# Is hydroxypyridonate 3,4,3-LI(1,2-HOPO) a good competitor of fetuin for uranyl metabolism?

## **Supplementary Information**

#### I. Determination of apparent affinity constant

For each HOPO concentration, the free part of  $UO_2^{2+}$  has been quantified by measuring the corresponding SPR signals in duplicates. From these signals, and thanks to a calibration curve (SPR signal = f ( $UO_2^{2+}$ ) previously established within the same experiment, the unbound  $UO_2^{2+}$  could be calculated. A binding isotherm has been obtained by plotting the response SPR signal (Resonance Units, RU) corresponding to the unbound  $UO_2^{2+}$  versus the ligand concentration (here log [HOPO]). This gives a sigmoid curve that could be fitted to a four parameter logistic curve to calculate the apparent  $K_D$ . These parameters are the maximum response corresponding to 50 nM  $UO_2^{2+}$ ), the minimum response corresponding to highest HOPO concentration leading to total  $UO_2^{2+}$  consumption, the slope and the inflexion point (see Table below and reference [15] for experimental details). This point corresponds to 50 % of  $UO_2^{2+}$  occupancy, thus giving the apparent  $K_D$ .



**Figure S1.** Determination of the apparent  $K_D$  of HOPO for  $UO_2^{2+}$  by SPR. Uranyl (0.5x10<sup>-7</sup> M) in TRIS buffer and  $Na_2CO_3$  (5x10<sup>-6</sup> M) was contacted with varied HOPO concentrations from 0.05 to 250x10<sup>-9</sup> M and in duplicates (diamonds). The curve of the binding isotherm (continuous black line) is fitted to a four parameter logistic curve. The parameters of this curve are given in the table

1 below. The apparent  $K_D$  of HOPO is deduced from the HOPO concentration giving 50 % of signal (i.e 50% occupancy).

Table S1. Best fit parameters for the binding isotherm of uranyl with HOPO

MIN (RU)	$3.7 \pm 0.2$	
MAX (RU)	18.1 ± 0.3	K <sub>D</sub> (50% signal)
Inflexion point	-7.49 ± 0.03	32 ± 0.3x10 <sup>-9</sup> M
(log)		
Slope	2.07	

#### II. Titration of HOPO by uranyl; UV-Vis spectroscopy

In order to further define the stoichiometry of the  $UO_2^{2^+}$ -HOPO complex, titration of HOPO by uranyl has been performed at 22 ± 2 °C. The evolution of the HOPO UV-Vis spectra is presented (Figure S2) within the 250-500 nm range. The aquo-ion of  $UO_2^{2^+}$  has a negligible absorption between within this range. HOPO at pH 7.4 presents a maximum absorbance at  $\lambda$  = 328 nm ( $\epsilon$  = 20.3 L.mol<sup>-1</sup>.cm<sup>-1</sup>).

Changes on the UV-Vis spectra were followed for successive additions of 0.1 equivalents of  $UO_2^{2+}$  from 0 up to 2  $UO_2^{2+}$  equivalents, and for additions of 0.5 equivalents up to 4 equivalents. The first successive additions from 0 to 2  $UO_2^{2+}$  equivalents led to a decrease of the 328 nm signal associated with a blue shift of the wavelength from 328 to 317 nm for 1 equivalent, and then to 314 nm for 2 equivalents. The ligand band at 328 nm is blue shifted upon the successive  $UO_2^{2+}$  addition because the  $\pi$  -  $\pi^*$  transition is affected by the complexation with uranyl. After this 2:1 ratio, the absorption increase at 314 and 328 nm remain proportional to  $UO_2^{2+}$  additions without modification of the wavelength maximum.

The ratio of absorbance at 328 over 314 nm decreases with a single slope up to  $2 \text{ UO}_2^{2^+}$  equivalents where a clear breakpoint is observed, followed then by a plateau. (Figure S2, insert). Knowingly, HOPO is an octadentate chelator composed of a spermine backbone coupled to four hydroxypyridinone groups for metal binding and the nature of  $\text{UO}_2^{2^+}$ -ligand complex formation comes to occupy in equatorial plane, that define HOPO as having four potential bidentate hydroxypyridinone rings for uranyl complexation.



**Figure S2.** UV-vis titration of HOPO by  $UO_2^{2+}$ . Evolution of HOPO spectra (5x10<sup>-5</sup> M in TRIS buffer at 22 ± 2 °C upon  $UO_2^{2+}$  additions. Successive additions of 0.1  $UO_2^{2+}$  equivalents from 0 (continuous black line) to 1 (continuous blue line) and 2 equivalents (continuous red line); then additions of 0.5 equivalents from 2 to 3 (continuous green line) and 4 equivalents (dashed green line). Insert: evolution of the absorbance ratio at 328 vs 314 nm (wavelengths corresponding to the 2 black arrows on the spectra) upon  $UO_2^{2+}$  additions.

## III. EXAFS data analysis



**Figure S3.** Moduli and imaginary parts of the Fourier tranforms of the EXAFS spectra of  $UO_2^{2+}$ -HOPO and  $UO_2^{2+}$ -fetuin (U- for clarity in the Figure) at pH = 6.0. Black curve = experiment, black dots = fit of the imaginary part, empty dots = fit of the modulus.



**Figure S4.** Comparison between the EXAFS spectra of U-HOPO, U-fetuin and U-HOPO-fetuin (enlarged spectral area between 4 and 9 Å<sup>-1</sup>).

# IV. DFT calculations

Table S2. xyz coordinates of the two calculated structures presented in Fig. 3 (in Å)

• Without water molecule:

С	2.08799200	1.58084900	3.51463700
С	1.63408600	2.15470100	4.68312700
С	0.80984800	1.39287300	5.55370900
С	0.47790300	0.08977400	5.25779800
С	0.94760500	-0.55540600	4.06864200
Ν	1.72688100	0.28486500	3.23406900
Н	0.44648700	1.84500800	6.46917500
Н	1.93524100	3.16229300	4.93381800
Н	-0.13757800	-0.50680700	5.91868500
0	2.29907600	-0.29669000	2.06035100
0	0.72307600	-1.75999500	3.70878000
С	3.07532200	2.30823100	2.62658800
0	4.24170700	2.49848900	3.10597400
Ν	2.71132600	2.78994500	1.41649500
С	1.32514300	2.73974800	0.89068900
н	1.40020400	2.71993900	-0.20057000
н	0.86147300	1.79459400	1.17705800
С	-4.22008000	-2.13835500	1.26281300
C	-4.61027900	-3.17719900	2.10187600
C	-3.84681600	-4.35625100	2.18168900
C	-2.68647900	-4.48809900	1.43371200
C	-2.27536300	-3.43748500	0.59020700
N	-3 07359300	-2 30560700	0 51831700
н	-4 16032100	-5 15798900	2 83863900
н	-5 50595500	-3 03557100	2 68814300
н	-2 06234100	-5 36998500	1 48201200
0	-2 61330300	-1 33908400	-0 38832100
0	-1 20006100	-3 41323500	-0 15570300
C C	-5 04847700	-0.88152300	1 23058400
0	-6 06047100	-0.81676200	2 00749600
N	-4 70168600	0.01070200	0 39197000
н	-3.86150600	-0.02822200	-0 177/3000
C C	-5.00100000	1 26851/00	0.17743000
с ц	-5.47508500	1 71862400	-0 70757500
ц	-5.43317000	1.71802400	0.58202400
C C	-0.30330200	2 80677700	-2 41409000
C C	-1.51351000	2.80077700	-2.41409000
C C	0 27820600	2 50228400	4 15274900
C C	0.37820000	2 21606800	4.13374800
C C	0.32947300	2.31090800	-4.01932700
N	1 22/62200	1.54595900	-3.06104500
	-1.23403200	1.02303100	-2.33307100
	-0.00174300	4.24130000	-4.87990000
n u	-2.07545700	4.00918100	-5.45790200
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0	-1.02083100	0.00049900	-1.4/230300
C C	0.47991200	0.13010300	-2.02083200
	-3.12100800	3.0851/300	-1.2281/000
	-4.21241900	3.34134400	-2.1/445/00
IN	-3.01084000	3.1/08/400	-0.21020400

С	-4.18957200	3.61567400	0.58109600
Н	-3.83205300	4.33820400	1.32124700
Н	-4.85804400	4.14074100	-0.10184400
С	4.45487100	-1.62706800	-0.79939000
С	5.31520900	-2.65703100	-0.43448500
С	4.82147800	-3.96332600	-0.25771700
С	3.47094100	-4.23399700	-0.43697500
С	2.58831700	-3.19648600	-0.80357700
Ν	3.12706300	-1.92972500	-0.98983300
Н	5.49696100	-4.75913900	0.03157600
Н	6.35634600	-2.41342400	-0.28079000
Н	3.05727500	-5.22339900	-0.29443400
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Н	5.46665700	2.14053000	-1.91491600
С	-4.95570800	2.47988300	1.28273200
Н	-4.34121900	2.02233200	2.06735000
Н	-5.80705800	2.94581600	1.79419500
С	-1.75128300	2.94882600	0.54568000
Н	-2.04455800	2.75616400	1.58187200
Н	-1.28551100	2.03179000	0.18051000
С	-0.79406500	4.15345200	0.49538100
Н	-1.31424500	5.05165500	0.85068000
Н	-0.51401500	4.35464900	-0.54687200
С	0.47922500	3.94981400	1.33923700
Н	0.22507300	3.83902900	2.40147900
Н	1.08552700	4.86070700	1.26047800
С	3.69007800	3.65795500	0.69856800
Н	4.12112100	4.35322700	1.42460900
Н	3.11355200	4.24586000	-0.02024700
С	4.85647200	2.93557200	-0.00758200
Н	5.35820700	2.26317500	0.69433600
Н	5.58140700	3.72323800	-0.25225800
U	-0.27323700	-1.46010000	-1.24617300
0	-0.84197815	-2.19961017	-2.79955712
0	0.21744205	-0.80597229	0.38090782
Н	1.53450100	-0.64265900	1.50074700

• With two water molecules

С	1.82058100	-1.37471800	2.67562900
С	1.75038500	-1.36351800	4.04884200
С	0.73385700	-2.11604000	4.69568600
С	-0.16262900	-2.86653700	3.97087400
С	-0.11251100	-2.92566700	2.53809200
Ν	0.90697200	-2.12418500	1.97460000
Н	0.67265500	-2.10076400	5.77746400
Н	2.46592900	-0.78389400	4.61530800
Н	-0.93272400	-3.45447200	4.45252200
0	1.11475500	-2.20357400	0.56397400
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Н	-1.34662700	5.81615500	-0.54821800
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н	8 85//7/00	2.14455400	-0 501/2800
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С	3.77968500	1.19385500	0.58174200
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Н	3.57205800	2.26189700	0.67822500
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Н	-1.97566600	-2.98549200	-4.16901700
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