Cooperative Iridium Complex Catalyzed Synthesis of Quinoxalines, Benzimidazoles and Quinazolines in Water

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1. General procedure and materials.

All the reactions were carried out under air or argon atmosphere by using standard schlenk line technique. Glasswares were oven dried prior to use. All the commercial reagents and metal precursors were purchased from Sigma-aldrich, Alfa-acear, Spectrochem, Avra, SD-fine chemical and arora matthey India. ¹H, ¹³C, ³¹P NMR spectra were recorded with CDCl₃, D₂O, CD₃OD and DMSO-d₆ in JEOL 500 MHz and 400 MHz Spectrometers. ESI-MS were recorded on a Waters Micromass Quattro Micro triple-quadrupole mass spectrometer. FT-IR were recorded on a Perkin Elmer IR spectrometer. All the GC analysis were performed using Perkein Elmer Clarus 600 and Agilent 7890 B Gas Chromatograph, where as GC-MS were measured using Agilent 7890 A Gas Chromatograph equipped with Agilent 5890 triple-quadrupole mass system.

2. Synthesis of ligands and metal complexes.

Ligands L1-L5 were synthesised according to the literature procedure.^{1, 2} Metal complexes 1a, 1c, 1d and 1e were synthesised following known literature procedure.²

2-(4,5-dihydro-1H-imidazol-2-yl)-6-methoxypyridine (L1)



In a two-neck round bottom flask, a solution of ethylenediamine (133 mg, 2.20 mmol) in *tert*butyl alcohol (20 mL) was added over 6-methoxypyridine-2-carboxaldehyde (250 mg, 1.83 mmol). Then, the mixture was stirred at room temperature under an argon atmosphere for 30 min. After that, K₂CO₃ (760 mg, 5.50 mmol) and solid I₂ (606 mg, 2.38 mmol) were added to the reaction mixture and stirred at 70 °C for 12 h. The mixture was quenched with sat. aq. Na₂SO₃ until the iodine colour almost disappeared and was extracted with CHCl₃. The organic layer was washed with sat. aq. NaHCO₃ solution and dried over anhydrous Na₂SO₄. After filtration, the mixture was evaporated under reduced pressure to provide crude product as a dark brown oil. It was then purified by column chromatography on alumina using EtOAc/MeOH (95:5) as eluent to give the desire product (271 mg, yield: 84%) as a yellow oil. ¹**H** NMR (500 MHz, CDCl₃): δ = 7.72 (dd, *J* = 9.3, 0.9 Hz, 1H), 7.62 (t, *J* = 10.1 Hz, 1H), 6.79 (d, *J* = 10.5 Hz, 1H), 3.93 (s, 3H), 3.81 (brs, 4H). ¹³**C** NMR (100 MHz, CDCl₃): δ = 164.40, 163.38, 145.98, 139.34, 115.34, 112.71, 53.63, 49.71. **GC-MS** (M⁺) = 177.1.

6-(4,5-dihydro-1H-imidazol-2-yl)pyridin-2-ol (L2)



In a two neck round bottom flask, a mixture of 2-(4,5-dihydro-1H-imidazol-2-yl)-6methoxypyridine (250 mg, 1.4 mmol) and 15 mL of aq. HBr (47 %) was stirred at 95 °C for 24 h. After that, the solvent was removed under reduced pressure. To the residue, 3 mL water was added. Then, the solution pH was adjusted to 6-7 with 0.5 M NaOH. The mixture was evaporated in vacuum, then 2 mL ethanol was added to remove the insoluble solid. The solution was concentrated to obtain the crude product. It was then purified by column chromatography on neutral alumina using MeOH/H₂O (7:1) as eluent to give the product (132.5 mg, yield: 58%) as a white solid. ¹**H NMR** (500 MHz, D₂O): $\delta = 7.64$ (t, J = 7.9 Hz, 1H), 7.11 (d, J = 7.0 Hz, 1H), 6.78 (d, J = 8.7 Hz, 1H), 3.96 (brs, 4H). ¹³**C NMR** (125 MHz, D₂O): $\delta = 165.73$, 163.14, 141.14, 135.02, 120.59, 114.21, 44.96. **HRMS** (**ESI**): Calcd. for C₈H₉N₃O [M+H]⁺: 164.0824; found: 164.0829.

2-(4,5-dihydro-1H-imidazol-2-yl)-6-methylpyridine (L3)



In a two-neck round bottom flask, ethylenediamine (150 mg, 2.49 mmol) was added over 6methylpyridine-2-carboxaldehyde (250 mg, 2.08 mmol) in *tert*-butyl alcohol (20 mL). Then, the mixture was stirred at room temperature under an argon atmosphere for 30 min. After that, K_2CO_3 (862 mg, 6.24 mmol) and solid I₂ (686 mg, 2.70 mmol) were added to the reaction mixture and stirred at 70 °C for 12 h. The mixture was quenched with sat. aq. Na₂SO₃ until the iodine colour almost disappeared, and was extracted with CHCl₃. The organic layer was washed with sat. aq. NaHCO₃ solution and dried over anhydrous Na₂SO₄. After filtration, the mixture was evaporated under reduced pressure to provide crude product as a yellow oil. It was then purified by column chromatography on alumina using ethyl acetate and hexane as eluent to give the desire product (242 mg, yield: 73%) as a yellow oil. ¹**H NMR** (500 MHz, CDCl₃): $\delta = 7.92$ (d, J = 7.8 Hz, 1H), 7.63 (t, J = 7.7 Hz, 1H), 7.18 (d, J = 7.6 Hz, 1H), 3.82 (brs, 4H), 2.54 (s, 3H). ¹³C NMR (125 MHz, CDCl₃): $\delta = 164.64$, 157.64, 147.89, 136.82, 124.72, 119.32, 24.39. **GC-MS** (M⁺) = 161.1.

2-(4,5-dihydro-1H-imidazol-2-yl)pyridine (L4)



In a two-neck round bottom flask, ethylenediamine (168 mg, 2.80 mmol) was added over pyridine-2-carboxaldehyde (250 mg, 2.33 mmol) in *tert*-butyl alcohol (20 mL). Then, the mixture was stirred at room temperature under an argon atmosphere for 30 min. After that, K₂CO₃ (967 mg, 7.00 mmol) and solid I₂ (770 mg, 3.03 mmol) were added to the reaction mixture and stirred at 70 °C for 12 h. The mixture was quenched with sat. aq. Na₂SO₃ until the iodine colour almost disappeared, and was extracted with CHCl₃. The organic layer was washed with sat. aq. NaHCO₃ solution and dried over anhydrous Na₂SO₄. After filtration, the mixture was evaporated under reduced pressure to provide crude product as a yellow oil. It was then purified by column chromatography on alumina using ethyl acetate and hexane as eluent to give the desire product (242 mg, yield: 73%) as a yellow solid. ¹H NMR (400 MHz, CDCl₃): $\delta = 8.46$ (d, J = 4.5 Hz, 1H), 8.05 (d, J = 7.9 Hz, 1H), 7.65 (t, J = 8.8 Hz, 1H), 7.24 (t, J = 8.7 Hz, 1H), 5.37 (brs, 1H), 3.73 (brs, 4H). ¹³C NMR (100 MHz, CDCl₃): $\delta = 164.33$, 148.60, 136.56, 125.05, 122.28, 50.52. GC-MS (M⁺) = 146.1.

2-(1-methyl-1H-imidazol-2-yl)pyridine (L5)



In a two-neck round bottom flask, a solution of pyridine-2-carboxaldehyde (107 mg, 1 mmol) in dichloromethane (10 mL) was dropwise added to *N*-methylethylenediamine (78 mg, 1.05 mmol) in 5 mL of DCM. The mixture was stirred for 1 h, and was then cooled to 0 °C. *N*-Bromosuccinimide (196 mg, 1.1 mmol) was added to the mixture and it was stirred overnight at room temperature. The reaction mixture was washed with 5% NaOH solution (10 mL) and then saturated Na₂S₂O₃ solution (10 mL), drying with Na₂SO₄ and removal of the dichloromethane under vacuum directly gave the desired crude product (142 mg, Yield: 90%) as yellow liquid. ¹H NMR (400 MHz, CDCl₃): $\delta = 8.60$ (d, J = 4.0 Hz, 1H), 7.82 (d, J = 8.0 Hz, 1H), 7.72 (td, J = 7.8, 1.7 Hz, 1H), 7.31-7.28 (m, 1H), 3.86 (t, J = 10.1, 2H), 3.47 (t, J = 10.0, 2H), 3.02 (s, 3H). ¹³C NMR (100 MHz, CDCl₃): $\delta = 165.80$, 150.71, 148.64, 136.67, 124.42, 124.26, 54.49, 53.07, 36.06. GC-MS (M⁺) = 161.1.

Synthesis of complex 1a



In a Schlenk flask a mixture of **L1** (50 mg, 0.28 mmol) and $[Cp*IrCl_2]_2$ (111.5 mg, 0.14 mmol) was stirred in dry DCM (20 mL) at room temperature for 24 h under an argon atmosphere. The resulting solution was filtered and insoluble part was removed. The filtrate was evaporated to dryness under reduced pressure and washed with diethyl ether to give pure compound (121.7 mg, Yield: 75%) as a yellow solid. ¹H NMR (500 MHz, CDCl₃): $\delta = 10.58$ (s, 1H), 8.84 (d, *J* = 7.5 Hz, 1H), 8.06 (t, *J* = 7.7 Hz, 1H), 7.15 (d, *J* = 8.6 Hz, 1H), 4.14 (s, 6H), 3.96-3.87 (m, 1H), 1.72 (s, 15H). ¹³C NMR (125 MHz, CDCl₃): $\delta = 170.17$, 163.95, 145.65, 143.72, 121.24, 110.67, 87.79, 57.98, 52.43, 46.06, 9.99. HRMS (ESI): Calcd. for C₁₉H₂₆ClIrN₃O [M-Cl]⁺: 540.1394; found: 540.1395. Anal. Calculated (C₁₉H₂₆Cl₂IrN₃O): C, 39.65; H, 4.55; N, 7.30; Found: C, 39.43; H, 4.33; N, 7.17.

Synthesis of complex 1b



In a round bottom flask a mixture of **L2** (50 mg, 0.3 mmol) and $[Cp*IrCl_2]_2$ (119.5 mg, 0.15 mmol) was taken. Then, dry MeOH (20 mL) was added to it and reflux for 24 h under argon atmosphere. The resulting solution was filtered and insoluble part was removed. The filtrate was evaporated to dryness under reduced pressure and washed with diethyl ether to give pure

compound (Yield: 121.03 mg, 65%) as a yellowish orange solid. ¹H NMR (500 MHz, DMSOd₆): $\delta = 9.28$ (s, 1H), 7.93 (t, J = 7.8 Hz, 1H), 7.68 (d, J = 7.1 Hz, 1H), 7.33 (d, J = 8.4 Hz, 1H), 4.35-4.29 (m, 1H), 3.92-3.86 (m, 2H), 3.75-3.68 (m, 1H), 1.59 (s, 15H). ¹³C NMR (125 MHz, CD₃OD): 168.02, 154.08, 142.87, 118.79, 117.59, 92.30, 87.31, 52.17, 45.34, 8.64. HRMS (ESI): Calcd. for C₁₈H₂₄ClIrN₃O [M-Cl]⁺: 526.1237; found: 526.1238. Anal. Calculated (C₁₈H₂₄Cl₂IrN₃O): C, 38.50; H, 4.31; N, 7.48; Found: C, 38.37; H, 4.19; N, 7.34.

Synthesis of complex 1c



A mixture of **L3** (50 mg, 0.31 mmol) and [Cp*IrCl₂]₂ (123.4 mg, 0.155 mmol) was stirred in dry DCM (20 mL) at room temperature for 24 h under argon atmosphere. The resulting solution was filtered and insoluble part was removed. The filtrate was evaporated to dryness under reduced pressure and washed with diethyl ether to give pure compound (124.9 mg, yield: 72%) as a yellow solid. ¹H NMR (500 MHz, CDCl₃): $\delta = 10.55$ (s, 1H), 8.98 (d, J = 7.5 Hz, 1H), 7.88 (t, J = 7.7 Hz, 1H), 7.55 (d, J = 7.8 Hz, 1H), 4.17-4.06 (m, 3H), 3.88-3.82 (m, 1H), 2.90 (s, 3H), 1.61 (s, 15H). ¹³C NMR (125 MHz, CDCl₃): 169.60, 161.97, 146.63, 139.96, 128.96, 125.16, 87.81, 52.03, 45.71, 28.84, 9.88. HRMS (ESI): Calcd. for C₁₉H₂₆ClIrN₃ [M-Cl]⁺: 524.1444; found: 524.1441. Anal. Calculated (C₁₉H₂₆Cl₂IrN₃): C, 40.78; H, 4.68; N, 7.51; Found: C, 40.43; H, 4.29; N, 7.39.

Synthesis of complex 1d



A mixture of **L4** (50 mg, 0.34 mmol) and $[Cp*IrCl_2]_2$ (135 mg, 0.17 mmol) was stirred in dry DCM (20 mL) at room temperature for 24 h under argon atmosphere. The resulting solution was filtered and insoluble part was removed. The filtrate was evaporated to dryness under reduced pressure and washed with diethyl ether to give pure compound (168 mg, yield: 82%) as a yellow solid. ¹H NMR (500 MHz, CDCl_3): $\delta = 10.58$ (s, 1H), 9.15 (d, J = 7.8 Hz, 1H), 8.71 (d, J = 5.5 Hz, 1H), 8.01 (t, J = 7.7 Hz, 1H), 7.65 (t, J = 6.1 Hz, 1H), 4.15-4.07 (m, 3H), 3.88-3.83 (m, 1H), 1.67 (s, 15H). ¹³C NMR (125 MHz, CDCl_3): 169.40, 151.15, 147.12, 140.19, 129.31, 128.00, 87.77, 52.02, 46.28, 9.26. HRMS (ESI): Calcd. for C₁₈H₂₄ClIrN₃ [M-Cl]⁺: 510.1288; found: 510.1276. Anal. Calculated (C₁₉H₂₄Cl₂IrN₃): C, 39.63; H, 4.43; N, 7.70; Found: C, 39.42; H, 4.27; N, 7.49.



Synthesis of complex 2

In a round bottom flask a mixture of **L2** (40 mg, 0.24 mmol) and $[Cp*Ir(H_2O)_3](OTf)_2$ (158.7 mg, 0.24 mmol) was taken under argon atmosphere. Then, H₂O (20 mL) was added to it and stirred at room temperature for 18 h. The resulting solution was filtered and insoluble part was removed. The filtrate was evaporated to dryness under reduced pressure and washed with diethyl ether to give pure compound (Yield: 134 mg, 68%) as a yellowish-orange solid. ¹H **NMR** (500 MHz, CD₃OD): $\delta = 7.88$ (t, J = 8.3 Hz, 1H), 7.40 (d, J = 7.2 Hz, 1H), 7.06 (d, J = 8.4 Hz, 1H), 3.98 (t, J = 9.7 Hz, 2H), 3.29-3.28 (m, 2H), 1.74 (s, 15H). ¹³C **NMR** (125 MHz, CD₃OD): 170.24, 165.10, 143.98, 141.30, 121.70, 119.17, 115.46, 88.08, 52.44, 45.49, 8.56. **HRMS (ESI)**: Calcd. for C₁₈H₂₅IrN₃O₂ [M-H₂O-2OTf-H]⁺: 490.1470; found: 490.1521. Anal. Calculated (C₂₀H₂₆F₆IrN₃O₈S₂): C, 29.77; H, 3.25; N, 5.21; Found: C, 29.43; H, 3.13; N, 5.09.

Synthesis of complex 1e



A mixture of **L5** (35 mg, 0.21 mmol) and $[Cp*IrCl_2]_2$ (86.5 mg, 0.108 mmol) was stirred in dry DCM (16 mL) at room temperature for 24 h under argon atmosphere. The resulting solution was filtered and insoluble part was removed. The filtrate was evaporated to dryness under reduced pressure and washed with diethyl ether to give pure compound (97.2 mg, yield: 75%) as a yellow solid. ¹H NMR (500 MHz, CDCl_3): $\delta = 8.90$ (d, J = 4.9 Hz, 1H), 8.52 (d, J = 7.8 Hz, 1H), 8.24 (t, J = 7.6 Hz, 1H), 7.85 (t, J = 5.2 Hz, 1H), 4.20-4.13 (m, 2H), 4.06-3.99 (m, 1H), 3.83-3.76 (m, 1H), 3.49 (s, 3H), 1.66 (s, 15H). ¹³C NMR (125 MHz, CDCl_3): 167.20, 153.12, 145.95, 140.70, 130.08, 127.28, 88.33, 55.86, 50.67, 36.02, 9.28. HRMS (ESI): Calcd. for C₁₉H₂₆ClIrN₃ [M-Cl]⁺: 524.1444; found: 524.1453. Anal. Calculated (C₁₉H₂₆Cl₂IrN₃): C, 40.78; H, 4.68; N, 7.51; Found: C, 40.47; H, 4.33; N, 7.43.

3. X-ray crystallographic studies.

Single crystal of complex **1b** was obtained from DCM-methanol-hexane mixture under slow evaporation. Single-crystal X-ray data of complex **1b** was collected at 100 K by using a Bruker SMART APEX II CCD diffractometer and Bruker D8 Quest Single Crystal diffractometer with graphite monochromatic MoK_{α} radiation (λ =0.71073 Å). The frames were indexed, integrated and scaled using SMART and SAINT software package and the data were corrected for absorption using the SADABS program. The structure was solved and refined using WINGX

and SHELX programs. The crystallographic figure has been generated using Diamond 3 software10 (30% probability thermal ellipsoids). The CCDC number of complex **1b** is CCDC 1878158. One molecule dichloromethane was co-crystallized in this method.



Fig. S1 Solid state structure of complex 1b (30% thermal ellipsoids, Cl counter anion was omitted for clarity).

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Identification code	1b
Empirical formula	C ₁₉ H ₂₆ Cl ₄ IrN ₃ O
Formula weight	646.43
Temperature/K	100(2)
Crystal system	monoclinic
Space group	$P2_1/n$
a/Å	12.768(3)
b/Å	14.165(3)
c/Å	13.588(3)
$\alpha/^{\circ}$	90
β/°	111.994(6)
$\gamma/^{\circ}$	90
Volume/Å ³	2278.6(10)

Z	4
$\rho_{calc}g/cm^3$	1.884
μ/mm^{-1}	6.344
F(000)	1256.0
Crystal size/mm ³	$0.010\times0.010\times0.010$
Radiation	MoKα ($\lambda = 0.71073$)
2Θ range for data collection/ ^c	^o 5.534 to 56.568
Index ranges	$-17 \le h \le 17, -18 \le k \le 18, -18 \le l \le 18$
Reflections collected	35066
Independent reflections	5634 [$R_{int} = 0.0556$, $R_{sigma} = 0.0324$]
Data/restraints/parameters	5634/0/259
Goodness-of-fit on F ²	1.069
Final R indexes [I>= 2σ (I)]	$R_1 = 0.0228, wR_2 = 0.0573$
Final R indexes [all data]	$R_1 = 0.0257, wR_2 = 0.0591$
Largest diff. peak/hole / e Å ⁻³	3 1.67/-1.05

4: Synthesis of quinoxaline derivatives

4A. General procedure for quinoxaline synthesis from di-amine derivatives: In a Schlenk tube 1,2-diamine derivatives (0.5 mmol), diol (2.5 mmol), Cat. **1b** (2.5 mol %), KOH (0.75 mmol) and H₂O (3.0 mL) were taken under air. The reaction mixture was heated at 120 $^{\circ}$ C in a preheated oil bath for 24 h. Then, it was cooled to room temperature and extracted with ethyl acetate. Finally, the quinoxaline derivatives were purified by column chromatography using hexane/ethyl acetate as eluent to afford the corresponding pure product.

4B. General procedure for quinoxaline synthesis from nitroamine derivatives: In a Schlenk tube 1,2-nitroamine derivatives (0.5 mmol), diol (2.5 mmol), Cat. **1b** (2.5 mol %), KOH (0.75 mmol) and H₂O (3.0 mL) were taken under air. The reaction mixture was heated at 120 °C in a preheated oil bath for 24 h. Then, it was cooled to room temperature and extracted with ethyl acetate. Finally, the quinoxaline derivatives were purified by column chromatography using hexane/ethyl acetate as eluent to afford the corresponding pure product.

5. Synthesis of benzimidazole derivatives

5A. General procedure for benzimidazole synthesis from di-amine derivatives: In a Schlenk tube 1,2-diamine derivatives (0.3 mmol), primary alcohol (0.45 mmol), Cat. **2** (2.5 mol %) and H₂O (2.0 mL) were taken under argon atmosphere. The reaction mixture was heated at 120 $^{\circ}$ C in a preheated oil bath for 24 h. Then, it was cooled to room temperature and extracted with ethyl acetate. Finally, the benzimidazole derivatives were purified by column chromatography using hexane/ethyl acetate as eluent to afford the corresponding pure product.

6. Synthesis of quinazolines derivatives

6A. General procedure for quinazolines synthesis from di-amine derivatives: In a Schlenk tube 2-aminobenzyl amine (0.3 mmol), primary alcohol (0.45 mmol), Cat. 2 (2.0 mol %) and H₂O (2.0 mL) were taken under argon atmosphere. The reaction mixture was heated at 120 $^{\circ}$ C in a preheated oil bath for 24 h. Then, it was cooled to room temperature and extracted with

ethyl acetate. Finally, the quinazolines were purified by column chromatography using hexane/ethyl acetate as eluent to afford the corresponding pure product.

7. Procedure for synthesis of pharmaceutically active quinoxaline derivatives

In a Schlenk flask 1,2-diamine/nitroamine derivatives (2 mmol), diol (10 mmol), Cat. **1b** (2.5 mol %), KOH (3 mmol) and H₂O (15 mL) were taken under air. The reaction mixture was heated at 120 $^{\circ}$ C in a preheated oil bath for 24 h. Then, it was cooled to room temperature and extracted with ethyl acetate. Finally, the quinoxalines were purified by column chromatography using hexane/ethyl acetate as eluent to afford the corresponding pure product.



Scheme S1 synthesis of pharmaceutically active quinoxaline derivatives.

8. Procedure for preparative scale reaction

8A. Quinoxaline derivatives synthesis

In a Schlenk flask 1,2-diamine/nitroamine derivatives (3 mmol), diol (15 mmol), Cat. **1b** (2.5 mol %), KOH (4.5 mmol) and H₂O (20 mL) were taken under air. The reaction mixture was heated at 120 $^{\circ}$ C in a preheated oil bath for 24 h. Then, it was cooled to room temperature and extracted with ethyl acetate. Finally, the quinoxaline derivatives were purified by column chromatography using hexane/ethyl acetate as eluent to afford the corresponding pure product.

8B. Benzimidazole derivatives synthesis

In a Schlenk flask 1,2-diamine derivatives (3 mmol), primary alcohol (4.5 mmol), Cat. **2** (2.5 mol%) and H₂O (25 mL) were taken under argon atmosphere. The reaction mixture was heated at 120 $^{\circ}$ C in a preheated oil bath for 24 h. Then, it was cooled to room temperature and extracted with ethyl acetate. Finally, the benzimidazole derivatives were purified by column chromatography using hexane/ethyl acetate as eluent to afford the corresponding pure product.

8C. Quinazoline derivatives synthesis

In a Schlenk flask 2-aminobenzyl amine (3 mmol), primary alcohol (4.5 mmol), Cat. **2** (2.0 mol%) and H_2O (25 mL) were taken under argon atmosphere. The reaction mixture was heated at 120 °C in a preheated oil bath for 24 h. Then, it was cooled to room temperature and extracted with ethyl acetate. Finally, the quinazolines derivatives were purified by column chromatography using hexane/ethyl acetate as eluent to afford the corresponding pure product.

a) From diamines



Scheme S2 Preparative scale reaction.

9. Control experiments and reusability

9A. Procedure for control experiments

In a Schlenk tube 1,2-diamine/nitroamine derivatives (0.5 mmol), diol (2.5 mmol), Cat. **1b** (2.5 mol %), KOH (0.75 mmol) and H₂O (3.0 mL) were taken under air. The reaction mixture was heated at 120 $^{\circ}$ C in a preheated oil bath for different time interval. Then, it was cooled to room temperature and extracted with ethyl acetate. Finally the yield of quinoxaline derivatives were confirmed by Gas chromatography using mesitylene as internal standard.

9B. Effect of proton responsive N-H functionality



Scheme S3 Effect of proton responsive N-H functionality.

9C. Synthesis of Ir-H species of Complex 1b

In a J-Young NMR tube, complex **1b** (0.02 mmol), KO_tBu (0.06 mmol) and ^{*i*}PrOH (0.5 mL) were taken and heated at 82 °C for 30 minutes. After that, solvent was removed in vacuum and ¹H NMR spectrum was recorded in DMSO-d6.



Fig. S2 ¹H NMR spectrum of Ir-H for complex 1b.

9D. Activity difference of cat 1b and 2 in aerobic and argon condition

Table S1. Catalytic activity difference of cat 1b and 2

Entry	Catalyst	Base	Condition	Conversion
1	1b	KOH (1.5 equiv.)	Aerobic	24
2	1b	KOH (1.5 equiv.)	Argon	4
3	2	-	Argon	55
4	2	-	Aerobic	23

Ir-Cat, H ₂ O, 120 [°] C	
 Argon/Air, 12 h	0

Reaction Conditions: Benzyl alcohol (0.15 mmol), Ir cat. (2 mol %), H₂O (1.0 mL), 120 $^{\circ}$ C (oil bath temperature), 12 h. Yield determined by GC analysis.

9E. Procedure for reusability

In a Schlenk tube benzene-1,2-diamine (0.5 mmol), propane-1,2-diol (2.5 mmol), Cat. **1b** (2.5 mol%), KOH (0.75 mmol) and H₂O (3.0 mL) were taken under air. The reaction mixture was heated at 120 $^{\circ}$ C in a preheated oil bath for 24 h. Then, it was cooled to room temperature and extracted with diethylether and the aqueous layer containing the catalyst was used for the next run. In the successive runs, benzene-1,2-diamine (0.5 mmol), propane-1,2-diol (2.5 mmol) and

KOH (0.75 mmol) were added to the aqueous layer. The yield of product C1 was determined by GC analysis using mesitylene as an internal standard.

Table S2. Reusability of catalytic system for synthesis of 2-methylquinoxaline

Run	1	2	3
Yield (%)	92	57	23

10. Proposed pathways of the reactions

10A. Stepwise mechanism of quinoxaline synthesis from diamine



Scheme S4 Stepwise mechanism of quinoxaline synthesis from diamine.

10B. Stepwise mechanism of quinoxaline synthesis from nitroamine



Scheme S5 Stepwise mechanism of quinoxaline synthesis from nitroamine.

10C. Stepwise mechanism of benzimidazole synthesis from diamine



2,3-dihydro-1*H*-benzoimidazole

Scheme S6 Stepwise mechanism of benzimidazole synthesis from diamine.

10D. Proposed mechanism of alcohol dehydrogenation by cat. 1b or 2



Scheme S7 Proposed mechanism of alcohol dehydrogenation by cat. 1b or 2.

11. Characterization of the products



2-methylquinoxaline $(C1)^3$

62 mg; 86% isolated yield from diamine and 63 mg; 87% isolated yield from nitroamine; yellow liquid. ¹**H NMR** (CDCl₃, 400 MHz): $\delta = 8.71$ (s, 1H), 8.04 (dd, J = 8.6 Hz, 1.6 Hz, 1H), 7.98 (dd, J = 7.6 Hz, 1.6 Hz, 1H), 7.73-7.65 (m, 2H), 2.74 (s, 3H). ¹³**C NMR** (CDCl₃, 100 MHz): $\delta = 153.87$, 146.10, 142.15, 141.04, 130.10, 129.26, 129.01, 128.74, 22.68. **GC-MS** (M⁺) = 144.1.



2,3-dimethylquinoxaline $(C2)^4$

70 mg; 89% isolated yield from diamine and 72 mg; 90% isolated yield from nitroamine; yellow solid. ¹H NMR (CDCl₃, 400 MHz): δ = 7.96 (dd, *J* = 6.3 Hz, 3.6 Hz, 2H), 7.65 (dd, *J* = 6.4 Hz, 3.5 Hz, 2H), 2.71 (s, 6H). ¹³C NMR (CDCl₃, 100 MHz): δ = 153.58, 141.15, 128.94, 128.38, 23.29. GC-MS (M⁺) = 158.1.



2-ethylquinoxaline $(C3)^5$

56 mg; 70% isolated yield from diamine; yellow oil. ¹**H NMR** (CDCl₃, 500 MHz): $\delta = 8.69$ (s, 1H), 8.02-7.97 (m, 2H), 7.68-7.62 (m, 2H), 2.98 (q, J = 7.6 Hz, 2H), 1.37 (t, J = 7.7 Hz, 3H). ¹³**C NMR** (CDCl₃, 125 MHz): $\delta = 158.49$, 145.61, 142.19, 141.26, 129.94, 129.21, 128.93, 128.90, 29.67, 13.46. **GC-MS** (M⁺) = 158.0.



2-butylquinoxaline $(C4)^5$

68 mg; 73% isolated yield from diamine and 72 mg; 77% isolated yield from nitroamine; yellow oil. ¹**H NMR** (CDCl₃, 400 MHz): $\delta = 8.72$ (s, 1H), 8.06-8.00 (m, 2H), 7.74-7.66 (m, 2H), 3.00 (t, J = 7.9 Hz, 2H), 1.85-1.77 (m, 2H), 1.49-1.39 (m, 2H), 0.95 (t, J = 7.4 Hz, 3H). ¹³**C NMR** (CDCl₃, 100 MHz): $\delta = 157.79$, 145.93, 142.27, 141.27, 130.00, 129.24, 128.99, 128.93, 36.35, 31.75, 22.67, 13.99. **GC-MS** (M⁺) = 186.1.

2-tert-butylquinoxaline (C5)⁶

63 mg; 68% isolated yield from diamine and 75 mg; 80% isolated yield from nitroamine; yellow oil. ¹H NMR (CDCl₃, 500 MHz): $\delta = 8.97$ (s, 1H), 8.06-8.03 (m, 2H), 7.73-7.66 (m, 2H), 1.50 (s, 9H). ¹³C NMR (CDCl₃, 125 MHz): $\delta = 163.77$, 143.50, 141.70, 140.87, 129.71, 129.37, 128.98, 128.94, 37.33, 29.83. **GC-MS** (M⁺) = 186.0.

2-phenylquinoxaline (C6)⁷

88 mg; 85% isolated yield from diamine and 95 mg; 92% isolated yield from nitroamine; yellow solid. ¹H NMR (CDCl₃, 400 MHz): $\delta = 9.31$ (s, 1H), 8.19-8.09 (m, 4H), 7.79-7.70 (m, 2H), 7.57-7.50 (m, 3H). ¹³C NMR (CDCl₃, 100 MHz): $\delta = 151.96$, 143.47, 142.40, 141.68, 136.88, 130.38, 130.28, 129.72, 129.64, 129.25, 127.65. GC-MS (M⁺) = 206.1.

2,3-diphenylquinoxaline $(\mathbf{C7})^7$

109 mg; 77% isolated yield from diamine and 116 mg; 83% isolated yield from nitroamine; cream yellow solid. ¹**H NMR** (CDCl₃, 400 MHz): $\delta = 8.17$ (dd, J = 6.5 Hz, 3.7 Hz, 2H), 7.76 (dd, J = 6.5 Hz, 3.4 Hz, 2H), 7.52-7.50 (m, 4H), 7.35-7.30 (m, 6H). ¹³**C NMR** (CDCl₃, 100 MHz): $\delta = 153.56$, 141.33, 139.18, 130.03, 129.93, 129.30, 128.88, 128.35. **GC-MS** (M⁺) = 282.1.

2,6,7-trimethylquinoxaline $(C8)^8$

62 mg; 72% isolated yield from diamine and 72 mg; 83% isolated yield from nitroamine; yellow solid. ¹H NMR (CDCl₃, 500 MHz): $\delta = 8.61$ (s, 1H), 7.77 (s, 1H), 7.73 (s, 1H), 2.71 (s, 3H), 2.45 (s, 6H). ¹³C NMR (CDCl₃, 125 MHz): $\delta = 152.77$, 145.09, 141.09, 140.53, 139.99, 139.32, 128.27, 127.84, 22.52, 20.44, 20.28. **GC-MS** (M⁺) = 172.1.



2,6-dimethylquinoxaline (C9a) and 2,7-dimethylquinoxaline (C9b)⁹

This two regio-isomers were inseparable in column chromatography. The mixture yield is 60 mg; 76% isolated yield from diamine; yellow solid. ¹H NMR (CDCl₃, 400 MHz): $\delta = 8.68$ -8.65 (s, 1H), 7.94-7.87 (m, 1H), 7.81-7.76 (s, 1H), 7.56-7.50 (m, 1H), 2.74 (s, 3H), 2.56 (s, 3H). ¹³C NMR (CDCl₃, 100 MHz): $\delta = 153.75$, 152.88, 145.98, 145.18, 142.26, 141.13, 140.62, 139.54, 139.41, 132.37, 131.27, 128.76, 128.26, 128.13, 127.67, 22.64, 22.54, 21.94, 21.78. **GC-MS** (M⁺) = 158.0.



6-methyl-2-phenylquinoxaline (C10a) and 7-methyl-2-phenylquinoxaline (C10b)¹⁰

This two regio-isomers were inseparable in column chromatography. The mixture yield is 90 mg; 82% isolated yield from diamine; yellow solid. ¹H NMR (CDCl₃, 400 MHz): δ = 9.26-9.24 (s, 1H), 8.18-8.15 (m, 2H), 8.03-7.97 (d, *J* = 8.6 Hz, 1H), 7.91-7.87 (s, 1H), 7.67-7.49 (m, 4H), 2.59 (s, 3H). ¹³C NMR (CDCl₃, 100 MHz): δ = 151.86, 151.15, 143.33, 142.54, 142.46, 141.72, 140.90, 140.85, 140.20, 140.16, 137.04, 132.69, 131.95, 130.14, 130.04, 129.20, 128.69, 128.55, 128.05, 127.57, 127.50, 21.96, 21.93. **GC-MS** (M⁺) = 220.1.

This two regio-isomers (C11a and C11b) were separated by column chromatography. The total (C11a + C11b) yield is 102 mg; 86% isolated yield from diamine with selectivity C11a:C11b = 3.4:1; the total (C11a + C11b) yield is 107 mg; 90% isolated yield from nitroamine with selectivity C11a:C11b = 4.1:1; cream yellow solid.



6-methoxy-2-phenylquinoxaline (C11a)³

¹**H NMR** (CDCl₃, 400 MHz): $\delta = 9.23$ (s, 1H), 8.15-8.13 (m, 2H), 8.02 (d, J = 9.1 Hz, 1H), 7.56-7.47 (m, 3H), 7.44-7.38 (m, 2H), 3.98 (s, 3H). ¹³**C NMR** (CDCl₃, 100 MHz): $\delta = 160.64$, 149.73, 143.31, 138.50, 137.09, 130.66, 129.80, 129.19, 127.29, 123.66, 106.60, 55.92. **GC-MS** (M⁺) = 236.1.

7-methoxy-2-phenylquinoxaline (C11b)³

¹**H NMR** (CDCl₃, 400 MHz): $\delta = 9.14$ (s, 1H), 8.15-8.13 (m, 2H), 7.96 (d, J = 9.1 Hz, 1H), 7.55-7.48 (m, 3H), 7.41-7.35 (m, 2H), 3.96 (s, 3H). ¹³**C NMR** (CDCl₃, 100 MHz): $\delta = 161.12$, 152.00, 144.03, 140.82, 137.84, 137.06, 130.12, 129.20, 127.59, 122.98, 106.93, 55.91. **GC-MS** (M⁺) = 236.1.



2,3,6-trimethylquinoxaline $(C12)^{11}$

66 mg; 77% isolated yield from diamine; pale yellow solid. ¹H NMR (CDCl₃, 400 MHz): δ = 7.83 (d, *J* = 8.5 Hz, 1H), 7.72 (s, 1H), 7.47 (d, *J* = 8.5 Hz, 1H), 2.68 (s, 6H), 2.53 (s, 3H). ¹³C NMR (CDCl₃, 100 MHz): δ = 153.38, 152.52, 141.18, 139.54, 139.23, 131.12, 127.88, 127.36, 23.25, 23.15, 21.83. **GC-MS** (M⁺) = 172.0.



6-methoxy-2,3-dimethylquinoxaline (C13)³

79 mg; 84% isolated yield from diamine and 84 mg; 89% isolated yield from nitroamine; yellow solid. ¹**H** NMR (CDCl₃, 400 MHz): $\delta = 7.82$ (d, J = 8.9 Hz, 1H), 7.29-7.25 (m, 2H), 3.90 (s, 3H), 2.67 (s, 3H), 2.65 (s, 3H). ¹³**C** NMR (CDCl₃, 100 MHz): $\delta = 160.01$, 153.42, 150.69, 142.51, 137.06, 129.31, 121.73, 106.22, 55.73, 23.18, 22.87. **GC-MS** (M⁺) = 188.0.

6-bromo-2,3-dimethylquinoxaline (C14)¹²

87 mg; 73% isolated yield from diamine; yellow solid. ¹**H** NMR (CDCl₃, 500 MHz): $\delta = 8.14$ (d, J = 1.9 Hz, 1H), 7.82 (d, J = 8.9 Hz, 1H), 7.72 (dd, J = 8.9 Hz, 2 Hz, 1H) 2.71 (s, 3H), 2.69 (s, 3H). ¹³**C** NMR (CDCl₃, 125 MHz): $\delta = 154.60$, 154.00, 141.82, 139.92, 132.40, 130.86, 130.81, 129.79, 122.56, 23.26. **GC-MS** (M⁺) = 237.0.

2,3-dimethylbenzo[g]quinoxaline (C15)¹³

65 mg; 62% isolated yield from diamine; yellow solid. ¹H NMR (CDCl₃, 400 MHz): δ = 8.51 (s, 2H), 8.05 (dd, *J* = 6.4 Hz, 3.2 Hz, 2H), 7.51 (dd, *J* = 6.4 Hz, 3.0Hz, 2H), 2.76 (s, 6H). ¹³C NMR (CDCl₃, 100 MHz): δ = 154.54, 138.15, 133.36, 128.46, 126.51, 126.40, 23.75. GC-MS (M⁺) = 208.1.

2-phenylbenzo[g]quinoxaline (C16)¹⁴

89 mg; 69% isolated yield from diamine; yellow solid. ¹H NMR (CDCl₃, 400 MHz): δ = 9.38 (s, 2H), 8.71 (s, 1H), 8.67 (s, 1H), 8.26 (d, *J* = 7.1 Hz, 2H), 8.12-8.09 (m, 2H), 7.60-7.54 (m, 5H). ¹³C NMR (CDCl₃, 100 MHz): δ = 151.92, 144.68, 138.86, 138.25, 136.89, 134.31, 133.74, 130.55, 129.31, 128.64, 128.56, 128.06, 127.75, 127.67, 127.02, 126.85. GC-MS (M⁺) = 256.0.

3-methylpyrido[2,3-b]pyrazine (C17)¹⁵

47 mg; 65% isolated yield from diamine; cream yellow solid. ¹H NMR (CDCl₃, 500 MHz): δ = 9.11 (t, *J* = 1.7 Hz, 1H), 8.81 (s, 1H), 8.41 (dd, *J* = 8.3 Hz, 1.8 Hz, 1H), 7.65 (dd, *J* = 8.3 Hz, 4.2 Hz, 1H), 2.84 (s, 3H). ¹³C NMR (CDCl₃, 125 MHz): δ = 157.76, 154.02, 150.86, 147.22, 138.26, 136.06, 124.51, 22.91. GC-MS (M⁺) = 144.9.



2,3-dimethylpyrido[2,3-b]pyrazine (C18)¹⁶

56 mg; 70% isolated yield from diamine; yellow solid. ¹H NMR (CDCl₃, 500 MHz): δ = 9.03 (s, 1H), 8.32 (d, *J* = 7.8 H, 1H), 7.60 (dd, *J* = 7.9 Hz, 3.8 Hz, 1H), 2.79 (s, 3H), 2.75 (s, 3H). ¹³C NMR (CDCl₃, 125 MHz): δ = 157.44, 155.12, 152.81, 150.30, 137.43, 135.98, 124.41, 23.58, 23.10. **GC-MS** (M⁺) = 159.1.



6,7-dimethoxy-2-phenylquinoxaline (C19)¹⁷

117 mg; 88% isolated yield from diamine; pale yellow solid. ¹H NMR (CDCl₃, 400 MHz): δ = 9.09 (s, 1H), 8.11 (d, *J* = 7.3 Hz, 2H), 7.54-7.44 (m, 3H), 7.40 (s, 1H), 7.35 (s, 1H), 4.50 (s, 3H), 4.04 (s, 3H). ¹³C NMR (CDCl₃, 100 MHz): δ = 153.11, 152.64, 149.93, 140.76, 139.58, 138.83, 137.30, 129.67, 129.15, 128.60, 127.24, 126.15, 107.11, 106.67, 56.47, 56.44. GC-MS (M⁺) = 266.1.

This two regio-isomers (C20a and C20b) were separated by column chromatography. The total (C20a + C20b) yield is 105 mg; 87% isolated yield from nitroamine with selectivity C11a:C11b = 1:1.7; cream yellow solid.

6-chloro-2-phenylquinoxaline $(C20a)^3$

¹**H** NMR (CDCl₃, 400 MHz): $\delta = 9.31$ (s, 1H), 8.18-8.15 (m, 2H), 8.10-8.06 (m, 2H), 7.71 (dd, J = 9.1 Hz, 2.3 Hz, 1H), 7.58-7.50 (m, 3H). ¹³**C** NMR (CDCl₃, 100 MHz): $\delta = 152.04$, 144.23, 141.87, 140.92, 136.43, 135.35, 131.43, 130.92, 130.54, 129.31, 128.15, 127.60. **GC-MS** (M⁺) = 240.0.

7-chloro-2-phenylquinoxaline (C20b)³

¹**H NMR** (CDCl₃, 400 MHz): $\delta = 9.36$ (s, 1H), 8.25-8.20 (m, 3H), 8.10 (d, J = 9.0 Hz, 1H), 7.73 (dd, J = 9.0 Hz, 2.3 Hz, 1H), 7.65-7.57 (m, 3H). ¹³**C NMR** (CDCl₃, 100 MHz): $\delta = 152.63$, 143.49, 142.74, 140.17, 136.38, 136.19, 130.64, 130.60, 130.43, 129.30, 128.58, 127.69. **GC-MS** (M⁺) = 240.0.



2-butyl-6,7-dimethylquinoxaline (C21)¹⁸

90 mg; 84% isolated yield from nitroamine; yellow solid. ¹H NMR (CDCl₃, 400 MHz): $\delta = 8.60$ (s, 1H), 7.77 (s, 1H), 7.75 (s, 1H), 2.94 (t, J = 8.0 Hz, 2H), 2.45 (s, 6H), 1.82-1.74 (m, 2H), 1.46-1.37 (m, 2H), 0.94 (t, J = 7.5 Hz, 3H). ¹³C NMR (CDCl₃, 100 MHz): $\delta = 156.73$, 144.91, 140.42, 140.19, 139.31, 128.24, 128.01, 36.24, 31.84, 22.65, 20.45, 20.31, 13.99. GC-MS (M⁺) = 214.0.

6,7-dimethyl-2-phenylquinoxaline (C22)¹⁰

106 mg; 90% isolated yield from nitroamine; yellow solid. ¹H NMR (CDCl₃, 400 MHz): δ = 9.19 (s, 1H), 8.15-8.13 (m, 2H), 7.87 (s, 1H), 7.82 (s, 1H), 7.55-7.45 (m, 3H), 2.84 (s, 6H). ¹³C NMR (CDCl₃, 100 MHz): δ = 151.06, 142.47, 141.30, 140.87, 140.62, 140.19, 137.20, 129.91, 129.15, 128.73, 128.21, 127.46, 20.48, 20.45. **GC-MS** (M⁺) = 234.0.



7-bromo-2,3,5-trimethylquinoxaline (C23)¹¹

98 mg; 78% isolated yield from nitroamine; yellow solid. ¹**H** NMR (CDCl₃, 500 MHz): $\delta =$ 7.96 (d, *J* = 1.9 Hz, 1H), 7.57 (d, *J* = 1.0 Hz, 1H), 2.72 (s, 3H), 2.69 (s, 6H). ¹³**C** NMR (CDCl₃, 125 MHz): $\delta =$ 153.98, 152.63, 141.83, 139.18, 138.78, 132.11, 128.53, 122.16, 23.39, 23.12, 16.93. **GC-MS** (M⁺) = 250.9.



7-bromo-5-methyl-2,3-diphenylquinoxaline (C24)

137 mg; 73% isolated yield from nitroamine; yellow solid. ¹H NMR (CDCl₃, 500 MHz): δ = 8.17 (s, 1H), 7.68 (s, 1H), 7.56-7.49 (m, 4H), 7.38-7.30 (m, 6H), 2.81 (s, 3H). ¹³C NMR (CDCl₃, 125 MHz): δ = 153.75, 152.14, 141.83, 139.61, 139.25, 139.13, 139.04, 133.15, 130.13, 129.88, 129.25, 129.05, 128.97, 128.38, 128.23, 123.63, 16.94. GC-MS (M⁺) = 374.9. HRMS (ESI): Calcd. for C₂₁H₁₅BrN₂ [M]⁺: 374.0419; found: 374.0417.

2,3-diphenylpyrido[3,4-b]pyrazine (C25)¹⁹

99 mg; 70% isolated yield from nitroamine; pale yellow solid. ¹H NMR (CDCl₃, 500 MHz): δ = 9.60 (s, 1H), 8.82 (d, *J* = 6.0 Hz, 1H), 7.99 (d, *J* = 5.8 Hz, 1H), 7.54-7.52 (m, 4H), 7.41-7.38 (m, 2H), 7.37-7.34 (m, 4H). ¹³C NMR (CDCl₃, 125 MHz): δ = 158.05, 155.48, 154.54, 147.39, 143.68, 138.37, 136.41, 129.95, 129.86, 129.77, 129.50, 128.50, 121.45. GC-MS (M⁺) = 283.1.



2-phenyl-1H-benzimidazole (E1)²⁰

51 mg; 87% isolated yield; white solid. ¹**H NMR** (DMSO-D₆, 500 MHz): δ = 12.89 (brs, 1H), 8.14 (d, *J* = 7.4 Hz, 2H), 7.62-7.43 (m, 5H), 7.17 (d, *J* = 4.6 Hz, 2H). ¹³**C NMR** (DMSO-D₆, 125 MHz): δ = 151.76, 144.28, 135.47, 130.64, 130.40, 129.49, 126.96, 123.10, 122.23, 119.35, 111.87. **GC-MS** (M⁺) = 194.0.



2-(4-methoxyphenyl)-1H-benzimidazole (E2)²¹

57 mg; 84% isolated yield; cream white solid. ¹H NMR (DMSO-D₆, 400 MHz): δ = 12.73 (brs, 1H), 8.09 (d, *J* = 7.7 Hz, 2H), 7.65-7.39 (m, 2H), 7.13-7.06 (m, 4H), 3.79 (s, 3H). ¹³C NMR (DMSO-D₆, 100 MHz): δ = 161.12, 151.58, 144.32, 135.37, 131.04, 128.54, 127.93, 123.20, 122.26, 118.92, 114.88, 55.83. **GC-MS** (M⁺) = 224.1.



2-(4-chlorophenyl)-1H-benzimidazole (E3)²²

52 mg; 75% isolated yield; white solid. ¹**H NMR** (DMSO-D₆, 400 MHz): δ = 12.95 (brs, 1H), 8.15 (d, *J* = 8.6 Hz, 2H), 7.70-7.49 (m, 4H), 7.22-7.10 (m, 2H). ¹³**C NMR** (DMSO-D₆, 100 MHz): δ = 150.68, 144.26, 135.54, 135.02, 129.59, 128.66, 123.30, 122.37, 119.49, 111.95. **GC-MS** (M⁺) = 228.2.



 $2-(3-chlorophenyl)-1H-benzimidazole (E4)^{23}$

54 mg; 78% isolated yield; white solid. ¹**H NMR** (DMSO-D₆, 400 MHz): $\delta = 13.02$ (brs, 1H), 8.18 (s, 1H), 8.10 (d, J = 7.1 Hz, 1H), 7.56-7.50 (m, 4H), 7.24-7.18 (m, 2H). ¹³**C NMR** (DMSO-D₆, 100 MHz): $\delta = 150.26$, 147.89, 134.31, 132.70, 131.48, 130.09, 126.54, 125.54, 123.47, 122.53, 119.60, 112.07. **GC-MS** (M⁺) = 228.0.



2-o-tolyl-1H-benzimidazole (E5)²⁴

35 mg; 56% isolated yield; cream white solid. ¹**H NMR** (DMSO-D₆, 400 MHz): δ = 12.58 (brs, 1H), 7.70 (d, *J* = 7.1 Hz, 1H), 7.64 (d, *J* = 7.5 Hz, 1H), 7.49 (d, *J* = 7.4 Hz, 1H), 7.36-7.30 (m, 3H), 7.20-7.13 (m, 2H), 2.57 (s, 3H). ¹³**C NMR** (DMSO-D₆, 100 MHz): δ = 152.47, 144.24, 137.54, 134.95, 131.78, 130.62, 129.97, 129.83, 126.48, 122.88, 121.92, 119.45, 111.78, 21.55. **GC-MS** (M⁺) = 208.1.



2-(benzo[d][1,3]dioxol-5-yl)-1H-benzimidazole (E6)²²

57 mg; 80% isolated yield; cream white solid. ¹**H NMR** (DMSO-D₆, 400 MHz): δ = 12.71 (brs, 1H), 7.69 (dd, *J* = 8.0 Hz, 1.7 Hz, 1H), 7.65 (d, *J* = 1.4 Hz, 1H), 7.58 (brs, 1H), 7.46 (brs, 1H), 7.14 (d, *J* = 5.0 Hz, 2H), 7.05 (d, *J* = 8.0 Hz, 1H), 6.08 (s, 2H). ¹³**C NMR** (DMSO-D₆, 100 MHz): δ = 151.68, 149.26, 148.40, 144.21, 135.43, 124.78, 122.79, 122.11, 121.42, 119.10, 111.61, 109.26, 107.02, 102.10. **GC-MS** (M⁺) = 238.1.



2-(thiophen-2-yl)-1H-benzimidazole $(\mathbf{E7})^{25}$

43 mg; 72% isolated yield; pale yellow solid. ¹**H NMR** (DMSO-D₆, 400 MHz): $\delta = 12.92$ (brs, 1H), 7.79 (dd, J = 3.6 Hz, 1.0 Hz, 1H), 7.68 (dd, J = 4.9 Hz, 0.72 Hz, 1H), 7.57 (d, J = 6.4 Hz, 1H), 7.46 (d, J = 6.0 Hz, 1H), 7.20-7.15 (m, 3H). ¹³**C NMR** (DMSO-D₆, 100 MHz): $\delta = 147.57$, 144.13, 135.21, 134.22, 129.27, 128.81, 127.22, 123.17, 122.30, 119.06, 111.63. **GC-MS** (M⁺) = 200.0.



2-(furan-2-yl)-1H-benzimidazole (E8)¹⁰

36 mg; 64% isolated yield; pale yellow solid. ¹H NMR (DMSO-D₆, 400 MHz): $\delta = 12.89$ (brs, 1H), 7.89 (s, 1H), 7.59 (d, J = 6.9 Hz, 1H), 7.45 (d, J = 6.8 Hz, 1H), 7.16 (d, J = 3.5 Hz, 3H), 6.69-6.67 (m, 1H). ¹³C NMR (DMSO-D₆, 100 MHz): $\delta = 146.10$, 145.14, 144.17, 134.76, 123.16, 122.33, 119.29, 112.85, 111.87, 111.00. **GC-MS** (M⁺) = 184.1.

2-pentyl-1H-benzimidazole (E9)²⁶

23 mg; 41% isolated yield; pale yellow solid. ¹H NMR (DMSO-D₆, 400 MHz): δ = 7.41 (dd, J = 5.9 Hz, 3.2 Hz, 2H), 7.05 (dd, J = 6.0 Hz, 3.1 Hz, 2H), 2.75 (t, J = 7.6 Hz, 2H), 1.75-1.68 (m, 2H), 1.27-1.24 (m, 4H), 0.81 (t, J = 6.9 Hz, 3H). ¹³C NMR (DMSO-D₆, 100 MHz): δ = 155.68, 139.31, 121.55, 114.85, 31.41, 29.01, 27.81, 22.37, 14.37. GC-MS (M⁺) = 188.0.



5-methyl-2-phenyl-1H-benzimidazole (E10)²⁷

51 mg; 81% isolated yield; pale yellow solid. ¹**H NMR** (DMSO-D₆, 400 MHz): $\delta = 12.72$ (brs, 1H), 8.12 (d, J = 7.2 Hz, 2H), 7.51-7.41 (m, 4H), 7.38-7.27 (m, 1H), 7.01-6.69 (m, 1H), 2.39 (s, 3H). ¹³**C NMR** (DMSO-D₆, 100 MHz): $\delta = 151.25$, 142.46, 135.78, 132.40, 130.82, 130.17, 129.44, 126.82, 124.51, 123.79, 118.97, 111.57, 21.89. **GC-MS** (M⁺) = 208.0.



5,6-dimethyl-2-phenyl-1H-benzimidazole (E11)²⁸

53 mg; 80% isolated yield; pale yellow solid. ¹H NMR (DMSO-D₆, 400 MHz): δ = 12.60 (brs, 1H), 8.10 (d, *J* = 7.2 Hz, 2H), 7.48 (t, *J* = 7.7 Hz, 2H), 7.41 (t, *J* = 7.1 Hz, 2H), 7.28-7.25 (m, 1H), 2.28 (s, 3H), 2.27 (s, 3H). ¹³C NMR (DMSO-D₆, 100 MHz): δ = 150.85, 143.03, 134.05, 131.69, 130.97, 130.44, 129.98, 129.38, 126.72, 119.45, 111.85, 20.56. **GC-MS** (M⁺) = 222.0.



5-methoxy-2-phenyl-1H-benzimidazole (E12)²⁹

55 mg; 82% isolated yield; pale yellow solid. ¹H NMR (DMSO-D₆, 400 MHz): $\delta = 12.72$ (brs, 1H), 8.09 (d, J = 7.6 Hz, 2H), 7.51-7.40 (m, 4H), 7.00 (brs, 1H), 6.79 (dd, J = 8.7 Hz, 2.0 Hz, 1H), 3.76 (s, 3H). ¹³C NMR (DMSO-D₆, 100 MHz): $\delta = 156.33$, 154.18, 151.16, 139.37, 130.85, 130.02, 129.43, 126.64, 119.09, 111.81, 101.75, 55.96. **GC-MS** (M⁺) = 224.1.



2-phenylbenzothiazole (E13)³⁰

33 mg; 52% isolated yield; pale yellow solid. ¹H NMR (CDCl₃, 400 MHz): $\delta = 8.10-8.08$ (m, 3H), 7.88 (d, J = 8.0 Hz, 1H), 7.50-7.47 (m, 4H), 7.37 (t, J = 7.8 Hz, 1H). ¹³C NMR (CDCl₃, 100 MHz): $\delta = 168.15$, 154.25, 135.17, 133.73, 131.06, 129.11, 127.66, 126.41, 125.28, 123.34, 121.71. **GC-MS** (M⁺) = 211.1.



methyl 2-phenyl-1H-benzimidazole-5-carboxylate

48 mg; 64% isolated yield; yellow solid. ¹**H NMR** (DMSO-D₆, 400 MHz): δ = 13.22 (brs, 1H), 8.23-8.09 (m, 3H), 7.80 (brs, 1H), 7.55-7.46 (m, 4H), 3.82 (s, 3H). ¹³**C NMR** (DMSO-D₆, 100 MHz): δ = 167.28, 154.79, 143.95, 139.06, 135.25, 130.98, 130.06, 129.57, 127.29, 124.25,

121.02, 113.53, 111.92, 52.51. $v_{C=0}$: 1715 cm⁻¹. **HRMS** (ESI): Calcd. for C₁₅H₁₂N₂O₂ [M+H]⁺: 253.0977; found: 253.0977.



2-phenylquinazoline (G1)³¹

58 mg; 94% isolated yield; pale yellow solid. ¹H NMR (CDCl₃, 400 MHz): δ = 9.45 (s, 1H), 8.61 (d, *J* = 7.4 Hz, 2H), 8.08 (d, *J* = 8.4 Hz, 1H), 7.91-7.86 (m, 2H), 7.59 (t, *J* = 7.9 Hz, 1H), 7.55-7.49 (m, 3H). ¹³C NMR (CDCl₃, 100 MHz): δ = 161.16, 160.57, 150.87, 138.14, 134.18, 130.69, 128.72, 128.67, 127.34, 127.20, 123.70. **GC-MS** (M⁺) = 206.1.



 $2-(4-methoxyphenyl)quinazoline (G2)^{32}$

64 mg; 90% isolated yield; pale yellow solid. ¹**H NMR** (CDCl₃, 400 MHz): δ = 9.39 (s, 1H), 8.56 (d, *J* = 8.8 Hz, 2H), 8.02 (d, *J* = 8.5 Hz, 1H), 7.87-7.83 (m, 2H), 7.54 (t, *J* = 7.4 Hz, 1H), 7.02 (d, *J* = 8.9 Hz, 2H), 3.87 (s, 3H). ¹³**C NMR** (CDCl₃, 100 MHz): δ = 161.92, 160.95, 160.46, 150.93, 134.08, 130.82, 130.29, 128.50, 127.20, 126.85, 123.40, 114.05, 55.46. **GC-MS** (M⁺) = 236.0.



2-p-tolylquinazoline (G3)³²

58 mg; 88% isolated yield; pale yellow solid. ¹H NMR (CDCl₃, 400 MHz): δ = 9.43 (s, 1H), 8.50 (d, *J* = 8.1 Hz, 2H), 8.05 (d, *J* = 8.4 Hz, 1H), 7.90-7.85 (m, 2H), 7.57 (t, *J* = 7.6 Hz, 1H), 7.33 (d, *J* = 8.0 Hz, 2H), 2.43 (s, 3H). ¹³C NMR (CDCl₃, 100 MHz): δ = 161.24, 160.51, 150.89, 140.94, 135.41, 134.11, 129.50, 128.62, 127.19, 127.10, 123.60, 21.60. GC-MS (M⁺) = 220.1.



2-(4-chlorophenyl)quinazoline (G4)³¹

61 mg; 85% isolated yield; pale yellow solid. ¹**H** NMR (CDCl₃, 400 MHz): δ = 9.41 (s, 1H), 8.56-8.54 (m, 2H), 8.04 (d, *J* = 8.8 Hz, 1H), 7.89 (d, *J* = 7.6 Hz, 2H), 7.59 (t, *J* = 7.6 Hz, 1H), 7.47 (d, *J* = 8.6 Hz, 2H). ¹³**C** NMR (CDCl₃, 100 MHz): δ = 160.60, 160.10, 150.76, 136.91, 136.60, 134.34, 129.98, 128.90, 128.68, 127.54, 127.23, 123.69. **GC-MS** (M⁺) = 240.1.



methyl 4-(quinazolin-2-yl)benzoate (G5)

47 mg; 60% isolated yield; yellow solid. ¹H NMR (CDCl₃, 400 MHz): $\delta = 9.48$ (s, 1H), 8.68 (d, J = 8.6 Hz, 2H), 8.18 (d, J = 8.6 Hz, 2H), 8.10 (d, J = 8.0 Hz, 1H) 7.95-7.92 (m, 2H), 7.64 (t, J = 7.3 Hz, 1H), 3.95 (s, 3H). ¹³C NMR (CDCl₃, 100 MHz): $\delta = 167.06$, 160.08, 160.11, 150.77, 142.21, 134.45, 131.78, 129.95, 128.86, 128.57, 127.91, 127.26, 123.85, 52.33. **v**_{C=0}: 1720 cm⁻¹. **HRMS (ESI)**: Calcd. for C₁₆H₁₂N₂O₂ [M+H]⁺: 265.0977; found: 265.0975.



2-(3-methoxyphenyl)quinazoline (G6)³³

63 mg; 88% isolated yield; pale yellow solid. ¹**H NMR** (CDCl₃, 400 MHz): δ = 9.43 (s, 1H), 8.22 (d, *J* = 7.8 Hz, 1H), 8.18-8.17 (m, 1H), 8.07 (d, *J* = 8.6 Hz, 1H), 7.90-7.85 (m, 2H), 7.58 (t, *J* = 7.5 Hz, 1H), 7.43 (d, *J* = 8.0 Hz, 1H), 7.05 (dd, *J* = 8.2, 2.6 Hz, 1H), 3.93 (s, 3H). ¹³**C NMR** (CDCl₃, 100 MHz): δ = 160.92, 160.51, 160.11, 150.81, 139.58, 134.18, 129.72, 128.75, 127.38, 127.19, 123.74, 121.25, 117.33, 113.12, 55.53. **GC-MS** (M⁺) = 236.1.



2-(benzo[d][1,3]dioxol-5-yl)quinazoline (G7)³⁴

65 mg; 87% isolated yield; pale yellow solid. ¹**H NMR** (CDCl₃, 400 MHz): δ = 9.37 (s, 1H), 8.21 (dd, *J* = 8.3, 1.7 Hz, 1H), 8.10 (d, *J* = 1.6 Hz, 1H), 8.00 (d, *J* = 9.1 Hz, 1H), 7.87-7.82 (m, 2H), 7.54 (td, *J* = 7.0, 0.8 Hz, 1H), 6.93 (d, *J* = 8.3 Hz, 1H), 6.02 (s, 2H). ¹³**C NMR** (CDCl₃, 100 MHz): δ = 160.61, 160.40, 150.83, 149.97, 148.27, 134.13, 132.60, 128.52, 127.18, 126.99, 123.60, 123.47, 108.80, 108.40, 101.51. **GC-MS** (M⁺) = 250.0.



2-(naphthalen-1-yl)quinazoline (G8)³⁵

64 mg; 83% isolated yield; pale yellow solid. ¹H NMR (CDCl₃, 400 MHz): δ = 9.58 (s, 1H), 8.69 (d, *J* = 9.3 Hz, 1H), 8.18-8.15 (m, 2H), 8.02-7.91 (m, 4H), 7.69 (t, *J* = 7.8 Hz, 1H), 7.62 (t, *J* = 7.6 Hz, 1H), 7.55-7.51 (m, 2H). ¹³C NMR (CDCl₃, 100 MHz): δ = 160.52, 158.41,

157.69, 150.68, 150.06, 134.43, 134.27, 129.72, 128.78, 128.58, 127.85, 127.25, 126.96, 125.99, 125.48, 123.24. **GC-MS** (M⁺) = 256.2.



2-(furan-2-yl)quinazoline (G9)³⁶

44 mg; 74% isolated yield; pale yellow solid. ¹H NMR (CDCl₃, 500 MHz): $\delta = 9.35$ (s, 1H), 8.07 (d, J = 6.6 Hz, 1H), 7.88-7.85 (m, 2H), 7.67 (d, J = 1.0, 1H), 7.56 (t, J = 5.8 Hz, 1H), 7.43 (t, J = 3.1 Hz, 1H), 6.60-6.58 (m, 1H). ¹³C NMR (CDCl₃, 100 MHz): $\delta = 160.83$, 154.18, 152.58, 150.51, 145.44, 134.62, 128.48, 127.37, 123.46, 114.19, 112.42. GC-MS (M⁺) = 195.9.



2-(thiophen-2-yl)quinazoline (G10)³⁷

50 mg; 78% isolated yield; pale yellow solid. ¹H NMR (CDCl₃, 400 MHz): $\delta = 9.32$ (s, 1H), 8.13 (d, J = 3.7 Hz, 1H), 7.98 (d, J = 8.9 Hz, 1H), 7.87-7.83 (m, 2H), 7.56-7.49 (m, 2H), 7.18-7.16 (m, 1H). ¹³C NMR (CDCl₃, 100 MHz): $\delta = 160.60$, 157.95, 150.71, 143.94, 134.43, 130.03, 129.32, 128.46, 128.27, 127.35, 127.07, 123.46. GC-MS (M⁺) = 212.1.



2-(pyridin-3-yl)quinazoline (G11)³⁸

43 mg; 68% isolated yield; pale yellow solid. ¹H NMR (CDCl₃, 500 MHz): δ = 9.81 (s, 1H), 9.48 (s, 1H), 8.86 (dt, *J* = 7.9, 1.7 Hz, 1H), 8.73 (d, *J* = 3.5 Hz, 1H), 8.10 (d, *J* = 8.4 Hz, 1H) 7.96-7.92 (m, 2H), 7.65 (t, *J* = 7.9 Hz, 1H), 7.46-7.44 (m, 1H). ¹³C NMR (CDCl₃, 100 MHz): δ = 160.75, 159.29, 151.25, 150.74, 150.32, 135.91, 134.49, 133.66, 128.76, 127.88, 127.29, 123.93, 123.51. **GC-MS** (M⁺) = 207.1.

12. Computational studies

All the calculations were performed using the Gaussian 09 package.³⁹ Full geometry optimization followed by frequency calculations on the stationary points were carried out to ascertain the nature of the stationary points as minima or first order saddle point. Hybrid functional, M062X was used with the LANL2DZ basis set⁴⁰ for Ir and 6-31G** basis set⁴¹ for non-metal elements. The transition states (TS) were further confirmed by performing intrinsic reaction coordinate (IRC) calculation using the same method. Solvent effect was incorporated using the conductor-like polarizable continuum model (CPCM) with water as solvent.⁴²

13. Cartesian coordinates and statistical thermodynamic analysis

Benzaldehyde



ion=		0.111133
to Gibbs Fre	e Energy=	0.080631
		-345.42344716
ent)		-345.42966126
1.98948800	0.47133900	0.0000300
2.82858600	-0.39799200	-0.0000400
2,27794100	1,54399900	0.0000000
0.52849700	0.21752900	0.0000200
-0.36149800	1.29085300	0.0000100
0.05113800	-1.09565300	0.00000200
-1.73300900	1.05561500	-0.00000100
0.02377600	2.30787200	0.0000000
-1.31704400	-1.32941400	0.0000000
0.77111100	-1.90820000	0.00000400
-2.20742700	-0.25387800	-0.00000100
-2.42937400	1.88750400	-0.00000300
-1.69607300	-2.34611900	-0.00000100
-3.27694700	-0.43945800	-0.00000200
	<pre>ion= to Gibbs Fre ent) 1.98948800 2.82858600 2.27794100 0.52849700 -0.36149800 0.05113800 -1.73300900 0.02377600 -1.31704400 0.77111100 -2.20742700 -2.42937400 -1.69607300 -3.27694700</pre>	<pre>ion= to Gibbs Free Energy= ent) 1.98948800 0.47133900 2.82858600 -0.39799200 2.27794100 1.54399900 0.52849700 0.21752900 -0.36149800 1.29085300 0.05113800 -1.09565300 -1.73300900 1.05561500 0.02377600 2.30787200 -1.31704400 -1.32941400 0.77111100 -1.90820000 -2.20742700 -0.25387800 -2.42937400 1.88750400 -1.69607300 -2.34611900 -3.27694700 -0.43945800</pre>

Benzyl alcohol



Zero-point correction=	0.135266
Thermal correction to Gibbs Free Energy=	0.104027
SCF done	-346.62490953
SCF done (for solvent)	-346.63115488

С	1.91991400	0.57487500	0.23926700
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С	-1.36074200	-1.34082900	0.05413700
Н	0.72936900	-1.84861600	0.24832700
С	-2.28779400	-0.30933900	-0.07151200
Н	-2.56650400	1.82025500	-0.21210700
Н	-1.69723900	-2.37261000	0.08064500
Н	-3.34708800	-0.53340300	-0.14582500
Н	2.18535500	0.69536100	1.29523200
Н	2.50137400	-0.62673900	-1.15954800

I1



Zero-point correction=	0.386466
Thermal correction to Gibbs Free Energy=	0.334557
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SCF done (for solvent)	-1043.35927815

С	2.31307100	3.60325700	-0.12140000
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С	2.93107300	-2.82038400	-0.00514100
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С	-1.80626900	-2.24433000	1.81163400
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Н	-1.07803300	-2.46465800	-2.13329900
Н	-2.83492000	-2.66545100	-1.94525200
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Н	0.45714700	3.36547700	1.04663600
Н	0.18531000	3.47433500	-0.70394800

TS1_{outer}



Zero-point correction=	0.516675
Thermal correction to Gibbs Free Energy=	0.456151
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SCF done (for solvent)	-1389.97607277

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С	-0.11189100	-2.41295000	-0.74641600
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С	-2.65741400	-2.29224200	-1.25204500
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Н	2,61877400	-2.01373700	-0.45935800
Н	-0 41194600	-1 56687800	3 62605700
н	2 50154200	-3 50957600	0 48493100
Н	-3 41976700	3 57566600	-0.50299100
Н	0 36023100	-3 13720300	3 40211500
Н	1 37059000	-2 41280800	-2 29672600
н	-3 77474800	3 25039800	-2 97457600
н	-2 53982000	-1 41058800	2 84445400
н	-2 43528400	1 51489700	-4 16571700
н	0 33936900	-3 85324400	-2 24390900
н	0 62958200	-0 00219100	-2 66700800
н	-0.28911800	-2 31763200	-2 88296500
н	-3 05562500	-3 01136300	2 30779500
н	-3 56896800	-1 57155500	1 41621900
н	-2 39044400	-1 79979600	-2 19044100
н ц	-3 51881200	-1 78096600	-0 81851900
н ц	-2 94936500	-3 32244500	-1 47989300
C	1 80151600	0 86487800	-1 35595300
ч	1 05650300	0.33781400	-0 10664300
н	1 14967600	1 74564600	-1 34809000
C	3 12080900	1 01368400	-0 71587900
C	3 29953700	1 99052000	0 26924300
C	4 18852400	0 20270000	-1 10878500
C	4 54129900	2 15185600	0 86616400
ч	2 45760100	2 61342700	0 56428400
C	5 43241800	0 36955100	-0.50920900
C	5 60739800	1 33835200	0.30311700
ч	4 68639800	2 91071900	1 62719700
п п	6 26877600	-0 24882300	-0.8163/200
п u	6 58017300	-0.24002300	-0.01034200
0	1 67259600	1.4073300	-2 31304400
U	1 07255000 1 03751600	-0 52895600	-2.JIJU4400 _1 89586/00
11 U	4.05751000	1 28202200	-1.03000400
11		1.20302200	2 10201000
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п		3.33840800	J.122000UU
п	-I.0/3226UU	Z. JIJ J0000	4.UZ3331UU



 Zero-point correction=
 0.524781

 Thermal correction to Gibbs Free Energy=
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 SCF done
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 SCF done (for solvent)
 -1390.01716087

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Н	-1.90282700	0.00817800	-3.29186900
Н	-4.12294300	2.86679700	-0.16223200
Н	3.30996600	-0.17995000	-2.70405200
Н	-3.49819300	0.76480900	-3.14991600

Н	-2.98950300	1.87332300	2.52817000
Н	3.42998800	-2.69783600	-2.74500800
Н	-1.72102900	-2.08325900	-2.48174600
Н	1.90590800	-4.00973100	-1.28363400
Н	-4.14242700	0.54968100	2.76389700
Н	0.21465300	-1.57143300	1.67160500
Н	-2.42291200	0.29623300	3.10528800
Н	-3.30996900	-2.68312000	-1.97871800
Н	-1.85855900	-3.08367500	-1.02757700
Н	-2.39513800	-1.93970000	2.54127400
Н	-2.10397100	-3.01979100	1.17497000
Н	-3.77306900	-2.61299300	1.65461600
0	0.05849900	-0.64419400	1.97807400
С	1.16129000	-0.10641100	2.71660300
Н	0.83954400	0.88455600	3.04708000
Н	1.30411100	-0.73207600	3.60505500
С	2.44356400	-0.00962300	1.92830900
С	3.03231300	1.22925600	1.67761900
С	3.07095000	-1.16876700	1.46197800
С	4.22458000	1.31568700	0.95866900
С	4.25460100	-1.08575800	0.73691000
Н	2.62717900	-2.14260000	1.66239600
С	4.83396000	0.15781900	0.48279300
Н	4.68965000	2.28396300	0.79359600
Н	4.72872600	-1.99221600	0.37463000
Н	5.76851000	0.21942200	-0.06580600
Н	2.56932200	3.58422400	-0.04686900
Н	1.81441900	4.31374200	-1.48763100
Н	0.30376600	3.36644200	0.74194400
Н	-0.35336900	3.47227000	-0.90297900
Н	2.56274500	2.13202000	2.06396500

TS1_{inner}



Zero-point correction=	0.518191
Thermal correction to Gibbs Free Energy=	0.457493
SCF done	-1389.85172227
SCF done (for solvent)	-1389.91274507

С	-1.67607900	-3.18426900	1.76878400
С	-1.74260900	-1.66650000	1.51815000
С	0.26168300	-2.41528000	0.87054800

С	-1.59752200	3.01760200	2.09982900
С	0.47506500	0.88700400	3.06547300
С	1.59878200	-2.34980000	0.26036300
С	2.50163600	-3.38812300	0.23985900
С	-0.35104800	2.69922500	1.34351000
С	0.53414300	1.53878800	1.71510200
С	3.68863500	-3.22311900	-0.49232400
С	2.90836500	-1.05857100	-1.15129400
С	3.88710200	-2.06810100	-1.21234000
С	0.10749600	3.20900100	0.17988300
С	-0.47593800	4.28420300	-0.67682700
С	1.69558400	1.61578300	0.89021800
С	3.06900200	1.12880900	1.23785800
С	1.29557900	2.39993500	-0.25292100
С	2.22458400	2.94975700	-1.30416700
Ir	0.33425200	0.32791600	-0.14214300
N	-0.22833300	-3.42884300	1.63197000
N	-0.53930200	-1.41390800	0.71075900
N	1.83949500	-1.16097700	-0.36155900
0	3.06358400	0.02387800	-1.89689800
Н	0.11188500	-4.37210600	1.51210400
Н	-2.30124200	2.17438100	2.05323000
Н	-0.55928500	0.69012500	3.36005900
Н	-2.10074600	3.89723200	1.69502200
Н	1.02398800	-0.05844700	3.08449700
H	-1.39191200	3.20562500	3.15858400
Н	2.29772600	-4.31277100	0.76603900
Н	0.90724300	1.54809700	3.82541200
Н	-1.35491500	4.73751400	-0.21572800
Н	4.42647900	-4.01735700	-0.52072600
Н	3.03787500	0.19448200	1.80457100
Н	4.75373200	-1.90685600	-1.84150600
H	0.25203400	5.08065700	-0.86392600
H	2.23549400	0.54169200	-1.90067000
H	-0.77010400	3.89305700	-1.65855200
H	3.55821600	1.88387400	1.86185700
H	3.68914100	0.98377100	0.35089800
H 	1.74415300	2.99206700	-2.28667600
H	3.15036700	2.3/832200	-1.39290800
H	2.50578800	3.9/546500	-1.04113300
0	-0.55383900	-0.19669200	-1.99819800
C	-1.53345700	0.59228200	-1.62185/00
H	-1.20986300	1.09/23600	-0.38034500
H	-1.59984500	1.58120000	-2.1025/700
C	-2.85154100	-0.04549600	-1.28945800
	-3.8/346000	0.69405500	-0.69303000
C	-3.03503700	-1.39/5/100	-1.5/591500
C	-5.07227900	0.0/432/00	-0.35775000
	-4.24217200	-2.01269400	-1.25148/00
п	-Z.ZZ6Z33UU	-1.941324UU	-2.05239000
U U	-J.ZJJJ9/UU _5 06615500	-T.20173ANA	-U.034336UU
п	UUCCL000.C-	0.04400000	U.IIZ483UU
л u	-4.39/329UU	-3.U0U0/UUU _1 76200500	-1.400933UU
n u	-0.19443200 _0.00017500	-1./0209000 _3.7200000	-U.JOLUOYUU
п	-2.2321/300	-3./3222200	U. YYX / 13UU
п	-2.03396300	-3.4/3U61UU	Z./J048ZUU

Н	-2.64742500	-1.35319600	0.99646800
Н	-1.64993200	-1.10139200	2.45324200
Н	-3.72365700	1.75126000	-0.48093400





Zero-point correction=	0.407485
Thermal correction to Gibbs Free Energy=	0.355433
SCF done	-1044.49636216
SCF done (for solvent)	-1044.55570476

С	-2.38690100	3.54563200	-0.30136700
С	-0.92056300	3.06949000	-0.17708400
С	-2.26863400	1.29062100	-0.01018900
С	3.19353400	0.08179500	-2.01258600
С	2.59017700	2.48277900	0.01724200
С	-2.60796100	-0.13466100	0.04071700
С	-3.86806300	-0.64690300	0.26021600
С	2.53392100	-0.03174900	-0.67454800
С	2.24518900	1.04142900	0.23009000
С	-4.02449800	-2.04015800	0.23561500
С	-1.68977400	-2.23779200	-0.25657500
С	-2.93831800	-2.84353100	-0.03402400
С	2.12051900	-1.25580300	-0.04030700
С	2.44857600	-2.63676100	-0.53339800
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С	1.48302200	1.24954800	2.73314200
С	1.70507600	-0.92732600	1.32353400
С	1.24311700	-1.94548800	2.31830100
Ir	0.39178600	0.07087700	-0.33778000
N	-3.12532100	2.33858400	0.11271500
N	-1.02859200	1.60744300	-0.17961400
N	-1.52069500	-0.91788400	-0.18019400
0	-0.64452600	-2.99804500	-0.55336000
Н	-4.09311400	2.23316800	-0.15683800
Н	3.00626100	1.05625200	-2.46608600
Н	2.51862100	2.75561800	-1.03751400
Н	2.82999500	-0.68464000	-2.69883800

Н	1.92953400	3.13977900	0.58756700
Н	4.27553600	-0.04169600	-1.90001800
Н	-4.70756900	0.00869000	0.45838300
Н	3.61489700	2.67685000	0.35017100
Н	2.31756500	-2.72344100	-1.61518800
Н	-4.99835600	-2.48308200	0.41225000
Н	0.96264400	2.18113900	2.49757400
Н	-3.00723300	-3.92251300	-0.09299200
Н	3.49468600	-2.86768800	-0.30869100
Н	0.08593300	-2.41338100	-0.83567700
Н	1.83760000	-3.40138800	-0.04745300
Н	2.40889000	1.51472600	3.25389300
Н	0.85920800	0.68160000	3.42492200
Н	0.61878500	-2.70461900	1.83877400
Н	0.66958800	-1.48880100	3.12619600
Н	2.10392300	-2.45715700	2.76098000
Н	0.09840000	0.12718500	-1.87471200
Н	-0.28352900	3.40768200	-0.99633500
Н	-0.47248500	3.39101300	0.77008000
Н	-2.63636000	3.81208900	-1.33366300
Н	-2.61397000	4.38789400	0.35198400

TS2



Zero-point correction=0.400392Thermal correction to Gibbs Free Energy=0.348665SCF done-1044.43918354SCF done (for solvent)-1044.50159852

С	-2.25932200	3.56089200	-0.22542200
С	-0.96720800	2.99018400	-0.85364800
С	-2.19353300	1.28653500	-0.06090200
С	3.48806800	0.51369600	-1.56161800
С	2.31808900	2.60376400	0.54368300
С	-2.51730100	-0.12943200	0.15779000
С	-3.76217600	-0.67451000	0.32928700
С	2.63284500	0.19610200	-0.37647000
С	2.11145000	1.12415200	0.56655900
С	-3.86800900	-2.08416600	0.21774500

С	-1.52760700	-2.20968000	-0.42084800
С	-2.79249400	-2.83976400	-0.16686000
С	2.25279300	-1.14395300	0.06057200
С	2.63858200	-2.42750300	-0.60285100
С	1.36438600	0.39477200	1.57899700
С	0.71670800	0.98646500	2.79036100
С	1.50718000	-1.01838900	1.27532400
С	0.91556200	-2.15543000	2.04266000
Ir	0.43486500	-0.00079600	-0.31797000
Ν	-3.04448700	2.33374300	-0.03355800
N	-0.99247400	1.57116800	-0.47554200
N	-1.41989800	-0.89664200	-0.06444800
0	-0.53028800	-2.74281900	-0.98344100
Н	-3.89338000	2.31925000	0.51131800
Н	3.30611700	1.52702900	-1.92275400
Н	2.34827900	2.99046100	-0.47699700
Н	3.29559200	-0.17744000	-2.38340200
Н	1.53552700	3.12465100	1.09853800
Н	4.54502400	0.43178600	-1.28920800
Н	-4.63825500	-0.05629600	0.48317500
Н	3.27434500	2.84010400	1.02140100
Н	2.84391300	-2.27321500	-1.66367400
Н	-4.83026000	-2.55940300	0.37680300
Н	0.24520900	1.94345500	2.55393600
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Н	3.54392200	-2.82772400	-0.13642000
Н	1.83690800	-3.16295100	-0.51935500
Н	1.45622000	1.15546300	3.57899900
Н	-0.05343300	0.32088100	3.18478700
Н	0.61613900	-2.95446400	1.36073200
Н	0.03761200	-1.83683200	2.60754200
Н	1.65276400	-2.55295300	2.74661300
Н	0.05110900	-1.46107600	-1.60874800
Н	0.19394600	-0.63793300	-1.99659800
Н	-2.77360100	4.26102700	-0.88380400
Н	-2.06516000	4.04777600	0.73725300
Н	-0.98292700	3.06082500	-1.94652300
Н	-0.06755500	3.48659900	-0.48957800
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15. Copies of ¹H and ¹³C NMR spectra of the products







8.7263 8.0672 8.0672 8.0672 8.0428 8.0126 8.0126 8.0126 8.0126 8.01412 9.9412

























180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 f1 (ppm)



















-9.0945 -9.0945 (8.1101 (7.5284 7.54090 7.75090 7.75090 7.74652 7.74652 7.74652 7.74652 7.74652 7.7463 7.7463 7.7463 7.7463 7.7463 7.75647 7.75647 7.75647 7.75647 7.75647 7.75647 7.75647 7.75647 7.75647 7.75647









$\begin{array}{c} -9.3109\\ 8.1825\\ 8.16285\\ 8.16785\\ 8.1004\\ 8.10$
















































































190 180 170 160 150 140 130 120 110 100 90 f1 (ppm)











G10



