

Supporting Information

Vanillin-derived amines for bio-based thermosets

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I. Characterization data

a. Dihydroxyaminopropane of bisphenol A (**4**)

Described in the literature:

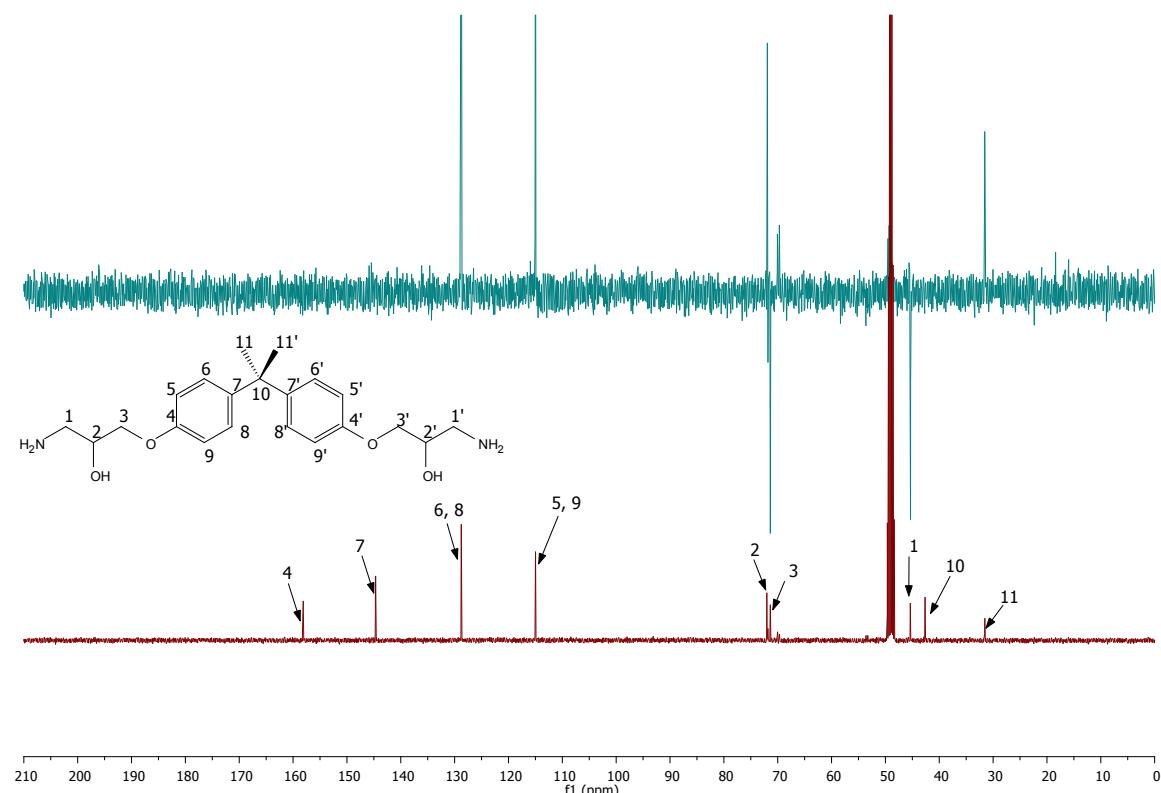
IR (neat, ν , cm⁻¹): 3371–3042, 2969–2925, 2866, 1604, 1582, 1508, 1240, 1029, 830.

¹H NMR (400 MHz, CD₃OD, ppm) δ : 7.13 and 6.85 (d, 8H, H_{ar}); 4.07–3.92 (m, 6H, 2xO-CH₂, 2xCH); 2.89–2.72 (m, 4H, 2xN-CH₂), 1.62 (s, 6H, 2xCH₃).

ESMS (in MeOH, positive ions): m/z $\frac{1}{2}$ 375.2 ([M + H⁺], calculated: 375.23). MS/MS: m/z $\frac{1}{2}$ 375.2, 357.2, 302.1, 208.0, 135.0.

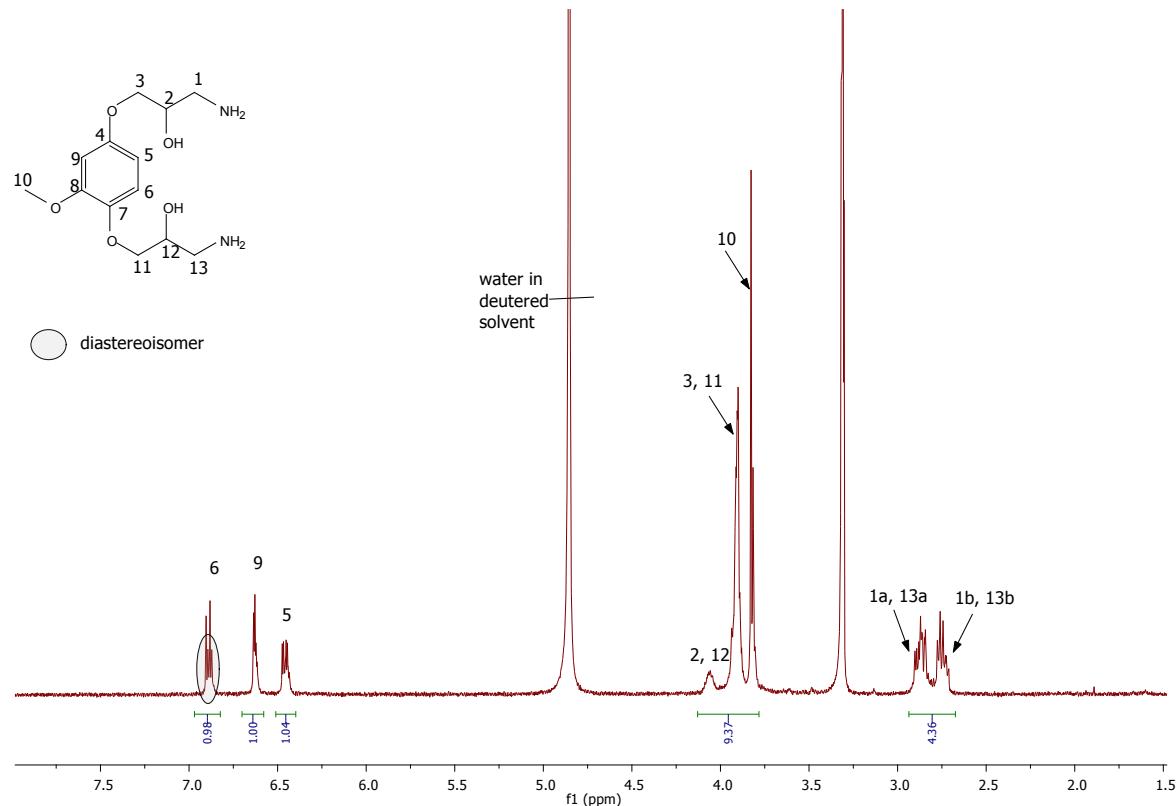
Recorded during this study:

¹³C NMR (101 MHz, MeOD) and DEPT 135

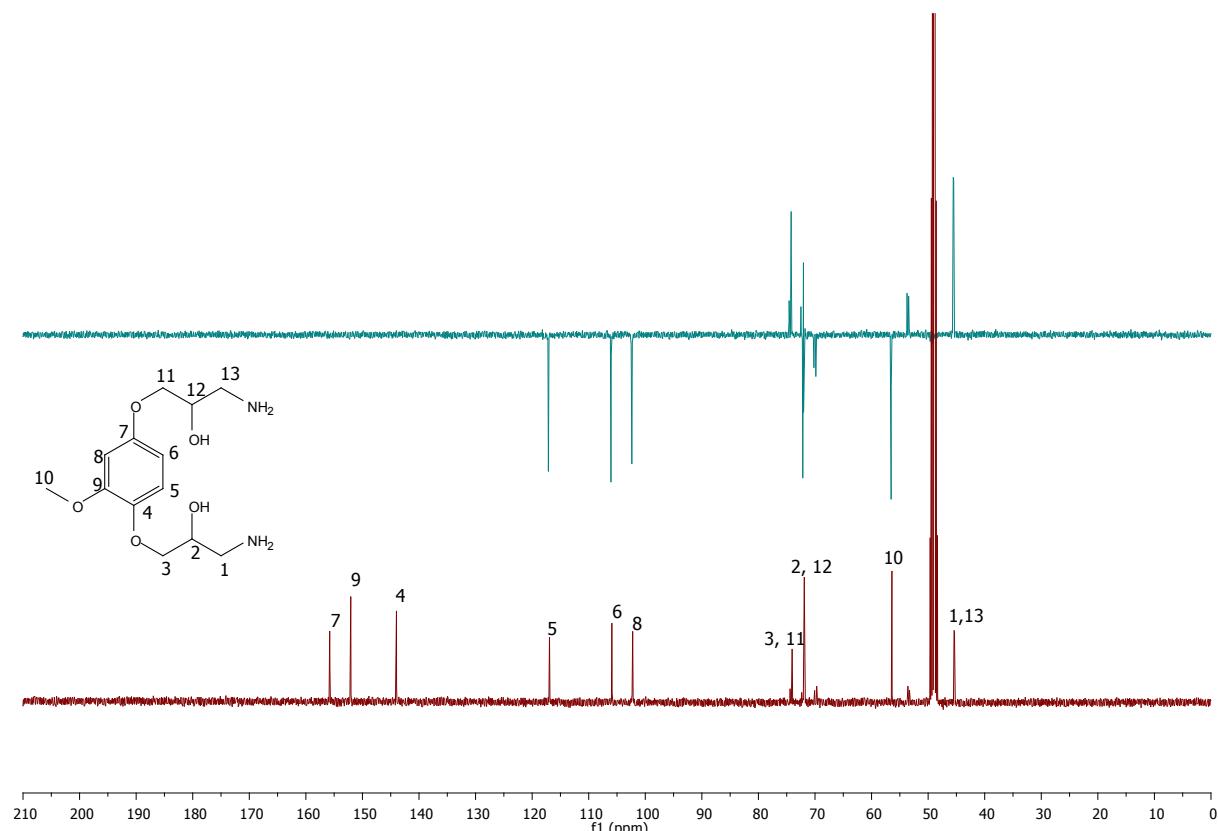


b. Dihydroxyaminopropane of methoxyhydroquinone (**5**)

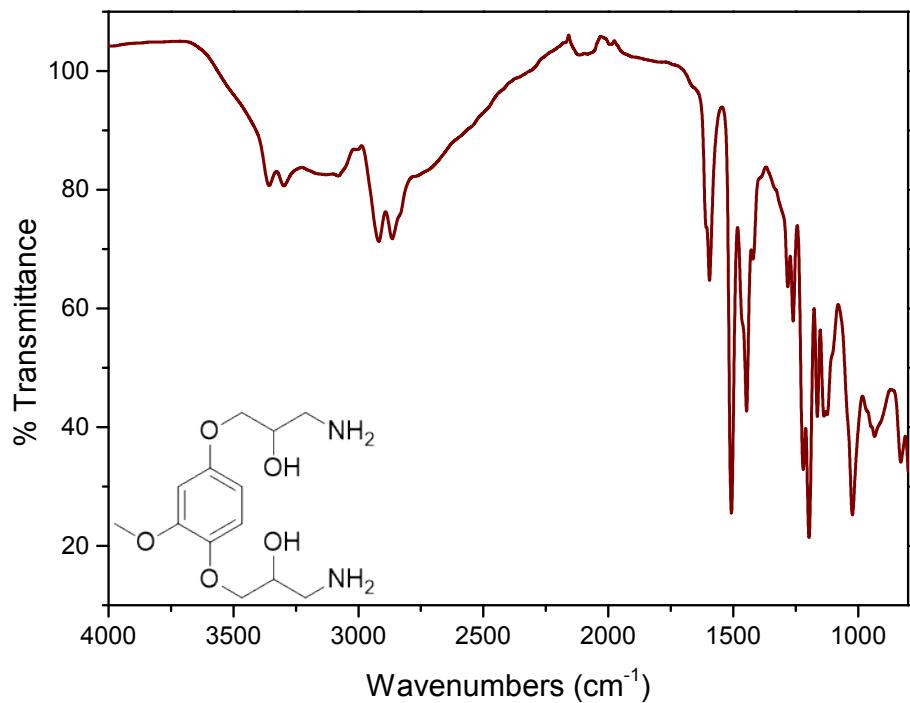
¹H NMR (400 MHz, MeOD)



¹³C NMR (101 MHz, MeOD) and DEPT 135

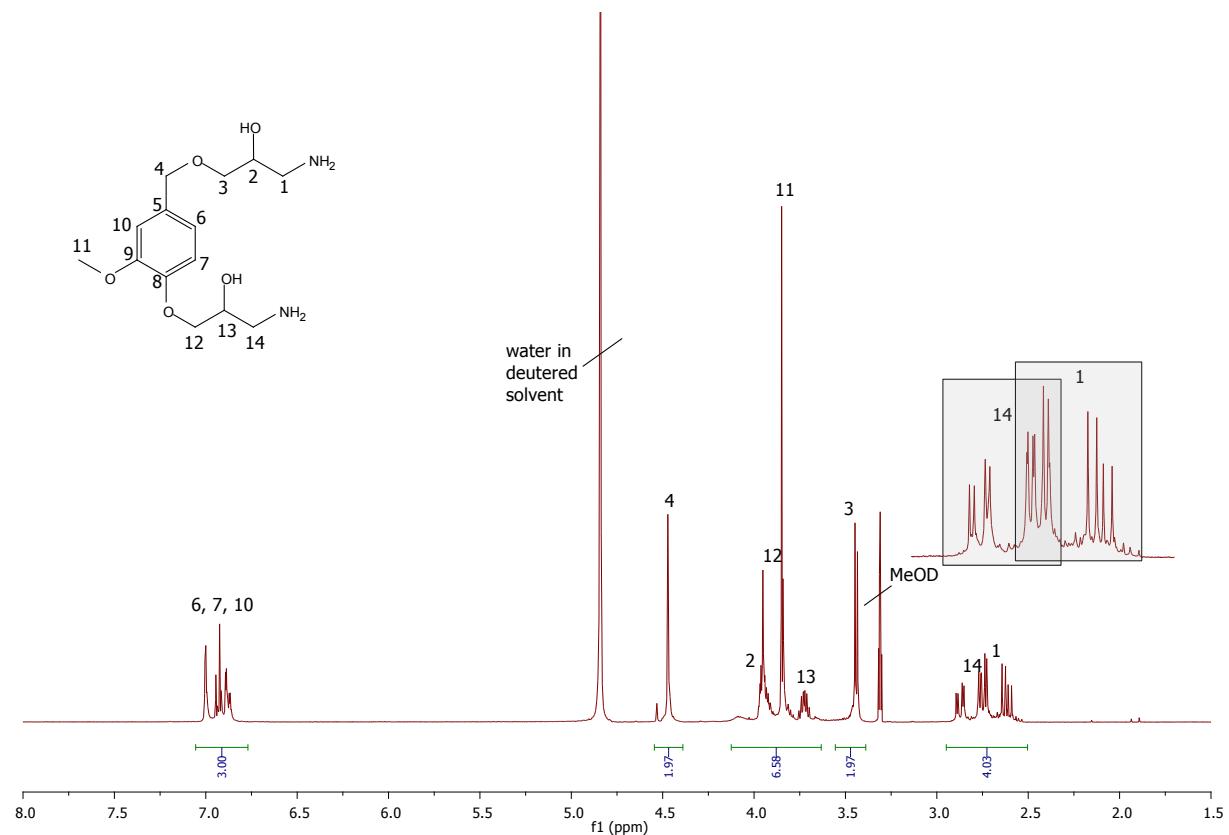


FTIR

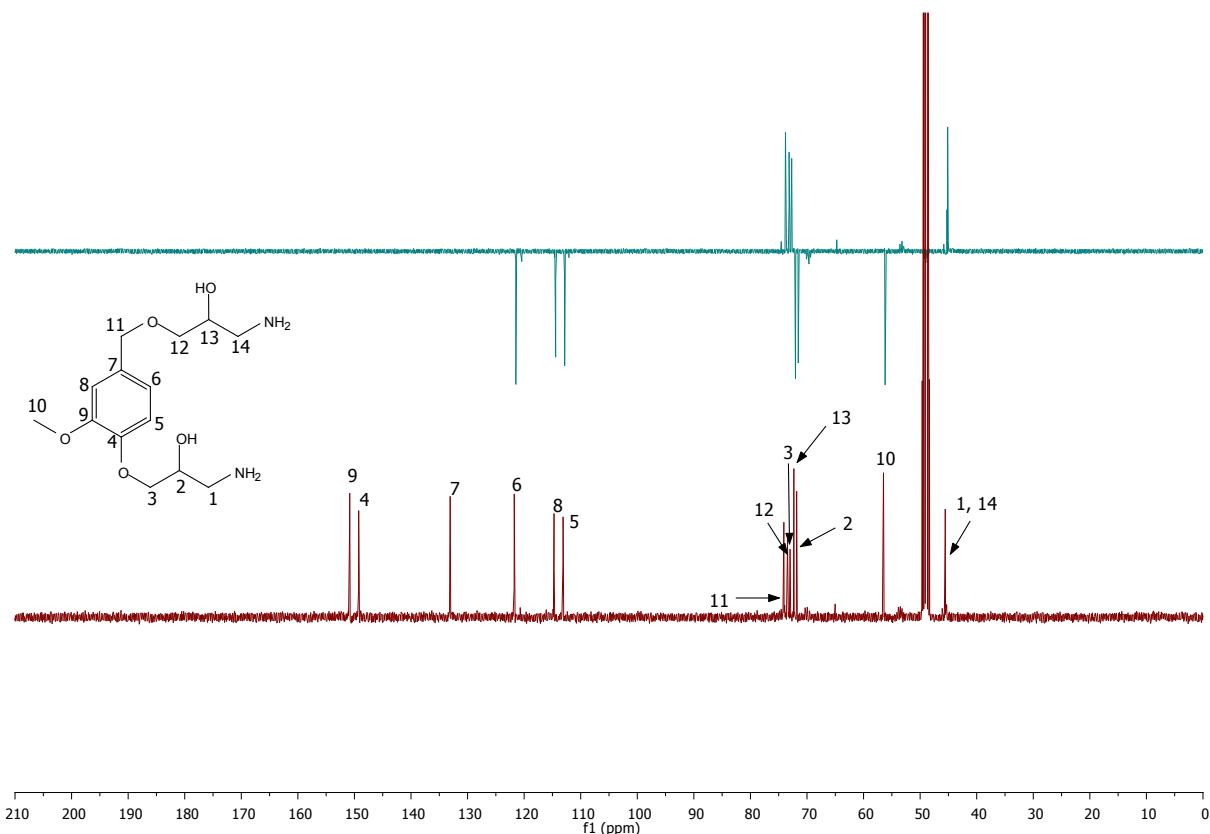


c. Dihydroxyaminopropane of vanillyl alcohol (**6**)

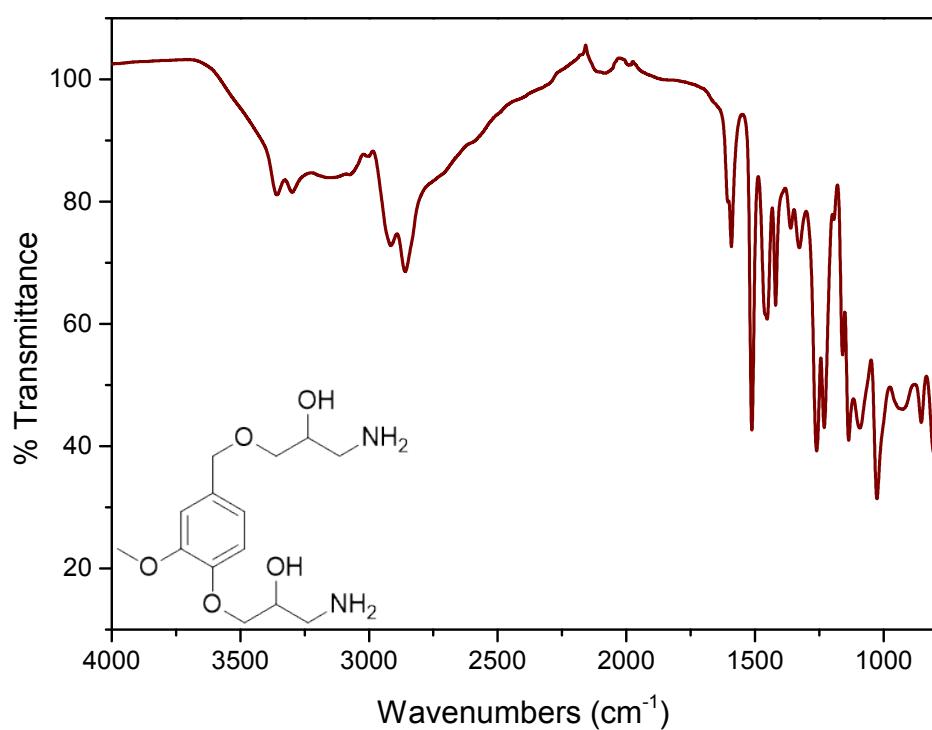
^1H NMR (400 MHz, MeOD)



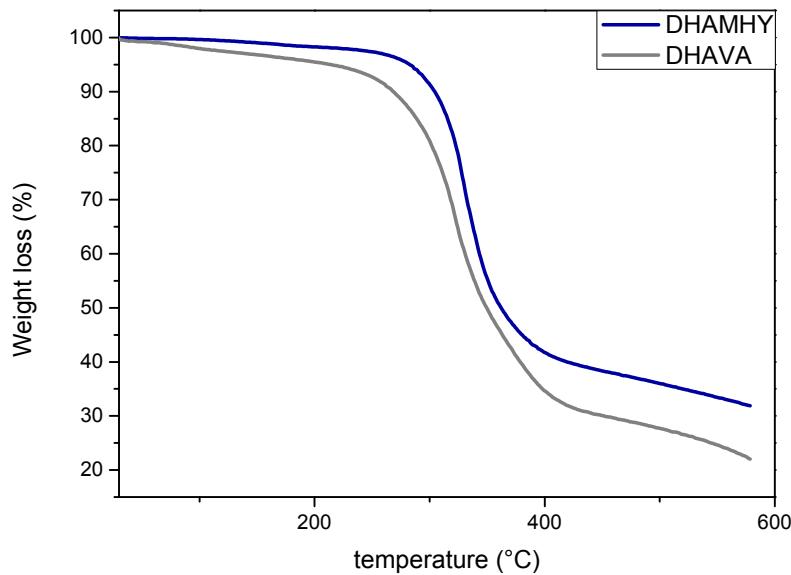
^{13}C NMR (101 MHz, MeOD) and DEPT 135



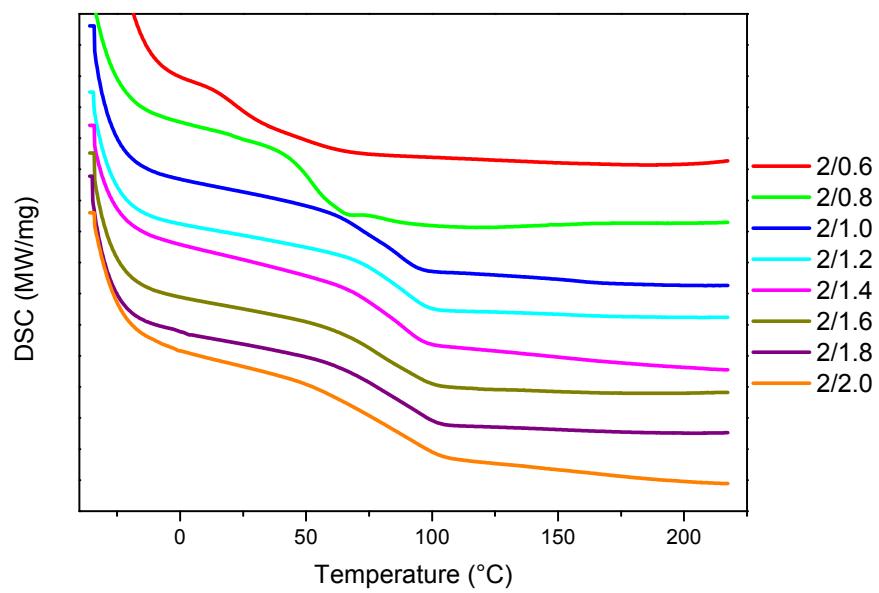
FTIR



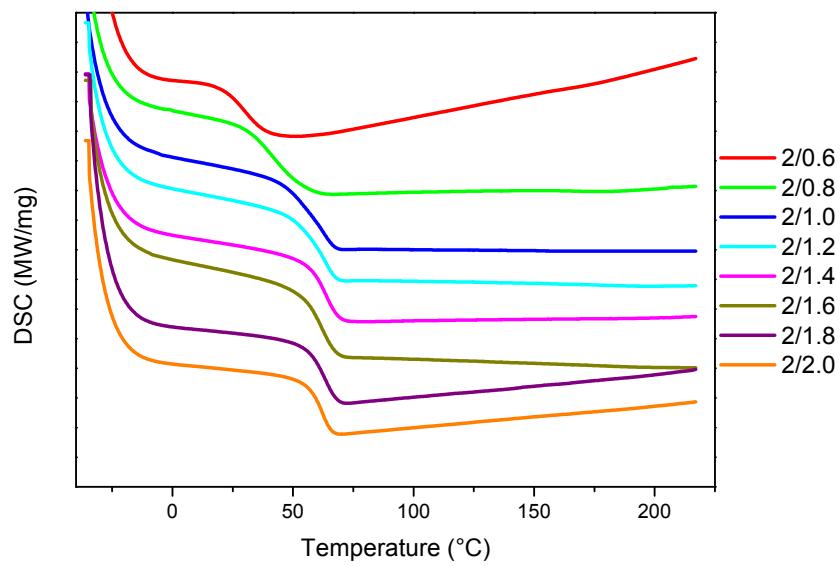
d. T_g measurements of DHAMHY and DAVA



e. Various T_g measurements of the DAVA/DGEBA materials



f. Various T_g measurements of the DAVA/DGEBA materials



g. T_g of the DHAVA materials as a function of the epoxy/amine ratio used

