Support information

The confined space electron transfer enhances phosphotungstate intercalated ZnAl-LDHs for photocatalytic oxidation/extraction desulfurization of fuel oil with air

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Catalyst		Wt %					
		Al	Р	W	Ν	Н	
Zn _{0.66} Al _{0.34} (OH) ₂ (NO ₃) _{0.34} • 2.3H ₂ O	28.8	6.1	-	-	3.2	4.44	
$Zn_{0.63}Al_{0.37}(OH)_2(PW_{12}O_{40})_{0.04}(NO_3)_{0.25} \bullet 2.1H_2O$	22.8	5.5	0.63	45.4	1.38	2.45	
$Zn_{0.64}Al_{0.36}(OH)_2(PW_{12}O_{40})_{0.06}(NO_3)_{0.19} \bullet 2.6H_2O$	20.3	4.7	0.75	54.5	0.98	2.27	
$Zn_{0.63}Al_{0.37}(OH)_2(PW_{12}O_{40})_{0.07}(NO_3)_{0.16} \bullet 1.7H_2O$	15.3	3.7	0.87	61.6	0.81	1.65	

Table S1 Elemental composition of different proportions of ZnAl-PW₁₂O₄₀.

Table S2 Asymmetric vibration bands position in FT-IR spectra								
Samples	N-O	P-Oa	$W = O_d$	W-O _b -W	W-O _C -W			
$H_{3}PW_{12}O_{40}$	-	1086	987	895	803			
ZnAl-(PW12O40)0.07-LDHs	1384	1064	960	895	803			
ZnAl-(PW12O40)0.06-LDHs	1384	1064	961	895	804			
ZnAl-NO ₃ -LDHs	1384	-	-	-	-			

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Fig. S1 Schematic illustration of the PODS reactor (1-cooling water, 2- air inlet, 3- stirrer, 4-reaction liquid, 5-mercury lamp, 6- condenser, 7-air outlet)

 Table S3 Electron binding energy of related elements.

Samples	Zn2p _{1/2}	Zn2p _{3/2}	Al2p _{3/2}	W4f _{7/2}	W4f _{5/2}
ZnAl-NO ₃ -LDHs	1045.0	1021.9	74.3	-	-
ZnAl-(PW12O40)0.07-LDHs	1045.6	1022.5	74.7	35.9	37.9
$H_3PW_{12}O_{40}$	-	-	-	36.3	38.4



Fig. S2 Experimental results of catalyst adsorption Condition: 500ppm oil(90 mL), catalyst dosage 1g/L



Fig. S3. GC–MS analysis of the acetonitrile phase after reaction.



Fig. S4. Mott-Schottky plots of LDHs, measured in 1 kHz at room temperature in the dark, Solution: 0.1 M NaSO₄.