO <sub>4</sub> /AC <sub>[S7]</sub>	15	400	95% after 1000 cycles
NVP/NVP <sub>[S8]</sub>	26	290	65 % after 10000 cycles
Li <sub>4</sub> Ti <sub>5</sub> O <sub>12</sub> /AC <sub>[S9]</sub>	10	1000	84% after 9000 cycles
Li <sub>4</sub> Ti <sub>5</sub> O <sub>12</sub> /Carbon clouds <sub>[S10]</sub>	10	9000	82% after 10000 cycles
LiCoO <sub>2</sub> //AC <sub>[S11]</sub>	32	100	40% after 3000 cycles
Li <sub>2</sub> MnSiO <sub>4</sub> /AC <sub>[S12]</sub>	37	1400	85% after 1000 cycles
LiMn <sub>0.5</sub> Ni <sub>0.5</sub> O <sub>4</sub> /AC <sub>[S13]</sub>	56	130	81% after 3000 cycles
LiMn <sub>2</sub> O <sub>4</sub> /AC <sub>[S11]</sub>	38	100	98% after 3000 cycles
LiCoPO <sub>4</sub> /AC <sub>[S14]</sub>	11	1607	67% after 1000 cycles
LiCo <sub>1/3</sub> Ni <sub>1/3</sub> Mn <sub>1/3</sub> O <sub>2</sub> /AC <sub>[S11]</sub>	42	100	90% after 500 Cycles
LiTi <sub>2</sub> (PO <sub>4</sub> ) <sub>3</sub> /AC <sub>[S15]</sub>	14	180	48% after 1000 cycles
Li <sub>3</sub> V <sub>2</sub> (PO <sub>4</sub> ) <sub>3</sub> /AC <sub>[S16]</sub>	27	255	66% after 1000 cycles

**Table S1** Comparison of performance of ONHC with various other systems.

## References

- S1. Z. Chen, V. Augustyn, X. Jia, Q. Xiao, B. Dunn and Y. Lu, *ACS Nano*, 2012, **6**, 4319-4327.
- S2. H. Kim, M.-Y. Cho, M.-H. Kim, K.-Y. Park, H. Gwon, Y. Lee, K. C. Roh and K. Kang, *Adv. Energy Mater.*, 2013, **3**, 1500-1506.
- S3. E. Lim, C. Jo, M. S. Kim, M.-H. Kim, J. Chun, H. Kim, J. Park, K. C. Roh, K. Kang, S. Yoon and J. Lee, *Adv. Funct. Mater.*, 2016, **26**, 3553.
- S4. J. Yin, L. Qi and H. Wang, *ACS Appl. Mater. Interfaces*, 2012, **4**, 2762-2768.
- S5. V. Aravindan, N. Shubha, W. C. Ling and S. Madhavi, *J. Mater. Chem. A*, 2013, **1**, 6145-6151.
- S6. Z. Le, F. Liu, P. Nie, X. Li, X. Liu, Z. Bian, G. Chen, H. B. Wu and Y. Lu, *ACS Nano*, 2017, **11**, 2952-2960.
- S7. R. Satish, V. Aravindan, W. C. Ling and S. Madhavi, *J. Power Sources*, 2015, **281**, 310-317.
- S8. Z. Jian, V. Raju, Z. Li, Z. Xing, Y. S. Hu and X. Ji, *Adv. Funct. Mater.*, 2015, **25**, 5778-5785.
- S9. V. Aravindan, W. Chuiling, M. V. Reddy, G. V. S. Rao, B. V. R. Chowdari and S. Madhavi, *Phys. Chem. Chem. Phys.*, 2012, **14**, 5808-5814.
- S10. A. Banerjee, K. K. Upadhyay, D. Puthusseri, V. Aravindan, S. Madhavi and S. Ogale, *Nanoscale*, 2014, **6**, 4387-4394.
- S11. Y. Wang, J. Luo, C. Wang, Y. Xia, *J. Electrochem. Soc.* **2006**, 153, A1425.
- S12. K. Karthikeyan, V. Aravindan, S. B. Lee, I. C. Jang, H. H. Lim, G. J. Park, M. Yoshio and Y. S. Lee, *J. Power Sources*, 2010, **195**, 3761-3764.

- S13. N. Arun, A. Jain, V. Aravindan, S. Jayaraman, W. Chui Ling, M. P. Srinivasan and S. Madhavi, *Nano Energy*, 2015, **12**, 69-75.
- S14. K. Karthikeyan, S. Amresh, K. J. Kim, S. H. Kim, K. Y. Chung, B. W. Cho and Y. S. Lee, *Nanoscale*, 2013, **5**, 5958-5964.
- S15. V. Aravindan, W. Chuiling, M. V. Reddy, G. V. S. Rao, B. V. R. Chowdari and S. Madhavi, *Phys. Chem. Chem. Phys.*, 2012, **14**, 5808-5814.
- S16. R. Satish, V. Aravindan, W. C. Ling and S. Madhavi, *J. Power Sources*, 2015, **281**, 310-317.