

Supporting Information

Copper catalysis: Ligand-controlled selective *N*-methylation or *N*-formylation of amines with CO₂ and phenylsilane

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1. General Experimental Section

The starting materials were commercially available and were used without further purification except solvents. The products were isolated by column chromatography on silica gel (200-300 mesh) using petroleum ether (60-90 °C) and ethyl acetate. All compounds were characterized by ^1H NMR, ^{13}C NMR and mass spectroscopy, and the results were consistent with those reported in the literature. NMR spectra were determined on Bruker 400 in CDCl_3 or C_6D_6 . ^1H NMR spectra was recorded on 400 MHz spectrometers using CDCl_3 as solvent referenced to tetramethylsilane (TMS, 0 ppm). The ^{13}C NMR chemical shifts were reported in ppm relative to the carbon resonance of CDCl_3 (central peak is 77.0 ppm). ^1H NMR peaks were labeled as singlet (s), doublet (d), triplet (t), and multiplet (m). The coupling constants, J , were reported in Hertz (Hz). GC-MS data were performed on Finnigan HP G1800 A. GC analyses were performed on a Shimadzu GC-2014 equipped with a capillary column (RTX-17 30 m $\times$ 0.25 μm) using a flame ionization detector.

Unless otherwise noted, carbon dioxide (99.999%), commercially available Cu salts, ligands all secondary amines were obtained from Aladdin, Alfa Aesar and Heowns and used as received. Hydrosilanes were obtained from Energy. All operations were carried out by using standard high vacuum and Schlenk technique unless otherwise noted.

2. General experimental Procedure

General procedure for reductive functionalization of CO₂ with amines to formylamines

Under inert atmosphere (Ar/N₂), a 10 mL Schlenk flask was charged successively with Cu₂(OH)₂CO₃ (3.3 mg, 3 mol%), Ph₂CyP (15.6 mg, 12 mol%), *N*-methylaniline (54 μL, 0.5 mmol), CH₃CN (2 mL) and phenylsilane (0.2 mL, 1.5 mmol). The reaction mixture was stirred at 60 °C for the desired time under an atmosphere of CO₂ (99.999%). After the reaction, 1,3,5-trimethoxybenzene (40.0 mg) was added as an internal standard and then a sample was taken for GC or ¹H NMR analysis to determine the conversion and yield. To identify the structure of the formylated product, the reaction mixture was concentrated and purified by silica gel column chromatography (petroleum ether-EtOAc) to afford the corresponding formamide.

General procedure for reductive functionalization of CO₂ with amines to methylamines

Under inert atmosphere (Ar/N₂), a 10 mL Schlenk flask was charged successively with Cu₂(OH)₂CO₃ (3.3 mg, 3 mol%), DPPB (12.8 mg, 6 mol%), *N*-methylaniline (54 μL, 0.5 mmol), CH₃CN (2 mL) and phenylsilane (0.2 mL, 1.5 mmol). The reaction mixture was stirred at 60 °C for the desired time under an atmosphere of CO₂ (99.999%). After the reaction, 1,3,5-trimethoxybenzene (40.0 mg) was added as an internal standard and then a sample was taken for GC or ¹H NMR analysis to determine the conversion and yield. To identify the structure of the methylated product, the reaction mixture was concentrated and purified by silica gel column chromatography (petroleum ether-EtOAc) to afford the corresponding methylamine.

General procedure for preparing [Cu(DPPB)₂]₂CO₃ and [Cu(Ph₂CyP)₂]₂CO₃

[Cu(DPPB)₂]₂CO₃: 1.71 g (4 mmol) DPPB was dissolved in 150 mL H₂O–iso-propanol (5:7) and then 0.22 g (1 mmol) Cu₂(OH)₂CO₃ was added. The mixture was heated and stirred at the temperature of 70 °C for 24 hrs. After that, the resulting turquoise green precipitate was collected on a filter, washed with Et₂O and dried in vacuo. ESI-MS (*m/z*): [CuC₅₆H₅₆P₄]₂²⁺ 915.26 (calculated), 915.34 (found).

[Cu(Ph₂CyP)₂]₂CO₃: 1.07 g (4 mmol) Ph₂CyP was dissolved in 150 mL H₂O–iso-propanol (5:7) and then 0.22 g (1 mmol) Cu₂(OH)₂CO₃ was added. The mixture was

heated and stirred at the temperature of 70 °C for 24 hrs. Then the resulting turquoise green precipitate was collected on a filter, washed with Et₂O and dried in vacuo. ESI-MS (m/z): [CuC₃₆H₄₂P₂]₂²⁺ 600.21 (calculated), 600.15 (found).

3. Screening Results

Table S1. Optimization of the solvent^a

$ \begin{array}{c} \text{Ph-NH-Me} + \text{CO}_2 \xrightarrow[\text{balloon}]{\begin{array}{l} \text{Cu}_2(\text{OH})_2\text{CO}_3 \text{ (2.5 mol\%)} \\ \text{DPPB (5 mol\%)} \\ \text{PhSiH}_3 \text{ (3 equiv.)} \\ \text{solvent, 50 }^\circ\text{C, 12 h} \end{array}} \begin{array}{c} \text{Ph-N-CHO-Me} \\ \text{2a} \end{array} + \begin{array}{c} \text{Ph-N-CH}_3\text{-Me} \\ \text{3a} \end{array} \\ \text{1a} \end{array} $				
Entry	Solvent/mL	Conv ^b /%	Yield ^b /%	
			2a	3a
1	CH ₃ CN (2)	90	11	77
2	PhMe (2)	87	27	14
3	<i>n</i> Bu ₂ O (2)	90	26	27
4	CH ₃ CN (1)	89	11	35
5	CH ₃ CN (3)	99	61	29

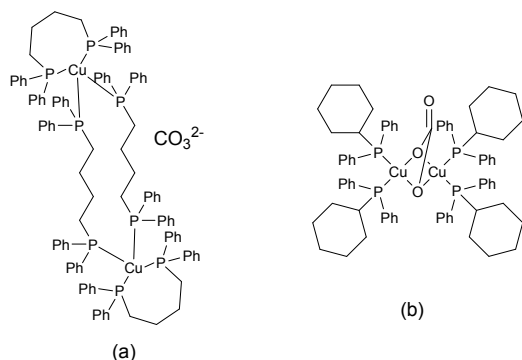
^a Reaction condition: **1a** (54 μ L, 0.5 mmol), Cu₂(OH)₂CO₃ (2.7 mg, 2.5 mol%), DPPB (10.5 mg, 5 mol%), PhSiH₃ (0.2 mL, 3 equiv.), CO₂ balloon, 50 $^\circ$ C for 12 h. ^b Determined by GC using 1,3,5-trimethoxybenzene as an internal standard.

Table S2. Optimization of the hydrosilane^a

 1a + CO₂ balloon $\text{Cu}_2(\text{OH})_2\text{CO}_3$ (3 mol%) DPPB (6 mol%) hydrosilane $\xrightarrow{\text{CH}_3\text{CN (2 mL), 60 }^\circ\text{C, 12 h}}$  2a +  3a				
Entry	Silane /eq	Conv ^b /%	Yield ^b /%	
			2a	3a
1	PhSiH ₃ (3)	99	6	93
2	Ph ₂ SiH ₂ (4.5)	90	18	72
3	PMHS(9)	<1	trace	trace
4	EtO ₃ SiH(9)	<1	trace	trace
5	PhSiH ₃ (2)	86	14	54

^a Reaction condition: **1a** (54 μL , 0.5 mmol), $\text{Cu}_2(\text{OH})_2\text{CO}_3$ (3.3 mg, 3 mol%), DPPB (12.8 mg, 6 mol%), CH_3CN (2 mL), CO_2 balloon, 60 $^\circ\text{C}$ for 12 h. ^b Determined by GC using 1,3,5-trimethoxybenzene as an internal standard.

4. Proposed Structure Active Catalyst and the



Scheme S1. Proposed structure for (a) $[\text{Cu}(\text{DPPB})_2]_2\text{CO}_3$ and (b) $[\text{Cu}(\text{Ph}_2\text{CyP})_2]_2\text{CO}_3$.

5. Verification of the interaction between allyl-group and copper

The interaction between Cu and C=C is also identified by DFT calculation. As shown in this figure, for *N*-allyl formanilide, both of the C=O in formyl group and C=C can coordinate to Cu, but methyl group in *N*-ethylformanilide cannot coordinate to Cu.

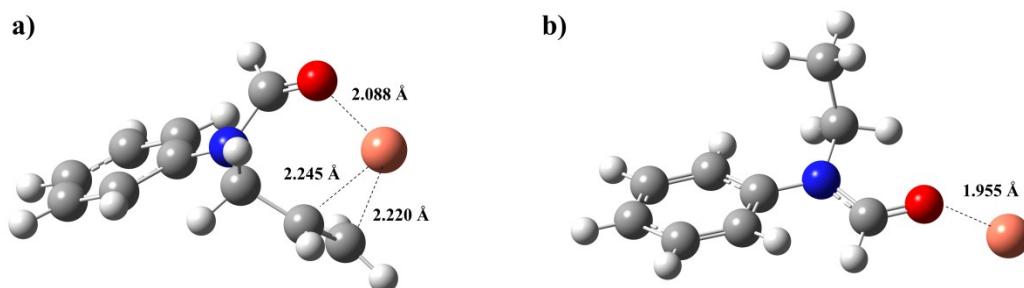
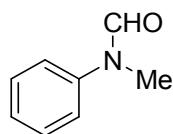


Figure S1. Interaction between copper and different substrates (Copper: orange; Oxygen: red; Nitrogen: blue; Carbon: gray; a: interaction between Cu and *N*-allyl formanilide; b: interaction between Cu and *N*-ethylformanilide)

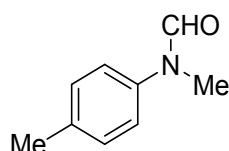
6. Characterization Data for the Products

N-Methylformanilide (2a)¹



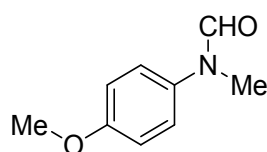
Yellow oil. ¹H NMR (400 MHz, CDCl₃) δ 8.48 (s, 1H), 7.42 (t, *J* = 7.8 Hz, 2H), 7.28 (t, *J* = 7.4 Hz, 1H), 7.18 (d, *J* = 7.6 Hz, 2H), 3.33 (s, 3H) ppm. ¹³C NMR (101 MHz, CDCl₃) δ 162.21, 142.03, 129.49, 126.26, 122.21, 31.90 ppm. GC-MS (EI, 70 eV) *m/z* (%) 135.15 (70.68), 106.10 (100.00), 94.10 (30.45), 79.10 (20.54), 77.10 (42.48), 66.05 (21.05).

N,4'-Dimethylformanilide (2b)²



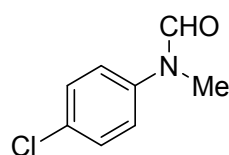
Yellow solid. ¹H NMR (400 MHz, CDCl₃) δ 8.42 (s, 1H), 7.21 (d, *J* = 8.0 Hz, 2H), 7.06 (d, *J* = 8.3 Hz, 2H), 3.30 (s, 3H), 2.37 (s, 3H) ppm. ¹³C NMR (101 MHz, CDCl₃) δ 162.34, 139.62, 136.33, 130.10, 122.50, 32.18, 20.82 ppm. GC-MS (EI, 70 eV) *m/z* (%) 149.20 (76.70), 120.15 (100.00), 108.10 (42.15), 91.10 (33.92), 65.10 (21.58).

N-Methyl-4-methoxyformanilide (2c)²



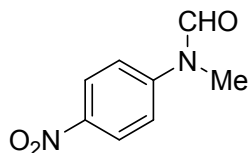
Brown oil. ¹H NMR (400 MHz, CDCl₃) δ 8.34 (s, 1H), 7.10 (d, *J* = 8.9 Hz, 2H), 6.93 (d, *J* = 8.9 Hz, 2H), 3.82 (s, 3H), 3.27 (s, 3H) ppm. ¹³C NMR (101 MHz, CDCl₃) δ 162.28, 158.11, 135.06, 124.47, 114.58, 55.38, 32.50 ppm. GC-MS (EI, 70 eV) *m/z* (%) 165.15 (100.00), 124.15 (62.20), 122.15 (95.98), 94.10 (33.80), 65.10 (21.69).

N-(4-Chlorophenyl)-*N*-methylformamide (2d)²



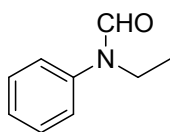
Yellow solid. ^1H NMR (400 MHz, CDCl_3) δ 8.45 (s, 1H), 7.39 (d, $J = 8.7$ Hz, 2H), 7.12 (d, $J = 8.7$ Hz, 2H), 3.30 (s, 3H) ppm. ^{13}C NMR (101 MHz, CDCl_3) δ 161.94, 140.68, 131.97, 129.71, 123.46, 32.01 ppm. GC-MS (EI, 70 eV) m/z (%) 169.10 (54.97), 142.10 (31.49), 140.10 (100.00), 128.10 (44.0), 77.10 (26.63).

***N*-(4-Nitrophenyl)-*N*-methylformamide (2e)**



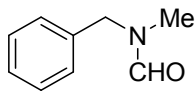
Yellow oil. ^1H NMR (400 MHz, CDCl_3) δ 8.73 (s, 1H), 8.31 (d, $J = 9.1$ Hz, 2H), 7.34~7.32 (m, 2H), 3.40 (s, 3H) ppm. ^{13}C NMR (101 MHz, CDCl_3) δ 161.45, 147.49, 145.03, 125.42, 120.40, 31.40 ppm. GC-MS (EI, 70 eV) m/z (%) 180.10 (55.01), 151.10 (100.00).

***N*-Ethyl-*N*-phenyl-formamide (2f)²**



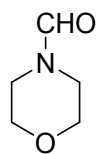
Yellow oil. ^1H NMR (400 MHz, CDCl_3) δ 8.35 (s, 1H), 7.42 (t, $J = 7.7$ Hz, 2H), 7.30 (t, $J = 7.6$ Hz, 1H), 7.17 (d, $J = 7.8$ Hz, 2H), 3.86 (q, $J = 7.2$ Hz, 2H), 1.16 (t, $J = 7.2$ Hz, 3H) ppm. ^{13}C NMR (101 MHz, CDCl_3) δ 162.13, 140.89, 129.71, 126.94, 124.34, 40.17, 13.14 ppm. GC-MS (EI, 70 eV) m/z (%) 150.06 (22.86), 149.04 (76.16), 121.06 (97.60), 120.18 (19.42), 107.13 (10.49), 106.18 (100.00).

***N*-Benzyl-*N*-methylformamide (2h)³**



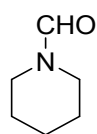
Purple oil. ^1H NMR (400 MHz, CDCl_3) δ 8.20 (s, 0.57H), 8.07 (s, 0.43H), 7.30~7.11 (m, 5H), 4.44 (s, 0.86H), 4.30 (s, 1.14H), 2.76 (s, 1.26H), 2.69 (s, 1.74H) ppm. ^{13}C NMR (101 MHz, CDCl_3) δ 162.78, 162.61, 136.06, 135.80, 128.93, 128.72, 128.27, 128.12, 127.67, 127.43, 53.50, 47.77, 29.74, 29.47 ppm. GC-MS (EI, 70 eV) m/z (%) 149.10 (100.00), 134.15 (17.80), 106.10 (29.35), 91.10 (57.23), 79.10 (28.00).

4-Morpholine carbaldehydede (2i)²



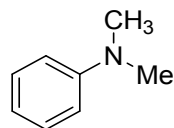
Colourless oil. ¹H NMR (400 MHz, CDCl₃) δ 8.07 (s, 1H), 3.75 – 3.63 (m, 4H), 3.62 – 3.53 (m, 2H), 3.49 – 3.31 (m, 2H) ppm. ¹³C NMR (101 MHz, CDCl₃) δ 160.80, 67.18, 66.39, 45.75, 40.55 ppm. GC-MS (EI, 70 eV) m/z (%) 115.15 (100.00), 100.10 (77.82), 86.10 (51.09), 85.10 (22.02).

Piperidine-1-carbaldehyde (2j)³



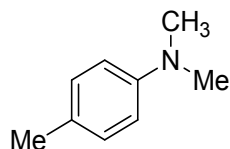
Yellow oil. ¹H NMR (400 MHz, CDCl₃) δ 8.01 (s, 1H), 3.55 – 3.43 (m, 2H), 3.38 – 3.28 (m, 2H), 1.70 (p, J = 5.7 Hz, 2H), 1.63 – 1.50 (m, 4H) ppm. ¹³C NMR (101 MHz, CDCl₃) δ 160.43, 46.36, 40.10, 26.00, 24.53, 24.08 ppm. GC-MS (EI, 70 eV) m/z (%) 113.15 (100.00), 112.15 (30.57), 98.10 (36.95), 84.15 (50.88), 70.10 (20.83).

N,N-Dimethylaniline (3a)¹



Yellow oil. ¹H NMR (400 MHz, CDCl₃) δ 7.24 (dd, J = 8.8, 7.3 Hz, 2H), 6.85 – 6.62 (m, 3H), 2.94 (s, 6H) ppm. ¹³C NMR (101 MHz, CDCl₃) δ 150.58, 129.03, 116.60, 112.62, 40.61 ppm. GC-MS (EI, 70 eV) m/z (%) 121.15 (81.29), 120.15 (100.00), 77.05 (29.56).

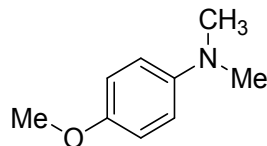
N,N-Dimethyl-p-toluidine (3b)¹



Yellow oil. ¹H NMR (400 MHz, CDCl₃) δ 7.05 (d, J = 8.1 Hz, 2H), 6.69 (d, J = 8.2 Hz, 2H), 2.89 (s, 6H), 2.25 (s, 3H) ppm. ¹³C NMR (101 MHz, CDCl₃) δ 148.77, 129.54,

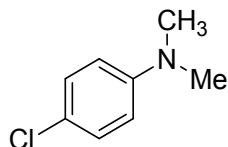
126.09, 113.19, 41.05, 20.22 ppm. GC-MS (EI, 70 eV) m/z (%) 35.20 (77.14), 134.15 (100.00), 91.10 (22.96).

4-Methoxy-*N,N*-dimethylaniline (3c)¹



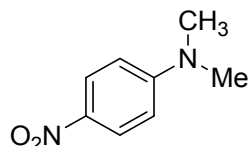
White solid. ¹H NMR (400 MHz, CDCl₃) δ 6.88 – 6.81 (m, 2H), 6.79 – 6.72 (m, 2H), 3.76 (s, 3H), 2.86 (s, 6H) ppm. ¹³C NMR (101 MHz, CDCl₃) δ 152.12, 145.52, 115.04, 114.60, 55.72, 41.92 ppm. GC-MS (EI, 70 eV) m/z (%) 151.25 (59.49), 136.20 (100.00).

4-Chloro-*N,N*-dimethylaniline (3d)¹



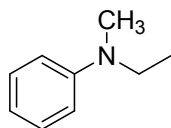
Yellow oil. ¹H NMR (400 MHz, CDCl₃) δ 7.17 (d, J = 8.9 Hz, 2H), 6.64 (d, J = 8.7 Hz, 2H), 2.92 (s, 6H) ppm. ¹³C NMR (101 MHz, CDCl₃) δ 149.10, 128.75, 121.36, 113.59, 40.64 ppm. GC-MS (EI, 70 eV) m/z (%) 157.10 (25.27), 156.10 (39.00), 155.10 (82.99), 154.10 (100.00).

4-Nitro-*N,N*-dimethylaniline (3e)⁴



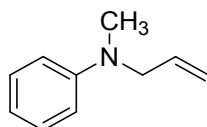
Yellow solid. ¹H NMR (400 MHz, CDCl₃) δ 8.13 (d, J = 8.9 Hz, 2H), 6.63 (d, J = 8.7 Hz, 2H), 2.12 (s, 6H) ppm. ¹³C NMR (101 MHz, CDCl₃) δ 154.13, 135.95, 125.11, 110.29, 40.32 ppm. GC-MS (EI, 70 eV) m/z (%) 166.25 (100), 136.25 (60.88), 119.20 (48.22), 105.20 (32.98), 104.20 (33.69), 79.15 (31.13), 77.10 (68.33).

***N*-Ethyl-*N*-methylaniline (3f)¹**



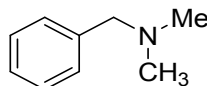
Yellow liquid. ^1H NMR (400 MHz, CDCl_3) δ 7.22 (t, $J = 7.7$ Hz, 2H), 6.84 – 6.52 (m, 3H), 3.39 (q, $J = 7.0$ Hz, 2H), 2.89 (s, 3H), 1.10 (t, $J = 7.1$ Hz, 3H) ppm. ^{13}C NMR (101 MHz, CDCl_3) δ 149.05, 129.11, 115.97, 112.33, 46.75, 37.38, 11.13 ppm. GC-MS (EI, 70 eV) m/z (%) 135.15 (31.98), 120.15 (100.00), 77.05 (33.12).

***N*-Allyl-*N*-methylaniline (3g)⁴**



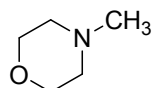
Yellow oil. ^1H NMR (400 MHz, CDCl_3) δ 7.25 – 7.20 (m, 2H), 6.74 – 6.68 (m, 3H), 5.89 – 5.80 (m, 1H), 5.18 – 5.13 (m, 2H), 3.92 (d, $J = 4.9$ Hz, 2H), 2.93 (s, 3H) ppm. ^{13}C NMR (101 MHz, CDCl_3) δ 149.43, 133.75, 129.07, 116.36, 116.12, 112.40, 55.23, 37.97 ppm. GC-MS (EI, 70 eV) m/z (%) 148.08 (70.11), 147.15 (100.00), 146.26 (25.67), 120.28 (34.38).

***N,N*-dimethyl-Benzenemethanamine (3h)¹**



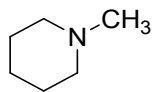
Yellow oil. ^1H NMR (400 MHz, CDCl_3) δ 7.32 – 7.25 (m, 5H), 3.42 (s, 2H), 2.24 (s, 6H) ppm. ^{13}C NMR (101 MHz, CDCl_3) δ 138.82, 129.10, 128.22, 127.03, 54.40, 45.35 ppm. GC-MS (EI, 70 eV) m/z (%) 135.25 (66.41), 134.25 (54.64), 91.15 (100), 65.10 (36.51).

4-Methylmorpholine (3i)⁵



Colourless oil. ^1H NMR (400 MHz, CDCl_3) δ 3.80 – 3.61 (m, 4H), 2.41 (s, 4H), 2.29 (s, 3H) ppm. ^{13}C NMR (101 MHz, CDCl_3) δ 66.67, 55.20, 46.23 ppm. GC-MS (EI, 70 eV) m/z (%) 101.15 (100.00), 100.15 (36.36), 71.10 (60.10).

1-Methylpiperidine (3j)⁶



Colourless oil. ¹H NMR (400 MHz, CDCl₃) δ 2.33 (s, 4H), 2.24 (s, 3H), 1.68 – 1.51 (m, 4H), 1.41 (s, 2H) ppm. ¹³C NMR (101 MHz, CDCl₃) δ 56.46, 46.86, 25.95, 23.70 ppm. GC-MS (EI, 70 eV) m/z (%) 99.15 (42.36), 98.15 (100.00), 71.10 (22.54).

Reference:

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7. NMR Spectral Copies of the Products

